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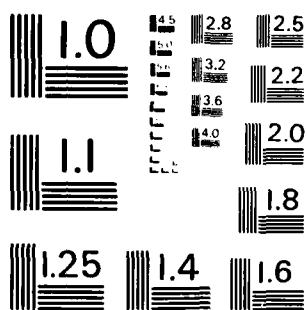
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**MICROSTRUCTURAL ORIGINS OF HOT SPOTS IN RDX EXPLOSIVE
AND SEVERAL REFERENCE INERT MATERIALS**

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W. L. ELBAN, C. S. COFFEY, AND R. W. ARMSTRONG*
RESEARCH AND TECHNOLOGY DEPARTMENT

AND

K. - C. YOO* AND R. G. ROSEMEIER*

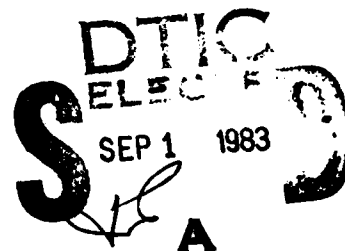
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MARCH 1983

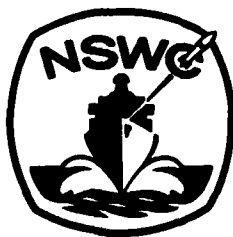
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20. ABSTRACT (Continue on reverse side if necessary and identify by block number) Hardness testing is being used to locally deform RDX explosive and several selected reference inert crystals in a controlled manner in order to investigate the microstructural origins of hot spots. The plastic deformation in RDX was found to be very inhomogeneous. The strain fields around Knoop hardness impressions in a reasonably perfect laboratory- grown crystal were studied using surface reflection Berg-Barrett		

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topography. Highly localized strain fields were observed, and a considerable anisotropy in Knoop hardness number was measured. A substantial variation in Vickers hardness numbers was obtained for several Holsten production-grade Class D RDX crystals. This is indicative of this material's inhomogeneous internal structure; perhaps most notable is the presence of numerous pores observed by microscopic examination.

Spherical ball hardness experiments at various applied loads were performed on a single crystal of MgO. Examination of the strain fields surrounding the resultant impressions by surface reflection Berg-Barrett topography revealed that, for increasing applied loads, the size of the strain fields was controlled by cracking. The strain fields around hardness indentations placed in a single crystal of NaCl were also investigated, in collaboration with other researchers, by synchrotron X-Ray topography.

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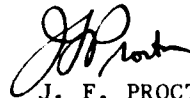
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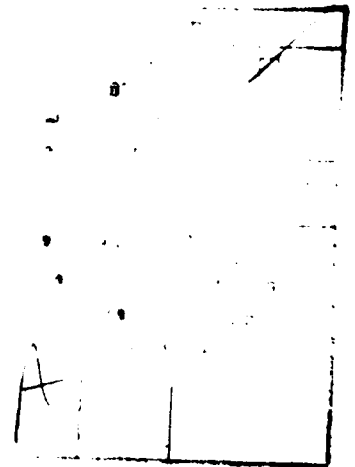
This work was sponsored by the Office of Naval Research under work request numbers N00014-82-WR-20129 and N00014-82-K-0263 as a cooperative effort between NSWC, White Oak, and the University of Maryland, College Park. The results and conclusions presented in this report concerning the microstructural characterization of deformed RDX explosive and selected reference inert (MgO and NaCl) crystals should be of interest to those studying plastic deformation and fracture in these materials. In particular, this work provides insight into their ability to locally concentrate energy as a result of being plastically deformed in a controlled manner. A listing of references appears on several pages after the body of the report.

Dr. J. W. Cleland of the Solid State Division of the Oak Ridge National Laboratory supplied the MgO crystal.

Approved by:



J. F. PROCTOR, Head
Energetic Materials Division



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INTRODUCTION

It is well known that for initiation to occur in a solid explosive under impact conditions, the energy transferred must be concentrated into small volumes of the explosive. The most widely-held view explaining this phenomenon involves the formation of "hot spots" as a result of the explosive experiencing a mechanical stimulus. Heat is generated within a fixed volume at a sufficient rate to cause the temperature to rise very rapidly, the kinetics being limited by the thermal conductivity to the surrounding medium. A number of mechanisms, including adiabatic compression of small entrapped bubbles of gas, friction (between explosive particles and between explosive and impact tools), shear deformation, and fracture, have been envisioned as leading to hot spot formation in crystalline explosives.^{1,2}

The role that localized behavior plays in the impact initiation of explosives has been investigated by striking single crystal targets (explosive) with tiny spherical particles (inert).³ Initiation resulted when critical conditions for particle size and velocity were exceeded. A nice feature of this experiment is the ability to examine and characterize the deformation that resulted when the impact conditions were slightly below the critical conditions necessary for initiation. Typically, the impact sites were smooth plastic indentations and few fractures were found. It was observed further that the deformation was highly localized in narrow bands of material, which were proposed to form by adiabatic shear. It was also suggested that the mechanism responsible for initiation was the production of high local temperatures as a result of the adiabatic shear deformation.

It is consistent with dislocation theory concepts that deformation in crystalline solids occurs by localized deformation twinning or along slip planes that form into local shear bands. Similar inhomogeneous deformation behavior occurs for partially crystalline or even amorphous material. The bulk of the material, which is not in the shear bands, remains undeformed. A dislocation description of localized shear deformation that occurs within a model slip band indicates that appreciable heating can occur where a pile-up is released suddenly in an avalanche mode.⁴ To date, no temperature measurements have verified directly that shear bands are hot spot sites. However, this may be inferred from related experiments. It has been reported that the temperature at the tip of a propagating crack, where highly localized plastic deformation is known to be occurring, is in excess of 500°K for polymethylmethacrylate,⁵ in excess of 3200°K for glass, and about 4700°K for quartz.⁶

In an effort to determine the temperatures generated when various energetic materials are impact loaded, a limited study⁷ was performed on HMX (cyclotetramethylenetetranitramine) explosive using a heat sensitive film as a detector. Substantial deformation and fracture accompanied by discoloration of the film (calibrated as indicating temperatures greater than 220°C for the impact loading conditions of this work) occurred when a single crystal of HMX (approximately 900 μ m) was impacted from 5 cm, but no reaction resulted. At a 10 cm drop height, enough heating took place to cause an ignition reaction which died out. The reaction occurred in the most heavily deformed region of the crystal. Very little heating resulted when a 30 mg pile of Class A HMX was impacted at a 10 cm drop height.

For these experiments the energy imparted to the explosive sample was insufficient to heat the entire sample to its ignition temperature. However, it is clear that large temperature rises can result when explosive crystals, such as HMX, are deformed on impact, and that this heat production must be localized, i.e., hot spots are formed. It is also clear that the heat generation is related to the deformation that takes place in the explosive crystal. The mechanics of deformation in explosive crystals remains to be determined, along with answering how the actual resultant crystal defect structure relates to hot spot formation.

Individual explosive crystals themselves are not typically of convenient size for many of the well-defined deformation experiments normally performed on other types of materials. Microhardness testing⁸ provides a controlled way of locally deforming these crystals offering information concerning the active slip systems and the degree of plastic anisotropy. Normally, two types of indenters are employed: Vickers or diamond pyramid and Knoop. The Vickers indenter has an apex angle of 136°. The Knoop indenter is more blunt than the Vickers; a diagonal indentation results that is seven times longer in one direction than in the other. It also yields a more shallow indentation which makes Knoop testing very suitable for studying brittle materials such as explosive crystals. It is the Knoop test that is used to investigate⁹ plastic anisotropy.

In previous work,^{10,11} the degree of localized plastic deformation was determined for RDX (cyclotrimethylenetrinitramine) explosive crystals grown at NSWC. Microhardness indentations were placed on various growth facets of several large (a few mm) size crystals that were grown from solution in acetone by evaporation at room temperature. Both Knoop and Vickers indentations were made at a 50g load with the orientation of the long axis of the Knoop indenter being allowed to vary with respect to specific surface directions in the growth facets. For the combined measurements on all of the growth facets, a considerable variation in Knoop hardness number was obtained (values ranged from 17 to 71 kgf/mm²), suggesting a limited number of operative slip systems. An etch pit technique¹² was used to determine the extent of the plastic zone or strain field associated with each indentation. Highly localized etch pit arrays centered on the indentations were observed with the surface area of the plastic zones being only about 11 times greater than the projected area of the indentations. In a preliminary comparative study of production-grade (Holsten) Class D RDX crystals (usually 1-2 mm), Knoop hardness numbers compared well with values obtained for laboratory-grown crystals.

