

TRAINING COURSE PRESENTATION

Contaminant Flux and Fate in Fractured Rock

ESTCP Project ER-201570

OCTOBER 2016

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REPORT DOCUMENTATION PAGE					<i>Form Approved</i> OMB No. 0704-0188	
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1. REPORT DATE (DD-MM-YYYY) 27-10-2016		2. REPORT TYPE ESTCP Training Course Presentation			3. DATES COVERED (From - To) 9/25/2015 - 6/30/2018	
4. TITLE AND SUBTITLE Contaminant Flux and Fate in Fractured Bedrock				5a. CONTRACT NUMBER 15-P-0099		
				5b. GRANT NUMBER		
				5c. PROGRAM ELEMENT NUMBER		
6. AUTHOR(S) Charles Schaefer				5d. PROJECT NUMBER ER-201570		
				5e. TASK NUMBER		
				5f. WORK UNIT NUMBER		
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) CDM Smith 110 Fieldcrest Ave., #8, 6th Floor Edison, NJ 08648					8. PERFORMING ORGANIZATION REPORT NUMBER ER-201570	
9. SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS(ES) Environmental Security Technology Certification Program 4800 Mark Center Drive, Suite 16F16 Alexandria, VA 22350-3605					10. SPONSOR/MONITOR'S ACRONYM(S) ESTCP	
					11. SPONSOR/MONITOR'S REPORT NUMBER(S) ER-201570	
12. DISTRIBUTION/AVAILABILITY STATEMENT DISTRIBUTION STATEMENT A. Approved for public release: distribution unlimited.						
13. SUPPLEMENTARY NOTES						
14. ABSTRACT The objectives of this presentation are to: •Identifying and characterizing DNAPL in bedrock -Edwards Air Force Base (AFB) •Abiotic Dechlorination in Rock Matrices (coupled diffusion & reaction) •Insights at Loring AFB						
15. SUBJECT TERMS Contaminant Flux and Fate, Fractured Bedrock						
16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF ABSTRACT UNCLASS	18. NUMBER OF PAGES 43	19a. NAME OF RESPONSIBLE PERSON Charles Schaefer	
a. REPORT UNCLASS	b. ABSTRACT UNCLASS	c. THIS PAGE UNCLASS			19b. TELEPHONE NUMBER (Include area code) 732-590-4633	

Contaminant Flux and Fate in Fractured Rock



October 27, 2016

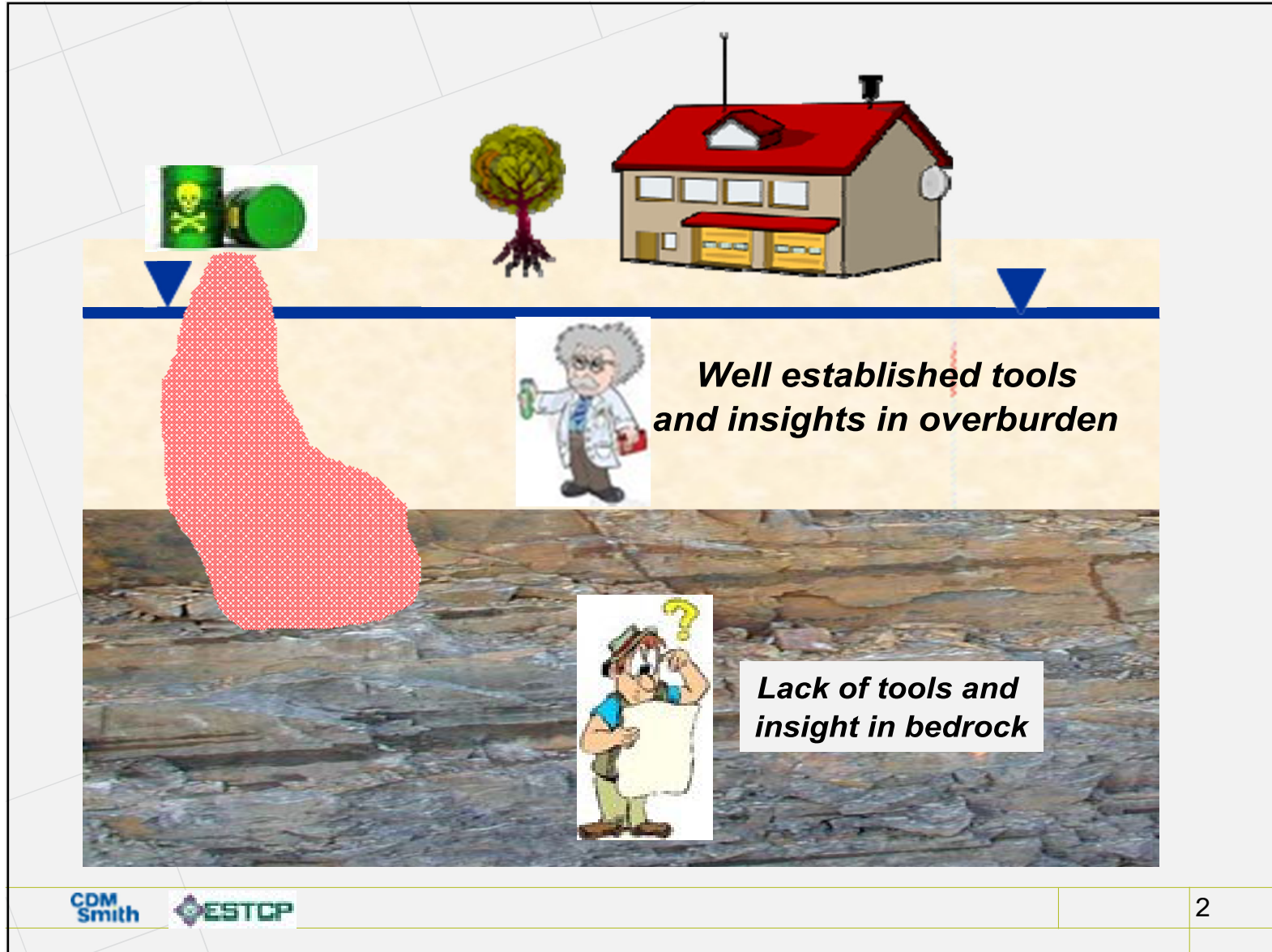
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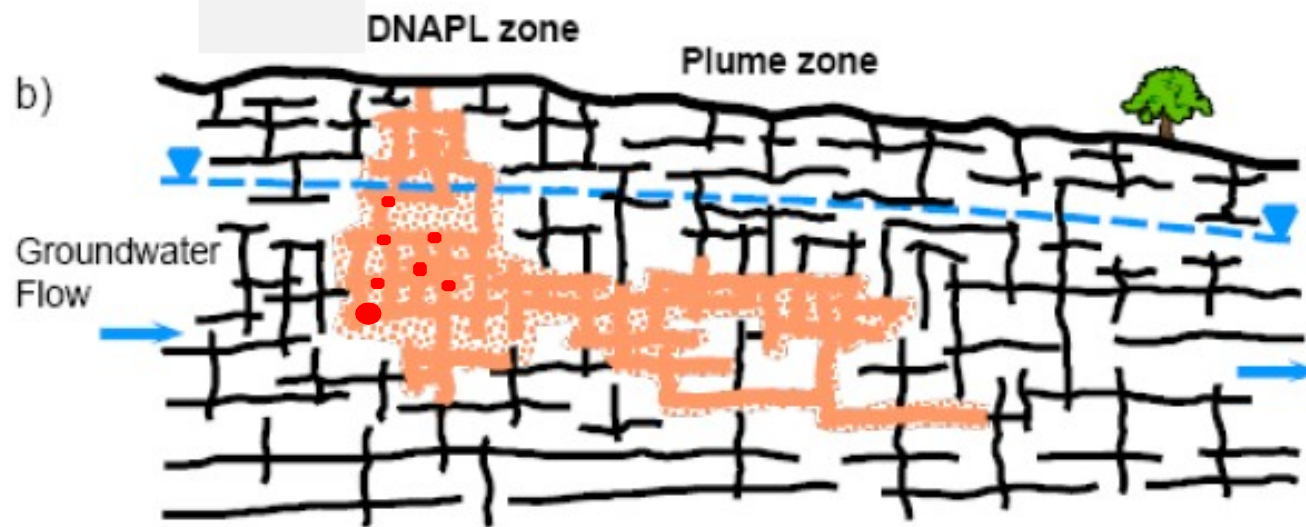
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Michael Bishop
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University of Florida
Prof. Mike Annable
Erin White

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Chlorinated Solvents in



Modified from Parker, B.L. Fractured Rock Conference: State of the Science and Measuring Success in Remediation, September 24-26, 2007, Portland, Maine, 2007

What is sustaining Perchloroethylene (PCE)/Trichloroethylene (TCE) plumes in bedrock?

- *Dense non-aqueous phase liquid (DNAPL)?*
- *Matrix back-diffusion?*
- *Low permeability fractures?*
- *Overburden impact?*

How quickly is contaminant mass attenuating?

