

Tuning the Functionality of Materials with Reversible Phase Transitions

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Tuning the Functionality of Materials with Reversible Phase Transitions

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Abstract

We investigated solid-state phase transitions of complex metal oxides as an effective way to dynamically control macroscopic material properties and their functionality via lattice and defect engineering. We demonstrated that careful film growth and the strain generated by buffer layers can provide an effective way for tuning the transition temperature of metal oxides with reversible phase transitions such as VO₂ films and in the process control their functionality. We also demonstrated an ultrafast switching capability (<400 fs) of the strained VO₂ films using a time-resolved optoelectronic autocorrelation measurement technique, indicating that it is possible to selectively activate the metallic phase without inducing the associated structural transition. Furthermore, to demonstrate the applicability of this basic research, we fabricated VO₂-based, layered solid-state radiators for spacecraft thermal control. The multilayer radiators in this work are well suited for spacecraft thermal control because they are passive and self-regulating. We also demonstrated that perovskite conducting oxides can provide a large tunable permittivity in the mid-wave IR (MWIR) range (2-7 μm) for MWIR plasmonic devices. These results are highly relevant to various Navy and DoD applications ranging from optical communications to low observables. Lastly, we were able to synthesize two different structural phases of epitaxial Ti₂O₃ films. The films grown at 485 °C are a corundum structure with p-type carriers while the films grown at 730 °C become an orthorhombic structure with n-type carriers. These results can open up the possibility of p-n homojunctions that will strongly impact future Navy and DoD applications including IR detection and thermal energy conversion.

Introduction

We present here the closeout report on our program, *Tuning the Functionality of Materials with Reversible Phase Transitions*, WU# 1J42. This program aimed to control solid-state phase transitions of complex metal oxides as mechanisms to dynamically adjust macroscopic material properties via investigation of the role of lattice structure, electrical and optical properties. Reversible solid-state phase transitions present a unique method to tune a wide range of functionalities such as electrical and thermal conductivity, magnetic susceptibility, dielectric polarizability and optical properties. This is in stark contrast to current material systems, which require non-reversible changes in composition or processing in order to adjust the above properties. Recently, both the number of materials and the range of applications for materials with reversible phase transformations have expanded significantly [1-4]. The ability to vary the crystal structure or the electron band configuration through the influence of an external magnetic, electric or strain field paves the way towards new functionality and applications.

A significant outcome of this work as described below has been the study of novel, emergent phenomena in complex metal oxide thin films with strongly interacting electronic, optic and lattice degrees of freedom. We investigated several material systems including VO₂, La-doped BaSnO₃, Sn-doped In₂O₃ (ITO) and Ti₂O₃ that exhibit novel properties in regions of the electromagnetic spectrum that are less explored, i.e. long wave IR (LWIR) and mid-wave IR (MWIR) [5-13]. These material systems were selected such that they provide a means to dynamically control electrical and optical properties. We focused on determining the underlying physical mechanisms of tunability, correlating electrical behavior and material properties in the systems, and evaluating fundamental performance limits for Naval applications. This research will provide material systems that will strongly impact future Navy and other DoD development of new types of low power, low loss RF switches, dynamic apertures for sensor tunability and protection (RF to visible), and new types of monolithically integrated optical and electrical sensors. At the end of this report, we also provide a list of publications (23 papers including two book chapters), presentations (20 presentations including 10 invited and 10 contributed) and patent (1) generated from the results achieved under this program.

Use of buffer layer to control the metal-to-insulator transition temperature (T_{MIT}) of VO₂

The first material system in this program was vanadium dioxide (VO₂). VO₂ undergoes a metal-insulator transition (MIT) at ~68 °C, which is associated with a structural phase transition (SPT) between an insulating monoclinic phase and a metallic tetragonal phase when it is triggered by thermal, electrical, or optical stimuli. This SPT leads to a sharp conductivity change by several orders of magnitude. Adjusting the MIT temperature (T_{MIT}) is advantageous from a device standpoint as it allows for the tuning of the device properties. Doping with high-valence metal ions into the VO₂ lattice is a commonly used method to achieve tuning of the T_{MIT} in VO₂ films. However, increasing doping level in VO₂ films generally leads to either a reduction in the magnitude of the transition or broadening of the transition width. Previously, the reduced T_{MIT} was achieved by suppressing the structural transition in ultrathin epitaxially strained VO₂ films (<10 nm) grown on lattice-matched TiO₂ substrates. However, ultrathin VO₂ films are difficult to manufacture due to the easy oxidation to V₂O₅ during the device fabrication process, limiting further applications. In addition, the films become relaxed as they are grown thicker (>10 nm), leading to an increase in the transition temperature. To overcome the strain relaxation issue in thicker VO₂ films, we utilized *buffer layers* to provide an additional degree of freedom to control the strain in thicker VO₂ films.

The first buffer layer we investigated was a tin dioxide (SnO₂), which functions as an excellent buffer because it has the same tetragonal crystal structure and space group as metallic VO₂ (P42/mnm). We have successfully deposited epitaxial VO₂/SnO₂ thin film heterostructures on m-cut sapphire substrates by pulsed laser deposition (PLD). By adjusting the growth conditions of the SnO₂ buffer layers (~150 nm), we were able to control the interfacial strain between VO₂ and SnO₂ buffer layer such that the T_{MIT} of the VO₂ films can be reduced from 68 °C to 52 °C without diminishing the magnitude or sharpness of their transition. It is shown that in-plane tensile strain and out-of-plane compressive strain of the VO₂ films lead to a decrease of T_{MIT} . Additionally, x-ray diffraction data shows that the VO₂ films grown on SnO₂ buffered Al₂O₃ suppress the monoclinic-VO₂ phase, inducing a structural phase transition from tetragonal-like VO₂ to tetragonal VO₂ during the transition. These results suggest that the strain generated by the SnO₂ buffer provides an effective way for tuning the T_{MIT} of VO₂ films. In summary, the

combination of growing thicker films and the suppression of the structural transition will help reduce resistive losses while maintaining sub-picosecond switching times (Figure 1).