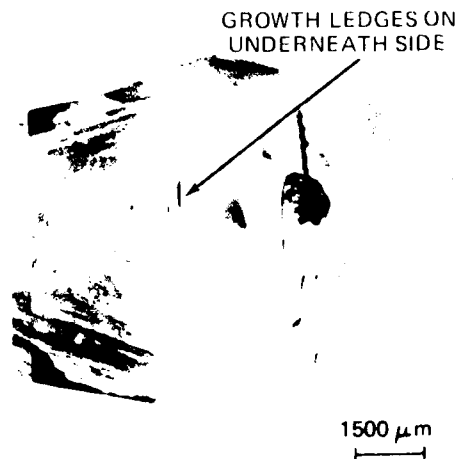
The objective of this current work is to further elucidate the fundamental microstructural reasons for hot spots being generated during the deformation of crystalline energetic and inert materials. The deformation behavior of RDX, the most common ingredient in Navy explosives, is being investigated at NSWC. Emphasis is being given to comparing various laboratory-grown crystals having different degrees of microstructural perfection and production-grade crystals. This work is closely allied with a companion research effort on selected model inert crystals at the University of Maryland, College Park.

CHARACTERIZATION AND DEFORMATION OF RDX

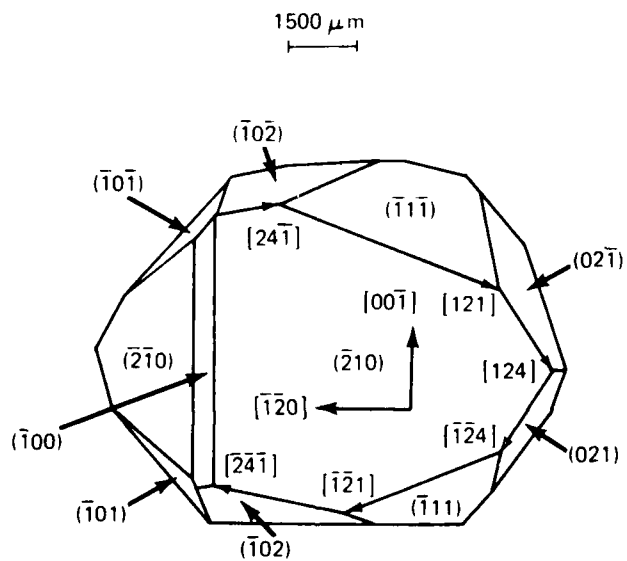
A large laboratory RDX crystal (Figure 1(a)) grown by Dr. R. Y. Yee (Naval Weapons Center, China Lake, CA) was studied using Laue and Berg-Barrett X-ray diffraction techniques. This crystal was grown by slow evaporation from acetone solution using a seed; the starting material was recrystallized Holsten production-grade crystals chemically manufactured¹³ by the nitration of hexamethylenetetramine. The morphology of this crystal, as determined from a zone analysis of its Laue back-reflection photograph, was found to be the same as for the non-standard morphology of some of the laboratory RDX crystals grown^{10,11} previously at NSWC. Various growth surfaces and crystal directions are specified in Figure 1(b). The degree of microstructural perfection of this crystal is significantly improved compared to crystals grown at NSWC, as evidenced by the absence of optical dispersion. In addition, most of the diffraction spots on the Laue photograph were sharp. However, some streaking for a few spots was observed. The lines visible in Figure 1(a) are associated with growth ledges on the bottom of the crystal. The overall crystal thickness was ~ 5 mm.

Preparatory to studying systematically the deformation properties of Dr. Yee's crystal, considerable effort was devoted to its microstructural characterization by surface reflection Berg-Barrett topography. This technique (Figure 2(a)) involves recording the structure in an individual Laue back-reflection spot under appropriate geometrical conditions on a fine-grained nuclear emulsion plate or film.¹⁴ Local variations in diffracted intensity (Figure 2(b)) can be associated directly with internal strains in the crystal surface layer from which the X-ray beam is diffracted (i.e., the extinction depth).¹⁵ The use of Berg-Barrett topography for examining the microstructure of crystalline materials has been discussed in detail.^{14,16,17}

Prior to obtaining the first topograph, two Knoop indentations (50 g load) were placed (Figure 3(a)) in the top portion of the $(\bar{2}10)$ growth surface to serve as recognizable features that would offer some variation in contrast in the topograph. The long axis of the indenter was aligned parallel to $[00\bar{1}]$. For one indentation, the $(\bar{2}10)$ surface fully supported the indenter. For the other indentation, the indenter was only partially supported by the $(\bar{2}10)$ surface, and a large crack resulted. No particular care was taken to insure orthogonality between the applied force axis of the indenter and the $(\bar{2}10)$ surface. The strain fields of these two indentations are readily apparent in the topograph for the $(\bar{7}2\bar{1})$ reflection appearing in Figure 3(b). The enhanced

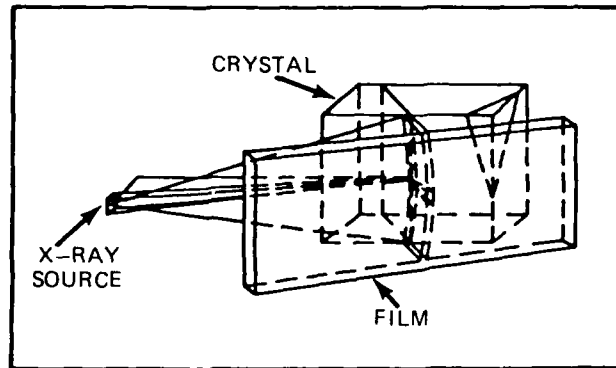


(a) LIGHT (ZEISS TESSOVAR) PHOTOGRAPH

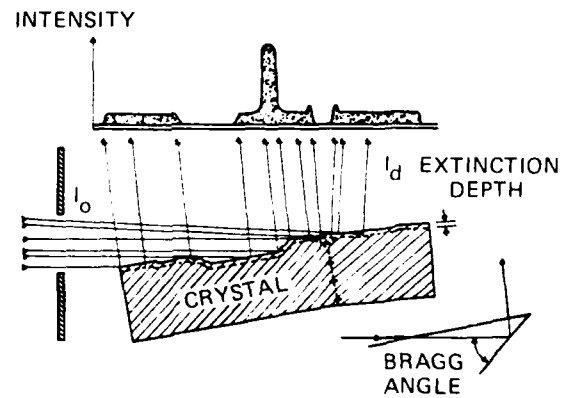


(b) SCHEMATIC IDENTIFYING EXPOSED GROWTH SURFACES AND DIRECTIONS

FIGURE 1. LABORATORY-GROWN CRYSTAL

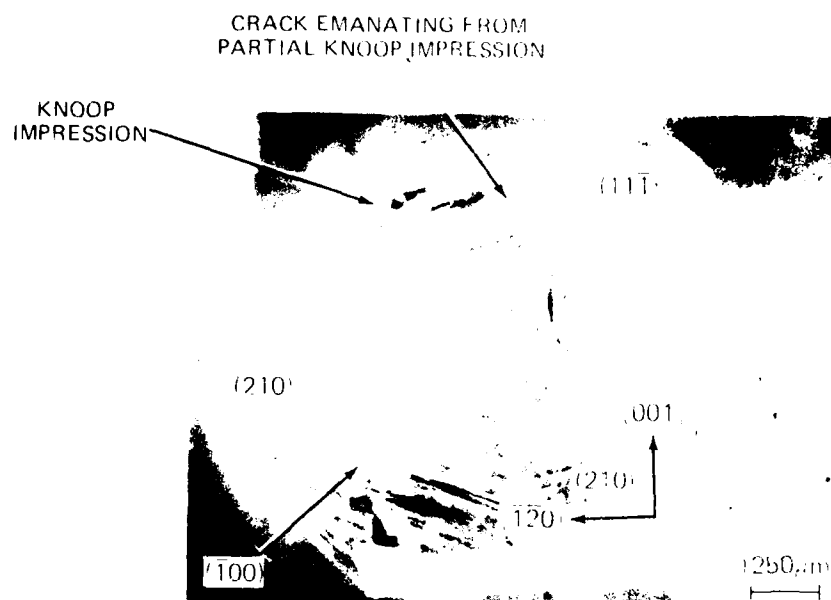


(a) EXPERIMENTAL CONFIGURATION
(AFTER REFERENCE 14)

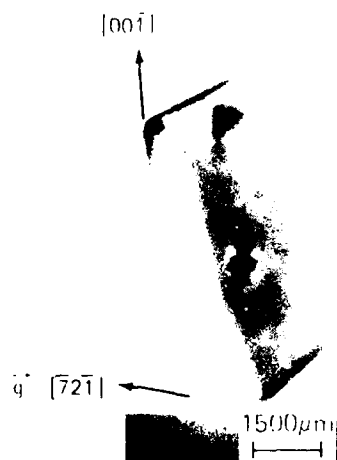


(b) CRYSTAL IMPERFECTION CONTRAST EFFECTS
(AFTER REFERENCE 15)

FIGURE 2. SCHEMATIC OF SURFACE REFLECTION BERG-BARRETT TOPOGRAPHY TECHNIQUE



(a) LIGHT (ZEISS TESSA 4A) PHOTOGRAPH



(b) X RAY SURFACE REFLECTION TOPOGRAPH

CuK α RADIATION AT 14 kV AND 20mA FOR
3 h ON ILFORD L4 50 μ m NUCLEAR PLATE

FIGURE 3. LABORATORY-GROWN RDX CRYSTAL AND BERG-BARRETT TOPOGRAPH
(721) REFLECTION)

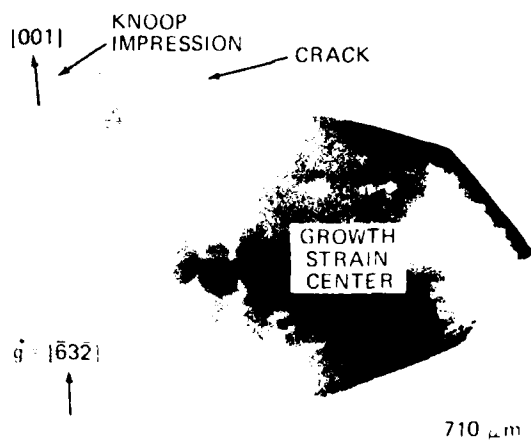
diffracted intensity occurs because of the cumulative residual dislocation strain fields that resulted from accommodating the indenter. Also visible in this topograph is a large growth strain center associated with grown-in dislocations emanating from the seed crystal to the $(\bar{2}10)$ growth surface.

