

AD

MTL TR 87-20

ENERGY

AD-A181 479

AD-A181 479

STATIC FATIGUE OF A SILICONIZED SILICON CARBIDE

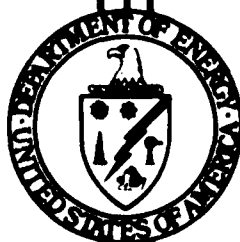
GEORGE D. QUINN

March 1987

DTIC
ELECTE
JUN 1 1 1987
S D
E

Prepared under
Interagency Agreement DE-A105-840R-21411

by
U.S. ARMY MATERIALS TECHNOLOGY LABORATORY
WATERTOWN, MASSACHUSETTS 02172-0001



U.S. DEPARTMENT OF ENERGY
Advanced Materials Development Program
Office of Transportation Systems

This document has been approved
for public release and sale; its
distribution is unlimited.

87

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

REPORT DOCUMENTATION PAGE		READ INSTRUCTIONS BEFORE COMPLETING FORM
1. REPORT NUMBER MTL TR 87-20	2. GOVT ACCESSION NO.	3. RECIPIENT'S CATALOG NUMBER
4. TITLE (and Subtitle) STATIC FATIGUE OF A SILICONIZED SILICON CARBIDE		5. TYPE OF REPORT & PERIOD COVERED Final Report
		6. PERFORMING ORG. REPORT NUMBER
7. AUTHOR(s) George D. Quinn		8. CONTRACT OR GRANT NUMBER(s)
9. PERFORMING ORGANIZATION NAME AND ADDRESS U.S. Army Materials Technology Laboratory Watertown, Massachusetts 02172-0001 SLCMT-MCC		10. PROGRAM ELEMENT, PROJECT, TASK AREA & WORK UNIT NUMBERS Interagency Agreement DE-AI05-840R-21411
11. CONTROLLING OFFICE NAME AND ADDRESS U.S. Department of Energy Office of Transportation Systems Washington, DC 20545		12. REPORT DATE March 1987
		13. NUMBER OF PAGES 14
14. MONITORING AGENCY NAME & ADDRESS (if different from Controlling Office)		15. SECURITY CLASS. (of this report) Unclassified
		15a. DECLASSIFICATION/DOWNGRADING SCHEDULE
16. DISTRIBUTION STATEMENT (of this Report) Approved for public release; distribution unlimited.		
17. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from Report)		
18. SUPPLEMENTARY NOTES		
19. KEY WORDS (Continue on reverse side if necessary and identify by block number) Mechanical properties , High temperature , Ceramic materials , Static fatigue , Silicon carbides , Fractography , Fatigue ,		
20. ABSTRACT (Continue on reverse side if necessary and identify by block number) (SEE REVERSE SIDE)		

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

Block No. 20

ABSTRACT

↓
The static fatigue resistance of a novel siliconized silicon carbide has been evaluated. The material is fabricated by the in-situ reaction of silicon with fine, uniform carbon skeletons produced from liquid polymer solutions. The process has potential to make net shape parts without final machining. A very fine microstructure results which, in turn, gives high strength. Flexural specimens tested in this study had high strength at room and elevated temperatures (up to 1300°C), and had good static fatigue and creep resistance.

(Keywords:)

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

CONTENTS

	Page
INTRODUCTION.	1
MATERIAL CHARACTERIZATION	2
EXPERIMENTAL PROCEDURE.	2
RESULTS	5
DISCUSSION.	9
CONCLUSION.	10
ACKNOWLEDGMENT.	10
REFERENCES.	11

Accession For	
NTIS GRA&I	☒
DTIC TAB	☐
Unannounced	☐
Justification	
By _____	
Distribution/	
Availability Codes	
Dist	Avail and/or Special
A-1	



INTRODUCTION

This work was part of a Department of Energy sponsored program to perform screening tests for room temperature strength, static fatigue, and creep resistance of new heat engine structural ceramics. Previous reports have been prepared for a wide variety of silicon carbide, silicon nitride, and zirconia materials.¹⁻¹⁰ In particular, References 6 and 10 include static fatigue and creep results for the siliconized silicon carbide family of materials.

The siliconized silicon carbide described in this report was prepared at the University of Michigan under contract (DAAG46-80-C-0056) to the Defense Advanced Research Agency. Its development is documented in a series of technical reports.¹¹⁻¹⁴ These reports had a limited distribution and therefore, a detailed introduction on materials processing is included herein. This material was developed using novel techniques intended to produce a uniform and very fine microstructure. The size of strength-limiting flaws is often commensurate with microstructural features. It was hoped that a finer and more uniform microstructure would lead to higher strength than heretofore reported for siliconized silicon carbide. The fabrication process is a novel variation of reaction bonding of silicon with carbon.¹¹⁻¹⁴ The key to this process is obtaining a uniform pure carbon skeleton which is made by casting a low viscosity liquid polymer that is subsequently carbonized. This carbon body is then infiltrated and reacted with silicon metal to form a silicon-silicon carbide composite. The carbon body structure can be accurately controlled by variation of polymer concentration, and polymerization times and temperatures. Conventional powder processes such as grinding, mixing, and pressing are not involved.¹¹⁻¹⁴

Preliminary work established that refining the structure of the carbon skeleton results in a finer final microstructure and that bulk density, pore, and carbon size distributions were key variables.^{11,12} A skeleton density of 0.87 g/cm³ was optimum. Higher values lead to incomplete silicon infiltration, whereas lower values cause strength reduction. Pore size uniformity is crucial so that silicon infiltration is uniform. An optimum pore size of 2 microns permitted up to 15 cm x 10 cm x 5 cm to be infiltrated in 1-2 hours. Carbon size is controlled to prevent skeleton sections that are too thick to react completely or which form coarse silicon carbide zones. Carbon "solid size" should be less than 10 microns. Green bodies with pore and carbon size less than 1 micron have also been produced. Silicon infiltration temperatures were between 1450°C and 1575°C. Residual-free silicon content ranged from 5% to 20%, with 10% typically the goal.¹³

The polymer precursor can be cast or machined prior to carbonization. Carbonization will cause shrinkage, but of a reproducible magnitude. However, warpage during carbonization has to be controlled. In principle, complex shapes can be fabricated by this process, although most of the effort was on plates or rectangular bars.

Processing problems include microstructure coarsening, silicon lake formation, nonuniform surface reaction layers, and incomplete carbon reaction. These difficulties must be overcome by careful control of processing parameters.

Previous mechanical property evaluations have been confined to room temperature flexure strength and fracture toughness.¹¹⁻¹⁴ Four-point flexure strength averages were as high as 714 MPa (105 ksi) with Weibull moduli in the 4-8 range.¹³ Fracture toughness was estimated from a Vickers indentation method to have been in the 3.5 MN/m^{1.5} range.¹³

MATERIAL CHARACTERIZATION

Forty rectangular bars, nominally 2.9 x 3.1 x 50.8 mm in size, were prepared for this study.* These bars were in an as-processed form and had surface irregularities so that they had to be machined to a final size of 2.1 x 2.5 x 50.8 mm. The average bulk density of the machined bars was 3.10 g/cm³ with a 0.02 standard deviation.

Metallographic inspection of several polished sections revealed a range of structures as shown in Figures 1a through 1c. The free silicon content is approximately 20 percent. A quantitative estimate by point or lineal analysis could not be made since the silicon (white phase) actually contains a very fine dispersed silicon carbide structure which would be hard to resolve. For comparison, Figure 2 is a polished section of NC 435 siliconized carbide, which is the finest grain material in the NC 430, NC 433, and NC 435 family.* Residual carbon in the University of Michigan material can be eliminated by high temperature heat treating, although such treatment can lead to microstructure coarsening. The residual carbon is reasonably uniform in size (~10 μ m) and is spherical. A chemical analysis by inductive coupled argon plasma spectroscopy is shown in Table 1. Elemental impurity levels are typical for siliconized silicon carbide.⁵

Table 1. ELEMENTAL IMPURITY ANALYSIS

AL	B	Ca	Co	Cr	Weight Percent		Mg	Mn	Ni	Ti	Zr
					Cu	Fe					
0.09	0.27	0.06	*	0.009	0.13	0.11	0.015	0.031	*	0.025	0.002

*Not Detected

EXPERIMENTAL PROCEDURE

Room temperature strength was evaluated to serve as a reference strength and to determine the nature of strength-limiting flaws. Four-point flexural testing was performed with B configuration (20-mm x 40-mm inner and outer spans) prescribed by MIL-STD-1942 (MR), but the final specimen size (2.1 x 2.5 x 50.8 mm) was not in accordance with the standard due to processing constraints.