Objectives

- Identifying and characterizing DNAPL in bedrock
 - *Edwards Air Force Base (AFB)*
- Abiotic Dechlorination in Rock Matrices
(coupled diffusion & reaction)
- Insights at Loring AFB

1. DNAPL ARCHITECTURE & DISSOLUTION

The DNAPL Challenge

- Most of the contaminant mass may be in the non-aqueous phase
- Dissolution rate may limit remedial effectiveness and mass discharge
- Locating and contacting DNAPL sources can be challenging

Complicating Factors in Bedrock

- Many of the technologies for locating and quantifying DNAPL sources are not appropriate, or have not been demonstrated, for bedrock
- DNAPL may be even more difficult to contact in fractured bedrock
- Costs



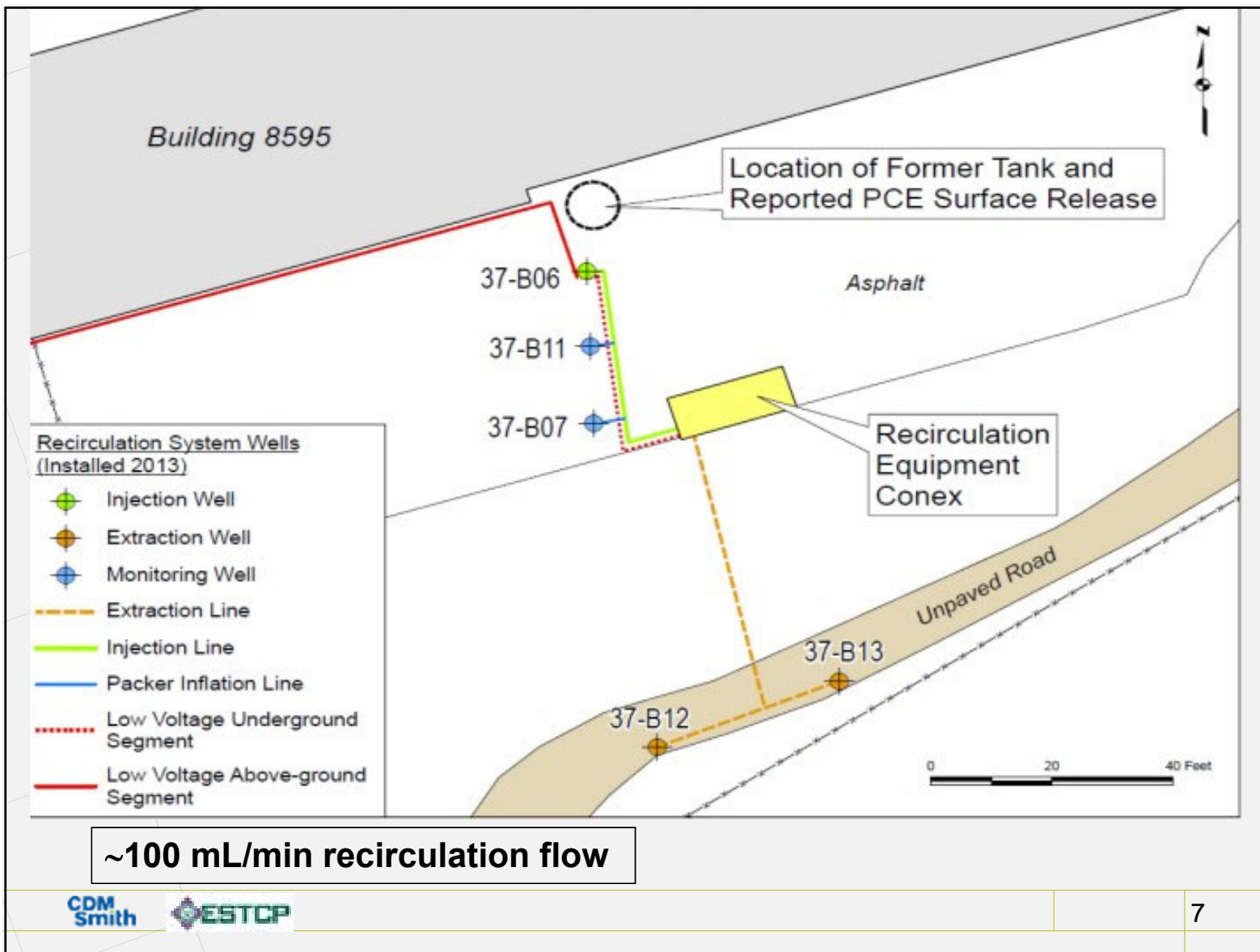
Illustrative Example – Key Insights

Site 37 Characteristics

- Large plume (390 acres)
- Deep (>200 feet)
- Granite bedrock (quartz/feldspar)
- Low transmissivity
- Fracture flow
- PCE at >10% solubility
- No direct evidence of DNAPL (prior to this project)

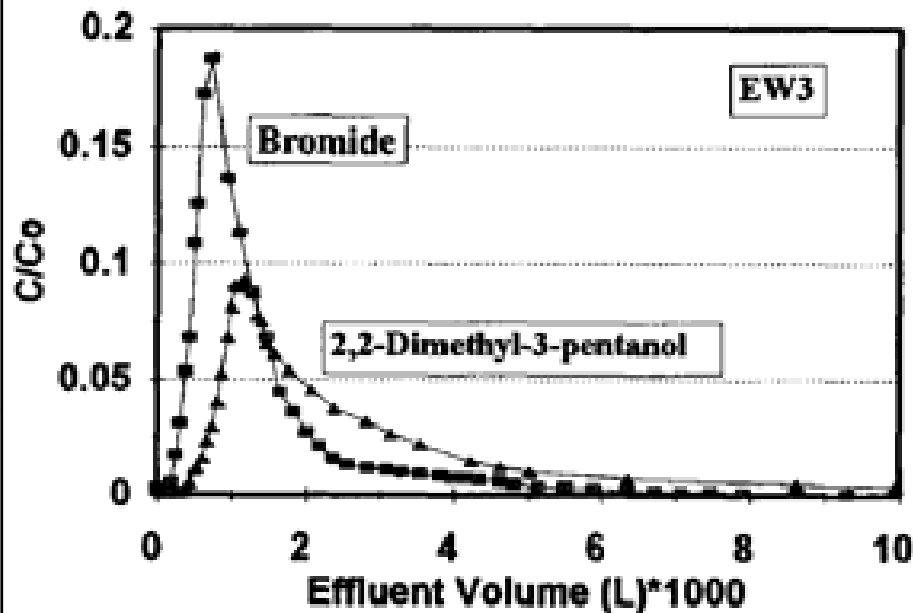
Demonstration Location -
Edwards AFB
(ESTCP 201210)





Partitioning Tracer Testing (PTT)

Applicable to Unconsolidated and Consolidated Materials

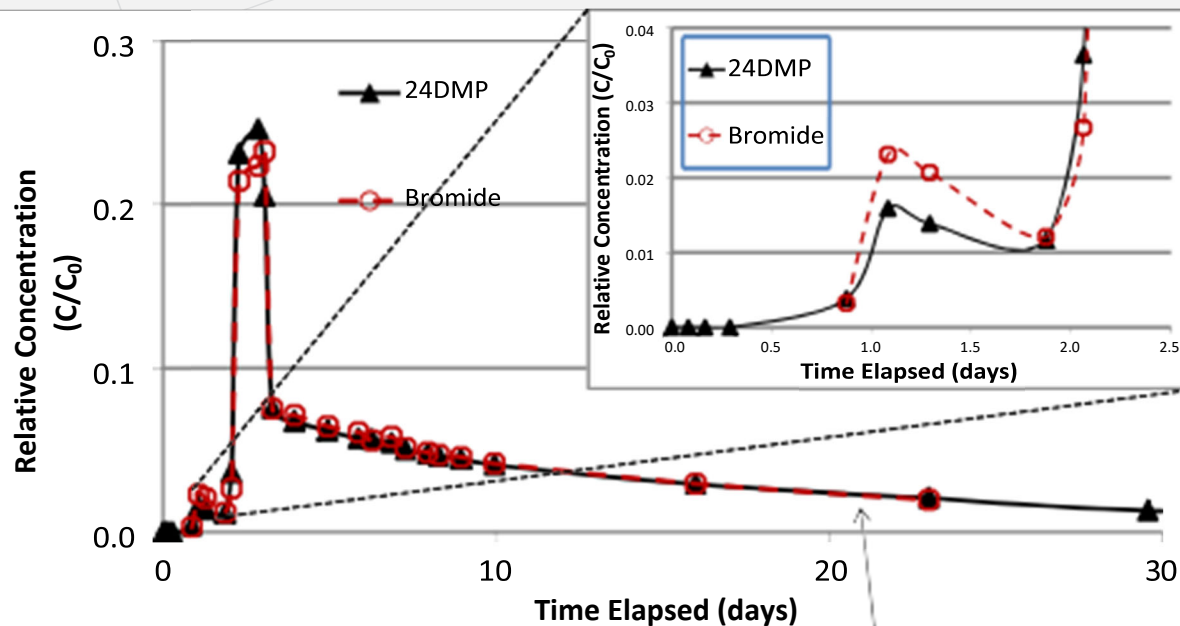


Compare elution of conservative and partitioning tracers (alcohols) to verify and quantify DNAPL

Annable et al., JEE, 1998

Tracer Results – Deep Zone

Bromide mass eluting through each zone proportional to transmissivity



Initial Peak (low T fracture)

- 1% of flow
- $S_n = 0.007$

Middle Peak

- 9% of flow
- $S_n = 0.0$

Late Peak

- 40% of flow
- $S_n = 0.0004$

Mass transfer-controlled tailing

What Else Does This Tell Us?