A topograph having significantly improved resolution appears in Figure 4(a) for the $(\bar{6}3\bar{2})$ reflection. This improvement was primarily the result of being able to reduce the film-to-specimen distance from 10 to 4 mm without overlapping images becoming a problem. A stereographic description for the Berg-Barrett X-ray topography alignment used in obtaining both a $(\bar{7}2\bar{1})$ and a $(\bar{6}3\bar{2})$ reflection image is given in Figure 5.

Extensive Knoop hardness testing (50 g load) was performed on the $(\bar{2}10)$ growth surface in regions not influenced by the large growth strain center to assess systematically the degree of plastic anisotropy. No effort had been made in previous hardness experiments^{10,11} on RDX to select regions of the crystal that were reasonably strain-free. Now, care was taken to maintain orthogonality between the applied force axis of the indenter and the $(\bar{2}10)$ growth surface by leveling the crystal each time prior to indenting it. The hardness anisotropy for this growth surface is shown in Figure 6, with maximum hardness near the $[124]$. Significant cracking and corresponding lower hardness values occurred when the indenter was aligned along the $[2\bar{4}1]$. This suggests that the amount of plastic deformation under the indenter for this particular arrangement is extremely limited. Consistent with hardness anisotropy measurements on NSWC laboratory-grown crystals,^{10,11} a considerable anisotropy in Knoop hardness number was obtained for this more perfect laboratory-grown RDX crystal. This anisotropy indicates that a limited number of slip systems are operative and that cross slip (dislocation maneuverability) is very difficult.¹⁸

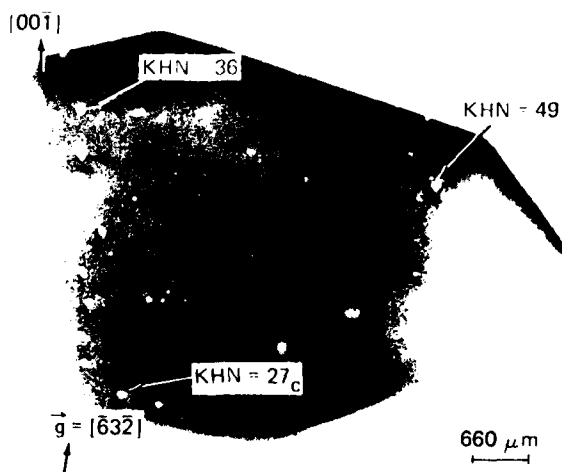
The strain fields of the indentations are delineated in the $(\bar{6}3\bar{2})$ reflection image in Figure 4(b). The absence of any diffracted intensity at the actual indentation sites is the result of the severe strains there. Immediately adjacent to these zero diffracted intensity zones are tiny black regions of enhanced diffracted intensity. The observed highly localized strain fields confirm a previous dislocation etch pit study¹⁰ on an indented NSWC crystal. This result and the large degree of plastic anisotropy indicate that plastic deformation is very inhomogeneous in RDX, even in reasonably perfect crystals.

Vickers hardness experiments (50 and 100 g load) were performed on the (001) growth face of a number of Holsten production-grade Class D RDX crystals, typically 1-2 mm in size (Figure 7(a)). Measurements at 100 g load allowed better resolution of the crystallographically-determined crack traces (Figure 7(b)) that emerged in the (001) surface. Vickers hardness numbers of 30 and 50 kgf/mm² were obtained for two crystals at 100 g load with one pyramid diagonal being roughly parallel (within 10-11½°) to $[110]$. Aligning the pyramid diagonal considerably off $[110]$ yielded badly distorted impressions that were not suitable for hardness measurements. However, varying the indenter diagonal alignment did not substantively alter the crystallographic alignment of the cracking pattern.



CuK α RADIATION AT 14 kV AND 20 mA FOR
~ 2 h ON ILFORD L4 50 μm NUCLEAR PLATE

(a) X-RAY SURFACE REFLECTION TOPOGRAPH
BEFORE KNOOP HARDNESS MEASUREMENTS



CuK α RADIATION AT 14 kV AND 20 mA FOR
5 1/2 h ON ILFORD L4 25 μm NUCLEAR PLATE

(b) X-RAY SURFACE REFLECTION TOPOGRAPH
AFTER KNOOP HARDNESS MEASUREMENTS

FIGURE 4. BERG-BARRETT TOPOGRAPHS ($\bar{6}32$ REFLECTION) BEFORE AND AFTER
KNOOP HARDNESS MEASUREMENTS ON THE (210) GROWTH SURFACE OF
A LABORATORY-GROWN RDX CRYSTAL

