

ADA146685

**MICROSTRUCTURAL ORIGINS OF HOT SPOTS IN RDX EXPLOSIVE  
AND A REFERENCE INERT MATERIAL**

**ANNUAL PROGRESS REPORT NO. 2 FOR PERIOD 1 OCT 1982 TO 30 SEP 1983  
WORK REQUEST NUMBER N00014-83-WR-30046 (WORK UNIT NUMBER NR659-797)  
AND FOR PERIOD 1 OCT 1982 TO 31 DEC 1983  
WORK REQUEST NUMBER N00014-82-K-0263 (WORK UNIT NUMBER NR659-798)**

**W. L. ELBAN, J. C. HOFFSOMMER, AND C. S. COFFEY  
RESEARCH AND TECHNOLOGY DEPARTMENT**

**AND**

**K. C. YOO\* AND R. G. ROSEMEIER\***

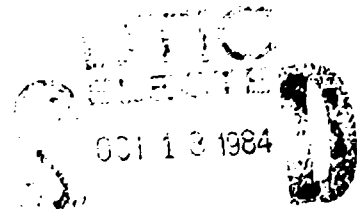
**\*MECHANICAL ENGINEERING DEPARTMENT  
UNIVERSITY OF MARYLAND, COLLEGE PARK**

**MAY 1984**

**REPRODUCTION IN WHOLE OR PART IS PERMITTED FOR ANY PURPOSE OF THE  
UNITED STATES GOVERNMENT.**

**APPROVED FOR PUBLIC RELEASE; DISTRIBUTION UNLIMITED**

**PREPARED FOR  
OFFICE OF NAVAL RESEARCH  
800 N. QUINCY STREET  
ARLINGTON, VA 22217**



**NAVAL SURFACE WEAPONS CENTER**

**Dahlgren, Virginia 22448 • Silver Spring, Maryland 20910**

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

REPORT DOCUMENTATION PAGE		READ INSTRUCTIONS BEFORE COMPLETING FORM
1. REPORT NUMBER NSWC MP 84-200	2. GOVT ACCESSION NO.	3. RECIPIENT'S CATALOG NUMBER
4. TITLE (and Subtitle) MICROSTRUCTURAL ORIGINS OF HOT SPOTS IN RDX EXPLOSIVE AND A REFERENCE INERT MATERIAL		5. TYPE OF REPORT & PERIOD COVERED Progress Report FY83
		6. PERFORMING ORG. REPORT NUMBER
7. AUTHOR(s) W. L. Elban, J. C. Hoffsommer, C. S. Coffey K.-C. Yoo, R. G. Rosemeier		8. CONTRACT OR GRANT NUMBER(s) N00014-83-WR-30046 N00014-82-K-0263
9. PERFORMING ORGANIZATION NAME AND ADDRESS Naval Surface Weapons Center (Code R13) White Oak, Silver Spring, MD 20910 and University of Maryland, College Park, MD 20742		10. PROGRAM ELEMENT, PROJECT, TASK AREA & WORK UNIT NUMBERS 61153N; RR024-02-0D; 0; 659-797(798)
11. CONTROLLING OFFICE NAME AND ADDRESS Office of Naval Research 800 N. Quincy Street Arlington, VA 22217		12. REPORT DATE May 1984
		13. NUMBER OF PAGES 36
14. MONITORING AGENCY NAME & ADDRESS (if different from Controlling Office)		15. SECURITY CLASS. (of this report) UNCLASSIFIED
		15a. DECLASSIFICATION/DOWNGRADING SCHEDULE
16. DISTRIBUTION STATEMENT (of this Report) Approved for public release; distribution unlimited.		
17. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from Report)		
18. SUPPLEMENTARY NOTES Reproduction in whole or part is permitted for any purpose of the United States Government.		
19. KEY WORDS (Continue on reverse side if necessary and identify by block number)		
Hot Spot RDX Explosive MgF <sub>2</sub> Berg-Barrett X-ray Topography	Laue X-ray Diffraction Hardness Microhardness Drop-Weight Testing	Chemical Analysis Fractoemission Ammonium Perchlorate
20. ABSTRACT (Continue on reverse side if necessary and identify by block number) The crystalline perfection in production-grade RDX material has been investigated using Berg-Barrett and Laue x-ray diffraction techniques. Particularly noteworthy is the observation by x-ray topography that pores within the crystals appear to be surrounded by a strain-free matrix. Fractoemission experiments on production-grade and NSWC laboratory-grown RDX were performed in collaboration with Washington State University researchers. The purpose was to study the early stages of decomposition resulting from fracture. Different electron emission behaviors were observed and attributed		

DD FORM 1473  
1 JAN 73

EDITION OF 1 NOV 68 IS OBSOLETE

S/N 0102-LF-014-6601

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

20. (Cont.)

to the nature of the fracture that occurred. Gas chromatographic analyses were performed on production-grade RDX samples that had been impact loaded to investigate whether any solid state decomposition had occurred. The trinitroso analog of RDX (called R-salt) was identified in impacted RDX residue. Hardness testing has been used to investigate the local deformation of  $MgF_2$  crystals (selected as a reference inert material). A considerable anisotropy in hardness was found; the extent of the strain fields differed appreciably with crystal orientation. Fractoemission experiments were also performed on  $MgF_2$  and revealed a strong crystallographic dependence for electron emission.  $NH_4ClO_4$  exhibited a substantial hardness anisotropy in an initial study that was conducted.

S-N 0102- LF-014-6601

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

FOREWORD

This work was sponsored by the Office of Naval Research under work request numbers N00014-83-WR-30046 and N00014-82-K-0263 as a cooperative effort between NSWC, White Oak, and the University of Maryland, College Park. The results and conclusions presented in this report concerning the microstructural characterization of deformed RDX explosive and a selected reference inert ( $MgF_2$ ) crystal should be of interest to those studying plastic deformation and fracture in these materials. In particular, this work provides insight into their ability to concentrate energy locally as a result of being plastically deformed and fractured, leading to the initial stages of chemical decomposition. A list of references is given after the body of the report.

The authors particularly want to thank J. R. Holden for his help with the Laue x-ray diffraction experiments on RDX, and R. W. Armstrong for helpful comments and suggestions regarding this work. V. F. DeVost helped with the drop-weight impact experiments on RDX.

Approved by:

*H. S. Ham*  
K. F. MUELLER, Head  
Energetic Materials Division

## CONTENTS

	<u>Page</u>
INTRODUCTION . . . . .	1
X-RAY DIFFRACTION STUDIES OF RDX . . . . .	2
FRACTOEMISSION EXPERIMENTS ON RDX . . . . .	6
CHEMICAL DECOMPOSITION IN IMPACTED RDX . . . . .	6
EXPERIMENTAL INVESTIGATION OF HOT SPOT FORMATION DURING IMPACT . . . . .	12
LOCAL PLASTIC DEFORMATION IN $MgF_2$ . . . . .	12
FRACTOEMISSION EXPERIMENTS ON $MgF_2$ . . . . .	16
PLASTIC ANISOTROPY IN $NH_4ClO_4$ . . . . .	18
SUMMARY . . . . .	18
REFERENCES . . . . .	21
PRESENTATIONS AND PUBLICATIONS . . . . .	23
DISTRIBUTION . . . . .	(1)

## ILLUSTRATIONS

<u>Figure</u>		<u>Page</u>
1	SCHEMATIC OF TRANSMISSION BERG-BARRETT X-RAY TOPOGRAPHY EXPERIMENTAL ARRANGEMENT . . . . .	3
2	TRANSMISSION BERG-BARRETT TOPOGRAPH OF HOLSTON CLASS D RDX CRYSTAL . . . . .	3
3	HOLSTON CLASS D RDX CRYSTALS . . . . .	4
4	TRANSMISSION LAUE PHOTOGRAPHS OF HOLSTON CLASS D RDX CRYSTALS . .	5
5	ELECTRON EMISSION FROM NSWC LABORATORY-GROWN RDX CRYSTAL FRACTURED BY THREE-POINT BENDING . . . . .	7
6	ELECTRON EMISSION FROM NSWC LABORATORY-GROWN RDX CRYSTAL FRACTURED IN COMPRESSION . . . . .	5
7	GAS CHROMATOGRAPHIC ANALYSES OF RECOVERED HOLSTON CLASS D RDX (NSWC LOT X924) THAT WAS DROP-WEIGHT IMPACTED . . . . .	10
8	HEAT SENSITIVE FILM RECORDS OF DROP-WEIGHT IMPACTED HOLSTON CLASS D RDX . . . . .	11
9	HARDNESS IMPRESSION (SPHERICAL INDENTER) IN THE (110) SURFACE OF $MgF_2$ ; AND, STEREOGRAPHIC PROJECTION DESCRIPTION . . . . .	13
10	HARDNESS IMPRESSION (SPHERICAL INDENTER) IN THE (111) SURFACE OF $MgF_2$ ; AND, STEREOGRAPHIC PROJECTION DESCRIPTION . . . . .	13
11	KNOOP HARDNESS IMPRESSIONS IN THE (110) SURFACE OF $MgF_2$ . . . . .	14
12	SURFACE REFLECTION BERG-BARRETT TOPOGRAPH OF KNOOP HARDNESS IMPRESSIONS IN THE (110) SURFACE OF $MgF_2$ . . . . .	14
13	SURFACE REFLECTION BERG-BARRETT TOPOGRAPH OF HARDNESS IMPRESSION (SPHERICAL INDENTER) IN THE (110) SURFACE OF $MgF_2$ . . . . .	17
14	SURFACE REFLECTION BERG-BARRETT TOPOGRAPH OF HARDNESS IMPRESSION (SPHERICAL INDENTER) IN THE (111) SURFACE OF $MgF_2$ . . . . .	17
15	KNOOP HARDNESS ANISOTROPY FOR LABORATORY-GROWN $NH_4ClO_4$ CRYSTAL . . . . .	19

## TABLES

<u>Table</u>		<u>Page</u>
1	SUMMARY OF DROP-WEIGHT IMPACT EXPERIMENTS ON CLASS D RDX . . . . .	9
2	KNOOP HARDNESS VALUES OBTAINED FOR VARIOUS $MgF_2$ CRYSTAL SURFACES . . . . .	15
3	ELECTRON EMISSION FROM $MgF_2$ CRYSTALS FRACTURED BY THREE-POINT BENDING . . . . .	16

## INTRODUCTION

It is well known that for initiation to occur in a solid explosive under impact conditions, the energy transferred must be concentrated into small volumes of the explosive. The most widely held view explaining this phenomenon involves the formation of "hot spots" as a result of the explosive experiencing mechanical forces. Heat is generated within a small volume at a sufficient rate to cause the temperature to rise very rapidly, the kinetics being limited by the thermal conductivity to the surrounding medium. The objective of this work is to investigate the microstructural mechanisms responsible for hot spot formation in RDX (cyclotrimethylenetrinitramine), the most common ingredient in Navy explosives, and in several reference inert materials.