Available
12/15/2010 10:00 AM

Dense Nonaqueous-Phase Liquid Architecture in Fractured Bedrock:
Implications for Treatment and Plume Longevity

Charles E. Schaefer,^{1,*} Erin B. White,² Graig M. Levenson,³ and Michael D. Annable²

DNAPL distribution

DNAPL present in high transmissivity fractures, but also in low transmissivity zones

Average fracture porosity

0.004

DNAPL mass

2.4 kg in 15 ft radius around injection well interval

DNAPL persistence under ambient conditions (dissolution only)

B11s (high T) – 65 years B11d (tail) – 13 years

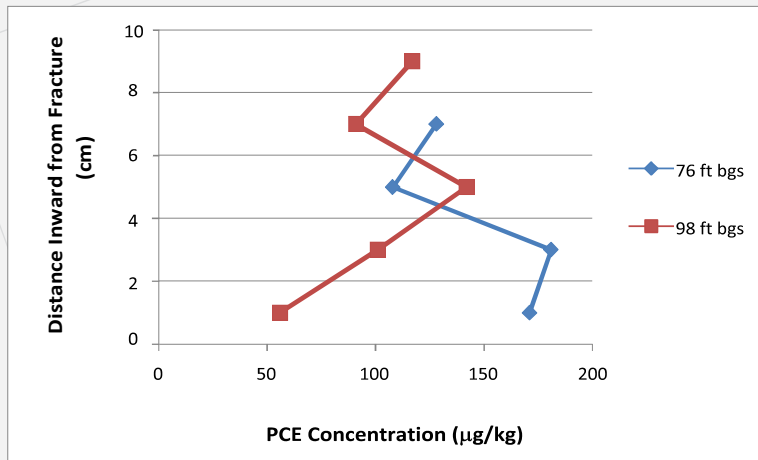
B11d (low T) – 194 years

PCE Mass Distribution and Flux

Rock Matrix

vs

Fractures



Based on PTT DNAPL

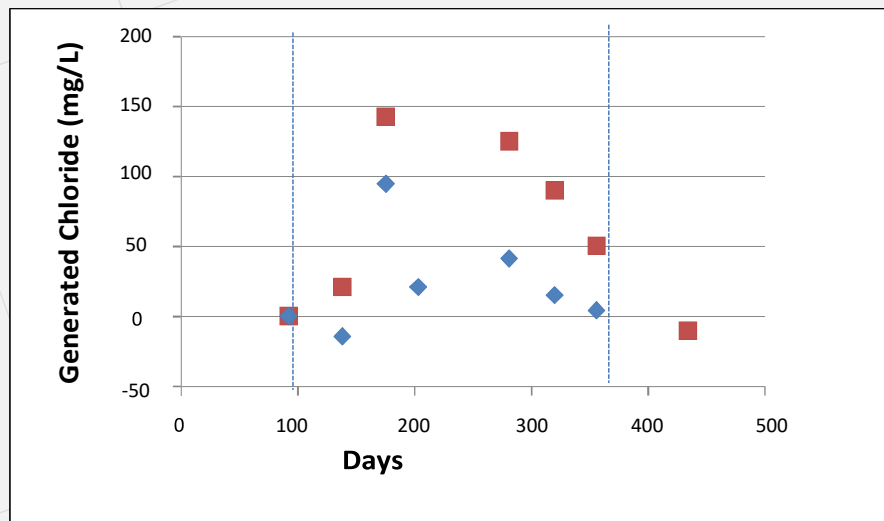
149 g PCE in rock matrix

**PCE concentration profile suggests
back-diffusion not occurring**

**2,400 g PCE as DNAPL
in fractures**

Treating to remove DNAPL might make sense

Bioaugmentation Results



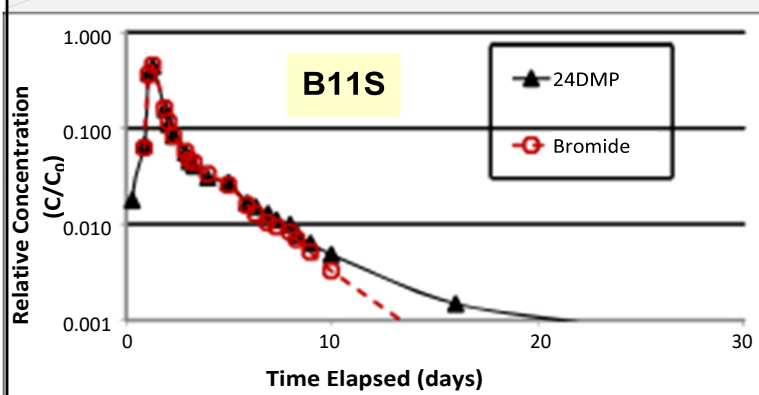
■ Shallow zone

◆ Deep Sone

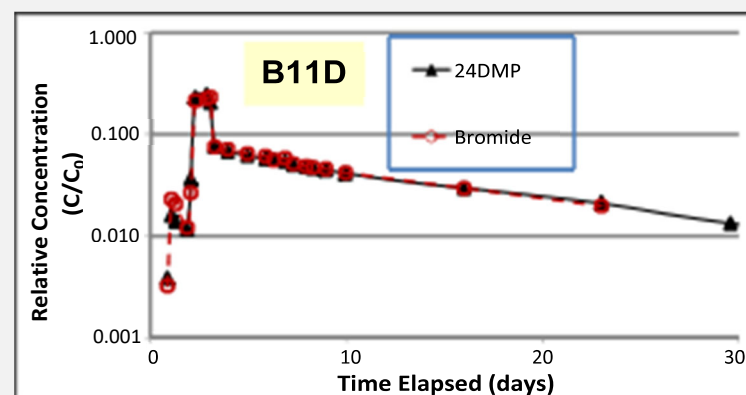
Substantial chloride generation indicated DNAPL dissolution (2.1 mM of PCE to Dichloroethylene [DCE])

Chlorinated volatile organic compounds (CVOC) & ethene are not the best indicator of DNAPL removal (Schaefer et al. ES&T, 2010)

Impacts of DNAPL Architecture on Treatment During Bioaugmentation



~100% DNAPL removal
Limited rebound (97% CVOC decrease)
High DNAPL diss. enhancement



Only 45% DNAPL removal
Rebound
Low DNAPL diss. enhancement

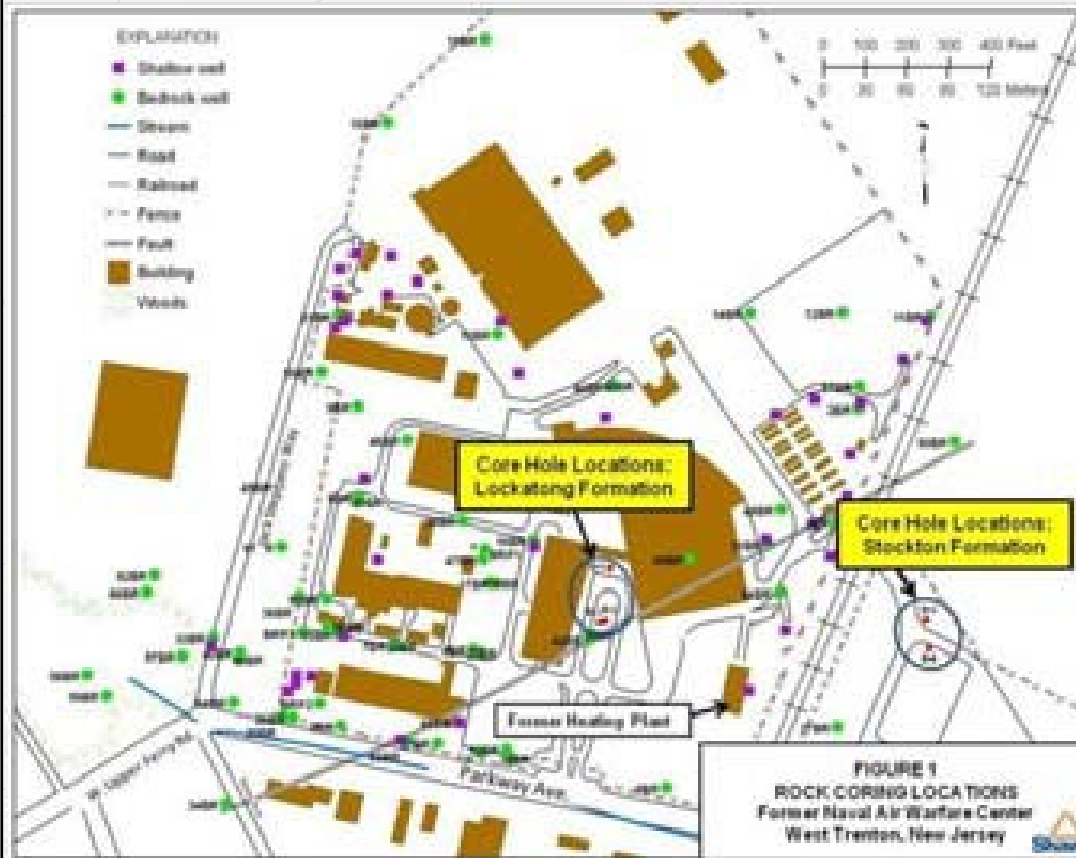
But a small fraction of DNAPL mass removal may result in a large improvement in groundwater

2. Rock Matrix Dechlorination Reactions

- Matrix diffusion is a SLOW process, so even a slow dechlorination reaction can be significant
- Several studies have focused on abiotic dechlorination reactions facilitated by naturally occurring ferrous minerals
 - *Bedrock matrices have received little attention*
 - *Ferrous minerals are present in many bedrock aquifers*
- **Are such reactions occurring, can they be measured, and are they important for my site?**

