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NEUTRALIZATION OF ORGANIC SUBSTANCES IN WASTE WATER BY PLANTS (--ETC(U)  
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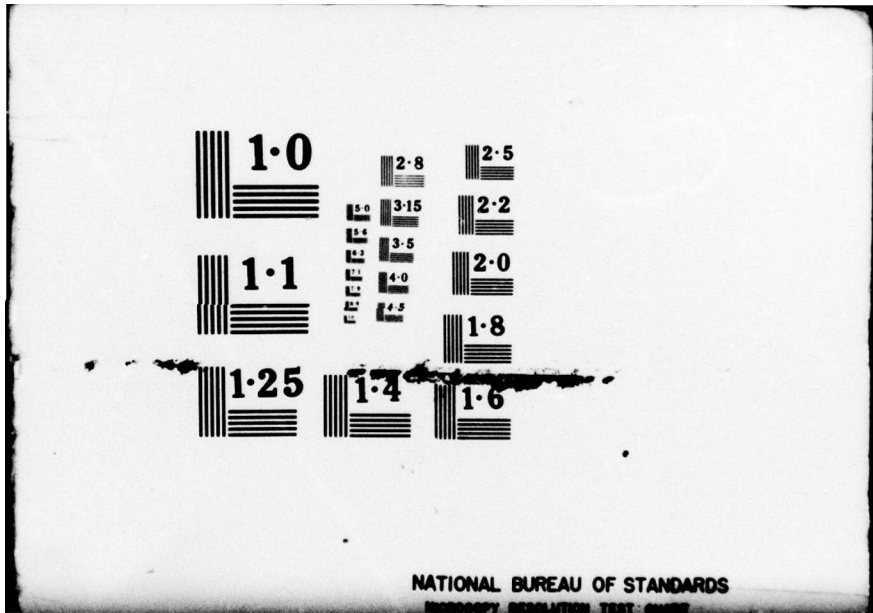
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## NEUTRALIZATION OF ORGANIC SUBSTANCES IN WASTE WATER BY PLANTS

Kupavna OBEZVREZHIVANIYE ORGANICHESKIKH VESHCHESTV STOCHNYKH VOD V RASTENIYAKH  
in Russian 1975 pp 3-7

The question of the purification of waste water is becoming ever more important in connection with the fast development of the chemical industry. Artificial biological irrigation with the help of water microorganisms is the most widespread method for the purification of waste water contaminated by organic matter. Thanks to the existence of a great variety of microorganisms in the soil, irrigation fields could also be used as a natural method for the biological purification of waste water. Here, along with the protection of water reservoirs from pollution, we resolve the problem of obtaining high crop yields. The danger arises, however, that the organic matter coming from the waste water may accumulate in the plants.

The works of Yu. V. Rakitin at the Institute of Plant Physiology of the USSR Academy of Sciences, as well as a number of other works published domestically and abroad, have proved that, penetrating the plants, herbicides and other phytotoxic substances may disturb one or another metabolic process in the course of which, however, they themselves become detoxified. Only in rare cases do compounds of higher toxicity develop as a result of the toxic substances introduced into the plants through biochemical processes. However, should the plant survive the existence of new toxic substances is only temporary, for, involved in the metabolic processes, they inevitably lose their toxicity.

The VNIISV (L. Ye. Kutepov and S. I. Khramova) has studied the neutralization of organic matter in the plant using the highly sensitive method of gas chromatography with a flame-ionizing detector.




It was determined that plants play an important role in the biological neutralization of organic matter in the utilization of waste water on irrigated fields. The organic substances introduced with the irrigation water were neutralized in the plants, depending on their chemical structure, over different time periods ranging from 1 to 16 days (Table 1). As a rule, the neutralization process in root and tuber crops developed somewhat more slowly than in the above-ground part of the plants. That is why it would be desirable to restrict the use of waste water for irrigation to root and tuber crops.

Aniline, acetone, benzaldehyde, dimethylamine, diethylamine, crotonaldehyde, furfural, cyclohexanol and cyclohexanone were totally neutralized in the plants within a 5-day period; acetaldehyde, benzene, butanol, p-xylene, methanol, n-propanol, toluene, and ethanol were neutralized in 5 to 10 days; caprolactam, dichlorethane, and carbon tetrachloride were neutralized in 10 to 16 days.

In the presence of such organic substances in waste water used for irrigation we must observe the quarantine period between the last watering and the harvest ranging from 2 to 4 weeks depending on the nature of waste water pollution.

Table 1

Name of substance and chemical formula	MPC b.p.s.* mg/liter	Concentration in irrigation water, mg/l	Irrigation norm	Plant type	Period of total decomposition in plant, days
1	2	3	4	5	6
Aniline $C_6H_5NH_2$	0.5	300	50	Corn Grasses	1 1
Acetaldehyde $CH_3CHO$	1000	100	100	Corn Grasses Potatoes (tubers) Potatoes (plants)	8 8 3 4
Acetone $CH_3COCH_3$	800	300	50	Corn Grasses Leguminous grasses	2 2 2
Benzaldehyde $C_6H_5CHO$	-	100	100	Corn Grasses Potatoes (tubers)	3 3 6
Benzene $C_6H_6$	100	100	70	Corn Grasses Carrots (root crop)	5 4 6
Butanol $CH_3(CH_2)_2CH_2OH$	420	100	70	Corn Grasses Celery (plant) Celery (root)	6 3 7 8

1	2	3	4	5	6
Dimethylamine (CH <sub>3</sub> ) <sub>2</sub> NH	0.7	300	70	Corn Grasses	1 1
Dimethyldioxane 	-	500	200	Corn	4
Diethylamine (C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub> NH	10	300	70	Corn Grasses	1 1
Caprolactam 	100	500	200	Corn	16
Crotonaldehyde CH <sub>3</sub> CH=CHCHO	250	100	100	Corn Grasses Potatoes (tubers)	2 2 5
p-Xylene C <sub>6</sub> H <sub>4</sub> (CH <sub>3</sub> ) <sub>2</sub>	1.0	100	100	Corn Grasses Carrots (root crop)	5 4 7
Methanol CH <sub>3</sub> OH	200	100	70	Corn Celery (root) Celery (plant)	2 7 3
n-Propanol CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> OH	62	100	70	Corn Grasses Celery (root) Celery (plant)	4 3 10 9
Toluene C <sub>6</sub> H <sub>5</sub> CH <sub>3</sub>	200	100	70	Corn Grasses Carrots (root)	5 3 6
Furfural 	1.0	400	50	Corn Grasses	2 2
Cyclohexanol C <sub>6</sub> H <sub>11</sub> OH	1.0	300	50	Corn Grasses Leguminous grasses	4 3 3

1	2	3	4	5	6
Cyclohexanone $C_6H_{11}O$	50	300	50	Corn Grasses Leguminous grasses	5 2 4
Dichlorethane $C_2H_4Cl_2$	200	100	50	Grasses Celery (plant) Celery (root)	13 8 12
Carbon tetra- chloride $CCl_4$	50	100	50	Grasses Celery (plant) Celery (root)	13 16 16
Ethanol $C_2H_5OH$	—	L))	&)	Corn Grasses Celery (plant) Celery (root)	5 4 5 10

\*Remark: MPC b.p.s. stands for maximum admissible concentration of matter in artificial biological purification systems.