In previous work,<sup>1</sup> surface reflection Berg-Barrett x-ray topography was used to characterize an RDX explosive single crystal, having reasonable microstructural perfection, that was grown by slow evaporation from acetone solution with seed crystals. Topographs of  $(\bar{7}2\bar{1})$  and  $(\bar{6}3\bar{2})$  reflections revealed a large growth strain center associated with grown-in dislocations emanating to the  $(\bar{2}10)$  natural growth surface. Extensive Knoop hardness testing (50 g load) was performed on the  $(\bar{2}10)$  surface in regions not influenced by the large growth strain center to assess systematically the degree of plastic anisotropy. The strain fields around the Knoop impressions were then studied using surface reflection Berg-Barrett topography ( $(\bar{6}3\bar{2})$  reflection). Highly localized strain fields were observed, confirming a previous dislocation etch pit study<sup>2</sup> on an indented laboratory-grown crystal of apparent lesser quality.

In addition, Vickers hardness experiments (50 and 100 g load) were performed on the (001) growth face of a number of Holston production-grade Class D RDX crystals.<sup>1,3</sup> A considerable variation in hardness was observed and attributed to internal porosity.

A companion study involving hardness experiments and x-ray topography was performed on an MgO crystal, selected as a reference inert material.<sup>1,4</sup> The strain fields around spherical ball hardness impressions (1 to 100 kg load) placed into the (001) cleavage surface of the single crystal were studied using surface reflection Berg-Barrett topography ( $(0\bar{2}2)$  reflection). The size of the strain fields was found to be controlled by cracking. In particular, there was a virtual absence of residual dislocations around an indentation placed at 100 kg load because they ran out (110) radial crack surfaces; this was confirmed by the inability to measure any systematic strain hardening by probing the strain fields with a Vickers indenter at low loads.

The current work further elucidates the fundamental microstructural reasons for hot spots and chemical decomposition occurring during the deformation of crystalline energetic and inert materials. The work on RDX is being primarily performed at NSWC and is closely allied with an accompanying research effort on selected model inert crystals at the University of Maryland, College Park.

#### X-RAY DIFFRACTION STUDIES OF RDX

Characterization of a large (several mm in size) Holston production-grade Class D RDX crystal (selected from material having NSWC designation X924) has been performed using transmission Berg-Barrett x-ray topography (Figure 1).<sup>5</sup> There was an absence of any appreciably enhanced diffraction intensity surrounding several pores present in the crystal (Figure 2). This result suggests that during crystallization, pore formation occurs without appreciably straining the neighboring lattice. Consequently, there is a near absence of internal strain energy available for subsequent release in these regions of the crystal. A substantial internal strain energy release is hypothesized to be necessary for the production of "hot spots."<sup>6,7</sup>

Using a Buerger precession camera operating in the stationary mode, transmission Laue photographs were obtained of the X924 crystal studied with x-ray topography and a crystal of comparable size selected from more recent Holston Class D material (NSWC designation X976). Both crystals exhibited the same tabular morphology reported previously by Connick and May<sup>8</sup> for RDX crystallized from cyclohexanone (Figure 3). However, the appearance of the diffraction spots was dramatically different (Figure 4). There was considerably more asterism for the crystal taken from X976 material, indicating that a wider spread of orientation exists in the mosaic blocks<sup>9</sup> (very small, slightly misoriented regions of the crystal) comprising the particular crystal. In addition, the crystal gave diffuse reflections close to some sharp reflections, a phenomenon normally attributed<sup>9</sup> to thermal vibration. Vickers hardness testing of the two crystals revealed a considerable difference in their hardnesses: 32-35 kgf/mm<sup>2</sup> for the crystal from X924 material versus 46 kgf/mm<sup>2</sup> for the crystal from X976 material.

Surface reflection Berg-Barrett topographic images were obtained of the strain fields around Vickers hardness impressions (50 to 450 g load) placed into a large RDX crystal that was grown from solution by Dr. H. Cady (Los Alamos National Laboratory, Los Alamos, NM). An analysis and discussion of the results will be developed in the next progress report.

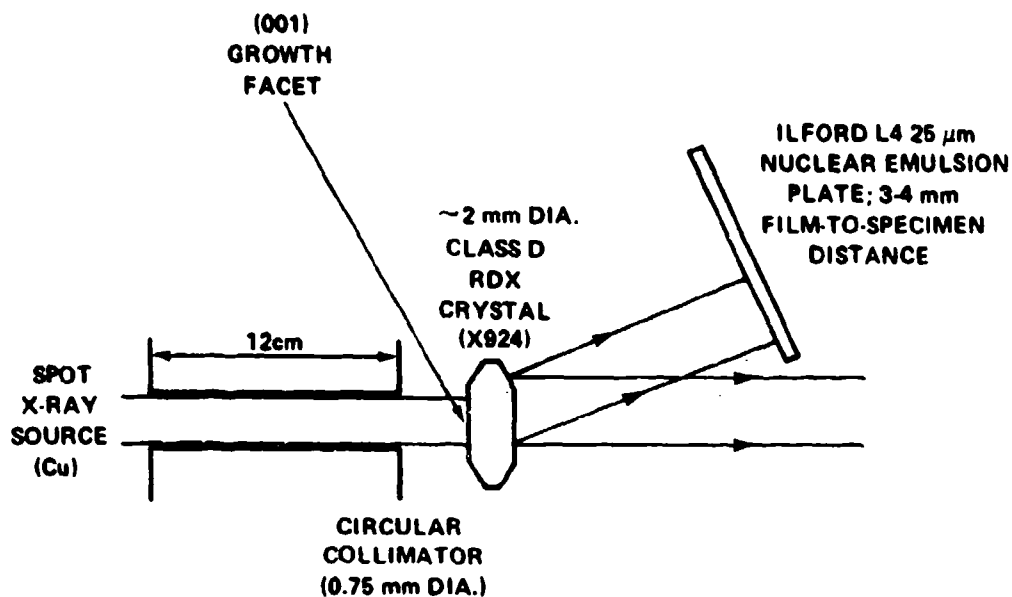
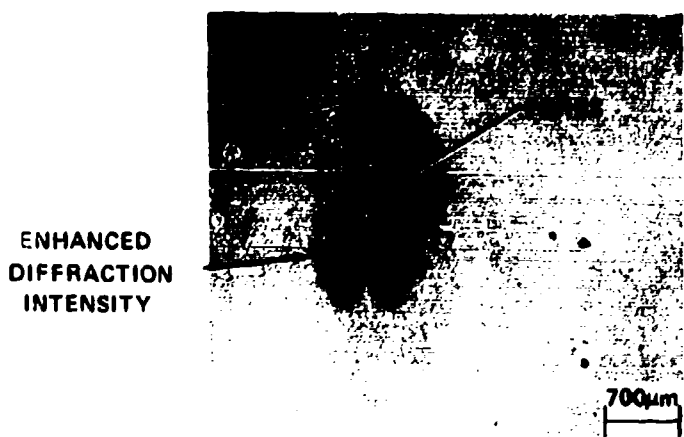
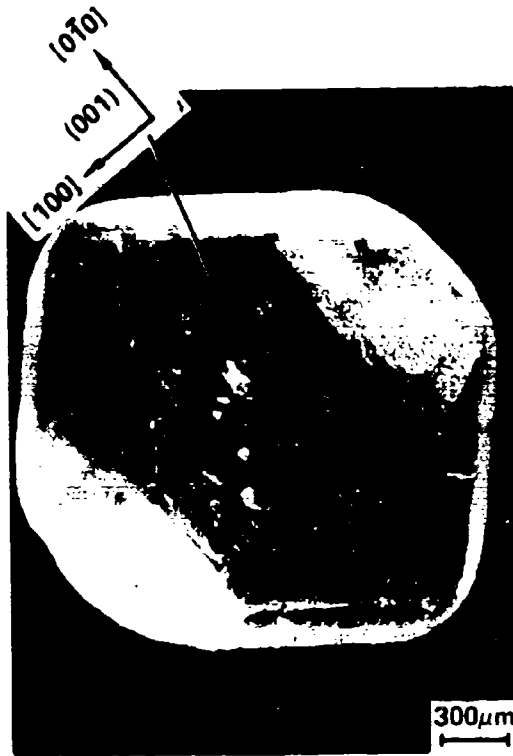


FIGURE 1. SCHEMATIC OF TRANSMISSION BERG-BARRETT X-RAY TOPOGRAPHY EXPERIMENTAL ARRANGEMENT



CuK $\alpha$  RADIATION AT 15 kV AND 20 mA FOR  
18 MIN ON ILFORD L4 25 μm NUCLEAR PLATE

FIGURE 2. TRANSMISSION BERG-BARRETT TOPOGRAPH OF HOLSTON CLASS D RDX CRYSTAL (NSWC LOT X924)