RuO₂ was also proposed as a functional buffer because it has the same crystal structure and space group as both TiO₂ and metallic VO₂ and at the same time allows the fabrication of out-of-plane thin-film-based devices. For example, the RuO₂ buffer layer can be used as a bottom oxide electrode for VO₂ based devices with out-of-plane configurations, considerably reducing the switching voltage and current required for operation as compared with VO₂-based planar type devices. Epitaxial VO₂ thin films were deposited on conductive RuO₂/TiO₂ templates by PLD. The MIT temperature (T_{MIT}) of VO₂/RuO₂/TiO₂ heterostructures can be tuned from 59 °C to 24 °C by adjusting the strain state of the films by decreasing the RuO₂ thickness from 50 nm to 10 nm. For comparison, the sheet resistance curves of 100 nm VO₂ films on TiO₂ substrates with and without the RuO₂ buffer layer are plotted in the same figure (Fig. 4c). Without the RuO₂ buffer layer, the T_{MIT} of the VO₂ film increases from 30 °C to 60 °C. The boundaries between RuO₂ structures are favorable regions responsible for the nucleation of dislocations, which can partially relieve the strain in the film, thereby sustaining intermediate strain states even with large thicknesses (~100 nm). This allows lowering of the T_{MIT} to near room temperature. These results suggest that the strain generated by the RuO₂ buffer layer can provide an effective way for tuning the T_{MIT} of VO₂ films and provide a route to realizing out-of-plane electrical switching devices (Figure 2).

Ultrafast transition dynamics in strained VO₂ films

Understanding the dynamic conductivity in VO₂ is of critical importance for both understanding the insulator-metal phase transition (IMT) physics and developing next generation configurable photonic devices. An ultrafast optoelectronic autocorrelation technique was used to directly measure the dynamical electrical properties of epitaxially strained VO₂ across the IMT. Strained and fully relaxed epitaxial films of VO₂ were produced by PLD utilizing different substrates and film thickness: 8 nm thick VO₂ on (001) TiO₂ and 60 nm thick VO₂ on (001) Al₂O₃ substrates. Both strained and fully relaxed films exhibit abrupt change in resistivity, indicating the IMT in each. By tailoring the film growth conditions, film thickness and substrate selection, various strain states can be exploited to reduce the IMT temperature to values ranging between 20 – 80°C [Fig. 3(a)].

We measured the dynamical resistivity using this optoelectronic autocorrelation technique on strained epitaxial films of VO₂ on TiO₂ (001) substrates [Fig.3(b)]. Our approach differs from purely optical methods, such as pump–probe measurements, where electrical properties of the material must be inferred from reflected or transmitted light. While the optoelectronic autocorrelation technique uses a combination of optical excitation and electrical sensing, the method differs significantly from purely electrical VO₂ switching measurements that use high amplitude voltage pulses to initiate the IMT. Purely electrical switching experiments have a relatively low temporal resolution due to the electrical readout, which is typically limited to nanosecond time scales. In contrast to the strain-relaxed VO₂ films on c-cut Al₂O₃, the resistance evolution of the strained VO₂ exhibits an ultrafast response that shows switching of the material to the metallic phase, and a return to the insulating state, lasting < 400 fs [Fig. 3(c)]. This short time scale limits the possibility of a structural change in the VO₂, which typically persists for nanoseconds after ultrafast optical excitation of the IMT. These results suggest that a selective excitation of the IMT is possible and support recent findings indicating the structural component of the

phase transition can be suppressed in epitaxially strained VO₂ films. These findings are particularly relevant for developing optical and optoelectronic devices for applications that require fast, high-contrast switching.

VO₂-based radiators for spacecraft thermal control

Building upon the knowledge gained earlier in the program on the growth of VO₂, we fabricated a multilayered solid-state radiator for spacecraft thermal control, as a demonstration of the applicability of this basic research. The ability to grow continuous high quality VO₂ thin film heterostructures enabled us to experimentally demonstrate a passive radiator for thermal control in a simulated space environment. Two VO₂-based radiator designs were demonstrated during this program for spacecraft thermal control applications.

The first radiator design was based on an inverted multilayer structure. In this design, the VO₂-based thermal radiator, which consisted of a BaF₂ dielectric spacer sandwiched between a VO₂ layer and an Au reflecting layer, achieved passive and switchable thermal emissivity control based on the VO₂ phase transition. By shifting the IR transparent Si substrate to the front of the window, the Si can then shield the multilayer thin film structure from the external environment, including exposure to atomic oxygen flux resulting in much improved device lifetimes. This new approach enabled the fabrication of high-quality thermal radiators for spacecraft applications much less likely to fail due to strain from thermal coefficient mismatch between layers or due to damage from exposure in outer space. When the VO₂ layer was in an insulating state at $T < 340$ K, the multilayer structure behaved like a simple infrared reflector, thereby minimizing radiative heat loss. However, when the VO₂ layer switched to a metallic state at $T > 340$ K, the multilayer structure behaved like a resonant absorber with high emissivity in the mid-infrared region (8-12 μm), thus providing a radiative cooling effect. A quantitative comparison between the emitted power using normal and hemispherical emissivity was performed. The radiator showed an excellent emissivity contrast of $\Delta\epsilon \sim 0.47$ comparable with previously published results using the normal emissivity values. However, using normal emissivity values leads to an overestimate of actual performance because it does not consider the angular dependence of the multilayer radiator design. While the measured hemispherical emissivity change of the radiator was $\Delta\epsilon \sim 0.35$, the experimentally measured radiated power increases by more than a factor 7 with a net radiated power difference of 480 W/m² between 300 K and 373 K (Figure 4).