Rock Core Collection (SERDP Project ER-1685)

former Naval Air Warfare Center, Trenton, NJ



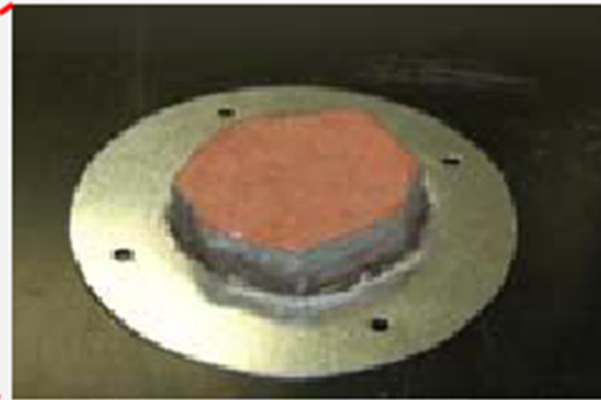
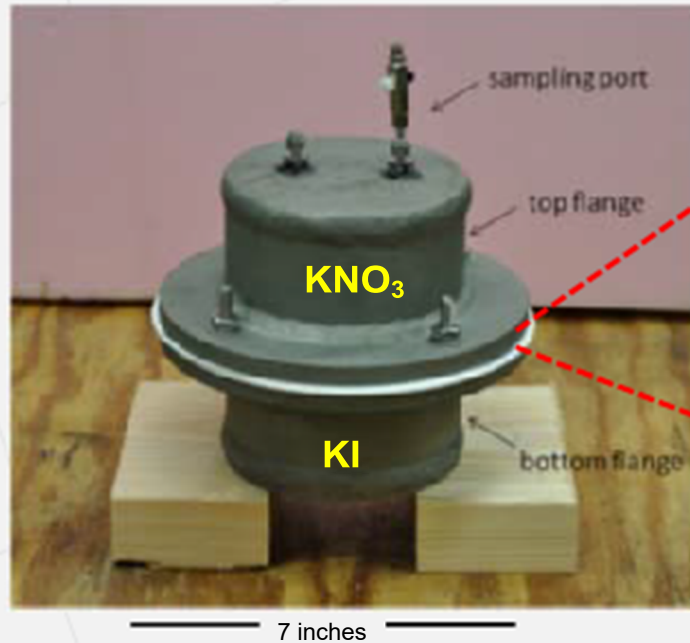
Lockatong Formation:

- Multiple types of mudstones and shale
- Multiple water bearing fractures
- Ferrous iron minerals

Stockton Formation

- Multiple water bearing fractures
- Red and white sandstones/siltstones

Diffusion Cell Experiments

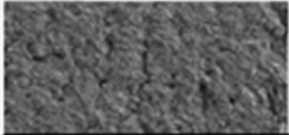
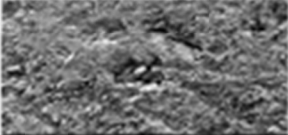
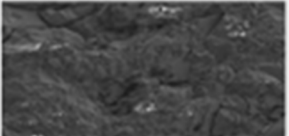
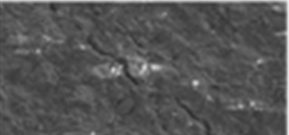
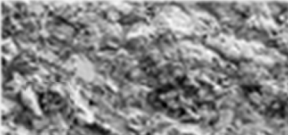
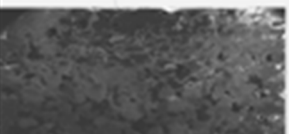
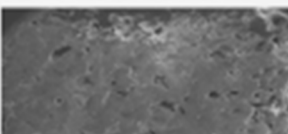


Mounted Rock Disc
(0.5 to 2 cm thick)

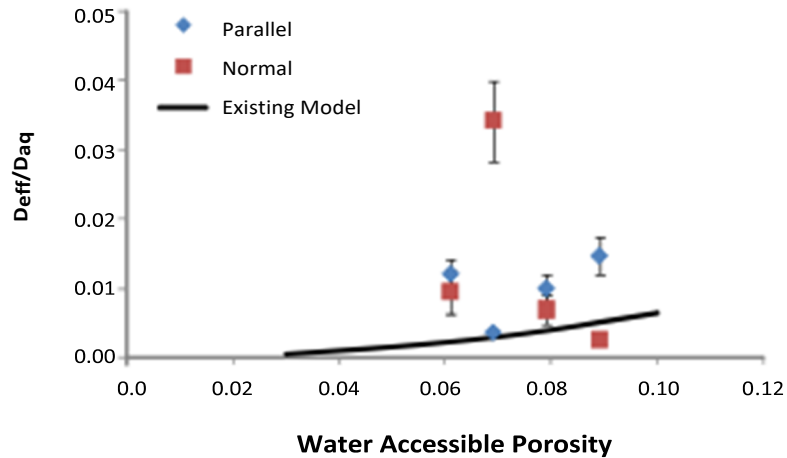
Iodide tracer experiments to assess diffusion parallel and normal to bedding
- four rock types

Iodide Diffusion Results: Impact of Mineral Bedding

Scanning Electron Microscope (SEM) Images

				SEM Porosity	Effective Diffusion Coefficient (cm ² /s)
Normal	Parallel				
 100 μm	 100 μm	Red Sandstone	Normal:	0.074	1.8 x 10 ⁻⁷
			Parallel:	0.069	2.3 x 10 ⁻⁷
 10 μm	 10 μm	Tan Sandstone	Normal:	0.068	1.3 x 10 ⁻⁷
			Parallel:	0.085	1.9 x 10 ⁻⁷
 10 μm	 10 μm	D.G. Mudstone	Normal:	0.050	0.49 x 10 ⁻⁷
			Parallel:	0.11	2.8 x 10 ⁻⁷
 100 μm	 100 μm	Quarry Sandstone	Normal:	0.085	6.5 x 10 ⁻⁷
			Parallel:	0.037	0.68 x 10 ⁻⁷

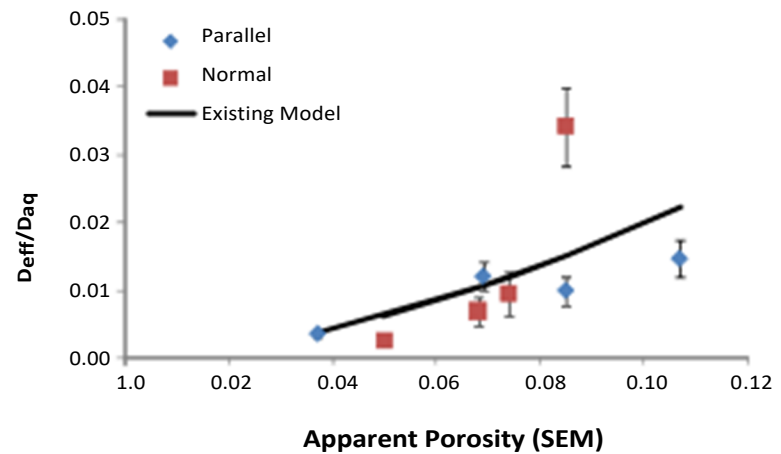
Predicting Diffusion Based on Apparent Porosity



$$\frac{D_{\text{eff}}}{D_{\text{aq}}} = \varepsilon^{2.2} \quad (\text{Boving and Grathwohl, 2001})$$

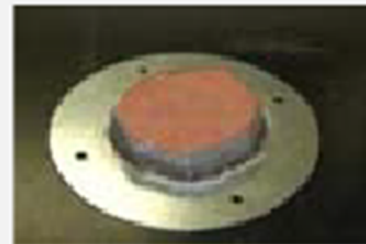
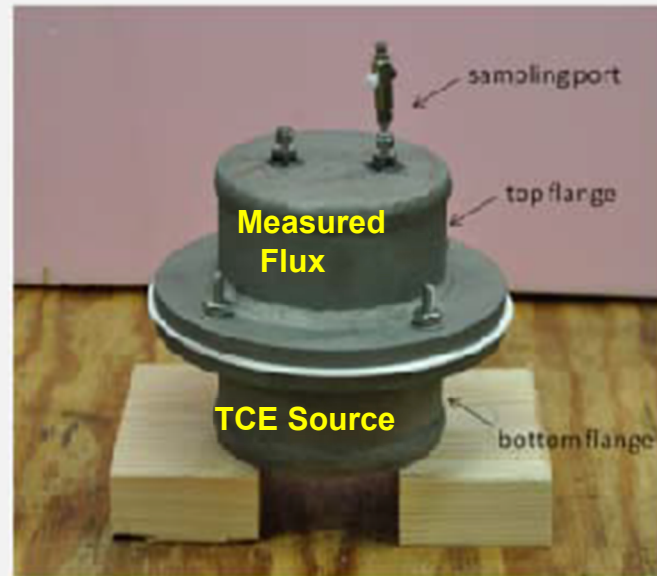
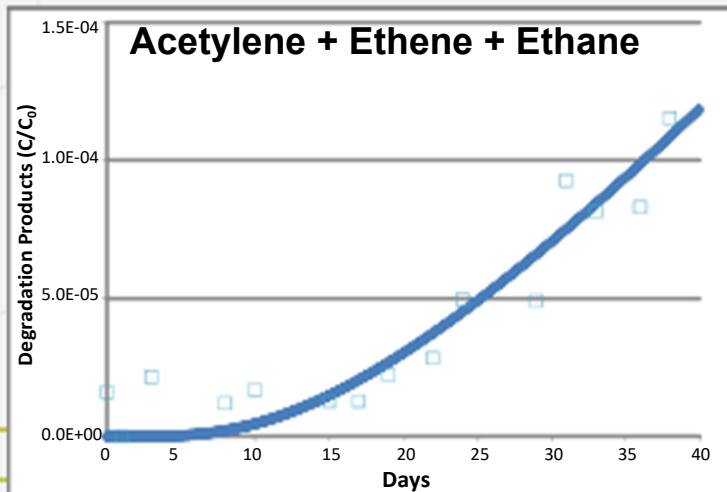
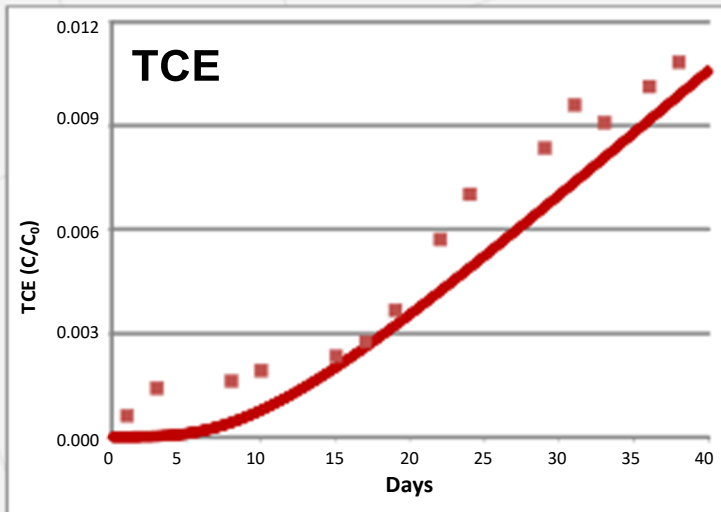