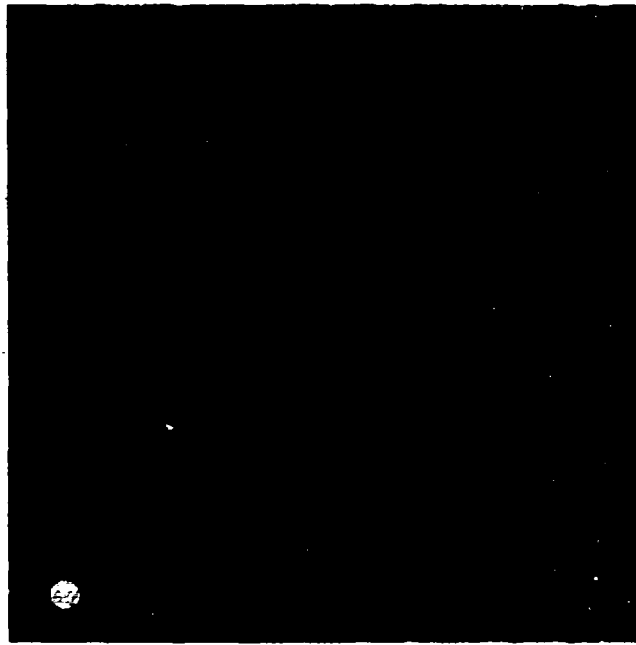


(a) NSWC LOT X924

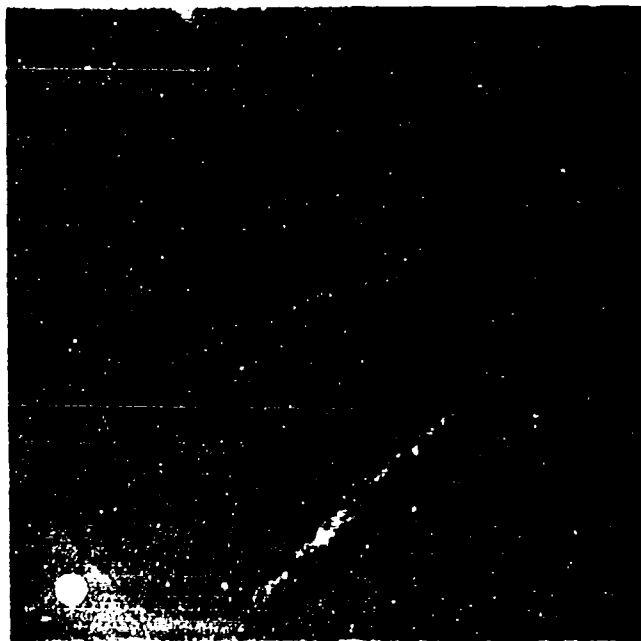


(b) NSWC LOT X976

FIGURE 3. HOLSTON CLASS D RDX CRYSTALS



(a) NSW LOT X924



(b) NSW LOT X976

Mo TUBE RADIATION AT 20 kV AND 20 mA FOR  
30 MIN ON ILFORD INDUSTRIAL G FILM

FIGURE 4. TRANSMISSION LAUE PHOTOGRAPHS OF HOLSTON CLASS D RDX CRYSTALS

## FRACTOEMISSION EXPERIMENTS ON RDX

Fractoemission experiments<sup>10</sup> on production-grade Class D (X924) and NSWC laboratory-grown<sup>2,3</sup> RDX crystals were performed at Washington State University in collaboration with Professors J. T. Dickinson and M. H. Miles in an effort to study the very initial stages of decomposition resulting from fracture. Fracture was achieved by compressive loading (both production and laboratory crystals) and by three-point bending (laboratory crystals only). "Cleavage" type fracture resulted for several crystals subjected to three-point bending; this "singular" fracture event yielded a sharp electron emission peak followed by rapid decay (Figure 5). Multiple fracture occurred for crystals that were compressed; in this case, a large electron emission resulted and continued several minutes after fracture had occurred (Figure 6).

The fracture surfaces of some of the larger recovered fragments from five of the laboratory crystals, subjected to three-point bending, were examined by Dr. M. K. Norr (NSWC, Code R34) with the scanning electron microscope (SEM).<sup>10</sup> The fracture surfaces for three samples were predominantly crystallographic. The fracture surfaces for the remaining two samples examined were mostly glassy; one of these samples gave the highest electron count measured for the three-point bending experiments.

## CHEMICAL DECOMPOSITION IN IMPACTED RDX

In an initial study, a series of drop-weight impact experiments (Table 1) were performed on production-grade Class D (X924) RDX crystals (0.041 g granular samples). A heat sensitive film technique<sup>11,12</sup> was used to detect hot spots and the generation of hot gaseous decomposition products. Gas chromatographic analyses, using a sensitive electron capture detector, were performed on the recovered, fractured samples (Figure 7(a)-(b), as examples). R-salt, the trinitroso analog of RDX (a known decomposition product of thermally degraded RDX), was conclusively identified as being formed in the production-grade RDX that had been impacted. This occurred in samples that were impacted not only at energy levels sufficient to cause the evolution of hot decomposition gases (Figure 8(a), as an example) but also at lower energy levels that resulted only in the occurrence of hot spots (Figure 8(b), as an example). However, based on the limited number of experiments performed, it is unclear how the nitroso compound was formed.

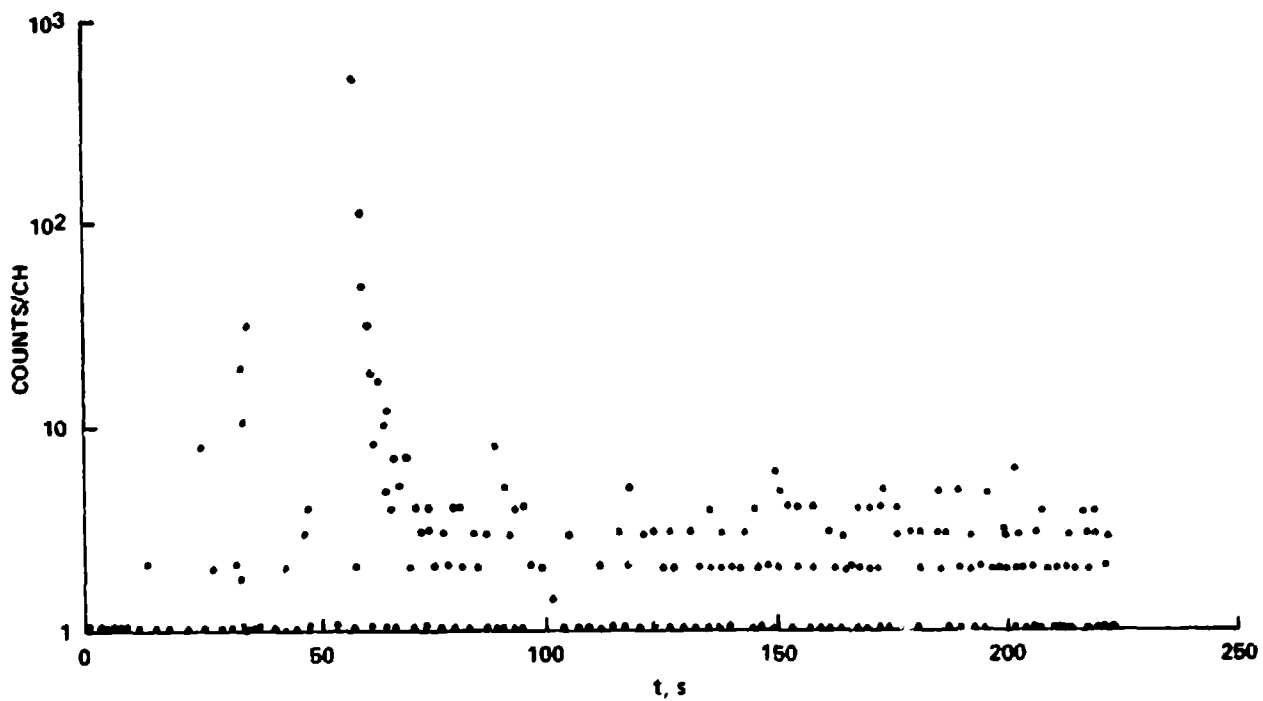


FIGURE 5. ELECTRON EMISSION FROM NSWC LABORATORY-GROWN RDX CRYSTAL FRACTURED BY THREE-POINT BENDING (AFTER REFERENCE 10)