The second radiator design was VO₂-based radiators using mechanically stable TiN and Al₂O₃ layers. In this design, the layered thin-film radiator consists of a TiN bottom infrared mirror, an Al₂O₃ dielectric spacer and a VO₂ top absorber layer to achieve dynamic thermal emissivity for spacecraft thermal control application. First, an analytical modeling approach was used to optimize the optical response of the layered radiators by varying the thickness of Al₂O₃ spacers and VO₂ layers for maximum emissivity change ($\Delta\epsilon$) between 25 °C and 80 °C. The results of this simulations showed that the thin-film radiators composed of the optimized thickness of VO₂ (30 – 50 nm) and Al₂O₃ (600 – 800 nm) can provide the highest emissivity change ($\Delta\epsilon \sim 0.48$) between these two temperature states. Experimental results validated the simulations for a radiator with a 50 nm thick VO₂ layer and a 600 nm of Al₂O₃ layer which exhibited a maximum emissivity change ($\Delta\epsilon \sim 0.46$) under the same temperature range. Our

experimental results agree very well with the modeling results obtained from the same radiator design. These results are of crucial importance for designing mechanically and thermally stable radiators for spacecraft thermal control due to the stability of both TiN and Al₂O₃ materials. The multilayer radiators in this work are well suited for spacecraft thermal control because they are passive and self-regulating (Figure 5).

Transparent conducting perovskite oxides for Near-IR (NIR) and Mid-IR (MIR) plasmonics

The second material system in this program was La-doped BaSnO₃ (LBSO). We focused on growth of epitaxial LBSO thin films that can serve as a transparent electrode for studying ferroelectric transitions in perovskite films, such as BaTiO₃. We deposited epitaxial LBSO thin films by PLD on two different substrates (SrTiO₃ and MgO). As a result of this work we were able to tune the permittivity over a mid-infrared range (2 μm to 6 μm) that enables the use of LBSO films as an electrically conducting, optically transparent buffer layer for active phase change based infrared plasmonic devices.

First, we completed our study on the epitaxial growth of LBSO thin films on (001) SrTiO₃ substrates by PLD. Specifically, we studied the tunable permittivity of La-doped BaSnO₃ (LBSO) epitaxial films in the mid-infrared range. By adjusting the oxygen pressure and substrate temperature during deposition, we were able to control the film crystallinity and strain, which modified the electrical and optical properties of the LBSO films. The LBSO films grown at an oxygen pressure of 100 mTorr and a growth temperature of 780 °C show the highest conductivity ($3.6 \times 10^3 \text{ S cm}^{-1}$) with a carrier concentration of $3.5 \times 10^{20} \text{ cm}^{-3}$ and a carrier mobility of $65 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$. The observed high conductivity corresponds to the film with the best crystallinity and the lowest strain (<0.2 %) state. The permittivity of the LBSO films can also be tuned as a function of the oxygen pressure and temperature during deposition allowing tuning their epsilon-near-zero (ENZ) wavelength from 2 μm to 5.6 μm (Figure 6).

In contrast to the LBSO films grown on perovskite SrTiO₃ substrates, we also investigated epitaxial LBSO films on non-perovskite MgO substrates. MgO was selected due to its low dielectric constant, which is much more efficient for electro-optic modulators operating at microwave frequencies. We were able to control the film crystallinity and strain by changing the oxygen partial pressure during deposition, which in turn enabled modifying the electrical and optical properties of the films. The LBSO films grown at oxygen pressures of 100 mTorr show reduced strain and low resistivity ($4.4 \times 10^{-4} \text{ } \Omega\text{-cm}$), while the films grown at 10 mTorr of oxygen show increased strain and high resistivity ($8.2 \times 10^{-2} \text{ } \Omega\text{-cm}$). Optical properties of the LBSO films were also adjusted by varying the oxygen deposition pressure. The average transmittance in the visible range is greater than 70% for both films. The optical band gap of the LBSO films increased from 3.81 eV to 3.87 eV with increasing oxygen deposition pressure from 10 mTorr to 100 mTorr. The permittivity of the LBSO films was modified as a function of the oxygen pressure during deposition allowing tuning their epsilon-near-zero (ENZ) wavelength from 2.2 μm to 7 μm. This large tunable permittivity in the mid-IR range makes LBSO films strong candidates as an electrically conducting, optically transparent buffer layer for MIR plasmonic devices that are highly relevant to various Navy and DoD applications ranging from optical communications to low observables (Figure 7).

