

AD-782 023

A COMPUTER CODE TO PREDICT THE EFFECTS
OF ELECTROPHILIC LIQUID INJECTION ON RE-
ENTRY PLASMA SHEATH PROPERTIES

Harold S. Pergament, et al

AeroChem Research Laboratories, Incorporated

Prepared for:

Air Force Cambridge Research Laboratories

January 1974

DISTRIBUTED BY:

NTIS

National Technical Information Service
U. S. DEPARTMENT OF COMMERCE
5285 Port Royal Road, Springfield Va. 22151

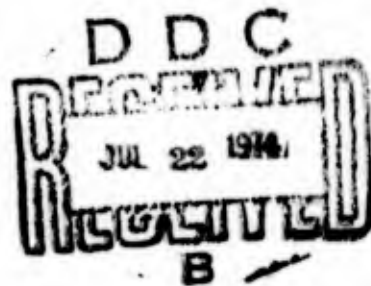
AD 782023

A COMPUTER CODE TO PREDICT THE EFFECTS OF ELECTROPHILIC LIQUID INJECTION ON RE-ENTRY PLASMA SHEATH PROPERTIES

by

Harold S. Pergament
Chung-jen Kau

AEROCHEM RESEARCH LABORATORIES, INC.
P.O Box 12
Princeton, New Jersey 08540



January 1974

Contract No. F19628-73-0045
Project No. 5635
Task No. 563504
Work Unit No. 56350401

Final Report for Period 1 October 1972 - 30 November 1973

Contract Monitor: Sheldon B. Herskovitz, Microwave Physics Laboratory

Approved for public release; distribution unlimited.

Prepared for

AIR FORCE CAMBRIDGE RESEARCH LABORATORIES
Air Force Systems Command
United States Air Force
Bedford, Massachusetts 01730

UNCLASSIFIED

AD-782023

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

REPORT DOCUMENTATION PAGE		READ INSTRUCTIONS BEFORE COMPLETING FORM
1. REPORT NUMBER AFCRL-TR-74-0074	2. GOVT ACCESSION NO.	3. RECIPIENT'S CATALOG NUMBER
4. TITLE (and Subtitle) A COMPUTER CODE TO PREDICT THE EFFECTS OF ELECTROPHILIC LIQUID INJECTION ON RE-ENTRY PLASMA SHEATH PROPERTIES		5. TYPE OF REPORT & PERIOD COVERED Final, 1 October 1972 to 30 November 1973
		6. PERFORMING ORG. REPORT NUMBER AeroChem TP-308
7. AUTHOR(s) Harold S. Pergament Chung-jen Kau		8. CONTRACT OR GRANT NUMBER(s) F19628-73-C-0045
9. PERFORMING ORGANIZATION NAME AND ADDRESS AeroChem Research Laboratories, Inc. P.O. Box 12 Princeton, New Jersey 08540		10. PROGRAM ELEMENT, PROJECT, TASK AREA & WORK UNIT NUMBERS 61102F, 5635, 563504, 56350401
11. CONTROLLING OFFICE NAME AND ADDRESS Air Force Cambridge Research Laboratories (LZ) L.G. Hanscom Field Bedford, Massachusetts 01730		12. REPORT DATE January 1974
		13. NUMBER OF PAGES 143
14. MONITORING AGENCY NAME & ADDRESS (if different from Controlling Office)		15. SECURITY CLASS (of this report) Unclassified
		15a. DECLASSIFICATION/DOWNGRADING SCHEDULE
16. DISTRIBUTION STATEMENT (of this Report) A - Approved for public release; distribution unlimited.		
17. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from Report)		
18. SUPPLEMENTARY NOTES		
19. KEY WORDS (Continue on reverse side if necessary and identify by block number) Re-entry Plasma Sheaths, Electrophilic Liquid Injection, Liquid Jet Breakup, Electron/Ion Recombination, Electron Attachment/Detachment, Computer Codes, Shock Layer Properties		
20. ABSTRACT (Continue on reverse side if necessary and identify by block number) A detailed description of a computer code to predict the influence of electrophilic liquid injection on re-entry plasma sheath properties is presented. The code requires input data on the liquid injection system parameters (i.e. electrophilic liquid properties, injection velocity, number of orifices, etc.) and the plasma sheath (shock layer) properties for no material addition, and predicts altered shock layer flow properties and species concentrations. Of primary interest is the ability of the		

DD FORM 1473 1 JAN 73 EDITION OF 1 NOV 69 IS OBSOLETE

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE(When Data Entered)

Block 2C

code to predict reductions in shock layer electron concentrations, which determine the effectiveness of the injectant in alleviating re-entry blackout.

The physical processes incorporated into the model include: flash vaporization of the liquid jet, penetration and breakup of the jet, droplet acceleration and vaporization, diffusion of the spray/vapor cloud into the shock layer, vapor decomposition, heterogeneous and homogeneous charged species recombination and homogeneous electron attachment/detachment kinetics. The influence of the electrophilic liquid and vapor is assumed to be limited to the volume contained within a droplet-vapor-air (DVA) tube, which extends in a streamwise direction from the location of liquid jet breakup to the end of the body. Flow properties and species concentrations are assumed to be constant over the tube cross section, so that only the streamwise influence of the droplets and vapor on integrated electron and ion concentrations is predicted.

This report includes the equations utilized to describe the model and serves as a Program User's Manual, incorporating instructions for preparing input data, sample input and output and Fortran listings for two versions of the code (specialized and generalized chemistry versions).

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE(When Data Entered)

ACKNOWLEDGMENTS

The authors gratefully acknowledge the efforts of J.T. Kelly in helping to formulate the model and for critically reviewing the computer code and J.S. Weber for the programming assistance needed to develop a working code.

We also thank the personnel at the AFCRL Microwave Physics Laboratory, particularly S.B. Herskovitz, J.F. Lennon and D.T. Hayes for their helpful suggestions, encouragement and support during the course of this work.

TABLE OF CONTENTS

	<u>Page</u>
I. INTRODUCTION	5
II. MATHEMATICAL MODEL	7
A. Phenomenological Description	7
B. Model Formulation	7
III. SHOCK LAYER PROPERTIES	9
IV. JET SHOCK, FLASH VAPORIZATION AND LIQUID JET BREAKUP	11
A. Jet Shock	11
B. Flash Vaporization	12
C. Liquid Jet Breakup	13
1. Aerodynamic Breakup ($We > We_{cr}$)	13
2. Vapor Pressure Breakup ($We < We_{cr}$)	15
a. $T_{l,o} < T_{sh}$	16
b. $T_{l,o} > T_{sh}$	16
V. INITIAL CONDITIONS FOR DVA TUBE CALCULATIONS	17
A. Lateral Spread at Breakup	17
B. Shock Layer Air/Flashed Vapor Mixing	19
VI. DVA TUBE ANALYSIS	20
A. Governing Equations	20
1. Vapor/Air Mixture	20
2. Droplets	22
B. Droplet Vaporization	23
C. Droplet Drag Coefficient	23
D. Normal Penetration and Entrainment Area	25
VII. CHEMICAL REACTION MECHANISMS	26
A. Specialized Chemistry Version (Version C)	26
B. Generalized Chemistry Version (Version D)	28
VIII. THERMODYNAMIC DATA	31
A. Program Input	31
B. Inversion of T from h(T)	32
IX. NUMERICAL TECHNIQUES	32
X. PROGRAM OPERATION	34

	<u>Page</u>
XI. PROGRAM INPUT DATA	
A. Version C - Specialized Chemistry	34
B. Version D - Generalized Chemistry	39
REFERENCES	45
NOMENCLATURE	47
Figure 1 Schematic of liquid injection into a re-entry plasma sheath	6
Figure 2 Schematic of Droplet-Vapor-Air (DVA) tube model	9
Figure 3 Significant dimensions for lateral spread calculations	18
APPENDIX A LATERAL SPREAD OF SPRAY	A-1
APPENDIX B SAMPLE INPUT DATA	B-1
1. Version C	
2. Version D	
APPENDIX C SAMPLE OUTPUT, Version C	C-1
APPENDIX D SAMPLE OUTPUT - EXPLANATIONS, Version C	D-1
APPENDIX E FORTRAN LISTINGS	E-1
1. Version C	
2. Version D	

I. INTRODUCTION

It has been established that the injection of electrophilic liquids into re-entry plasma sheaths can significantly reduce electron and positive ion concentrations^{1,2} and alleviate re-entry blackout for signal frequencies ranging from VHF to X-band.^{1,2} Prior to the present study, however, a systematic computational technique to predict the alteration of plasma sheath properties as a result of liquid injection (and to subsequently interpret flight data^{1,2}) from given injection system parameters was not available. In this report we discuss an analysis and computer code which can be used to predict altered re-entry plasma sheath fluid mechanic, chemical and electrical properties when flow field (shock layer) properties for no material injection are given as input data. The computer code is a considerable advance over the one developed by Pergament, et al,³ which has been used to analyze electrostatic probe and signal recovery data taken during the RAM C-III flight.²

The fundamental assumption in the present model is that the influence of the injected liquid (and the electrophilic vapor) is limited to the volume contained within a droplet-vapor-air (DVA) tube which extends in a streamwise direction from the location of liquid jet breakup to the end of the body (see Fig. 1). The boundaries of the tube normal to and circumferentially around the body are defined by the expansion and spread of the electrophilic vapor. Flow properties and species concentrations are assumed to be constant over the tube cross section. Therefore only the streamwise influence of the injected liquid on integrated electron and ion concentrations within the plasma sheath can be predicted. This appears to be a reasonable first approach considering the overall complexity of this problem and the number of physical processes which must be incorporated into a realistic model.

¹ Hayes, D.T., Herskovitz, S.B., Lennon, J.R., and Poirier, J.L., "Preliminary Report on the Trailblazer II Chemical Alleviation Flight of 28 July 1972," AFCRL-72-0640, 25 October 1972; see also, "Inflight Electrostatic Probe Measurements of the Effect of Chemical Injection on the Properties of the Re-Entry Flow Field," AIAA Paper 73-692, AIAA 6th Fluid and Plasma Dynamics Conference, Palm Springs, 16-18 July 1973.

² Schroeder, L.C. and Akey, N.D., "Material Injection Alleviation during the RAM C-III Flight," J. Spacecraft Rock. 10, 170 (1973).

³ Pergament, H.S., Mikatarian, R.R., and Kurzius, S.C., "Fluid Mechanic and Chemical Kinetic Effects of Water and Freon E-3 Injection into Re-Entry Plasma Sheaths," Final Report, AeroChem TP-291, NASA CR-132309, October 1973.

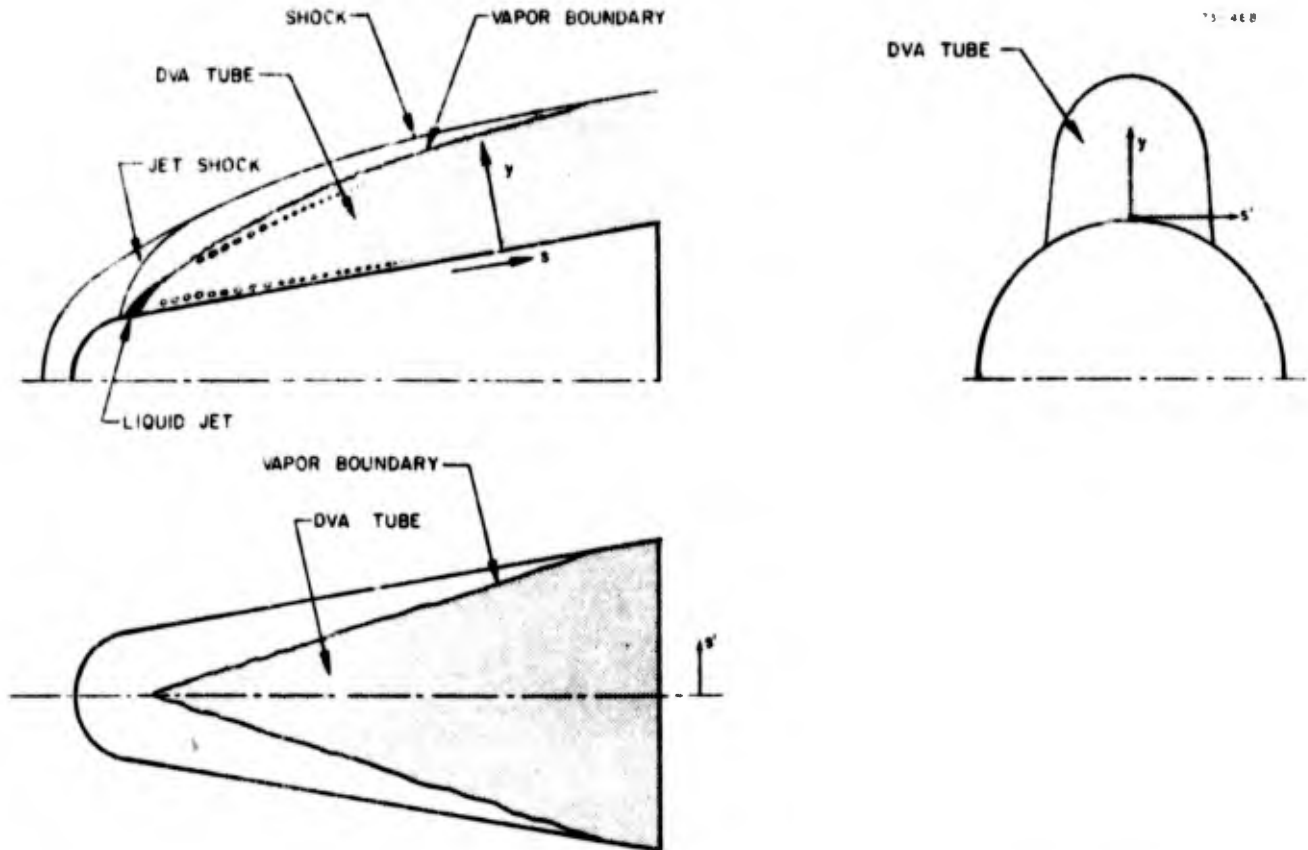


Figure 1. Schematic of liquid injection into a re-entry plasma sheath

The processes included in the model are: flash vaporization, penetration and breakup of the liquid jet, droplet acceleration and vaporization, diffusion of the spray/vapor cloud into the external (shock layer) flow field, decomposition of the vapor, heterogeneous and homogeneous charged species recombination kinetics and homogeneous electron attachment/detachment kinetics.

This report is a detailed description of the analysis, including all equations utilized in the computer code. The code is written in such a manner that it should be relatively easy to replace existing pieces of the model (e.g. droplet vaporization, vapor diffusion, etc.) with more sophisticated treatments, when these become available. There are two basic versions of the code (versions C and D), which are almost identical except for the chemistry routine. Version C incorporates a specialized chemistry routine which considers specific heterogeneous and homogeneous electron/ion recombination reactions and electron attachment/detachment reactions and does not allow for vapor decomposition. Version D incorporates a generalized chemistry routine in which up to 50 different species involved in 100 different reactions can be employed. (Unfortunately, version D has not yet been completely debugged.)

This report also serves as a Program User's Manual, incorporating instructions for preparing input data, sample input data, sample output and Fortran listings for versions C and D.

II. MATHEMATICAL MODEL

A. Phenomenological Description

A schematic of liquid injection into a re-entry plasma sheath is shown in Fig. 1. For blunt re-entry bodies, shock layer Mach numbers immediately downstream of the sphere-cone junction (the location of injection orifices for both Trailblazer¹ and RAM C-III² flight test vehicles) are, typically, about 2 to 3. Thus the presence of the liquid jet induces a strong shock which is nearly normal at the surface and rapidly decays to a weak shock before it finally coalesces into the air shock surrounding the body. The liquid jet is therefore exposed to air pressures and temperatures which are greater than the undisturbed shock layer values at the injection orifice. If the temperature of the injected liquid is greater than its boiling point at the post-shock pressure a fraction of the liquid will "flash" into vapor (flash vaporization) in order to reach pressure equilibrium with the surroundings. The remaining liquid is deformed and turned downstream by the air flow while simultaneously disintegrating into droplets; the droplets are accelerated and vaporized by the momentum and heat transferred from the high velocity, high temperature shock layer.

Mixing between the droplet/vapor spray and shock layer air influences the plasma sheath chemistry by: (i) reducing the shock layer temperature (which reduces ionization rates), (ii) enhancing the rate of electron/ion recombination over the dominant homogeneous recombination reaction, $\text{NO}^+ + e^- \rightarrow \text{N} + \text{O}$ via heterogeneous recombination, and (iii) attaching electrons. It is anticipated that the efficiency of electrophilic vapor attachment will exceed that of heterogeneous electron/ion recombination in reducing electron concentrations.³

The top of Fig. 1 shows the normal penetration of vapor and droplets into the shock layer, while the bottom of Fig. 1 shows the circumferential (lateral) spread of the vapor and droplets around the body. Flight tests^{1,2} have shown that the extent of the lateral spread of the vapor off the injection axis can be substantial.

B. Model Formulation

In principle, the accurate prediction of plasma sheath properties during liquid injection cycles requires a three-dimensional analysis. However the effort to model the complex interactions between the liquid jet, droplets, vapor and shock layer air in three dimensions was beyond the scope of the present study;* therefore a one-dimensional approach was taken. The present model

* In addition, the uncertainties in some of the input data, e.g. droplet sizes, entrainment rates, rate coefficients, etc., are sufficiently large that a three-dimensional analysis is not warranted at this time. However after a number of calculations have been performed with the present computer code to interpret flight data^{1,2} and obtain a feeling for the relative importance of all the input parameters on altered plasma sheath properties, it may be desirable to extend the present model to three dimensions.

should give an adequate description of the significant physical/chemical processes without unnecessarily complicating the analysis.

Referring to Fig. 2, the normal shock air passing through a cross sectional area determined by the liquid jet penetration and the spray lateral spread distance is assumed to mix perfectly with the "flashed" vapor. This mixture, together with the droplets, enters the DVA tube at the point of liquid jet breakup, establishing the initial gas and droplet conditions for the DVA tube calculations. Droplet vaporization and droplet/vapor/air chemical reactions, occurring within the DVA tube, are assumed to be functions only of streamwise distance, parallel to the re-entry body. Shock layer air is continually ingested into the DVA tube (Fig. 2) and instantaneously mixed with the air/vapor mixture flowing within the tube. The neutral species entering the tube are assumed to be frozen at their shock layer values at the point of entry, whereas the charged species (e.g. NO^+ , e^-) will, of course, be depleted via the recombination and attachment reactions taking place within the tube.

The pressure along the tube must be specified as input data. The initial pressure is that behind the normal (jet) shock, and the streamwise distribution is assumed to exponentially decay* to the shock layer pressure for no material injection.

Expressions for momentum, energy and mass (vaporization) transfer rates between the droplets and the vapor/air mixture, together with a suitable chemical reaction mechanism are incorporated into the DVA tube governing equations. During vaporization the droplets are assumed to remain at their (initial) boiling point temperature. The problem is completely formulated once the pressure distribution, initial conditions at the point of liquid jet breakup and the shock layer properties for no material injection are known; the gas and droplet conservation equations are then integrated along the DVA tube in the streamwise direction.

* The "relaxation" length is an input parameter to the computer code. Preliminary calculations show that the streamwise property distributions are relatively insensitive to this parameter if a reasonable value is chosen (say, 1/2 the body length).

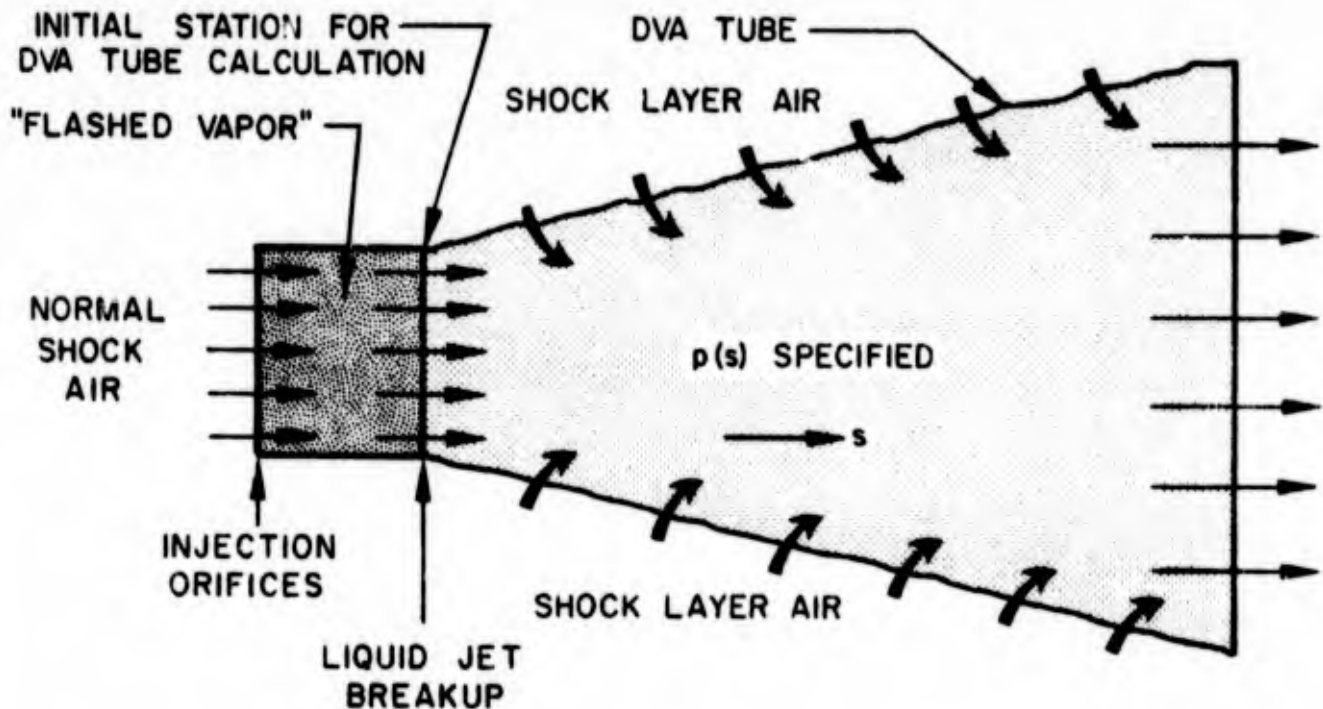


Figure 2. Schematic of Droplet-Vapor-Air (DVA) tube model.

III. SHOCK LAYER PROPERTIES

Several programs have been used⁴⁻⁷ to calculate the shock layer properties for no material injection for different re-entry test vehicles. For example, the method of Lew⁴ has been applied to the Trailblazer flight tests; the fully viscous shock layer theory of Kang, et al⁵ has been used to make RAM C-III

⁴ Lew, H.G., "A Study of Electron Density Distribution in Viscous Flows," AFCRL-72-0718, November 1972

⁵ Kang, S.W., Jones, W.L., and Dunn, M.G., "Theoretical and Measured Electron-Density Distributions at High Altitudes," AIAA J. 11, 141 (1973).

⁶ Schexnayder, C.J., Jr., Huber, P.W., and Evans, J.S., "Calculations of Electron Concentration for a Blunt Body at Orbital Speeds and Comparison with Experimental Data," NASA TN D-6294, 1971.

⁷ Kiel, R.E., Bergenn, R.F., Jr., Kessler, T.J., and Kaplan, A.E., "Boundary Layer Plasma of a Re-entry Vehicle: A Comparison of Prediction Models and Flight Measurements," AIAA J. 11, 1235 (1973).

flight calculations,* as has the inviscid flow/boundary layer analysis of Schexnayder, et al.⁶ Kiel, et al.⁷ have combined several programs to interpret electrostatic probe flight data taken in the Re-entry Measurements Program-Phase B.

The shock layer properties required as input data are, pressure, temperature, velocity and the following species concentrations: NO^+ , e^- , O , N , O_2 , N_2 , NO . These properties must be specified (input cards 10 and 11) as a function of distance normal to the body for a selected number of streamwise locations measured from the liquid injection station. Average property values are determined by the program at each input value of s as follows:

$$\bar{p}_e(s) = \frac{1}{y_{\max}} \int_0^{y_{\max}} p_e(s,y) dy \quad (1)$$

with similar definitions for $\bar{T}_e(s)$, $\bar{u}_e(s)$ and $\bar{F}_{e_i}(s)$. The molecular weight of the entrained air is then determined from

$$\bar{w}_a(s) = \left[\sum_i \bar{F}_{e_i} \right]^{-1} \quad (2)$$

The entrained air density is then computed from the equation of state,

$$\bar{\rho}_e(s) = \frac{\bar{p}_e(s) \bar{w}_a(s)}{R \bar{T}_e(s)} \quad (3)$$

and the stagnation enthalpy is computed from,

$$h_{s_e}(s) = \bar{h}_e(s) + \bar{u}_e^2/2 \quad (4)$$

where[†]

$$\bar{h}_e(s) = \bar{h}_e[\bar{\Gamma}_e(s)] \quad (5)$$

A linear interpolation scheme is used to obtain shock layer properties at values of s other than the input points.

* Recently, Miner and Lewis⁸ have also reported shock layer calculations for the RAM C-III vehicle.

† The enthalpy-temperature relation used in the program is described in Section VIII.

⁸ Miner, E.W. and Lewis, C.H., "Hypersonic Chemically Reacting Viscous Shock Layers over Sphere-Cones and Cylinder-Wedges," AIAA Paper 74-172, AIAA 12th Aerospace Sciences Meeting, Washington DC, 30 January-1 February 1974.

The actual pressure distribution along the DVA tube is different than \bar{p}_e due to the displacement of the shock layer air by the liquid jet and electrophilic vapor. At the injection station the pressure increases to the value immediately downstream of the normal shock and is assumed to remain constant until the point of liquid jet breakup (i.e. the start of the DVA tube calculations). Downstream of this point the pressure is assumed to decay to the undisturbed shock layer value, according to,

$$\bar{p} = \bar{p}_e + \bar{\rho}_e \bar{u}_e^2 \left(1 - \frac{\rho_x}{\rho_y}\right) e^{-(s - s_{bk})/L_s} \quad (6)$$

where L_s is a relaxation length within which the dynamic pressure decreases to $1/e$ of its local normal shock value. L_s is input to the code and should be taken to be some fraction of the body length.

IV. JET SHOCK, FLASH VAPORIZATION AND LIQUID JET BREAKUP

In this section analyses are described for determining (i) conditions downstream of the liquid jet shock, (ii) the amount of liquid that "flashes" into vapor, (iii) the position of the liquid jet at breakup and (iv) the initial droplet properties. The general procedure is to compute properties downstream of the jet (normal) shock using, first, the integrated average properties in the shock layer at the injection station (see Section III) as the pre-shock properties. The computed post-shock properties are then used to compute flash vaporization and liquid jet breakup. After the distance normal to the body at which breakup occurs (y_{bk}) is determined, all calculations are repeated using as input data to the normal shock calculations integrated average property values up to y_{bk} instead of y_{max} . This single iteration* accounts for the fact that the jet shock is normal only in a region relatively near the body and that y_{max} is approximately equal to the total shock layer thickness.

A. Jet Shock

The properties immediately downstream of the jet shock are computed via solution of the conservation equations through a normal shock. These equations are:

* This iteration is performed by the program even if all the liquid flashes into vapor. In that case the y_{bk} is used only to establish a reasonable value for the initial entrainment area (see Eq. (35)).

$$p_y = p_x + \rho_x u_x^2 \left(1 - \frac{\rho_x}{\rho_y}\right) \quad (7)$$

$$h_y = h_x + (u_x^2/2) \left(1 - \frac{\rho_x}{\rho_y}\right) \quad (8)$$

$$u_y = u_x \frac{\rho_x}{\rho_y} \quad (9)$$

$$T_y = T(h_y) \quad (10)$$

The method of solution is to assume ρ_x/ρ_y and solve for p_y and h_y from Eqs. (7) and (8). The enthalpy-temperature polynomial is then inverted (via the procedure described in Section VIII.B) to determine T_y , and ρ_y is computed from the equation of state

$$\rho_y = \frac{W_y p_y}{R T_y} \quad (11)$$

A solution is obtained when the computed value of ρ_x/ρ_y equals the assumed value of ρ_x/ρ_y .

B. Flash Vaporization

Assuming that the vapor pressure of the superheated liquid jet is in equilibrium with the surrounding pressure, p_y , the fraction of liquid that "flashes" into vapor can be computed from,

$$f = \frac{C_{p,\ell}(T_{\ell,o} - T_{bp})}{\Delta H_v} \quad (12)$$

If the temperature of the liquid jet is lower than T_{bp} there is no flash vaporization. For large temperature differences f can be computed from Eq. (12) to be greater than unity; in that case the program will set $f = 1$.*

The boiling point of the electrophilic liquid is usually known at 1 atm. The program uses this value as input data and computes T_{bp} at the local pressure, p_y , from the following expression,⁹

* The program will also set $f = 1$ if f is greater than some input number (FFVI, Card 2, Cols. 51-60).

⁹ Weast, R.C., ed., Handbook of Chemistry and Physics, 52nd edition (Chemical Rubber Co., Cleveland, 1971), p. D-144.

$$T_{bp} = \left(\frac{1}{T_{bp}(1)} - \frac{R}{\Delta H_v} \ln p_y \right)^{-1} \quad (13)$$

where ΔH_v is assumed to be a constant equal to its value at 1 atm, and $T_{bp}(1)$ is the boiling point at 1 atm.

C. Liquid Jet Breakup

The program allows for the occurrence of either aerodynamic breakup or vapor pressure breakup, depending on whether the Weber number exceeds or is less than the critical Weber number. The value for the critical Weber number is input to the program; from the literature,¹⁰ a reasonable value appears to be $We_{cr} = 6.2$. The Weber number is defined as,

$$We = \rho_y u_y^2 d_{eq} / 2\sigma_l \quad (14)$$

where d_{eq} is the diameter of an equivalent single orifice having the same total mass flow as all the in-line (with the flow direction) orifices (i.e. $d_{eq} = \sqrt{N} d_o$).^{*} It is necessary to introduce the concept of an equivalent single orifice since both the Air Force (Trailblazer)¹ and NASA (RAM C-III)² flight test vehicles used multiple orifices for increasing the penetration of the jet into the shock layer.

1. Aerodynamic Breakup ($We > We_{cr}$)

Aerodynamic breakup of the liquid jet is assumed to occur via the jet stripping mechanism discussed in Ref. 3. From that analysis we obtain the following expression for the jet breakup time,

$$t_{bk} = 0.536 \left(\frac{\rho_l}{\rho_y} \right)^{2/3} \left(\frac{\mu_y}{\mu_l} \right)^{1/3} \frac{d_{eq}}{2u_y} \sqrt{\frac{Re_y}{1 + \frac{2We}{Re_y}}} \quad (15)$$

In Eq. (15) the air viscosity is computed from Sutherland's law,¹¹

¹⁰Lane, W.R., "Shatter of Drops in Streams of Air," I and EC 43, 1312 (1951).

* The justification for the use of this equivalent orifice diameter is given in Ref. 3.

¹¹Sutherland, W., "The Viscosity of Gases and Molecular Force," London, Edinburgh and Dublin Phil. Mag. J. Sci.--5th Series 36, 507 (1895).

$$\mu = \mu_{\text{ref}} \left(\frac{T}{T_{\text{ref}}} \right)^{3/2} \left[\frac{T_{\text{ref}} + 110}{T + 110} \right] \quad (16)$$

or

$$\mu = 1.21 \times 10^{-6} \frac{T^{3/2}}{T + 110} \text{ g/cm-sec} \quad (16a)$$

where T is in $^{\circ}\text{K}$. The Reynolds number, Re_y is defined as,

$$Re_y = \frac{\rho_y u_y d_{\text{eq}}}{2 \mu_y} \quad (17)$$

From a force balance on the jet the location of jet breakup ($s_{\text{bk}}, y_{\text{bk}}$) is³

$$s_{\text{bk}} = \frac{C_D \rho_y}{\pi d_{\text{eq}} \rho_l} (u_y - V_l \sin \Theta)^2 t_{\text{bk}}^2 + V_l \sin \Theta t_{\text{bk}} \quad (18)$$

$$y_{\text{bk}} = V_l \cos \Theta t_{\text{bk}} \quad (19)$$

The streamwise jet velocity component at breakup is,

$$u_{\text{bk}} = \frac{2 C_D \rho_y}{\pi d_{\text{eq}} \rho_l} (u_y - V_l \sin \Theta)^2 t_{\text{bk}} + V_l \sin \Theta \quad (20)$$

The vertical jet velocity component is assumed to be constant, i.e.

$$V_{\text{bk}} = V_l \cos \Theta \quad (21)$$

The drag coefficient C_D is taken to be that for a cylinder in hypersonic cross flow ($C_D = 1.2$).

A modified form of the Volynsky correlation^{1,2} is used to estimate the mass-averaged droplet diameter, i.e.

^{1,2} Kurzius, S.C. and Raab, F.H., "Measurement of Droplet Sizes in Liquid Jets Atomized in Low-Density Supersonic Streams," AeroChem TP-152, NASA CR-1242, March 1967.

$$\frac{d_{30}}{d_{eq}} = 48 W_L^{-3/8} (Re_y/M_y)^{-1/4} \quad (22)$$

where M_y is the air Mach number downstream of the normal shock, defined as

$$M_y = u_y / (\gamma_y R T_y)^{1/2} \quad (23)$$

and W_L is defined as $2(\rho_l/\rho)We$. The specific heat ratio is computed from,

$$\gamma = \left(1 - \frac{R}{W_y C_{py}}\right)^{-1} \quad (24)$$

where W_y and C_{py} are evaluated assuming the composition of the air remains frozen through the jet shock at its input value at the injection location ($s = 0$).

The mean droplet size, d_{32} , is then obtained from the expression given by Gooderum and Bushnell,¹³

$$d_{32} = 1.38 d_{30} \quad (25)$$

The droplet diameter computed from Eq. (25) is the initial droplet diameter for the DVA tube calculations.

2. Vapor Pressure Breakup ($We < We_{cr}$)

For Weber numbers less than the critical Weber number breakup occurs via "boiling" of the liquid jet after emerging from the nozzle. It has been observed¹⁴ that breakup is initiated at some characteristic temperature of the jet, designated the "shatter temperature". Gooderum and Bushnell¹⁴ have correlated this temperature with orifice diameter for water jets and have obtained the following relation,

$$\frac{T_{sh} - T_{bp}}{T_{sh}} = \xi \quad (26)$$

where $\xi \approx 0.1$ for orifice diameters from 0.01 in. to 0.04 in. Droplet

¹³ Gooderum, P. and Bushnell, D.M., "Atomization, Drop Size and Penetration for Cross-Stream Water Injection During High Altitude Re-entry with Application to the Ram C-I Flight," NASA TN D-6747, July 1972.

¹⁴ Bushnell, D.M. and Gooderum, P.B., "Atomization of Superheated Water Jets at Low Ambient Pressures," J. Spacecraft Rock. 5, 231 (1968).

diameters are calculated from additional results of Gooderum and Bushnell¹⁵ who correlated water droplet sizes with jet injection temperature and orifice diameter. A curve fit of the data given in Ref. 15 results in the following expression

$$\frac{d_{32}}{d_o} = A_k \exp(- B_k T) \quad , \quad (26a)$$

where $A_k = 612$ and $B_k = 0.0244$ and T is in °K. The extrapolation to electrophilic liquid jets must obviously be made with caution.

a. $T_{\ell,o} < T_{sh}$ - For liquid injection temperatures less than the shatter temperature the breakup time can be calculated from the time to heat the jet from $T_{\ell,o}$ to T_{sh} , i.e.,

$$t_{bk} = \frac{C_{p,\ell} (T_{sh} - T_{\ell,o}) \rho_{\ell} d_{eq}}{4 q''} \quad (27)$$

The heat transfer to the jet is computed from the expression for free molecular heat transfer rate given in Section VI.B. The position and velocity of the jet at breakup are then obtained from Eqs. (18) - (21).

b. $T_{\ell,o} > T_{sh}$ - In this case the jet will 'flash' immediately into vapor, reaching the boiling point at the post-shock pressure. The jet breakup time is then computed assuming the jet is heated from T_{bp} to T_{sh} .

The present version of the program automatically goes into the vapor pressure breakup routine if $We < We_{cr}$, after calculating liquid jet breakup parameters from the aerodynamic breakup analysis and completing the DVA tube calculations. This is done because the precise value of We_{cr} is not well known and, if We is only marginally less than We_{cr} , it appears worthwhile to calculate DVA tube properties for both jet breakup modes and compare the predicted distributions of charged species concentrations for the two cases.

¹⁵Gooderum, P.B. and Bushnell, D.M., "Measurement of Mean Drop Sizes for Sprays from Superheated Waterjets," J. Spacecraft Rock. 6, 201 (1969).

V. INITIAL CONDITIONS FOR DVA TUBE CALCULATIONS

The initial conditions for DVA tube calculations are determined via an analysis that assumes perfect mixing between 'flashed' vapor and the shock layer air passing through a semi-elliptical cross sectional area, defined by the maximum jet penetration,* y_{bk} , and the lateral spread of the spray at the point of jet breakup, s'_{bk} .