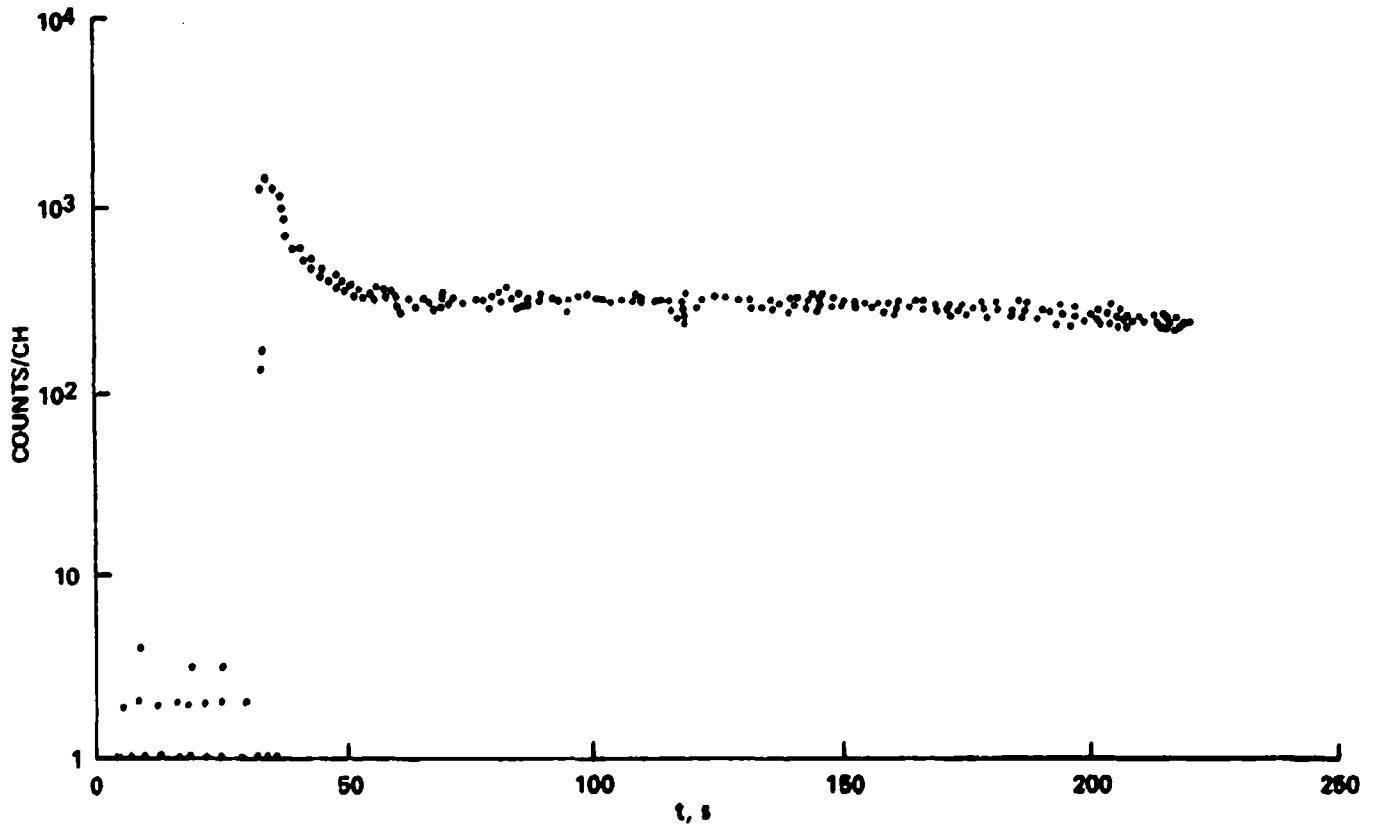


FIGURE 6. ELECTRON EMISSION FROM NSWC LABORATORY-GROWN RDX CRYSTAL FRACTURED IN COMPRESSION (AFTER REFERENCE 10)

TABLE 1. SUMMARY OF DROP-WEIGHT IMPACT EXPERIMENTS  
ON CLASS D RDX (NSWC LOT X924)

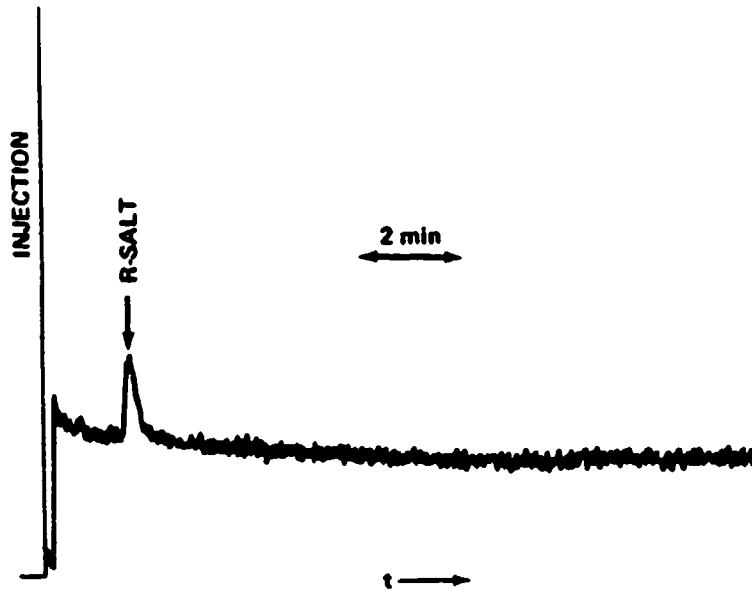
(0.041 g Sample: Loose Pile)

<u>Experiment Number</u>	<u>Drop* Height (cm)</u>	<u>Go/No Go</u>	<u>Comments</u>
1	10	Go**	---
2	5	No Go	Tiny Amount of Browning at Center***
3	8	No Go	Tiny Amount of Browning at Center***
4	9	No Go	Tiny Amount of Browning at Center***
5	9.5	No Go	Tiny Amount of Browning at Center***
6	9.8	Go**	---
7	9.7	No Go	Considerable Amount of Browning at Center***

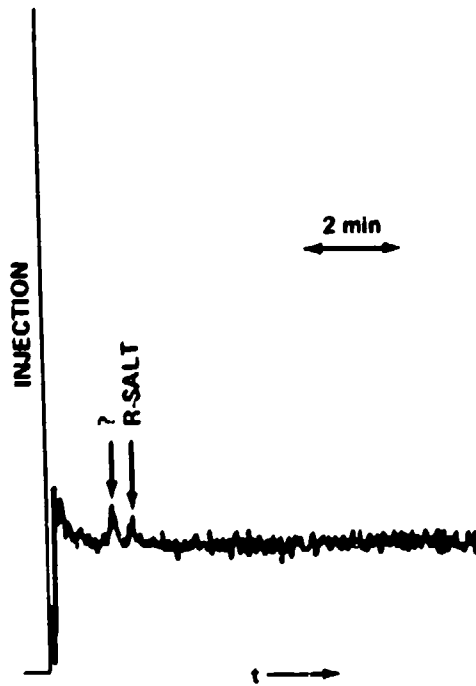
\*5 kg mass

\*\*As indicated by browning of film beyond the extent of the impacted sample  
(Figure 8(a))

\*\*\*Visual observation of heat sensitive film discoloration

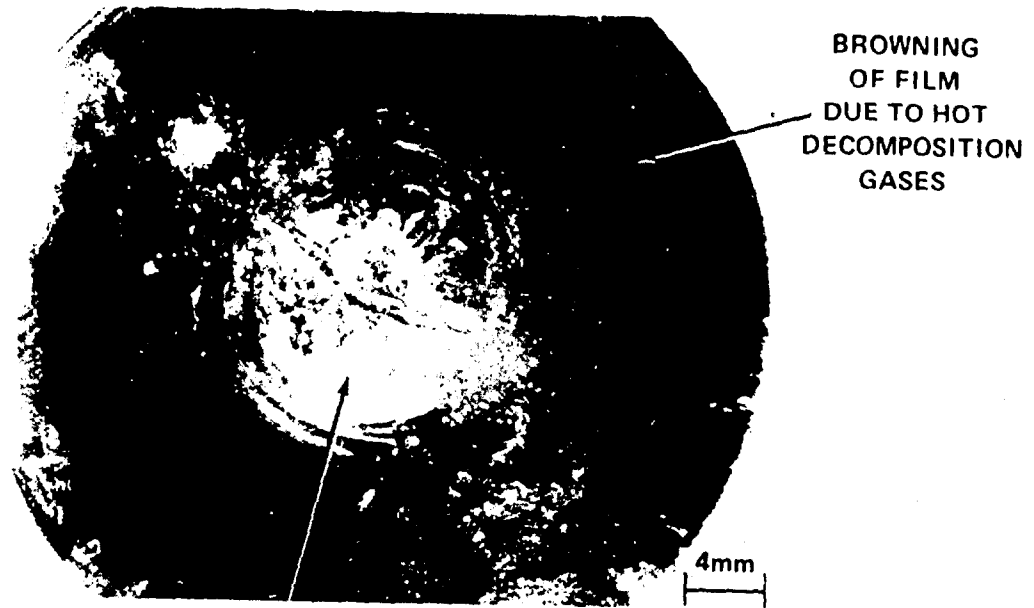


(a) EXPERIMENT #1 (10 cm DROP HEIGHT: GO)



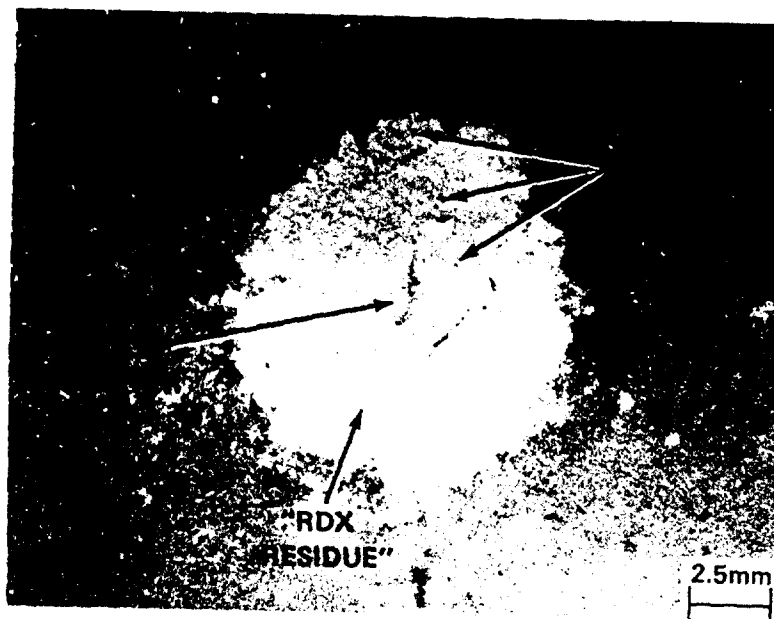
(b) EXPERIMENT #2 (5 cm DROP HEIGHT: NO GO)

FIGURE 7. GAS CHROMATOGRAPHIC ANALYSES OF RECOVERED HOLSTON CLASS D RDX (NSWC LOT X924) THAT WAS DROP-WEIGHT IMPACTED



"RDX  
RESIDUE"

(a) EXPERIMENT #1 (10 cm DROP HEIGHT: GO)



"RDX  
RESIDUE"

(b) EXPERIMENT #2 (5 cm DROP HEIGHT: NO GO)

FIGURE 8. HEAT SENSITIVE FILM RECORDS OF DROP-WEIGHT IMPACTED HOLSTON CLASS D RDX (NSWC LOT X924)

## EXPERIMENTAL INVESTIGATION OF HOT SPOT FORMATION DURING IMPACT

In a separate study, drop-weight impact experiments were performed on pressed samples (0.035 g) of PETN (pentaerythritoltetranitrate) and Comp A-3 (RDX/wax = 91/9) using the well instrumented<sup>13</sup> NSWC impact machine. The heat generated on impact was measured by a fast, broadband (1-13  $\mu\text{m}$ ) infrared detector. Initial, low level stages ( $\Delta T \approx 200^\circ\text{C}$ ) of chemical reaction were observed for both materials while being impacted. However, to date it has not been possible to detect heating prior to the initial stages of chemical reaction. This may be a matter of detector sensitivity, and the problem is being addressed.