Ti₂O₃ films

The third material system studied was Ti₂O₃. Unlike VO₂ or V₂O₃ phase changing materials, Ti₂O₃ shows the metal-insulator transition (MIT) electronically without a structural phase transition. Thus, the timescale of the electronic MIT in Ti₂O₃ can be much faster than the MIT induced by the structural changes (such as VO₂), resulting in ultrafast switching responses to the external stimuli. These unique properties can be used in switching devices for numerous applications, including IR detection and thermal energy conversion. We deposited Ti₂O₃ thin films on single crystal c-Al₂O₃ substrates via PLD. No additional processing gases were introduced during growth to maintain low pressure and eliminate oxygen from the system. Ti₂O₃ films were grown at both 485°C and 730°C resulting in two different structural phases. The resulting films, grown at different temperatures, were determined to have different carrier types, with the low temperature (LT) film (485°C) being p-type and the higher temperature (HT) film (730°C) being n-type. The ability to grow Ti₂O₃ films with different carrier types opens up the possibility for p-n homojunctions. (Figure 8).

Furthermore, the XRD and Raman data indicated that the LT-Ti₂O₃ films had an elongated c-axis and the electrical and optical measurements demonstrated that these films showed metallic conduction that did not transition to insulating at any temperature. At room temperature, the optical conductivity of these films matched that of the bulk high temperature metallic Ti₂O₃ as did the position of the Raman A_{1g} mode. This indicated that our film was strain-locked in the metallic state at all temperatures. A likely explanation for this lattice deformation is due to the grain structure of the films. In Ti₂O₃ nanoparticles, it has been reported that there is an increase in the c-axis and a decrease in the a-axis length as the nanoparticle size is reduced. The grain size in our films is approximately 15 – 33 nm, which should yield a (c/a) ratio of ~2.69 based on the previous results on Ti₂O₃ nanoparticles. This is consistent with our observed c/a ratio of 2.7. In bulk Ti₂O₃, this c/a ratio occurred at T ~550 K, well into the metallic state.

From our data it is clear that our films were strained to a metallic (c/a) ratio at room temperature due to the grain size causing local c-axis elongation and a-axis contraction. This local strain was preventing the shorting of the c-axis to a point where the Ti³⁺-Ti³⁺ dimers can fully isolate to a $a_{1g}a_{1g}^*$ singlet while also keeping the a-axis contracted to allow the conduction along the a-b plane through the e_g^π band. This grain size induced strain locks the film into a metallic state and prevents the bandgap from opening at all temperatures. (Figure 9).

Laser surface structuring of metal oxide thin films

We have studied laser induced crystallization and surface structuring of metal oxide thin films. In contrast to the use of epitaxial strain where the strain is set by the lattice constants at the unit cell level, laser induced crystallization and surface structuring allow for spatially varying morphologies and access to thermodynamic conditions that are outside of conventional growth techniques. Rather than relying on conventional excimer laser-based crystallization methods, we investigated the coherent interference of two pulsed laser beams, also referred to as direct laser interference patterning (DLIP). This technique allows for control of the spatial period by adjusting the interference angle. We demonstrated highly ordered linear crystallization in indium tin oxide (ITO) thin films, which yielded a spatial period of 650 nm. Furthermore, unlike DLIP, a single laser beam can also produce periodic patterns with sub-

micrometer feature sizes, also referred to as laser-induced periodic surface structures (LIPSS). We were able to produce the LIPSS structure (a period of 75 nm) on the ITO films (Figure 10).

We utilized a UV picosecond laser pulses to produce simultaneous DLIP and LIPSS features in a single laser step on ITO films (~150 nm). Results show that the DLIP formation is more dominant than the LIPSS formation when high laser fluences were applied, while the LIPSS formation is dominant at low fluences. By utilizing optimized laser conditions (laser polarization, fluence, number of laser pulses, pulse duration) of a picosecond laser system, we were able to produce both LIPSS and DLIP structures of the ITO films. The combination of picosecond laser pulses and interfering beams allowed for hierarchical patterning in ITO thin films with feature sizes spanning two orders of magnitude, from 75 nm up to 50 μm , which is the size of the laser spot. After laser patterning, the ITO thin films exhibited an anisotropic electrical resistance of 50,000:1 and a strong polarization contrast in the LWIR (Figure 10).

Future Plans

Perovskite oxides such as BaTiO_3 have become attractive materials to extend the functionalities of silicon-based photonics platforms due to their large Pockels coefficient ($> 1000 \text{ pm/V}$). For example, BaTiO_3 -based integrated photonics potentially provide a wide spectral operating range (visible to mid-IR), higher power handling, rapid frequency tunability, low linear losses and nonlinear behavior. The epitaxial film growth and defect/ strain engineering approaches developed under this WU led to a new program (WU# 1AC4: Functional Materials for Integrated Photonics) that will identify the basic material properties of the perovskite oxides as new functional materials for integrated photonics for broadband application (UV to SWIR). In addition, we plan to pursue further research on other oxide systems including thermochromic VO_2 , phase transition Ti_2O_3 and perovskite conducting oxides for improving their functionality in various applications under WU# 1AC4.