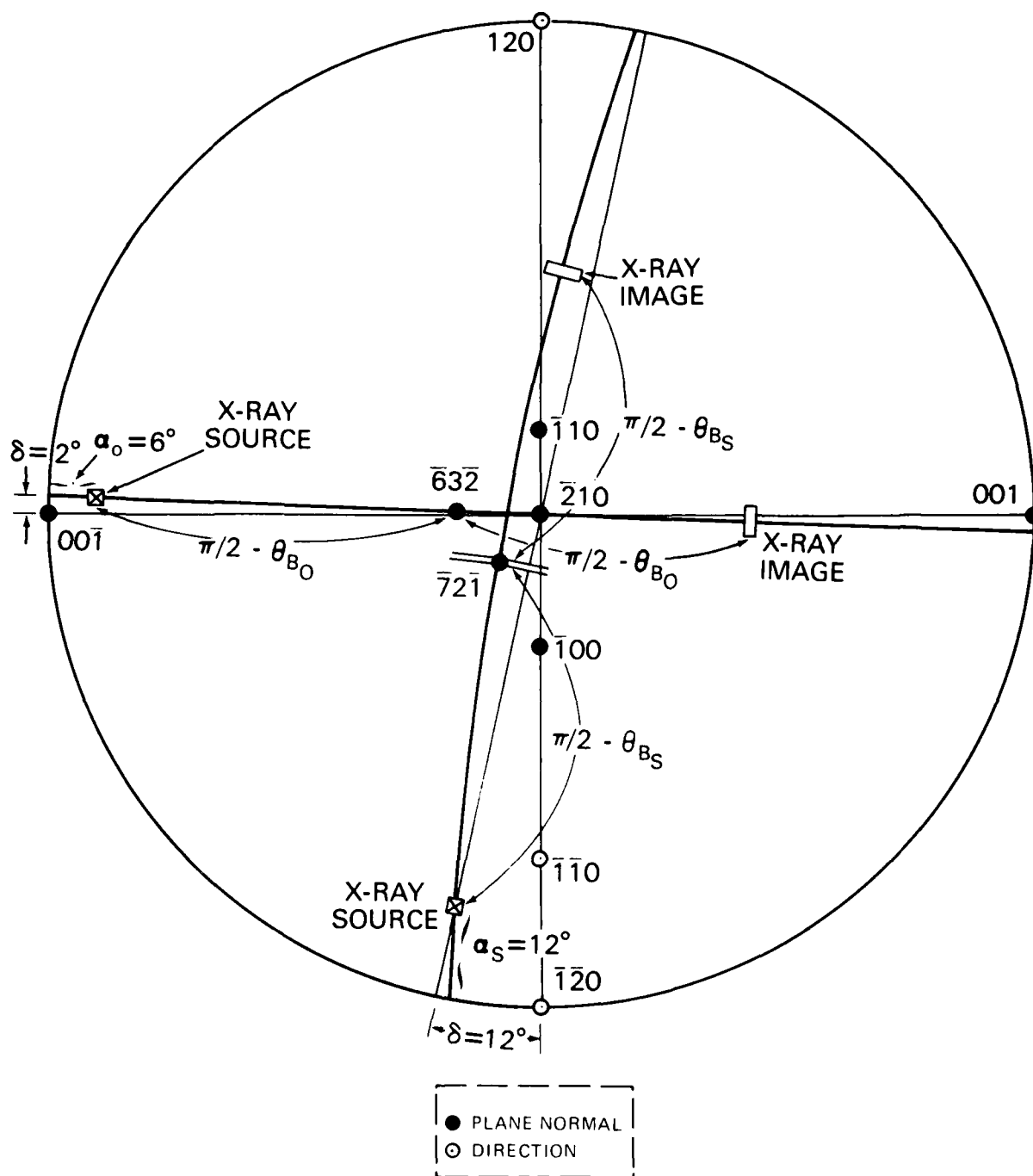


FIGURE 5. STEREOGRAPHIC DESCRIPTION OF X RAY TOPOGRAPHY ALIGNMENT FOR RDX

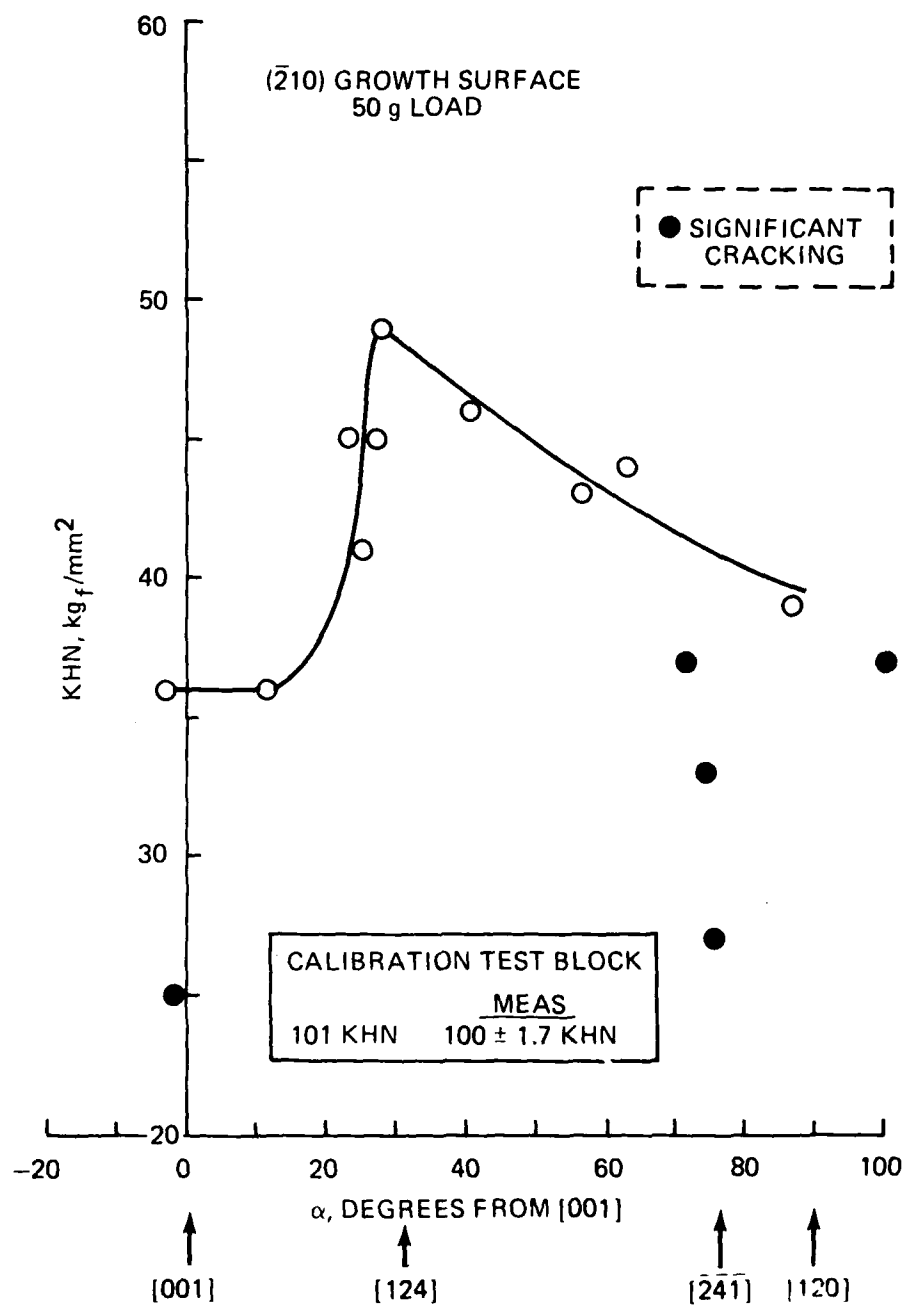
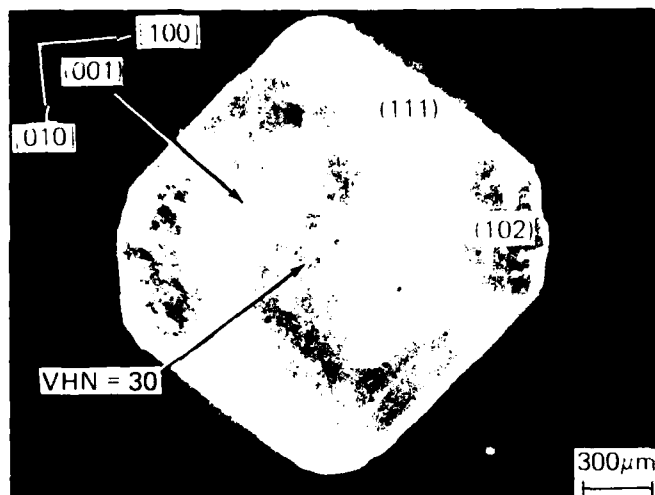
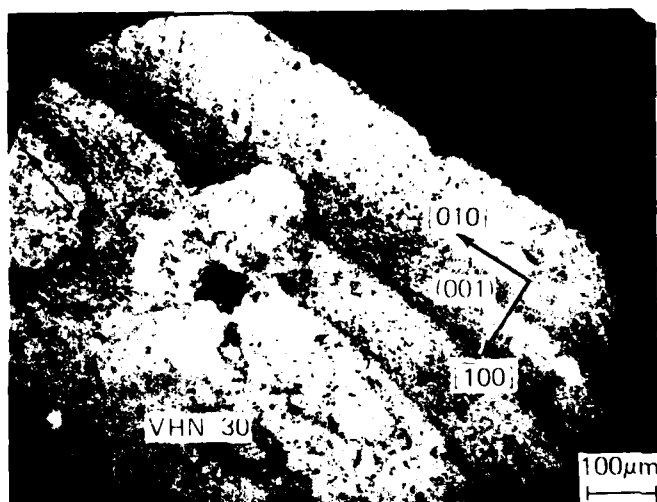


FIGURE 6. KNOOP HARDNESS ANISOTROPY FOR LABORATORY-GROWN RDX CRYSTAL



(a) LIGHT (ZEISS TESSOVAR) PHOTOGRAPH



(b) REFLECTED LIGHT PHOTOGRAPH

FIGURE 7. VICKERS HARDNESS INDENTATION IN HOLSTEN CLASS D RDX CRYSTAL

Microscopic examination of as-received/indented (Figure 7(a)) and fractured (Figure 8) Holsten Class D crystals revealed the presence of numerous pores. Porosity of this type would explain the large variation in Vickers hardness numbers for these crystals. In addition to performing scanning electron microscopy (SEM) on indentation-induced fracture surfaces of Class D crystals, Dr. M. K. Norr (NSWC, Code R34) examined a number of indentations (e.g., Figure 9) in crystals for evidence of well-defined slip. No surface relief caused by cumulative slip centered on the indentations was observed. However, there was what appeared to be a thin sheared layer that was displaced into the large oblong pore near where the Vickers indentation had been placed into the (001) surface (Figure 8). Interestingly, this sheared layer is curved rather than crystallographic in appearance, suggesting possibly that wavy slip¹⁸ occurred.

FEASIBILITY OF MONITORING CHEMICAL DECOMPOSITION IN FRACTURED RDX CRYSTALS

A brief feasibility study was conducted by Dr. J. C. Hoffsommer (NSWC, Code R16) to assess using various instrumental methods to monitor chemical decomposition in crystalline energetic materials that have been deformed and/or fractured in a controlled manner. Assuming that the amount of gas evolved during the fracture of RDX is the same as that for β -lead azide,¹⁹ it is estimated that a 0.05 g crystal of RDX would experience a 0.00017% weight loss corresponding to 0.0024 mole % decomposition. Thin-layer chromatography, high performance liquid chromatography, and gravimetric methods would all be too insensitive to detect such a small amount of decomposition. However, gas chromatographic analysis (electron capture detector) appears feasible to measure chemical decomposition in fractured RDX crystals, provided an organic (e.g., nitroso) derivative is formed with a retention time sufficiently different from RDX. In the event that RDX is completely degraded to gaseous products, a specially adapted mass spectrometer, such as that described by Fox and Soria-Ruiz,¹⁹ would be needed.

In addition, arrangements were made with Prof. J. T. Dickinson, Washington State University (WSU), to perform fracto-emission experiments at WSU on a number of Holsten production-grade Class D RDX crystals, on a few laboratory-grown RDX crystals that are to be characterized at NSWC, and on various model inert crystals.