LOCAL PLASTIC DEFORMATION IN  $\text{MgF}_2$ 

The nature of the plastic deformation resulting from performing hardness experiments on single crystals of  $\text{MgF}_2$  (supplied by Harshaw) was investigated at the University of Maryland, College Park, using light microscopy and surface reflection Berg-Barrett x-ray topography.<sup>5,14,15</sup>

Optical examination of hardness impressions put into the (110), (100), (101), (001), and (111) surfaces of  $\text{MgF}_2$  using a 1.59 mm (0.0625 in.) diameter spherical indenter (10 kg load) revealed very different deformation features. However, it was determined that these hardness impressions were all reproducible. As examples, a hardness impression on the (110) and (111) surfaces appears in Figures 9 and 10, respectively, together with an appropriate stereographic projection providing angular information for certain directions (open symbol) and plane normals (closed symbol).

Knoop hardness measurements (50 g load) were obtained for different directions in the (110), (001), (101), (100), and (111) crystal surfaces of  $\text{MgF}_2$ . The Knoop hardness indentations put into the (110) surface appear at low magnification in Figure 11. The results for all of the surfaces appear in Table 2 and show that a considerable hardness anisotropy exists for each surface; to this extent,  $\text{MgF}_2$  should serve as an appropriate model material for RDX since it also exhibited<sup>1-3</sup> a significant hardness anisotropy. With one exception, no cracking occurred for those indentations in which the long axis of the indenter was aligned parallel to  $[\bar{1}15]$ , which was also the hard direction for this surface; hardness impressions for the other directions had cracks surrounding them. This result also relates to previous results<sup>1,2</sup> obtained on RDX crystals as well.

The residual strain fields surrounding various hardness indentations in  $\text{MgF}_2$  were investigated using surface reflection Berg-Barrett topography. Figure 12 is an (040) reflection topograph of the area shown optically in Figure 11; the topographic strain fields appear as black regions of enhanced diffracted intensity. The strain fields for [001] impressions were reasonably circular while the strain fields for the other impressions were anisotropic. Higher



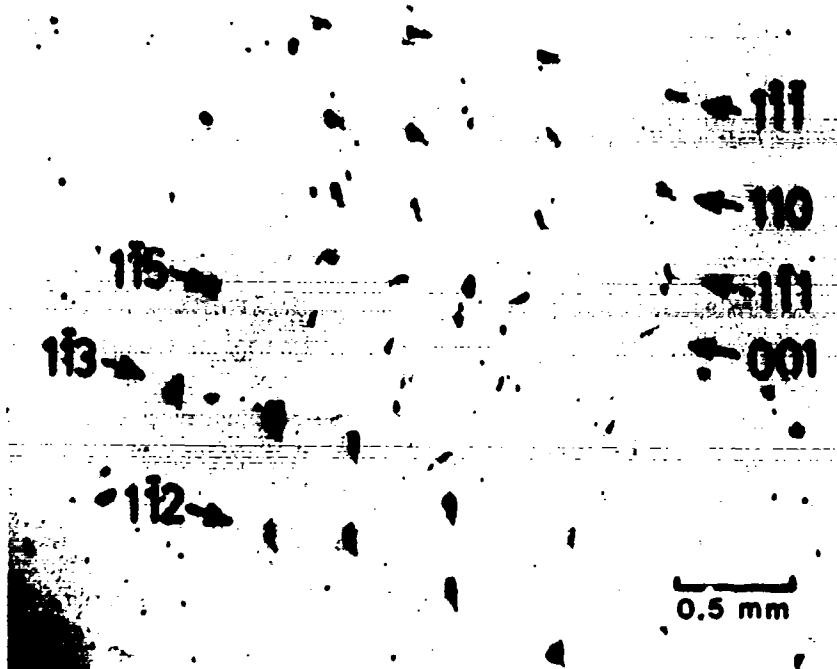


FIGURE 11. KNOOP HARDNESS IMPRESSIONS IN THE (110) SURFACE OF  $MgF_2$

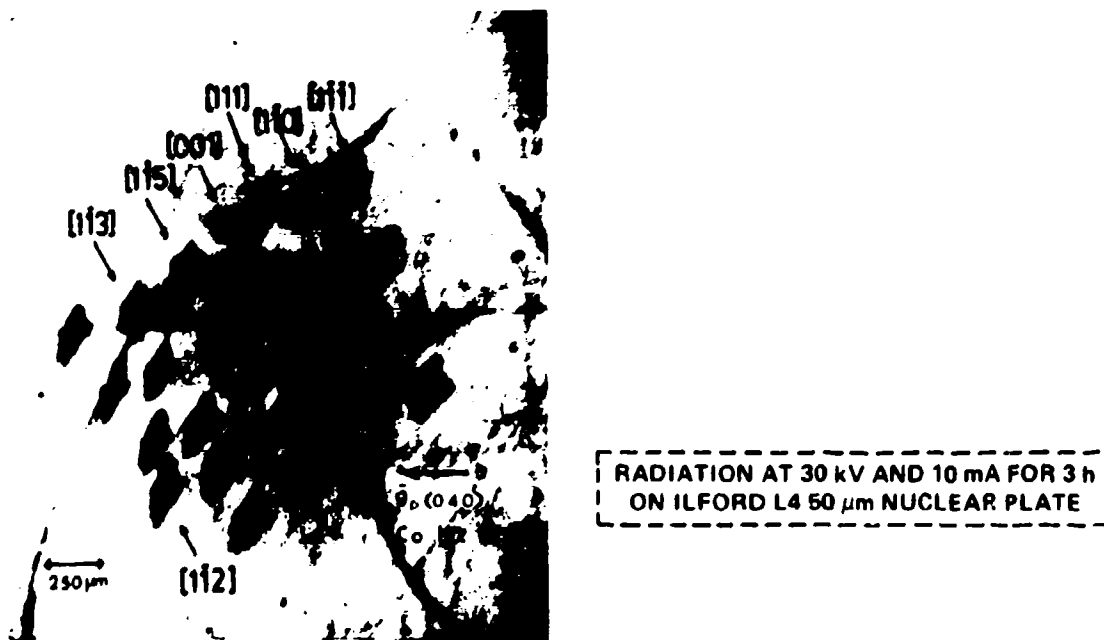


FIGURE 12. SURFACE REFLECTION BERG-BARRETT TOPOGRAPH OF KNOOP HARDNESS IMPRESSIONS IN THE (110) SURFACE OF  $MgF_2$

TABLE 2. KNOOP HARDNESS VALUES OBTAINED FOR VARIOUS  
MgF<sub>2</sub> CRYSTAL SURFACES

(50 g Load)

<u>Surface</u>	<u>Hardness (kgf/mm<sup>2</sup>)</u>	<u>Orientation of Indenter Axis</u>
(110)	268	[001]
	458	[115]
	232	[113]
	139	[112]
	280	[111]
	214	[110]
	242	[111]
(001)	122	[010]
	455	[100]
(101)	444	[010]
	94.4	[111]
	163	[101]
(100)	211	[001]
	133	[011]
	270	[010]
	416	[011]
(111)	310	[101]
	128	[011]
	94.5	[110]

magnification topographs of the hardness impressions placed in the (110) and (111) surfaces using a spherical indenter are shown in Figures 13 and 14, respectively. These two topographs also show a contrast in the degree of anisotropy of the strain fields. In addition, the extent of the strain field for each impression differs appreciably. The strain fields around hardness impressions in  $MgF_2$  are not similar in this respect to those observed in RDX, where there was a virtual absence of enhanced diffracted intensity<sup>1</sup> which is consistent with the findings from etch pit studies.<sup>2,16</sup>

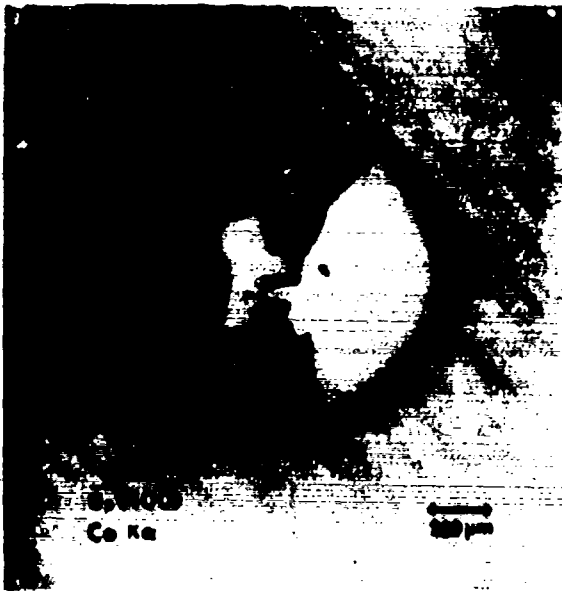
### FRACTOEMISSION EXPERIMENTS ON $MgF_2$

Fractoemission experiments on single crystals of  $MgF_2$  (supplied by Harshaw) were performed at Washington State University in a collaborative effort between Professor J. T. Dickinson and researchers at the University of Maryland, College Park. The specimen dimensions were 10 mm x 5 mm x 1 mm thick. Fracture was achieved by three-point bending using a knife edge; the knife axis was applied normal to either the (110) or (101) surfaces. This resulted in either the (101) or (110) fracture surfaces, respectively. The electron emission results are given in Table 3 for a number of samples and show a marked dependence on the crystallographic orientation. The fracture surfaces were examined in the SEM. The (101) fracture surface was found to consist of two (101) planes while only a single, flat (110) fracture surface resulted. This behavior is opposite to that observed for RDX where a "simple" fracture gave a lower total emission.<sup>10</sup>