Figures

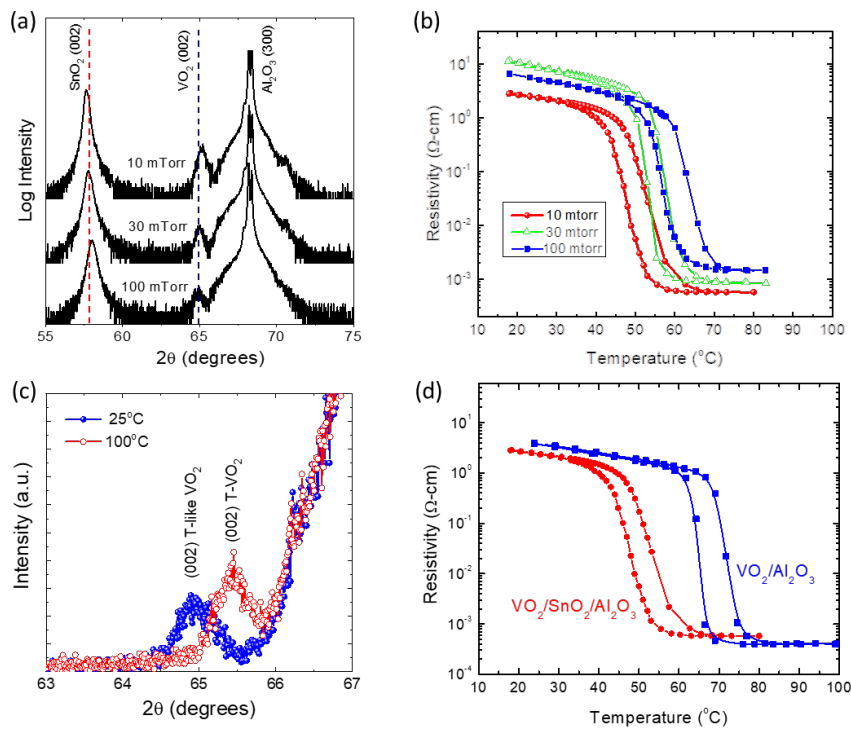


Figure 1. (a) θ -2 θ XRD scans and (b) electrical resistivity vs temperature of VO₂/SnO₂ thin films (~50 nm) grown on m-cut sapphire substrates, where the SnO₂ buffer layers were grown at various oxygen pressures (10-100 mTorr). The red broken line represents the bulk SnO₂ (002) peak position and the blue broken line represents the bulk VO₂ (002) peak position in (a). (c) Temperature-dependent θ -2 θ XRD scans of VO₂ film on 30 mTorr SnO₂/Al₂O₃ buffer layer at 25 °C and 100 °C. (d) Electrical resistivity vs temperature for 50 nm VO₂ film on SnO₂/Al₂O₃ (red) and 50 nm VO₂ film on Al₂O₃ (blue). [Adapted from Ref. 5].

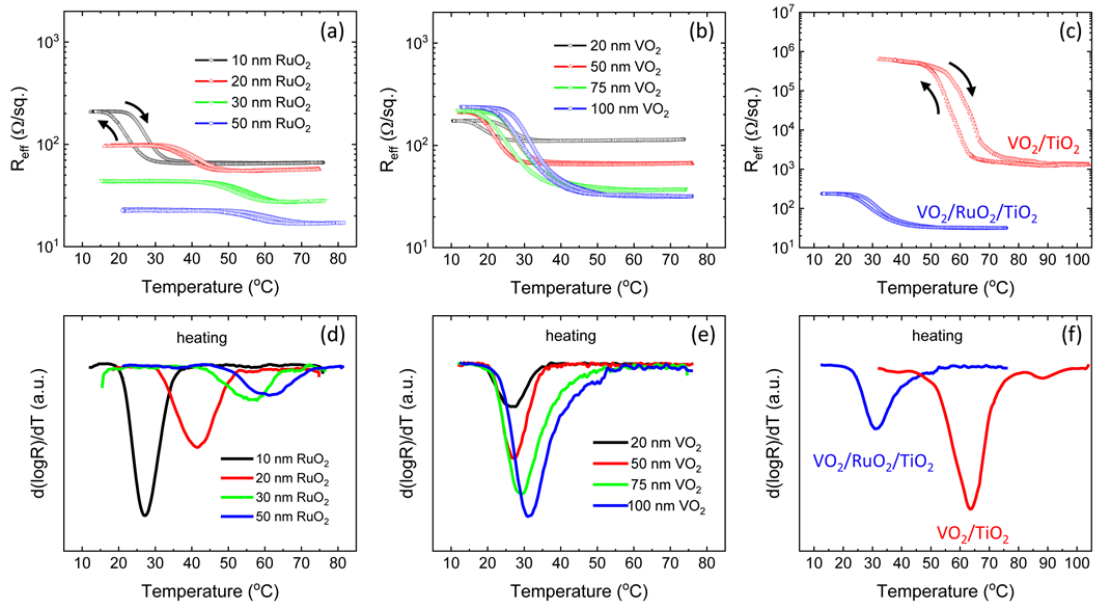


Figure 2. (a) Effective sheet resistance vs temperature for 50 nm thick VO_2 films grown on $\text{RuO}_2/\text{TiO}_2$ as a function of RuO_2 buffer layer thickness (10 – 50 nm). (b) Effective sheet resistance vs temperature for various thicknesses of VO_2 films (20, 50, 75 and 100 nm) grown on RuO_2 (10 nm)/ TiO_2 templates. (c) Effective sheet resistance vs temperature for 100 nm VO_2 films grown on TiO_2 substrates with /without RuO_2 buffer layer. Arrows show measurement direction. (d) - (f) Corresponding derivative curves during heating process for plots in (a), (b) and (c), respectively. [Adapted from Ref. 6].