EXPERIMENTAL INVESTIGATION OF HOT SPOT FORMATION DURING IMPACT

This work is in support of another effort investigating hot spot formation in crystalline explosives, plastic bonded explosives, and propellants when drop-weight impact loaded. An effort was made to verify theoretical predictions^{4,20} that hot spot temperature is proportional to dislocation velocity. Originally, it was planned to use an infrared technique to measure heating as a function of loading rate in samples that were impact loaded. Unfortunately, an equipment problem prevented this. Instead, the time of ignition (based upon measurement of the sample's electrical resistance during impact (with a 5 kg drop weight) was determined as a function of the applied load and the loading rate.



FIGURE 8. SEM PHOTOMICROGRAPHS OF FRACTURE SURFACE OF INDENTED HOLSTEN CLASS D RDX CRYSTAL

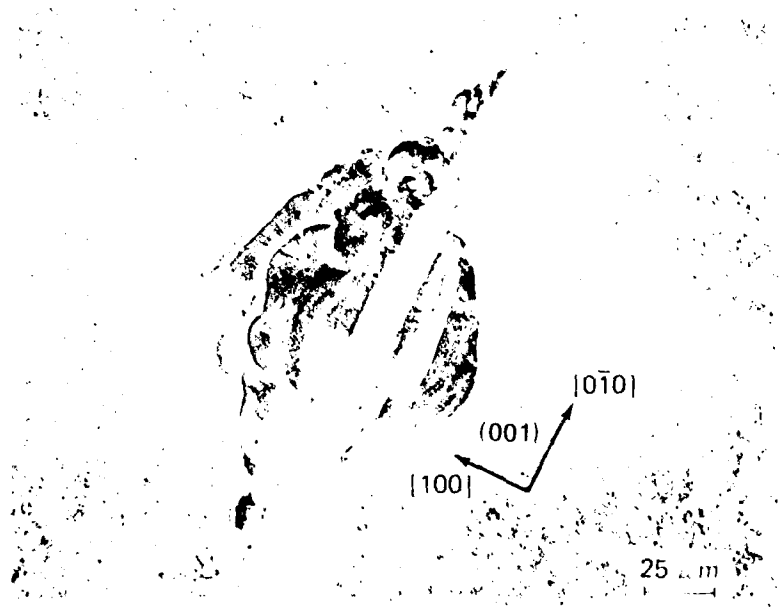
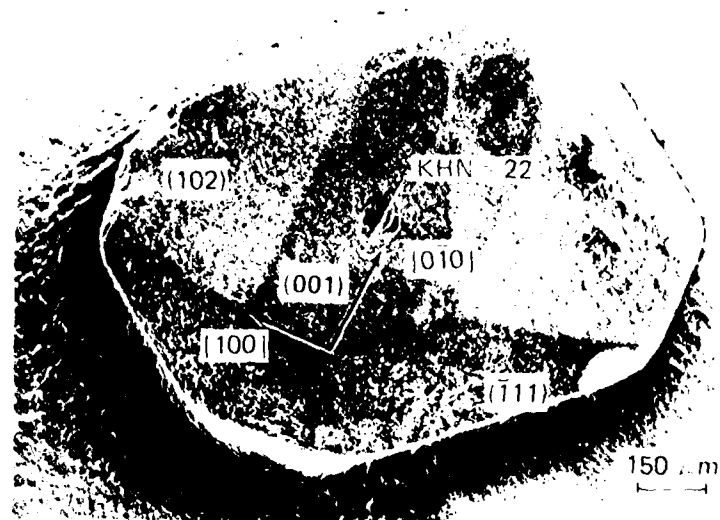


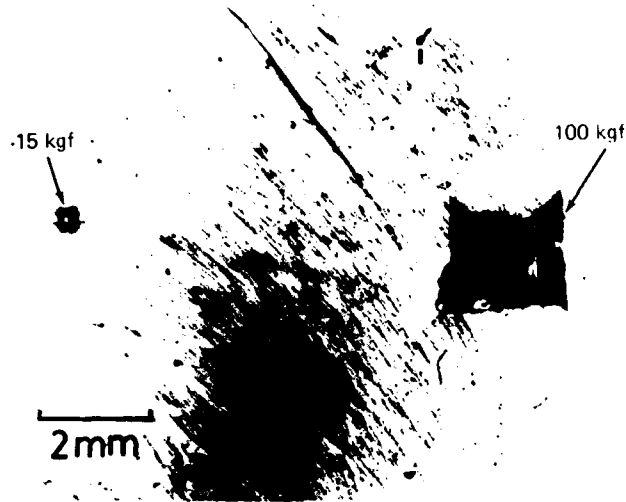
FIGURE 9 SEM PHOTOMICROGRAPHS OF INDENTED HOLSTEN CLASS D RDX CRYSTAL

A number of neat crystalline and plastic bonded explosives, including RDX, HMX, and PETN (pentaerythritetranitrate), were studied; the crystalline explosive samples were 30-35 mg pressed discs (5 mm diameter x 1 mm high). For a given threshold loading rate, corresponding to the first observation of ignition, it was noted that ignition occurred at or just after the maximum in the applied force-time history, measured by a strain gage mounted on the anvil. As the loading rate was increased, the force level at which ignition occurs was found to decrease. Most of this decrease occurs just beyond threshold. For PETN, which has been studied most extensively, doubling the loading rate reduces the average force on the sample at ignition by about a factor of 5. Although needing additional verification, this average force seems to be approaching a limiting value for progressively higher loading rates. In a directly related way, as the loading rate increases, ignition occurs with decreasing amounts of sample deformation. Still needed are strain measurements that will allow stress level comparisons.

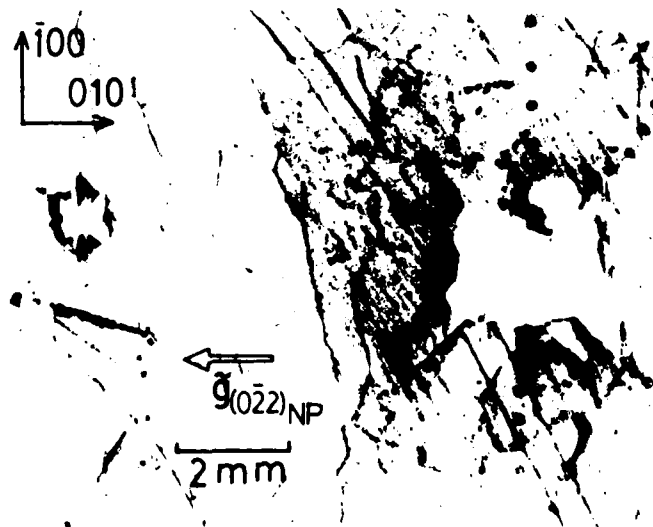
DEFORMATION OF MODEL INERT MATERIALS

The nature of the plastic deformation resulting from performing hardness experiments on selected model inert crystals was investigated at the University of Maryland, College Park, using Berg-Barrett topography. Among the crystals chosen for eventual study are LiF, NaCl, KCl, and MgO, all having the rock salt structure. They span a significant range in deformation behavior for this crystal structure, from being reasonably ductile, as is LiF, to being relatively brittle, as is MgO. Further, these materials have dislocation structures capable of being characterized by X-ray topography.¹⁷ Previous results obtained on LiF crystals have indicated that it should not be prone to hot spot formation because of its ductility, preventing a discontinuous strain gage response.²¹ Alternatively, specific dislocation interactions are known to promote cracking in MgO. Thus, it is of interest to determine whether this cracking behavior²² is associated with a discontinuous stress-strain response, and how this then relates to hot spot formation. Particular emphasis was given to studying MgO during this reporting period. In addition to the work at the University of Maryland, synchrotron X-ray topography was performed²³ on an indented NaCl crystal at the Cornell High Energy Synchrotron Source (CHESS); this particular aspect of the work was done in collaboration with Mr. R. C. Dobbyn and Dr. M. Kuriyama of the National Bureau of Standards.

Hardness experiments using a 1.59 mm (0.0625 in) diameter spherical indenter (1, 3.5, 15, and 100 kg load) were performed on the (001) cleavage surface of a large (several cm) MgO crystal. A Vickers indenter (25 and 100 g load) was then used to probe the strain field caused by the spherical indenter (100 kg load) in an effort to assess strain hardening. The indentations placed at 15 and 100 kg load are shown in Figure 10(a). A Berg-Barrett topograph (022) reflection) of the same area appears in Figure 10(b). The absence of diffracted intensity (white region) centered at either indentation is readily apparent. Immediately surrounding the white region for the indentation placed at 15 kg load is a region of dislocation enhanced diffracted intensity. By comparison, only a minimal region of enhanced intensity exists for the 100 kg load indentation, even though its size might be expected to scale. This surprising lack of enhanced intensity is caused by an absence of residual dislocations that were able to run out to the $\{110\}$ radial crack surfaces (Figure 11).