High temperature flexural stress rupture and stepped temperature stress rupture (STSR) testing were performed to assess the static fatigue and creep resistances. Isothermal stress rupture experiments were performed at 1200°C in air for comparison to previous results.¹⁻¹⁰ STSR experiments¹⁵ were under deadweight loading, but with each specimen exposed to a range of temperatures from 1000°C to 1300°C. A similar schedule has been used for the testing of other siliconized carbides: 24-hour holds at 1000°C, 1100°C, and 1200°C, and a 72-hour hold at 1300°C. The purpose of this procedure is to quickly assess the temperature and stress levels that static fatigue and creep processes are active. The applied stresses were computed on the basis of the elastic formulation.

*University of Michigan Lot 332-22.

†Norton Company, Worcester, MA, see Reference 5 for additional details.

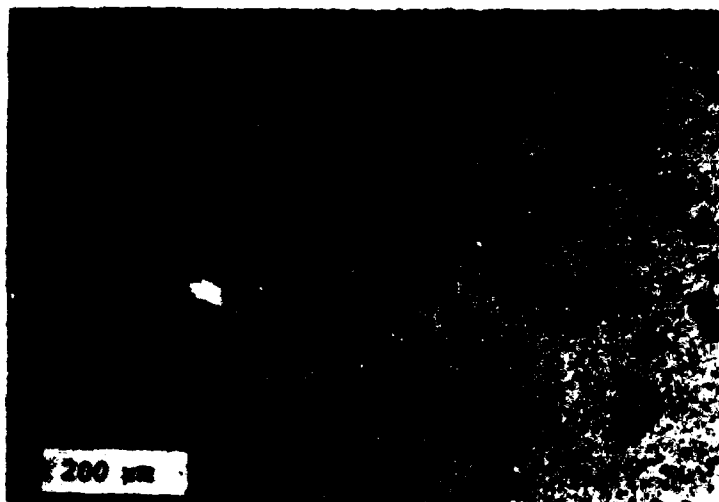


Figure 1a. Polished section illustrating the silicon carbide (grey), silicon (white), and carbon/void (black) phases.

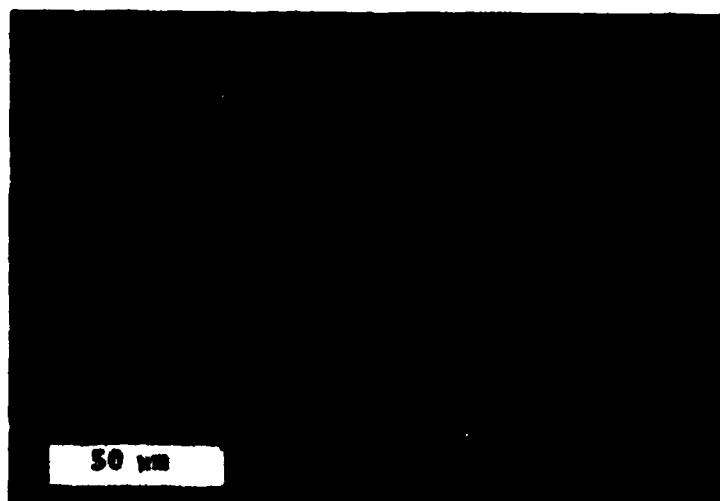


Figure 1b. Higher magnification showing that the white phase is actually complex silicon/silicon carbide structure.

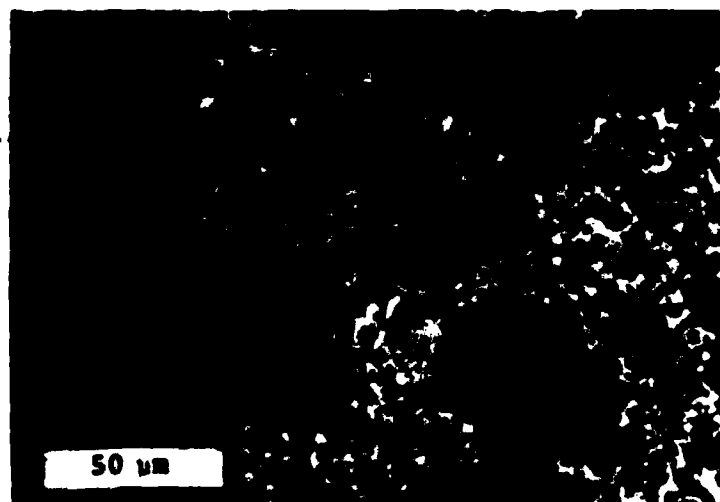


Figure 1c. A zone with excessive silicon carbide grain growth.

Three new high temperature, hot-pressed silicon carbide fixtures were used for this study (Figure 3). The fixture is designed for well-machined specimens and has inner and outer spans of 20 mm and 40 mm, respectively. The inner span loading is through fixed knife edges; the outer span, through rollers that were permitted to roll during a test in order to minimize friction constraints. The rollers were butted against a shoulder at room temperature and tacked in place with a household cement (which subsequently burns off). The specimen and upper load block are inserted and a slight preload applied prior to heating in the furnace. This design is an interim one, pending a more sophisticated jig that has inner span rollers as well.

Specimens which survived stress rupture or STSR testing intact were subsequently measured for weight change, final creep strain, and retained strength. The creep strain was determined by photographic enlarging the inner span curvature, and measuring the midspan deflection relative to the inner span points.* Retained strength was measured at room temperature with the MIL-STD-1942 (MR) B fixtures.

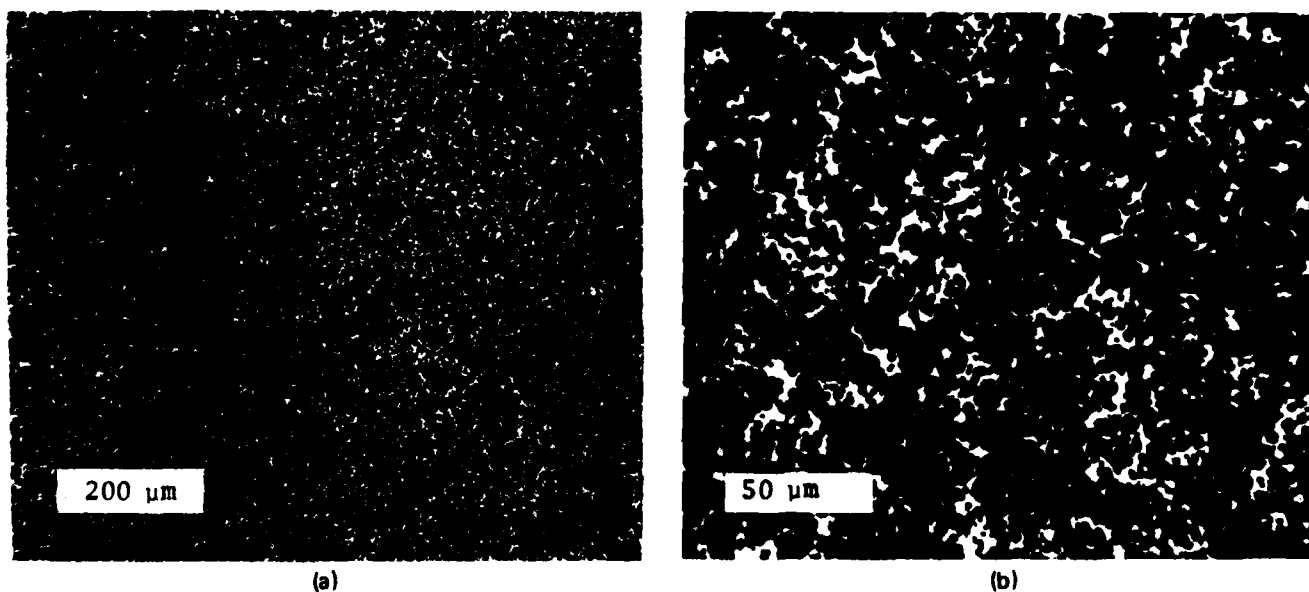


Figure 2. Polished sections of NC 435. Silicon carbide is the grey phase and silicon is in the white phase.



Figure 3. High temperature test fixture. A white alumina specimen is in place.

*Strain = $(4 \times \text{specimen thickness} \times \text{midspan deflection}) / (\text{inner span length})^2$.