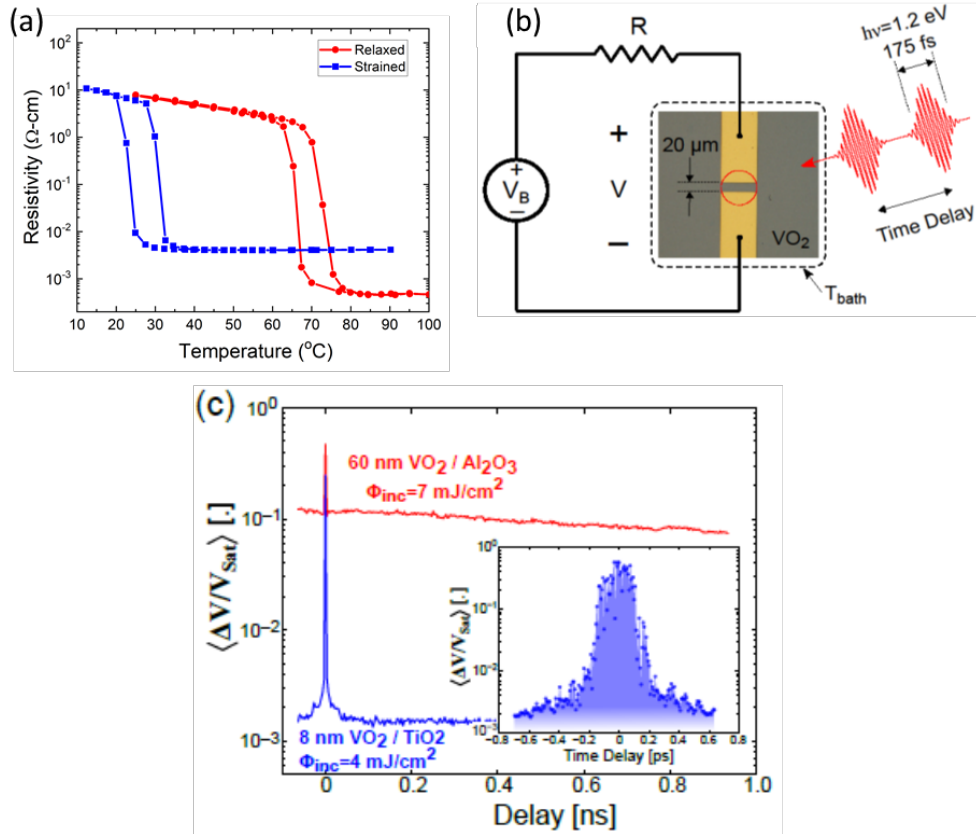


Figure 3. Resistivity of relaxed (red) and strained (blue) VO_2 films as a function of temperature. (b) Schematic showing the autocorrelation measurement and an optical micrograph of the VO_2 channel formed between two gold contacts. (c) Average normalized change in voltage across the VO_2 channel plotted as a function of time delay for strained (blue) and relaxed (red) films. The inset shows detail of the strained film response around zero-time delay. [Adapted from Ref. 7].

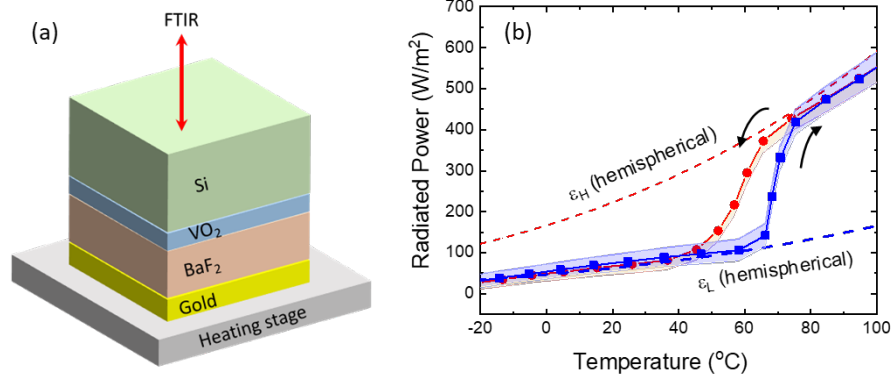


Figure 4. (a) Schematic illustration of the multilayer structure mounted upside down to the temperature-controlled stage inside an FTIR microscope. (b) Radiated thermal power of a multilayer structure with a 60 nm thick VO₂ layer. The direction of heating and cooling are indicated by arrows. [Adapted from Ref. 8].

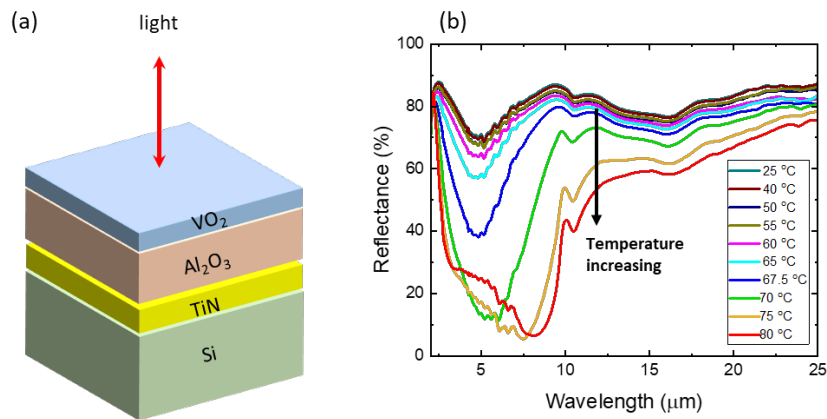


Figure 5. (a) Schematic illustration of the thermal radiator. (b) Experimental reflectance spectra of VO₂-based multilayer structures at various temperatures from 25°C to 80°C during heating. The thickness of VO₂, Al₂O₃, and TiN layers is 50 nm, 600 nm and 70 nm, respectively. [Adapted from Ref. 9].