Modified Correlation: $\frac{D_{\text{eff}}}{D_{\text{aq}}} = \varepsilon_o^{1.7}$



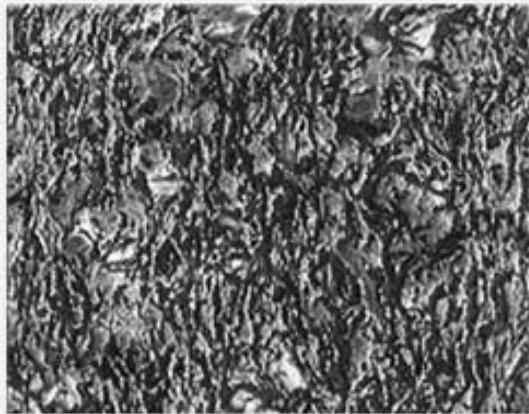
Coupled TCE Diffusion and Reaction

Light Gray Mudstone

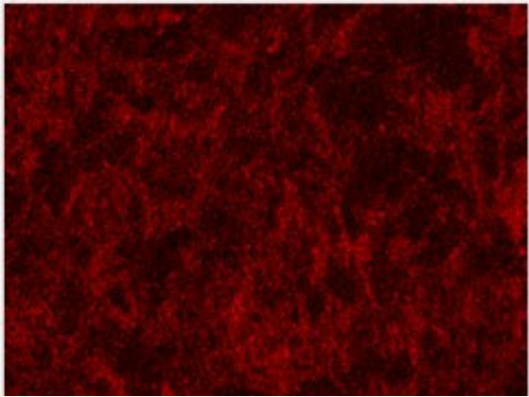


HgCl₂ added to inhibit microbial activity

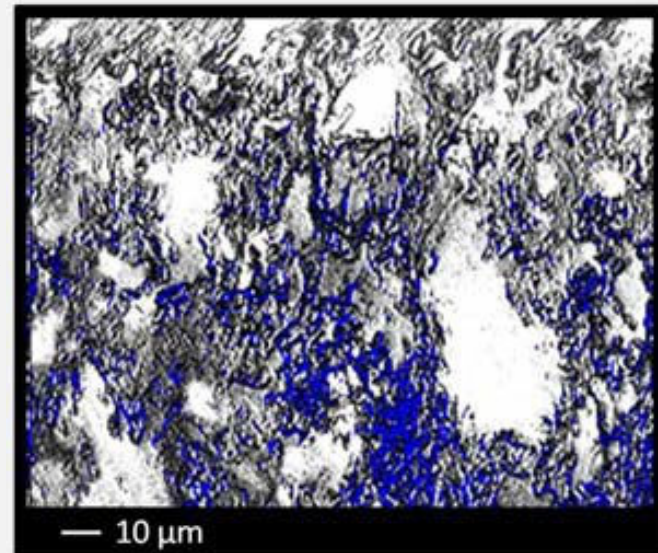
Available Ferrous Iron



SEM for porosity distribution



Electron Dispersive Spectroscopy for Fe²⁺ distribution



"Available" Fe²⁺ indicated by blue

Coupled Diffusion and Reaction of TCE through Rock

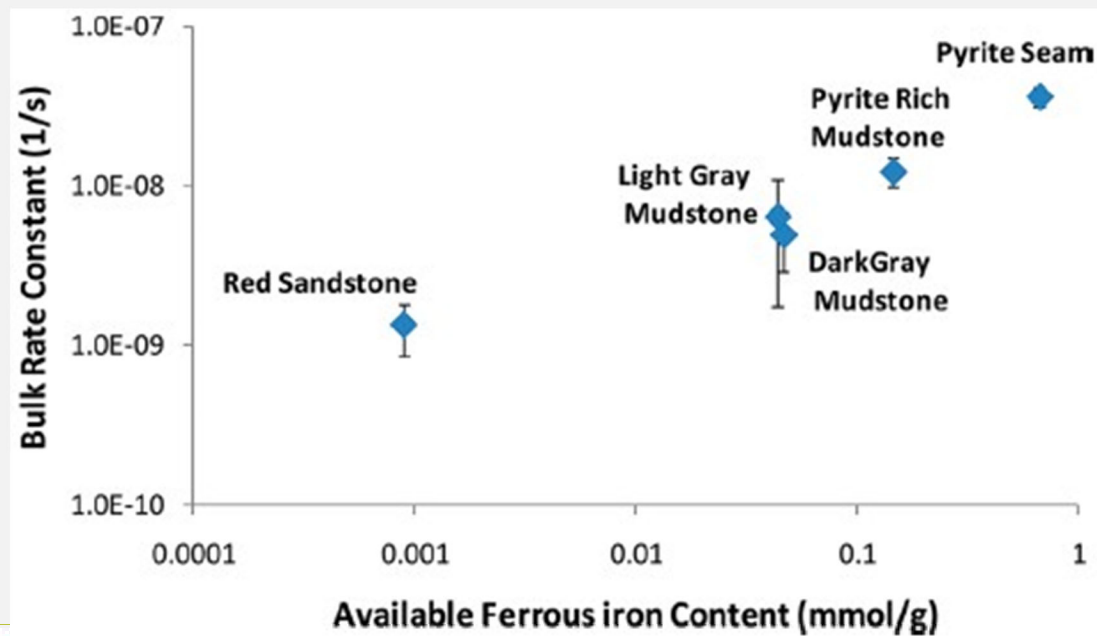
Coupled Diffusion and Abiotic Reaction of Trichlorethene in Minimally Disturbed Rock Matrices

Charles E. Scharley,^{1,2} Rachael M. Trione,² David R. Lippincott,² Vello Lammik,²
Timothy B. Fischer,² Michael E. Bishop,³ and Hailiang Dong¹

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³Department of Geology, Miami University, Oxford, Ohio 45056, United States



- Up to 10 months exposure at elevated TCE concentrations (200 mg/L) did not result in any decrease in the observed dechlorination rate constant
- Low stoichiometric Fe^{2+} consumption (<2%)

**But are these reactions really
occurring in historically contaminated
bedrock aquifers?**

Collect Core from within the TCE Plume



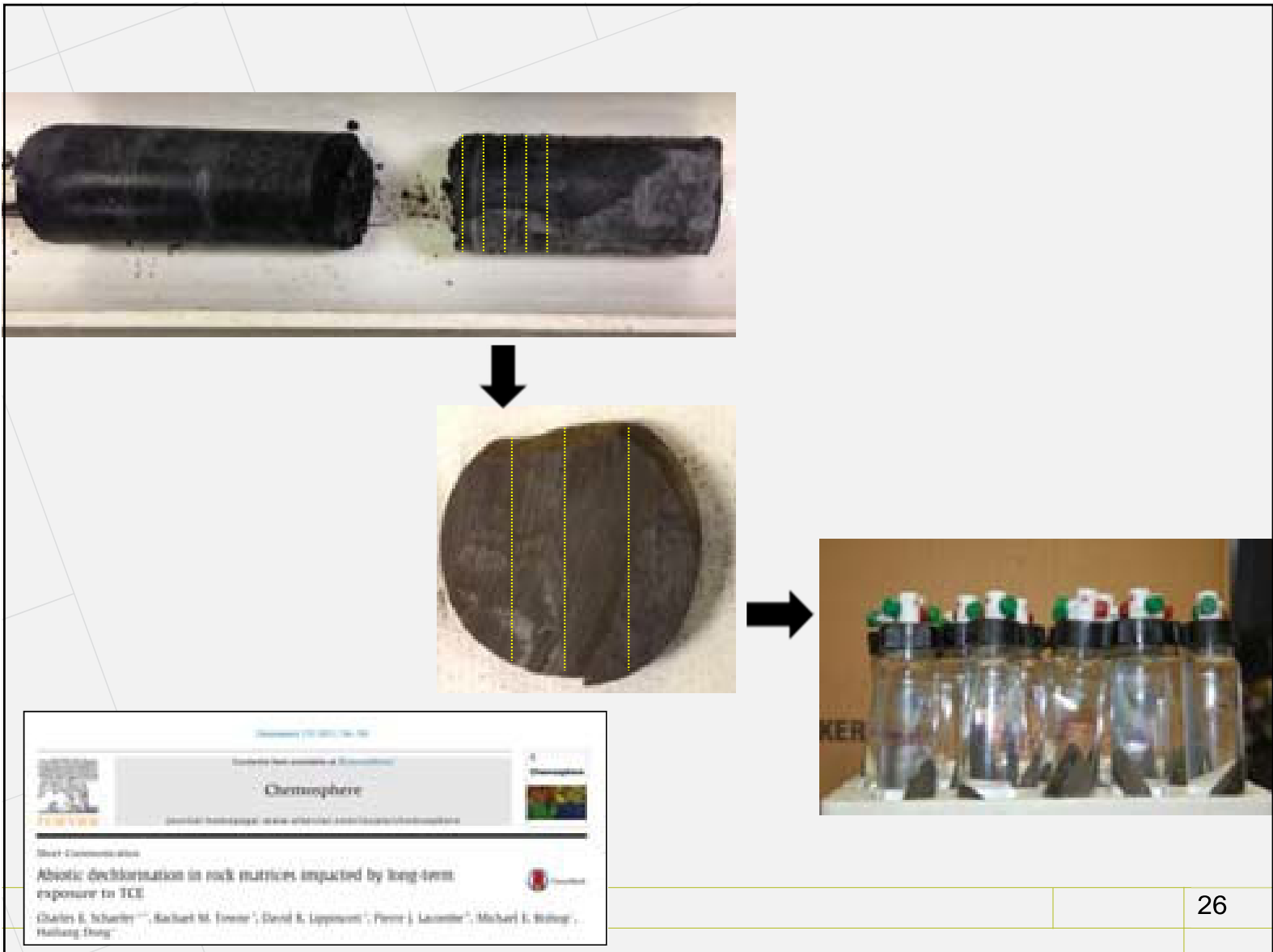
- Collection at the former Naval Air Warfare Center, Trenton, NJ, USA
- Core collected adjacent to fractures at three depth intervals