RESULTS

The average room temperature strength for 16 specimens was 609 MPa (88 ksi). One very low strength specimen (174 MPa) fractured from a large crack already present (but not noticed prior to testing) as shown in Figure 4. If this datum is deleted, the average strength was 638 MPa (92.5 ksi) with a 144 MPa (21 ksi) standard deviation. The highest strength was 805 MPa (117 ksi). A Weibull two-parameter analysis is shown in Figure 5. The strength values were ordered and assigned probabilities of $i/n+1$ where i is the i th datum and n is the total number of specimens. A least-square fitted line matched the data very well with a slope of 7.5 and a characteristic strength of the bend specimen of 678 MPa.

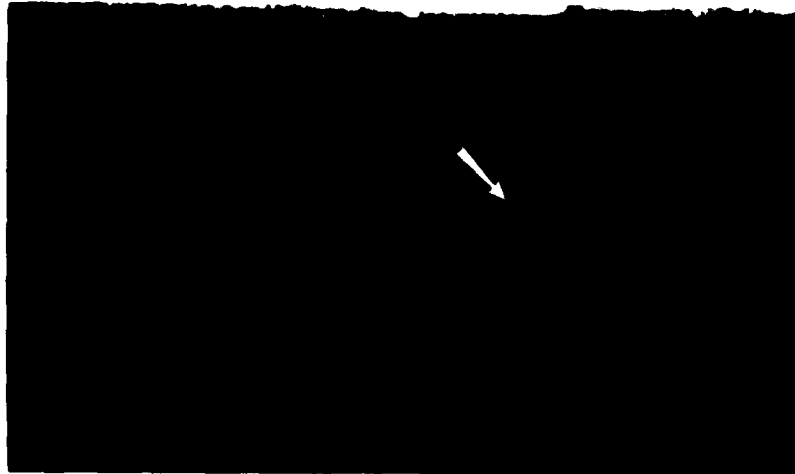


Figure 4. Tensile face of a flexure specimen which fractured at very low strength (174 MPa). The arrow highlights the origin of failure, which is the crack, even though it is oriented 90° to the applied stress.

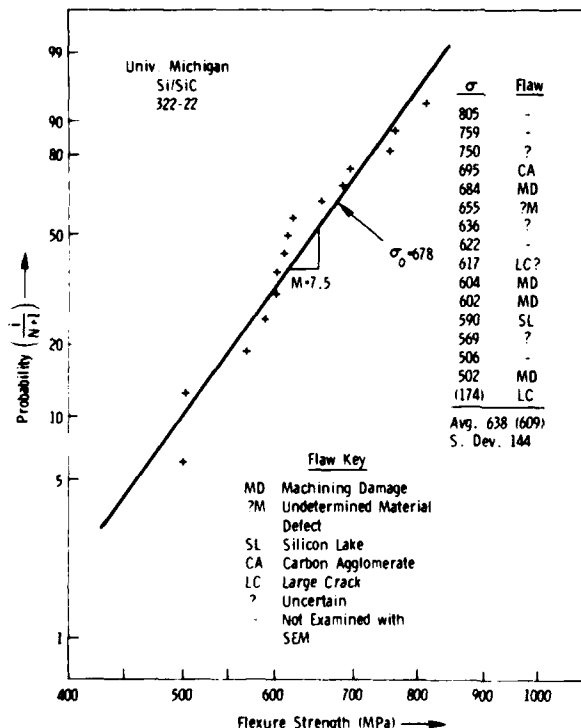


Figure 5. Weibull two-parameter representation of the room temperature reference strengths.

The fracture surfaces had small mirrors which facilitated location of the strength-limiting flaws which unfortunately were too small to characterize optically. A representative group was examined with the SEM which revealed a range of possible flaw types including: machining damage, carbon agglomerates, silicon agglomerates, and sometimes, planar seams of material different in microstructure than the bulk (see Figures 6 and 7). In two instances, it was possible to estimate the fracture toughness K_{IC} based upon the machining flaw size and shape and failure stress (assuming no residual stress). Values of 3.1 and 3.2 $\text{MN/m}^{1.5}$ were obtained. The best estimates of the flaw type are listed on the side of the Weibull plot (Figure 5). Similar defects have been reported previously in the progress reports on the development of this material.^{13,14}

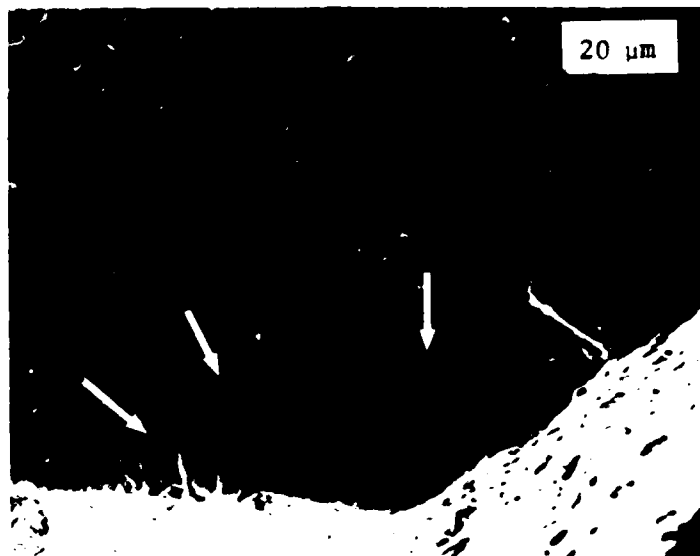


Figure 6. SEM photo of the fracture surface of a specimen revealing the strength-limiting defect is machining damage.

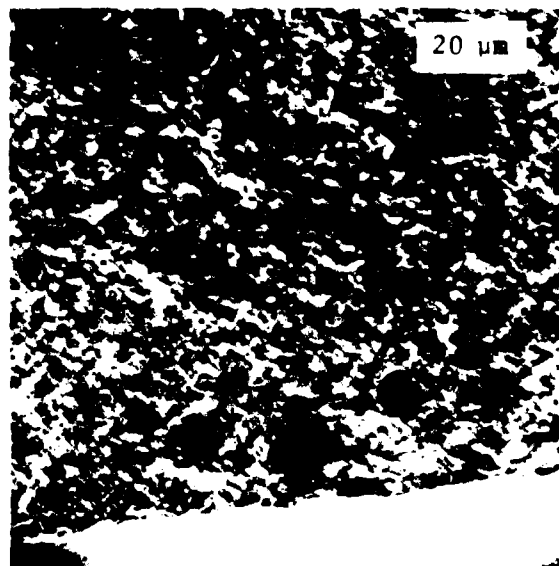


Figure 7. SEM photo of the fracture surface revealing three carbon agglomerates which were strength limiting.

STSR results are given in Figure 8. Three specimens failed in a time-dependent manner, two failed immediately upon loading at 1000°C, and three specimens survived intact. The results suggest that time-dependent failure can occur, but only for highly stressed specimens in the 1000°C to 1300°C range. The fracture origin for the specimen which broke at 0.04 hour at 1000°C was a material defect, probably a pore or a region originally carbon-rich. The origin looked like a fast fracture site and had no slow crack growth (SCG) markings. The 530 MPa specimen, which failed at one hour at 1200°C (after accumulating 24 hours at 1000°C and 1100°C) had machining damage at the origin and also did not have clear time-dependent markings (Figure 9). The fracture surface on the third time-dependent failure, 0.15 hour on the 1300°C step on a 500 MPa stressed specimen, was heavily damaged by impact on the flexural fixtures.

The three STSR survivors had negligible changes in mass (0.26% loss to 0.03% gain). Permanent creep strains were small; 0.20%, 0.13%, and 0.19% for the specimens loaded to 350, 400, and 500 MPa, respectively. Retained strengths at room temperature were 639, 617, and 689 MPa.

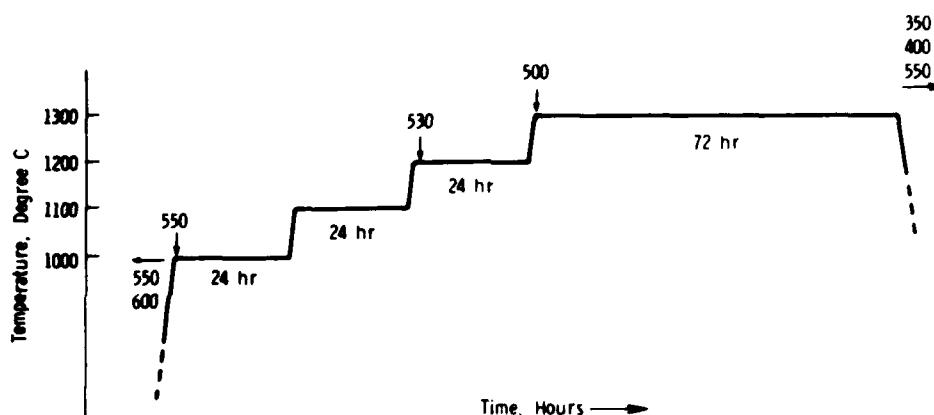


Figure 8. STSR results. Each arrow denotes one (or more) specimen and is labeled with the applied elastic stress in MPa.