A. Lateral Spread at Breakup

The lateral spread of the spray has been estimated via interpretation of data obtained for water sprays by Horn and Reichenbach,¹⁶ who showed that the lateral spread distance scaled by the orifice diameter (at $s = 150 d'_{eq}$) depends approximately on the inverse square root of the free stream Mach number. In order to interpret the data of Ref. 16 in terms of the lateral spread of electrophilic liquid sprays, a simple dimensional analysis of the lateral motion of a droplet was formulated (see Appendix A). The results of this analysis indicate that,

$$\frac{s'}{d'_{eq}} = K' \frac{d'_{eq}}{r_p} \quad (28)$$

Using Eq. (28) to interpret the data¹⁶ shows that the proportionality constant is approximately unity. The equivalent diameter, d'_{eq} , to be used in Eq. (28) is one having a mass flow equal to the total mass flow from all the side-by-side orifices,† i.e.

$$d'_{eq} = \sqrt{N'} d_o \quad (29)$$

In Ref. 16 it was found that at a downstream distance (from the orifice) of about $70 d'_{eq}$ the lateral spread distance approached its asymptotic value. In order to calculate the lateral spread downstream of this point we

* If all the liquid flashes into vapor ($f = 1$) y_{bk} will still be computed via the methods described in Section IV.C. However, in this case y_{bk} will be a fictitious value, used only to determine an approximate initial cross sectional area for the DVA tube calculations.

¹⁶ Horn, K.P. and Reichenbach, R.E., "Further Experiments on Spreading of Liquids Injected into a Supersonic Flow," AIAA J. 7, 358 (1969).

† Note that this equivalent diameter is different from d_{eq} , used to compute penetration distances and liquid jet breakup.

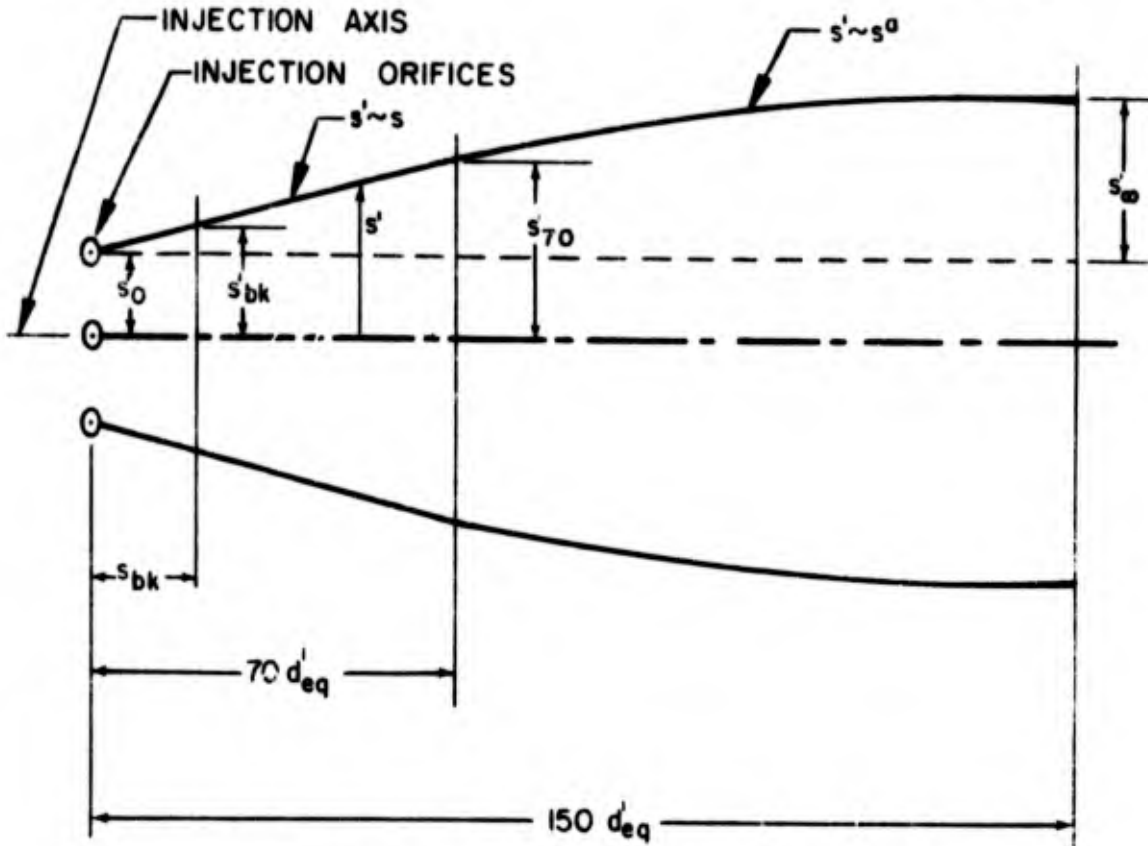


Figure 3. Significant dimensions for lateral spread calculations.

assume that the growth of the spray is similar to that for the wake growth¹⁷ behind circular cylinders and spheres. Thus for $s > 70 d'_{eq}$ (see Fig. 3),

$$s' = s'_o + C s^a \tag{30}$$

We assume that the lateral growth of the spray is linear from the orifice to $70 d'_{eq}$. Thus, for $s' < 70 d'_{eq}$

$$s' = s'_o + \frac{s'_{70} - s'_o}{70 d'_{eq}} s \tag{31}$$

The constant C in Eq. (30) is evaluated in terms of a at $s = 150 d'_{eq}$, the position at which the data of Ref. 16 were taken and where the correlation expressed by Eq. (28) applies. Thus,

$$C = s'_{\infty} / (150 d'_{eq})^a \tag{32}$$

¹⁷ Herrman, J., Slattery, R.E., and Clay, W.G., "Measured Properties of the Wakes of Hypersonic Cones," AIAA Paper No. 68-687, 1968.

where

$$s'_\infty = d'_{eq}{}^2 / r_{p_0} \quad (33)$$

Then, we get for $s' > 70 d'_{eq}$

$$s' = s'_0 + s'_\infty \left(\frac{s}{150 d'_{eq}} \right)^a \quad (30a)$$

Now s'_{0} is evaluated from Eq. (30a),

$$s'_{0} = s' + s'_\infty \left(\frac{70}{150} \right)^a \quad (34)$$

and, finally, we get for $s' < 70 d'_{eq}$

$$s' = s'_0 + s'_\infty \left(\frac{70}{150} \right)^a \left(\frac{s}{70 d'_{eq}} \right) \quad (31a)$$

Usually the streamwise position of jet breakup will be less than $70 d'_{eq}$. Thus, substituting s'_{bk} in Eq. (31a) will give the lateral spread distance at breakup, s'_{bk} . The entrainment area from which the mass flow of shock layer air is computed is then given by

$$A_{bk}^* = \frac{\pi}{2} y_{bk} s'_{bk} \quad (35)$$

for the assumed semi-elliptical cross section.

B. Shock Layer Air/Flashed Vapor Mixing

Assuming perfect mixing between the shock layer air and flashed vapor, the conservation equations are solved across a control volume bounded by the jet shock and the droplet/vapor/air mixture at the streamwise location of jet breakup. The mass flow of vapor is computed from

$$\dot{m}_v = f \dot{m}_l \quad (36)$$

and the mass flow of air is determined by

$$\dot{m}_a = \bar{\rho}_x \bar{u}_x A_{bk}^* \quad (37)$$

The conservation equations are:

Continuity

$$\dot{m}_a + \dot{m}_v = \rho_o u_o A_o \quad (38)$$

Momentum (assuming $p = p_y$ throughout the control volume)

$$\dot{m}_a u_y + \dot{m}_v u_\ell = \rho_o u_o^2 A_o \quad (39)$$

Energy

$$\dot{m}_a H_{s_a} + \dot{m}_v H_{s_v} = (\dot{m}_a + \dot{m}_v) H_{s_o} \quad (40)$$

$$H_{s_v} = C_{p_\ell} T_{bp} + \Delta H_v + v_\ell^2/2 \quad (41)$$

The momentum and energy equations are first solved to get u_o and H_{s_o} , respectively; the static temperature is then computed by inverting the H_{s_o} enthalpy-temperature relation. The equation of state is then used to compute ρ_o , noting that the molecular weight of the mixture now includes the molecular weight of the vapor. Finally the initial DVA tube area is obtained from the continuity equation, Eq. (38). This area is larger than the entrainment area, A_{bk}^* , because it must accommodate the vapor, which pushes the streamlines outward.

VI. DVA TUBE ANALYSIS

The DVA tube calculations are initiated at the point of liquid jet breakup using the droplet and air/vapor properties computed in Sections IV and V, respectively. One-dimensional droplet and gas conservation equations are solved along the tube with a pressure distribution specified by Eq. (6). Shock layer air is entrained continuously, the entrainment rate being a function of the entrainment area which is computed via the wake growth law given by Eqs. (30a) and (31a). The shock layer air is assumed to mix perfectly and instantaneously with the air/vapor mixture within the tube while undergoing no change in composition.

A. Governing Equations

1. Vapor/Air Mixture

The conservation equations applied in the DVA tube analysis are:

$$du = \frac{1}{\rho u A} \left[(u_p - u) \dot{\Phi}_v A ds + (\bar{\rho}_e \bar{u}_e^2 - \bar{\rho}_e \bar{u}_e u) dA^* - A dp - dF_p \right] \quad (42)$$

$$dH_s = \frac{1}{\rho u A} \left[(H_{p_s} - H_s) \dot{\phi}_v \text{ Ads} + (\bar{\rho}_e \bar{u}_e H_{s_e} - \bar{\rho}_e \bar{u}_e H_s) dA^* - \rho_e u_p A dH_{p_s} \right] \quad (43)$$

$$dY_i = \frac{1}{\rho u A} \left[W_i \dot{w}_i \text{ Ads} + (\bar{\rho}_e \bar{u}_e \bar{Y}_{e,i} - \bar{\rho}_e \bar{u}_e Y_i) dA^* - Y_i \dot{\phi}_v \text{ Ads} \right] \quad (44)$$

$$dY_v = \frac{1}{\rho u A} \left[(1 - Y_v) \dot{\phi}_v \text{ Ads} - Y_v \bar{\rho}_e \bar{u}_e dA^* \right] \quad (45)$$

(Eq. (45) is only used for version C, in which the vapor is assumed to undergo no decomposition. For version D all species mass fractions are computed from Eq. (44)).

$$dh = \left[(H_s + dH_s) - \frac{(u + du)^2}{2} \right] - h \quad (46)$$

Then, $T = T(h)$ by inverting the enthalpy-temperature polynomial, and

$$d\rho = \frac{\rho + d\rho}{R(T + dT)} \left[\sum_j \frac{Y_j + dY_j}{W_j} \right]^{-1} - \rho \quad (47)$$

where the summation over all j species in Eq. (47) includes the vapor. Finally the change in area is computed from,

$$dA = \frac{1}{\rho u} \left[\dot{\phi}_v \text{ Ads} + \bar{\rho}_e \bar{u}_e dA^* - Au d\rho - A\rho du \right] \quad (48)$$

The droplet properties in the above equations, u_p , dF_p , H_{p_s} and dH_{p_s} are determined via the droplet conservation equations; the vaporization rate $\dot{\phi}_v$ is computed from the heat transfer rate to the droplet and the chemical production terms, $\dot{w}_i (= \sum_j \dot{w}^{(j)})$ are computed from the specified chemical reaction

mechanism and rate coefficients.

2. Droplets

We assume that (i) the droplet temperature remains constant at its boiling point and, therefore, the energy equation does not have to be solved and (ii) the normal droplet velocity component remains constant at the normal liquid injection velocity, $V_\ell \cos\Theta$. The following equations are employed:

$$dF_p = \frac{9}{2} \frac{\rho_p}{\rho_\ell} \frac{f_p \mu}{r_p^2} (u - u_p) A ds \quad (49)$$

$$du_p = \frac{dF_p}{\rho_p u_p A} \quad (50)$$

$$\rho_p = \frac{\dot{m}_i - \rho Y_v u A}{u_p A} \quad (51)$$

$$H_{p_s} = C_{p_\ell} T_{bp} + u_p^2/2 \quad (52)$$

$$dH_{p_s} = C_{p_\ell} T_{bp} + \frac{(u_p + d u_p)^2}{2} - H_{p_s} \quad (53)$$

The term f_p in Eq. (49) is defined as the ratio of the actual droplet drag coefficient to the drag coefficient from Stokes law. In Eqs. (52) and (53) T_{bp} is the droplet boiling point temperature evaluated from Eq. (13) at the pressure behind the (normal) jet shock.

The change in droplet radius is determined from

$$dr_p = \frac{\dot{d}}{2 u_p} ds \quad (54)$$

where \dot{d} is determined from the free molecular heat transfer rate (see next section).

B. Droplet Vaporization

Free molecular heat transfer to the droplet is assumed, which gives the rate of change of droplet diameter as,³

$$\dot{d} = - \frac{\alpha_D (T_s - T_{bp}) p (u - u_p)}{\rho_l \Delta H_v T} \left[1 + \frac{\alpha_T}{\alpha_D} \left(\frac{v}{(u - u_p)} \right) \frac{T - T_{bp}}{T_s - T_{bp}} \right] \quad (55)$$

where T_s is the relative stagnation temperature between the droplets and gas determined from the relative stagnation enthalpy,

$$H_{s_r} = h + u_r^2/2 \quad (56)$$

and the enthalpy-temperature polynomial. The thermal mean velocity in Eq. (55) is computed from,

$$\dot{\phi}_v = (3p/\rho)^{1/2} \quad (57)$$

The vaporization rate (typically, in units of g/cm³-sec) is then expressed as,

$$\dot{\phi}_v = 4\pi r_p^2 \frac{\dot{d}}{2} n \rho_l \quad (58)$$

where n is the droplet number density, defined as,

$$n = \rho_p / (4/3) \pi r_p^3 \rho_l \quad (59)$$

Combining Eqs. (58) and (59) gives,

$$\dot{\phi}_v = \frac{3}{2} \frac{\rho_p}{r_p} \dot{d} \quad (60)$$

C. Droplet Drag Coefficient

The expression for the drag coefficient used in the program was taken from the work of Crowe,¹⁸ who correlated data on small diameter spheres in flow regimes from free molecular to continuum in terms of a normalized drag coefficient, defined as,

¹⁸ Crowe, C.T., "On the Momentum and Heat Transfer Equations for Two-Phase Plumes," JANNAF 6th Plume Technology Meeting, CPIA Publ. No. 209 (Applied Physics Lab., Johns Hopkins Univ., Silver Spring, 1971), p. 101.

$$\bar{C}_D = (C_D - C_{D_I}) / (C_{D_{FM}} - C_{D_I}) \quad (61)$$

where C_{D_I} is the incompressible high Reynolds number drag coefficient (taken to be 0.) and $C_{D_{FM}}$ is the free molecular drag coefficient. \bar{C}_D is expressed as a function of Knudson number and particle Reynolds number,

$$\bar{C}_D = G(Kn) D(Kn, Re_p) \quad (62)$$

where,

$$G(Kn) = \frac{Kn^{0.3} \exp(1.2 Kn^{0.5})}{1 + Kn^{0.3} \exp(1.2 Kn^{0.5})} \quad (63)$$

and

$$D(Kn, Re_p) = 1 - \exp \left[- \frac{Re_p}{8} Kn^{0.7} e^{Kn} (C_{D_0} - 0.48) \right] \quad (64)$$

$$Kn = 1.26 \sqrt{\gamma} \frac{M_p}{Re_p} \quad (65)$$

$$M_p = \frac{u - u_p}{a} \quad (66)$$

$$Re_p = \frac{\rho r_p (u - u_p)}{\mu} \quad (67)$$

C_{D_0} is obtained from Stokes drag law, (note that Re_p is defined in terms of r_p), i.e.

$$C_{D_0} = 12/Re_p \quad (68)$$

$C_{D_{FM}}$ is obtained from,

$$C_{D_{FM}} = \frac{\exp(-S_1^2/2)}{\sqrt{\pi} S_1^3} (1 + 2S_1^2) + \frac{4(S_1^4 - S_1^2) - 1}{2S_1^4} \operatorname{erf}(S_1) + \frac{2\sqrt{\pi}}{3 S_2} \quad (69)$$

where

$$S_1 = \sqrt{\gamma/2} M_p \quad (70)$$

and

$$S_2 = S_1 \sqrt{T/T_{bp}} \quad (71)$$

Finally, C_D is evaluated from Eq. (61) and $f_p = C_D/C_{D_0}$ is used in Eq. (49).

D. Normal Penetration and Entrainment Area

The penetration of the spray into the shock layer normal to the body as a function of streamwise distance, s , is computed by adding the penetration due to mixing with shock layer air to the local droplet penetration. The contribution due to mixing, y_{mix} , is assumed to be the same as the lateral spread distances ($s' - s'_0$) computed via Eq. (30a) or (31a). The droplet penetration is computed from,

$$y_{pen} = v_l \cos\theta \int \frac{ds}{u_p} \quad (72)$$

Thus the total penetration is $y_{tot} = y_{mix} + y_{pen}$, and the entrainment area is computed from

$$A^* = \frac{\pi}{2} s' y_{tot} \quad (73)$$

VII. CHEMICAL REACTION MECHANISMS

A. Specialized Chemistry Version (Version C)

This version of the code incorporates the following one-way reactions and rate coefficients.

	<u>Reaction</u>	<u>Rate Coefficient Form</u>
(1)	$\text{NO}^+ + \text{e}^- \rightarrow \text{N} + \text{O}$	$C_1 T^{-N_1}$
(2)	$\text{NO}^+ + \text{F}^- \rightarrow \text{NO} + \text{F}$	$C_2 T^{-N_2}$
(3)	$\text{NO}^+ + (\text{e}^- + \text{S}) \rightarrow \text{N} + \text{O} + (\text{S})$	k_3
(4)	$\text{NO}^+ + (\text{F}^- + \text{S}) \rightarrow \text{N} + \text{O} + (\text{S})$	k_4
(5)	$\text{Q} + \text{e}^- \rightarrow \text{F}^- + \text{Products}$	$C_5 T^{-N_5}$
(6)	$\text{F}^- + \text{M} \rightarrow \text{F} + \text{e}^- + \text{M}$	$C_6 \exp(-B_6/RT)$
(7)	$\text{F}^- + \text{O} \rightarrow \text{OF} + \text{e}^-$	$C_7 \exp(-B_7/RT)$
(8)	$\text{N} + \text{O} \rightarrow \text{NO}^+ + \text{e}^-$	$C_8 T^{N_8} \exp(-B_8/RT)$

The following points should be noted: (i) Reaction (8) is the reverse of Reaction (1); (ii) Reactions (3) and (4) are heterogeneous recombination reactions which occur at the surface of the droplets; expressions for k_3 and k_4 are given below*, (iii) Q represents the electrophilic vapor which is assumed not to be decomposed; (iv) The rate coefficients are incorporated into the program in the form shown, with the coefficients, temperature exponents and activation energies being input data (cards 8 and 9); (v) The justification for the use of this mechanism, the form of the rate coefficients and the specific values recommended in Section XI are given by Pergament, et al.³

Expressions for the production rates for each of the above reactions are:

Homogeneous

$$\dot{w}^{(1)} = k_1 \rho^2 F_{\text{NO}^+} F_{\text{e}^-} \quad (74)$$

$$\dot{w}^{(2)} = k_2 \rho^2 F_{\text{NO}^+} F_{\text{F}^-} \quad (75)$$

$$\dot{w}^{(5)} = k_5 \rho^2 F_{\text{Q}} F_{\text{e}^-} \quad (76)$$

$$\dot{w}^{(6)} = k_6 (\rho^2/W) F_{\text{F}^-} \quad (77)$$

$$\dot{w}^{(7)} = k_7 \rho^2 F_{\text{F}^-} F_{\text{O}} \quad (78)$$

$$\dot{w}^{(8)} = k_8 \rho^2 F_{\text{N}} F_{\text{O}} \quad (79)$$

* A discussion of the derivation of these equations is given in Ref. 3.

Heterogeneous

$$\dot{w}^{(3)} = k_3 \rho F_{NO^+} \quad (80)$$

$$\dot{w}^{(4)} = k_4 \rho F_{NO^+} \quad (81)$$

where

$$k_3 = f_3 k_s \left[\frac{\rho v_{e^-} F_{e^-}}{\rho v_{e^-} F_{e^-} + (v_{F^-} + u - u_d) \rho F_{F^-}} \right] \quad (82)$$

$$k_4 = f_4 k_s \left[\frac{(v_{F^-} + u - u_d) \rho F_{F^-}}{\rho v_{e^-} F_{e^-} + (v_{F^-} + u - u_d) \rho F_{F^-}} \right] \quad (83)$$

$$k_s = \frac{3}{4} \frac{(1 + \eta) \dot{m}_i (r_p / r_{p0})^3}{r_p \rho_l u_d} (v_{NO^+} + u - u_d) \quad (84)$$

The factors f_3 and f_4 (card 8) are coefficients by which the rates of the heterogeneous recombination reactions can be arbitrarily varied. η must be evaluated from the following transcendental equation,

$$\exp(-\eta) = \frac{(v_{+} + u - u_d) \rho F_{NO^+} (1 + \eta)}{\rho v_{e^-} F_{e^-} + (v_{F^-} + u - u_d) \rho F_{F^-}} \quad (85)$$

Expressions for the thermal velocities are:

$$v_{NO^+} = 1.88 \times 10^5 (T/5000)^{1/2} \text{ cm/sec} \quad (86)$$

$$v_{e^-} = 4.40 \times 10^7 (T/5000)^{1/2} \text{ cm/sec} \quad (87)$$

$$v_{F^-} = 2.36 \times 10^5 (T/5000)^{1/2} \text{ cm/sec} \quad (88)$$

where T is in °K.

B. Generalized Chemistry Version (Version D)

Reaction types are given for one direction only; both forward and backward reactions and rate coefficients must therefore be specified. The generalized form of the one-way reaction (card 9) is $A + B + C + D \rightarrow E + F + G + H$. Individual reaction types (card 9, cols. 41 and 42) are specified below.

<u>Reaction Type</u>	<u>Reaction</u>
(1)	$A + B \rightarrow E + F$
(2)	$A + B + M \rightarrow E + M$
(3)	$A + B \rightarrow E + F + G$
(4)	$A + B \rightarrow E$
(5)	$A + M \rightarrow E + F + M$
(6)	$A + B + C + M \rightarrow E + M$
(7)	$A + M \rightarrow E + F + G + M$
(8)	$A + B + C \rightarrow E + F$
(9)	$A + B + C + D \rightarrow E + F$
(10)	$A + B \rightarrow E + F + G + H$
(11)	$A \rightarrow \text{Products}$

Examples: For reaction type (2) on card 9, $C \equiv M$ and $F \equiv M$. For reaction type (5) on card 9, $B \equiv M$ and $G \equiv M$.

Rate coefficient types (card 9, cols. 43-44) are specified below; data must be given in cm-molecule-sec units.

<u>Rate Coefficient Type</u>	<u>Rate Coefficient, k</u>
(1)	A (constant)
(2)	$A T^{-1}$
(3)	$A T^{-2}$
(4)	$A T^{-1/2}$
(5)	$A \exp(B/RT)$
(6)	$A T^{-1} \exp(B/RT)$
(7)	$A T^{-3/2}$
(8)*	$A T^N \exp(B/T)$
(9)*	$A T^N [(K/T) + 1] \exp(B/T)$
(10)*	$\delta A T^N \exp(B/T)$

* Note that for these reactions the exponential term is expressed as $\exp(B/T)$ rather than $\exp(B/RT)$ as in rate coefficient types (5) and (6).

Expressions for the production rates for each of the reaction types are given below:

$$\dot{w}^{(1)} = k_f \rho^2 F_A F_B \quad (89)$$

$$\dot{w}^{(2)} = \frac{k_f \rho^3 F_A F_B}{W} \quad (90)$$

$$\dot{w}^{(3)} = k_f \rho^2 F_A F_B \quad (91)$$

$$\dot{w}^{(4)} = k_f \rho^2 F_A F_B \quad (92)$$

$$\dot{w}^{(5)} = \frac{k_f \rho^2 F_A}{W} \quad (93)$$

$$\dot{w}^{(6)} = k_f \rho^4 F_A F_B F_C / W \quad (94)$$

$$\dot{w}^{(7)} = k_f \rho^2 F_A / W \quad (95)$$

$$\dot{w}^{(8)} = k_f \rho^3 F_A F_B F_C \quad (96)$$

$$\dot{w}^{(9)} = k_f \rho^4 F_A F_B F_C F_D \quad (97)$$

$$\dot{w}^{(10)} = k_f \rho^2 F_A F_B \quad (98)$$

$$\dot{w}^{(11)*} = \delta k_f \rho^2 F_A F_H \quad (99)$$

* F_H in this reaction refers to the moles/g of hydrogen atoms that might be introduced into the flow due to an ablating heat shield.

Note that reaction-type (11) only specifies one reactant (with arbitrary products) and yet $\dot{w}^{(11)}$ has the form of a 2-body reaction. This special reaction-type is introduced because of the manner in which the rate coefficients were specified³ for the decomposition reactions involving the electrophilic vapor and hydrogen atoms, i.e.



where

$$(k)_Q = 4 \times 10^{-11} \exp(-4000/RT)$$

$$(k)_H = 9.2 \times 10^{-10} \exp(-4000/RT)$$

$$(k)_{HF} = 9.2 \times 10^{-10} \exp(-4000/RT)$$

$(k)_Q$ is the rate coefficient for depletion of Q by H; $(k)_H$ is the rate coefficient for the depletion of H by Q and $(k)_{HF}$ is the rate coefficient for the production of HF from Q. In this program these reactions would be handled by specifying the reaction $Q + H \rightarrow HF$ as reaction-type (4) with rate coefficient type (5), having $A = 4 \times 10^{-11}$; $B = -4000$. In the normal method of program operation this would give identical depletion rates for Q and H, (equal to the production rate of HF), via Eq. (92) above. But because of the manner of specifying the rate coefficients for this reaction, the above values of A and B only give the correct depletion rate for Q. Now we must invoke reaction type (11) and rate coefficient type (10) in the following way:

$$Q \rightarrow H \quad \delta = -(9.2 - 0.4)10^{-10} \text{ for depletion rate of H}^*$$

$$A = 1$$

$$N = 0$$

$$B = -2000$$

$$Q \rightarrow HF \quad \delta = (9.2 - 0.4)10^{-10} \text{ for production rate of HF}$$

$$A = 1$$

$$N = 0$$

$$B = -2000$$

* δ is negative because in the reaction of interest H is actually on the left hand side.

VIII. THERMODYNAMIC DATA

A. Program Input

The thermodynamic data have been incorporated directly into the specialized chemistry version of the program (version C) via a set of block data. The specific heat of each species is expressed in the following form,

$$C_{p_i} = L_{1_i} + L_{2_i}\Theta + L_{3_i}\Theta^2 + L_{4_i}\Theta^3 + L_{5_i}\Theta^{-2} \quad \text{cal/mole-}^\circ\text{K} \quad (100)$$

where $\Theta = T(^{\circ}\text{K})/1000$. The enthalpy is then expressed as,

$$h_i = \int_0^T C_{p_i} dT + L_{6_i} \quad (101)$$

where $L_{6_i} = H_{f_i}^{\circ} - \int_0^{298} C_{p_i} dT$; $H_{f_i}^{\circ}$ being the heat of formation at 298°K. Thus

the block data are L_{1_i} through L_{6_i} and the molecular weight of each species, W_i . The values of the coefficients used in version C were obtained by curve fits of the JANNAF data.¹⁹

For the generalized chemistry version (version D) the above data have to be input (card type 8), up to a maximum of 50 species, in the manner discussed in Section XI.B.

It should be noted that thermodynamic data are used in the program only to determine the enthalpy and temperature of the mixture. They are not used to determine equilibrium constants (via the Gibbs free energy) from which backward reaction rate coefficients could be evaluated (since only one-way reactions are employed in the kinetics scheme). Thus for species with small concentrations, and for which thermodynamic data are not available, it is recommended that only some reasonable value of L_1 (i.e., constant specific heat) be input and the remaining coefficients be set equal to zero. This should result in a negligible error in the enthalpy of the system.

¹⁹ JANNAF Thermochemical Tables (Dow Chemical Co., Midland, continuously updated).

B. Inversion of T from h(T)

As noted above the mixture enthalpy is expressed as $h = f(\Theta)$. The following numerical procedure is used to find a given value of h . Let

$$G = f(\Theta) - h \quad (102)$$

$$G_{n+1} = G_n + \left(\frac{\partial G}{\partial \Theta} \right)_n \Delta \Theta_n \quad (103)$$

where n is the number of the iteration step. Setting $G_{n+1} = 0$ gives

$$\Delta \Theta_n = -G_n / (\partial G / \partial \Theta)_n \quad (104)$$

Since

$$\Delta \Theta_n = \Theta_{n+1} - \Theta_n, \quad \text{we get}$$

$$\Theta_{n+1} = \Theta_n - G_n / (\partial G / \partial \Theta)_n \quad (105)$$

The above recursion formula is used to solve for Θ from $h(\Theta)$. The iteration stops when $G_n \leq \epsilon h$, where ϵ is a very small number.

IX. NUMERICAL TECHNIQUES

The ordinary differential equations given in Section VI (with the exception of the species continuity equation) are solved via a simple explicit marching procedure in which a nominal integration step size, Δs , is specified as input data. During the first 10 integration steps, however, the step size is reduced at least a factor of 10 less than the input value, and will be reduced still further if the vaporization rate is very large. This has been found to be necessary to obtain a smooth start to the calculations. After the first 10 steps the step size is allowed to increase up to 5 times the input value, but will still be controlled by the vaporization rate if $\dot{\phi}_v$ is large.

The species mass fractions, Y_i are evaluated via an implicit integration technique; i.e. these variables are evaluated from information at the forward step ($n + 1$) as well as at the present step (n). This is accomplished by linearizing the production terms in the individual species continuity equations, Eq. (44). The sum of all species continuity equations then results in a matrix equation in which known coefficients (elements of the matrix) multiply the unknown species concentrations (which form a vector). This product is equal to a known vector on the right hand side of the equation. The linearizations involving species F_i and F_j (for a two-body reaction) or F_i , F_j and F_k (for a three-body reaction) at station $n + 1$ (all variables are known at station n), are given by

$$\begin{aligned}
 (F_i F_j)_{n+1} &= (F_i F_j)_n + F_{jn} \left[(F_i)_{n+1} - (F_i)_n \right] + F_{in} \left[(F_j)_{n+1} - (F_j)_n \right] \\
 &= - \underline{(F_i F_j)_n} + \underline{(F_j)_n} (F_i)_{n+1} + \underline{(F_i)_n} (F_j)_{n+1}
 \end{aligned}$$

$$\begin{aligned}
 (F_i F_j F_k)_{n+1} &= - \underline{2(F_i F_j F_k)_n} + \underline{(F_j F_k)_n} (F_i)_{n+1} + \underline{(F_i F_k)_n} (F_j)_{n+1} \\
 &\quad + \underline{(F_i F_j)_n} (F_k)_{n+1}
 \end{aligned}$$

The terms underscored by a single line contribute to the elements of the coefficient matrix, while the terms underscored by a double line contribute to the known column matrix on the right hand side of the matrix equation for the linearized system. Thus the matrix equation takes the form (for N species),

$$\begin{bmatrix} a_{11} & a_{12} & a_{13} & & a_{1N} \\ & a_{21} & & & \\ & & & & \\ & & & & \\ a_{N1} & & & & a_{NN} \end{bmatrix} \begin{bmatrix} F_1 \\ F_2 \\ \\ F_N \end{bmatrix} = \begin{bmatrix} Q_1 \\ Q_2 \\ \\ Q_N \end{bmatrix}$$

It should be noted that the size of the matrix is determined by the number of species in the chemical system.

This implicit technique is used for both versions C and D. For the version D integration procedure, however, if any species mass fraction is predicted to be negative the step size will be halved until a positive value is obtained.* The recommended value of the initial step size (card 6) of 0.01 cm was determined from several trial calculations and should produce a stable solution. However, if the solution appears to be unstable this value should be reduced.

* It is a straightforward matter to add this step size control to version C and to incorporate a step size control on the other dependent variables (u, H_s, etc.).

X. PROGRAM OPERATION

The program was originally written in Fortran IV for the IBM 360-91 and converted to the CDC 6600. It can readily be converted to other machines with a sufficient core storage (Version C occupies a core of 40.6 K and version D occupies a core of 54.4 K on the CDC 6600). There are no special routines which must be provided by the system, other than the usual functions.

XI. PROGRAM INPUT DATA

If two complete (identical) data sets are used the program will make complete DVA tube calculations for both aerodynamic and vapor pressure breakup if $We < We_{cr}$. If only one data set is used the program will consider only aerodynamic breakup in establishing initial conditions for the DVA tube calculations (even if $We < We_{cr}$).

A. Version C - Specialized Chemistry

Card No.	Columns	Description	Format
1	1-80	Run identification	8 A 10
2	1-10	Factor which multiplies entrainment area, A^* (usually 1.0)	E 10.3
	11-20	Factor which multiplies DVA tube static pressure, p (usually 1.0)	E 10.3
	21-30	Factor which multiplies vaporization rate, $\dot{\phi}_v$ (usually 1.0)	E 10.3
	31-40	Factor which multiplies particle drag coefficient, C_D (usually 1.0)	E 10.3
	41-50	Pressure relaxation length, L_s (cm), see Eq. (6)	E 10.3
	51-60	FFVI, fraction of liquid that has flashed into vapor (f) is set equal to 1.0 if $f \geq FFVI$	E 10.3
3	1-10	Molecular weight of liquid injectant	E 10.3
	11-20	Specific heat of liquid (assumed constant) (cal/g-°K)	E 10.3
	21-30	Injectant boiling point at one atm (°K)	E 10.3

Card No.	Columns	Description	Format
3	31-40	Injectant heat of vaporization (cal/g)	E 10.3
	41-50	Density of liquid (g/cm ³)	E 10.3
	51-60	Injectant mass flux (g/sec)	E 10.3
	61-70	Velocity of liquid jet (cm/sec)	E 10.3
	71-80	Jet injection angle (deg), measured from normal to surface	E 10.3
4	1-10	Temperature of liquid jet (°K)	E 10.3
	11-20	Surface tension of liquid jet (dyne/cm)	E 10.3
	21-30	Viscosity of liquid jet (poise)	E 10.3
	31-40	Diameter of single orifice (cm)	E 10.3
	41-50	A _k { Coefficients in expression for d ₃₂ for B _k { vapor pressure breakup (see Eq. 26a)	E 10.3
	51-60		E 10.3
5	1- 5	Number of input data points per streamline (NPS)	I5
	6-10	Number of streamlines + 1 (NSL)	I5
	11-15	Number of in-line orifices	I5
	16-20	Debugging output index, If 0: No debug output 1: All debug output printed	I5
	21-25	Number of side-by-side orifices	I5
6	1-10	Value of s at injection site (<u>must</u> be 0)	E 10.3
	11-20	Initial integration step size DS (cm) (recommended value, 0.01 cm)	E 10.3
	21-30	Maximum integration distance (cm)	E 10.3
	31-40	Number of integration steps between print output (DSP), in multiples of DS; e.g. to print at every 10 steps, set DSP = 10 DS	E 10.3
7	1-10	Directed thermal accommodation coefficient, α _D (recommended value, 0.3)	E 10.3
	11-20	Random thermal accommodation coefficient, α _T (recommended value, 0.2)	E 10.3
	21-30	s' ₀ , maximum lateral spacing of orifices (cm)	E 10.3
	31-40	a, lateral spread parameter, (recommended value, 0.333)	E 10.3
	41-50	Critical Weber number (recommended value, 6.2)	E 10.3

Card No.	Columns	Description	Format
----------	---------	-------------	--------

Cards 8 and 9 contain input data on the rate coefficients for the reactions in the specialized chemistry version (see Section VII.A). Values used in previous calculations³ are given in parentheses (cm-molecule-sec units).

8	1-10	$C_1 (1 \times 10^{-2})$	E 10.3
	11-20	$C_2 (1 \times 10^{-3})$	E 10.3
	21-30	$C_3 (3 \times 10^{-1})$	E 10.3
	31-40	$C_6 (3 \times 10^{-10})$	E 10.3
	41-50	$C_7 (1 \times 10^{-10})$	E 10.3
	51-60	$C_8 (2.3 \times 10^{-16})$	E 10.3
	61-70	f_3 (usually 1.0)	E 10.3
	71-80	f_4 (usually 1.0)	E 10.3
9	1-10	$N_1 (1.5)$	E 10.3
	11-20	$N_2 (1.5)$	E 10.3
	21-30	$N_5 (2.0)$	E 10.3
	31-40	$N_8 (1.5)$	E 10.3
	41-50	$B_6 (8.0 \times 10^4)$	E 10.3
	51-60	$B_7 (3.0 \times 10^4)$	E 10.3
	61-70	$B_8 (6.4 \times 10^4)$	E 10.3

The following cards specify the free stream (shock layer) properties. Card-type 10 gives the streamwise location of each data set; these cards are noted, 10.1, 10.2...10.NPS. Card-type 11 gives the streamline location normal to the body and the property values. These cards are noted, 11.1.1.1, 11.1.1.2...11.1.(NSL-1)*.2 ...11.NPS.(NSL-1).2.