TABLE 3. ELECTRON EMISSION FROM  $MgF_2$  CRYSTALS FRACTURED BY THREE-POINT BENDING

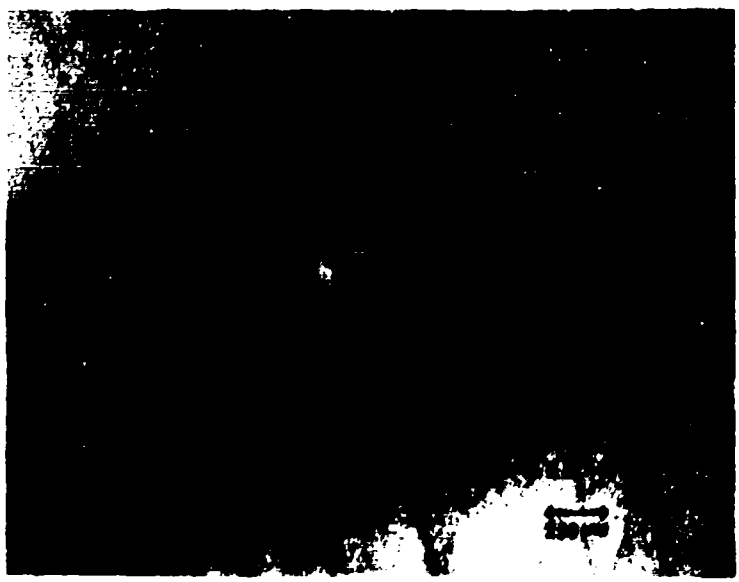
<u>Fracture Surface</u>	<u>Electron Emission (Total Count)</u>			
	<u>Sample #1</u>	<u>#2</u>	<u>#3</u>	<u>#4</u>
(101)	5,500	6,700	5,000	10,000
(110)	433,000*	22,900	22,500	34,700

\*A part of the fractured sample was on the detector.



RADIATION AT 30 kV AND 10 mA FOR 3 h  
ON ILFORD L4 50  $\mu$ m NUCLEAR PLATE

FIGURE 13. SURFACE REFLECTION BERG-BARRETT TOPOGRAPH OF HARDNESS IMPRESSION (SPHERICAL INDENTER) IN THE (110) SURFACE OF  $MgF_2$



RADIATION AT 30 kV AND 10 mA FOR 3 h  
ON ILFORD L4 50  $\mu$ m NUCLEAR PLATE

FIGURE 14. SURFACE REFLECTION BERG-BARRETT TOPOGRAPH OF HARDNESS IMPRESSION (SPHERICAL INDENTER) IN THE (111) SURFACE OF  $MgF_2$

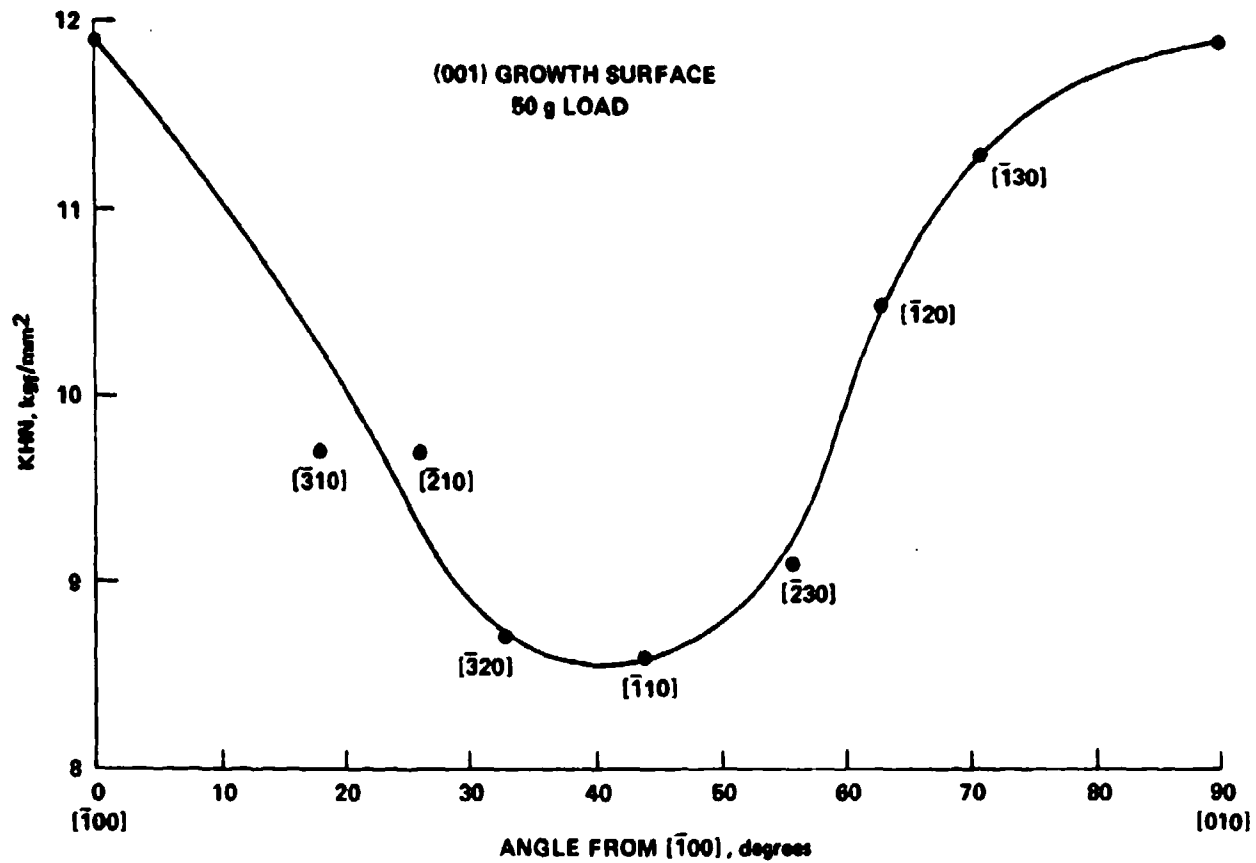
PLASTIC ANISOTROPY IN  $\text{NH}_4\text{ClO}_4$ 

In an initial attempt to assess the plastic anisotropy of ammonium perchlorate ( $\text{NH}_4\text{ClO}_4$ ), Knoop hardness testing (50 g load) was performed on the (001) growth surface of a high quality, pure single crystal (Apache: Code No. 1-3-20-0) supplied by T. Boggs (Naval Weapons Center, China Lake, CA). The hardness anisotropy for this surface is shown in Figure 15, with a minimum hardness being measured for the  $[\bar{1}\bar{1}0]$  direction. The hardness was highest for  $[100]$  and  $[010]$  directions. Although this hardness anisotropy is substantial, it is not as pronounced as has been obtained previously for RDX.<sup>1-3</sup>

## SUMMARY

Through-transmission Berg-Barrett and Laue x-ray diffraction techniques have been used to make significant microstructural observations about the state of growth perfection in Class D (Holston) production-grade RDX crystals. It was determined that pore formation has occurred in these crystals without appreciably straining the surrounding lattice structure. It was also possible to detect a variation in the spread of orientation for mosaic blocks in two lots of crystals. In collaboration with researchers at Washington State University, fractoemission experiments were performed on production-grade and laboratory-grown RDX crystals to investigate the events that take place in the initial stages of decomposition in fractured crystals. Work commenced to identify chemically any solid state decomposition products in impacted production-grade RDX crystals at energy levels below and above that required to cause gaseous decomposition. The trinitroso analog of RDX, R-salt, was successfully identified by gas chromatographic analyses of impacted "RDX residues". The combined work aids our understanding, on a microstructural level, into how "hot spots" can form in crystalline explosives. Further, knowledge of the sub-initiation events that occur in fractured RDX crystals is being acquired.

A companion study involving hardness experiments and Berg-Barrett topography (to assess the strain fields surrounding hardness impressions) was performed on  $\text{MgF}_2$ . This material was selected as a model inert, for one reason, because it has a non-cubic unit cell. A considerable hardness anisotropy was measured for  $\text{MgF}_2$  as was the case for RDX. The significant observation was made that the strain fields around hardness impressions in  $\text{MgF}_2$  were not localized as in RDX. Fractoemission experiments were also performed on  $\text{MgF}_2$  at Washington State University; electron emission was found to have a strong crystallographic dependence. An initial study of the plastic anisotropy of  $\text{NH}_4\text{ClO}_4$  was performed; a substantial hardness anisotropy was measured, although not as great as for RDX.

FIGURE 15. KNOOP HARDNESS ANISOTROPY FOR LABORATORY-GROWN  $\text{NH}_4\text{ClO}_4$  CRYSTAL

## REFERENCES

1. Elban, W. L., Coffey, C. S., Armstrong, R. W., Yoo, K.-C., and Rosemeier, R. G., Microstructural Origins of Hot Spots in RDX Explosive and Several Reference Inert Materials, NSWC MP 83-116, Mar 1983.
2. Elban, W. L., and Armstrong, R. W., "Microhardness Study of RDX to Assess Localized Deformation and Its Role in Hot Spot Formation," in Proceedings Seventh Symposium (International) on Detonation, 16-19 Jun 1981, NSWC MP 82-334, pp. 976-985.
3. Elban, W. L., Hoffsommer, J. C., and Armstrong, R. W., "X-ray Orientation and Hardness Experiments on RDX Explosive Crystals," Journal of Materials Science, Vol. 19, No. 2, 1984, pp. 552-566.
4. Yoo, K.-C., Rosemeier, R. G., Elban, W. L., and Armstrong, R. W., "X-ray Topography Evidence for Energy Dissipation at Indentation Cracks in MgO Crystals," Journal of Materials Science Letters, Vol. 3, No. 6, 1984, pp. 560-562.
5. Tanner, B. K., X-ray Diffraction Topography (Oxford: Pergamon Press, 1976), pp. 24-28.
6. Coffey, C. S., and Armstrong, R. W., "Description of 'Hot Spots' Associated with Localized Shear Zones in Impact Tests," in Shock Waves and High-Strain-Rate Phenomena in Metals, Eds.: Meyers, M. A., and Murr, L. E. (New York: Plenum Publishing Corporation, 1981), pp. 313-324.
7. Armstrong, R. W., Coffey, C. S., and Elban, W. L., "Adiabatic Heating at a Dislocation Pile-Up Avalanche," Acta Metallurgica, Vol. 30, No. 12, 1982, pp. 2111-2116.
8. Connick, W., and May, F. G. J., "Dislocation Etching of Cyclotrimethylene Trinitramine Crystals," Journal of Crystal Growth, Vol. 5, No. 1, 1969, pp. 65-69.
9. Jeffrey, J. W., Methods in X-ray Crystallography (London and New York: Academic Press, 1971), p. 62.
10. Dickinson, J. T., Miles, M. H., Elban, W. L., and Rosemeier, R. G., "Fractoemission from Cyclotrimethylenetrinitramine (RDX) Explosive Single Crystals," Journal of Applied Physics, Vol. 55, No. 11, 1984, pp. 3994-3998.