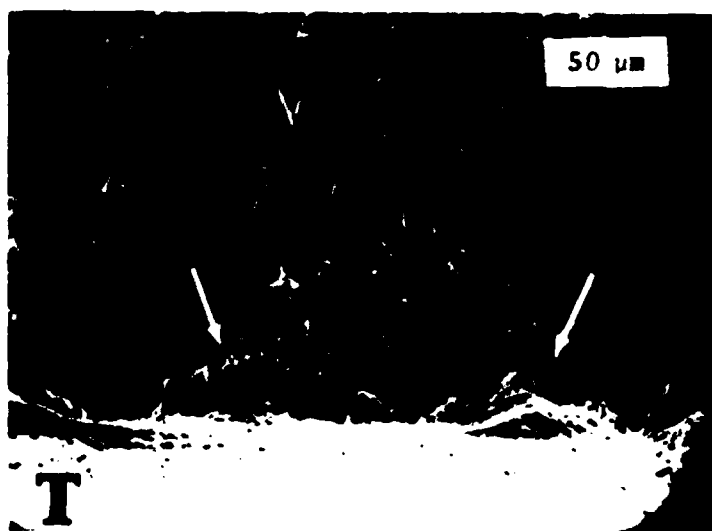


Figure 9. SEM photo of the fracture origin of an STSR specimen which failed at one hour in the 1200°C step. The arrows highlight machining damage.

Isothermal stress rupture experiments at 1200°C are given in Figure 10. Two specimens failed upon loading (arrows left), four failed in time-dependent manner (solid points), and five did not fail (arrows right). One of the latter accumulated 8559 hours (0.98 year) whereupon a furnace malfunction occurred causing temperature to soar to 1725°C. The specimen failed and there was excessive melting and oxidation of the free silicon. The fracture surfaces of the four time-dependent failures were difficult to interpret, sometimes due to excessive impact damage sustained when the specimen struck the fixture after fracture. In one instance, the origin was a zone or seam of material different than the bulk (Figure 11). The origin of the specimen which failed at 2400 hours was a surface-connected pore associated with a zone of microporosity. The three 550 MPa, 1000-hour survivors and the 600 MPa survivor had negligible mass changes from 0.2% loss to 0.04% gain. Permanent creep strains were 0.32%, 0.33%, 0.29% and 0.21%, respectively. The room temperature retained strengths were generally higher than the reference strength (638 MPa); 679, 624, 709, and 966 MPa, respectively.

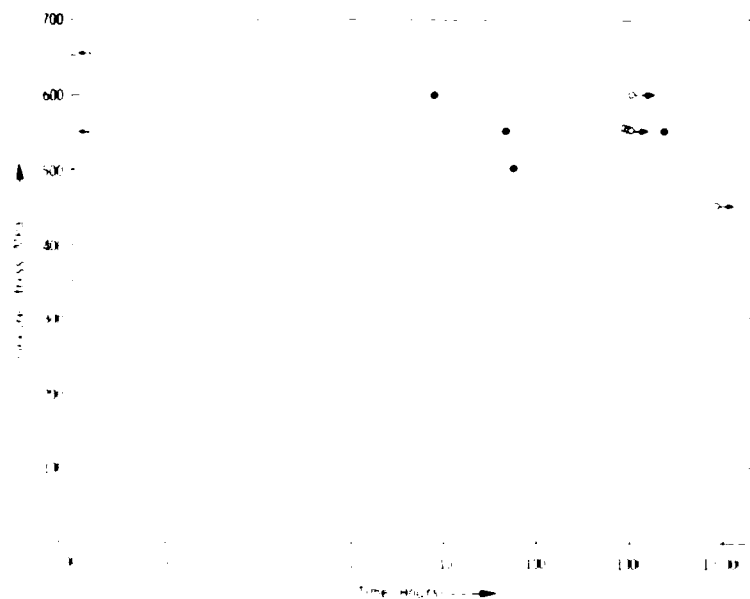


Figure 10. Flexural stress rupture at 1200°C.

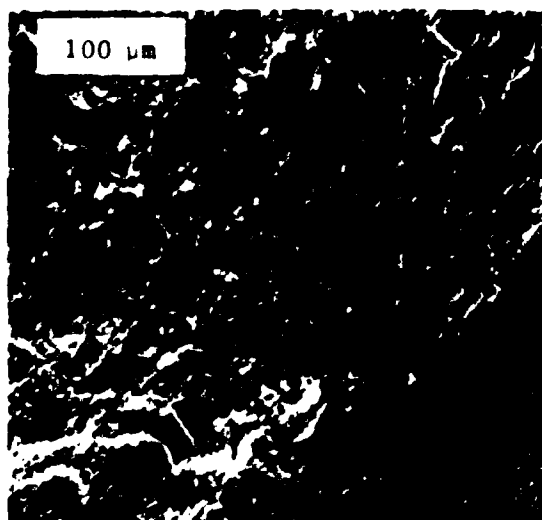


Figure 11. Fracture origin of specimen which failed at 58.5 hours at 1200°C with 500 MPa loading.

Both the STSR and 1200°C stress rupture survivors had a coherent thin surface oxide layer that was interrupted by numerous small spherical pores. These presumably result from the oxidation of surface-connected, spherical carbon pockets such as shown in Figure 1.

DISCUSSION

This material is the latest in a series of siliconized silicon carbides dating back to the REFEL grades¹⁴ and more recently, the NC 430 series¹⁵ and the Silcomp grades.¹⁶ The University of Michigan material uses a novel process involving carbonization of a polymer precursor in order to achieve a much finer microstructure. In principle, this should result in higher strength if strength-limiting flaws scale in size with microstructural features. The results in this report substantiate this approach. The University of Michigan material has the highest strength reported for siliconized silicon carbide and also the finest scaled microstructure. Room temperature flexural strengths reported herein are in good agreement with those presented in the earlier progress reports on this material's development.¹¹⁻¹⁴ Unfortunately, to maintain consistently high strength, the uniformity of the microstructure must be maintained. This is a common and difficult problem in the siliconized silicon carbide family.^{5,14} More careful processing can presumably alleviate this problem, but the presence of coarser surface reaction layers may be difficult to overcome. The low Weibull modulus (7.5) is also consistent with earlier findings^{13,14} and is partly a consequence of a nonuniform microstructure. Very complex multiple flaw populations no doubt contribute to the strength scatter as well.

The flexure specimen preparation in this study was fairly careful and induced subsurface cracks of rather small size (~10 microns). Nevertheless, such flaws often were strength limiting, because of low inherent fracture toughness (~3 MN/m^{1.5}) of siliconized silicon carbide. The implication is that if the microstructure was made even finer and more uniform, machining preparation would limit strength. If specimens were not machined and were tested in an as-processed shape, then surface irregularities (common to siliconized silicon carbide) would limit strength. Thus, the practical upper limit of strength for siliconized silicon carbide may indeed be in the 600-800 MPa range. In any case, the University of Michigan grade evaluated in this report has achieved this practical limit. Any further improvements should be directed at improving the Weibull modulus, eliminating surface coarsened regions, or improving fracture toughness.

The University of Michigan material has high static fatigue and creep resistance. This is common to siliconized silicon carbides wherein time-dependent failures in the 1000°C to 1300°C range only occur at stresses close to the fast fracture strength. Thus, since the University of Michigan material has a very high reference strength, it is able to sustain very high flexural stresses up to 1300°C. In this context, the material outperforms many other silicon carbides and nitrides.^{5,6} The few time-dependent fractures had unclear fracture origin features making identification of a mechanism of static fatigue difficult. Creep fracture could be ruled out due to the small strains (0.2%) involved, unlike NC 435, the finest grain member of the NC 430 series.^{5,6} Indeed, a comparison of the coarser NC 433 to NC 435 showed the latter was more sensitive to slow crack growth and creep fracture.⁵ The higher creep rate in the finer grained material was not surprising. It may therefore be expected that the University of Michigan material which possesses an even finer structure, would have higher creep strains. It did not, despite the very

high applied stresses. The creep resistance of this material, thus, is superior to the other siliconized silicon carbides and many other silicon carbides and nitrides in the 1000°C to 1300°C range.