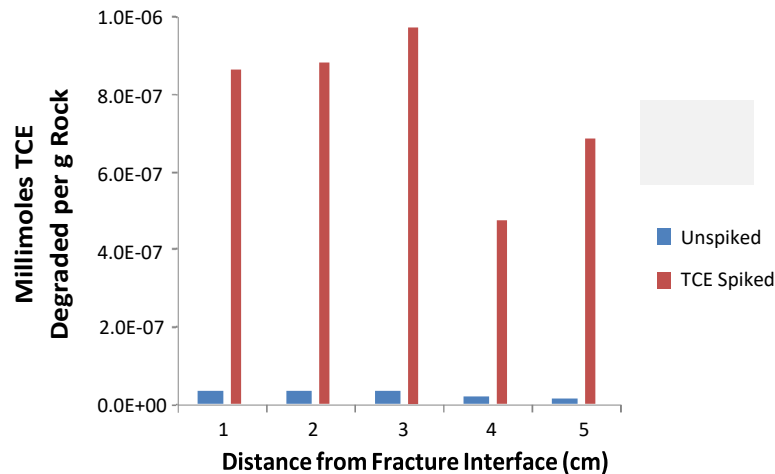
Groundwater Data as Screening for Potential Abiotic Dechlorination Reactions

Groundwater Packer Sampling (μg/L)

Compound	49'	53'	83'
TCE	203	463	24,800
<i>cis</i> -DCE	59	410	260
vinyl chloride	2	31	6.2
methane	28	31	4.8
ethane	3.2	2.0	0.48
ethene	-	3.6	-
propane	2.4	2.4	0.3
acetylene	-	-	4.8



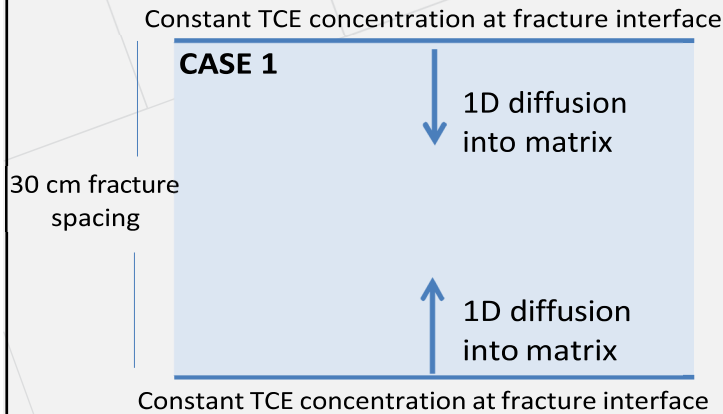
Abiotic TCE Dechlorination in Rock Collected within the TCE Plume



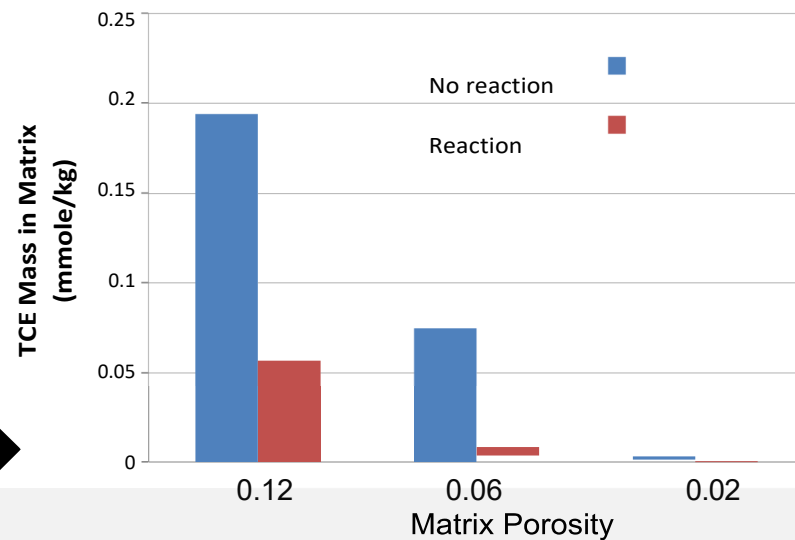
***Comparison of
TCE spiked vs. non TCE spiked
samples in laboratory system***

***Rock minerals still able
to dechlorinate TCE
(similar rate constant)***

How Important are These Reactions?



Dechlorination half-life = 2.1 years



Simulated 30 years of uptake

How Important are the Reactions?

Diffusion Simulations

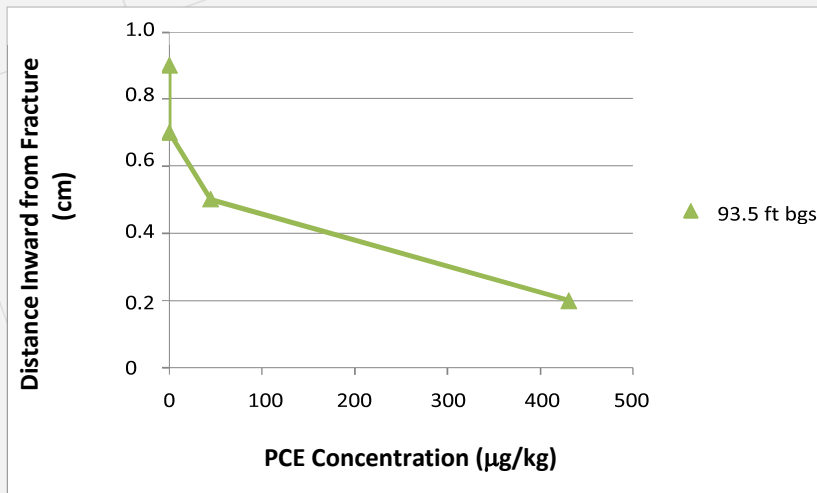
- After 30 years constant TCE exposure, 30 years of back-diffusion (maximum gradient: no TCE in fractures)



Can help determine if matrix back-diffusion is sustaining the plume

More Evidence

PCE/TCE concentrations profiles in the rock matrix moving inward from the fracture interface (cm-scale):

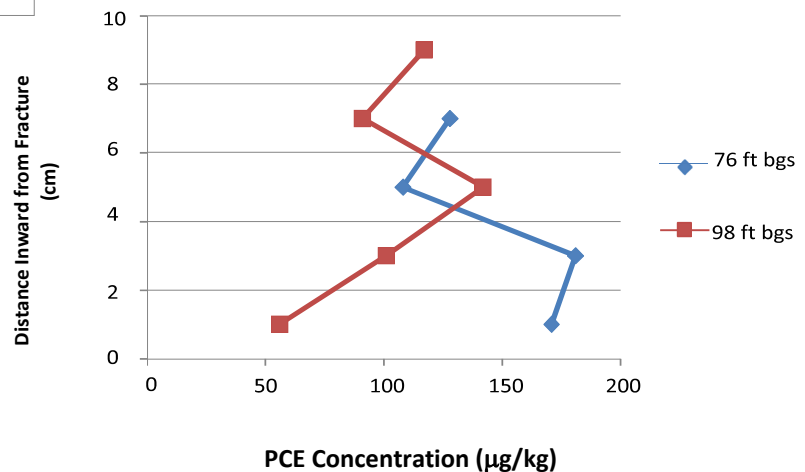


- Loring AFB Limestone

- Batch testing showed substantial abiotic reactivity

- Edwards AFB Granite

- Batch testing showed negligible abiotic reactivity



How Readily Can This Information Be Attained?