10.1	1-10	Streamwise location of first data set (cm), must be at injection site, $s \equiv 0$	E 10.3
11.1.1.1	1-10	Normal distance from body of first streamline (cm)	E 10.3
	11-20	Pressure (dyne/cm ²)	E 10.3
	21-30	Temperature (°K)	E 10.3
	31-40	Velocity (cm/sec)	E 10.3

* Note that on Card 5, Cols. 6-10, NSL is the number of streamlines + 1.

Card No.	Columns	Description	Format
11.1.1.1	41-50*	F_{NO^+} (mole/g)	E 10.3
	51-60	F_{e^-}	E 10.3
	61-70	F_O	E 10.3
	71-80	F_{F^-} (usually 0)	E 10.3
11.1.1.2	1-10	F_N (mole/g)	E10.3
	11-20	F_{O_2}	E 10.3
	21-30	F_{N_2}	E 10.3
	31-40	F_{NO}	E 10.3
11.1.2.1	1-10	Normal distance from body of second streamline (cm)	E 10.3
	11-20	Pressure (atm)	E 10.3
	⋮		
	71-80	F_{F^-} (usually 0)	E 10.3
11.1.2.2	1-10	F_N (mole/g)	E 10.3
	⋮		
	⋮		
	⋮	31-40	F_{NO}
11.1.(NSL-1).2.	⋮		
	31-40	F_{NO}	E 10.3
10.2	1-10	Streamwise location of second data set (cm)	E 10.3

* Note that $F_i = Y_i/W_i = X_i/W$, where Y_i = mass fraction; X_i = mole fraction; W_i = molecular weight of i th species and W = mixture molecular weight.

<u>Card No.</u>	<u>Columns</u>	<u>Description</u>	<u>Format</u>
11.2.1.1	1-10	Normal distance from body of first streamline	E 10.3
.	.		
.	.		
.	.		
11.2.(NSL-1).2	31-40	F _{NO}	E 10.3
.			
.			
.			
10.NPS	1-10	Streamwise location of last (NPS) data set	E 10.3
.			
.			
.			
.			
11.NPS.(NSL-1).2	31-40	F _{NO} for last data set (NPS) at last streamline (NSL-1) location	E 10.3

1 REPEAT ABOVE DATA SET

2

3

.

.

.

B. Version D - Generalized Chemistry

Card No.	Columns	Description	Format	
1	1-80	Run identification	8 A 10	
2	1-10	Factor which multiplies entrainment area, A^* (usually 1.0)	E 10.3	
	11-20	Factor which multiplies DVA tube static pressure, p (usually 1.0)	E 10.3	
	21-30	Factor which multiplies vaporization rate, $\dot{\phi}_V$ (usually 1.0)	E 10.3	
	31-40	Factor which multiplies drag coefficient, C_D (usually 1.0)	E 10.3	
	41-50	f_3 , factor which multiplies k_S in Eq. (82) (usually 1.0)	E 10.3	
	51-60	f_4 , factor which multiplies k_S in Eq. (83) (usually 1.0)	E 10.3	
	61-70	Pressure relaxation length, L_S (cm), see Eq. (6)	E 10.3	
	71-80	FFVI, fraction of liquid that has flashed into vapor (f) is set equal to 1.0 if $f \geq \text{FFVI}$	E 10.3	
	3	1-10	Molecular weight of liquid injectant	E 10.3
		11-20	Specific heat of liquid (assumed constant) (cal/g-°K)	E 10.3
21-30		Injectant boiling point at one atm (°K)	E 10.3	
31-40		Injectant heat of vaporization (cal/g)	E 10.3	
41-50		Density of liquid (g/cm ³)	E 10.3	
51-60		Injectant mass flux (g/sec)	E 10.3	
61-70		Velocity of liquid jet (cm/sec)	E 10.3	
71-80		Jet injection angle (deg), measured from normal to surface	E 10.3	
4	1-10	Temperature of liquid jet (°K)	E 10.3	
	11-20	Surface tension of liquid jet (dyne/cm)	E 10.3	
	21-30	Viscosity of liquid jet (poise)	E 10.3	
	31-40	Diameter of single orifice (cm)	E 10.3	
	41-50	A_k Coefficients in expression for d_{32}	E 10.3	
	51-60	B_k for vapor pressure breakup (see Eq. 26a)	E 10.3	

Card No.	Columns	Description	Format
5	1- 5	Number of input data points per streamline (NPS)	I5
	6-10	Number of streamlines + 1 (NSL)	I5
	11-15	Number of in-line orifices	I5
	16-20	Debugging output index, If 0: No debug output 1: All debug output printed	I5
	21-25	Number of homogeneous reactions (IKINE)	I5
	26-30	Number of species (NS1)	I5
	31-35	Number of side-by-side orifices	I5
	6	1-10	Value of s at injection site (<u>must</u> be 0)
11-20		Initial integration step size (DS), (cm) (recommended value, 0.01 cm)	I5
21-30		Maximum integration distance (cm)	E 10.3
31-40		Number of integration steps between print output	E 10.3
7	1-10	Directed thermal accommodation coefficient, α_D (recommended value, 0.3)	E 10.3
	11-20	Random thermal accommodation coefficient, α_T (recommended value, 0.2)	E 10.3
	21-30	s'_0 maximum lateral spread of orifices (cm)	E 10.3
	31-40	a, lateral spread parameter (recommended value, 0.333)	E 10.3
	41-50	Critical Weber number (recommended value, 6.2)	E 10.3

The following cards (card-type 8) contain the specific heat data for each species. Data must be included for each species. The first 8 species must be in the following order: NO^+ , e^- , O, F^- , N, O_2 , N_2 , NO. The remaining species can be in any order. Specific heats are expressed in the following form:

$$C_p = L_{1,i} + L_{2,i} \Theta + L_{3,i} \Theta^2 + L_{4,i} \Theta^3 + L_{5,i} \Theta^{-2} \quad \text{cal/mole-}^\circ\text{K where } \Theta = T(^{\circ}\text{K})/1000$$

<u>Card No.</u>	<u>Columns</u>	<u>Description</u>	<u>Format</u>
-----------------	----------------	--------------------	---------------

The enthalpy of each species is then expressed as

$$h_i = \int_0^T C_{P_i} dT + L_{e_i}$$

where $L_{e_i} = \Delta H_{f_i}^{\circ} - \int_0^{298} C_{P_i} dT$

$\Delta H_{f_i}^{\circ}$ = heat of formation at 298°K

8.1	1-10	First species (NO ⁺) name identification	A4
	11-20	Molecular weight	E 10.3
	21-30	L ₁	E 10.3
	31-40	L ₂	E 10.3
	41-50	L ₃	E 10.3
	51-60	L ₄	E 10.3
	61-70	L ₅	E 10.3
	71-80	L ₆	E 10.3
8.2	1-10	Second species (e ⁻) name identification	A4
.	.		
.	.		
.	.		
.	71-80	L ₆ for second species (e ⁻)	E 10.3
.	.		
.	.		
8.NS1	71-80	L ₆ for NS1 species	E 10.3

Card-type 9 specifies the reaction mechanism and rate coefficients (see Section VII.B). No particular order is required.

9.1	1- 4	Species A (Reaction 1)	A4
	5	+ sign	
	6- 9	Species B	A4
	10	+ sign	
	11-14	Species C	A4
	15	+ sign	
	16-19	Species D	A4

Card No.	Columns	Description	Format
9.1	20	= sign	
	21-24	Species E	A4
	25	+ sign	
	26-29	Species F	A4
	30	+ sign	
	31-34	Species G	A4
	35	+ sign	
	36-39	Species H	A4
	40	Blank	
	41-42	Reaction type (1-11)	I2
	43-44	Rate coefficient type (1-10)	I2
	45-53	A (Pre-exponential factor, cm-molecule-sec units)	E 9.2
	54-58	N (Temperature exponent)	F 5.2
	59-66	B (Activation energy, cal/mole)	E 8.1
	67-73	K (Reaction type 9 only)	E 7.1
	74-80	δ (Reaction type 10 only)	E 7.1
9.2	1- 4	Species A (Reaction 2)	A4
.			
.			
.			
9. IKINE	1- 4	Species A (Reaction IKINE)	A4
10.1	1-10	Streamwise location of first data set (cm), must be at injection site, $s \equiv 0$	E 10.3
11.1.1.1	1-10	Normal distance from body of first streamline (cm)	E 10.3
	11-20	Pressure (dyne/cm ²)	E 10.3
	21-30	Temperature (°K)	E 10.3
	31-40	Velocity (cm/sec)	E 10.3
	41-50*	F_{NO} (mole/g)	E 10.3
	51-60	F_{e^-}	E 10.3
	61-70	F_O	E 10.3
	71-80	F_{F^-} (usually 0)	E 10.3

* Note that $F_i = Y_i/W_i = X_i/W_i$, where Y_i = mass fraction; X_i = mole fraction; W_i = molecular weight of i th species and W = mixture molecular weight.

<u>Card No.</u>	<u>Columns</u>	<u>Description</u>	<u>Format</u>
11.1.1.2	1-10	F_N (mole/g)	E 10.3
	11-20	F_{O_2}	E 10.3
	21-30	F_{N_2}	E 10.3
	31-40	F_{NO}	E 10.3
11.1.2.1	1-10	Normal distance from body of second streamline (cm)	E 10.3
	11-20	Pressure (atm)	E 10.3
	.		
	.		
	.		
	71-80	F_{F^-} (usually 0)	E 10.3
11.1.2.2	1-10	F_N (mole/g)	E 10.3
.	.		
.	.		
.	31-40	F_{NO}	E 10.3
.			
.			
11.1.(NSL-1).2	.		
	.		
	31-40	F_{NO}	E 10.3
10.2	1-10	Streamwise location of second data set (cm)	E 10.3
11.2.1.1	1-10	Normal distance from body of first streamline	E 10.3
.	.		
11.2.(NSL-1).2	31-40	F_{NO}	E 10.3
.			
.			
10.NPS	1-10	Streamwise location of last (NPS) data set	E 10.3
.			
.			
.			
.			
11.NPS.(NSL-1).2	31-40	F_{NO} for last data set (NPS) at last streamline (NSL-1) location	E 10.3
12	1-10	Initial mass fraction of H atoms (This card <u>must</u> be included even if the value is zero)	E 10.3

<u>Card No.</u>	<u>Columns</u>	<u>Description</u>	<u>Format</u>
1		REPEAT ABOVE DATA SET	
2			
3			
.			
.			
.			
.			
.			
.			
.			
.			
.			
12			

XII. REFERENCES

1. Hayes, D.T., Herskovitz, S.B., Lennon, J.F., and Poirier, J.L., "Preliminary Report on the Trailblazer II Chemical Alleviation Flight of 28 July 1972," AFCRL-72-0640, 25 October 1972; see also, "In-flight Electrostatic Probe Measurements of the Effect of Chemical Injection on the Properties of the Re-Entry Flow Field," AIAA Paper 73-692, AIAA 6th Fluid and Plasma Dynamics Conference, Palm Springs, 16-18 July 1973.
2. Schroeder, L.C. and Akey, N.D., "Material Injection Alleviation during the RAM C-III Flight," J. Spacecraft Rock. 10, 170 (1973).
3. Pergament, H.S., Mikatarian, R.R., and Kurzius, S.C., "Fluid Mechanic and Chemical Kinetic Effects of Water and Freon E-3 Injection into Re-Entry Plasma Sheaths," Final Report, AeroChem TP-291, NASA CR-132309, October 1973.
4. Lew, H.G., "A Study of Electron Density Distribution in Viscous Flows," AFCRL-72-0718, November 1972.
5. Kang, S.W., Jones, W.L., and Dunn, M.G., "Theoretical and Measured Electron-Density Distributions at High Altitudes," AIAA J. 11, 141 (1973).
6. Schexnayder, C.J., Jr., Huber, P.W., and Evans, J.S., "Calculations of Electron Concentration for a Blunt Body at Orbital Speeds and Comparison with Experimental Data," NASA TN D-6294, 1971.
7. Kiel, R.E., Bergenn, R.F., Jr., Kessler, T.J., and Kaplan, A.E., "Boundary Layer Plasma of a Re-entry Vehicle: A Comparison of Prediction Models and Flight Measurements," AIAA J. 11, 1235 (1973).
8. Miner, E.W. and Lewis, C.H., "Hypersonic Chemically Reacting Viscous Shock Layers over Sphere-Cones and Cylinder-Wedges," AIAA Paper 74-172, AIAA 12th Aerospace Sciences Meeting, Washington, DC, 30 January - 1 February 1974.
9. Weast, R.C., ed., Handbook of Chemistry and Physics, 52nd edition (Chemical Rubber Co., Cleveland, 1971), p. D-144.
10. Lane, W.R., "Shatter of Drops in Streams of Air," I and EC 43, 1312 (1951).
11. Sutherland, W., "The Viscosity of Gases and Molecular Force," London, Edinburgh and Dublin Phil. Mag. J. Sci. --5th Series 36, 507 (1895).
12. Kurzius, S.C. and Raab, F.H., "Measurement of Droplet Sizes in Liquid Jets Atomized in Low-Density Supersonic Streams," AeroChem TP-152, NASA CR-1242, March 1967.

13. Gooderum, P. and Bushnell, D.M., "Atomization, Drop Size and Penetration for Cross-Stream Water Injection During High Altitude Re-entry with Application to the Ram C-I Flight," NASA TN D-6747, July 1972.
14. Bushnell, D.M. and Gooderum, P.B., "Atomization of Superheated Water Jets at Low Ambient Pressures," J. Spacecraft Rock. 5, 231 (1968).
15. Gooderum, P.B. and Bushnell, D.M., "Measurement of Mean Drop Sizes for Sprays from Superheated Waterjets," J. Spacecraft Rock. 6, 201 (1969).
16. Horn, K.P. and Reichenbach, R.E., "Further Experiments on Spreading of Liquids Injected into a Supersonic Flow," AIAA J. 7, 358 (1969).
17. Herrman, J., Slattery, R.E., and Clay, W.G., "Measured Properties of the Wakes of Hypersonic Cones," AIAA Paper No. 68-687, 1968.
18. Crowe, C.T., "On the Momentum and Heat Transfer Equations for Two-Phase Plumes," JANNAF 6th Plume Technology Meeting, CPIA Publ. No. 209 (Applied Physics Lab., Johns Hopkins Univ., Silver Spring, 1971), p. 101.
19. JANNAF Thermochemical Tables (Dow Chemical Co., Midland, continuously updated).
20. Weaver, W.L. "Multiple-Orifice Liquid Injection into Hypersonic Air-streams and Application to RAM C-III Flight," NASA TM X-2486, 1972.
21. Kush, E.A., Jr. and Schetz, J.A., "Decomposition of a Liquid Jet Injected Normal to a Supersonic Air Stream," VPI-E-10-72 (AFOSR TR-72-1180) Virginia Polytechnic Institute, June 1972.

NOMENCLATURE

A	DVA tube area; also constant in rate coefficient expression for generalized chemistry version of code (version D)
A*	entrainment area
A ₀	initial area for DVA tube calculations
a	exponent in expression for lateral spread, Eq. (30); also speed of sound
B	activation energies for rate coefficient types (5) and (6); activation energy/2 for rate coefficient types (8), (9), (10) (version D)
B ₆ , B ₇ , B ₈	activation energies in rate coefficient expressions
C	constant, defined by Eq. (32)
C ₁ , C ₂ ...C ₈	constants in rate coefficient expressions
C _D	drag coefficient
\bar{C}_D	normalized drag coefficient, defined by Eq. (61)
C _{D_I}	incompressible, high Reynolds number drag coefficient
C _{D_{FM}}	free molecular drag coefficient
C _{D₀}	drag coefficient from Stokes law, $C_{D_0} = 12/Re_p$
C _p	specific heat
C _{p,l}	specific heat of liquid
\dot{d}	rate of change of droplet diameter
d _{eq}	equivalent orifice diameter based on number of in-line orifices
d' _{eq}	equivalent orifice diameter based on number of side-by-side orifices
d _o	orifice diameter
d ₃₀	mass-averaged droplet diameter
d ₃₂	mean droplet diameter
erf	error function
F ₁	mass fraction/molecular weight (moles/g) of ith species; $F_i = Y_i/W_i$
f	fraction of liquid that flashes into vapor
f ₃ , f ₄	factors which multiply k _s in Eqs. (82) and (83) respectively
dF _p	defined by Eq. (49)
f _p	defined as C _D /C _{D₀}
H _f ^o	heat of formation at 298°K

H_s	stagnation enthalpy
H_{s_0}	initial stagnation enthalpy for DVA tube calculations
ΔH_v	heat of vaporization of liquid injectant
h	static enthalpy
K	constant in rate coefficient type (9) (version D)
Kn	Knudson number
k	rate coefficient
k_s	proportional to heterogeneous rate coefficients, k_3 and k_4 ; defined by Eq. (84)
$L_1, L_2,$... L_6	coefficients in enthalpy-temperature polynomial, Eq. (101)
L_s	pressure relaxation length, see Eq. (6)
M	Mach number
M_p	droplet Mach number, defined by Eq. (66)
\dot{m}_a	mass flow of shock layer air passing through A_{bk}^*
\dot{m}_i	mass flow of liquid injectant
\dot{m}_v	mass flow of 'flashed' vapor
N'	number of side-by-side orifices
N	number of in-line orifices; also temperature exponent in rate coefficient expression for generalized version D
$N_1, N_2,$ N_5	temperature exponents in rate coefficient expression
n	droplet number density
p	DVA tube static pressure, defined by Eq. (6)
q''	heat transfer rate
R	universal gas constant
Re	Reynolds number, defined by Eq. (17)
Re_p	droplet Reynolds number, defined by Eq. (67)
r_p	droplet radius
r_{p_0}	droplet radius at liquid jet breakup; initial value for DVA tube calculations
s	streamwise distance (along the DVA tube)
s'	lateral spread distance, measured from injection axis to DVA tube boundary
s'_∞	lateral spread distance at $s = 150 d'_{eq}$
s'_0	initial lateral spread distance

s'_0	lateral spread distance at $s = 70 d'_{eq}$
T	static temperature
$T_{l,o}$	temperature of injected liquid
T_s	stagnation temperature
T_{sh}	shatter temperature, see Eq. (26)
t_{bk}	time for liquid jet to break up
u	velocity in s direction
u_0	initial velocity for DVA tube calculations
u_d	total droplet velocity, $u_d = [u_p^2 + (V_l \cos \Theta)^2]^{1/2}$
u_r	velocity relative to droplet, $u_r = [(u - u_p)^2 + (V_l \cos \Theta)^2]^{1/2}$
u_l	streamwise component of injection velocity
V_l	liquid jet injection velocity
v	thermal velocity of neutrals
v_{e^-}	thermal velocity of electrons
v_{F^-}	thermal velocity of negative (F^-) ions
v_{NO^+}	thermal velocity of positive (NO^+) ions
W	mixture molecular weight
\bar{W}_a	average molecular weight of shock layer species; defined by Eq. (2)
W_L	liquid jet Weber number, defined as $W_L = 2We(\rho_l/\rho)$
We	Weber number, defined by Eq. (14)
We_{cr}	critical Weber number for aerodynamic breakup
W_i	molecular weight of ith species
\dot{w}_i	chemical production rate of ith species
$\dot{w}^{(j)}$	chemical production rate from jth reaction
Y_i	mass fraction of ith species
Y_v	mass fraction of vapor
y	distance normal to body
y_{max}	maximum distance normal to body at which shock layer input data are given
y_{mix}	expansion distance of DVA tube in both normal and lateral directions due to mixing
y_{pen}	droplet penetration distance normal to surface
y_{tot}	normal distance of DVA tube boundary from surface

Greek

α_D	directed thermal accommodation coefficient
α_T	random thermal accommodation coefficient
γ	specific heat ratio
η	defined by Eq. (85)
Θ	liquid jet injection angle (measured from normal to body); also $T(^{\circ}\text{K})/1000$ in Eq. (100)
μ	viscosity
ρ	density
ρ_0	initial density for DVA tube calculations
σ	surface tension
$\dot{\phi}_v$	vaporization rate

Subscripts

bk	value at point of liquid jet breakup
bp	boiling point
e	shock layer
i	ith species
j	jth reaction
l	liquid
p	droplet
r	relative to droplets
ref	reference value
x	upstream of normal (jet) shock
y	downstream of normal (jet) shock

Superscripts

_____ average value of shock layer property, defined, e.g. by Eq. (11)

APPENDIX A

LATERAL SPREAD OF SPRAY

In order to derive an approximate expression for the lateral spread of the spray we employ the following simplified dimensional analysis. The force acting on a droplet due to a pressure gradient in the s' direction is,

$$F = \frac{4}{3} \pi r_p^3 \frac{dp}{ds'} \quad (A-1)$$

From normal momentum and dimensional considerations, the pressure gradient across the droplet formed from the jet fragment can be expressed as,

$$\frac{dp}{ds'} \propto \frac{\rho u^2}{r_p} \quad (A-2)$$

The equation of lateral motion of the droplet is,

$$m_\ell \frac{\Delta w}{\Delta t} = F \quad (A-3)$$

where w is the droplet velocity in the s' direction and $\Delta t \propto L/u_\ell$ where L is a characteristic break-up length. Combining Eqs. (A-1), (A-2) and (A-3) gives the following expression for w ,

$$w \propto \frac{L}{r_p} \frac{\rho u^2}{\rho_\ell u_\ell} \quad (A-4)$$

Taking $L \sim y_{pen}$, we get

$$w \propto \left(\frac{y_{pen}}{d_o} \right)^2 \left(\frac{d_o}{r_p} \right) \frac{\rho u^2}{\rho_\ell u_\ell} \quad (A-4a)$$

The lateral distance the particle will move is,

$$s' = w \Delta t \quad (A-5)$$

Combining Eqs. (A-4) and (A-5) gives

$$s' \propto \left(\frac{y_{pen}}{d_o} \right) \frac{\rho u^2}{\rho_\ell u_\ell^2} \frac{d_o}{r_p} \cdot d_o \quad (A-6)$$

From the results of Weaver²⁰ we get $y_{pen}/d_o \propto [\rho_\ell u_\ell^2 / \rho u^2]$. Thus it is reasonable to express s' as

$$s'/d_o = C d_o/r_p \quad (A-7)$$

By interpretation of available experimental data on the lateral spread of liquid sprays,^{16,21} the proportionality constant (C) in Eq. (A-7) was found to be approximately unity.

APPENDIX B

SAMPLE INPUT DATA

1. Version C
2. Version D



COMPUTER CODE TO PREDICT THE EFFECTS OF
ELECTROPHILIC LIQUID INJECTION INTO RE-ENTRY PLASMA SHEATHS

Version C

```

1 *****J108 RAM C-III 200KFT FREON E-3 (CASE 6-TLO=360) 12/13/73C
2 1.0 1.0 1.0 1.0 50.0 0.8
3 618. 0.243 428. 14.5 1.723 152. 2.40E+3 39.0
4 380. 284.0 2.2E-2 0.051 612.0 0.0244
5 10 7 3 0 3
6 0.0 0.01 100.0 2.0
7 0.3 0.2 0.112 0.333 6.2
8 1.0E-2 1.0E-3 3.0E-1 3.E-10 1.E-10 2.3E-18 1.0
9 1.5 1.5 2.0 1.5 8.0E4 3.0E4 6.4E4

```

Note: All cards for 10 and 11 can be found under Version D input 10.1 thru 11.10

Duplicate above data set to reproduce sample output (Appendix C)



FURTRAN Coating Furnace

COMPUTER CODE TO PREDICT THE EFFECTS OF
ELECTROPHILIC LIQUID INJECTION INTO RE-ENTRY PLASMA SHEATHS

1 7

version D

Line	Code	RAM	C-III	200KFT	PREON	F-3	(CASE	6-720=360)	12/13/73C	50.0	0.8
1		1.0	1.0	1.0	1.0	1.0	1.0	1.0	1.0	50.0	0.8
2		618.	0.243	428.	14.5	1.723	152.0	2.40E+3	2.40E+3	39.00	
3		380.	284.0	2.2E-2	0.057	612.0	0.0244				
4		10	3	6	3						
5		0.0	0.01	100.0	200.0						
6		0.3	0.2	0.112	0.333	6.2					
7											
8	NO+	30.0	6.260E+0	1.927E+0	1.927E+0	-4.576E-1	3.634E-2	9.764E-3	9.764E-3	2.347E+2	
9	E-	5.487E-4	4.968E+0	-3.076E-4	-3.076E-4	7.913E-5	-6.637E-6	-1.709E-5	-1.709E-5	-1.481E+0	
10	O	16.0	5.062E+0	-1.174E-1	-1.174E-1	3.941E-2	-2.053E-3	1.813E-2	1.813E-2	5.812E+1	
11	F-	18.99	4.968E+0	-7.323E-5	-7.323E-5	2.247E-5	-2.775E-6	-1.045E-5	-1.045E-5	-6.368E+1	
12	N	14.0	5.046E+0	-8.613E-2	-8.613E-2	8.062E-3	6.133E-3	-5.908E-3	-5.908E-3	1.114E+2	
13	O2	32.0	7.241E+0	1.239E+0	1.239E+0	-1.844E-1	1.050E-2	-5.960E-2	-5.960E-2	-2.421E+0	
14	N2	28.0	6.264E+0	1.932E+0	1.932E+0	-4.615E-1	3.708E-2	9.185E-3	9.185E-3	-1.924E+0	
15	NO	30.0	6.752E+0	1.700E+0	1.700E+0	-4.176E-1	3.412E-2	-1.677E-2	-1.677E-2	1.942E+1	
16	Q	618.0	2.152E+2	0.0	0.0	0.0	0.0	0.0	0.0	0.0	
17	NO+	+E-	+0			1 7	1.1E-2	-1.5			
18	NO+	+F-	+0			1 7	1.0E-3	-1.5			
19	Q	+E-	+Q			1 7	3.0E-1	-2.0			
20	F-	+M	+E- +M			5 5	3.E-10				-8.0E+4
21	F-	+O	+E-			1 5	1.E-10				-3.0E+4
22	N	+O	+E-			1 8	2.3E-18	1.5			-3.2E+4
10.1		0.11	7.80E-3	1610.	1.00E5	3.920E-8	3.920E-8	1.45E-2	1.45E-2	0.0	
11.1.1											

COMPUTER CODE TO PREDICT THE EFFECTS OF
ELECTROPHILIC LIQUID INJECTION INTO RE-ENTRY PLASMA SHEATHS

FORTHAN - 2000 - Form

11.1.2.1	3.00E-2	4.80E-8	1.23E-2	4.40E-7	2.922E-6	2.922E-6	1.45E-2	0.0
11.1.2.1	0.69	8.25E-3	3410.	3.95E+5				
	2.40E-2	2.70E-8	1.53E-2	8.90E-6	5.264E-6	5.264E-6	1.45E-2	0.0
	1.04	8.25E-3	3840.	4.51E+5				
	2.03E-2	2.90E-7	1.72E-2	4.70E-5	7.927E-6	7.927E-6	1.45E-2	0.0
	1.47	8.25E-3	4500.	5.01E+5				
	1.55E-2	2.20E-6	1.95E-2	1.90E-4	1.084E-5	1.084E-5	1.29E-2	0.0
	3.12	7.80E-3	5320.	5.80E+5				
	5.40E-3	9.90E-5	2.38E-2	1.90E-3	7.958E-7	7.958E-7	6.40E-3	0.0
	4.86	1.23E-2	5825.	6.38E+5				
	2.30E-4	3.60E-3	2.66E-2	1.30E-3				
	6.48							
10.2	0.15	6.37E-3	1700.	1.00E5	4.526E-8	4.526E-8	1.45E-2	0.0
11.2.1.1	3.00E-2	4.80E-8	1.23E-2	4.4E-7				
	0.83	6.60E-3	3300.	4.01E5	1.980E-6	1.980E-6	1.45E-2	0.0
	2.40E-2	2.70E-8	1.53E-2	8.9E-6	3.659E-6	3.659E-6	1.45E-2	0.0
	1.21	6.60E-3	3720.	4.61E5				
	2.03E-2	2.90E-7	1.72E-2	4.7E-5	5.214E-6	5.214E-6	1.45E-2	0.0
	1.68	6.60E-3	4215.	5.11E5				
	1.55E-2	2.20E-6	1.95E-2	1.9E-4	7.659E-6	7.659E-6	1.32E-2	0.0
	3.39	6.18E-3	5025.	5.90E5				
	5.40E-3	9.90E-5	2.38E-2	1.9E-3	4.782E-7	4.782E-7	6.60E-3	0.0
	5.42	7.85E-3	5825.	6.50E5				
	2.30E-4	3.60E-3	2.66E-2	1.3E-3				

IBM

FURTRAN Coding Form.

3 7

10.3		11.3.1.1		10.4		11.4.1.1	
11.54	5.73E-3	1775.	1.00E5	4.971E-8	4.971E-8	1.45E-2	0.0
0.16	4.8E-8	1.23E-2	4.4E-7	1.915E-6	1.915E-6	1.45E-2	0.0
3.00E-2	5.83E-3	3265.	4.03E5	3.378E-6	3.378E-6	1.45E-2	0.0
0.93	2.7E-8	1.53E-2	8.9E-6	4.757E-6	4.757E-6	1.45E-2	0.0
2.40E-2	5.83E-3	3690.	4.66E5	6.840E-6	6.840E-6	1.33E-2	0.0
1.33	2.9E-7	1.72E-2	4.7E-5	4.479E-7	4.479E-7	6.65E-3	0.0
2.03E-2	5.83E-3	4075.	5.16E5	5.714E-8	5.714E-8	1.45E-2	0.0
1.80	2.2E-6	1.95E-2	1.9E-4	1.729E-6	1.729E-6	1.45E-2	0.0
1.55E-2	5.59E-3	4900.	5.93E5	2.876E-6	2.876E-6	1.45E-2	0.0
3.45	9.9E-5	2.38E-2	1.9E-3	4.009E-6	4.009E-6	1.45E-2	0.0
5.40E-3	6.62E-3	4925.	6.54E5	5.814E-6	5.814E-6	1.35E-2	0.0
5.65	3.6E-3	2.66E-2	1.3E-3				
2.30E-4							
21.67							
0.19	4.90E-3	1900.	1.00E5				
3.00E-2	4.8E-8	1.23E-2	4.4E-7				
1.11	4.80E-3	3200.	4.07E5				
2.40E-2	2.7E-8	1.53E-2	8.9E-6				
1.52	4.80E-3	3635.	4.70E5				
2.03E-2	2.9E-7	1.72E-2	4.7E-5				
2.02	4.80E-3	3875.	5.23E5				
1.55E-2	2.2E-6	1.95E-2	1.9E-4				
3.56	4.78E-3	4700.	6.00E5				
5.40E-3	9.9E-5	2.38E-2	1.9E-3				

IBM

FORTRAN Coding Sheet

4 7

FORTRAN STATEMENT

10.5	2.30E-4	5.30E-3	4600.	6.60E5	3.982E-7	3.982E-7	6.73E-3	0.0
	31.79	3.6E-3	2.66E-2	1.3E-3				
11.5.1.1	0.21	4.16E-3	2025.	9.90E4	6.320E-8	6.320E-8	1.45E-2	0.0
	3.00E-2	4.8E-8	1.23E-2	4.4E-7				
	1.25	4.08E-3	3160.	4.00E5	1.572E-6	1.572E-6	1.45E-2	0.0
	2.40E-2	2.7E-8	1.53E-2	8.9E-6				
	1.67	4.08E-3	3585.	4.70E8	2.516E-6	2.516E-6	1.45E-2	0.0
	2.03E-2	2.9E-7	1.72E-2	4.7E-5				
	2.18	4.08E-3	3740.	5.25E5	3.463E-6	3.463E-6	1.45E-2	0.0
	1.55E-2	2.2E-6	1.95E-2	1.9E-4				
	3.68	4.16E-3	4565.	6.05E5	4.957E-6	4.957E-6	1.37E-2	0.0
	5.40E-3	9.9E-5	2.38E-2	1.9E-3				
	6.25	4.48E-3	4400.	6.65E5	3.637E-7	3.637E-7	6.80E-3	0.0
	2.30E-4	3.6E-3	2.66E-2	1.3E-3				
10.6	41.92							
11.6.1.1	0.24	3.70E-3	2130.	9.70E4	6.663E-8	6.663E-8	1.45E-2	0.0
	3.00E-2	4.8E-8	1.23E-2	4.4E-7				
	1.36	3.55E-3	3030.	4.03E5	1.461E-6	1.461E-6	1.45E-2	0.0
	2.40E-2	2.7E-8	1.53E-2	8.9E-6				
	1.80	3.55E-3	3540.	4.65E5	2.228E-6	2.228E-6	1.45E-2	0.0
	2.03E-2	2.9E-7	1.72E-2	4.7E-5				
	2.33	3.55E-3	3650.	5.22E5	3.093E-6	3.093E-6	1.45E-2	0.0
	1.55E-2	2.2E-6	1.95E-2	1.9E-4				

IBM

FORTRAN Coding Form

5 7

FORTRAN STATEMENT

5	3.80	3.70E-3	4460.	6.00E5	4.241E-6	4.241E-6	1.38E-2	0.0
	40E-3	9.9E-5	2.38E-2	1.9E-3				
	6.45	3.90E-3	4255.	6.68E5	3.689E-7	3.689E-7	6.80E-3	0.0
	2.30E-4	3.6E-3	2.66E-2	1.3E-3				
10.7	52.04							
11.7.1.1	0.26	3.35E-3	2225.	9.35E4	6.696E-8	6.696E-8	1.45E-2	0.0
	3.00E-2	4.8E-8	1.23E-2	4.4E-7				
	1.44	3.22E-3	3115.	3.97E5	1.201E-6	1.201E-6	1.45E-2	0.0
	2.40E-2	2.7E-8	1.53E-2	8.9E-6				
	1.90	3.22E-3	3495.	4.55E5	1.776E-6	1.776E-6	1.45E-2	0.0
	2.03E-2	2.9E-7	1.72E-2	4.7E-5				
	2.44	3.22E-3	3585.	5.18E5	2.732E-6	2.732E-6	1.45E-2	0.0
	1.55E-2	2.2E-6	1.95E-2	1.3E-4				
	3.91	3.36E-3	4375.	6.11E5	3.700E-6	3.700E-6	1.39E-2	0.0
	5.40E-3	9.9E-5	2.38E-2	1.9E-3				
	6.62	3.50E-3	4150.	6.71E5	3.706E-7	3.706E-7	6.80E-3	0.0
	2.30E-4	3.6E-3	2.66E-2	1.3E-3				
10.8	72.29							
11.8.1.1	0.31	2.88E-3	2385.	8.70E4	6.046E-8	6.046E-8	1.45E-2	0.0
	3.00E-2	4.8E-8	1.23E-2	4.4E-7				
	1.54	2.77E-3	3110.	3.78E5	1.219E-6	1.219E-6	1.45E-2	0.0
	2.40E-2	2.7E-8	1.53E-2	8.9E-6				
	2.02	2.77E-3	3415.	4.28E5	1.784E-6	1.784E-6	1.45E-2	0.0
	2.03E-2	2.9E-7	1.72E-2	4.7E-5				

IBM

6 7

10.9	2.60	2.77E-3	3540.	4.98E5	2.278E-6	2.278E-6	1.45E-2	0.0
11.9.1.1	1.55E-2	2.2E-6	1.95E-2	1.9E-4				
	4.19	2.95E-3	4270.	6.15E5	3.109E-6	3.109E-6	1.41E-2	0.0
	5.40E-3	9.9E-5	2.38E-2	1.9E-3				
	6.87	3.02E-3	4025.	6.74E5	3.923E-7	3.923E-7	6.80E-3	0.0
	2.30E-4	3.6E-3	2.66E-2	1.3E-3				
	87.48							
	0.33	2.70E-3	2335.	8.50E4	5.083E-8	5.083E-8	1.45E-2	0.0
	3.00E-2	4.8E-8	1.23E-2	4.4E-7				
	1.58	2.59E-3	3125.	3.59E5	1.079E-6	1.079E-6	1.45E-2	0.0
	2.40E-2	2.7E-8	1.53E-2	8.9E-6				
	2.10	2.59E-3	3370.	4.19E5	1.541E-6	1.541E-6	1.45E-2	0.0
	2.03E-2	2.9E-7	1.72E-2	4.7E-5				
	2.68	2.59E-3	3555.	4.87E5	1.874E-6	1.874E-6	1.45E-2	0.0
	1.55E-2	2.2E-6	1.95E-2	1.9E-4				
	4.33	2.80E-3	4225.	6.17E5	2.586E-6	2.586E-6	1.41E-2	0.0
	5.40E-3	9.9E-5	2.38E-2	1.9E-3				
	6.99	2.82E-3	3975.	6.75E5	3.861E-7	3.861E-7	6.80E-3	0.0
	2.30E-4	3.6E-3	2.66E-2	1.3E-3				
10.10	100.0							
11.10.1.1	0.34	2.60E-3	2265.	8.50E4	4.185E-8	4.185E-8	1.45E-2	0.0
	3.00E-2	4.8E-8	1.23E-2	4.4E-7				
	1.62	2.55E-3	3130.	3.39E5	9.697E-7	9.697E-7	1.45E-2	0.0
	2.40E-2	2.7E-8	1.53E-2	8.9E-6				

IBM

FURTHAN Loading Form

7 7

FINANCIAL STATEMENT

2.15	2.55E-3	3330.	4.05E5	1.374E-6	1.374E-6	1.45E-2	0.0
2.03E-2	2.9E-7	1.72E-2	4.7E-5				
2.75	2.55E-3	3590.	4.82E5	1.591E-6	1.591E-6	1.45E-2	0.0
1.55E-2	2.2E-6	1.95E-2	1.9E-4				
4.48	2.73E-3	4200.	6.19E5	2.178E-6	2.178E-6	1.42E-2	0.0
5.40E-3	9.9E-5	2.38E-2	1.9E-3				
7.08	2.73E-3	3950.	6.77E5	3.889E-7	3.889E-7	6.40E-3	0.0
2.30E-4	3.6E-3	2.66E-2	1.3E-3				
0.0							

Duplicate above data set for vapor pressure breakup calculations (We < WeCr)

12

12

APPENDIX C

SAMPLE OUTPUT,* Version C[†]

(see pages E-2 - E-14)

* Sample output is not given for Version D since it was not completely debugged at the time this report was prepared.