The specimens which survived the stress rupture and STSR exposures had no strength degradation. Indeed, one such specimen had the highest strength measured in this study (966 MPa).

CONCLUSION

A new siliconized silicon carbide, fabricated by a novel processing technique, has demonstrated very high flexural strength. This is a consequence of the processing method which produces a very fine microstructure. This material probably had recorded the maximum practical fast fracture strength for the siliconized silicon carbide family of materials. Strength scatter is somewhat high, which is typical of siliconized silicon carbides due to microstructural nonuniformity and multiple flaw populations. Static fatigue and creep resistances were very high and are superior to other siliconized silicon carbides, and many silicon carbides and nitrides in the 1000°C to 1300°C range. Failures only occurred for specimens loaded nearly to the fast fracture strength.

ACKNOWLEDGMENT

This research was sponsored by the Advanced Materials Development Program, Office of Transportation Systems, U.S. Department of Energy IEA DE-AI05-84OR 21411 with DOE's Oak Ridge Operation.

REFERENCES

1. QUINN, G. D., and KATZ, R. N. *Stepped Temperature Stress-Rupture Testing of Silicon-Based Ceramics*. American Ceramic Society Bulletin, v. 57, no. 11, 1978, p. 1057.
2. QUINN, G. D., and KATZ, R. N. *Time Dependence of the High Temperature Strength of Sintered Silicon Carbide*. U.S. Army Materials Technology Laboratory, AMMRC TN 79-5, June 1979.
3. QUINN, G. D. *Stress Rupture of Sintered Alpha Silicon Carbide*. U.S. Army Materials Technology Laboratory, AMMRC TN 81-4, December 1981.
4. QUINN, G. D., and KATZ, R. N. *Time-Dependent High Temperature Strength of Sintered Alpha SiC*. J. Am. Cer. Soc., v. 63, no. 1-2, January-February 1980, p. 117.
5. QUINN, G. D. *Characterization of Turbine Ceramics After Long-Term Environmental Exposure*. U.S. Army Materials Technology Laboratory, AMMRC TR 80-15, April 1980.
6. QUINN, G. D. *Review of Static Fatigue in Silicon Nitride and Silicon Carbide*. Ceramic Engineering and Science Proceedings, v. 3, no. 1-2, p. 77-98.
7. SCHIOLER, L. J., QUINN, G. D., and KATZ, R. N. *Time-Temperature Dependence of the Strength of Commercial Zirconia Ceramics*. Proceedings of the 21st Automotive Technology Development - Contractors Coordinator Meeting, Dearborn, MI, November 1983. Published by Society of Automotive Engineers, Warrendale, PA, March 1984; also U.S. Army Materials Technology Laboratory, AMMRC TR 84-16, April 1984.
8. GRESKOVICH, C., PASCO, W., and QUINN, G. D. *Thermomechanical Properties of a New Composition of Sintered Si_3N_4* . American Ceramic Society Bulletin, v. 63, no. 9, September 1984, p. 1165-1170.
9. QUINN, G. D. *Static Fatigue of a Sintered Silicon Nitride*. U.S. Army Materials Technology Laboratory, AMMRC TR 84-40, October 1984.
10. SLAVIN, M. J., and QUINN, G. D. *Mechanical Property Evaluation at Elevated Temperatures for Sintered Beta Silicon Carbide*. U.S. Army Materials Technology Laboratory, AMMRC TR 86-9, March 1986.
11. HUCKE, E. *Process Development for Silicon Carbide Based Structural Ceramics*. University of Michigan, Contract DAAG46-80-C-0056, Interim Report, U.S. Army Materials Technology Laboratory, AMMRC TR 81-13, March 1981. NTIS ADA 098158.
12. HUCKE, E. *Process Development for Silicon Carbide Based Structural Ceramics*. University of Michigan, Contract DAAG46-80-C-0056, Interim Report, U.S. Army Materials Technology Laboratory, AMMRC TR 82-10, February 1982. NTIS ADA 114834.
13. HUCKE, E. *Process Development for Silicon Carbide Based Structural Ceramics*. University of Michigan, Contract DAAG46-80-C-0056, Semiannual Report, U.S. Army Materials Technology Laboratory, AMMRC TR 83-5, January 1983.
14. FORREST, C., KENNEDY, P., and SHENAAN, J. *The Fabrication and Properties of Self Bonded Silicon Carbide Bodies*. Special Ceramics 5, Published by the British Ceramic Research Association Stoke on Trent, June 1972, p. 99-123.
15. TORTI, M., LUCEK, J., and WEAVER, G. *Densified Silicon Carbide - An Interesting Material for Diesel Applications*. SAE Paper 780071, ISSN-0148-7191, Society of Automotive Engineers, Warrendale, PA, 1978.
16. MEHAN, R. *Effect of SiC Content and Orientation on the Properties of Si/SiC Ceramic Composite*. J. Mat. Sci., v. 13, 1978, p. 358-366.