(a) REFLECTED LIGHT PHOTOGRAPH



(b) X RAY SURFACE REFLECTION TOPOGRAPH

CuK α RADIATION AT 18 kV AND 20 mA FOR
2 h ON ILFORD L4 50 μ m NUCLEAR PLATE

FIGURE 10. HARDNESS INDENTATIONS (SPHERICAL INDENTER) IN THE (001) CLEAVAGE SURFACE OF MgO



FIGURE 11. ENLARGED VIEW (REFLECTED LIGHT) OF INDENTATION (SPHERICAL INDENTER AT 100 kg LOAD) IN (001) CLEAVAGE SURFACE OF MgO

Measurements were made (Figure 12) of indentation size and crack length that resulted for the various loads on the ball and Vickers indenters. Microindentation results obtained by Armstrong and Raghuram⁹ and dynamic ball measurements reported by Chaudhri, Wells, and Stephens²⁴ are included. Results in Figure 12 show that a power law relationship exists between force and diagonal crack length with the exponent equaling $3/2$ in agreement with a fracture mechanics analysis of Lawn and Fuller.²⁵

An effort was made to quantitatively assess the relationship between the strain field surrounding the hardness impression and the cracking that occurred at the indentation site as a function of indenter load. Measurements of the diagonal $\{110\}$ crack length, d_c , and the diagonal length for enhanced X-ray diffraction (i.e., the extent of the strain field), d_x , were made from Berg-Barrett topographs. For the indentation made with the spherical indenter at 100 kg load, $d_x = 3-4$ mm, $d_c = 4.2$ mm, and $d_x \lesssim d_c$. At 15 kg load, $d_x = 1.8$ mm, $d_c = 0.8$ mm, and $d_x \approx 2.3 d_c$. For the Vickers impression made at 0.1 kg load, $d_x = 0.18$ mm, $d_c = 0.05$ mm, and $d_x = 3.6 d_c$. Armstrong and Raghuram⁹ obtained for a Vickers impression made at 0.05 kg load (the lowest load for which cracking occurred) $d_x = 0.2$ mm, $d_c = 0.03$ mm, and $d_x = 6.7 d_c$. These results indicate that an inverse relationship exists between the size of the strain field surrounding the hardness impression and the extent of $\{110\}$ cracking for MgO single crystals.

Hardness impressions were also placed into the (001) cleavage surface of a large (2x2x1 cm) NaCl (Harshaw) crystal using a 1.59 mm (0.0625 in) diameter spherical indenter (3, 10, and 15 kg load). Synchrotron radiation topographs were obtained^{23,26} at the CHESS using the system depicted in Figure 13. A topograph ((044) reflection) of a region of the (001) surface containing three ball indentations appears in Figure 14. The eight-lobed regions of zero diffraction intensity are caused by slip-induced crystal lattice rotations under the spherical indenter. A stereographic projection description of this appears in Figure 15. Observation of the zero intensity regions centered on the indentations is a function of the angular position of the sample with respect to the incident X-ray beam (Figure 16); this sequence of images was photographed from a video screen (the zero diffraction region now appears black).