† These results were obtained with π inadvertently omitted from the denominator in Eqs. (18) and (20). When this correction is made in the code, values of s_{bk} and u_{bk} will be correspondingly reduced.

APPENDIX D

SAMPLE OUTPUT-EXPLANATIONS

VERSION C

- Line 1 Input data on Card 1
- Line 2 Input data on Card 2
- Lines 3-15 Input data on Cards 3 and 4
- Line 17 Input data on Card 5
- Line 19 Input data on Card 6
- Line 20 Input data on Card 7
- Lines 22-23 Input data on Cards 8 and 9
- Lines 25-36 Starts listing input for first value of S (Card-types 10 and 11)
- J = streamline number (first streamline is designated J = 2)
- Y = normal distance from body (cm)
- PE = pressure (dyne/cm²)
- TE = temperature (°K)
- UE = velocity (cm/sec)
- FE(I) = F_i (moles/g of ith species)--in order NO⁺, e⁻, O, F⁻,
 N, O₂, N₂, NO
- Line 38 AV = integrated average properties (\bar{G}) from the body (Y = 0)
 to the position of the last streamline (Y = Y_{max});
- $$\bar{G}(S) = \frac{1}{Y_{\max}} \int_0^{Y_{\max}} G(s,y) dy$$
- First column gives input value of S (Card-type 10) at
 which average properties are computed.
- Line 41 I = Input data set number
- WA = mixture molecular weight
- HE = static enthalpy, cal/g
- HSE = stagnation enthalpy, cal/g
- REUE = $\bar{\rho}_e \bar{u}_e$ (g/cm²-sec)
- REUE2 = $\bar{\rho}_e \bar{u}_e^2$ (g/cm-sec²)

$$\begin{aligned} \text{REUEHSE} &= \bar{\rho}_e \bar{u}_e \bar{h}_{s,e} \text{ (cal/cm}^2\text{-sec)} \\ \text{RUF1E} &= \bar{\rho}_e \bar{u}_e \bar{F}_{\text{NO}^+} \text{ (moles/cm}^2\text{-sec)} \\ \text{RUF2E} &= \bar{\rho}_e \bar{u}_e \bar{F}_{e^-} \text{ (moles/cm}^2\text{-sec)} \\ \text{RUF3E} &= \bar{\rho}_e \bar{u}_e \bar{F}_O \text{ (moles/cm}^2\text{-sec)} \\ \text{RUF4E} &= \bar{\rho}_e \bar{u}_e \bar{F}_{F^-} \text{ (moles/cm}^2\text{-sec)} \\ \text{RUF5E} &= \bar{\rho}_e \bar{u}_e \bar{F}_N \text{ (moles/cm}^2\text{-sec)} \\ \text{RUF6E} &= \bar{\rho}_e \bar{u}_e \bar{F}_{O_2} \text{ (moles/cm}^2\text{-sec)} \\ \text{RUF7E} &= \bar{\rho}_e \bar{u}_e \bar{F}_{N_2} \text{ (moles/cm}^2\text{-sec)} \\ \text{RUF8E} &= \bar{\rho}_e \bar{u}_e \bar{F}_{\text{NO}} \text{ (moles/cm}^2\text{-sec)} \end{aligned}$$

Line 42

Line 46 and
following

Input for second value of S

⋮

through tenth value of S (I = 10)

*CONDITIONS IN FRONT OF SHOCK (integrated average values from $Y = 0$ to $Y = Y_{\text{max}}$ at the injection site, $S = 0$)

PX = pressure (dyne/cm²)
 UX = velocity
 HSX = stagnation enthalpy, cal/g
 RHOX = density, g/cm³
 TX = static temperature, °K
 WAX = mixture molecular weight

MASS FRACTIONS F(I) = mass fractions (= $W_i F_i$) of species in following order NO⁺, e⁻, O, F⁻, N, O₂, N₂, Q (vapor)

*CONDITIONS BEHIND SHOCK

PY = pressure (dyne/cm²)
 UY = velocity (cm/sec)
 HY = static enthalpy (cal/g)
 RHOY = density (g/cm³)
 TY = static temperature (°K)
 TBPY = boiling point temperature at PY

***JET BREAK-UP INFORMATION**

WE = Weber number
 REJ = Jet Reynold's Number[†]
 TBK = Jet breakup time (sec)
 UBK = Jet velocity in s direction at breakup (cm/sec)
 XBK = S breakup distance (cm)
 YBK = Distance normal to body at breakup (cm)
 WEL = Liquid jet Weber number[†]
 D30 = Droplet diameter based on mass (cm)[†]
 D32 = Mean droplet diameter (cm)
 DMAX = Maximum droplet diameter (cm)[†]

****MODIFIED FREE STREAM CONDITIONS IN SHOCK FRONT BY SINGLE ITERATION****
 (integrated average values--from Y = 0 to Y = YBK (above)--upstream of shock at the injection site, S = 0

Repeat of Lines 38 to 44, with new integrated average values

***CONDITIONS IN FRONT OF SHOCK**

Repeat of above with new integrated average values

***CONDITIONS BEHIND SHOCK**

Repeat of above with new values

***JET BREAKUP INFORMATION[§]**

Repeat of above with new values

INITIAL FLOW RATE (G/SEC)

XMV = Mass flow of " flashed " vapor (g/sec)
 XML = Mass flow of liquid (g/sec)
 XMA = Mass flow of entrained air (g/sec)
 XMM = Mass flow of entrained air plus vapor (g/sec)
 AE = Initial entrainment area (cm²)

† Set = 0 in first calculation of Jet Breakup Information if $We < We_{cr}$.

§ Always assumes aerodynamic breakup in calculations for first data set.

**CONDITIONS 0. THE INITIAL MIXTURE AT S =

(The following are initial conditions for the DVA tube calculations at the position of jet breakup (XBK))

U = Velocity (cm/sec)
 HSV = Stagnation enthalpy of vapor (cal/g)
 HSR = Relative stagnation enthalpy of droplets (cal/g)
 H = Static enthalpy (cal/gm)
 TSR = Relative stagnation temperature of droplets (°K)
 RHO = Density (g/cm³)
 A = DVA tube area (cm²)

MASS FRACTIONS OF SPECIES ($Y_i = F_i W_i$) (given in the usual order)

P = Pressure (dyne/cm²)
 WM = Mixture molecular weight
 RHO = Density (g/cm³)
 HS = Stagnation enthalpy (cal/g)
 FV = Mass fraction of vapor (mass flow of vapor/mass flow of air + vapor)
 UP = Droplet velocity in S direction (cm/sec)
 VP = Droplet velocity in Y direction (cm/sec)
 TP = Droplet (boiling point) temperature (°K)
 RP = Droplet radius (cm)
 SE = Lateral spread distance (cm)
 YE = Penetration distance (cm)

INITIAL THERMODYNAMIC INFORMATION

GAMA = Specific heat ratio
 CP = Specific heat
 SS = Speed of sound (cm/sec)
 MA = Gas Mach number
 MP = Droplet Mach number
 REP = Droplet Reynold's number

The results of the DVA tube calculations follow

RUN IDENTIFICATION

*S(cm) = Distance from injection site

All symbols same as above, with the addition of:

RHOP = Particle cloud density (g/cm³)

ELECTRON DENSITY = (cm⁻³)

The following line gives the values of F_i (moles/g) for each species in the same order as the mass fractions.

CHANGES OF MASS FRACTION OVER DS =

The following line gives the changes of mass fraction (derivatives) over the present integration step, in the usual order--ending with the mass fraction of Q (vapor).

The following quantities, starting with DP are evaluated at the previous value of S

- DP = Pressure derivative (dynes/cm²)
- DRHO = Density derivative (gm/cm³)
- DU = Velocity derivative (cm/sec)
- DHS = Stagnation enthalpy derivative (cal/g)
- DFV = Vapor mass fraction derivative
- PHADS = (PHIV) (A) (DS) (g/cm³-sec)
- DA = DVA tube area derivative (cm²)
- DAE = Entrainment area derivative (cm²)
- DUP = Droplet velocity derivative (cm/sec)
- DRP = Droplet radius derivative (cm)
- QP = Free molecular heating rate (erg/sec-cm²)
- PHIV = Vaporization rate (g/cm³-sec)
- DFP = DUP/(RHOP)(UP)(A)
- DQ = Total heat transfer over DS (ergs)

PRODUCTION RATES FOR EACH REACTION (MOL/SEC-CM3)

The following line gives the molar production rates of Reactions 1-8 (see Section VII.A) in order.

PRODUCTION RATES FOR EACH SPECIE

The following line gives the molar production rates (mole/sec-cm³) for NO⁺, e⁻, O, F⁻, and N, in order.

LOCAL THERMODYNAMIC INFORMATION

The first line following gives the same data as noted above at the local value of S, with the addition of, # P/sec = total number of droplets/sec within the DVA tube.

The second line following gives the following information in order: DS, ETA, SE, YE, VP, XX,a,XX, XX, XX, where

DS	=	Present integration step size (cm)
ETA	=	Used for heterogeneous rate coefficients (see Section VII.A)
SE	=	Lateral spread (cm)
YE	=	Penetration depth (cm)
VP	=	Droplet velocity normal to body (cm/sec)
a	=	Exponent in expression for lateral spread, Eq. 30
XX	=	Not used in present version of program

After printout at maximum integration distance (Card 6, cols. 21-30) program will print, AERODYNAMIC BREAK-UP IS IMPOSSIBLE, WE = (if $We < We_{cr}$). Then, if a second data set is included, all calculations will be repeated using the vapor pressure breakup equations to establish initial conditions for the DVA tube calculations.

APPENDIX E

FORTRAN LISTINGS*
(AND SAMPLE OUTPUT)

SAMPLE OUTPUT	p. E-2
VERSION C	p. E-15
VERSION D	p. E-42

* These listings do not contain π in the denominator in Eqs. (18) and (20). Corrections should be made to the code in subroutine BREAK; expressions for UBK (version C), lines 108 and 139 and XBK (version C), lines 112 and 142; UBK (version D), lines 111 and 143 and XBK (version D), lines 115 and 146.

***** J1 3 RAM C-III 2...KFT FREQN-E-3 (CASE 6-T-0=3-0) ***** 12/13/73 C
 .100E+1 .100E+1 .100E+1 .100E+1 .100E+1 .100E+1 .100E+1 .100E+1
 --- PROPERTIES OF THE INJECTED LIQUID ---
 MOLECULAR WT.= 518.00
 SPECIFIC HEAT (CAL/G-DEG) = 2.0E-1
 BOILING PT. (K) = 4.28E+02
 HEAT OF VAPORIZATION (CAL/G) = 1.450E+1
 DENSITY (G/CM3) = 1.723E+00
 MASS FLUX (G/SEC) = 1.520E+1
 JET VELOCITY (CM/SEC) = 2.400E+03
 JET ANGLE (RAD.) = 6.007E-01
 TEMPERATURE (K) = 3.900E+02
 SURFACE TENSION (DYN/CM) = 2.800E+02
 VISCOSITY (GM/CM-SEC) = 2.200E-02
 ORIFICE DIAMETER (CM) = 5.100E-02
 LIQUID DROP COEFFICIENTS
 AK= 6.012E+2
 BK= 2.444E-02

NPS= 10 NDL= 7 JDS= 3 IBUG= -C NSOF= 3
 INITIAL S= 3.000 INCREMENT= 1.000E-02 FINAL S= 1.000E+02 PRINT INCREMENT= 2.000E+00
 .300E+00 .100E+00 .333E+00 .622E+01
 C1= 1.100E-02 C2= 1.100E-03 C3= 3.000E-01 C4= 3.000E-10 C5= 2.000E-10 C6= 3.000E-10 C7= 1.000E-10 C8= 2.000E-10 C9= 1.000E-10
 N1= 1.000E+00 N2= 1.000E+00 N3= 2.000E+00 N4= 1.000E+00 N5= 1.000E+00 N6= 1.000E+00 N7= 1.000E+00 N8= 1.000E+00 N9= 1.000E+00 N10= 1.000E+00
 FF= 1.000E+00
 FF1= 1.000E+00
 FF2= 1.000E+00
 FF3= 1.000E+00
 FF4= 1.000E+00
 FF5= 1.000E+00
 FF6= 1.000E+00
 FF7= 1.000E+00
 FF8= 1.000E+00
 FF9= 1.000E+00
 FF10= 1.000E+00

J	Y	PE	TE	JE	FE(I)	(MOLE/GM)	REUEN	REUEN	REUEN	RUF2E	RUF3E	
2	.11	7.301E+03	1.511E+03	1.000E+05	3.900E-08	3.900E-08	1.450E-12	1.450E-12	1.450E-12	4.800E-08	1.230E-02	4.400E-07
3	.09	5.317E+03	3.100E+03	3.950E+05	2.922E-06	2.922E-06	1.450E-12	1.450E-12	1.450E-12	2.400E-02	2.700E-08	1.530E-06
4	1.04	8.357E+03	3.340E+03	4.510E+05	5.264E-06	5.264E-06	1.450E-12	1.450E-12	1.450E-12	2.030E-02	2.900E-07	1.720E-02
5	1.47	8.357E+03	4.500E+03	2.010E+05	7.927E-06	7.927E-06	1.450E-12	1.450E-12	1.450E-12	1.000E-02	2.200E-06	1.950E-02
6	3.12	7.911E+03	5.320E+03	5.800E+05	1.064E-05	1.064E-05	1.294E-02	1.294E-02	1.294E-02	5.400E-03	9.900E-05	2.380E-02
7	4.06	1.246E+04	5.015E+03	5.836E+05	7.998E-7	7.998E-7	6.400E-03	6.400E-03	6.400E-03	2.300E-04	3.600E-03	2.550E-02
AV=	0.00	9.661E+03	5.810E+03	5.804E+05	5.372E-06	5.372E-06	1.100E-02	1.100E-02	1.100E-02	5.257E-03	1.317E-03	2.259E-02

J	Y	PE	TE	JE	FE(I)	(MOLE/GM)	REUEN	REUEN	REUEN	RUF2E	RUF3E	
2	.15	6.000E+03	1.000E+03	1.000E+05	4.526E-08	4.526E-08	1.450E-12	1.450E-12	1.450E-12	4.900E-08	1.230E-02	4.400E-07
3	.03	6.686E+03	3.300E+03	4.010E+05	1.900E-06	1.900E-06	1.450E-12	1.450E-12	1.450E-12	2.400E-02	2.700E-08	1.530E-06
4	1.01	6.686E+03	3.720E+03	4.010E+05	3.609E-06	3.609E-06	1.450E-12	1.450E-12	1.450E-12	2.030E-02	2.900E-07	1.720E-02
5	1.58	6.000E+03	4.215E+03	5.811E+05	9.214E-06	9.214E-06	1.450E-12	1.450E-12	1.450E-12	1.000E-02	2.200E-06	1.950E-02
6	3.39	6.250E+03	5.000E+03	5.800E+05	7.659E-6	7.659E-6	1.324E-02	1.324E-02	1.324E-02	5.400E-03	9.900E-05	2.380E-02
7	5.42	7.952E+03	5.225E+03	6.500E+05	4.782E-07	4.782E-07	6.600E-03	6.600E-03	6.600E-03	2.300E-04	3.600E-03	2.550E-02
AV=	6.48	7.000E+03	4.500E+03	5.811E+05	5.550E-06	5.550E-06	1.100E-02	1.100E-02	1.100E-02	5.342E-03	1.317E-03	2.259E-02

I	WA	ME	MSE	REUE	FEUE2	REUEHSE	RJF1E	RJF2E	RUF3E
2	2.255E+01	RUF5E	RUF6E	RUF7E	RUF8E	1.20E+3	7.419E-7	7.419E-7	2.324E-3
2	1.6	PE	UE	FE(I) (MOLE/GM)=					
2	5.504E+03	1.775E+03	1.000E+05	4.971E-08	4.971E-08	1.450E-02	3.400E-02	4.800E-02	1.23 E-2
3	0.906E+03	3.20E+03	0.30E+05	1.91E-05	1.91 E-05	1.450E-02	2.400E-02	2.700E-02	1.530E-02
4	5.906E+03	3.550E+03	4.660E+05	3.379E-06	3.379E-06	1.450E-02	2.030E-02	2.900E-02	1.720E-02
5	5.906E+03	4.75E+03	5.16E+05	4.757E-06	4.757E-06	1.450E-02	1.000E-02	2.200E-02	1.000E-02
6	5.663E+03	4.800E+03	5.930E+05	5.840E-06	5.840E-06	1.330E-02	5.400E-03	9.800E-05	2.380E-02
7	6.706E+03	4.825E+03	6.540E+05	4.473E-07	4.473E-07	6.650E-03	2.30E-4	3.60E-3	2.66 E-2
AV=	5.10E+03	5.120E+05	3.047E-06	1.101E-02	1.101E-02	1.101E-02	9.452E-03	1.421E-03	2.242E-02

I	WA	ME	MSE	REUE	FEUE2	REUEHSE	RUF1E	RUF2E	RUF3E
3	2.255E+01	RUF5E	RUF6E	RUF7E	RUF8E	1.161E+03	5.845E-07	5.845E-07	2.113E-03
3	1.11	PE	UE	FE(I) (MOLE/GM)=					
2	4.862E+03	1.300E+03	1.000E+05	5.714E-08	5.714E-08	1.450E-02	3.400E-02	4.800E-02	1.230E-02
3	4.862E+03	3.200E+03	4.070E+05	1.729E-06	1.729E-06	1.450E-02	2.400E-02	2.700E-02	1.530E-02
4	4.862E+03	3.550E+03	4.760E+05	2.879E-06	2.879E-06	1.450E-02	2.030E-02	2.900E-02	1.720E-02
5	4.862E+03	3.750E+03	5.230E+05	4.009E-06	4.009E-06	1.450E-02	1.550E-02	2.200E-02	1.950E-02
6	4.842E+03	4.70E+03	6.00E+05	5.814E-06	5.814E-06	1.351E-02	5.40E-3	9.900E-05	2.380E-02
7	5.363E+03	4.800E+03	6.600E+05	3.982E-07	3.982E-07	6.730E-03	2.300E-04	3.600E-03	2.660E-02
AV=	5.063E+03	4.890E+05	2.435E-06	1.100E-02	1.100E-02	1.100E-02	9.732E-03	1.474E-03	2.227E-02

I	WA	ME	MSE	REUE	FEUE2	REUEHSE	RUF1E	RJF2E	RUF3E
4	2.247E+01	RUF5E	RUF6E	RUF7E	RUF8E	1.001E+03	4.063E-07	4.063E-07	1.835E-03
4	2.214E+03	2.325E+03	3.900E+04	5.323E-08	5.323E-08	1.450E-02	3.600E-02	4.800E-02	1.23 E-2
3	0.133E+03	3.100E+03	0.860E+05	1.572E-06	1.572E-06	1.450E-02	2.400E-02	2.700E-02	1.530E-02
4	0.133E+03	3.365E+03	4.700E+05	2.515E-06	2.515E-06	1.450E-02	2.030E-02	2.900E-02	1.720E-02
5	0.133E+03	3.74E+03	5.25E+05	3.463E-06	3.463E-06	1.450E-02	1.000E-02	2.200E-02	1.000E-02
6	0.214E+03	4.565E+03	6.050E+05	4.957E-06	4.957E-06	1.370E-02	5.400E-03	9.800E-05	2.380E-02
7	4.630E+03	4.800E+03	6.650E+05	3.637E-07	3.637E-07	6.600E-03	2.300E-4	3.600E-3	2.660E-2
AV=	0.322E+03	0.310E+05	2.036E-06	1.100E-02	1.100E-02	1.100E-02	9.940E-03	1.491E-03	2.215E-02

I	WA	ME	MSE	REUE	FEUE2	REUEHSE	RUF1E	RJF2E	RUF3E
4	2.247E+01	RUF5E	RUF6E	RUF7E	RUF8E	1.001E+03	4.063E-07	4.063E-07	1.835E-03

[F]

ω

J	Y	PE	TE	UE	FE(I) (MOLE/GM)	REUSE	RUF1E	RUF2E	RUF3E
2	1.33	2.735E+03	2.439E+03	8.500E+04	5.003E+06	5.003E+06	1.450E+02	3.000E+02	1.23E-02
3	1.6	2.02E+03	3.12E+03	3.990E+03	1.079E+06	1.079E+06	1.450E+02	2.400E+02	1.53E-02
4	2.10	2.624E+03	3.370E+03	4.191E+03	1.541E+06	1.541E+06	1.450E+02	2.000E+02	1.72E-02
5	2.00	2.524E+03	3.555E+03	4.07E+03	1.074E+06	1.074E+06	1.450E+02	1.900E+02	1.900E+03
6	4.33	2.030E+03	4.225E+03	5.174E+03	2.506E+06	2.506E+06	1.410E+02	5.400E+03	2.34E-02
7	5.9	2.857E+03	3.475E+03	5.755E+03	3.061E+07	3.061E+07	6.000E+03	2.30E+04	2.65E-02
AV=	8.8	2.730E+03	3.20E+03	4.044E+03	1.211E+06	1.211E+06	1.136E+02	9.709E+03	2.17E-02

I	MA	HE	HSE	REUE	REUE2	REUSE	RUF1E	RUF2E	RUF3E
9	2.212E+01	2.85E+03	5.658E+03	9.574E+02	4.639E+04	5.419E+02	1.160E+07	1.160E+07	1.000E+03
10	3.57E+01	1.322E+04	2.682E+03	2.682E+03	9.142E+05				

J	Y	PE	TE	UE	FE(I) (MOLE/GM)	REUSE	RUF1E	RUF2E	RUF3E
2	1.34	2.033E+03	2.27E+03	8.000E+04	4.105E+08	4.105E+08	1.450E+02	3.000E+02	1.23E-02
3	1.02	2.583E+03	3.130E+03	3.390E+03	9.697E+07	9.697E+07	1.450E+02	2.400E+02	1.53E-02
4	2.45	2.530E+03	3.30E+03	4.546E+03	1.374E+06	1.374E+06	1.450E+02	2.000E+02	1.72E-02
5	2.75	2.503E+03	3.394E+03	4.020E+03	1.591E+06	1.591E+06	1.450E+02	1.550E+02	1.95E-02
6	4.40	2.705E+03	4.200E+03	5.19E+03	2.170E+06	2.170E+06	1.420E+02	5.40E+03	2.30E-02
AV=	3.0	2.705E+03	3.45E+03	5.774E+03	3.089E+07	3.089E+07	6.000E+03	2.300E+04	2.65E-02
AV=	3.0	2.697E+03	3.70E+03	4.709E+03	1.079E+06	1.079E+06	1.140E+02	9.313E+03	2.16E-02

I	MA	HE	HSE	REUE	REUE2	REUSE	RUF1E	RUF2E	RUF3E
10	2.206E+01	2.873E+03	5.605E+03	9.239E+02	4.419E+04	5.179E+02	3.966E+08	9.365E+08	1.161E-03
11	3.158E+01	1.321E+04	2.681E+03	2.681E+03	6.000E+05				

CONDITIONS IN FRONT OF SHOCK
 PK= 9.661E+03 JX= 5.040E+05 MSX= 6.128E+03 RHOX= 5.237E-07 TX= 5.013E+03 MAX= 2.257E+01
 MASS FRACTIONS F(I)= 0.612E-04 2.348E-03 1.762E-01 ..
 1.155E-01 4.215E-02 6.321E-01 3.301E-02 ..

CONDITIONS BEHIND SHOCK
 PY= 1.102E+03 JY= 9.402E+04 MY= 6.422E+03 RHOY= 2.811E-06 TY= 1.141E+04 TBPY= 3.555E+02

JET BREAK-UP INFORMATION
 WE= 3.065E+03 REJ= 0.
 WEL= 0.
 U3= 0.
 TK= 1.127E+03 UBK= 1.800E+03 XBK= 1.751E+03 YBK= 1.915E+00
 U32= 2.935E+03 OMAX= 0.

----- MODIFIED FREE STREAM CONDITIONS IN SHOCK FRONT BY SINGLE ITERATION -----

J	Y	PE	TE	UE	FE(I) (MOLE/GM)	REUSE	RUF1E	RUF2E	RUF3E
AV=	8.8	6.37E+03	3.167E+03	4.598E+03	5.762E+06	5.762E+06	1.450E+02	1.947E+02	3.362E-07

I	MA	HE	HSE	REUE	REUE2	REUSE	RUF1E	RUF2E	RUF3E
1	4.435E+01	4.233E+03	6.76E+03	2.259E+01	1.139E+05	1.527E+03	1.302E+05	1.302E+06	3.27E-03
11	3.98E+03	7.93E+03	3.981E+03	1.22E+01					

CONDUCTIONS FROM FRONT OF SHOCK
P= 3.374E+03 J= 4.593E+05 MSK= 6.760E+05 RHO= 4.312E-07 TX= 3.967E+03 MAX= 1.935E+4
MASS FRACTION: F1= 2.714E-04 F2= 3.152E-05 F3= 2.320E-01 F4= 2.725E-01 F5= 1.075E-02
+ .93E-01 1.022E-03 0.

CONDUCTIONS BEHIND SHOCK
P= 3.775E+04 J= 1.08E+05 MY= 6.62E+3 KMJ= 1.84E-6 TY= 9.611E+03 TBPV= 3.474E+02

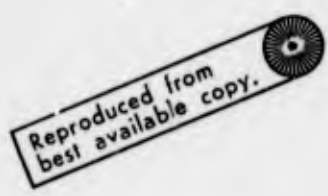
JET BREAK-UP INFORMATION
M= 1.37E+04 KEJ= 8.42E+04 TPK= 1.107E+03 USK= 1.926E+03 XBK= 1.901E+01 YBK= 2.064E+01
WEL= 5.295E+05 U3= 5.34E-03 D32= 7.475E-03 DMAX= 1.739E-2

INITIAL FLOW RATE (G/SEC)
XMY= 3.315E+01 XML= 5.345E+01 XMA= 4.653E-01 XMM= 6.361E+01 AE= 2.063E+02

CONDITIONS OF THE INITIAL MIXTURE AT S= 1.000
U= 2.54E+02 MSV= 9.893E+01 MSK= 1.363E+02 M= 1.353E+02 TSR= 3.432E+02 F= 3.431E+02 RHO= 1.630E-03 A= 2.045E+01
MASS FRACTIONS OF SPECIES
NO= 3.619E-07 E= 1.759E-11 O= 1.291E-03 F= 1.0 N= 1.517E-03 O2= 5.335E-08 N2= 2.745E-3 MO= 9.17E-6 FV= 9.944E-01

P= 8.315E+04 MM= 5.273E+02 RHO= 1.630E-03 MS= 1.360E+02 FV= 9.944E-01 UP= 1.926E+03 VP= 1.866E+03
TP= 3.474E+02 RP= 3.737E-03 SE= 6.354E-01 YE= 2.164E+00

INITIAL THERMODYNAMIC INFORMATION
XMA= 1.011 O(CAL/G)= 3.480E-01 SS= 7.395E+03 MA= 3.312E-01 MAP= 2.642E-01 REP= 7.113E+01



***** J108 RAM C-III 200KFT FREDON-E-3 (CASE 6-TLO=360)**** 12/13/73 C

S(CM) = 1.9-20
T(K) = 343.16 P(OYNE/CM2) = 8.815E+04 K40(GM/ML) = 1.529E-03 U(CM/SEC) = 2.511E+03 HS(CAL/GM) = 1.361E+02 FV = 3.944E-01 WM = 5.272E+02
A = 2.144E+01 UP = 1.926E+03 TP = 3.474E+02 RP = 3.737E-03 RHOP = 1.740E-03 AE = 2.562E+00 TSR = 3.432E+02 HF = 1.35 E+2

MASS FRACTIONS OF SPECIES
UO = 3.645E-07 E = 2.999E-17 O = 1.292E-03 F = 5.599E-07 N = 1.517E-03 O2 = 1.231E-07 N2 = 2.743E-03 MO = 3.470E-06

ELECTRON DENSITY = 3.144E+13
.321E-07 .321E-07 .807E-04 .1.0E-03 .107E-08 .981E-04 .301E-05

CHANGES OF MASS FRACTION OVER DS = 1.000E-03
-7.730E-03 -1.759E-11 1.250E-06 5.559E-07 5.000E-07 6.907E-06 2.986E-06 5.849E-09 -5.513E-05
DP = -1.139E+01 DRHU = -5.823E-06 DU = .300E+01 DHS = .364E-01 DFV = -5.502E-05 PHADS = -.471E-04
DA = -1.049E-02 DAE = 4.433E-03 DUP = 3.870E-03 DRP = 8.29E-10 O2P = -4.103E-02 PHIV = -2.304E-03
DUP = 5.111E-01 DQ = -3.549E-15

PRODUCTION RATES FOR EACH REACTION (MOL/SEC-CM3)
2.545E-03 .0 1.827E-06 .0 2.113E+02 .0 .0 .0 .0

PRODUCTION RATES FOR EACH SPECIE
-2.547E-03 -2.133E+01 2.847E-03 2.133E+02 2.847E-03

LOCAL THERMODYNAMIC INFORMATION

GAMA = 1.011 CP(CAL/G) = 3.480E+01 SS = 7.396E+03 MA = 3.391E-01 MAP = 2.542E-01 REP = 7.013E+01 P/SEC = 1.827E+09
.1.0E-02 .330E+01 .636E+00 .2.7E+01 .187E+04 .192E+13 .333E+00 .177E+22 .157E+01 .540-210

S(CM) = 1.9-20
T(K) = 524.04 P(OYNE/CM2) = 8.525E+04 RHO(GM/ML) = 7.965E-04 U(CM/SEC) = 9.712E+03 HS(CAL/GM) = 2.253E+02 FV = 9.603E-01 WM = 3.915E+02
A = 1.120E+01 UP = 2.071E+03 TP = 3.474E+02 RP = 3.705E-03 RHOP = 2.692E-03 AE = 7.726E+00 TSR = 5.157E+02 HF = 2.024E+2

MASS FRACTIONS OF SPECIES
UO = 5.147E-08 E = 2.255E-19 O = 4.205E-03 F = 5.094E-08 N = 4.346E-03 O2 = 2.352E-04 N2 = 1.058E-02 MO = 2.358E-04

ELECTRON DENSITY = 1.959E+05
.258E-08 .449E-15 .262E-03 .310E-03 .838E-05 .377E-03 .783E-03

CHANGES OF MASS FRACTION OVER DS = 5.533E-03
2.176E-10 1.374E-21 6.025E-06 1.377E-10 7.131E-06 1.106E-06 2.344E-05 8.567E-07 -4.057E-05
DP = -5.68E+01 DRHU = -1.139E-05 DU = .205E+02 DHS = .252E+00 DFV = -4.35E-04 PHADS = .134E-01
DA = -2.042E-03 DAE = 1.741E-02 DUP = 7.338E-01 DRP = -1.941E-07 O2P = 1.885E+03 PHIV = 1.757E-01
DUP = 4.936E+11 DQ = 3.539E-07

PRODUCTION RATES FOR EACH REACTION (MOL/SEC-CM3)
4.77E-13 2.427E-07 1.35E-12 3.969E-08 2.988E-07 .0 .0 .0 .0

PRODUCTION RATES FOR EACH SPECIE
-2.824E-07 -2.939E-07 2.824E-07 1.144E-06 1.442E-12

LOCAL THERMODYNAMIC INFORMATION

GAMA = 1.015 CP(CAL/G) = 3.471E+01 SS = 1.542E+04 MA = 9.299E-01 MAP = 7.519E-01 REP = 1.044E+02 P/SEC = 1.827E+09
.3.5E-02 .1.0E+01 .121E+01 .403E+01 .187E+04 .192E+13 .333E+00 .177E+22 .157E+01 .550-219

S(CM) = 1.9-20
T(K) = 582.52 P(OYNE/CM2) = 8.445E+04 RHO(GM/ML) = 5.038E-04 U(CM/SEC) = 1.310E+04 HS(CAL/GM) = 2.664E+02 FV = 9.736E-01 WM = 3.52E+02
A = 1.120E+01 UP = 2.210E+03 TP = 3.474E+02 RP = 3.660E-03 RHOP = 2.621E-03 AE = 1.074E+01 TSR = 5.164E+02 HF = 2.044E+2

MASS FRACTIONS OF SPECIES
UO = 1.190E-07 E = 5.502E-19 O = 5.502E-03 F = 7.535E-08 N = 5.446E-03 O2 = 4.634E-04 N2 = 1.452E-02 MO = 3.927E-04

ELECTRON DENSITY = 3.663E+05

-3.17E-10 7.5E-11 3.12E-10 6.08E-11 3.42E-10
 LOCAL THERMODYNAMIC INFORMATION
 JAMB= 1.442 CP(CAL/G)= 3.43E-11 SS= 3.53E+04 IA= 2.75E+04 MAP= 1.79E+01 REP= 2.14E+01 #P/SEC= 1.02E+01
 5.5E-11 8.5E-11 4.3E+01 1.92E+02 1.67E+04 1.92E+13 3.33E+03 1.77E+22 1.07E+01 5.1E-21

SICM)= 99.90+9
 I(K)= 1.45E+03 P(DYNE/CM2)= 7.44E+03 R(D/GM/ML)= 8.25E-05 U(CM/SEC)= 1.0E+01 M(SICAL/G)= 6.25E+22 EV= 6.70E+01 MW= 1.0E+02
 A= 2.50E+02 UP= 2.94E+04 TP= 3.47E+02 RP= 1.99E-03 RMOP= 1.54E-06 IE= 1.34E+02 ISE= 1.55E+03 ME= 7.34E+02

MASS FRACTIONS OF SPECIES
 40+ 9.17E-07 E= 2.89E-17 O= 2.257E-02 F= 3.909E-07 N= 1.692E-02 O2= 5.43E-03 N2= 7.582E-02 NO= 3.52E-03
 ---- ELECTRON DENSITY= 2.61E+05
 8.6E-7 5.6E-13 1.41E-2 2.0E-7 1.01E-2 1.64E-3 2.71E-02 1.18E-03

CHANGE OF MASS FRACTION OVER DS= 5.00E-02
 1.50E-13 1.59E-24 3.73E-06 9.93E-11 2.82E-06 6.69E-17 1.24E-05 5.60E-07 -2.10E-05
 UP= 1.59E+01 DRHO= 1.31E-06 O= 1.72E+02 OMS= 1.1E+03 O2V= 2.3E-04 PMADS= 2.1E-2
 OA= 1.0E-01 OAE= 1.30E+02 OUP= 1.92E+00 ORP= -1.33E-07 Q= 1.957E+03 PHIV= 1.83E-04
 DFP= 7.43E+11 DQ= 5.14E-08

PRODUCTION RATES FOR EACH REACTION (MOL/SEC-CM3)
 6.42E-15 2.60E-1 3.307E-13 9.672E-12 3.0E-2E-10 6.6E-2E-18 3.618E-12 6.990E-11

PRODUCTION RATES FOR EACH SPECIE
 -2.962E-13 -3.774E-11 2.894E-10 8.119E-11 2.650E-15

LOCAL THERMODYNAMIC INFORMATION
 JAMB= 1.442 CP(CAL/G)= 3.429E-11 SS= 3.308E+04 MA= 2.76E+00 MAP= 1.93E+01 REP= 2.098E+03 #P/SEC= 1.02E+01
 5.5E-11 8.5E-11 4.41E+1 1.94E+2 1.67E+4 1.92E+13 3.33E+03 1.77E+22 1.07E+01 5.1E-21

--- AERODYNAMIC BREAK-UP IS IMPOSSIBLE, WE= 3.807E+00

***** J108 RAN C-111 250KFT FREON-E-3 (CASE 6-TLO=360)*** 12/13/73 -

.1.0E+1 .1.0E+01 .1.0E+01 .5JJE+J2 .8JCE+CC
 --- PROPERTIES OF THE INJECTED LIQUID -----

MOLECULAR WT.= 518.00
 SPECIFIC HEAT(CAL/G-K)= 2.430E-J1
 BOILING PT. (K)= 4.280E+02
 HEAT OF VAPORIZATION(CAL/G)= 1.450E+1
 DENSITY (G/CM3)= 1.723E+CC
 MASS FLUX(G/SEC)= 1.520E+02
 JET VELOCITY(CM/SEC)= 2.430E+J3
 JET ANGLE(RAD.)= 6.807E-J1
 TEMPERATURE(K)= 4.000E+J2
 SURFACE TENSION(DYNE/CM)= 2.840E+02
 VISCOSITY(CM/CM-SEC)= 2.200E-J2
 ORIFICE DIAMETER(CM)= 5.100E-02
 LIQUID DROP COEFFICIENTS
 AK= 6.120E+02
 BK= 2.440E-02