DISTRIBUTION LIST

No. of Copies	To	No. of Copies	To
1	Office of the Under Secretary of Defense for Research and Engineering, The Pentagon, Washington, DC 20301		Department of Energy, 1000 Independence Avenue, Washington, DC 20585
	Commander, Defense Technical Information Center, Cameron Station, Building 5, 5010 Duke Street, Alexandria, VA 22314	3	ATTN: Mr. Richard T. Alpaugh, Dr. R.B. Schultz, and Mr. William McDonough, Office of Vehicle & Engine R&D, CE-131 FORSTL
2	ATTN: DTIC-FDAC	1	Mr. Albert A. Chesnes, Director, Technology Development and Analysis, Office of Vehicle & Engine R&D, CE-131 FORSTL
	Commander, U.S. Army Laboratory Command, 2800 Powder Mill Road, Adelphi, MD 20783-1145	1	Dr. J. J. Eberhardt, Office of Energy Systems Research, CE-142 FORSTL
1	ATTN: SCLIS-IM-TL		Department of Energy, Washington, DC 20545
	Commander, U.S. Army Materiel Command (AMC), 5001 Eisenhower Avenue, Alexandria, VA 22333	1	ATTN: Mr. Stanley J. Dapunkas, Office of Technical Coordination, Fossil Energy Technical Coordination Staff, FE-14 GTN
1	ATTN: AMCLD	1	Dr. John W. Fairbanks, Office of Advanced Energy Conversion, FE-22 GTN
	Commander, U.S. Air Force Wright Aeronautical Laboratories, Wright-Patterson Air Force Base, OH 45433	1	Dr. Robert J. Gotschall, Office of Material Sciences ER-131 GTN
1	ATTN: AFWAL/MLLM, Robert Ruh, Metals and Ceramics Division		Department of Energy, 550 2nd Street, Idaho Falls, ID 83401
1	AFWAL/MLLM, Dr. H. Graham	1	ATTN: Mr. W. H. Thielbahr, Chief, Energy Programs Branch, Idaho Operations Office
1	AFWAL/MLLM, Dr. Allan Katz		Battelle Columbus Laboratories, 505 King Avenue, Columbus, OH 43201
1	AFWAL/POOS, Mr. Everett A. Lake	1	ATTN: Winston Duckworth
1	AFWAL/POX, Mr. Donald J. Campbell	1	Dinesh Shetty
1	AFWAL/POTC, Mr. Don Zabierek	1	Dr. Dale E. Niesz, Manager, Materials Department
	Commander, U.S. Army Belvoir Research and Development Center, Fort Belvoir, VA 22060		Argonne National Laboratory, Building 362, 9700 South Cass Avenue, Argonne, IL 60439
1	ATTN: FTRBE-EMP, Mr. James P. Arnold	1	ATTN: Mr. Larry Johnson, Director, Center for Transportation Research
	Commander, U.S. Army Tank-Automotive Command, Warren, MI 48090		Los Alamos National Laboratory, P.O. Box 1663, Los Alamos, NM 87545
1	ATTN: Mr. W. Bryzik	1	ATTN: Dr. S. S. Hecker, Deputy Division Leader, Material Science and Technology Division, G-756
	Director, U.S. Army Research and Technology Laboratory (AVSCOM), Applied Technology Laboratory, Fort Eustis, VA 23604	1	Frank D. Gac, CMB-6, MS 6770
1	ATTN: SAVDL-ATL-ATP, Mr. Graydon A. Elliot	1	Dr. J. Petrovic, CMB-5, MS 730
	Commander, U.S. Army Mobility Equipment Research and Development Command (MERADCOM), Electrical Power Laboratory, Fort Belvoir, VA 22060	1	Mr. S. R. Skaggs, Program Office, MS F-682
1	ATTN: Mr. Arthur D. Kinser		Oak Ridge National Laboratory, P.O. Box X, Oak Ridge, TN 37831
	Commander, U.S. Army Research Office, P.O. Box 12211, Research Triangle Park, NC 27709-2211	1	ATTN: A.C. Schaffhauser, Building 4508, MS 113
1	ATTN: Dr. J. Hurt	1	Dr. R. Johnson, Building 4508, MS 113
1	Dr. Robert R. Reeber	1	V. J. Tennerly, Building 4500S, B-160
	Chief of Naval Research, 500 North Quincy Street, Arlington, VA 22217	2	J. Carpenter and Dr. K. Lio
1	ATTN: Dr. R. Pohanka		Sandia National Laboratories, P.O. Box 5800, Albuquerque, NM 87185
	Naval Research Laboratory, Washington, DC 20375	1	ATTN: Dr. E. Beaucamp, Ceramics Development Division
1	ATTN: Mr. David Lewis, Code 6363	1	Robert E. Cuthrell
	Chief, Naval Air Propulsion Center, Science and Technology Group, Systems Technology Division, Box 7176, PE 34 Trenton, NJ 08628	1	R. J. Egan
1	ATTN: Mr. Joseph W. Glatz		National Research Council, National Materials Advisory Board, 2101 Constitution Avenue, Washington, DC 20418
	U.S. Bureau of Mines, 2401 Eye Street, N.W., Washington, DC 20241	1	ATTN: Dr. Richard M. Spriggs
1	ATTN: Mr. Murray A. Schwartz	1	Dr. Klaus M. Zwiisky, Executive Director
	National Bureau of Standards, Building 223, Gaithersburg, MD 20899		National Science Foundation, Division of Materials Research, 1800 G Street, N.W., Washington, DC 20550
1	ATTN: Dr. N. Tighe, Deformation and Fracture Division	1	ATTN: Dr. Louis E. Toth
1	Dr. D. Freiman		National Aeronautics and Space Administration, Energy Systems Office, Code REC-1, Washington, DC 20546
1	S. Wiederhorn, Inorganic Materials Division	1	ATTN: Dr. Phillip R. Compton
2	Dr. E. Fuller and T. Mechalske		National Aeronautics and Space Administration, Goddard Space Center, Greenbelt, MD 20771
	Appropriations Committee, Subcommittee on Interior and Related Events, B-308 Rayburn Building, Washington, DC 20515	1	ATTN: Thomas Heslin, Code 313.1
1	ATTN: Mr. Neal Sigmon		National Aeronautics and Space Administration, Lewis Research Center, 21000 Borrkpark Road, Cleveland, OH 44135
	U.S. House of Representatives, Committee on Science and Technology, Suite 2321, Rayburn House Office Building, Washington, DC 20515	1	ATTN: Mr. Thomas P. Herbell, and Dr. S. Dutta, Mail Stop 49-1
1	ATTN: Mr. Harlan L. Watson, Subcommittee on Energy Research and Production	1	Dr. S. Dutta, Mail Stop 49-1
	U.S. House of Representatives, Committee on Science and Technology, Room 2321 Rayburn Building, Washington, DC 20515	1	Dr. H. B. Probst, Chief, Non-Metallic Materials Branch, Materials Division, Mail Stop 49-3
1	ATTN: Mr. Anthony C. Taylor, Staff Director, Subcommittee on Transportation, Aviation and Materials	1	Dr. Michael Millard, Materials Division, 5131
	Department of Defense, 1400 Wilson Boulevard, Arlington, VA 22209	1	Mr. Robert C. Evans, Assistant Manager, Vehicular Gas Turbine and Diesel Project Office
1	ATTN: Dr. Steven G. Wax, Materials Science Division, Advanced Research Projects Agency		Alfred University, 22 South Main Street, Alfred, NY 14802
		1	ATTN: Prof V. D. Frechette

No. of Copies	To
2	Alfred University, New York State College of Ceramics, 22 South Main Street, Alfred, NY 14802
1	ATTN: James R. Varner, Associate Prof. Dr. W. Richard Ott
1	Brown University, Department of Chemical Engineering, Providence, RI 02912
1	ATTN: Prof. Thomas Rockett
2	Case Western Reserve University, Department of Metallurgy, Cleveland, OH 44106
2	ATTN: Prof. A. Cooper and Prof. Arthur H. Heuer
1	Columbia University, Director of Center for Strategic Materials, 1137 S.W., Mudd Building, New York, NY 10027
1	Dr. John K. Tien
1	Georgia Institute of Technology, Atlanta, GA 30331
1	ATTN: EMSL/EES, J. D. Walton
1	Iowa State University, Department of Materials Science and Engineering, Ames, IA 50011
1	ATTN: Dr. Thomas D. McGee
2	Johns Hopkins University, Applied Physics Laboratory, Laurel, MD 20810
2	ATTN: R. Kelly Frazer and Louis D. Weckesser
1	Massachusetts Institute of Technology, Department of Metallurgy and Materials Science, Cambridge, MA 02139
1	ATTN: Prof. K. Bowen, Room 12-009
1	Prof. R. L. Coble
1	Michigan Technical University, Department of Metallurgical Engineering, Houghton, MI 49931
1	ATTN: Dr. B. Pletka
1	North Carolina State University, Materials Engineering Department, 232 Riddick Laboratory, Raleigh, NC 27650
1	ATTN: Prof. Robert F. Davis
1	North Carolina State University, Engineering Research Services Division, 2158 Burlington Engineering Laboratories, P.O. Box 5995, Raleigh, NC 27607
1	ATTN: Dr. Hayne Palmour III
1	Ohio State University, Ceramic Engineering Department, 2041 College Road, Columbus, OH 43210
1	ATTN: Prof. Dennis Readey, Department Chairman
2	Pennsylvania State University, Department of Materials Science and Engineering, Steidle Building, University Park, PA 16802
2	ATTN: Prof. Richard Tressler, and Dr. J. Mecholsky
1	Rensselaer Polytechnic Institute, Materials Engineering Department, Troy, NY 12181
1	ATTN: Prof. R. Doremus
1	Rutgers University, Busch Campus, Box 909, Bowser Road, Piscataway, NJ 08854
1	ATTN: Dr. Malcolm G. McLaren, Head, Department of Ceramics
1	Prof. R. Cannon
1	Rutgers University, Department of Ceramics, P.O. Box 901, Piscataway, NJ 08854
1	ATTN: Dr. John B. Wachtman, Jr.
1	Texas A&M University, Civil Engineering Department, College Station, TX 77843
1	ATTN: Dr. Richard Schapery
1	University of California, Lawrence Berkeley Laboratory, 264 Hearst Mining Building, Berkeley, CA 94720
1	ATTN: Dr. Anthony G. Evans
2	University of California, Department of Materials Science, Berkeley, CA 94720
2	ATTN: Prof. D. Marshall, and Dr. A. Evans
1	University of California, Lawrence Livermore Laboratory, P.O. Box 808, Livermore, CA 94550
1	ATTN: Mr. R. L. Landingham
1	University of California, Department of Materials Science and Engineering, Los Angeles, CA 90024
1	Prof. George Sines
1	University of Delaware, Center for Composite Materials, 2001 Spencer Laboratory, Newark, DE 19716
1	ATTN: Prof. R. Byron Pipes