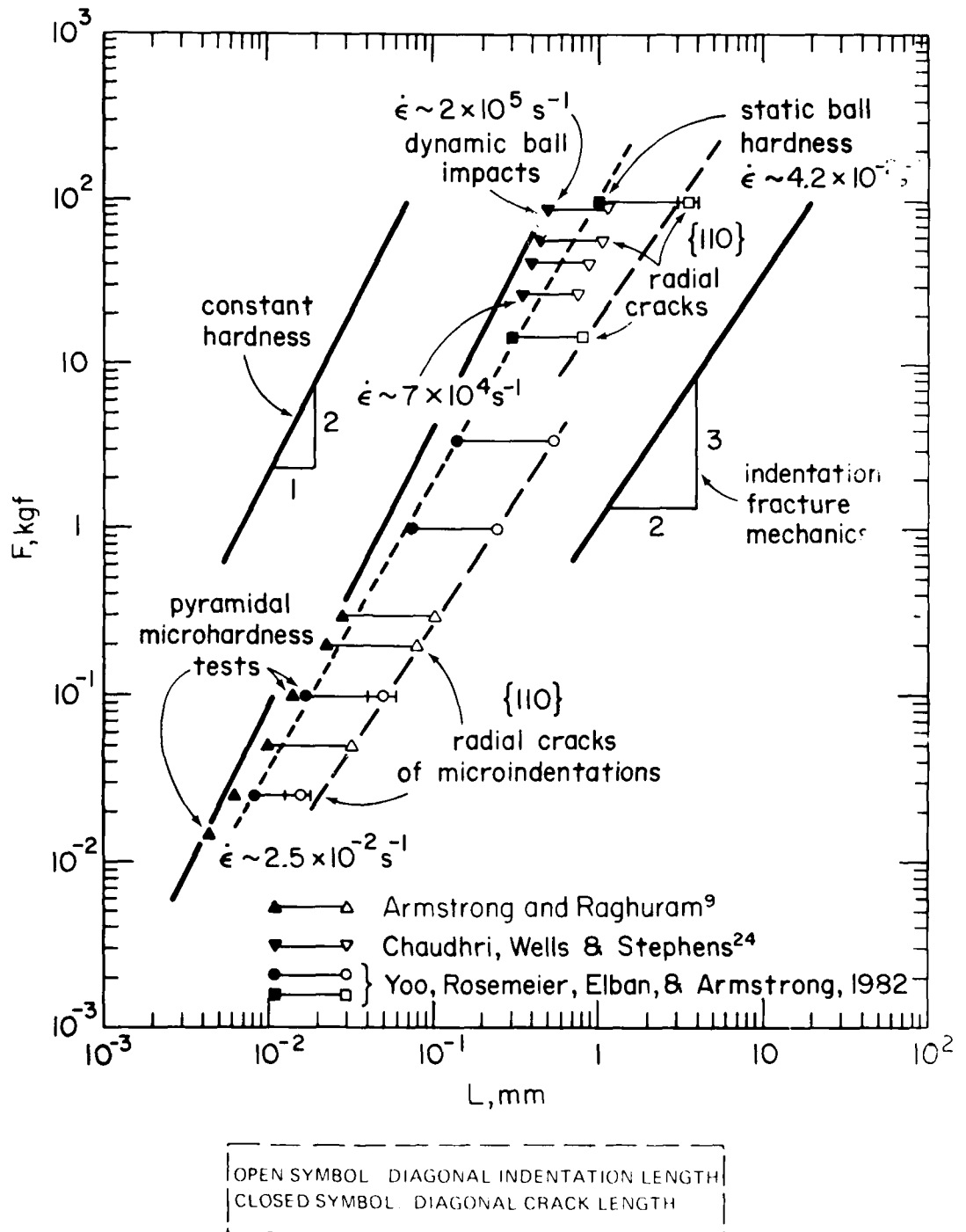


FIGURE-12. LOGARITHMIC VARIATION OF APPLIED FORCE ON THE INDENTER VERSUS DIAGONAL INDENTATION AND CRACK LENGTHS FOR MgO

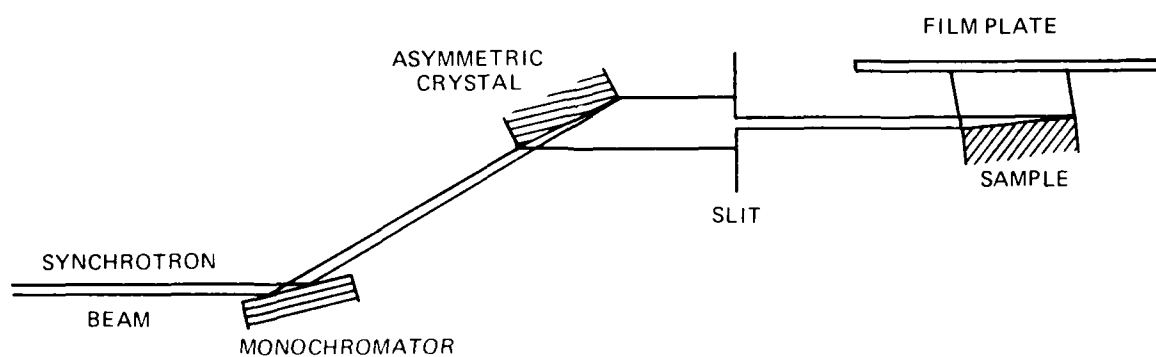
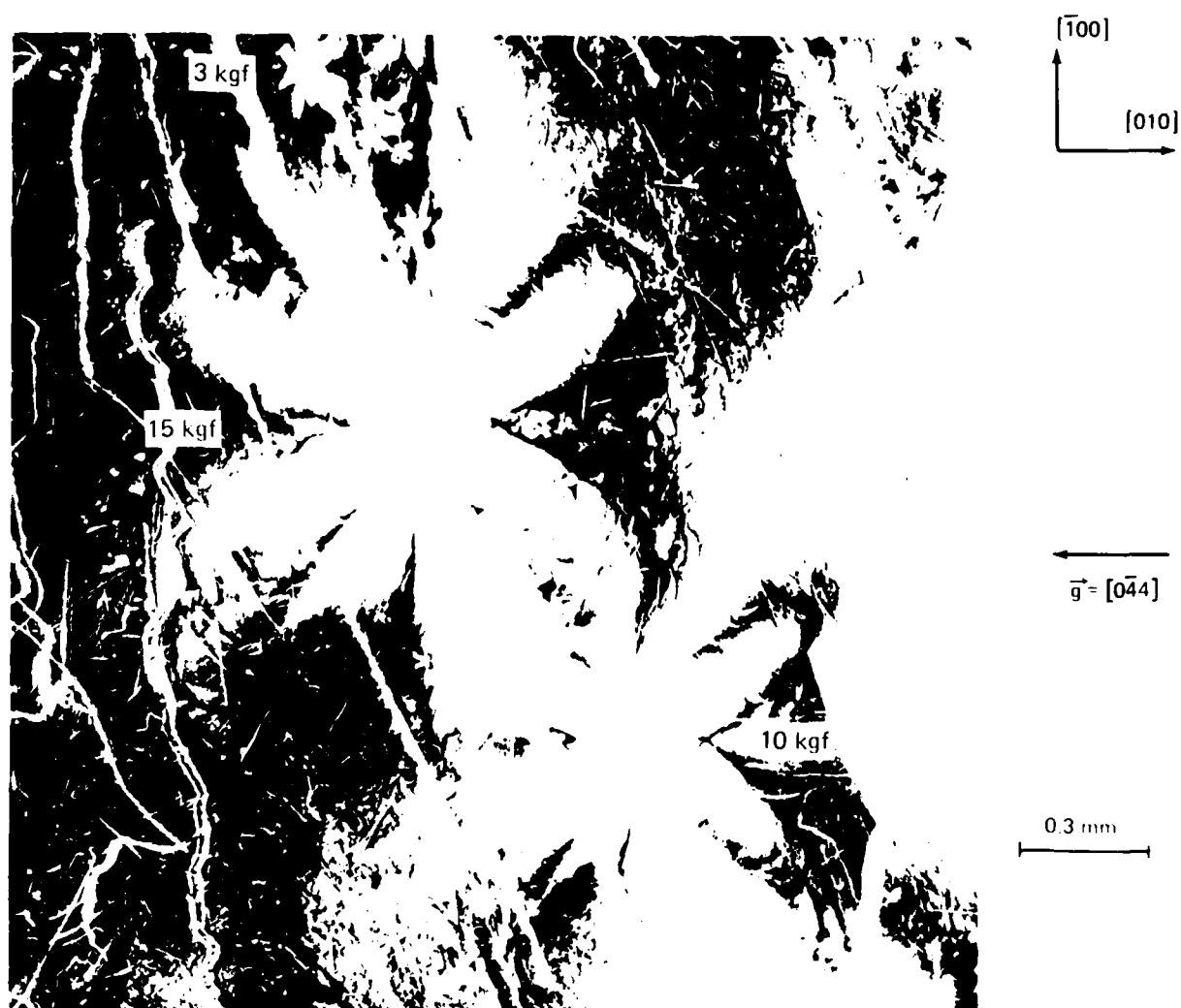


FIGURE 13. SCHEMATIC OF SYNCHROTRON X-RAY TOPOGRAPHY SYSTEM
(AFTER REFERENCE 23)



8 keV SYNCHROTRON RADIATION AT 5.25 GeV AND 16 mA
FOR 5 MIN ON ILFORD L4 50 μ m NUCLEAR PLATE

FIGURE 14. SYNCHROTRON X-RAY TOPOGRAPH (044 REFLECTION) OF INDENTATIONS (SPHERICAL INDENTER) IN THE (001) CLEAVAGE SURFACE OF NaCl (AFTER REFERENCES 23 AND 26)

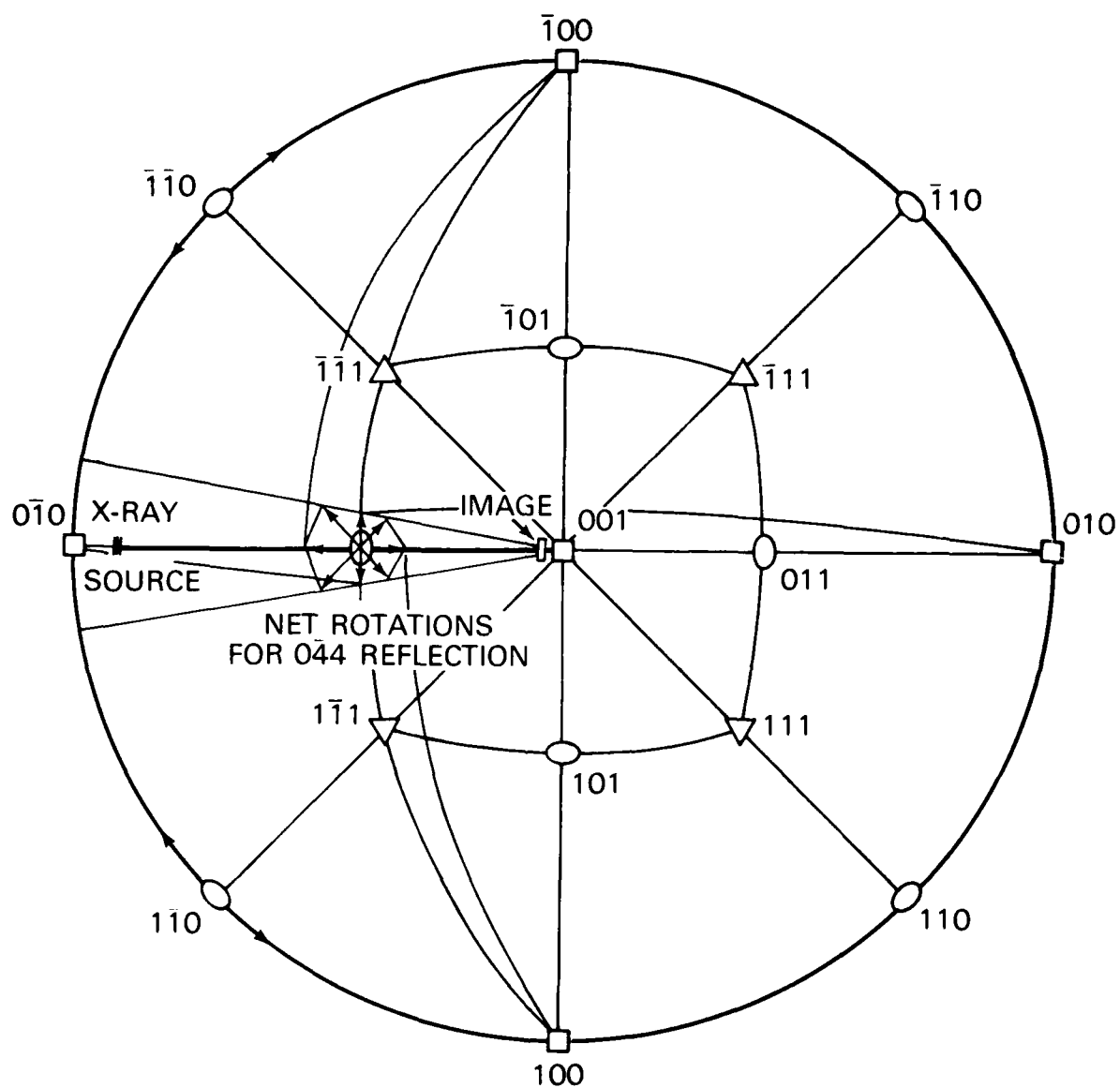


FIGURE 15 STEREOGRAPHIC DESCRIPTION OF X RAY TOPOGRAPHY ALIGNMENT AND SLIP INDUCED LATTICE ROTATION FOR NaCl (AFTER REFERENCE 23)