NPS= 7 NSL= 7 NS= 3 IBUG= - NSO= 3
 INITIAL S= 0.000 S INCREMENT= 1.000E-02 FINAL S= 1.000E+02 PRINT INTERVAL= 2.000E+00
 .300E+00 .200E+00 .110E+00 .333E+00 .920E+J1

C1= 1.100E-02 C2= 1.000E-03 C3= 3.000E-01 C4= 3.000E-10 C5= 3.000E-10 C6= 3.000E-10 C7= 1.000E-10 C8= 2.330E-10 FF3= 1.000E+00 FF4= 1.000E+00
 N1= 1.500E+00 N2= 1.500E+00 N3= 2.000E+00 N4= 1.500E+00 N5= 2.000E+00 N6= 1.500E+00 N7= 9.000E+04 N8= 3.000E+04 N9= 3.000E+04

J	Y	PE	TE	JE	FE(I) (MOLE/CM)	REU	REU2	REU3	RUF1	RUF2	RUF3	
2	.11	7.931E+03	1.310E+03	1.000E+05	3.920E-08	3.920E-08	1.450E-02	1.450E-02	3.000E-02	4.800E-09	1.230E-02	4.400E-07
3	.69	8.357E+03	3.410E+03	3.950E+05	2.920E-06	2.920E-06	1.450E-02	1.450E-02	2.400E-02	2.700E-06	1.530E-02	9.300E-06
4	.4	8.357E+03	3.410E+03	4.000E+05	5.260E-06	5.260E-06	1.450E-02	1.450E-02	2.000E-02	2.300E-07	1.720E-02	4.700E-05
5	1.7	6.357E+03	4.500E+03	5.000E+05	7.920E-06	7.920E-06	1.450E-02	1.450E-02	1.550E-02	2.200E-06	1.950E-02	1.900E-04
5	3.12	7.301E+03	5.300E+03	5.800E+05	1.080E-05	1.080E-05	1.450E-02	1.450E-02	5.400E-03	9.900E-01	2.300E-02	1.900E-03
7	4.86	1.246E+04	5.825E+03	5.380E+05	7.950E-07	7.950E-07	5.400E-03	5.400E-03	2.300E-04	3.600E-03	2.560E-02	1.300E-03
AV=	3.00	9.661E+03	5.110E+03	5.040E+05	5.370E-06	5.370E-06	1.100E-02	1.100E-02	8.257E-03	1.317E-03	2.250E-02	1.127E-03

J	Y	MA	ME	HSE	REU	REU2	REU3	RUF1	RUF2	RUF3		
1		RUF4	RUF5	RUF6	RUF7	RUF8	RUF9					
1	.15	2.257E+01	3.080E+03	6.120E+03	2.643E-01	1.330E+05	1.020E+03	1.420E-05	1.420E-06	2.910E-03		
2	.15	6.453E+03	1.700E+03	1.000E+05	4.526E-08	4.526E-08	1.450E-02	1.450E-02	3.000E-02	4.800E-09	1.230E-02	4.400E-07
3	.03	5.500E+03	3.300E+03	4.000E+05	1.980E-06	1.980E-06	1.450E-02	1.450E-02	2.400E-02	2.700E-06	1.530E-02	9.300E-06
4	1.21	6.605E+03	3.720E+03	4.610E+05	3.650E-06	3.650E-06	1.450E-02	1.450E-02	2.000E-02	2.300E-07	1.720E-02	4.700E-05
5	1.6	5.500E+03	4.210E+03	5.110E+05	5.210E-06	5.210E-06	1.450E-02	1.450E-02	1.550E-02	2.200E-06	1.950E-02	1.900E-04
6	3.09	6.250E+03	5.125E+03	5.900E+05	7.650E-06	7.650E-06	1.320E-02	1.320E-02	5.400E-03	9.900E-05	2.300E-02	1.900E-03
7	5.42	7.952E+03	5.225E+03	6.500E+05	4.700E-07	4.700E-07	6.000E-03	6.000E-03	2.300E-04	3.500E-03	2.560E-02	1.300E-03
AV=	5.48	7.519E+03	4.530E+03	5.111E+05	3.530E-06	3.530E-06	1.100E-02	1.100E-02	8.342E-03	1.317E-03	2.250E-02	1.127E-03

I	WA	HE	HSE	REJE	REUEZ	REUEHS	RUF1E	RUF2E	RUF3E
2	2.255E+01	2.297E+03	6.091E+03	2.102E+01	2.311E-04	1.280E+03	7.419E-07	7.419E-07	2.324E-03
2	5.800E+03	1.775E+03	1.000E+03	4.971E-08	4.971E-06	1.450E-02	3.000E-02	4.830E-08	1.239E-02
3	5.930E+03	3.265E+03	4.030E+03	1.915E-06	1.915E-06	1.450E-02	2.400E-02	2.700E-08	1.530E-02
4	5.930E+03	3.330E+03	4.660E+03	3.378E-06	3.378E-06	1.450E-02	2.030E-02	2.900E-07	1.721E-02
5	5.900E+03	4.175E+03	5.160E+03	4.757E-06	4.757E-06	1.450E-02	1.950E-02	2.200E-06	1.350E-02
6	5.563E+03	4.000E+03	7.930E+03	5.800E-06	5.800E-06	1.330E-02	5.400E-03	9.900E-05	2.380E-02
7	6.736E+03	4.825E+03	6.540E+03	4.479E-07	4.479E-07	6.650E-03	2.300E-04	3.600E-03	2.550E-02
AV=	6.14E+03	4.444E+03	5.120E+03	3.47E-06	3.47E-06	1.101E-02	6.452E-03	1.421E-03	2.27E-02

I	WA	HE	HSE	REJE	REUEZ	REUEHS	RUF1E	RUF2E	RUF3E
3	2.255E+01	2.913E+03	6.000E+03	1.918E-01	9.825E-04	1.161E+03	6.845E-07	6.845E-07	2.113E-03
4	4.862E+03	1.300E+03	1.000E+03	5.714E-08	5.714E-08	1.450E-02	3.000E-02	4.800E-08	1.239E-02
5	4.862E+03	3.200E+03	4.070E+03	1.729E-06	1.729E-06	1.450E-02	2.400E-02	2.700E-08	1.530E-02
6	4.862E+03	3.339E+03	4.700E+03	2.876E-06	2.876E-06	1.450E-02	2.030E-02	2.900E-07	1.721E-02
7	4.802E+03	3.175E+03	5.200E+03	4.009E-06	4.009E-06	1.450E-02	1.950E-02	2.200E-06	1.900E-04
AV=	4.842E+03	4.710E+03	6.000E+03	5.814E-06	5.814E-06	1.350E-02	5.900E-03	9.900E-05	2.380E-02
2	4.369E+03	4.000E+03	8.000E+03	3.982E-07	3.982E-07	5.730E-03	2.800E-04	3.600E-03	2.550E-02
AV=	4.663E+03	4.190E+03	5.139E+03	2.430E-06	2.435E-06	1.160E-02	8.732E-03	1.474E-03	2.227E-02

I	WA	HE	HSE	REJE	REUEZ	REUEHS	RUF1E	RUF2E	RUF3E
4	2.247E+01	2.872E+03	6.000E+03	1.665E-01	8.548E-04	1.001E+03	4.063E-07	4.063E-07	1.635E-03
2	4.210E+03	2.322E+03	3.900E+03	5.320E-08	5.320E-08	1.450E-02	3.000E-02	4.800E-08	1.239E-02
3	4.133E+03	3.160E+03	4.080E+03	1.572E-06	1.572E-06	1.450E-02	2.400E-02	2.700E-08	1.530E-02
4	4.133E+03	3.180E+03	4.700E+03	2.515E-06	2.515E-06	1.450E-02	2.030E-02	2.900E-07	1.721E-02
5	4.133E+03	3.740E+03	5.250E+03	3.463E-06	3.463E-06	1.450E-02	1.950E-02	2.200E-06	1.900E-04
6	4.214E+03	4.565E+03	6.000E+03	4.957E-06	4.957E-06	1.370E-02	5.400E-03	9.900E-05	2.380E-02
7	4.538E+03	4.400E+03	6.650E+03	3.637E-07	3.637E-07	6.800E-03	2.300E-04	3.600E-03	2.550E-02
AV=	4.322E+03	4.145E+03	5.107E+03	2.636E-06	2.636E-06	1.144E-02	8.944E-03	1.491E-03	2.216E-02

I	WA	HE	HSE	REJE	REUEZ	REUEHS	RUF1E	RUF2E	RUF3E
1									

J	Y	PE	RUF5E	RUF6E	RUF7E	RUF8E	RUF9E
5	2.24E+1	2.84E+3	5.96E+3	9.74E+4	1.47E-1	7.59E+4	2.99E-7
	1.31E-3	2.19E-4	3.25E-3	5.66E-8	3.25E-3	1.47E-4	1.12E-3
J	Y	PE	TE	JE	FE(I) (MOLE/GM)		
2	.24	3.74E+3	2.13E+3	9.74E+4	5.66E-8	5.66E-8	1.45E-2
3	1.36	3.19E+3	3.13E+3	4.03E+3	1.46E-6	1.46E-6	2.70E-2
4	1.00	3.59E+3	3.54E+3	4.05E+3	2.22E-6	2.22E-6	2.30E-2
5	2.03	3.19E+3	3.14E+3	5.22E+3	3.09E-6	3.09E-6	1.55E-2
6	3.00	3.74E+3	4.46E+3	6.00E+3	4.24E-6	4.24E-6	5.40E-3
7	5.45	3.95E+3	4.25E+3	6.68E+3	3.68E-7	3.68E-7	2.30E-3
AV	41.92	3.76E+3	3.91E+3	5.46E+3	1.76E-6	1.76E-6	3.10E-3

I	MA	HE	HSE	REU2	REU3E	RUF1E	RUF2E	RUF3E
6	2.23E+1	2.83E+3	5.89E+3	1.33E-01	6.69E+4	2.32E-7	2.32E-7	1.457E-3
	1.24E-3	1.95E-4	2.89E-3	1.26E-4	1.26E-4			
J	Y	PE	TE	JE	FE(I) (MOLE/GM)			
2	.26	3.39E+3	2.25E+3	9.35E+4	5.69E-8	5.69E-8	1.45E-2	4.40E-7
3	1.44	3.26E+3	3.11E+3	3.97E+3	1.20E-6	1.20E-6	2.70E-2	1.53E-6
4	1.70	3.26E+3	3.39E+3	4.00E+3	1.77E-6	1.77E-6	2.30E-2	4.70E-5
5	2.44	3.26E+3	3.36E+3	5.18E+3	2.73E-6	2.73E-6	1.55E-2	1.30E-4
6	3.91	3.44E+3	4.35E+3	5.11E+3	3.70E-6	3.70E-6	1.40E-3	2.38E-2
7	6.02	3.45E+3	4.15E+3	6.71E+3	3.76E-7	3.76E-7	2.30E-3	2.55E-2
AV	52.44	3.41E+3	3.46E+3	5.26E+3	1.52E-6	1.52E-6	3.38E-3	2.19E-2

I	MA	HE	HSE	REU2	REU3E	RUF1E	RUF2E	RUF3E
7	2.22E+1	2.83E+3	5.89E+3	1.33E-01	6.69E+4	1.82E-7	1.82E-7	1.329E-3
	1.11E-3	1.77E-4	2.62E-3	1.15E-4	1.15E-4			
J	Y	PE	TE	JE	FE(I) (MOLE/GM)			
2	.31	2.91E+3	2.35E+3	8.70E+4	5.46E-8	5.46E-8	1.45E-2	4.40E-7
3	1.24	2.80E+3	3.11E+3	3.78E+3	1.21E-6	1.21E-6	2.70E-2	1.53E-6
4	2.02	2.80E+3	3.15E+3	4.26E+3	1.76E-6	1.76E-6	2.30E-2	4.70E-5
5	2.0	2.80E+3	3.00E+3	4.98E+3	2.27E-6	2.27E-6	1.55E-2	1.30E-4
6	4.19	2.98E+3	4.27E+3	6.15E+3	3.10E-6	3.10E-6	5.40E-3	2.38E-2
7	5.07	3.09E+3	4.26E+3	6.74E+3	3.92E-7	3.92E-7	2.30E-3	2.55E-2
AV	72.29	2.92E+3	3.76E+3	4.91E+3	1.39E-6	1.39E-6	3.62E-3	2.16E-2

I	MA	HE	HSE	REU2	REU3E	RUF1E	RUF2E	RUF3E
6	2.21E+1	2.84E+3	5.76E+3	1.03E-01	5.65E+4	1.43E-7	1.43E-7	1.161E-3
	1.09E-3	1.45E-4	2.24E-3	9.85E-5	9.85E-5			

J	Y	PE	MA	HE	TE	JE	MOLE/GM)	SM)
2	1.33	2.735E+03	2.435E+03	5.043E+04	5.063E+04	5.063E+04	1.45E-10	1.45E-10
3	1.36	2.624E+03	3.125E+03	3.59E+03	1.179E+06	1.090E+02	1.090E+02	1.090E+02
4	2.10	2.524E+03	3.370E+03	4.190E+03	1.541E+06	1.490E+02	1.490E+02	1.490E+02
5	2.68	2.624E+03	3.355E+03	4.674E+03	1.674E+06	1.450E+02	1.450E+02	1.450E+02
6	3.3	2.436E+03	4.285E+03	5.174E+03	2.586E+06	1.430E+02	1.430E+02	1.430E+02
7	6.09	2.637E+03	3.975E+03	6.751E+05	3.661E+07	6.80E+08	2.66E+02	2.66E+02
AV=	67.48	2.768E+03	3.125E+03	4.674E+03	1.211E+06	1.136E+02	1.136E+02	1.136E+02

I	MA	HE	HSE	REUE	REUE2	REUE3	RUF1E	RUF2E	RUF3E
4	2.212E+01	2.01E+03	5.58E+03	4.639E+04	5.419E+02	1.160E-07	1.160E-07	1.084E+03	
4	2.212E+01	4.357E+04	1.322E+04	2.062E+03	9.142E-05				
J	Y	PE	TE	JE	MOLE/GM) <td>SM) <td></td> <td></td> <td></td> </td>	SM) <td></td> <td></td> <td></td>			
2	1.4	2.834E+03	2.265E+03	8.100E+04	4.187E+08	1.470E+02	3.000E-02	1.235E+02	4.400E+07
3	1.12	2.593E+03	3.130E+03	3.395E+03	9.697E-07	1.450E+02	2.400E-02	2.740E+08	1.535E+02
4	2.15	2.543E+03	3.33E+03	4.5E+03	1.374E+06	1.45E+02	2.03E+02	2.700E+07	1.72E+02
5	2.75	2.583E+03	3.390E+03	4.821E+03	1.591E+06	1.450E+02	1.650E+02	2.940E+06	1.95E+02
6	4.48	2.765E+03	4.200E+03	6.191E+03	2.176E+06	1.42E+02	5.44E+03	9.9E+05	2.38E+02
7	6.6	2.625E+03	3.330E+03	6.74E+03	3.889E-07	6.80E+03	2.300E+04	3.630E+03	2.55E+02
AV=	100.00	2.697E+03	3.715E+03	4.783E+05	1.079E+06	1.140E+02	9.313E+03	1.332E+03	2.165E+02

I	MA	HE	HSE	REUE	REUE2	REUE3	RUF1E	RUF2E	RUF3E
10	2.225E+01	2.073E+03	5.505E+03	3.233E+02	4.419E+04	5.179E+02	9.966E-08	9.966E-08	1.061E-03
10	2.225E+01	5.158E+04	1.231E+04	2.041E+03	8.804E-05				

CONDITIONS IN FRONT OF SHOCK
 PX= 3.651E+03 UX= 5.041E+05 HSX= 6.128E+03 RHOX= 5.237E-07 TX= 5.010E+03 MAX= 2.257E+01
 MASS FRACTIONS FILE= ..612E+04 2.348E+03 1.762E+01 ..

CONDITIONS BEHIND SHOCK
 PY= 1.182E+05 JY= 9.402E+04 MY= 6.122E+03 RHOY= 2.811E+06 TY= 1.141E+04 TOPY= 3.555E+02

JET BREAK-UP INFORMATION
 ME= 3.865E+04 REJ= 8.417E+00 TBK= 2.692E+03 UBK= 2.532E+03 XBK= 5.442E+03 YBK= 4.650E+00
 MEL= .. OSL= 0. D32= 5.332E+03 DMAX= 0.

----- QUALIFIED FREE STREAM CONDITIONS IN SHOCK FRONT BY SINGLE ITERATION -----

J	Y	PE	MA	HE	TE	JE	MOLE/GM)	SM)
AV=	100.0	3.661E+03	5.112E+03	5.543E+05	5.393E+06	5.395E-06	1.1.6E-05	2.267E-03

I	MA	HE	HSE	REUE	REUE2	REUE3	RUF1E	RUF2E	RUF3E
1	2.225E+01	3.093E+03	6.441E+03	2.633E+01	1.329E+03	1.017E+03	4.44E+06	4.44E+06	4.44E+03
1	2.225E+01	2.183E+03	3.482E+04	1.966E+03	2.979E-05				

CONDITIONS IN FRONT OF SHOCK
 PX= 3.561E+03 JX= 5.043E+05 MSX= 6.141E+03 RHOX= 5.215E-07 TX= 5.610E+02 MAX= 2.243E+1
 MASS FRACTIONS F(1)= 1.619E-04 2.961E-9 1.769E-01 0. 1.111E-01 1.111E-01 3.333E-02 0.

CONDITIONS BEHIND SHOCK
 PY= 1.178E+05 JY= 3.411E+04 MY= 6.055E+3 RHOY= 2.797E-6 TY= 1.144E+4 TRPY= 3.554E+2

JET BREAK-UP INFORMATION
 ME= 3.053E+00 REJ= 9.417E+00 TBK= 2.633E-03 UBK= 2.529E+03 XBK= 5.433E+01 YBK= 4.656E+04
 MEL= . . D32= 5.332E-03 DMAX= . .

INITIAL FLOW RATE (G/SEC)
 XMV= 2.920E-01 XML= 1.517E+02 XMA= 4.441E+00 XMM= 4.733E+06 AE= 1.687E+01

----- CONDITIONS OF THE INITIAL MIXTURE AT S= 331
 U= 1.504E+05 MSV= 1.009E+02 HSR= 5.759E+03 H= 5.469E+03 TSP= 1.119E+04 I= 1.075E+04 RHO= 2.625E-09 A= 1.134E+01
 MASS FRACTIONS OF SPECIES
 MU+= 1.519E-04 E= 2.778E-03 J= 1.666E-01 F= . . N= 1.069E-1 O2= 3.973E-02 N2= 5.956E-01 MO= 3.185E-02 FV= 6.171E-02
 P= 9.615E+04 MM= 2.394E+01 RMD= 2.625E-06 MS= 5.756E+03 FV= 6.171E-02 UP= 2.529E+03 V2= 1.665E+03
 TP= 3.554E+02 KP= 2.696E-03 SE= 2.210E+00 YE= 4.564E+00

INITIAL THERMODYNAMIC INFORMATION
 GAMA= 1.150 CPICAL(0)= 6.365E-01 SS= 2.074E+05 MA= 7.635E-01 MAP= 7.514E-01 REP= 8.741E-01

S(CM)= 1.4334
T(K)= 1.739E+15 P(DYN/CM2)= 9.815E+04 RHO(GM/ML)= 2.033E+06 U(CM/SEC)= 1.591E+15 HS(CAL/GM)= 5.756E+03 FV= 5.369E-02 MM= 2.345E+01
A= 1.146E+01 UP= 2.529E+03 TP= 3.554E+02 RP= 2.566E-03 RHOP= 5.269E-03 AE= 1.687E+01 ISR= 1.119E+04 ME= 5.438E+1.3

MASS FRACTIONS OF SPECIES
M0= 1.245E-04 E= 2.773E-04 Q= 1.656E-01 F= 7.301E-09 N= 1.887E-01 O2= 3.933E-02 N2= 5.944E-01 MO= 3.173E-02
---- ELECTRON DENSITY= 6.024E+12
.5 6E-03 .5 6E-03 .144E-11 .1 .778E-02 .124E-02 .243E-01 .100E-02

CHANGES OF MASS FRACTION OVER DS= 2.822E-05
-2.726E-07 -5.192E-12 -3.507E-04 7.001E-09 -2.301E-04 -8.306E-05 -1.258E-03 -6.730E-05 1.982E-03
DP= -1.63E+00 DRH0= .348E-08 DU= -.331E+03 DHS= -120E+02 DFV= .193E-01 PHADS= .11E-1
DA= 1.29E-02 DAE= 1.931E-04 DDP= .808E-06 DP= 1.312E+02 PHIV= 3.113E+01
DFP= 2.333E+11 DQ= 1.618E-09

PRODUCTION RATES FOR EACH REACTION (MOL/SEC-CM3)
1. 49E-06 .. 1.769E-06 .0. 5.442E-06 0. .0. .230E-01

PRODUCTION RATES FOR EACH SPECIE
2.396E-05 1.852E-05 -2.396E-05 5.442E-06 -2.396E-05

LOCAL THERMODYNAMIC INFORMATION
GAMA= 1.153 CP(CAL/G)= 6.345E-01 SS= 2.271E+05 MA= 7.647E-01 MAP= 7.514E-01 REP= 6.701E-01 P/SEC= 1.119E+09
.2 2E-04 .35E+0 .221E+1 .485E+1 .187E+4 .192E+13 .333E+0 .177E+22 .17E+1 .33C-218

S(CM)= 5.4456
T(K)= 557.12 PDYNE/CM2)= 9.812E+04 RHO(GM/ML)= 4.602E-06 U(CM/SEC)= 1.118E+15 HS(CAL/GM)= 4.84E+3 FV= 3.444E-1 MM= 3.345E+01
A= 1.316E+01 UP= 2.033E+03 TP= 3.554E+02 RP= 2.566E-03 RHOP= 4.461E-03 AE= 1.691E+01 ISR= 3.931E+03 ME= 3.931E+03

MASS FRACTIONS OF SPECIES
M0= 1.114E-04 E= 1.741E-04 Q= 1.167E-01 F= 1.252E-05 N= 7.657E-02 O2= 2.792E-02 N2= 4.187E-01 MO= 2.239E-02
---- ELECTRON DENSITY= 6.025E+12
.372E-03 .319E-03 .733E-02 .534E-06 .548E-02 .874E-03 .150E-01 .748E-03

CHANGES OF MASS FRACTION OVER DS= 3.372E-05
-1.571E-07 -6.165E-12 -1.732E-04 1.146E-07 -1.136E-04 -4.113E-03 -6.213E-04 -3.324E-05 9.790E-04
DP= -.21E+00 DRH0= .118E-07 DU= -.163E+03 DHS= -.593E+01 DFV= .973E-03 PHADS= .100E-01
DA= 5.113E-03 DAE= 2.175E-04 DDP= 1.845E-01 DDP= 5.909E-06 DP= 9.530E+01 PHIV= 1.923E+01
DFP= 2.762E+11 DQ= 2.246E-09

PRODUCTION RATES FOR EACH REACTION (MOL/SEC-CM3)
2.762E-06 3.163E-05 1.597E-05 1.941E-06 9.025E-05 5.592E-07 9.526E-07 2.165E-05

PRODUCTION RATES FOR EACH SPECIE
3.748E-06 -8.54E-05 -1.601E-06 6.873E-05 -3.798E-05

LOCAL THERMODYNAMIC INFORMATION
GAMA= 1.109 CP(CAL/G)= 4.102E-01 SS= 1.579E+05 MA= 7.031E-01 MAP= 5.931E-01 REP= 1.204E+01 P/SEC= 1.119E+09
.37E-04 .349E+ .221E+1 .485E+1 .187E+4 .192E+13 .333E+0 .177E+22 .17E+01 .33C-218

S(CM)= 5.4456
T(K)= 5.8134 PDYNE/CM2)= 9.807E+04 RHO(GM/ML)= 7.335E-06 U(CM/SEC)= 8.681E+15 HS(CAL/GM)= 3.175E+3 FV= 4.933E-1 MM= 4.261E+01
A= 1.39E+1 UP= 2.033E+03 TP= 3.554E+02 RP= 2.566E-03 RHOP= 4.418E-03 AE= 1.696E+01 ISR= 7.125E+03 ME= 3.931E+03

MASS FRACTIONS OF SPECIES
M0= 3.317E-05 E= 3.803E-05 Q= 9.016E-02 F= 4.076E-05 N= 5.917E-02 O2= 2.159E-02 N2= 3.235E-01 MO= 1.73E-02
---- ELECTRON DENSITY= 3.069E+12

TRACE

```

*DECK MAIN
PROGRAM J103INPUT,OUTPUT,IAPLS=INPUT,IAPCS=OUTPUT,TAPE=
----- THIS DECK FOR CJC USE -----
----- 12/18/73 -----
C
C
C

```

```

C ONE DIMENSIONAL REACTING DROPLET-VAPOUR-AIR TUBE ANALYSIS
C A SELF-INITIALISED REACTING DVA TUBE (0/8/73)
C
C
C

```

```

C ***** MODIFIED TUBE PRESSURE (0/30/73)
C ***** ITERATED SHOCK FRONT CONDITIONS (9/13/73)
C *****
COMMON SPA(40),CPV(40),MS(40),RHO(40),PSI(40),MSL(40),DE(40),
1 S(40),T(40),PT(40)
2 RUF(40),RUF2(40),RUFSE(40),Y(40),RUFSE(40)
3 SO,OS,SMAX,SMIN,FU,FM,ATQ,VP,RP,CP,ACTE,RSI,HRZ,AV,C,AI,VE
4 NY,SY,AE,AP,TP,CHOP,EP,CP,LAE,RES,IS,OKP,DRM,DP,DSB,CYE,DI,OMS
5 YBU,CUB,DFV,DMO,UP,DA,DS,DMO,UM,SI,SKO,REFA
6 RUA1,UEI,RUFIE1,RUF21,RUF31,PUF,LI,RUF5E1,PHATS
COMMON RUF4E(20),RUF2E(20),RUF3E(20),RUF4E(20),RUF5E(20),
1 RUF2E(20),RUF3E(20),F01, F(9,20),F(8,20),F(7,20),
2 F(9),F(8),F(7),F(6),F(5),F(4),F(3),F(2),F(1),F(0),
3 F1,F2,F3,F4,F5,F6,F7,F8,F9,F10,F11,F12,F13,F14,F15,F16,
4 F17,F18,F19,F20,F21,F22,F23,F24,F25,F26,F27,F28,F29,F30,
5 F31,F32,F33,F34,F35,F36,F37,F38,F39,F40,F41,F42,F43,F44,F45,
6 RUF6E(20),RUF7E(20),RUF8E(20)
7 RUF6E(20),RUF7E(20),RUF8E(20)
8 RUK,FK,FKK,PEKK,RUZBK,SHOKY,SOP,NSO,NSO,NSO,NSO,NSO,NSO,
COMMON STR(30),TT(30),TCT(30),U(20),F(30),F(30),F(30)
COMMON /HEAT/ MLE(9),W(9)
COMMON /ICUD/ CPL,TEP,DMV,PHUL,MV,TPC,SIG,ILU,JML,DO,MECR,
1 FFI,AK,AKME,ITAK,FFV,ITAPY
COMMON /ITEG/ NPS,PT,IO(20),NSL,IOPT,KO,NS,NSI,IPUG,IO
COMMON /SHOCK/ YG(20),PT(20),RUF(20),F(20),F(20),F(20),F(20),
1 F2(20),F3(20),F4(20),F5(20),F6(20),F7(20),F8(20),F9(20),
2 F10(20),F11(20),F12(20),F13(20),F14(20),F15(20),F16(20),

```

```

3 CONTINUE
WRITE(6,19)
WRITE(6,19) (IO(I),I=1,20)
READ(5,100) (ID(I),I=1,20)
IF (E0F5) 3 90+
3 CONTINUE

```

```

PI=3.14159
SOPF=SQR(PI)
AV=6.02E+23
RR=82.05
RR1=1.037
RR2=8.313E7
CTE=4.13E7
AIO=1.013E6
DO 6 I=1,9

```

```

1000 CONTINUE
READ(5,100) (ID(I),I=1,20)
IF (E0F5) 3 90+
3 CONTINUE

```

```

WRITE(6,19)
WRITE(6,19) (IO(I),I=1,20)
READ(5,102) FFAE,FFPE,FEV,FFE,PPXL
WRITE(6,102) FFAE,FFFL,FEV,FFE,PPXL
C ***** CONSTANTS
PI=3.14159
SOPF=SQR(PI)
AV=6.02E+23
RR=82.05
RR1=1.037
RR2=8.313E7
CTE=4.13E7
AIO=1.013E6
DO 6 I=1,9

```

```

10 ***** MODIFIED TUBE PRESSURE (0/30/73)
15 ***** ITERATED SHOCK FRONT CONDITIONS (9/13/73)
20 *****
25 COMMON SPA(40),CPV(40),MS(40),RHO(40),PSI(40),MSL(40),DE(40),
30 S(40),T(40),PT(40)
35 RUF(40),RUF2(40),RUFSE(40),Y(40),RUFSE(40)

```

```

60 DF(I)=0.0
   IOK=0
   XK=0.0
   6 CONTINUE
   OFV=3.0
   OFQ=0.0
   OU=0.0
   OP=0.0
   ORHO=0.0
   OT=0.0
   OMS=0.0
   OIS=0.0
   ORHOP=0.0
   DUP=0.0
   ORP=0.0
   OEP=0.0
   OA=0.0
   OSE=0.0
   OVL=0.0
   OAZ=0.0
   O0 27 I=1,8
   MOOT(I)=J.0
   27 CONTINUE
   EIA=1.0
   80 C*** RR IN 243-ATM/G-40LL-K
   CALL PUTIN
   YPEN=YE
   SEQ=SE
   EP= CPL*TF
   85 RPS= EPAJF*UP/1.0*3*CFE1
   DS1=DS
   DSMAX=5.1*CS
   OSMIN=05/10.0
   WRITE(6,39)
   90 WRITE(6,100) (IDD(I),I=1,20)
   IPS=0SE/JS
   IS=-1
   1 CONTINUE
   US=DS1
   AXI=(IJ-UP)**2+JP**2)
   UR = SORT(AXI)
   95 RSR=RS-(JL-UR)*UR 142.9*CFE1
   TI=TS
   CALL IIN/HSR,IS +II,F,MS1,CPSR)
   RHOP=0.0
   XNP=0.0
   100 IF (RPS-1.1*JL-9) GO TO 40
   RHOP= IXALX*HC*FINS111J+ALJUF*AL)
   XNP= 3.0*Q-OP*UP*4/1.0*PI*3*P*P*PHCL)
   105 XXI=RHDL*CHV*0.5
   XX2=ALD*UR*(TS-TP)
   V= SORT(X,C*P/RHO)
   C***** V IS MEAN THERMAL VELOCITY OF MIXTURE BY KINETIC THEORY
   XX31=0.5*AL*V*II-IP/XX2
   110 OP=0.5*X(2)*P*XX3/CTE *EFEV

```

```

0001=-09/XXI
PHIV=1.58RHO*0001/PP
IF (PHIV.GT. 1.0E-6) US=DS1/(PHIV*A)
IF ((IS.LT.16).AND.(DS.LT.05MIN)) JS=DSMIN
IF (DS.LT.05MAX) JS=05MAX
40 CONTINUE
SI=SI+DS
IF((SI.GT. S(NFS)).OR.(SI.LT.S(1))) GO TO 999
DO 21 IJ=2,NPS
IF (SI.GT.S(IJ-1) .AND. SI.LT.S(IJ)) GC TO 22
21 CONTINUE
WRITE(6,260)I,S,SI,SI
STOP
200 FORMAT (16,' OUT OF RANGE IO=',I9,' S=',F9.3/)
22 CONTINUE
CALL LIP,M(SI,RUC,S,RUCI,I)
CALL LIP,M(SI,RUF,S,RUFI,I)
CALL LIP,M(SI,RUFE,S,RUFEI,I)
CALL LIP,M(SI,RUF2,S,RUF2I,I)
CALL LIP,M(SI,RUF3,S,RUF3I,I)
CALL LIP,M(SI,RUF4,S,RUF4I,I)
CALL LIP,M(SI,RUF5,S,RUF5I,I)
CALL LIP,M(SI,RUF6,S,RUF6I,I)
CALL LIP,M(SI,RUF7,S,RUF7I,I)
CALL LIP,M(SI,RUF8,S,RUF8I,I)
CALL LIP,M(SI,PTJ,S,PTJ,I)
XXXX=(SI-ARK)/PRX
IF(XXXX.GT.0) XXXX=J.0
PI=-PI-RUC2E1+I.9-RMGM)*-XK(-XXXX)
PI*PI*PI*PI
IF (IQUA.NE. 0) WRITE(6,10) RUCI,RUFEI,RUFEI,RUFEI,RUFEI,
RUC2E1,RUC2E1,RUC2E1,RUC2E1,RUC2E1,RUC2E1,PI
DS=PI-9
ADS=ADS
SP = S
SE = SOP + S10*(SI/150.**2/PI)**AI
IF(SI.GE.47) *GEQPI) S=SOP+S10*(70./150.**AI)*SI/150.**GEQPI
YPEN=YCN*Y/00*DS
YE = YPEN*ISE-SPI
SE = FFA*YCE
CALL 0.5*PI*SE*YE*4E
IF(IJUS.15.0) WRITE(6,10) AC,SE,Y,YPEN,C,DI,SI,DS
SOGAN= SRTIGAMA1
XMAP=XMS-LP/U
C***** VISCOSITY FOR AIR IN (G/CM-SEC) BY SUTHERLAND LAW
UM1.31E-5*(T+.5)/(T+110)
REPRAGE=UM*PR*U
PHAD=0.4
DFPE=0
DRE=0
OD=0.0
IF(DP.LT. 1.0E-9) GO TO 7
C CALCULATE DRAG COEFFICIENT
COD=2./REF

```



```

00 29 I=INSI
MM=MM+F(I)/MI(I)
29 CONTINUE
MM=1.0/M
C CALCULATE LOCAL SPECIFIC HEAT
GAMA=CP/(Cp-R/R/M)
C LOCAL SPEED OF SOUND
SS= SQRT(GAMA*R2/T/M)
C LOCAL MACH NUMBER
XMA=U/SS
DRHO=PI*RM/R2/T-RHO
DA=(PHAD3+RUEI*DAE-AMU*3+HO-4*RH0*DU)/R2+U
AE=AE+DA
RP=RP+DR
IF (RP.LT.0.0) RP=0.0
UP=UP+DJ
U=U+DU
HPS=HPS+DMPS
RHO=RHO+DRHO
A=A+DA
P=P1
IS=IS+1
IF (4*IS+1)ST*6F,SMAX,1-50,14,9
IF ((1/IS)*IPS)*IPS,NE, IS) GO TO 2
9 CONTINUE
C CALCULATING PRODUCTION RATES
EONE=R4+J+F2
W00T(1)= AMH2*RKI(1)*F1*F2
W00T(2)= AMH2*RKI(2)*F1*F2
W00T(3)= RHO*RKI(3)*F1
W00T(4)= RHO*RKI(4)*F1
W00T(5)= AMH2*RKI(5)*F01*F2
W00T(6)= AMH2*RKI(6)*F2/M4
W00T(7)= AMH2*RKI(7)*F3*F4
W00T(8)= AMH2*RKI(8)*F3*F5
W00T(9)= -W00T(1)+W00T(2)+W00T(3)+W00T(4)+W00T(5)
W00T(10)= -W00T(1)+W00T(3)+W00T(5)-W00T(6)-W00T(7)+W00T(8)
W00T(11)= W00T(1)+W00T(2)+W00T(3)+W00T(4)+W00T(5)+W00T(6)+W00T(7)
W00T(12)= W00T(1)+W00T(2)+W00T(3)+W00T(4)+W00T(5)+W00T(6)+W00T(7)
W00T(13)= W00T(1)+W00T(2)+W00T(3)+W00T(4)+W00T(5)+W00T(6)+W00T(7)
WRITE(6,104) SI, P, RHO, G, MS, F, INS, Y, M
WRITE (5,110) A,UP,IP,02,RMCP,AL,IS,M
WRITE(6,105) (F(I),I=1,5),EONE
WRITE(6,100) SI, Y, P, U, F(1), F(2), EONE
400 FORMAT(2F11.4,IP,5E10.3)
WRITE(6,103) F1,F2,F3,F4,F5,F6,F7,F8
WRITE(6,104) DS,10F411.1,14S11
WRITE(6,104) UP,DRHO,DU,DMS,DF,INSI),PHADS
WRITE(6,111) G4,032,GUP,02F,0P,CHIV,3FF,00
WRITE(6,105) (W00T(I),I=1,3)
WRITE(6,107) (W00T(I),I=1,5)
WRITE(6,118) GAMA,CP,SS,XMA,AMH2*F,RP,XMP
WRITE(6,101) 05,2E14,55,40,2,0,0,1,5,0,0,0,14
1,3E14