No. of Copies	To
1	University of Illinois at Urbana-Champaign, Department of Ceramic Engineering, 105 South Goodwin Avenue, Urbana, IL 61801
1	ATTN: Prof. Clifton G. Bergeron, Head, Department of Ceramic Engineering, 204 Ceramics Building
1	Prof. Sherman D. Brown, 203 Ceramics Building
1	Dr. Subhash Risbud
1	University of Massachusetts, Department of Mechanical Engineering, Amherst, MA 01003
1	ATTN: Prof. J. Ritter, Jr.
1	Dr. K. Jakus
2	University of Michigan, Ann Arbor, MI 48109
2	ATTN: Prof. T. Y. Tien and Prof. J. Wallace, Department of Materials and Metallurgical Engineering
1	Prof. E. Huckle
1	University of Missouri - Rolla, 120 Fuller Hall, Rolla, MO 65401-0249
1	ATTN: Prof. Robert Moore
1	University of Washington, Seattle, WA 98195
1	ATTN: Prof. R. Bradt, Mining, Metallurgical and Ceramic Group
3	Dr. A. F. Emery, Prof. Albert S. Kobayashi, and Dr. Raymond Taggart, Department of Mechanical Engineering
1	Charles W. Newquist, Ceramic Engineering Division, FB-10
1	Prof. A. Miller
1	University of Wyoming, Composite Materials Research Group, Mechanical Engineering Department, Laramie, WY 82071
1	ATTN: Prof. Donald F. Adams
1	Virginia Polytechnic Institute and State University, Blacksburg, VA 24061
1	ATTN: Dr. D. P. H. Hasselman, Department of Materials Engineering
1	Prof. K. L. Reifsnider, Department of Engineering Science and Mechanics
1	Washington University at St. Louis, Lindell and Skinker, Box 1087, St. Louis, MO 63130
1	ATTN: Dr. H. T. Hahn, Department of Mechanical Engineering
2	Admiralty Materials Technology Establishment, Polle, Dorset BH 16 6JU, United Kingdom
2	ATTN: Dr. D. Godfrey and Dr. M. Lindley
1	Admiralty Research Establishment, Holton Heath, Polle, Dorset, BH 16 6JU, United Kingdom
1	ATTN: Mr. Keith Pitman
1	CSIRO Advanced Materials Laboratory, P.O. Box 4331, Melbourne, Victoria, Australia 3001
1	ATTN: Dr. M. V. Swain
1	Denki Kagaku Kogyo Kabushiki Kaisha, Pan American Build- ing, 200 Park Avenue, New York, NY 10166
1	ATTN: Mr. Yukio Ohhara
1	Kyocera Corporation, Central Research Laboratory, 1-4 Yamashita - Cho, Kokubo, Kagoshima, 899-43, Japan
1	ATTN: Dr. Y. Hamano
2	Lucas Cookson Syalon, Ltd., Cranmore Boulevard, Shirley- Sulihull, West Midlands, B904LL, United Kingdom
2	ATTN: Dr. E. G. Butler, and J. Lumby
1	Materials Research Laboratories, P.O. Box 50, Ascot Vale, VIC 3032, Australia
1	ATTN: Dr. C. W. Weaver
1	National Institute for Materials Research, P.O. Box 395, Pretoria, 0001, South Africa
1	ATTN: Dr. B. Berger
2	Norton Company, 8001 Daly Street, Niagara Falls, Ontario, Canada L2G 6S2
2	ATTN: Dr. Chris Knapp and Dr. R. Trischuk
1	SNECMA Laboratorie - Section YKOT5, BP 81 91003, Eury Eedex, France
1	ATTN: Dr. Daniel Laroche
1	Stiftelsen Svensk Detonikforskning, Tellusborgsvagen 34-26, P.O. Box 32058, S-126 11 Stockholm, Sweden
1	ATTN: Dr. Finn Ouchterlong

No. of Copies	To
1	GA Technologies, Inc., P.O. Box 81608, San Diego, CA 92138 ATTN: Dr. T. D. Gulden, Manager, Ceramics and Chemistry
1	Gas Research Institute, 8600 West Bryn Mawr Avenue, Chicago, IL 60631 ATTN: Dr. Winston W. Liang, Project Manager, Industrial Materials Research
1	General Electric Company, P.O. Box 527, King of Prussia, PA 19406 ATTN: Dr. Solomon Musikant, Program Manager, Advanced Energy Programs Department
1	General Electric Company, Aircraft Engine Group, Mail Drop H-99, Cincinnati, OH 45215 ATTN: Warren A. Nelson
1	General Electric Company, Mail Drop MB2, I-75 and Neumann Way, Cincinnati, OH 45215 ATTN: Dr. John S. Erickson
3	General Electric Company, P.O. Box 8, Schenectady, NY 12345 ATTN: Dr. Curtis A. Johnson, Dr. R. Giddings, and Dr. C. Greskovich, Research and Development Center Dr. R. J. Charles
1	General Electric Space Division, Rm. U4212, Bldg. 100, P.O. Box 8555, Philadelphia, PA 19101 ATTN: Dr. A. Ismail Abdel-Ilaf
1	General Dynamics Corporation, P.O. Box 2507, Pomona, CA 91766 ATTN: Dr. R. Hallse, Pomona Division
1	AC Spark Plug Division of General Motors, 1300 North Dort Highway, Flint, MI 48556 ATTN: Bernard J. Busovne, Jr. Dr. Frederick L. Kennard, III, Supervisor, Ceramics Research
2	General Motors Technical Center, Research Laboratories, Department 43, General Motors Research, Warren, MI 48090 ATTN: Mr. Bill Alcini, and Jerry A. Barendrecht
3	General Motors Corporation, Allison Gas Turbine Operations, P.O. Box 420, Indianapolis, IN 46206-0420 ATTN: Dr. Peter W. Heitman, W-5, Mr. H. E. Helms, and Michael Janovicz Mr. Harold L. Stocker, T-23, Manager, Low Heat Rejection Program Dr. Pramod Khandelwal W-5
1	Gobal, 50 Milk Street, 15th Floor, Boston, MA 02109 ATTN: Dr. Bruce Rubinger
1	GTE Laboratories, 40 Sylvan Road, Waltham, MA 02154 ATTN: Dr. J. Thomas Smith, Director, Precision Materials Technology Dr. Arvid E. Pasto, Mr. J. Neal, Dr. M. L. Huckabee, Dr. J. G. Baldoni, and Dr. T. Butler
3	GTE Wesgo, 477 Harbor Boulevard, Belmont, CA 94002 ATTN: Mr. C. Edmiston, Dr. D. Milovic, and Dr. Howard Mizuhara
1	Honeywell Ceramics Center, 1885 Douglas Drive, Golden Valley, MN 55422 ATTN: Mr. W. Harrison
2	Honeywell, Inc., 10701 Lyndale Avenue South, Bloomington, MN 55420 ATTN: Dr. B. Koepke, and Dr. K. McHenry
1	Howmet Turbine Components Corporation, 1600 South Warner Road, Whitehall, MI 49461 ATTN: Mr. Larry G. McCoy, Ceramic Products Division
1	IBM Corporation, 1701 North Street, Endicott, NY 13732 ATTN: Dr. Brent Knifer, U48/022-3
1	Jet Propulsion Laboratories, 4800 Oak Grove Drive, Pasadena, CA 91103 ATTN: William A. Edmiston, Building 506, Room 432 James G. Zwissler, Member Tech. Staff Dr. Richard Smoak
1	Kennametal, Inc., Box 639, Greensburg, PA 15601 ATTN: Dr. Bernard North
1	Kyocera International, 2300 East Devon Avenue, Suite 253, Des Plaines, IL 60018 ATTN: Nick Kamiya