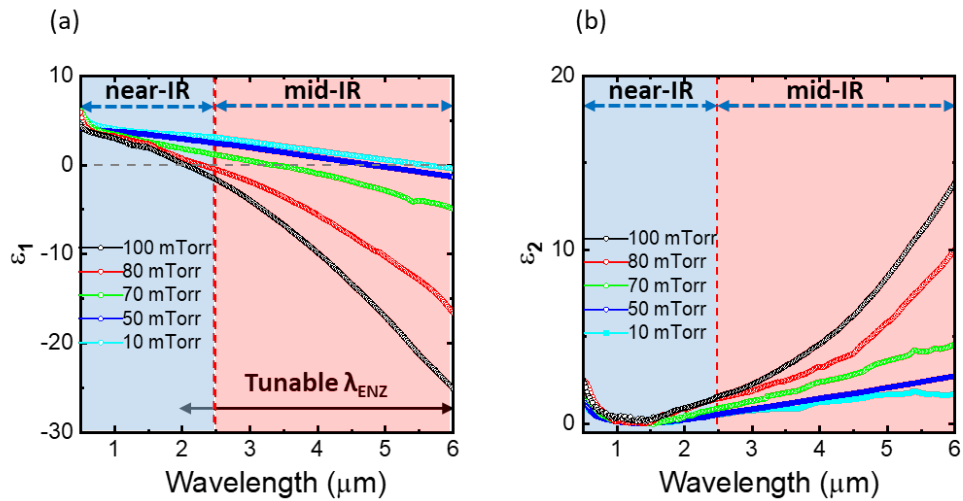


Figure 6. (a) Real (ϵ_1) and (b) imaginary (ϵ_2) parts of permittivity of 100 nm thick La-doped BaSnO₃ (LBSO) films deposited on SrTiO₃ substrates at various oxygen pressures (10 - 100 mTorr). The permittivity of the LBSO films can be tuned as a function of the oxygen pressure allowing tuning their epsilon-near-zero (ENZ) wavelength from 2 μm to 5.6 μm . [Adapted from Ref. 10].

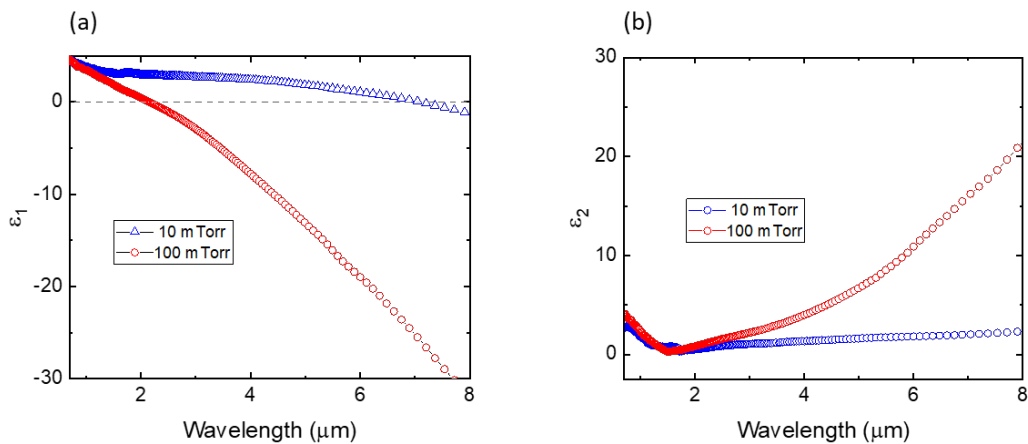


Figure 7. (a) Real (ϵ_1) and (b) imaginary (ϵ_2) parts of permittivity of 100 nm thick La-doped BaSnO₃ (LBSO) films deposited on MgO substrates at two different oxygen pressures (10 and 100 mTorr). The permittivity of the LBSO films can be tuned as a function of the oxygen pressure allowing tuning their ENZ wavelength from 2 μm to 7 μm . [Adapted from Ref. 11].

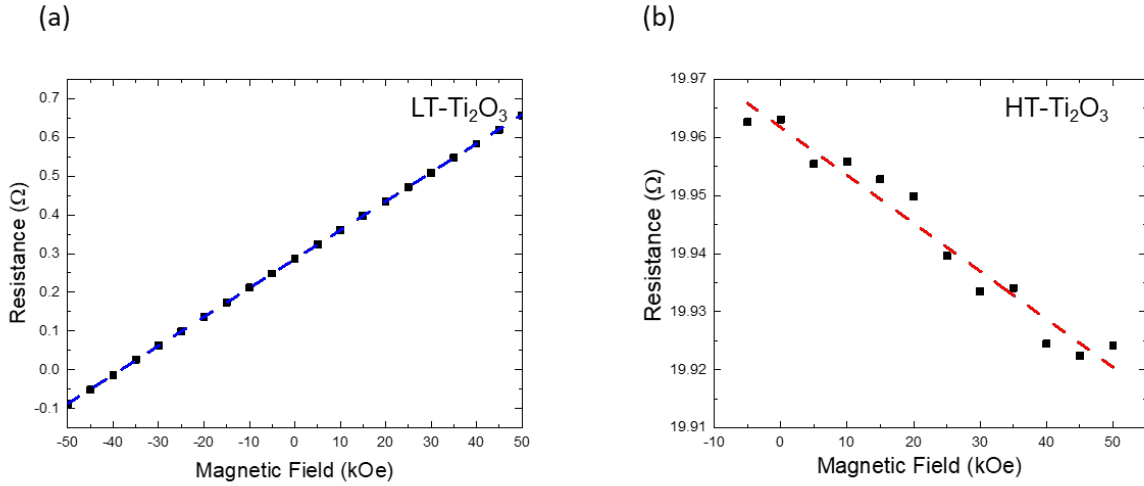


Figure 8. Resistance vs magnetic field of Ti_2O_3 films grown at two different temperatures: (a) 485 $^\circ\text{C}$ (LT) and (b) 730 $^\circ\text{C}$ (HT). LT- Ti_2O_3 film yields trigonal (corundum) p -type film, while HT- Ti_2O_3 film yields an orthorhombic n -type film.