Initial Screening

- Porosity analysis
 - <\$400/sample **Several commercial labs**
- Reduced gases in groundwater (acetylene, propane, ethane, ethene)
 - <\$100/sample **Several commercial labs (ask for gases)**
- Ferrous iron in crushed rock (EPA Method SM3500-Fe B.4.c)
 - ~\$75/sample **ALS Environmental & McCampbell Analytical**

Next Steps - Using Intact Rock Core:

- Field crushing and methanol extraction of rock “slices”
 - \$10K (**not including drilling**) **Several labs**
- Batch reactivity and sorption testing
 - ~\$25K lab cost (assuming core available) **Published methods**

3. Impacts of Matrix Diffusion and Reaction on the Dissolved Plume: Loring AFB – Groundwater Management Zone (GMZ) 4



Remedial efforts as part of sitewide performance based contract (CB&I)

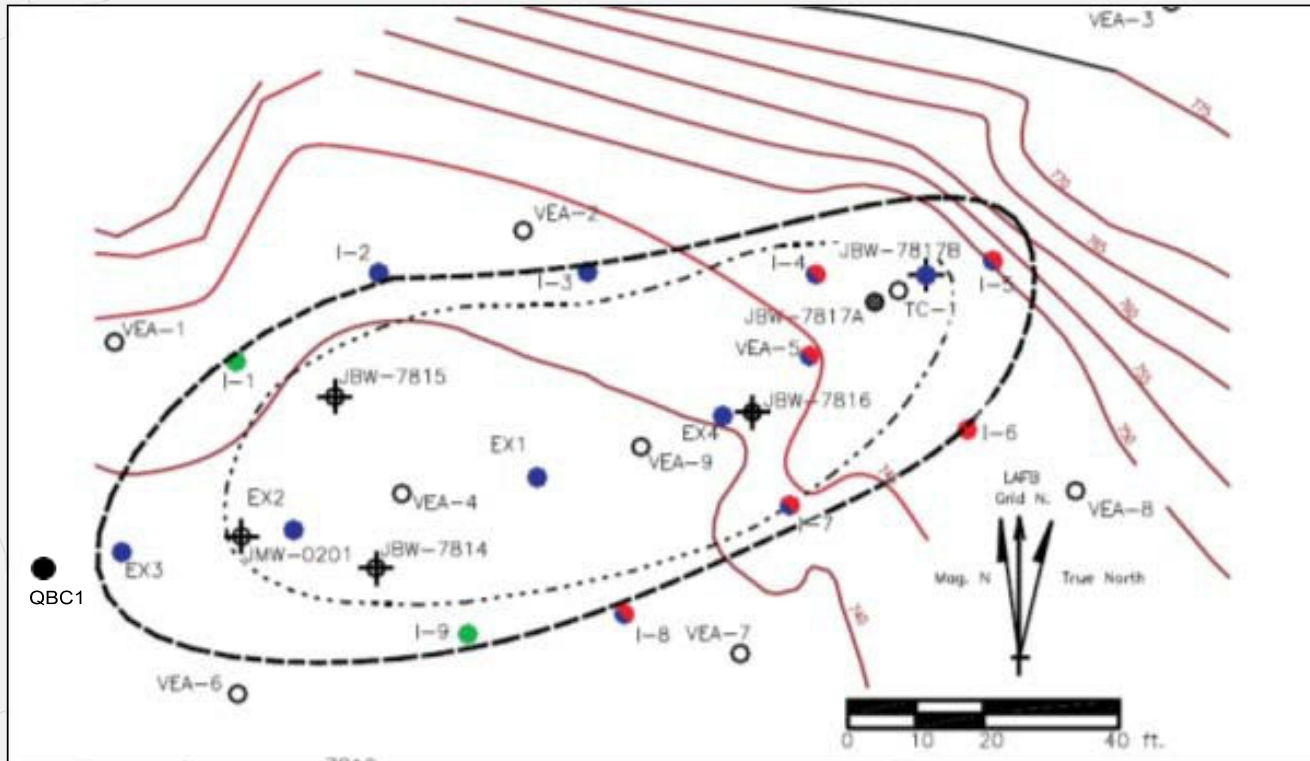
USAF, Maine DEP and USEPA very involved

GMZ 4

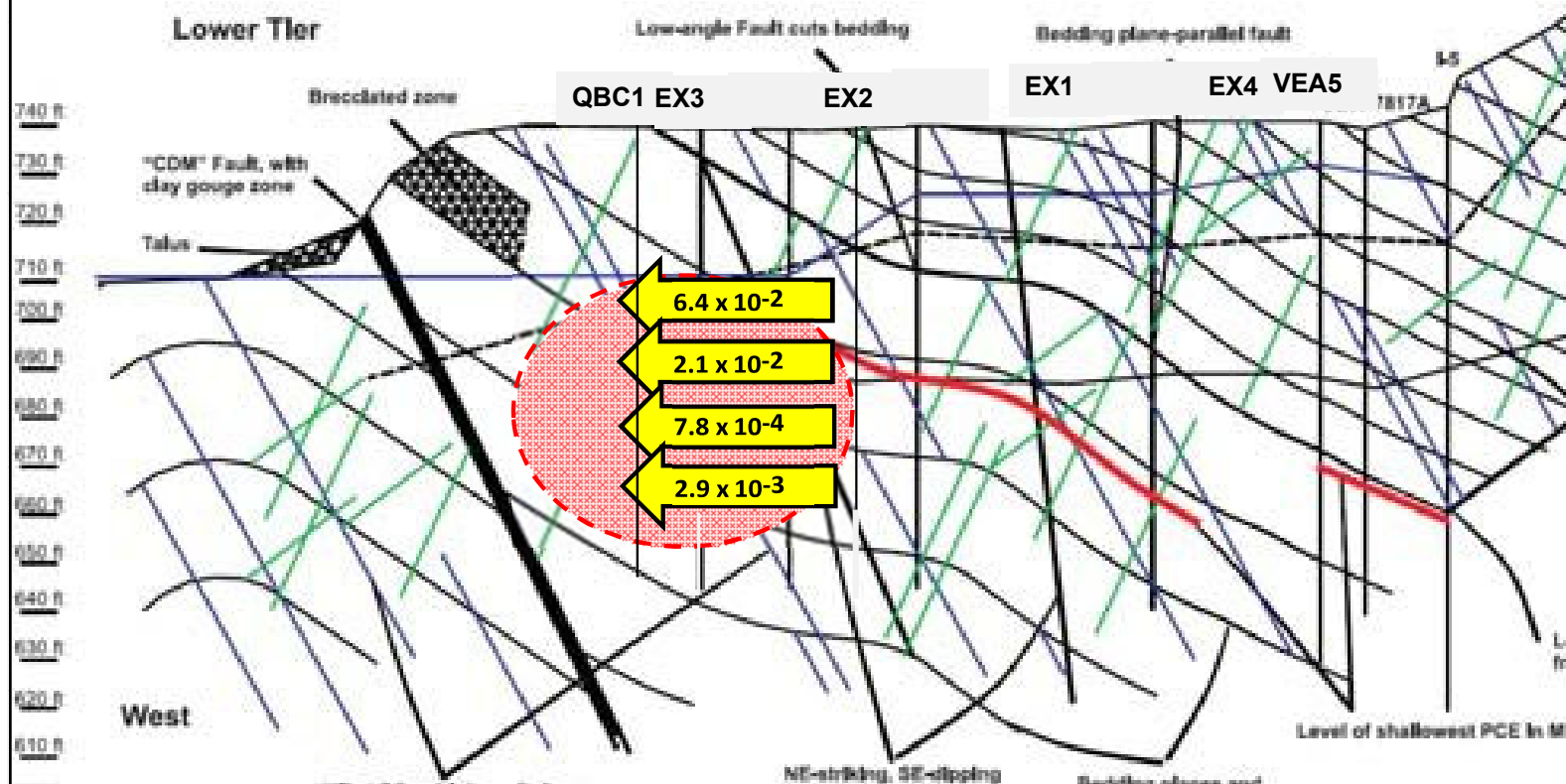
- fractured limestone
- complex fracture flow
- former drum disposal
- PCE (DNAPL likely)
- prior steam injection
- no groundwater data for ~10 years