## REFERENCES (Cont.)

11. Coffey, C. S., and Jacobs, S. J., "Detection of Local Heating in Impact or Shock Experiments with Thermally Sensitive Films," Journal of Applied Physics, Vol. 52, No. 11, 1981, pp. 6991-6993.
12. Coffey, C. S., Elban, W. L., and Jacobs, S. J., "Detection of Local Heating and Reaction Induced by Impact," in Proceedings of the Sixteenth JANNAF Combustion Meeting, Vol. I, 10-14 Sep 1979, CPIA Publ. 308, pp. 205-219.
13. Coffey, C. S., and DeVost, V. F., "Evaluation of Equipment Used to Impact Test Small-Scale Explosive and Propellant Samples," NSWC TR 81-215, 1 Nov 1981.
14. Armstrong, R. W., and Wu, C. Cm., "X-ray Diffraction Microscopy," in Microstructural Analysis: Tools and Techniques, Eds.: McCall, J. L., and Mueller, W. M. (New York-London: Plenum Press, 1973), pp. 169-219.
15. Armstrong, R. W., "Laboratory Techniques for X-ray Reflection Topography," in Characterization of Crystal Growth Defects by X-ray Methods, Eds.: Tanner, B. K., and Bowen, D. K. (New York and London: Plenum Press, 1980), pp. 349-367.
16. Halfpenny, P. J., Roberts, K. J., and Sherwood, J. N., "Dislocations in Energetic Materials--Part 3--Etching and Microhardness Studies of Pentaerythritol Tetranitrate and Cyclotrimethylenetrinitramine," Journal of Materials Science, Vol. 19, No. 5, 1984, pp. 1629-1637.

## PRESENTATIONS AND PUBLICATIONS

1. "Microstructural Origins of Hot Spots and Decomposition in Deformed RDX Explosive," W. L. Elban, R. G. Rosemeier, K.-C. Yoo, and R. W. Armstrong, Oral Presentation before the GNR Workshop on Dynamic Deformation, Fracture, and Transient Combustion, Chestertown, MD, 28-29 Jul 1983.
2. "Continuing Investigation of Hot Spots in Model Inert Crystals," K.-C. Yoo, R. G. Rosemeier, R. W. Armstrong, and W. L. Elban, Oral Presentation before the ONR Workshop on Dynamic Deformation, Fracture, and Transient Combustion, Chestertown, MD, 28-29 Jul 1983.
3. "Surface Reflection Berg-Barrett Topographic Study of Strain Fields Around Hardness Impressions in RDX Explosive and MgO Single Crystals," W. L. Elban, R. W. Armstrong, K.-C. Yoo, R. G. Rosemeier, and R. Y. Yee, Oral Presentation before the Summer Meeting of the American Crystallographic Association, Snowmass Village, CO, 1-5 August 1983.
4. "Microstructural Origins of Hot Spots in RDX Explosive and Several Reference Inert Materials," W. L. Elban, C. S. Coffey, R. W. Armstrong, K.-C. Yoo, and R. G. Rosemeier, NSWC MP 83-116, Mar 1983. (Annual Progress Report No. 1 prepared for ONR).
5. "X-ray Orientation and Hardness Experiments on RDX Explosive Crystals," W. L. Elban, J. C. Hoffsommer, and R. W. Armstrong, Journal of Materials Science, Vol. 19(2), 1984, pp. 552-566.
6. "X-ray Topography Evidence for Energy Dissipation at Indentation Cracks in MgO Crystals," K.-C. Yoo, R. G. Rosemeier, W. L. Elban, and R. W. Armstrong, Journal of Materials Science Letters, Vol. 3(6), 1984, pp. 560-562.

## DISTRIBUTION

	<u>Copies</u>		<u>Copies</u>
Assistant Secretary of the Navy (R, E, and S), Room 5E 731 Attn: Dr. L. V. Schmidt Pentagon Washington, DC 20350	1	AFATL - DLDL Attn: Mr. O. K. Heiney Elgin AFB, FL 32542	1
Commandant of the Marine Corps Attn: Code RD-1, Dr. A. L. Slafkosky, Scientific Advisor Washington, DC 20380	1	AFRPL Attn: MKP/MS24, Mr. R. Geisler Edwards AFB, CA 93523	1
Office of Naval Research Attn: Code 432, Dr. R. S. Miller Arlington, VA 22217	10	AFRPL Attn: MKPA, Dr. F. Roberto Edwards AFB, CA 93523	1
Office of Naval Research Attn: Code 260, Mr. D. Siegel Arlington, VA 22217	1	Air Force Office of Scientific Research Attn: Dr. L. H. Caveny Directorate of Aerospace Sciences Bolling Air Force Base Washington, DC 20332	1
Office of Naval Research Attn: Dr. R. J. Marcus Western Office 1030 East Green Street Pasadena, CA 91106	1	Air Force Office of Scientific Research Attn: Dr. D. L. Ball Directorate of Chemical Sciences Bolling Air Force Base Washington, DC 20332	1
Office of Naval Research Attn: Code 431N, Dr. L. Peebles Arlington, VA 22217	1	FJSRL/NC Attn: Dr. J. S. Wilkes, Jr. USAF Academy, CO 80840	1
Office of Naval Research Attn: Dr. P. A. Miller, Suite 601 San Francisco Area Office One Hallidie Plaza San Francisco, CA 94102	1	Aerojet Strategic Propulsion Co. Attn: Dr. R. L. Lou P.O. Box 15699C Sacramento, CA 95813	1
		Anal-Syn Lab, Inc. Attn: Dr. V. J. Keenan P.O. Box 547 Paoli, PA 19301	1

## DISTRIBUTION (Cont.)

	<u>Copies</u>		<u>Copies</u>
Army Ballistic Research Labs Attn: Code DRDAR-BLT, Dr. P. Howe ARRADCOM Aberdeen Proving Ground, MD 21005	1	Hercules, Inc. Eglin Attn: Dr. R. L. Simmons AFATL/DLDD Eglin AFB, FL 32542	1
Army Ballistic Research Labs Attn: Code DRDAR-BLI, Mr. L. A. Watermeier ARRADCOM Aberdeen Proving Ground, MD 21005	1	Hercules, Inc. Attn: Dr. E. H. Debutts Baccus Works P.O. Box 98 Magna, UT 84044	1
Commander U.S. Army Missile Command Attn: DRSMI-RKL, Dr. W. W. Wharton Redstone Arsenal, AL 35898	1	Hercules, Inc. Magna Attn: Dr. J. H. Thacher Baccus Works P.O. Box 98 Magna, UT 84004	1
Commander Army Missile Command Attn: DRSMI-R, Dr. R. G. Rhoades Redstone Arsenal, AL 35898	1	Johns Hopkins University APL Attn: Mr. T. M. Gilliland Chemical Propulsion Info. Agency Johns Hopkins Road Laurel, MD 20810	1
Atlantic Research Corp. Attn: Dr. W. D. Stephens Pine Ridge Plant 7511 Wellington Rd. Gainesville, VA 22065	1	Lawrence Livermore Laboratory University of California Attn: Dr. R. McGuire, Code L-324 Livermore, CA 94550	1
Ballistic Research Laboratory ARRADCOM Attn: DRDAR-BLP, Dr. A. W. Barrows Aberdeen Proving Ground, MD 21005	1	Lockheed Missiles & Space Co. Attn: Dr. J. Linsk, Code 83-10, Bldg. 154 P. O. Box 504 Sunnyvale, CA 94088	1
Chemical Systems Division Attn: Dr. C. M. Frey P.O. Box 358 Sunnyvale, CA 94086	1	Armament Development and Test Center Attn: Dr. B. G. Craig Eglin Air Force Base, FL 32542	1
Cornell University Attn: Professor F. Rodriguez School of Chemical Engineering Clin Hall, Ithaca, NY 14853	1	Los Alamos National Lab Attn: Dr. R. L. Rabie, WX-2, MS-952 P.O. Box 1663 Los Alamos, NM 87545	1
Defense Technical Information Center DTIC-DDA-2 Cameron Station Alexandria, VA 22314	12	Los Alamos Scientific Lab Attn: Dr. R. Rogers, WX-2 P.O. Box 1663 Los Alamos, NM 87545	1