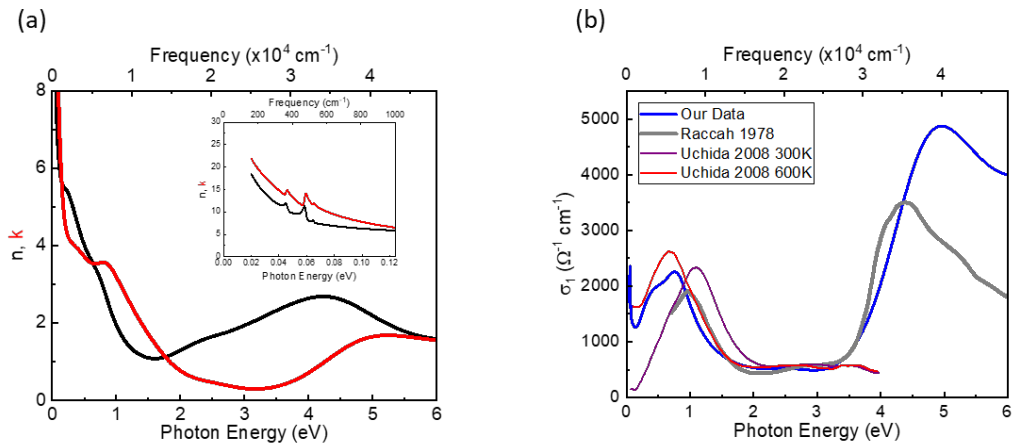


Figure 9. (a) Optical constants (n , k) and (b) broadband optical conductivity of 140nm LT- Ti_2O_3 films deposited on $c\text{-Al}_2\text{O}_3$ substrate at 485 $^\circ\text{C}$. Inset in (a) shows the zoomed in region in the far infrared showing the Drude feature from free carrier contributions. [Adapted from Ref. 12].

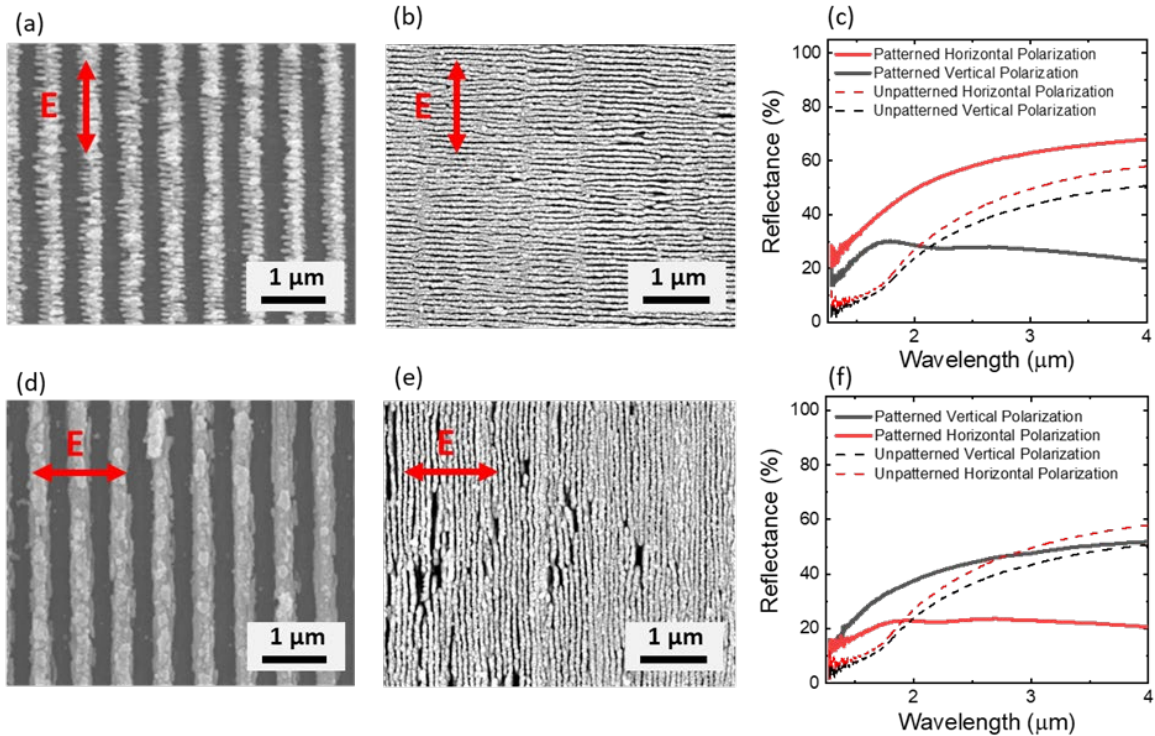


Figure 10. SEM images showing combination of LIPSS and DLIP structures in ITO films irradiated with (a,d) an intermediate laser fluence (90 mJ/cm^2 , 200 kHz and 5 mm/s) and (b,e) a low laser fluence (32 mJ/cm^2 , 500 kHz and 0.1 mm/s) with an electric field parallel (a,b) and perpendicular (d,e) to the DLIP. (c,f) Polarized infrared reflectance of laser-structured ITO films with (b) horizontal and (e) vertical LIPSS structures. [Adapted from Ref. 13].

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