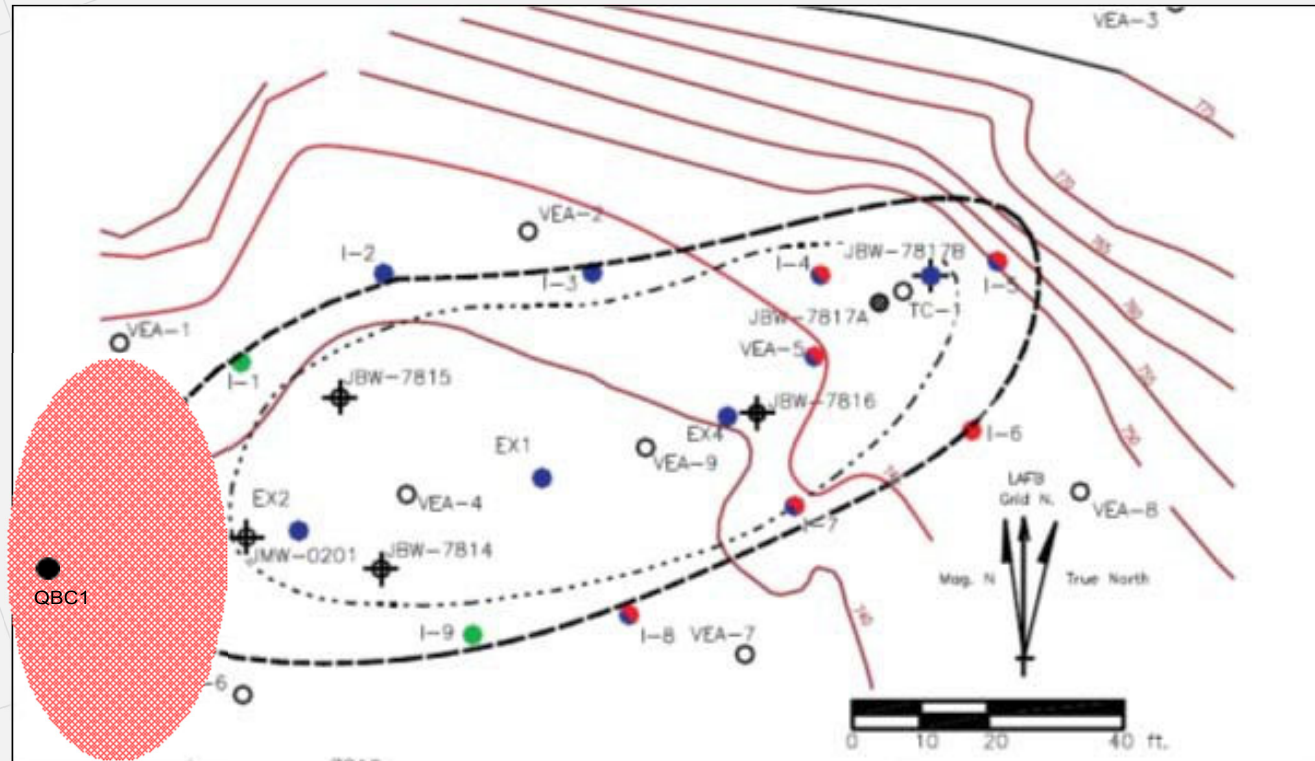
PCE Source Area in Fractured Limestone



Identify High Contaminant Flux Zones



Target Treatment Area



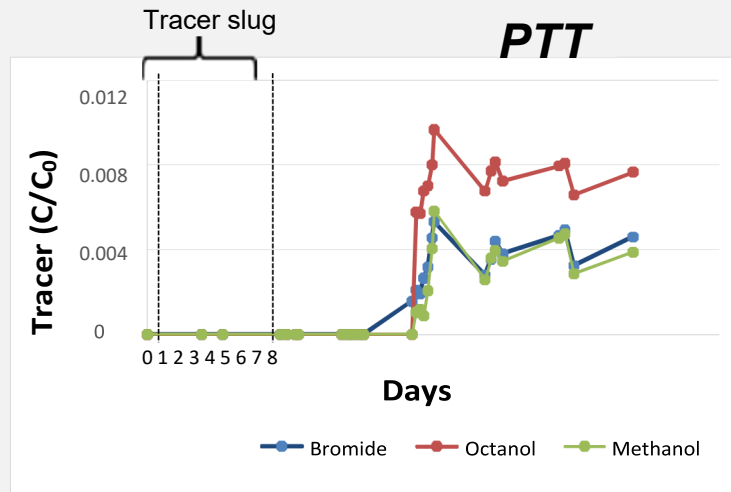
Bioaugmentation to Target High PCE Flux Zone



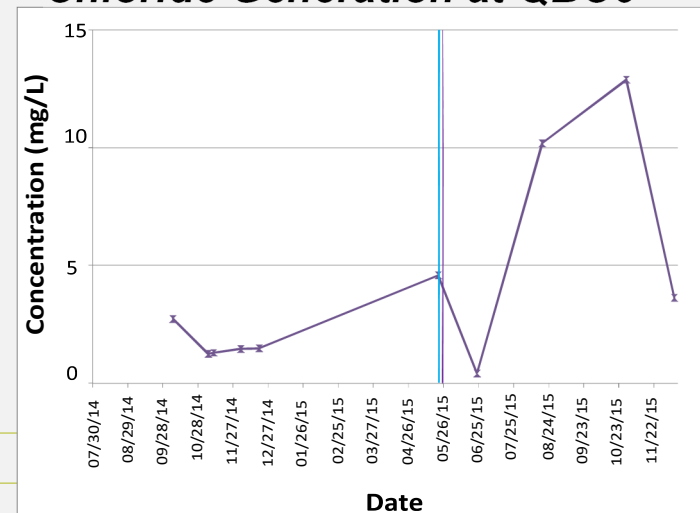
Treatment is on-going

CDM
Smith

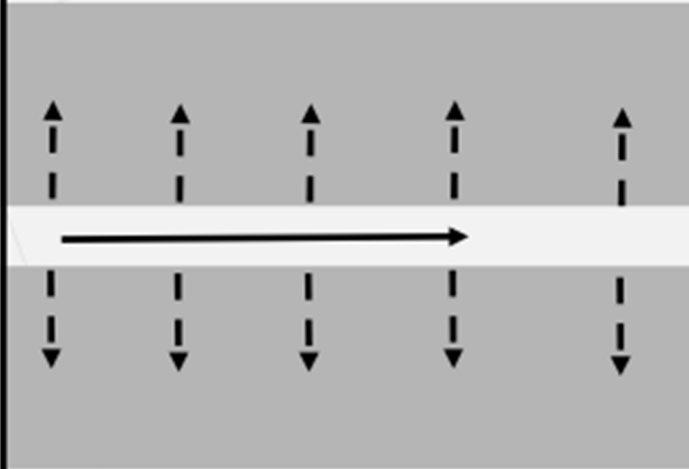
ESTCP



Chloride Generation at QBC6



Preliminary Assessment – Role of Matrix Dechlorination Reactions

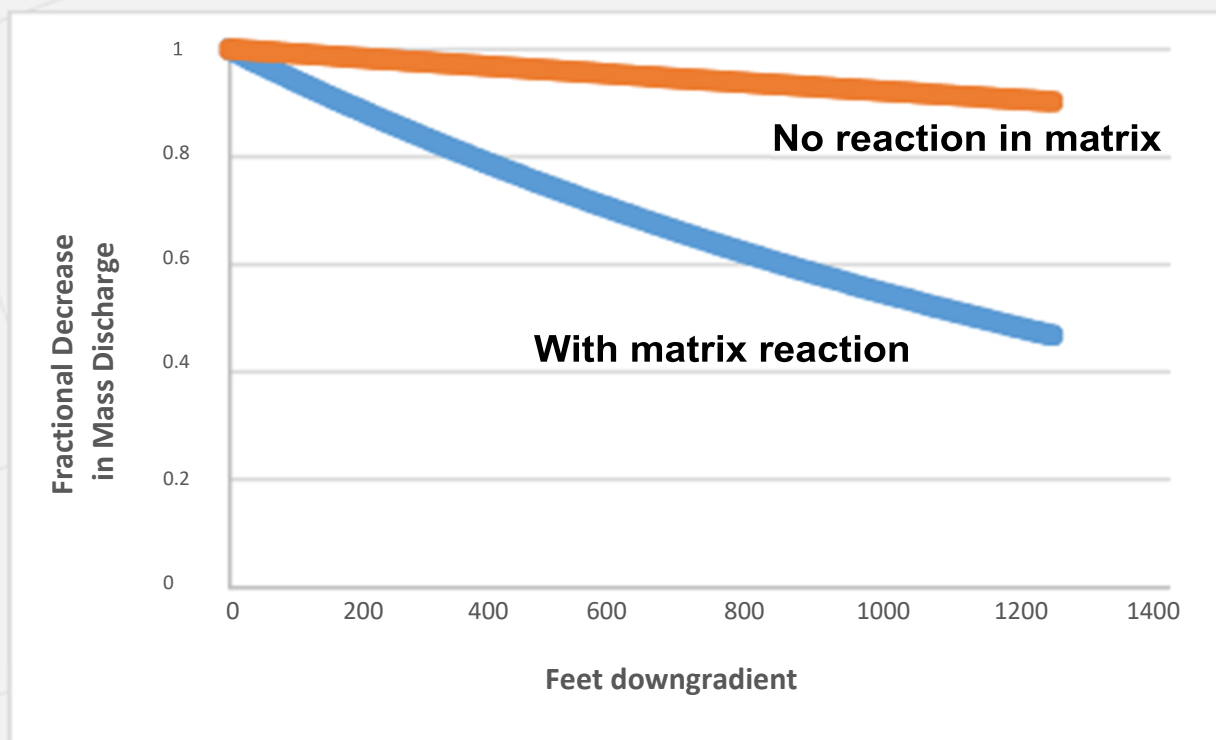


Determine:

- Fracture aperture (cubic law and transmissivity) [0.021 cm]
- Superficial velocity (Darcy's Law and fracture porosity) [400 cm/day]
- Rock matrix effective diffusion coefficient estimated from rock matrix porosity
[$3.6 \times 10^{-9} \text{ cm}^2/\text{s}$]
- First order dechlorination rate constant
[$6.6 \times 10^{-7} \text{ s}^{-1}$]

Flux at fracture surface = $2 \operatorname{erf} \left[\sqrt{\frac{C}{n}} \right]$ (Crank, 1995)

Role of the Rock Matrix for Natural Attenuation of the Downgradient Dissolved Plume



*Model
assumes no
dilution
due to
dispersion*

Conclusions

- Residual DNAPL can be identified and quantified in fractured rock
- DNAPL in low transmissivity fracture zones may be sustaining bedrock plumes
- In DNAPL source areas, most of the contaminant mass and flux might be due to DNAPL
 - *thus, treatment may be worthwhile*
- Abiotic dechlorination reactions in rock matrices may be mitigating back-diffusion and contaminant migration at many sites
- Identifying and quantifying these dechlorination reactions can be readily performed

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