```



```

*DECK PUTIN
SURROUT14E PUTIN
DIMENSION D(16),Y(12)
COMMON C(4)0,CY(4)0,WA(4)0,MOE(4)0,PSIV(4)0,MSE(4)0,UE(4)0,
5    1 S(4)0,T(4)0,PT(4)0
2 RUF(4)0,RUFSE(4)0,ROMSE(4)0,Y(4)0,RUFSE(2)0
3 RO,OS,SMX,RHO,F,U,T,MS,ATO,PP,ALSP,CIL,ARRI,PR2,4,C,AI,VE
4 MM,SE,AE,PO,TP,PHOP,EP,DEP,C42,GT,ST,0DP,DMOP,DS,0VE,GE,DS
5 DU,DP,DFV,DZHO,OP,OA,OS,ARMZ,UM,SI,XO,BETA
6 RUAI,RUJ,RUFEI,RUFSEI,RUFSEI,ROFSEI,RUFSEI,PHADS
COMMON RUF(4)0,D3(2)0,DDT(3)0,DOCT(1)0,DOCTN(5)0,UL(2)0,PUF1E(2)0,
1 RUFSEI(2)0,RUFSEI(2)0,F9I,FE(3)0,PE(2)0,TE(2)0
2 F(9)0,F(9)0,DZII,DZII,XMLI,XMDU,ETI,ALO,ALT,DE,VL,I,IMETA
3 F1,F2,F3,F4,F5,PI,411,36,KMFB,CF3,UF,VP,GAMA,CL,C2,C3,C4,C5,C6,
4 C7,C8,C9,C10,C11,XMS,XMAP,SS,UF,OUR,MS5,MSR,XML3
5 RUF6E(2)0,RUF7E(2)0,RUF8E(2)0
6 RUF9E1,RUF7E1,RUF8E1,F9,F9,F9,F9,F9,F9,F9,F9,F9
7 YFE,FE3M,YCEI,GE2,GE2,GE2,GE2,CF7,XNEI,XNF2,XNF5,95,87,FF3,FF4,XMOT
8 X6K,Y6K,TK,K,P4K,RUZBK,PH0XY,SO,NSO,XMWR,DCP,S12
COMMON SI(3)0,TI(3)0,RHOI(3)0,MI(3)0,FI(18)0,3)
COMMON ZHEAT,XL(9)0,MI(9)
COMMON ZIGUO,ZP,ZP,ZP,ZP,ZP,ZP,ZP,ZP,ZP,ZP,ZP,ZP,ZP
1 FEVI,K,K,K,K,K,K,ITEK,FEV,199Y
COMMON ZIIEGA,MS,NPI,10,4,20,MSE,IOP1,40,MS,MSI,IEU,10
COMMON ZHECK7,YO(2)0,PT(3)0,CE,UF9(2)0,TI(3)0,FI(2)0,
2 F1(8)0(2)0,F3(8)0(2)0,F+TE(2)0,F5(3)0(2)0,F7(2)0(2)0,
  F1(2)0(2)0,F3(2)0(2)0
  READ(5,10) W,CPL,TBP,DM,RHOL,XML,XLI,VEI,IMETA
  1 TLO,SI,UML,DC
  ZHEAT=ZHEAT*100
  V2=VL*ZCS(IMETA)
  WRITE(6,11) W,CPL,TBP,DM,RHOL,XML,XMLI,VEI,IMETA
  1 TLO,SI,UML,DC
  READ(5,11) NPS,NSL,NOS,IRUF,NSO,NMWR
  WRITE(6,12) NPS,NSL,NOS,IRUF,NSO,NMWR
  A1=AI-3.0
  XNOS=NOS
  DE=SGRT(XNOS)*.00
  XNSO=XSO
  ZCEP=SGRT(XNSO)*.00
  READ(5,13) C1,CF2,CF3,CF4,CF5,CF6,CF7,CF8,CF9,FF3,FF4
  WRITE(6,14) C1,CF2,CF3,CF4,CF5,CF6,CF7,CF8,FF3,FF4
  READ(5,15) XNF1,XNF2,XNF3,XNF4,XNF5,XNF6,XNF7,XNF8
  WRITE(6,16) XNF1,XNF2,XNF3,XNF4,XNF5,XNF6,XNF7,XNF8
  C ----- EIT INPUT AS MOLE/GM OF MIXTURE (-----
  C ASSUME ALL POSITIVE IONS ARE 1.0* KHRT=1.0
  XNHI=1.0
  C CALCULATE MOLECULAR WEIGHT OF THE INITIAL MIXTURE

```

2


```

12 CONTINUE
00 14 II=1,NS
FT3(II,I)=FT3(II,I)/SUM
14 CONTINUE
C***** NORMALIZED IN MOLE/GM OF MIXTURE
115 W4(I)=W4(I)
00 30 II=1,NS
W4(I)=W4(I)+FT3(II,I)
30 CONTINUE
W4(I)=1.0/W4(I)
RHO(I)=PT3(I)+W4(I)*RHO2+FT3(I)
00 31 JJ=1,6
00(JJ)=0.0
00 32 II=1,NS
00(JJ)=00(JJ)+XL(JJ,II)*FT3(II,I)
32 CONTINUE
31 CONTINUE
110=110+110-1036+9
HE=1.000*(-00(5)/10+00(6)+10*(00(1)+11)*00(2)/2.0+110*(00(3)/3.
1+110*00(4)/4.0))
HSE(I)=HE+0.5*UT3(I)*UT3(I)/CTE
XXX1=RHO(I)*UT3(I)
RUE(I)=XXX1
RUE(I)=XXX1+UT3(I)
0UMSE(I)=XXX1*MSE(I)
RUF1(I)=XXX1*FT3(1,I)
RUF2(I)=XXX1*FT3(2,I)
RUF3(I)=XXX1*FT3(3,I)
RUF4(I)=XXX1*FT3(4,I)
RUF5(I)=XXX1*FT3(5,I)
RUF6(I)=XXX1*FT3(6,I)
RUF7(I)=XXX1*FT3(7,I)
RUF8(I)=XXX1*FT3(8,I)
WRITE(6,126) SII,PT3(I),T13(I), (FT3(II,I),II=1,NS)
WRITE(6,102)
WRITE(6,103) I,W4(I),MSE(I),RUE(I),RUE(I)
1 RUMSE(I),RUF1(I),RUF2(I),RUF3(I),RUF4(I),RUF5(I)
2,RUF6(I),RUF7(I),RUF8(I)
5 CONTINUE
CALL BREAK
GAMA=CP7(CP-PR1/MY)
C CALCULATE INITIAL SPEED OF SOUND
SS=SQRT(GAMA**22/T/M)
C CALCULATE INITIAL MACH NUMBER
XMA=U/SS
AXI=(IJ-UB)*2+J+2)
UR=SQRT(AXI)
UM=1.21E-5*(I+5)/(I+110)
REP=PHO*UR*P/UM
XMAP=XMA*UR/U
WRITE(6,116)P,MM,RHO,MS,F(NS1),UP,VP
WRITE(6,118) GAMA,CP,SS,XMA ,XMAF,REP
999 CONTINUE
100 FORMAT(4E-10,3I
101 FORNAT(10I5)

```



```

*DECK*WATEF
SUBROUTINE RAT.F
COMMON /PA(40),CVI(40),WA(40),RUF(40),PSIV(40),HSE(40),VE(40)
1 S(40),I(40),PI(40)
2 RUC(40),RUC(40),RUC(40),Y(40),RUFEE(40)
3 ISO,OS,SMAX,RO,FOY,MAI9,PAE,DI5,IS,DPF,DRHDP,OSE,JOE,DT,QMS
4 VM,SE,AE,AP,TP,PHOP,TP,CAE,DI5,IS,DPF,DRHDP,OSE,JOE,DT,QMS
5 JU,CO,CFV,DRHDP,DP,DA,OS,ARW22,UM,SI,3XG,BETA
6 RUAI,XJCI,KUFEE,RUFEE,PI,KI(8),MOUTN(9),UL(20),RUFEE(20)
COMMON RUFEE(20),J32(20),KDI(8),FE(20),TE(20)
7 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
8 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
9 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
10 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
11 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
12 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
13 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
14 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
15 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
16 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
17 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
18 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
19 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
20 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
21 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
22 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
23 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
24 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
25 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
26 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
27 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
28 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
29 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
30 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
31 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
32 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
33 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
34 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
35 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
36 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
37 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
38 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
39 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
40 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
41 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
42 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
43 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
44 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
45 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
46 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
47 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
48 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
49 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
50 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
51 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
52 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
53 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
54 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)
55 RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20),RUFEE(20)

```

SUBROUTINE NAME TRACE

DATE: 1967 JUN 13 14:45:30

PAGE 5

CARD NO: 1001-1000-1000-1000

1001-1000-1000-1000

01= AMOZERKI(1)

02= AMOZERKI(1)

03= AMOZERKI(1)

04= AMOZERKI(1)

05= AMOZERKI(1)

06= AMOZERKI(1)

07= AMOZERKI(1)

08= AMOZERKI(1)

09= AMOZERKI(1)

10= AMOZERKI(1)

60

65

END

TRACE

```

*DFCK,ETAGS
SUBROUTINE ETAGS(ETA,CG,SIP
IF ((CG.LT.1.0) .AND. (CG.GT.5.0)) GO TO 4
5 CONTINUE
WRITE(6,100) CG,SIP
STOP
100 FORMAT(14,'*----- CG=0, SE=0.3)
4 CONTINUE
E01=0.01
DETA=0.5*ETA
ETA=1.0E-5*DETA
DELT=1.0
1 CONTINUE
ETA=ETA *DETA
XLMSE=EXP(-ETA)
XRMS=CG*(1.0+ETA)
DEL=XLM5-XRMS
EPS=ABS(XLMSE-DEL)
IF (ABS(DEL).LE.EPS) GO TO 3
IF ((DEL*DEL).GE.0.0) GO TO 2
ETA=ETA-DETA
DETA=0.5*DETA
GO TO 1
2 DELT=DEL
GO TO 1
3 CONTINUE
RETURN
END

```



```

60      B(4,4)=-2A*F3
        QX(1)=QX(1)*ADM+QUAI*((RUF1EI-RUEI)*F1)+JAE-F1*PHADS)
        QX(2)=2A(2)*ADM+QUAI*((RUF2EI-RUEI)*F2)+JAE-F2*PHADS)
        QX(3)=2A(3)*ADM+QUAI*((RUF3EI-RUEI)*F3)+JAE-F3*PHADS)
        QX(4)=QX(4)*ADM+QUAI*((RUF5EI-RUEI)*F5)+JAE-F5*PHADS)
        DO 2 I=1,4
          DO 2 J=1,4
            B(I,J)=A(I,J)*ADM
            IF (I.EI.,J) B(I,J)=1.0+3(I,J)
        2 CONTINUE
        IF (IRUC.EQ.0) GO TO 6
        DO 4 I=1,4
          WRITE(6,100) (B(I,J),J=1,4),QX(I)
        4 CONTINUE
        6 CONTINUE
        CALL SLP(IX,R,4)
        IF (IRUC.EQ.0) GO TO 7
        DO 5 I=1,4
          WRITE(6,100) (B(I,J),J=1,4),QX(I)
        5 CONTINUE
        7 CONTINUE
        C      QX(I) IS NOW IN MOLE/54 UNIT
        DO 3 I=1,3
          3 9=I+1
          -----> CHANGE BALANCE IMPOSED (-----)
          DF(4)=2X(1)-QX(2)*WI(4)
          DF(5)=QX(4)*WI(5)
          DF(6)=QUAI*((RUF5EI-RUEI)*F5)+JAE-F5*PHADS)*WI(6)
          DF(7)=QUAI*((RUF7EI-RUEI)*F7)+JAE-F7*PHADS)*WI(7)
          DF(8)=QUAI*((RUF9EI-RUEI)*F9)+JAE-F9*PHADS)*WI(9)
          DF(I) NOW IN MASS FRACTION UNIT
        C      100 FORMAT( 2X, 1P16E12.3)
          RETURN
        END

```


SUBROUTINE BREAK TRACE

```

60      F(I)=F(I)*W(I)/RUEX
        PSI(I)=F(I)
        C 44 CONTINUE
        ---- F(I) IS NUM IN MASS FRACTIONS -----
        MSX=MSX/RUEX
        UAZ=UAZ*F
        RHOX=ROE*K/LX
        HX=MSX*JA2/I2*G/GIEP
        ---- CALCULATING GAS PROPERTIES PLMINE JET SHOCK ----
        ITRK=0
        RHOXY=0.25
        SMOX=SMO4/RHOXY
        C 50 CONTINUE
        WRITE(6,121) PX,UX,MSX,RHOX,IX,MAX,(F(I),I=1,NS1)
        DO 40 I=1,100
        PX=PX+2*UX*UX*(1.-RHOXY)
        MY=MSX*JA2*(RHOXY*RHOXY)/I2.*GIEP
        ITRK=ITRK+1
        I1=TRK/PX*RHOXY
        CALL TIM(IH,IV,I1,F,NS1,CBY)
        RHO=PY*IX/RR2/IV
        IF (IRJ5.NE.6)
        IWRITE (5,101) 240X,RHOY,RHOXY,2HJ,2Y,IV ,IV
        IF (ABS(4440/RHOY)-1.5)*IF, 9.30IF 50-IF 41
        RHOY=0.5*(RHOY+RHO)
        RHOXY=RHO/RHOY
        C 40 CONTINUE
        IF (IRJ5.NE.6) WRITE(6,128) RHOX,RHOY,RHOXY,RHO,2Y,MY
        129 FORMAT (7X,6 --- SHOCK CALCULATION DIVERGED*/P1(L1,3/)
        C 41 CONTINUE
        UY=UX*RHODAY
        RHOV=RI/(CHV*W)
        TPOI=1./TBP
        P=PY/RI
        TRPV=1./TBPDI -20H*ALCG(2V)
        WRITE (5,122) 2V,UY,MY,RHOY,IV ,IRJ5
        ---- COMPUTE FLASH VAPORIZATION AT SHOCKED PRESSURE ----
        UMS1=21.5*(Y*1.5)/IV+113
        FV=C.0
        IF (TRPV .LT. IL0) FV=C.0*(TLG-TDGV)/CH
        IF (FV .GT. 1.0) FV=1.0
        ---- JET BREAK ----
        ULI=ULI+ SING(MT4)
        MERRHOP JVUV*G*DE/SIS
        IF (MERRHOP) WRITE(5,126) ME
        REJ=RHOP*UY*G*DE/UM
        XXX=REJ/(1.0+2.0*REJ/ SORT(P))
        XXX=SQRT(XXX)
        IF (IRJ5.NE.0)
        IWRITE(6,101) RHOX,RHOY,RHO,UL,DE,UY,XXX,IV
        TRK=TRK+(RHO/RHOY)*J3.656*(UM/UMI)*J3.333)*0.5*J/JV*XXX
        XXX=TRK*UY*G*DE/UM
        UTK=0.5*XXX*TRK+ULI
        RHO=TRK*TRK*TRK+ULI*TRK

```

```

110 I=(URK*.T.CE) *BK=DE
    XK=DK*50
    ----- CALCULATING DROPLET SIZE -----
    MEL=DE*JY*CV*RHOL/SIG
    GAMA=CV/(CV+1/MAX)
    SSS= Sqrt(GAMA*DE*2*PI/MH)
    XH=UY/ZE
    D3D=8.3E-16*(DE/0.375)**3*(DE/J*H)*DE**(-0.25)*DE
    D3Z=1.2E-13
    DMX=D3D**4.5*(D3D/D3Z)**(0.16)
    IF(NCR.EQ.0) GO TO 62
    ***** VAFOR PRESSURE BREAK UP
    61 XZPS=0.1
    ISH=TAP/(1.0-KLPS)
    C***** TSH= SPAT19 TEMPERATURE
    IF (TLO.GT.TSH) GO TO 63
    AX1= ((JY-VLI+ SIN(META))**2 +YD**2)
    YZ= Sqrt(AX1)
    HSY=HY*JY*U/(L.*J*CL)
    IIEY
    CALL TIV(MSY,TSY,II,F,MSI,CSY)
    XZ= ALJQU*(SY-TLO)
    YZ= SRTII.COPY(RHUY)
    XZ=1E-4*YZ*(IV-FL)*Y*2
    C***** F2.2 MOLECULAR HEAT TRANSFER
    QP=0.5*Y*2*BY*XZ/Y/CT*EFFV
    FZ=COU*(TSH-TLO)*RHOL*Z/(1.+D*ZP)
    XAX1=BY**C*(JY-JLI)**2*DE/DMEL
    UK=2.E-5*XAX1*FZ*YLI
    XK=V*Y*Z
    140 IF(YK.GT.(HMSL)) YK=V*(HSL)
    XK=(XAX1*FZ*UR I)*YK*50
    DMX=0.1
    MEL=0.6
    D3D=0.2
    C***** INPUT INITIAL PARTICLE SIZE
    READ(9,10) D3Z
    GO TO 52
    C***** VAFOR PRESSURE BREAK UP OCCUR INSTANTLY INITIAL CONDITION
    C***** 43S TO BE INPUT
    63 READ(5,10) D3Z1,URK,XK,MEL,DIC,D3Z1,DMX
    62 CONTINUE
    WRITE (6,123) MEL,REJ,FB,URK,XK,YK,MEL,DIC,D3Z1,DMX
    IF (IFK.GT.0) GO TO 45
    IFK=1
    C***** MODIFYING SHOCK FRONT CONDITIONS
    46 CONTINUE
    47 CONTINUE
    IF(1)=F130(1)
    IF(2)=F130(2)
    IF(3)=F130(3)
    F130(1)=F130(2)

```

SUBROUTINE BREAK TRACE

```

F2130(11) = F2130(2)
F3140(11) = F3140(2)
F4150(11) = F4150(2)
F5160(11) = F5160(2)
F6170(11) = F6170(2)
F7180(11) = F7180(2)
F8190(11) = F8190(2)
CALL LIPLN(Y3K, Y130, Y0, 2X, I3)
CALL LIPLN(Y3K, Y130, Y0, 1X, I3)
CALL LIPLN(Y3K, Y130, Y0, 0X, I3)
CALL LIPLN(Y3K, Y130, Y0, F(1), I3)
CALL LIPLN(Y3K, Y130, Y0, F(2), I3)
CALL LIPLN(Y3K, Y130, Y0, F(3), I3)
CALL LIPLN(Y3K, Y130, Y0, F(4), I3)
CALL LIPLN(Y3K, Y130, Y0, F(5), I3)
CALL LIPLN(Y3K, Y130, Y0, F(6), I3)
CALL LIPLN(Y3K, Y130, Y0, F(7), I3)
CALL LIPLN(Y3K, Y130, Y0, F(8), I3)
C***** F(I) ARE IN WALE/CM
MAY=5.5
DO 49 II=1,NS
MAX=MAX*F(II)
49 CONTINUE
R40X=PX*MAX/(P+2*IX)
M4(1)=MAX
PT3(1)=PX
UT3(1)=JK
VT3(1)=K
R40I=240-R40X
F3(1,1)=F(1)
F3(2,1)=F(2)
F3(3,1)=F(3)
F3(4,1)=F(4)
F3(5,1)=F(5)
F3(6,1)=F(6)
F3(7,1)=F(7)
F3(8,1)=F(8)
DO 31 JJ=1,5
O7(JJ)=0.0
O0 72 II=1,NS
O94J33=O94J33+K(9JJ,II)*F3(II)
32 CONTINUE
31 CONTINUE
T10=TT(1)/I00.2
M4=100.0*(O0(5)*T10+O0(6)+F10*(30(11)+M3*(O0(2)/2.0+T10*(O0(3)/7.
1+T10*O0(4)/74.3)))
M4=K4
M5K=MY+3.5*UT(1)*UT3(1)/OTE
M5E(1)=M5K
XXX1=K4*OT3(1)*UT3(1)
KUE(1)=K4*XI
R02E(1)=K4*XI*UT3(1)
R4UMSE=100-XXX1+M5E(1)
RUF12(1)=XXX1*FT3(1,1)

```

170

175

180

185

190

195

200

205

210

215

220

```

225 RUF5(I)=XX1*FT3(2,I)
    RUF3(I)=XX1*FT3(3,I)
    RUF4(I)=XX1*FT3(4,I)
    RUF5(I)=XX1*FT3(5,I)
    RUF6(I)=XX1*FT3(6,I)
    RUF7(I)=XX1*FT3(7,I)
    RUF9(I)=XX1*FT3(9,I)
    WRITE(6,120)
    I=1
230 WRITE(6,100)
    WRITE(6,100) S(I),FT3(I),FT3(I),FT3(I),FT3(I),I,I=1,MS)
    WRITE(6,137)
    WRITE(6,103) I,MA(I),MS(I), RUF5(I),RUF3(I),
1 RUMS(I),RUF1(I),RUF2(I),RUF3(I),RUF4(I),RUF5(I)
2,RUF5(I),RUF7(I),RUF9(I)
    DO 49 II=1,MS
    F(II)=F(II)*MI(II)
    PSIV(II)=F(II)
43 CONTINUE
240 C***** F(II) NOW IS IN MASS FRACTION
    UX2=UX*JX
    GO TO 50
45 CONTINUE
245 RPO=0.5*JZ
    S150 = 1.05*J2*DEOP/RPO
    SE = SO + S150 * ((XK)/(150. * DEOP))**4
    IF(XK .LE. (70.*DEOP)) SE = SOP + S150*((70./150.))**4)*X3K/(70.
1 *DEOP)
    SEFFAE*SE
    VE=XK
250 AC=0.5*PI*VE*SE
    ***** RELAX TO EXTERNAL PRESSURE RAPIDLY *****
    DO 52 ID=2,NPS
    IF (XK, SE, S(ID-1),AN),XK,LT,S(ID)) GO TO 53
52 CONTINUE
    WRITE (5,200) IV, S(1),XK
    STOP
53 CONTINUE
    XW=FF*XLI
    XNL=(1.-FFV)*XLI
    CALL LIPUN (XK,PT,S,DEK,TD)
    CALL LIPUN(XK,PT,S,DEK,TD)
    BETA=C.3*PI
    A=AE
265 P=23K+123K*( S14(4,14)**2)*(1.0-RHOXY)
    P=FFPE
    ***** MIXING OF SHOCKED AIR WITH FLASH VAPOR *****
    XNG=2MOKAUBA
    XW=XNG*XW
    WRITE(6,124) XW,XNL,XM3,XM4,AE
    F(NS1)=XW/XM4
    U=XNG*U+XW*UAK+(PY-D)*A1/XM4
    U=XNG*U+XW*UAK+(PY-D)*A1/XM4
    MS=CBLT55K+3M+10K*U44/12+8*Gte
    MS=(XMG*MS+XW*U)/XM4
    
```

```

00 24 I=1,NS
24 F(I)=PSI(V(I))*(1.0-F(NS,I))
C***** NORMALIZED MASS FRACTION
WM=0.0
00 29 I=1,NS1
C F(I)-44E IN MASS FRACTION
MEMP+F(I)/MI (1)
29 CONTINUE
WMS1.0/M
WMS=(WP)/2.0*CTE
II=IX
CALL JNMM(I,II,F,MS,IS)
RHO=PA/RRT/T
C***** INITIAL OVA TURE AREA AFTER MIXING
A=XM/(R*COU)
CA=4-4E
R=2*0
T=2*PI*RY
UP=URK
SI=XRK
XLO=XM
CALL TTM(MS,TS,II,F,SI,PSR)
WRITE(6,105) (F(I),I=1,NS)
100 FORMAT(10F3)
101 FORMAT(2X,10E12,3)
102 FORMAT(2X,15,1P10,12,*)
103 FORMAT(14,1D,1P12,3,1X,1D,12,3)
104 FORMAT(24,EMASS FRACTIONS OF SPECIES,1X,4D10,3,1X,4D10,3,1
1 210,3,1X,0D,10,3,1X,4E-10,3,1X,4E-10,3,1X,0D,10,3,1X,0D,10,3,1
2X,R2=,E10,3,1X,RHO=,E10,3,1X,PA=,E10,3,1X,
105 FORMAT(1X,AV=,E0,2(1P10,11,3//)
106 FORMAT(1X,AV=,E0,2(1P10,11,3//)
1 9X,REJ=,7X,REUF=,7X,REUFC=,7X,REUFS=,7X,REUFC=,7X,REUFS=,7X,REUFC=
2X,2X,REUFC=,7X,REUFS=,7X,REUFC=,7X,REUFS=,7X,REUFC=,7X,REUFS=
106 FORMAT(1X,SI,SI,2X,PE=,8X,PI=,8X,PIE=
1 2X,PE= (MOL/GM)=)
120 FORMAT(1X,-----) 400 FIEL FREE STREAM CONDITIONS IN SHOCK FR
10NT BY SINGLE ITERATION -----)
121 FORMAT(2X, CONDITIONS IN FRONT OF SHOCK *2X, *2X, *2X, *2X,
1 2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X,
2 2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X,
2 MASS FRACTIONS F(I)=,4,12,3//)
122 FORMAT(2X, CONDITIONS BEHIND SHOCK *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X,
1 210,3,2X,AV=,E10,3,2X,RHO=,E10,3,2X,PA=,E10,3,2X,PI=,E10,3,2X,PIE=,
2 210,3//)
123 FORMAT(2X, jet data *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X,
1 2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X, *2X,

```

SUBROUTINE M33AK TRACE

```

2 *YK=,E10.3/2X,*WEL=,E10.3,2X,*D3L=,E11.3,2X,*D72=,E11.3,2X,
3 *DMX=,E10.3/)
124 FORMAT(2X,'INITIAL FLOW RATE (G/SEC)*2X,'AMV=,1PE10.3,2X,'XML=,
1 E10.3,2X,'XMA=,E10.3,2X,'XMA=,E10.3,2X,'AL=,E10.3/)
125 FORMAT(2X,'----- CONDITIONS OF THE INITIAL MIXTURE AT 5*F10.4
1 /2X,'U=,E10.3,2X,'MSV=,E10.3,2X,'MSK=,E10.3,2X,'MST=
1E10.3,2X,'ISR=,E10.3,2X,'T=,E10.3,2X,'TMO=,E10.3,2X,'A=,E10.3)
126 FORMAT(2X,' --- ACFDY-AMIC BREAK-UP IS IMPOSSIBLE, WE=,1PE10.3
1/)
127 FORMAT(2X,' -- INITIAL MIXING CALCULATION DOES NOT CONVERGE TO
1CORRECT SOLUTION',15/1P:11.3)
RETURN
END

```

335

340

*DECK.LIPLN

SUBROUTINE LIPLN (SK,I,S,RA,K)

C A SCHEME FOR LINEAR INTERPOLATION

DIMENSION I(20),S(20)

22 RA= (R(K-1)+S(K)-SX) * A(K) + (SX-S(K-1)) / (S(K)-S(K-1))

RETURN

END

5

*CHECK DATA

BLOCK DATA

COMMON /HEAT/ XL(5,9),MI(9)

***** COEFFICIENTS FOR CURVE 4 POLYNOMIALS

***** FROM JANNAF CURVE FITTING

***** XL(I) IS THE I-TH COEFFICIENT AND XL(6,1) AND XL(6,2)

DATA XL

- 1 6.2631E+0, 1.4256E+0, -7.9751E-1, 3.6333E-2, 7.7635E-3, 2.3473E+2
- 2, 4.9634E+0, -3.6173E-4, 7.9172E-5, -6.8372E-6, -1.7591E-5, -1.0313E+0
- 3, 5.9017E+0, -1.6173E-1, 3.3745E-2, -2.0531E-3, 1.3127E-2, 5.6115E+1
- 4, 4.3042E+0, -7.2222E-3, 1.2705E-5, -2.7745E-0, 1.2451E-2, -6.3041E+1
- 5, 5.0069E+0, -9.0135E-2, 4.9729E-3, 4.1333E-3, -5.7493E-3, 1.1140E+2
- 6, 7.2410E+0, 1.2535E+0, -1.5545E-1, 1.0505E-2, -5.4603E-2, -2.4210E+0
- 7, 6.2631E+0, 1.4256E+0, -7.9751E-1, 3.7883E-2, 9.1651E-3, -1.9245E+0
- 8, 6.7523E+0, 1.7207E+0, -1.1765E-1, 3.4323E-2, -1.9375E-2, 5.3420E+1
- 9, 2.1523E+0, 0.0, 0.0, 0.0, 0.0, 0.0

***** MI(I) ARE MOLECULAR WEIGHT FOR EACH SPECIES

DATA MI(10)5.007E+01,10.017E+01,19.019E+01,32.042E+01,61.067E+01

END

TRACE

```

*DECK,ERF
FUNCTION ERF(X)
----- THIS CALCULATES THE ERROR FUNCTION AND ITS COMPLEMENT -----
DATA EPSILM/1.0E-10,M/100,CONST/1.28791674557
KOMP=+1
GO TO 5
ENTRY ERF
KOMP=-1
5 IF (X) 10,59,19
10 Z=-X
15 Z=X
20 IF (Z=5.) 25,25,45
25 ZSQ=Z**2
    TERM=CONST*EXP(-ZSQ)
    I=C
    R=1.0
30 I=I+1
    K=R**2.0
    TER=Z**2*250*I**4/R
    ERF=ERF+TERM
    IF (I-N) 35,43,43
35 IF (TERM-EPSILM=0) 40,36,39
43 IF (ERF-1.0) 40,53,45
45 ERF=1.0
50 IF (X) 57,55,62
55 ERF=0.0
    GO TO 65
60 ERF=-ERF
65 IF (KOMP) 70,59,59
----- COMPUTE COMPLEMENTARY ERROR FUNCTION ERF -----
70 ERF=1.0-ERF
93 RETURN
END
    
```

```

*DECL, INV
SUBROUTINE INV (M, I, II, F, NS1, CP)
C ----- THIS SUBPROGRAM WILL INVERT TEMPERATURE FROM ENTHALPY BY THE
C REACTION METHOD -----
5 DIMENSION J(6), F(6)
COMMON /HEAT/ TOL, IML, JML, JOL, MV, TPC, JLU, JML, JOL, JLC,
1 FV1, AK, BK, ME, IT, FV, FV1, FV2
1 CONTINUE
10 DO 10 J=1,6
11 F(J)=0.0
12 DO 11 I=1, NS1
13 F(I) = F(I) + F(I) * F(I)
14 CONTINUE
15 EPS=1.0E-6
16 I=1
17 DO 17 IF=1, 100
18 F10=F/100.0
19 G=100.0 * (-D(5)/T10 + J(5)) + T10 * (D(2) + I10 * (D(2)/2.0 + I10 * (D(2)/
20 * 3.0 * F10) + D(4) / (4.0 * J(10))) - H
21 CP = (D(6) / (T10 * F10) + J(10) * T10 + D(2) + I10 * (D(2) + I10 * D(2)
22 * 1.0))
23 IF (ABS(G).LT. EPS) GO TO 24
24 F=F/G
25 CONTINUE
26 WRITE (5, 119) II, MV, T, CP, FI, F, EPS
27 STOP
28 CONTINUE
29 F=H/CP
30 PC=AL(1, NS1) * F(NS1) / W(NS1) + (VL(1, 0) * F. 251 / W(0) + VL(1, 1) * 0.759 /
31 W(1)) * F
32 CP=PC
33 T=H/CP
34 CONTINUE
35 119 FORMAT(12A, *--ITERATION DOES NOT CONVERGE TO THE CORRECT TEMPERA
RETURN
END

```

```

*DECK MAIN
PROGRAM J108(INPUT,OUTPUT,TAPE5=INPUT,TAPE6=OUTPUT,TAPE4)
C----- 12/16/73 -----
C----- THIS IS VERSION D -----
C----- THIS DECK FOR CDC USE -----
C-----
C
C ONE DIMENSIONAL REACTING DROPLET-VAPOR-AIR TUBE ANALYSIS
C A SELF-INITIALIZED REACTING DVA TUBE (8/8/73)
C-----
C***** MODIFIED TUBE PRESSURE (11/30/73)
C***** ITERATED SHOCK FRONT CONDITIONS (4/13/73)
C***** GENERALIZED REACTION TYPE AND SPECIES (1/24/73)
C-----
COMMON CPA(40),CPV(40),MA(40),RHOE(40),PSIV(51),MEL(40),UE(40),
1 SC(4),TB(40),PTB(40)
2 RUF(40),RUF2(40),RUMSE(40),Y(40),RUF5E(20)
3 FSO,DS,SMAX,RWD,F40,T,MS,ATD,P,KK,OSP,CF,PKL,R2,AV,G,AI,VE
4 MM,SEAE,RP,TP,RDP,EP,DP,DAE,UTS,IS,URP,KRHP,R3E,OVE,DT,D45
5 DU,DUP,DFV,DRHO,DP,DA,ADS,AKM02,UM,ZI,SKO,BETA
6 KMAI,RUEI,RUF1E1,RUF2E1,RUF3E1,KJF4E1,RUF5E1,P4D3
COMMON RUF4E(20),D32(2),
1 RUF2E(20),RUF3E(20),FOI,F5(8,20),PE(20),TE(20)
2 F(51),DF(51),D32I,D32II,XMLI,XMDU,ETA,ADD,A,I,ZI,VL1,TMETHA
3 F1,F2,F3,F4,F5,PI,AI1,BB,XNF,CF,JP,VP,GAMA,CI,Z2,C3,C4,C5,C6,
4 C7,C8,CP,AP,KMA,XMAP,SS,UR,OUR,H5R,CHSR,XMLC
5 RUF6E(20),RUF7E(20),KJF5E(20)
6 RUF6E1,RUF7E1,RUF8E1,F5,F7,F8,F9,FFPE,FFTE,FFFE
7 FFF,FFV,C9,C2,CF5,C6,C7,KNF1,KNF2,KNF5,B3,B7,FF3,FF4,KNF7
8 XK,YK,TBK,PEK,RJ2BK,KDXY,SOP,ASO,AMCR,DEP,SI:
COMMON STB(30),TTB(30),RHO7B(30),UTB(30),FTB(6,30)
COMMON /HEAT/ XL(6,30),MI(51)
COMMON/LIQUID/ CPL,TBP,JAV,RHOL,MV,TB-O,SIG,TU,JK,J0,WSCP,
9 FFI,AK,BK,ME,ITK,FFJ,TBP
COMMON/INTEG/PS,NPI,I00(20),NS--1DPT,K3,NS,WSI,IBJ,IO,KINE
COMMON /SHOCK/ YO(20),PTB0(20),UTB0(20),FTB0(20),FITB0(20),
1 FTB0(20),FTB0(20),FTB0(20),FTB0(20),FTB0(20),
2 FTB0(20),FTB0(20)
COMMON/CHEM2/CM(51,51),FI(51),MFI(51),MMI(51),MDO(51),DX(51)
1 RC(100,5)
COMMON /SPI0/ I24,I21,I22,I23,I20,I25, I4,INDP,I14,IOK,IF4,IN
COMMON/CHE1/IR(10),IRT(10),IKER(1,8),ED(8),IDENT(51)
100 CONTINUE
17BK=0
20 TO 1002
1001 ITBK=2
1002 LEAD(5,100) (I00(I),I=1,20)
1003 IF(EOF(5)) GO TO 3
3 CONTINUE
WRITE(6,99)
1004 WRITE(6,100) (I00(I),I=1,20)
1005 LEAD(5,102) FFAE,FFPE,FFV,FF3,FF4,PRAL,FFVI
1006 WRITE(6,102) FFAE,FFPE,FFV,FF3,FF4,PRAL,FFVI
C***** CONSTANTS
91=3.14159

```

```

SQUPI= SQRT(PI)
AV=6.02E23
KR=02.06
KRL= 1.987
KRZ= 8.313E7
CIE= 4.106E7
ATD=1.0J13E6
1UK=0
XDK=..
DFV=0.0
DFJ=0.0
DU=0.0
UP=..
URMO=0.0
DT=0.0
DMS=0.0
DTS=..
DKMOP=0.0
DUP=0.0
DRP=0.0
DEP=..
DA=0.0
USE=0.0
DYE=0.0
DAE=..
ETA=0.0
**** RR IN CM3-AFM/G-HOLE-K
CALL PUTIN
DO 35 I=1,NS1
DF(I)=0.0
FI(I)=F(I)/MI(I)
35 CONTINUE
YPEN=YE
SEU=SE
EP= CPL*TP
TPS= EP*UP*UP/(2.0*CTE)
LS1=DS
JSMAX=5.0*DS
DSMIM=DS/10.0
WRITE(6,39)
WRITE(6,100) (IDJ(I),I=1,20)
IPS=0SP
IS=-1
1 CONTINUE
US=DS1
AXX1=((U-UP)**2+J)**2)
JR = SQRT(AXX1)
HSP=MS-(U*U-JR*UR )/(2.0*CTE)
II=IS
CALL TIMV(MSR,IS ,II,F,MSI,CPSP)
KMP=..
KMP=J00
IF (KPL*1.0JE=9) GO TO 40
KMP= (XLI-KM)*MSI)*J00)/(J00*1)
KMP= 3.0*KMP*UP*1/(...)*I*RP*...