## DISTRIBUTION (Cont.)

	<u>Copies</u>		<u>Copies</u>
Naval Air Systems Command Attn: Code 330, Mr. R. Brown Washington, DC 20360	1	Naval Sea Systems Command Attn: Mr. R. Beauregard, SEA-64E Washington, DC 20362	1
Naval Air Systems Command Attn: Dr. H. Rosenwasser, AIR-310C Washington, DC 20360	1	Naval Sea Systems Command Attn: Mr. G. Edwards, SEA-62R3 Washington, DC 20362	1
Naval Air Systems Command Attn: Code O3P25, Mr. B. Sobers Washington, DC 20360	1	Naval Ship Engineering Center Attn: Mr. J. Boyle, Materials Branch Philadelphia, PA 19112	1
Naval Weapons Station Attn: Dr. L. R. Rothstein, Assistant Director Naval Explosives Dev. Engineering Dept. Yorktown, VA 23691	1	Naval Sea Systems Command Attn: Mr. C. Christensen, SEA-62R2 Washington, DC 20362	1
Naval Explosives Ordnance Disposal Tech. Center Attn: Code D, Dr. L. Dickinson Indian Head, MD 20640	1	Naval Weapons Center Attn: Code 388, Dr. D. R. Derr China Lake, CA 93555	1
Naval Ordnance Station Attn: Code PM4, Mr. C. L. Adams Indian Head, MD 20640	1	Naval Weapons Center Attn: Code 3205, Mr. L. N. Gilbert China Lake, CA 93555	1
Naval Ordnance Station Attn: Mr. S. Mitchell, Code 5253 Indian Head, MD 20640	1	Naval Weapons Center Attn: Code 3858, Dr. E. Martin China Lake, CA 93555	1
Dean of Research Attn: Dr. W. Tolles Naval Postgraduate School Monterey, CA 93940	1	Naval Weapons Center Attn: Code 3272, Mr. R. McCarten China Lake, CA 93555	1
Naval Research Lab Attn: Code 6100 Washington, DC 20375	1	Naval Weapons Center Attn: Code 385, Dr. A. Nielsen China Lake, CA 93555	1
Naval Research Lab Attn: Code 6510, Dr. J. Schnur Washington, DC 20375	1	Naval Weapons Center Attn: Code 388, Dr. R. Reed, Jr. China Lake, CA 93555	1
		Naval Weapons Center Attn: Code 3205, Dr. L. Smith China Lake, CA 93555	1

## DISTRIBUTION (Cont.)

	<u>Copies</u>		<u>Copies</u>
Naval Weapons Support Center Attn: Code 5042, Dr. B. Douda Crane, IN 47522	1	Thiokol Corporation Attn: Dr. D. A. Flanigan Huntsville Division Huntsville, AL 35807	1
Chief of Naval Technology Attn: MAT-0716, Dr. A. Faulstich Washington, DC 20360	1	Thiokol Corporation Attn: Mr. G. F. Mangum Huntsville Division Huntsville, AL 35807	1
Chief of Naval Material Office of Naval Technology Attn: MAT-0712, LCDR J. Walker Washington, DC 20360	1	Thiokol Corporation Attn: Mr. E. S. Sutton Elkton Division P.O. Box 241 Elkton, MD 21921	1
Naval Ocean Systems Center Attn: Mr. J. McCartney San Diego, CA 92152	1	Thiokol Corporation Attn: Dr. G. Thompson Wasatch Division MS 240, P.O. Box 524 Brigham City, UT 84302	1
Naval Ocean Systems Center Attn: Dr. S. Yamamoto Marine Sciences Division San Diego, CA 91232	1	Thiokol Corporation Attn: Dr. T. F. Davidson Government Systems Group P.O. Box 9258 Ogden, UT 84409	1
Naval Ship Research & Development Center Attn: Dr. G. Bosmajian Applied Chemistry Division Annapolis, MD 21401	1	Thiokol Elkton Division Attn: Dr. C. W. Vriesen P.O. Box 241 Elkton, MD 21921	1
Rohn and Haas Company Attn: Dr. H. Shuey Huntsville, AL 35801	1	Thiokol Wasatch Division Attn: Dr. J. C. Hinshaw P.O. Box 524 Brigham City, UT 83402	1
KAMAN Sciences Suite 1200 1911 S. Jefferson Davis Highway Attn: Dr. J. F. Kincaid Arlington, VA 22202	1	U.S. Army Research Office Chemical & Biological Sciences Division P.O. Box 12211 Research Triangle Park NC 27709	1
Strategic Systems Project Office Attn: Propulsion Unit, Code SP2731 Department of the Navy Washington, DC 20376	1	USA ARRADCOM Attn: Dr. R. F. Walker, DRDAR-LCE Dover, NJ 07801	1
Strategic Systems Project Office Department of the Navy Attn: Mr. E. L. Throckmorton, Room 1048 Washington, DC 20376	1		

## DISTRIBUTION (Cont.)

	<u>Copies</u>		<u>Copies</u>
Propellants and Explosives Defense Equipment Staff Attn: Dr. R. Kelly, Munitions Directorate British Embassy 3100 Massachusetts Ave. Washington, DC 20008	1	Graduate Aeronautical Lab Attn: Professor W. G. Knauss California Institute of Tech. Pasadena, CA 91125	1
AFROL/LK Attn: LTC B. Loving Edwards AFB, CA 93523	1	Georgia Institute of Tech. Attn: Professor E. Price School of Aerospace Engr. Atlanta, GA 30332	1
Institute of Polymer Science Attn: Professor A. N. Gent University of Akron Akron, OH 44325	1	Hercules Aerospace Division Attn: Dr. K. O. Hartman Hercules Incorporated P.O. Box 210 Cumberland, MD 21502	1
Army Ballistic Research Labs ARRADCOM Attn: DRDAR-BLI, Mr. J. M. Frankle Aberdeen Proving Ground, MD 21005	1	IBM Research Lab Attn: Dr. T. L. Smith, D42-282 San Jose, CA 95193	1
Army Ballistic Research Labs Attn: Dr. I. W. May, DRDAR-BLI ARRADCOM Aberdeen Proving Ground, MD 21005	1	Lockheed Missile & Space Co. Attn: Dr. J. P. Marshall Dept. 52-35, Bldg. 204/2 3251 Hanover Street Palo Alto, CA 94304	1
California Institute of Tech. Attn: Professor N. W. Tschoegl Dept. of Chemical Engineering Pasadena, CA 91125	1	Los Alamos National Lab Attn: Ms. J. L. Janney, Mail Stop 920 Los Alamos, NM 87545	1
University of California Attn: Professor M. D. Nicol Dept. of Chemistry 405 Hilgard Avenue Los Angeles, CA 90024	1	Los Alamos National Lab Attn: Dr. J. M. Walsh Los Alamos, NM 87545	1
University of California Attn: Professor A. G. Evans Berkeley, CA 94720	1	Naval Postgraduate School Attn: Prof. R. A. Reinhardt Physics & Chemistry Dept. Monterey, CA 93940	1
Catholic Univ. of America Attn: Professor T. Litovitz Physics Department 520 Michigan Ave., NE Washington, DC 20017	1	Northwestern University Attn: Professor J. D. Achenbach Dept. of Civil Engineering Evanston, IL 60201	1

## DISTRIBUTION (Cont.)

	<u>Copies</u>		<u>Copies</u>
Office of Naval Research Attn: Dr. N. L. Basdekas Mechanics Program, Code 432 Arlington, VA 22217	1	Washington State University Attn: Professor G. D. Duvall Dept. of Physics Pullman, WA 99163	1
Pennsylvania State Univ. Attn: Professor K. Kuo Dept. of Mechanical Engineering University Park, PA 16802	1	Washington State University Attn: Prof. J. T. Dickinson Dept. of Physics Pullman, WA 99163	1
Sandia Laboratories Attn: Dr. S. Sheffield, Division 2513 P.O. Box 5800 Albuquerque, NM 87185	1	University of Maryland Attn: Prof. R. W. Armstrong Dept. of Mechanical Engineering College Park, MD 20742	5
Space Sciences, Inc. Attn: Dr. M. Farber 135 Maple Avenue Monrovia, CA 91016	1	Naval Sea Systems Command Attn: SEA-06R, Dr. D.J. Pastine Washington, DC 20362	1
Washington State University Attn: Prof. Y. M. Gupta Dept. of Physics Pullman, WA 99163	1	Office of Naval Technology Attn: MAT-0712, Dr. E. Zimet Arlington, VA 22217	1
SRI International Attn: Mr. M. Hill 333 Ravenswood Avenue Menlo Park, CA 94025	1	Washington State University Attn: Prof. M. H. Miles Dept. of Physics Pullman, WA 99163	1
Texas A&M Univ. Attn: Prof. R. A. Schapery Dept. of Civil Engineering College Station, TX 77843	1	Solarex Corporation Attn: Dr. K.-C. Yoo 1335 Piccard Drive Rockville, MD 20850	10
Univ. of Utah Attn: Dr. S. Swanson Dept. of Mech. & Industrial Engineering MEB 3008 Salt Lake City, UT 84112	1	Brimrose Corporation of America Attn: Dr. R. G. Rosemeier P.O. Box 6204 Baltimore, MD 21206	10
Thiokol Corporation Attn: Mr. J. D. Byrd Huntsville Division Huntsville, AL 35807	1	National Bureau of Standards Attn: Dr. M. Kuriyama Bldg. 223, Room B266 Washington, DC 20234	1
		Naval Weapons Center Attn: Code 3858, Dr. R. Y. Yee China Lake, CA 93555	1

## DISTRIBUTION (Cont.)

	<u>Copies</u>		<u>Copies</u>
Naval Weapons Center		R11 (Kamlet)	1
Attn: Code 389, Mr. T. Boggs	1	R11 (Leahy)	1
China Lake, CA 93555		R11 (Ringbloom)	1
		R12	1
University of Delaware		R10B	1
Attn: Prof. T. B. Brill	1	R10C	1
Department of Chemistry		R10D	1
Newark, DE 19711		R10F	1
		R13	1
Strategic Systems Project Office		R13 (Clairmont)	1
Department of the Navy		R13 (Coffey)	5
Attn: Mr. J. Culver, Rm 1048	1	R13 (Coleburn)	1
Washington, DC 20376		R13 (DeVost)	1
		R13 (Elban)	25
Naval Research Laboratory		R13 (Forbes)	1
Attn: Code 6360, Dr. C. Cm. Wu	1	R13 (Jacobs)	1
Washington, DC 20375		R13 (Jones)	1
		R13 (Kim)	1
Los Alamos National Lab		R13 (Price)	1
Attn: Dr. H. Cady	1	R13 (Sandusky)	1
Mail Stop 920		R14	1
Los Alamos, NM 87545		R14 (Nicholson)	1
		R15	1
<u>Internal Distribution:</u>		R16	1
R	1	R16 (Hoffsommer)	10
R10	1	R34 (Norr)	1
R11	1	E35	1
R11 (Doherty)	1	E431	9
R11 (Holden)	1	E432	3