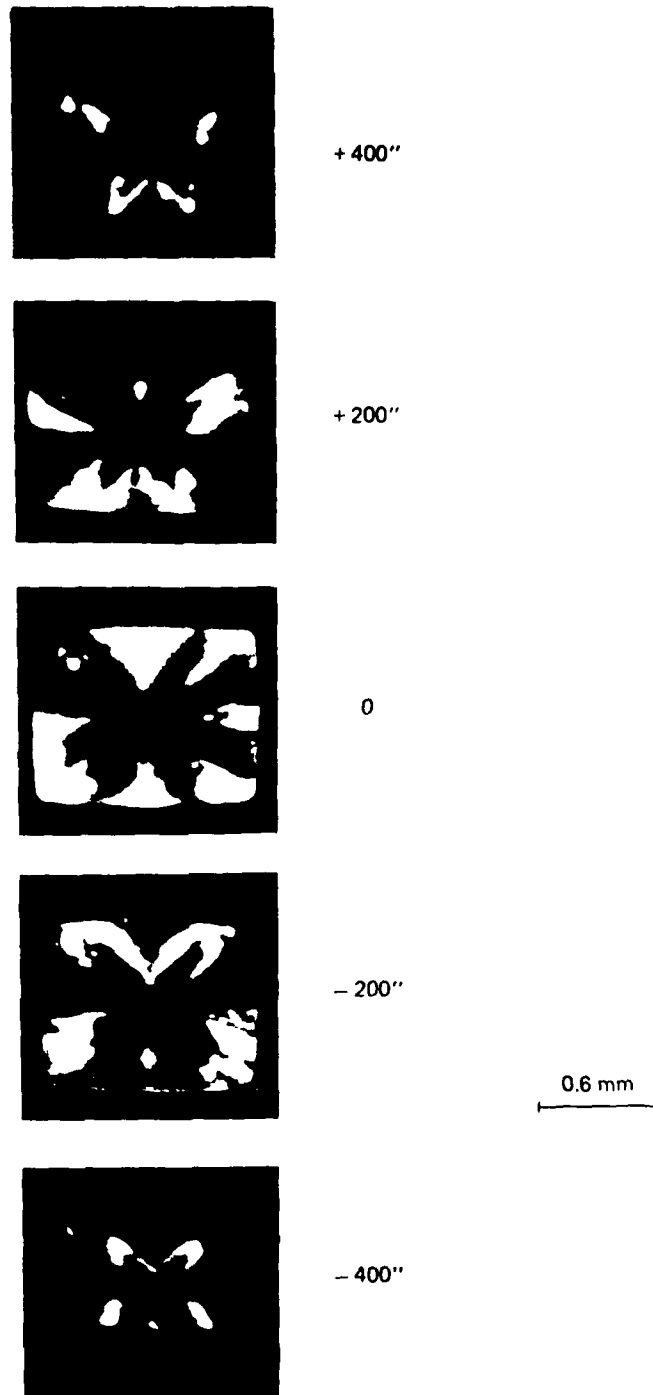


FIGURE 16. SEQUENCE OF SYNCHROTRON X-RAY TOPOGRAPHS (044 REFLECTION) OF INDENTATION (SPHERICAL INDENTER) IN THE (001) CLEAVAGE SURFACE OF NaCl AS A FUNCTION OF ANGULAR POSITION OF CRYSTAL (AFTER REFERENCE 23)

SUMMARY

Berg-Barrett X-ray topography has been used to characterize growth perfection and the extent of deformational zones associated with induced imperfections (i.e., hardness impressions) in a laboratory-grown RDX crystal having reasonable microstructural perfection. Large, localized internal stress concentrations are predicted to occur in RDX, helping to explain its propensity for hot spot formation under drop-weight impact conditions. The considerable variation in Vickers hardness observed for Holsten production-grade Class D RDX crystals was attributed to porosity.

A companion study involving hardness experiments and Berg-Barrett topography (to assess the strain fields surrounding the hardness impressions) was performed on MgO. This material was selected as a model inert that should exhibit a discontinuous stress-strain response. It was observed, for applied loads increasing from 1 to 100 kgf, that the size of the strain fields centered on the impressions was controlled by cracking. In particular, there was a virtual absence of dislocations around the indentation placed at 100 kg load because they ran out $\{110\}$ radial crack surfaces; this was confirmed by the inability to measure any systematic strain hardening by probing the strain field with a Vickers indenter at low loads.

REFERENCES

1. Bowden, F. P., and Yoffe, A. D., Fast Reactions in Solids (London: Butterworths Scientific Publications, 1958).
2. Afanas'ev, G. T., and Bobolev, V. K., Initiation of Solid Explosives by Impact (Jerusalem: Israel Program for Scientific Translations, 1971).
3. Winter, R. E., and Field, J. E., "The Role of Localized Plastic Flow in the Impact Initiation of Explosives," Proceedings of the Royal Society of London, Ser. A, Vol. 343, No. 1634, 1975, pp. 399-413.
4. Armstrong, R. W., Coffey, C. S., and Elban, W. L., "Adiabatic Heating at a Dislocation Pile-Up Avalanche," Acta Metallurgica, Vol. 30, No. 12, 1982, pp. 2111-2116.
5. Fuller, K. N. G., Fox, P. G., and Field, J. E., "The Temperature Rise at the Tip of Fast Moving Cracks in Glassy Polymers," Proceedings of the Royal Society of London, Ser. A, Vol. 361, No. 1705, 1978, pp. 245-263.
6. Weichert, R., and Schönert, K., "Heat Generation at the Tip of a Moving Crack," Journal of the Mechanics and Physics of Solids, Vol. 26, No. 3, 1978, pp. 151-161.
7. Coffey, C. S., Elban, W. L., and Jacobs, S. J., "Detection of Local Heating and Reaction Induced by Impact," in Proceedings of the Sixteenth JANNAF Combustion Meeting, Vol. 1, 10-14 Sep 1979, CPIA Publ. 308, pp. 205-219.
8. Westbrook, J. H., and Conrad, H., Eds., The Science of Hardness Testing and Its Research Applications (Metals Park, OH: American Society for Metals, 1973).
9. Armstrong, R. W., and Raghuram, A. C., "Anisotropy of Microhardness in Crystals," in The Science of Hardness Testing and Its Research Applications, Eds.: Westbrook, J. H., and Conrad, H. (Metals Park, OH: American Society for Metals, 1973), pp. 174-186.
10. Elban, W. L., and Armstrong, R. W., "Microhardness Study of RDX to Assess Localized Deformation and Its Role in Hot Spot Formation," in Proceedings Seventh Symposium (International) on Detonation, 16-19 Jun 1981, NSWC MP 82-334, pp. 976-985.
11. Elban, W. L., Armstrong, R. W., and Hoffsommer, J. C., "X-Ray Orientation and Hardness Experiments on RDX Explosive Crystals," to be published in Journal of Materials Science.
12. Connick, W., and May, F. G. J., "Dislocation Etching of Cyclotrimethylene Trinitramine Crystals," Journal of Crystal Growth, Vol. 5, No. 1, 1969, pp. 65-69.

REFERENCES (Cont.)

13. Urbanski, T., Chemistry and Technology of Explosives, Vol. 111, First English Edition (Oxford: Pergamon Press, 1967), pp. 87-98.
14. Armstrong, R. W., "Laboratory Techniques for X-Ray Reflection Topography," in Characterization of Crystal Growth Defects by X-Ray Methods, Eds.: Tanner, B. K., and Bowen, D. K. (New York and London: Plenum Press, 1980), pp. 349-367.
15. Farabaugh, E. N., Parker, H. S., and Armstrong, R. W., "Skew Reflection X-Ray Microscopy of the Vapor-Growth Surface of an Al_2O_3 Single Crystal," Journal of Applied Crystallography, Vol. 6, Part 6, 1973, pp. 482-486.
16. Tanner, B. K., X-Ray Diffraction Topography (Oxford: Pergamon Press, 1976).
17. Armstrong, R. W., and Wu, C. Cm., "X-Ray Diffraction Microscopy," in Microstructural Analysis: Tools and Techniques, Eds.: McCall, J. L., and Mueller, W. M., (New York-London: Plenum Press, 1973), pp. 169-219.
18. Stokes, R. J., "Fracture in Ceramics," in Fundamental Phenomena in the Materials Sciences, Vol. 4, Eds.: Bonis, L. J., Duga, J. J., and Gilman, J. J. (New York: Plenum Press, 1967), pp. 151-175.
19. Fox, P. G., and Soria-Ruiz, J., "Fracture-Induced Thermal Decomposition in Brittle Crystalline Solids," Proceedings of the Royal Society of London, Ser. A, Vol. 317, No. 1528, 1970, pp. 79-90.
20. Coffey, C. S., "Phonon Generation and Energy Localization by Moving Edge Dislocations," Physical Review B, Vol. 24, No. 12, 1981, pp. 6984-6990.
21. Heavens, S. N., and Field, J. E., "The Ignition of a Thin Layer of Explosive by Impact," Proceedings of the Royal Society of London, Ser. A, Vol. 338, No. 1612, pp. 77-93.
22. Armstrong, R. W., and Wu, C. Cm., "Lattice Misorientation and Displaced Volume for Microhardness Indentations in MgO Crystals," Journal of the American Ceramic Society, Vol. 61, Nos. 3-4, 1978, pp. 102-106.
23. Yoo, K.-C., Armstrong, R. W., Dobbyn, R. C., and Kuriyama, M., "An Analysis of Local Deformation in Sodium Chloride by Synchrotron Radiation Topography," to be published.
24. Chaudhri, M. M., Wells, J. K., and Stephens, A., "Dynamic Hardness, Deformation and Fracture of Simple Ionic Crystals at Very High Rates of Strain," Philosophical Magazine, Vol. 43, No. 3, 1981, pp. 643-664.

REFERENCES (Cont.)

25. Lawn, B. R., and Fuller, E. R., "Equilibrium Penny-Like Cracks in Indentation Fracture," Journal of Materials Science, Vol. 10, No. 12, 1975, pp. 2016-2024.
26. Kuriyama, M., and Boettinger, W. J., "Application of Synchrotron Radiation to Materials Science," Materials Research Council (DARPA) Workshop on Nondestructive Microstructural Characterization, 6-7 Jul 1982, pp. 166-187.

PRESENTATIONS AND PUBLICATIONS

1. "Investigation of the Origin of Hot Spots in Deformed Materials,"
W. L. Elban, Workshop on the Fundamentals of Initiation of Chemical
Decomposition of Energetic Materials, Chestertown, MD, 4-6 May 1982.
2. "X-Ray Orientation and Hardness Experiments on KDX Explosive Crystals,"
W. L. Elban, R. W. Armstrong, and J. C. Hoffsommer, to be published in
Journal of Materials Science.
3. "X-Ray Diffraction Topography of Energetic and Inert Crystals,"
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