```

```

115  XX1=RHU*OHV**5
      XX2=ALD*UK*(TS-TP)
      V= SRT(3.0*P/R40)
G***** V IS MEAN THERMAL VELOCITY OF MIXTURE BY KINETIC THEORY
      XX3=1.0+ALT*V*(T-TP)/XX2
      DP=0.5*XX2*P*XX3/T/CTE*FFEV
      UOUT=-DP/XX1
      PHAV=-1.5*KHC*DUUT/RP
      IF (PHIV.GT. 1.0E-3) DS=DS1/(PHIV**4)
      IF ((IS..T.IG).AND.(DS.GT.DSMIN)) DS=DSMIN
      IF (US.GT.DSMAX) DS=DSMAX
120  41 CONTINUE
      SI=SI+DS
      IF((SI.GT. S(NP3)).OR.(SI.LT.S(1))) GO TO 999
      DO 21 I=2,NPS
125  IF (SI.GC.S(I)-1) .AND. S4.LT.S(I)) GO TO 22
      21 CONTINUE
      WRITE(6,20)I,S(I),SI
      STOP
130  20C FORMAT (1H *, 20T OF RANGE 10**,15, * 5**,E9.3/)
      22 CONTINUE
      CALL LIPN(SI,RUE,S,RUEI, ID)
      CALL LIPN(SI,RUE2,S,RUE2I, ID)
      CALL LIPN(SI,RUE3,S,RUE3I, ID)
      CALL LIPN(SI,RUF1,S,RUF1E, ID)
      CALL LIPN(SI,RUF2,S,RUF2E, ID)
      CALL LIPN(SI,RUF3,S,RUF3E, ID)
      CALL LIPN(SI,RUF4,S,RUF4E, ID)
      CALL LIPN(SI,RUF5,S,RUF5E, ID)
      CALL LIPN(SI,RUF6,S,RUF6E, ID)
      CALL LIPN(SI,RUF7,S,RUF7E, ID)
      CALL LIPN(SI,RUF8,S,RUF8E, ID)
      CALL LIPN(SI,PT9,S,PEI, ID)
      XXX=(SI-XXX)/PRX
      IF(XXX.GT.0.0) XXX=...
      PI=PEI+RUEI*(1.0-RHOXY)* EXP(-XXXX)
      PI=PI*FFEV
      IF (IBUG .NE. 0) WRITE(6,10) RUEI,RUE2,RUE3,RUF1E,RUF2E,
140  RUF3E,RUF4E,RUF5E,RUF6E,RUF7E,RUF8E,PT9,PEI
      DP=PI-P
      ADS=A*DS
      SP = SE
      SE=SOP*SL9*((SI/115.**DECP)**AI)
      IF (SI.LE.(7.**DECP)) SE=SOP*SL5*(71./15.0)**AI)*SI/(70.**DECP)
      YCEN=YPEN*VP/JP*J3
      VE = VPEN*SLC-SP)
      SE=FFAE*SE
      DAE = C*SE*J*SE*VE -AE
      IF(BUG.NE.0) WRITE(6,11) A,SE,VE,YCEN,C,A,SI,VE
      SQDM= SRT(3.0*P)
      XMAP=XMA* UR/U
G***** VISCOSITY FOR AIR IN (MCM-SEC) BY SUTHERLAND LAW
      JM=1.21E-5*(T+115)/(T+11)
      FEP=MO*U*RO/34
      PMAOS=U.

```

```

17.  DFPE...
      DRPE...
      DQ...
      IF (RPA.LT. 1.3E-4) GO TO 7
      CALCULATE DRAG COEFFICIENT
      CDO=24./REP
      COL=0.46
      CUB=1.0
      CD=CDI
      KMK=XMAP/REP
      KKM=1.26 * SQAM*NR
      IF (KKN .GT. 2.00) GO TO 4
      IF (KKN .LT. 1.E-9) GO TO 5
      XN1=KKM** .3
      XN2= EXP(1.2* KKN** .5)
      XN3=KKM**0.7
      XN4= EXP(KKN)
      DRC=1.0- EXP(-KK3*KK4*(CDO-.48)*PEP/8.)
      CDB=5KN*DRE
      4 CONTINUE
      S1= SORT(.5*GAMA)*XMAP
      S2=S1*51
      S3=S2*51
      S4=S3*51
      S32=S2*.5
      S4S1= SORT(I/TP)
      CDFM= EXP(-S32*(1.0+2.0*S2)/50TP)/53
      1 1.4 .54.J*(34+S2)-1.0/2.J/50* EXP(S1)
      2 1. .667*SQPI/SQ
      3 CDB*(CDFM-CUI)*CUI
      5 FP=CD/24.*REP*FFF
      DRPE=4.5*RHOB*FP*U4*(U-UP)*ADS/RHUL/PP/RP
      UQ=3.0*RHOB*JP*AU3/JS/(JP*RHOL*UP)
      IF (IBUG.NE.0) WRITE(6,101) KKN,GK4,DRE,CDFM,CD,FP,S1,SQ
      PHADS=PHIV*ADS
      7 CONTINUE
      KUA=RH0*J*A
      KUAI=1.0/RUA
      KUAP=RHOP*UP*4
      DUP= .4
      IF (FFV.EQ.1.0) GO TO 5
      DUP = DFP / KUAP
      5L IF (IBUG.NE.0)
      .WRITE(6,101) KUAI,UP,U,PHADS,KUZEI,ROEI,DAE,A,DP,DFP
      DU=RUAI*((UP-J)*PHADS+(KUZEI-KUEI*U)*DAE-B*DP-DFP)
      DMS= EP*(UP+DUP)*(UP+DUP)/(2.0*CTE) -HPS
      HPS=RUAI*(HPS-HS)*PHADS+(RUHS-I-KUEI*HS)*UAI-RUI*(J+P3)
      UF(NS1)= RUAI*(1.0-F(NS1))*PHADS-F(NS1)*ROEI*DAE
      CALL CHEMF
      CALCULATING DFI
      F(NS1)=F(NS1)+DF(NS1)
      GO 33 I=1,NS1
      IF (F(I) .LT. .001) GO TO 999

```

PROGRAM JLL TRACE

```

33 CONTINUE
DO 3J I=1,NS1
  FI(I)=F(I)/MI(I)
3L CONTINUE
MS=MS+DMS
MH=MH+(U+DU)*(U+DU)/(2.*CTE)
II=T
IF (IBUG.NE.0)
  WRITE(6,103) DP,DU,DMS,DF(NS1),BETA,DUUP,MH,MS,DS
CALL TINV(M,T,FI,F,NS1,CP)
MM=J.0
DO 29 I=1,NS1
  MM=MM+FI(I)
29 CONTINUE
MM=1.0/MM
  CALCULATE LOCAL SPECIFIC HEAT
  GAMA=CP/(CP-KR1/MM)
  LOCAL SPEED OF SOUND
  SS= SQRT(GAMA*RR2*T/MM)
  LOCAL MACH NUMBER
  XMA=U/SS
  DKHO=PI*MM/RR2/T-RHO
  DA=(PHADS+RUEI*DAE-1+U*ORHO-A*RH0*UJ)/(RH0*U)
  DA=(PHADS+RUEI*DAE+RUA)/(RH0+DRHO)*(U+DU) -A
  AE=AE+DAE
  RP=RP+DRP
  IF (RP.LT.0.0) RP=0.0
  JP=JP+DUP
  U=U+DU
  HPS=HPS+DHPS
  RHU=RHU+DRHU
  A=A+DA
  P=PI
  IS=IS+1
  IF ((SI+JS).GT. SMAX ) GO TO 9
  IF(((IS/IPS)*IPS).NE. IS) GO TO 2
  5 CONTINUE
  EDEN= RHO*AV*FI(IEM)
  WRITE(6,103) SI,T,P,RHO,U,MS,F(NS1),MM
  WRITE (6,110) A,UP,TP,RP,RHJ,AE,TS,M
  WRITE(6,135)
  DO 42 I=1,NS1
    WRITE(6,119) IDENT(I),F(I),FI(I),WDOT(I)
  42 CONTINUE
  WRITE(6,105) EDEN
  WRITE(6,114) DS, DP,DRHO,DU,DMS,DF(NS1),PHADS
  WRITE(6,111) DA,DAE,DUUP ,DRP,DP,PHIV,DFP,DDQ
  WRITE(6,118) GAMA,CP,SS,XMA ,XMAP,REP,XNP
  WRITE(6,101) DS,ETA,SE,VE,VP,C,AI,SXO ,BETA
  BETA1
  IF ((SI+DS) .GT. SMAX ) GO TO 999
  2 CONTINUE
  50 T J 1
  91 WRITE(6,11.9)
  946 CONTINUE

```

```

00 34 I=4,NS1
F(I)=FI(I)*MI(I)
34 CONTINUE
SI=SI-DS
YREN=YREN-JP/UP*DS
DS1=DS1*0.5
IKT=1
999 CONTINUE
IF(IIBK.GT.2) CALL EXIT
IF(WE-GE.WECR) CALL EXIT
WRITE (6,126) WE
GO TO 1001
99 FORMAT(141)
100 FORMAT(26A )
101 FORMAT (2X,1JE12.3)
102 FORMAT (6E10.3)
103 FORMAT(/,/, SICH)=,F6.4/1X,*TK)=,F8.2/1X,*PIDYNE(M2)=,1PE10.3
104 *1X,*RHOD(M/ML)=,E10.3/1X,*UICH/SEC)=,E10.3/1X,*VISCAL/G*ML)=,E10
2.3/1X,*FV)=,E10.3/1X,*HM)=,E10.3)
105 *1X,*2X,*DU)=,E10.3/1X,*2X,*DRHO)=,E10.3/1X,*2X,*DU)=,
E10.3/1X,*2X,*DHS)=,E10.3/1X,*2X,*DFV)=,E10.3/1X,*2X,*PHADS)=,E10.3)
106 FORMAT (2X,* --- ELECTRON DENSITY)=,E10.3)
107 FORMAT (2X,* --- PRODUCTION RATES FOR EACH REACTION (MOL/SEC-CM3)/2X,1P
11E12.3)
1 7 FORMAT(/2X,* PRODUCTION RATES FOR EACH SPECIE*,/2X,1P,12.3)
1.8 FORMAT(/, CHANGES OF MASS FRACTION OVER DS=,1PE10.3/4X,1P10E12.3)
1)
105 FORMAT(1X,* --- REACHING THE LOCAL SONIC SPEED ----*)
11 FORMAT (2X,*A=,1PE10.3/2X,*UP)=,E10.3/2X,*TP)=,E10.3/1X,*RP)=,
E10.3/2X,*RHOP)=,E10.3/2X,*AE)=,E10.3/2X,*TSR)=,E10.3/2X,*H)=,
E10.3/)
111 FORMAT(2X,*OA)=,1PE10.3/2X,*OAE)=,E10.3/2X,*OUP)=,
E10.3/2X,*ORP)=,E10.3/2X,*OP)=,E10.3/2X,*PHIV)=,E10.3/2X,*DFP)=,
E10.3/2X,*OJ)=,E10.3/)
114 FORMAT(2X,* LOCA_ THERMODYNAMIC INFORMATION /, 2X,*GAMA=*
1 F6.3/2X,* CP(CAL/G)=,1PE10.3/2X,*SS)=,E10.3/2X,*AA)=,E10.3/2X,*
*MAP)=,E10.3/2X,*R2)=,E10.3/2X,*P/SEC)=,E10.3/)
126 FORMAT(2X,* --- AERODYNAMIC BREAK-UP IS IMPOSSIBLE, WE=,1PE10.3
1/)
119 FORMAT(9X,84,3X,1PE12.3)
13. FORMAT(7X,*SPECIES*,5X,*F(I),5X,*FI(I),5X,*MD3(I)*)
90 REMIND 4
STOP
END

```

TRACE

CDC 6300 FTY V5.0-300 OPT=0

*DECK.PUTIN

```

SUBROUTINE PJTIN
DIMENSION DD(6),Y5(2),
DIMENSION YOMAX(1,40)
COMMON CPA(40),CPV(40),MA(40),RMODE(40),PSIV(50),TSE(40),JE(40),
1 S(40),TB(40),PTB(40)
2 ,RUE(40),KUE(40),RUHSE(40),Y(40),RUFSE(2)
3 ,SO,DS,SMAX,RHO,FV,U,I,HS,ATD ,P,K,DS,CFE,RN1,R22,AV,C,41,VE
+ ,MH,SE,AE,RPO,TP,R4OP,EP,DEP,DAE,DTIS,DKP,RUR4D,JE,OVE,DT,CH45
5 ,DU,DUP,DFV,DRHJ,DRJ,ADA,ADS,ARKD2,J4,51,SKU,SET4
6 ,NUAI,RJEL,RUF1E1,RUF2E1,RUF3E1,RUF4E1,RUF5E1,R4C7E
COMMON RUF E(20),D32(20), RKI(5),MOM(5),... (20),RUF1E(20),
1 RUF2E(20),RUF3E(20),FQI, FE(6,20),PE(20),TE(20)
2 , F(51),DF(51),D32I,D32II,XMLI,XMDU,ETA,ALD,A,I,JE,V,I,THETA
3 ,F1,F2,F3,F4,F5,PI,A11,88,XNF8,CF8,JP,VP,SAMA,C1,22,3,32,CF,40,
+ CF,C0 ,CP,A,RP ,XMA,XMAP,SS,UR,OUR,MS,X,DMSR,XMLO
5 ,RUF6E(20),RUF7E(20),RUF8E(20)
6 ,RUF6E1,RUF7E1,RUF8E1,F6,F7,F8 ,F4 ,FPEE,FFTE,FFFE
7 ,FFF,FFE1 ,CF1,CF2,CF5,CF6,CF7,XNF1,XNF2,XNF ,SE,PT,FF3,FF-,XNRT
8 ,XBK,YBK,TBK,PEBK,RJ2BK,RMXY,SOP,NSO,NMCR,DECP,S32
COMMON STB(30),TTB(30),RHTB(30),UTB(5),FTB(6,31)
COMMON /HEAT/ XL(6,21),WI(51)
COMMON/LIQUID/ CPL,TBP,DHV,RHOL,MV,TSFO,SIS,TLO,JML,DO,MECR,
1 FFEVI,AK,BK,ME,ITK,FFV,TBP
COMMON/INTEG/NP5,NPI,LOU(20),NSL,IOP1,KO,NS,N31,IEJ,IO,IKINE
COMMON /SHOCK/ Y0(20),PIB0(20),UTB0(20),F1B0(20),F1B0(20),F1B0(20),
1 F2B0(20),F3B0(20),F1B0(20),F1B0(20),F1B0(20),F1B0(20),
2 F7B0(20),F8B0(20)
COMMON/CHEM2/CM(51,51),F1(51),MP(51),M1(51),MOUT(51),TX(51)
1 KC(10,5)
COMMON /SPID/ IZ4,IZ1,IZ2,IZ3,IZ-,IZ-, I4,INJP,IE4,IX,IFM,IM
COMMON/CHEM1/IMR(100),IRT(100),IKRR(100,3),ITD(5),IEMT(51)
READ(5,100) MV,CPL,TBP,OHV,RHOL,XMLI,MLI,THETA
1 ,TLO,SIG,UM,DO,AK,BK
THETA=THETA*PI/180
V=VLI* COS(THETA)
WRITE(6,115) MV,CPL,TBP,OHV,RHOL,XMLI,MLI,THETA
1 ,TLO,SIS,UM,DO,AK,BK
READ(5,101) NPS,NSL,NUS,IBUG,IKINE,NSI,NSO,NMCR
WRITE(6,107) NPS,NSL,NCS,IBUG,IKINE,NSI,NSO,NMCR
NS=NS1-1
DO 27 I=1,NS1
F(I)=...
F(I)=0.0
MOUT(I)=0.0
27 CONTINUE
WI(NS1)=WJ
READ(5,100) SO,DS,SMAX,OSP
WRITE(6,100) SO,DS,SMAX,OSP
READ (5,100) ALD,ALT,SOP,AI ,MECP
WRITE(6,100) ALD,ALT,SOP,AI ,MECR
AI=AI-1.0
XNUS=NOS
DE= SORT(XNUS) *DO
XNSO=NSO

```

25
48

```

UEQP=SQRT(XNS0)
CALL CPUIN(IKINE,NS1)
C -----> F(I) INPUT AS MOLE/FM OF MIXTURE <-----
C ASSUME ALL POSITIVE IONS ARE NO+ I.E. XNRT=1.0
KNRT=1.0
C CALCULATE MOLECULAR WEIGHT OF THE INITIAL MIXTURE
DO 5 I=1,NPS
  READ(5,100) S(I)
  COMPUTE AVERAGED FLUX OF ENTRAINMENT
  TT8(I)=0.0
  PT8(I)=0.0
  JTB(I)=0.0
  DO 11 II=1,NS
    FT8(II,I)=...
  11 CONTINUE
  Y(I)=0.0
  YO(I)=0.0
  JE(I)=...
  WRITE(6,105)
  DO 1 J=2,NSL
    READ(5,100) Y(J),PE(J),TE(J),UE(J),(FE(II,J),II=1,NS)
    ***** FE(II) ARE IN MOLE/FM-OF-MIXTURE
    ***** PE(J) INPUT AS ATM AND CONVERTED TO DYNE/CM2
    IF(I.GT.1) GO TO 1
    YMAX(II,J)=Y(J)
  1 CONTINUE
  DO 3 J=2,NSL
    J1=J
    DY= Y(J)-Y(J-1)
    DY5= 0.5*DY
    PT8(I)=PT8(I)+DY*PE(J)
    FT8(I)=FT8(I)+DY*TE(J)
    UT8(I)= UT8(I)+DY5*(UE(J)+UE(J-1))
    DO 10 II=1,NS
      FT8(II,I)= FT8(II,I)+DY*FE(II,J)
  10 CONTINUE
  IF (I.GT.1) GO TO 3
  YO(J)=Y(J)
  PT8(J)= PT8(I)/Y(J)
  TT8(J)= TT8(I)/Y(J)
  JTB(J)= JTB(I)/Y(J)
  FT8(J)= FT8(I)/Y(J)
  FT80(J)= FT8(2,I)/Y(J)
  FT800(J)= FT8(3,I)/Y(J)
  FT8000(J)= FT8(4,I)/Y(J)
  FT80000(J)= FT8(5,I)/Y(J)
  FT800000(J)= FT8(6,I)/Y(J)
  FT8000000(J)= FT8(7,I)/Y(J)
  FT80000000(J)= FT8(8,I)/Y(J)
  3 CONTINUE
  4 JTB(I)= JTB(I)/Y(J)
  FT8(I)=PT8(I)/Y(J)

```

```

115 FTB(I)=FTB(I)/Y(J1)
    SUM=0.0
    DO 12 II=1, NS
      FTB(II,I)= FTB(II,I)/Y(J1)
      SUM=SUM+FTB(II,I)*WL(II)
12 CONTINUE
    DO 14 II=1, NS
      FTB(II,I)=FTB(II,I)/SUM
14 CONTINUE
    ***** NORMALIZED IN MOLE/GM OF MIXTURE
    MA(I)=0.0
    DO 31 II=1, NS
      MA(I)=MA(I)+FTB(II,I)
31 CONTINUE
    MA(I)= 1.0/MA(I)
    RHO(TB(I))= PTB(I)*MA(I)/(RR2*TTB(I))
    DO 31 JJ=1,6
      DO(JJ)=0.0
    DO 32 II=1, NS
      DO(JJ)= DO(JJ)+ XL(JJ,II)*FTB(II,I)
32 CONTINUE
31 CONTINUE
    TL=TTB(I)/ 1000.0
    HE=1000.*(-DO(5)/TL+DO(6)+TL*(DO(1)+TL*(DO(2)/2.+TL*(DO(3)/3.
    1 +10*DO(4)/TL))
    HSE(I)=HE+.5*UTB(I)*UTB(I)/CTE
    XXXI=RHO(TB(I))*UTB(I)
    RUE(II)=XXXI
    FUF2(I)=XXXI*UTB(I)
    RUMSE(I)=XXXI*HSE(I)
    RUF1E(I)=XXXI*FTB(1,I)
    RUF2E(I)=XXXI*FTB(2,I)
    RUF3E(I)=XXXI*FTB(3,I)
    RUF4E(I)=XXXI*FTB(4,I)
    RUF5E(I)=XXXI*FTB(5,I)
    RUF6E(I)=XXXI*FTB(6,I)
    RUF7E(I)=XXXI*FTB(7,I)
    RUF8E(I)=XXXI*FTB(8,I)
    WRITE(6,126) S(I),PTB(I),TTB(I),UTB(I), FTB(II,I), II=1, NS)
    WRITE(6,132)
    WRITE(6,133) I, MA(I), HE, HSE(I), RUE(I), RUF2E(I),
1 RUMSE(I), RUF1E(I), RUF2E(I), RUF3E(I), RUF4E(I), RUF5E(I),
2, RUF6E(I), RUF7E(I), RUF8E(I)
5 CONTINUE
    YMAX=YOMAX(1, NSL)
    CALL BREAK(YMAX)
    SAMA=CP(ICP-RR1/4M)
    C CALCULATE INITIAL SPEED OF SOUND
    SS= SQR(GAMA*RR2*T/MM)
    C CLACULATE INITIAL MACH NUMBER
    XMA=U/SS
    AXX1=((U-UP)**2+JP**2)
    UR = SQR(AXX1)
    JN=1.21E-5*(T+.15)/(T+110)
    FEP=RMO* UK*RP/U4

```

```

XMAP=XMA* UK/U
WRITE(6,116)P,MM,RHD,MS,F(NS1),UP,JP
WRITE(6,118) GAMA,C,SS,XMA ,XMAP,REP
999 CONTINUE
100 FORMAT(8E10.3)
101 FOKMAT(10I5)
102 FORMAT(/ 5X, *I*, 5X,*MA*,10X, *HE*,10X,*HSE*,
1 9X,*REUE*, 7X,*REUE2*,7X,*REUEHSE*,7X,*RUF1E*,7X,*RUF2E*,7X,*RUF3
2E*,7X,*RUF4E*,7X,*RUF5E*,7X,*RUF6E*,7X,*RUF7E*,7X,*RUF8E*)
103 FOKMAT(1H ,I5,1P)3=12.375X,0E12.3)
104 FOKMAT( // 3X,*T(K)*,7X,*CPA*, 7X,*CPV*)
105 FOKMAT(/ 2X,*J*, 4X,*I*,10X,*PE*,8X,*TE*,8X,*UE*
1,4X,*FE(1) (MOLE/GM)=)
106 FOKMAT(1X,12,1X,F6.2,2(1P11E11.3/))
107 FOKMAT(14 ,*NPS=*,I,3X,*NSL=*,I ,3X,*NOS=*,I5,3X,*I3UG=*,I5,3X,
1 *NO. OF REACTIONS=*,I5,3X,*NO. OF SPECIES=*,I5,3X,
2 *NSO=*,I5,3X,*BREAK-UP PARAMETER=*,I2/)
108 FOKMAT(1H ,*INITIAL S*,F7.3,3X,* V.R. STEP SIZE(G/SEC)=*,1PE10.3
1 ,3X,*FINAL DISTANCE=*,E10.3,3X,
2 *PRINT INCREMENT=*,E10.3)
109 FOKMAT(/2X,* C1=*,1PE10.3,2X,*C2=*,E10.3,2X,*C5=*,E10.3,2X,*C6=*,
1E10.3,2X,*C7=*,E10.3,2X,*C8=*,E10.3,2X,*FF3=*,E10.3,2X,*FF4=*,E10.
23)
110 FOKMAT(1H ,* N1=*,1PE10.3,2X,*N2=*,E10.3,2X,*N5=*,E10.3,2X,*N6=*,
1E10.3,2X,*N8=*,E10.3,2X,*N7=*,E10.3,2X,*N8=*,E10.3)
111 FOKMAT(1H ,* NNO*/N=*,1PE10.3,2X,*FNO=*,E10.3,2X,*FE=*,E10.3,2
1X,*FO=*,E10.3,2X,*F=*,E10.3,2X,*FN=*,E10.3,2X,*D2=*,E10.3/
2 3X,*FN2=*,E10.3,2X,*FNO=*,E10.3)
112 FOKMAT(1H ,* ----- KINETIC CALCULATION ONLY -----//)
115 FOKMAT(8X,* --- PROPERTIES OF THE INJECTED LIQUID -----//
1 2X,*MOLECULAR WT.=*,F10.2/20X,*SPECIFIC HEAT(CAL/G-K)=*,1PE10.3/
2 2X,*BOILING PT. (K)=*,E10.3/20X,*HEAT OF VAPORIZATION(CAL/G)=*,E
31.3 /20X,*DENSITY (G/CM3)=*,E10.3/20X,*MASS FLUX(G/SEC)=*,E10.3/
/20X,*JET VELOCITY(CM/SEC)=*,E10.3/20X,*JET ANG-E(RAD.)=*,E10.3/
3 2X,*TEMPERATURE(K)=*,E10.3/20X,*SURFACE TENSION(DYNE/CM)=*,E10.3
5 /20X,*VISCOSITY(G/CM-SEC)=*,E10.3 / 20X,*ORIFICE DIAMETER(CM)=*
7,E10.3/20X,*LIQUID DROP COEFFICIENTS*/20X,*AK=*,E10.3/20X,
3 *BK=*,E10.3/)
116 FOKMAT (/2X, *P=*,1PE10.3
1,3X,*MM=*,E10.3,3X,*RHO=*,E10.3,3X,*HS=*,E10.3,3X,*F=*,E10.3,3X,
2 *UP=*,E10.3,2X,*VP=*,E10.3/ 2X,
2,E10.3,3X,*R2=*,E10.3,3X,*SE=*,E10.3,3X,*YE=*,E10.3)
118 FOKMAT(/2X, * INITIAL THERMODYNAMIC INFORMATION */ 2X,*GAMA=*,
1 F6.3,2X,* Cp(CAL/G)=*,1PE10.3,2X,*SS=*,E10.3,2X,*A=*,E10.3,2X,
1*MAP=*,E10.3,2X,*RE=*,E10.3)
119 FOKMAT (/2X,*---- TS CALCULATION DOES NOT CONVERSE (N IT=*,I5/)
126 RETURN
END

```

```

CHECK.CPUTIN
SUBROUTINE CPUTIN (IKINE, NS1)
COMMON /HEAT/ XL(6,51), MI(51)
COMMON /CHEM2/ CM(51,51), FI(51), WP(51), MM1(51), WDOF(51), QX(51)
COMMON /SPID/ IZH, IZ1, IZ2, IZ3, IZ4, IZ5, IM, INOP, IE4, IOX, IFM, IN
COMMON /CHEM1/ IRR(103), IRT(103), ISKR(100,0), IZD(0), IDENT(51)
AV=6.03E23
IM=NS1+1
INOP=NS1+1
IEM=NS1+1
IOX=NS1+1
IFM=NS1+1
IN=NS1+1
DO 10 L=1, NS1
  READ(5,102) IDENT(L), MI(L), (XL(II,L), II=1,6)
  WRITE(6,103) IDENT(L), MI(L), (XL(II,L), II=1,6)
  IF (IDENT(L).EQ.IZH) IM=L
  IF (IDENT(L).EQ.IZ1) INOP=L
  IF (IDENT(L).EQ.IZ2) IE4=L
  IF (IDENT(L).EQ.IZ3) IOX=L
  IF (IDENT(L).EQ.IZ4) IFM=L
  IF (IDENT(L).EQ.IZ5) IN=L
10 CONTINUE
DO 2 I=1, IKINE
  READ (5,100) (IZD(J), J=1,8), IRR(I), IRT(I), (RC(I,K), KR=1,5)
  WRITE(6,101) I, (IZD(J), J=1,8), IRR(I), IRT(I), (RC(L,K), KR=1,5)
DO 5 J=1,8
  IRR(I,J)=NS1+1
DO 3 L=1, NS1
  IF (IZD(J).EQ.IDENT(L)) 3,4,3
4 IRR(I,J)=L
3 CONTINUE
: CONTINUE
2 CONTINUE
KC(I,1)= AV*RC(I,1)
1: FORMAT(4,1X), 2I2, E9.2, F5.2, E8.1, 2E7.1)
101 FORMAT(2X, I2, 2X, 3(1X, A4, 1X, E9.2), 1X, A4, 1X, E8.1, 3(1X, A4, 1X, E7.1))
102 FORMAT (4, 6X, 7E13.3)
103 FORMAT (1X, A4, 6X, 7E13.3)
RETURN
END

```

*DECK.HKATE

```

SUBROUTINE HKATE
COMMON CPA(40),CPV(40),WA(40),RMOE(40),PSIV(51),ASE(40),UE(40),
1 S(4),TB(4),PTB(4.)
2 ,RUE(40),RUZE(40),RUHSE(40),Y(40),RUFIE(20)
3 ,SO,DS,SMAX,RMO,FY,U,T,HS,AID ,PARR,DSP,CTE,RR1,R32,AV,C,AI,VE
4 ,MH,SE,AE,RPO,TP,R1OP,EP,DEP,DAE,DTIS,TS,DRP,DRHJP,DSSE,DYE,DT,DHS
5 ,OU,DUP,DFY,DRHO,D>,DA,ADS,ARH02,UM,SI,SKO,BETA
6 ,RUAI,RUEI,RUF1E1,RUF2E1,RUF3E1,RUF4E1,RUF5E1,P+ADS
COMMON RUF4E(20),R32(20),
1 RUF2E(20),RUF3E(20),FQI, FE(8,20),PE(20),TE(20)
2 , F(51),DF(51),D32I,D32II,XMLI,XMDU,ETA,ALD,A-I,JE,VLI,THETA
3 ,C7,CA ,CP,A,RP ,XMA,XMAP,SS,UR,DUR,MSR,XMLD
4 ,RUF6E(20),RUF7E(20),RUF8E(20)
5 ,FFF,FFEV ,CF1,CF2,CF3,CF5,CF7,XNF1,XNF2,XNF5,8,37,FF3,FF4,XNRT
6 ,XBK,YBK,TBK,PEBK,RJ2BK,RHOXY,SOP,NSO,NMGR,DEQ,S153
COMMON STB(30),TTB(30),RHOXB(30),UTB(30),FTB(8,30)
COMMON /HEAT/ YL(6,51),HI(51)
COMMON/LIQUID/ CPL,TBP,DHV,RMOL,MV,IBPO,SIG,TLD,JH.,JO,WECR,
1 FFVI,AK,BK,WE,ITBK,FFV,ITBP
COMMON/INTES/NPS,NPT,IOU(20),NSL,IQPT,KO,NS,NS1,IBJ5,IO,IKIN
COMMON/CHEM2/CH(51,51),FI(51),WP(51),MH(51),WDOT(51),QX(51)
1 ,RC(100,2)
COMMON /SPID/ IZH,I21,I22,I23,I24,I25, IM,INOP,IE4,IOX,IFM,IN
COMMON/CHEM1/IRR(100),IRI(100),IRRR(100,0),IZD(8),I3:NT(51)
F1=FI(INOP)
F2=FI(IEM)
F3=FI(IOX)
F4=FI(IFM)
F5=FI(IN)
RKI(3)=0.0
RKI(4)=0.0
IF(RP .LT. 1.0E-9) GO TO 25
ULI= SQRT(UP*VP*VP)
XMOU=0.5*XMLJ*RP*RP/(ULI*A*RPO**3)
TI50= T /500.
VPLUS= 1.0E5* SQRT(TI50)
JEMIUS= 4.4E7* SQRT(TI50)
VMIUS= 2.36E5* SQRT(TI50)
DNM= (VEMIUS*RHO *F2 + (VMIUS+U -ULI)*RHO *F4 )
CC= (VPLUS+U -ULI)*RHO *F1 /DNM
CALL ETAGT(ETA,CC,SI)
RKS = 1.5*(1.0+ETA)*XMOU *(VPLUS+U -ULI) / RHO.
RKI(3)=FF3*RKS *VEMIUS*RHO *F2 /DNM
RKI(4)=FF4*RKS *(VMIUS+U -ULI)*RHO *F4 /DNM
25 CONTINUE
M3=RKI(3)*RHO*FI(INJP)
M4=RKI(4)*RHO*FI(INJP)
QX(INOP)=-M3-M4
QX(IEM)=-M3
QX(IFM)=-M4
QX(IOX)=M3+M4
QX(IN)=-M3

```

SUBROUTINE HRAFE TRACE

RETURN
END

CDC 6500 FTN VJ.J-308+ OPT=0 03/22/74 18.41.12.

PAGE 2

TRACE

```

*DECK,ETAGT
SUBROUTINE ETAGT(ETA,CC,SI)
IF ((CC.LT.1.0) .AND. (CC.GT.C.0)) GO TO 4
5 CONTINUE
WRITE(6,1...) CC,SI
STOP
100 FORMAT(1H ,*----- CC=, 5E10.3)
4 CONTINUE
EPI=.01
DETA=0.5*ETA
ETA=1.0E-5-DETA
DELI=1.0
1 CONTINUE
ETA= ETA +DETA
XLMS= EXP(-ETA)
XRMS= CC*(1.0+ETA)
DEL= XLMS-XRMS
EPS= ABS(XLMS*EPI)
IF (ABS(DEL).LE.EPS) GO TO 3
IF ((DEL*DEL).GE.0.0) GO TO 2
ETA= ETA-DETA
50 TO 1
2 DELI= DEL
50 TO 1
3 CONTINUE
RETURN
END

```

TRACE

```

*DECK,CHEMF
SUBROUTINE CHEMF
COMMON CPA(40),CPV(40),WA(40),RHOE(40),PSIV(51),VSE(40),UE(40),
1 S(40),TB(40),PTB(40)
2 RUE(40),RUZE(40),RUHSE(40),Y(40),RUFEE(20)
3 SD,DS,SMAX,RHD,FV,U,T,MS,ATD ,P,PK,OSP,CTE,RMI,RR2,AV,C,AI,VE
4 WM,SE,AE,RPO,TP,R1OP,EP,DE,DAE,UTS,TS,DRP,DRHCP,ISE,DVE,DT,DMS
5 DU,DUP,DFV,DRHD,DS,DA,ADS,ARMH2,UM,SI,SKO,BETA
6 KUAI,RUEI,RUFIEI,RUFZEI,RUF3EI,RUF4EI,RUF5EI,PR435
COMMON RUF4E(20),D32(20),
1 RUF2E(20),RUF3E(20),FQI, FE(8,20),PE(20),TE(20)
2 F(51),DF(51),D32I,XCL1,X4DU,ETA,ALO,A,T,J,VLI,THETA
3 F1,F2,F3,F4,F5,PI,A11,B8,X,F8,CF9,UP,VP,GAMA,C1,C2,C3,C4,C5,C6,
4 C7,C8 ,CP,A,RP ,XMA,XMAP,S5,UR,DUR,MSK,DMSR,XMLO
5 RUF6EI,RUF7EI,RUF8EI,F6,F7,F8 ,F9 ,FFPE,FFTE,FFAE
7 ,FFF,FFEV ,CF1,CF2,CF3,CF4,CF5,CF6,CF7,XMF1,XMF2,XNF5,B5,B3,FF3,FF4,XNRT
8 ,XBK,YBK,TBK,PEBK,XL2,X4,RHOXY,SOP,NSO,MHCR,DECP,S1,S2
COMMON STB(30),TTB(30),RHO7B(30),UTB(30),FTB(8,30)
COMMON/HEAT/ XL(6,51),MI(51)
COMMON/LIQUID/ CPL,TBP,DMV,R40L,MV,TBPO,SIG,TL0,JM.,J0,MECR,
1 FFI,AK,BK,ME,ITBK,FFV,TBPY
COMMON/INTEG/NPS,NPI,DDO(20),NSL,IOPT,KO,NS,NSI,IBJ,IO,IKINE
COMMON/CHEM2/CH(51,51),FI(51),WP(51),MM1(51),MDOF(51),OX(51)
2 ,KC(100,5)
COMMON /SPID/ IZH,I21,I22,I23,I24,I25,
COMMON/CHEM1/IRR(10),IRI(100),IARR(11,8),IZD(8),IDCNT(51)
DIMENSION RQ(80),RM(80)
KRT=1.987*T
5 G45 CONSTANT IN 82.56 CH3-ATM/G-MOLE-K
KRT1=82.06*T
XLOGT=ALOG(T)
DO 3 IR=1,NS1
OX(IR)=0.0
WP(IR)=0.0
WH(IR)=3.0
MDOF(IR)=J..
DO 4 JR=1,NS1
4 CH(IR,JR)=0.0
3 CONTINUE
CALL HRATE
DO 1 IR=1,IKINE
J1=NS1+1
J2=NS1+1
J3=NS1+1
J4=NS1+1
J5=NS1+1
J6=NS1+1
J7=NS1+1
J8=NS1+1
KIRI=IRT(IR)
50 TO (841,842,843,844,845,846,847,848,849),KIRT
841 KKF=RC(IK,1)
50 TO 850
842 KKF=RC(IR,1)/T

```

```

60  GO TO 85J
    8*3 KKF=RC(IR,1)/T/T
    GO TO 85.
    8*4 KKF=RC(IR,1)/ SQRT(T)
    GO TO 85D
    8*5 KKF=RC(IR,1)* EXP(RC(IR,3)/RRT)
    GO TO 85J
    8*6 KKF=KC(IR,1)* EXP(RC(IR,3)/RRT)/T
    GO TO 85J
    8*7 KKF=RC(IR,1)/T / SQRT(T)
    GO TO 85J
    8*8 KKF=RC(IR,1)*(T **RC(IR,2))* EXP(RC(IR,3)/T )
    GO TO 85J
    8*9 KK=RC(IR,4)
    KKF=RC(IR,1)*(T**RC(IR,2))*(KK/T+1.0)* EXP(KC(IR,3)/T)
    GO TO 85.
75  85C CONTINUE
    KIRK=IRR(IK)
    GO TO (851,852,853,854,855,856,857,858,859,860,851),<IRR
    851 J1=IRRR(IR,1)
    J2=IRRR(IR,2)
    J5=IRRR(IR,5)
    J6=IRRR(IR,6)
    CRR=KKF*(RHO**3)/MM*AV
    KQ(IR)=CRR*FI(J1)*FI(J2)
    KM(IR)=6.0
    DO 778 J=1,6
    SIGN=1.
    IF (J.GT. 4) SIGN=-1.
    IROW=IRRR(IR,J)
    CM(IROW,J1)=CM(IROW,J1)+SIGN*CRR*FI(J2)
    CM(IROW,J2)=CM(IROW,J2)+SIGN*CRR*FI(J1)
    QX(IROW)=QX(IROW)+SIGN*Q(IR)
778 CONTINUE
    GO TO 86C
    852 J1=IRRR(IR,1)
    J2=IRRR(IR,2)
    J5=IRRR(IR,5)
    CRR=KKF*(RHO**3)/MM*AV
    FQ(IR)=CRR
    KM(IR)=6.0
    DO 779 J=1,5
    SIGN=1.
    IF (J.GT.4) SIGN=-1.
    IROW=IRRR(IR,J)
    CM(IROW,J1)= CM(IROW,J1)+SIGN*CRR
    CM(IROW,J2)= CM(IROW,J2)+SIGN*CRR
    QX(IROW)=QX(IROW)+SIGN*Q(IR)
779 CONTINUE
    GO TO 86J
1.5  853 J1=IRRR(IR,1)
    J2=IRRR(IR,2)
    J5=IRRR(IR,5)
    J6=IRRR(IR,6)
    J7=IRRR(IR,7)

```

```

115      CKK=RKF*RHO*RHO
      KQ(IR)=CRR*FI(J1)*FI(J2)
      KM(IR)=0.0
      DO 780 J=1,7
      SIGN=1.0
      IF (J.GT.4) SIGN=-1.0
      IKOM=IRRR(IR,J)
      CM(IROM,J1)=CM(IROM,J1)+SIGN*CRR*FI(J2)
      CM(IROM,J2)=CM(IROM,J2)+SIGN*CRR*FI(J1)
      QX(IROM)=QX(IROM)+SIGN*Q(IR)
120      /dL CONTINUE
      GO TO 880
125      J1=IRRR(IR,1)
      J2=IRRR(IR,2)
      J5=IRRR(IR,5)
      CRR=RKF*RHO*RHO
      KQ(IR)=CRR*FI(J1)*FI(J2)
      KM(IR)=0.0
      DO 776 J=1,6
      SIGN=1.0
      IF (J.GT.4) SIGN=-1.0
      IROM = IRRR(IR,J)
      CM(IROM,J1)=CM(IROM,J1)+SIGN*CRR*FI(J2)
      CM(IROM,J2)=CM(IROM,J2)+SIGN*CRR*FI(J1)
      QX(IROM)=QX(IROM)+SIGN*Q(IR)
130      /dL CONTINUE
      GO TO 88
135      J1=IRRR(IR,1)
      J5=IRRR(IR,5)
      J6=IRRR(IR,6)
      CKK=RKF*RHO*RHO/MM
      KQ(IR)=CRR*FI(J1)
      KM(IR)=0.0
      DO 777 J=1,6
      SIGN=1.0
      IF (J.GT.4) SIGN=-1.0
      IROM = IRRR(IR,J)
      CM(IROM,J1)=CM(IROM,J1)+SIGN*CRR*FI(J2)
      CM(IROM,J2)=CM(IROM,J2)+SIGN*CRR*FI(J1)
      QX(IROM)=QX(IROM)+SIGN*Q(IR)
140      /dL CONTINUE
      GO TO 880
145      J1=IRRR(IR,1)
      J2=IRRR(IR,2)
      J3=IRRR(IR,3)
      J5=IRRR(IR,5)
      CKK=RKF*(RHO**4)*AV*AV
      KQ(IR)=CRR*FI(J1)*FI(J2)*FI(J3)/MM
      KM(IR)=0.0
      DO 781 J=1,5
      SIGN=1.0
      IF (J.GT.4) SIGN=-1.0
      IROM=IRRR(IR,J)
      CM(IROM,J1)=CM(IROM,J1)+SIGN*CRR*FI(J2)*FI(J3)/MM
      CM(IROM,J2)=CM(IROM,J2)+SIGN*CRR*FI(J1)*FI(J3)/MM
      CM(IROM,J3)=CM(IROM,J3)+SIGN*CRR*FI(J1)*FI(J2)/MM
      QX(IROM)=QX(IROM)+SIGN*2.0*Q(IR)
150      /dL CONTINUE

```

SUBROUTINE CHEF TRACE

781 CONTINUE

GO TO 880

857 J1=IRRR(IR,1)

J5=IRRR(IR,5)

J6=IRRR(IR,6)

J7=IRRR(IR,7)

CKR=RKF*(RHO**4)*AV

KQ(IR)=CRZ*FI(J1)/MM

KM(IR)=0.0

DO 782 J=1,7

SIGN=1.0

IF (J.GT. 4) SIGN=-1.0

IRUM=IRRR(IR,J)

CM(IROM,J1)=CM(IROM,J1)+SIGN*CKR

782 CONTINUE

GO TO 880

858 J1=IRRR(IR,1)

J2=IRRR(IR,2)

J3=IRRR(IR,3)

J5=IRRR(IR,5)

J6=IRRR(IR,6)

CKR=RKF*(RHO**3)*AV

KQ(IR)=CKR*FI(J1)*FI(J2)*FI(J3)

KM(IR)=0.0

DO 783 J=1,6

SIGN=1.0

IF (J.GT. 4) SIGN=-1.0

IRUM=IRRR(IR,J)

CM(IROM,J1)=CM(IROM,J1)+SIGN*CKR*FI(J2)*FI(J3)

CM(IROM,J2)=CM(IROM,J2)+SIGN*CKR*FI(J1)*FI(J3)

CM(IROM,J3)=CM(IROM,J3)+SIGN*CKR*FI(J1)*FI(J2)

QX(IROM)=QX(IROM)+SIGN*CKR*QX(I,R)

783 CONTINUE

GO TO 880

859 J1=IRRR(IR,1)

J2=IRRR(IR,2)

J3=IRRR(IR,3)

J4=IRRR(IR,4)

J5=IRRR(IR,5)

J6=IRRR(IR,6)

CKR=RKF*(RHO**4)*AV*AV

RQ(IR)=CKR*FI(J1)*FI(J2)*FI(J3)*FI(J4)

KM(IR)=J.0

DO 784 J=1,6

SIGN=1.0

IF (J.GT.4) SIGN=-1.0

IRUM=IRRR(IR,J)

CM(IROM,J1)=CM(IROM,J1)+SIGN*CKR*FI(J2)*FI(J3)*FI(J4)

CM(IROM,J2)=CM(IROM,J2)+SIGN*CKR*FI(J1)*FI(J3)*FI(J4)

CM(IROM,J3)=CM(IROM,J3)+SIGN*CKR*FI(J1)*FI(J2)*FI(J4)

CM(IROM,J4)=CM(IROM,J4)+SIGN*CKR*FI(J1)*FI(J2)*FI(J3)

QX(IROM)=QX(IROM)+SIGN*CKR*QX(I,R)

784 CONTINUE

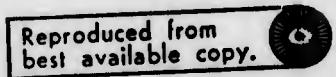
GO TO 880

860 J1=IRRR(IR,1)

```

J2=4*KK(I2,2)
J5=I*KK(I2,5)
J6=I*KK(I2,6)
J7=I*KK(I2,7)
J8=I*KK(I2,8)
J9=I*KK(I2,9)
J10=I*KK(I2,10)
J11=I*KK(I2,11)
J12=I*KK(I2,12)
J13=I*KK(I2,13)
J14=I*KK(I2,14)
J15=I*KK(I2,15)
J16=I*KK(I2,16)
J17=I*KK(I2,17)
J18=I*KK(I2,18)
J19=I*KK(I2,19)
J20=I*KK(I2,20)
J21=I*KK(I2,21)
J22=I*KK(I2,22)
J23=I*KK(I2,23)
J24=I*KK(I2,24)
J25=I*KK(I2,25)
J26=I*KK(I2,26)
J27=I*KK(I2,27)
J28=I*KK(I2,28)
J29=I*KK(I2,29)
J30=I*KK(I2,30)
J31=I*KK(I2,31)
J32=I*KK(I2,32)
J33=I*KK(I2,33)
J34=I*KK(I2,34)
J35=I*KK(I2,35)
J36=I*KK(I2,36)
J37=I*KK(I2,37)
J38=I*KK(I2,38)
J39=I*KK(I2,39)
J40=I*KK(I2,40)
J41=I*KK(I2,41)
J42=I*KK(I2,42)
J43=I*KK(I2,43)
J44=I*KK(I2,44)
J45=I*KK(I2,45)
J46=I*KK(I2,46)
J47=I*KK(I2,47)
J48=I*KK(I2,48)
J49=I*KK(I2,49)
J50=I*KK(I2,50)
J51=I*KK(I2,51)
J52=I*KK(I2,52)
J53=I*KK(I2,53)
J54=I*KK(I2,54)
J55=I*KK(I2,55)
J56=I*KK(I2,56)
J57=I*KK(I2,57)
J58=I*KK(I2,58)
J59=I*KK(I2,59)
J60=I*KK(I2,60)
J61=I*KK(I2,61)
J62=I*KK(I2,62)
J63=I*KK(I2,63)
J64=I*KK(I2,64)
J65=I*KK(I2,65)
J66=I*KK(I2,66)
J67=I*KK(I2,67)
J68=I*KK(I2,68)
J69=I*KK(I2,69)
J70=I*KK(I2,70)
J71=I*KK(I2,71)
J72=I*KK(I2,72)
J73=I*KK(I2,73)
J74=I*KK(I2,74)
J75=I*KK(I2,75)
J76=I*KK(I2,76)
J77=I*KK(I2,77)
J78=I*KK(I2,78)
J79=I*KK(I2,79)
J80=I*KK(I2,80)
J81=I*KK(I2,81)
J82=I*KK(I2,82)
J83=I*KK(I2,83)
J84=I*KK(I2,84)
J85=I*KK(I2,85)
J86=I*KK(I2,86)
J87=I*KK(I2,87)
J88=I*KK(I2,88)
J89=I*KK(I2,89)
J90=I*KK(I2,90)
J91=I*KK(I2,91)
J92=I*KK(I2,92)
J93=I*KK(I2,93)
J94=I*KK(I2,94)
J95=I*KK(I2,95)
J96=I*KK(I2,96)
J97=I*KK(I2,97)
J98=I*KK(I2,98)
J99=I*KK(I2,99)
J100=I*KK(I2,100)

```



```

2.0 QX(2)=QX(2)*ADSM+RUAI*((RUF2E1-RUEI*FI(2))*DAE-FI(2)*PHADS)
      QX(3)=QX(3)*ADSM+RUAI*((RUF3E1-RUEI*FI(3))*DAE-FI(3)*PHADS)
      QX(4)=QX(4)*ADSM+RUAI*((RUF4E1-RUEI*FI(4))*DAE-FI(4)*PHADS)
      QX(5)=QX(5)*ADSM+RUAI*((RUF5E1-RUEI*FI(5))*DAE-FI(5)*PHADS)
      QX(6)=QX(6)*ADSM+RUAI*((RUF6E1-RUEI*FI(6))*DAE-FI(6)*PHADS)
      QX(7)=QX(7)*ADSM+RUAI*((RUF7E1-RUEI*FI(7))*DAE-FI(7)*PHADS)
      QX(8)=QX(8)*ADSM+RUAI*((RUF8E1-RUEI*FI(8))*DAE-FI(8)*PHADS)
      DO 2 I=1,NS
      QX(I)=QX(I)+F(I)
      DO 2 J=1,MS
      CM(I,J)=CM(I,J)+ADSM
      IF(I.EQ.1) CM(I,J)=1.0+CM(I,J)
2 CONTINUE
      IF (QX(4) .GT. 0.0) GO TO 6
      IF (IBUG.EQ.0) GO TO 5
      DO 24 I=1,MS
      WRITE(6,100) (CM(I,J),J=1,NS),QX(I)
24 CONTINUE
      6 CONTINUE
      CALL SLOP(QX,CM,NS)
      IF (IBUG.EQ.0) GO TO 7
      DO 5 I=1,MS
      WRITE(6,100) (CM(I,J),J=1,NS),QX(I)
      5 CONTINUE
      7 CONTINUE
      DO 23 I=1,MS
      F(I)=QX(I)*A(I)
23 CONTINUE
      F(I) NJ4 IN MASS FRACTION UNIT
      1.1 FORMAT( 27, 1P1E12.3)
      RETURN
      END
3.5
    
```

TRACE

```

*DECK.SLDP
SUBROUTINE S-OP(X,A,N)
THIS PROGRAM FINDS THE SOLUTIONS TO A SET OF N SIMULTANEOUS LINEAR
EQUATIONS BY USING THE GAUSS-GORDAN REDUCTION ALGORITHM WITH THE
DIAGONAL PIVOT STRATEGY
DESIGNED PARTICULARLY FOR IMPLICIT CHEMISTRY ---CJK
DIMENSION A(51,51),X(51)
DO 9 K=1,N
IF(ABS(A(K,K)) .GT. 1.0-1) GO TO 5
WRITE (6,101) K,A(K,K)
DO 10 IR=1,N
WRITE(6,100)(A(IR,JR),JR=1,N),X(IR)
10 CONTINUE
100 FORMAT(1X, IP10E12.3)
101 FORMAT( + ERROR--- SMALL PIVOT *,15,1PE12.5)
STOP
5 KPI= K+1
DO 6 J= KPI, N
A(K,J)= A(K,J)/A(K,K)
X(K)= X(K)/A(K,K)
A(K,K)= 1.0
DO 9 I=1,N
IF (1.EQ.K .OR. A(I,K).EQ.0.) GO TO 9
DO 6 J=KPI,N
A(I,J)= A(I,J)- A(I,K)*A(K,J)
X(I)= X(I)- A(I,K)*X(K)
A(I,K)=0.
9 CONTINUE
99 CONTINUE
RETURN
END
    
```

5

10

15

20

25

30

SUBROUTINE BREAK TRACE

```

F(NS1)=0.0
---- F(I) ARE IN MOLE/CM. -----
UX=RUEX/RUEX
DO 44 I=1,NS
F(I)=F(I)*MI(I)/RUEX
PSIV(I)=F(I)
44 CONTINUE
---- F(I) IS NOW IN MASS FRACTIONS -----
MSX=MSX/RUEX
UX2=UX*UX
KHX=RUEX/UX
4X=MSX-UX2/(2.*CTE)
---- CALCULATING GAS PROPERTIES BEHIND JET SHOCK -----
RHOXY=0.25
RHOY=RHOX/RHOXY
5 CONTINUE
WRITE(6,121) PX,UX,TSX,RHOX,IX,MAX,(F(I),I=1,NS1)
DO 44 I=1,111
5Y=PX+RHOX*UX2*(1.0-RHOXY)
4Y=MSX-UX2*(RHOXY+RHOXY)/(2.*CTE)
TBPY=TBPO
TI=TX+PY/PX+RHOXY
CALL TINV(HY,TY,II,F,NS1,CPY)
RHO=PY*WAX/R22/TY
IF (IBUG.NE.0)
1WRITE (6,101) RHOX,KHOY,RHOXY,RHO,PY,MY,TY
IF ( ABS((RHO/RHOY)-1.0).LT. 0.001) GO TO 41
RHOY=0.5*(RHOY+RHO)
RHOXY=RHOX/RHOY
40 CONTINUE
IF (IBUG.NE.0) WRITE(6,128) RHOX,RHOY,RHOXY,RHO,PY,4Y
128 FORMAT (/2X,* --- SHOCK CALCULATION DIVERGED*/1P1E12.3/)
41 CONTINUE
UY=UX+RHOXY
KOHV=RR1/(OHV*WV)
TPOI=1./TBP0
PV=PY/ATO
TBPY=1.0/(TBP0I -R04V*ALOG(PV))
WRITE (6,122) PY,UY,MY,RHOY,TY,TBPY
---- COMPUTE FLASH VAPORIZATION AT SHOCKED PRESSURE -----
JM=1.21E-5*(TY**1.5)/(TY+110)
FFV=0.0
IF (TBPY .LT. T-0) FFV=CPL*(T0-TBPY)/DHV
IF (FFV.GE.FFVI) FFV=1.0
---- JET BREAK-UP -----
JLI=VLI* SIN(THETA)
WE=RHOY*UY*UY*J.5*DE/SIG
IF ((NE.LT.WEGR).AND.(ITBK.NE.1)) GO TO 31
KEJ=RHOY*UY*J.5*DE/JM
XXX=REJ/(1.0+2.3*WE/ SQRT(REJ))
XXX=SQRT(XXX)
IF (IBUG.NE.0)
1WRITE(6,101) RHOX,40Y,UM,UML,DE,UY,XXX,TY
T8K=0.536*(RHOX/RHOY)**(0.6556)*(UM/UGL)**(0.3333)*J.5*DE/UY*XXX
XXX1=RHOY*CD*(UY-UGL)**2/DE/RHOX

```

SUBROUTINE BREAK TRACE

```

115 JBK=2.0*XXX1*TBK+J.LI
    YBK=YP*TBK
    DBK=(XXX1*TBK+ULI)*TBK
    IF (DBK.LT.DE) DBK=DE
    XBK=DBK*SO
    ----- CALCULATING DRUPLET SIZE -----
    MEL=DE*UY*UY*RHOL/SIG
    GAMMA=CPY/(CPY-RR1/MAX)
    SS= SQR((GAMA*RR2*TY/MAX)
    XMA=UY/SS
    D3u=48.0*(MEL**(-0.375))*(REJ/XMA)**(-0.25)*DE
    D32I=1.29*03
    DMAX=030*4.6*(030/DE)**(0.15)
    READ (5,*) FIH
    GO TO 62
125 ***** VAPOR PRESSURE BREAK UP
    61 KEPS=0.1
    ISH=IBPY/(1.-XEPS)
    ***** TSH= SMATTER TEMPERATURE
    IF (TLO.GT.TS4) TLO=IBPY
    XXX1= (UY-VLI* SIN(THETA))**2 +YP**2
    UR= SQR(TAXX1)
    MSRY=HY*UR*UR/(2.0*STE)
    II=IY
    CALL TINV(MSRY,TSY,I,F,NS1,CPSY)
    XX2= ALD*UR*(TSY-TLJ)
    V= SQR(3.0*CPY/RHOY)
    XX3=1.0+ALD*V*(TY-TLJ)/XX2
    QP=0.5*XX2*PY*XX3/TY/CTE*FFEJ
    TBK=CPL*(TSH-TLO)*R40L*DE/(1.0*03P)
    XXX1=RH0Y*CD*(UY-ULI)**2/DE/XMOL
    JBK=2.0*XXX1*TBK+ULI
    YBK=YP*TBK
    IF (YBK.GT.YMAX) YBK=YMAX
    XBK=(XXX1*TBK+ULI)*TBK*SO
    DMAX=U.L
    MEL=...
    D3C=0.0
    D32I=(AK*EXP((-BK)*TLO))*00
130 CONTINUE
    WRITE (6,123) ME,REJ,TBK,UBK,XBK,YBK,MEL,030,D32I,JMAX
    IF ((ITBK.EQ.1).OR.(ITBK.EQ.3)) GO TO 45
    ITBK=ITBK+1
    ***** MODIFYING STACK FRONT CONDITIONS *****
    DO 46 ID=2,NSL
    IF (YBK.GT.Y(ID-1).AND. YBK.LT.Y(ID)) GO TO 47
    46 CONTINUE
    ID=NSL
    +7 CONTINUE
    PT80(1)=PT80(2)
    IT80(1)=IT80(2)
    JT80(1)=J80
    F1T80(1)=F1T80(2)
    F2T80(1)=F2T80(2)
135

```

SUBROUTINE BREAK TRACE

```

F3TB0(1)=F3TB0(2)
F4TB0(1)=F4TB0(2)
F5TB0(1)=F5TB0(2)
F6TB0(1)=F6TB0(2)
F7TB0(1)=F7TB0(2)
F8TB0(1)=F8TB0(2)
CALL LIPLN(YBK,PTB0,Y0,PX,1D)
CALL LIPLN(YBK,TTB0,Y0,TX,1D)
CALL LIPLN(YBK,UTB0,Y0,UX,1D)
CALL LIPLN(YBK,FTB0,Y0,F(1),1D)
CALL LIPLN(YBK,F2TB0,Y0,F(2),1D)
CALL LIPLN(YBK,F3TB0,Y0,F(3),1D)
CALL LIPLN(YBK,F4TB0,Y0,F(4),1D)
CALL LIPLN(YBK,F5TB0,Y0,F(5),1D)
CALL LIPLN(YBK,F6TB0,Y0,F(6),1D)
CALL LIPLN(YBK,F7TB0,Y0,F(7),1D)
CALL LIPLN(YBK,F8TB0,Y0,F(8),1D)

```

***** F(I) ARE IN HOLE/GH

```

MAX=0.0
DO 48 II=1,NS
  MAX=MAX+F(II)
+6 CONTINUE

```

```

MAX=1.0/MAX
RHOX=PX*MAX/(RR2*TX)

```

```

WA(1)=MAX
PTB(1)=PX
ITB(1)=TX
RMOTB(1)=RHOX
FTB(1,1)=F(1)
FTB(2,1)=F(2)
FTB(3,1)=F(3)
FTB(4,1)=F(4)
FTB(5,1)=F(5)
FTB(6,1)=F(6)
FTB(7,1)=F(7)
FTB(8,1)=F(8)
DO 31 JJ=1,6
  UD(JJ)=0.0
DO 32 II=1,NS
  DD(JJ)=UD(JJ)+ XL(JJ,II)*FTB(II,1)

```

```

32 CONTINUE
31 CONTINUE

```

```

T1=TTB(1)/14.7
M2=1000.*(-DD(5)/T13+DD(5)+T10*(DD(1)+T10*(DD(2)+T10*(DD(3)/3.
  1.+T10*DD(4)/4.0)))
HE=HX
MSX=HX +C.5*UTB(1)*UTB(1)/CTE
MSE(1)=MSX
XXX1=RMOTB(1)*UTB(1)
RUE(1)=XXX1
RUSE(1)=XXX1*UTB(1)
RUMSE(1)=XXX1*MSE(1)
RUFIE(1)=XXX1*FTB(1,1)
RUFZE(1)=XXX1*FTB(2,1)

```

1 0

175

1 0

1 05

1 0

1 05

2 0

2 5

210

215

220

```

199 3015(1)=XXX1*PIB(3,2)
    3015(2)=XXX1*PIB(3,2)
    3015(3)=XXX1*PIB(3,2)
    3015(4)=XXX1*PIB(3,2)
    3015(5)=XXX1*PIB(3,2)
    3015(6)=XXX1*PIB(3,2)
    3015(7)=XXX1*PIB(3,2)
    3015(8)=XXX1*PIB(3,2)
    3015(9)=XXX1*PIB(3,2)
    3015(10)=XXX1*PIB(3,2)
    3015(11)=XXX1*PIB(3,2)
    3015(12)=XXX1*PIB(3,2)
    3015(13)=XXX1*PIB(3,2)
    3015(14)=XXX1*PIB(3,2)
    3015(15)=XXX1*PIB(3,2)
    3015(16)=XXX1*PIB(3,2)
    3015(17)=XXX1*PIB(3,2)
    3015(18)=XXX1*PIB(3,2)
    3015(19)=XXX1*PIB(3,2)
    3015(20)=XXX1*PIB(3,2)
    3015(21)=XXX1*PIB(3,2)
    3015(22)=XXX1*PIB(3,2)
    3015(23)=XXX1*PIB(3,2)
    3015(24)=XXX1*PIB(3,2)
    3015(25)=XXX1*PIB(3,2)
    3015(26)=XXX1*PIB(3,2)
    3015(27)=XXX1*PIB(3,2)
    3015(28)=XXX1*PIB(3,2)
    3015(29)=XXX1*PIB(3,2)
    3015(30)=XXX1*PIB(3,2)
    3015(31)=XXX1*PIB(3,2)
    3015(32)=XXX1*PIB(3,2)
    3015(33)=XXX1*PIB(3,2)
    3015(34)=XXX1*PIB(3,2)
    3015(35)=XXX1*PIB(3,2)
    3015(36)=XXX1*PIB(3,2)
    3015(37)=XXX1*PIB(3,2)
    3015(38)=XXX1*PIB(3,2)
    3015(39)=XXX1*PIB(3,2)
    3015(40)=XXX1*PIB(3,2)
    3015(41)=XXX1*PIB(3,2)
    3015(42)=XXX1*PIB(3,2)
    3015(43)=XXX1*PIB(3,2)
    3015(44)=XXX1*PIB(3,2)
    3015(45)=XXX1*PIB(3,2)
    3015(46)=XXX1*PIB(3,2)
    3015(47)=XXX1*PIB(3,2)
    3015(48)=XXX1*PIB(3,2)
    3015(49)=XXX1*PIB(3,2)
    3015(50)=XXX1*PIB(3,2)
    3015(51)=XXX1*PIB(3,2)
    3015(52)=XXX1*PIB(3,2)
    3015(53)=XXX1*PIB(3,2)
    3015(54)=XXX1*PIB(3,2)
    3015(55)=XXX1*PIB(3,2)
    3015(56)=XXX1*PIB(3,2)
    3015(57)=XXX1*PIB(3,2)
    3015(58)=XXX1*PIB(3,2)
    3015(59)=XXX1*PIB(3,2)
    3015(60)=XXX1*PIB(3,2)
    3015(61)=XXX1*PIB(3,2)
    3015(62)=XXX1*PIB(3,2)
    3015(63)=XXX1*PIB(3,2)
    3015(64)=XXX1*PIB(3,2)
    3015(65)=XXX1*PIB(3,2)
    3015(66)=XXX1*PIB(3,2)
    3015(67)=XXX1*PIB(3,2)
    3015(68)=XXX1*PIB(3,2)
    3015(69)=XXX1*PIB(3,2)
    3015(70)=XXX1*PIB(3,2)
    3015(71)=XXX1*PIB(3,2)
    3015(72)=XXX1*PIB(3,2)
    3015(73)=XXX1*PIB(3,2)
    3015(74)=XXX1*PIB(3,2)
    3015(75)=XXX1*PIB(3,2)
    3015(76)=XXX1*PIB(3,2)
    3015(77)=XXX1*PIB(3,2)
    3015(78)=XXX1*PIB(3,2)
    3015(79)=XXX1*PIB(3,2)
    3015(80)=XXX1*PIB(3,2)
    3015(81)=XXX1*PIB(3,2)
    3015(82)=XXX1*PIB(3,2)
    3015(83)=XXX1*PIB(3,2)
    3015(84)=XXX1*PIB(3,2)
    3015(85)=XXX1*PIB(3,2)
    3015(86)=XXX1*PIB(3,2)
    3015(87)=XXX1*PIB(3,2)
    3015(88)=XXX1*PIB(3,2)
    3015(89)=XXX1*PIB(3,2)
    3015(90)=XXX1*PIB(3,2)
    3015(91)=XXX1*PIB(3,2)
    3015(92)=XXX1*PIB(3,2)
    3015(93)=XXX1*PIB(3,2)
    3015(94)=XXX1*PIB(3,2)
    3015(95)=XXX1*PIB(3,2)
    3015(96)=XXX1*PIB(3,2)
    3015(97)=XXX1*PIB(3,2)
    3015(98)=XXX1*PIB(3,2)
    3015(99)=XXX1*PIB(3,2)
    3015(100)=XXX1*PIB(3,2)

```

```

2.0 24 I=1,NS
2.1 24 F(I)=PSIV(I)*(1.-F(NS1))
2.2 ***** NORMALIZED MASS FRACTION
2.3 24 I=1,NS1
2.4 24 F(I) ARE IN MASS FRACTION
2.5 24 F(I)=F(I)/MI(I)
2.6 ***** INITIAL OVA TUBE AREA AFTER MIXING
2.7 24 UA=1.0/MM
2.8 24 MS=(U*U)/(2.*CTE)
2.9 24 I=TX
2.10 24 CALL TINV(M,T,I,F,NS1,CP)
2.11 24 FMO=FM/KR2/T
2.12 ***** INITIAL OVA TUBE AREA AFTER MIXING
2.13 24 UA=1.0/MM
2.14 24 MS=(U*U)/(2.*CTE)
2.15 24 I=TX
2.16 24 CALL TINV(M,T,I,F,NS1,CP)
2.17 24 FMO=FM/KR2/T
2.18 ***** INITIAL OVA TUBE AREA AFTER MIXING
2.19 24 UA=1.0/MM
2.20 24 MS=(U*U)/(2.*CTE)
2.21 24 I=TX
2.22 24 CALL TINV(M,T,I,F,NS1,CP)
2.23 24 FMO=FM/KR2/T
2.24 ***** INITIAL OVA TUBE AREA AFTER MIXING
2.25 24 UA=1.0/MM
2.26 24 MS=(U*U)/(2.*CTE)
2.27 24 I=TX
2.28 24 CALL TINV(M,T,I,F,NS1,CP)
2.29 24 FMO=FM/KR2/T
2.30 ***** CALCULATING INITIAL RELATIVE STAGNATION TEMPERATURE
2.31 24 UR = SQRT(AXX1)
2.32 24 HSK=MS-(U*U-UR*UR)/(2.*CTE)
2.33 24 II=TY
2.34 24 CALL TINV(HSR,TS,TI,F,NS1,CP,SR)
2.35 24 WRITE(6,125) SI,U,HSV,HSR,M,TS,T,RHO,A
2.36 24 WRITE(6,135)
2.37 24 DO 72 I=1,NS1
2.38 24 WRITE(6,119) IDENT(I),F(I),FI(I),WGT(I)
2.39 72 CONTINUE
2.40 1.1 FORMAT(8E1.3)
2.41 1.1 FORMAT(2X,10E12.3)
2.42 1.2 FORMAT(2X,15,1P10E12.4)
2.43 1.3 FORMAT(1H,15,1P9E12.3/5X,6E12.3)
2.44 1.5 FORMAT(2X,*,MASS FRACTIONS OF SPECIES*/1X,*,NO=*/1P1.3,1X,*,E==
2.45 1,*,E10.3,1X,*,0=*,E10.3,1X,*,F==*,E10.3,1X,*,V=*,E10.3,1X,*,O2=*,E10.3,1
2.46 2X,*,N2=*,E10.3,1X,*,N=*,E10.3,*,FV=*,E10.3)
2.47 1.7 FORMAT(/ 5X,*,I*,*,OX,*,WA,*,10X,*,HE,*,1X,*,HSE*,
2.48 1 9X,*,REUE*, 7X,*,REUE2*,7X,*,REUEHSE*,7X,*,RUF1E*,7X,*,RUF2E*,7X,*,RUF3
2.49 2E*9X,*,RUF4E*,7X,*,RUF5E*,7X,*,RUF6E*,7X,*,RUF7E*,7X,*,RUF8E*)
2.50 1.8 FORMAT(/
2.51 1,7X,*,FE(I) (MOLE/GM)=*)
2.52 1.9 FORMAT(9X,*,A,*,3X,1P8E12.3)
2.53 1.20 FORMAT(1X,*,-----* MODIFIED FREE STREAM CONDITIONS IN SHOCK FR
2.54 1.21 FORMAT(/2X,*,* CONDITIONS IN FRONT OF SHOCK */2X,*,*,*,1P10.3,2X,
2.55 1 *UX=*,E10.3,2X,*,HXS=*,E10.3,2X,*,P+OX=*,E10.3,2X,*,PX=*,E10.3
2.56 2 *UX,*,MAX=*,E10.3/ 2X,
2.57 2 *MASS FRACTIONS F(I)=*,9E12.3/)
2.58 1.22 FORMAT(/2X,*,* CONDITIONS BEHIND SHOCK*/2X,*,*,*,1P10.3,2X,*,UY=*,

```

```

1 E10.3,2X,*MY=*E10.3,2X,*MOT=*E10.3,2X,*TY=*E10.3,2X,*TBPY=*E
2 E10.3/)
123 FORMAT(2X,* JET BREAK-UP INFORMATION */2X,*ME=*E10.3,2X,
1 *REJ=*E10.3,2X,*TBK=*E10.3,2X,*UBK=*E10.3,2X,*XK=*E10.3,2X,
2 *YBK=*E10.3,2X,*MEL=*E10.3,2X,*D3C=*E10.3,2X,*D42=*E10.3,2X,
3 *DHAX=*E10.3/)
124 FORMAT(2X,* INITIAL FLOW RATE(G/SEC)*/2X,*XMV=*E10.3,2X,*XML=*E
1 E10.3,2X,*XMA=*E10.3,2X,*XMH=*E10.3,2X,*AE=*E10.3/)
125 FORMAT(2X,*----- CONDITIONS OF THE INITIAL MIXTURE AT S=*E10.4
/2X,*J=*E10.3,2X,*MSJ=*E10.3,2X,*MSR=*E10.3,2X,*H=*,
1 E10.3,2X,*YSR=*E10.3,2X,*T=*E10.3,2X,*RMO=*E10.3,2X,*A=*E10.3)
127 FORMAT(2X,* -- INITIAL MIXING CALCULATION DOES NOT CONVERGE TO
1CORRECT SOLUTION*,I5/IPIE11.3)
135 RETURN
END
END
3 5

```

TRACE

```

*DECK,ERF
FUNCTION ERF(X)
----- THIS CALCULATES THE ERROR FUNCTION AND ITS COMPLEMENT -----
DATA EPSILN/1.E-17,N/1.07,CONST/1.12e379167955e7
5  KOMPL=+1
   GO TO 5
ENTRY ERF
  KOMPL=-1
10 Z=-X
   GO TO 20
15 Z= X
20 IF (Z-5.) 25,25,45
21 ZSQ=Z*Z
   TERM=CONST*Z*EXP(-ZSQ)
   ERF=TERM
   I=J
   R=1.
30 I=I+1
   K=K+2.0
   TERM=2.0*ZSQ*TERM/R
   ERF=ERF+TERM
   IF (I-10) 35,40,40
35 IF (TERM-EPSILN*ERF) 40,30,30
40 IF (ERF-1.0) 50,50,45
45 ERF=1.0
50 IF (X) 0.0,5,5,5
55 ERF=0.0
   GO TO 65
60 ERF=-ERF
65 IF (KOMPL) 7.,99,99
----- COMPUTE COMPLEMENTARY ERROR FUNCTION ERF -----
70 ERF=1.0-ERF
95 RETURN
END

```

TRACE

CDC 660J FTM V2.1-308+ OPT=0 03/22/74 18.41.12.

PAGE

1

```
*DECK.LIPLM
SUBROUTINE LIPLM (SX,A,S,RA,K)
C 4 SCHEME FOR LINEAR INTERPOLATION
DIMENSION A(20),S(20)
22 RA= (A(K-1)*(S(K)-SX) + A(K)*(SX-S(K-1)))/(S(K)-S(K-1))
RETURN
END
```

5

```

*DECK,INV
SUBROUTINE TIMV (M,T,TF,NS1,CP)
----- THIS SUBPROGRAM WILL INVERT TEMPERATURE FROM ENTHALPY BY THE
NEWTON METHOD -----
5 DIMENSION DD(5),F(51)
COMMON /HEAT/ XL(5,51),MI(51)
COMMON/LIQUID/ CPL,TBP,DH,RHOL,MV,TBPU,SLG,TLQ,JM,JQ,WECR,
1 FFL,AK,BK,WE,IYK,FFV,TBP
10 CONTINUE
DO 10 J=1,6
DD(J)=0.0
DO 11 I=1,NS1
F(I) ARE IN MASS FRACTION
J0(J)=DD(J)+XL(J,1)*F(I)/MI(L)
11 CONTINUE
1 CONTINUE
EPS=1.0E-6*M
I=1
DO 12 IT=1,100
T1=TF
Z=1000.*(-DD(5)/T1+DD(5)+T1.*(DD(1)+T1*(DD(2)/2.+.110*(DD(3)/
+ 3.CAT10*DD(4)/4.0)))**
ZP= (DD(5)/(T10*T10)+DD(1)+T1*(DD(2)+T1*(DD(3)+T1*DD(4)
1 )))
IF (ABS(G)-LT. EPS) GO TO 999
I=I+5/CP
12 CONTINUE
WRITE (6,119) IT ,M,T,CP,VI ,G ,EPS
STOP
20 CONTINUE
FAA=1.0-F(NS1)
PC=XL(1,NS1)*F(NS1)/MI(NS1) +.KL(1,5)*.251/MI(5)+(-.11,7)*.769/
1 MI(7))*FAA
CP=PC
I=H/CP
999 CONTINUE
119 FORMAT(12X, '---ITERATION DOES NOT CONVERGED TO THE CORRECT TEMPERA
TURE *,I5/2X,1P10E12.3)
RETURN
END

```