No. of Copies	To
1	Lanxide Corporation, Tralee Industrial Park, Newark, DE 19711 ATTN: Mr. Stanley Lutz
1	Lockheed Missiles and Space Company, Inc., P.O. Box 504, Sunnyvale, CA 94086 ATTN: C. D. Himmelblau, Sr. Materials Engineer, 0/81-33, B/154
1	LTV Vought Missiles and Advanced Programs Division, P.O. Box 225907, Dallas, TX 75265 ATTN: Mr. Douglas Freitag, Mail Stop TH-85
1	Maremont Corporation, 291 North Street, Saco, ME 04072 ATTN: Ernest Bunning, Saco Defense Systems Division
1	Martin Marietta Aerospace, P.O. Box 5837, Orlando, FL 32855 ATTN: Earle A. Welsh, 305/352-3435, Mail Point 275
1	Martin Marietta Laboratories, 1450 South Rolling Road, Baltimore, MD 21227-3898 ATTN: Dr. Albert R. C. Westwood, Corporate Director Dr. J. Venables
2	Mechanical Technology, Inc., 968 Albany-Shaker Road, Latham, NY 12110 ATTN: Mr. Warren F. Bessler, Manager, Advanced Programs Branch S. Gray and W. P. Sumigray
1	Norton Company, High Performance Ceramics, Goddard Road, Northboro, MA 02153 32-1545 ATTN: Mr. Robert J. Russell, Divisional Vice President, Technology and Planning, High Performance Ceramics Dr. Anil K. Agarwall, Product Manager Dr. Joseph Panzarino, Dr. P. O. Charreyron, Dr. James Hannoosh, and Maurice L. Torti Mr. R. Yeckley, Mr. N. Corbin, and Mr. L. Chuck
1	ORI, Inc., 1375 Piccard Drive, Rockville, MD 20910 ATTN: Mr. William L. Cleary, Associate Division Director
1	Pratt and Whitney Aircraft, P.O. Box 2691, West Palm Beach, FL 33402 ATTN: Mr. Mel Mendelson
3	Raytheon Missile Systems Division, Bedford, MA 01730 ATTN: Dr. Peter L. Boland, Joel Spearr, and Bruce C. Center, Staff Engineer
2	Raytheon Research Division, 131 Spring Street, Lexington, MA 02173 ATTN: Dr. R. Gentilman and Dr. E. Maguire
1	Rockwell International, 6633 Canoga Avenue, Canoga Park, CA 91304 ATTN: Mr. Harold Carpenter, M/S D539/BA92
1	Rockwell International Science Center, P.O. Box 1085, 1049 Camino Dos Rios, Thousand Oaks, CA 91360 ATTN: Dr. Frederic Lange
1	Signal UOP Corporate Research Center, 10 UOP Plaza, Des Plaines, IL 60016 ATTN: Dr. Steven Gonczy
1	SONIO Engineered Materials Company, P.O. Box 1054, Niagara Falls, NY 14302 ATTN: Dr. Jonathan W. Hinton, Vice President and General Manager, Structural Ceramics
3	SONIO Engineered Materials Company, P.O. Box 832, Niagara Falls, NY 14302 ATTN: Dr. M. Srinivasan, Dr. S. Seshadri, and G. Weber, Advanced Materials Division
1	Solar Energy Research Institute, 1617 Cole Boulevard, Golden, CO 80401 ATTN: Mr. Gary Jorgensen
1	Solar Turbines, Inc., P.O. Box 80966, 2200 Pacific Highway, San Diego, CA 92138 ATTN: Dr. Arthur Metcalfe Andrew Russell, and J. Napier, MS R-1
1	Southern Research Institute, P.O. Box 3307-A, Birmingham, AL 35205 ATTN: Mr. H. S. Starrett
1	Southwest Research Institute, Culebra Road, San Antonio, TX 78284 ATTN: Dr. J. Lankford

No. of Copies	To
1	Toyota Central Research and Development Laboratories, 502 Group, Nagakute-Cho, Aichi-Gun, Aichi-Ken 480-11, Japan ATTN: Dr. Shigeki Kobayashi
1	University of Newcastle Upon Tyne, Cookson Group, Newcastle on Tyne, NE1 7 RU, United Kingdom ATTN: Prof. K. Jack
1	Dr. Maxine Savitz, 5019 Lowell Street, Washington, DC 20016
1	Adiabatics, Inc., 630 S. Mapleton, Columbus, IN 47201 ATTN: Dr. Roy Kamo, President
1	Advanced Refractory Technologies, Inc., 699 Hertel Avenue, Buffalo, NY 14207 ATTN: Dr. Peter T. B. Shaffer
1	The Aerospace Corporation, P.O. Box 92957, Los Angeles, CA 90009 ATTN: W. U. Roessler, M3-392
1	Air Products and Chemicals, Inc., Box 538, Allentown, PA 18105 ATTN: Mr. Ken Baumert
4	AiResearch Coating Company, 19800 Van Ness Avenue, Torrance, CA 90509 ATTN: Mr. Karsten H. Styhr, R. Hengst, D. Schwab, and D. Kotchick
4	AiResearch/Garrett Turbine Engine Company, P.O. Box 5217, 111 South 34th Street, Phoenix, AZ 85010 ATTN: Dr. D. Carruthers, Dave G. Finger, Dr. J. Smythe, and Dr. J. Wimmer
1	AiResearch Manufacturing Company, 2525 West 190th Street, Torrance, CA 90509 ATTN: David Kotchick, 3-42, Department 93-18
1	ALCOA Laboratories, Alcoa Center, PA 15069 ATTN: Mr. Richard Rolf, Product Engineering Division
1	Amercom, Inc., 8948 Fullbright Avenue, Chatsworth, CA 91311 ATTN: Mr. R. E. Fisher, President
1	AM General Corporation, 14250 Plymouth Road, Detroit, MI 48232 ATTN: Steven L. Sanborn
2	ARCO Metals Company, Ceramic Materials & Process Technology, Silag Operation, Route 6, Box A, Greer, SC 29651 ATTN: Mr. James F. Rhodes and Dr. Ronald L. Beatty
1	AYCO, Lowell Industrial Park, Lowell, MA 01887 ATTN: Dr. T. Vasilos, Applied Technology Division
5	AYCO Lycoming Division, 550 South Main Street, Stratford, CT 06497 ATTN: Dr. Sharon Trombley, D. E. Wilson, Hsianmin Jen, Robert Johnston, and Mr. Zoltal Libertini
1	Babcock and Wilcox, P.O. Box 1260, Lynchburg, VA 24505 ATTN: Mr. Michael Rigdon
1	Bell Laboratories, Inc., 555 Union Blvd., Allentown, PA 18103 ATTN: Bhaskar B. Ghate
1	The Bendix Corporation, 900 W. Maple Road, Troy, MI 48064 ATTN: Paul C. Becker, Ph.D., Manager, Engineering Materials, Materials Technology Center, Automotive Technical Center
1	The Boeing Company, Seattle, WA 98124 ATTN: Mr. J. Van Wyk, Materials & Processes Laboratory, Aerospace Group
1	Brunswick Corporation, 150 Johnston Road, Marion, VA 24354 ATTN: Harold R. Mabe, Defense Division
2	Garry D. Leonard and Robert L. Copeland
1	Caterpillar Tractor Company, 100 N.E. Adams, Peoria, IL 61629 ATTN: Mr. John M. Bailey, Research Consultant, Research Department, Technical Center
1	Ceradyne Inc., 3030-A South Red Hill Avenue, Santa Ana, CA 92705 ATTN: Dr. Richard Palicka

No. of Copies	To
1	Ceramic Finishing Company, P.O. Box 498, State College, PA 16801 ATTN: Dr. H. P. Kirchner
2	Champion Spark Plug Company, 20000 Conner Avenue, Detroit, MI 48234 ATTN: Mr. William Manning and Mr. Paul Rempes
1	Chrysler Corporation, Box 1118, Detroit, MI 48288 ATTN: John M. Corwin, CIMS 418-36-15
1	Combustion Engineering Inc., 911 West Main Street, Chattanooga, TN 37402 ATTN: C. H. Sump
2	Coors Porcelain Company, 600 Ninth Street, Golden, CO 80401 ATTN: Kenneth Green, Eng. IV Supervisor, and Mr. Donald Roy
1	Coors Porcelain Company, 17750 North 32nd Street, Golden, CO 80401 ATTN: Dr. David Wirth, Vice President, Technical Operations Research and Development
1	Corning Glass Works, Sullivan Park, Corning, NY 14830 ATTN: Mr. J. G. Lanning, Advanced Engine Components, MP-88-2
5	Herbert A. Miska, Main Plant, MS 8-05-5 Dr. K. Chyung, Dr. M. Taylor, Donna L Taylor, Ms. S. Stuckley, and Mr. D. Larsen
2	Cummins Engine Company, Inc., Box 3005, Columbus, IN 47202 ATTN: Dr. James W. Patten, Director, Materials Engineering, Mail Code 50183, and Dr. George Wolter
1	Dr. T. Yonushonis
1	Deere and Company Technical Center, 3300 River Drive, Moline, IL 61265 ATTN: Dr. Gopal Revankar, Metals Research
1	Delta Q Labs, P.O. Box 64, West Nottingham, NH 03291 ATTN: Dr. Dale Rice
1	DHR, Inc., 6858 Old Dominion Drive, McLean, VA 22101 ATTN: Dr. Richard Silbergliitt
1	Dow Corning Corporation, 3902 S. Saginaw Road, Midland, MI 48640 ATTN: Dr. Frank Chi
1	Dow Corning Corporation, 3901 S. Saginaw Road, Midland, MI 48640 ATTN: Mr. H. W. Foglesong
1	Dow Corning Corporation, Midland, MI 48000 ATTN: Mr. Neal Langley
1	The Dow Chemical Company, M. E. Pruett Bldg. - 1776, Central Research Inorganic Lab., Midland, MI 48640 ATTN: William Rafaniello
1	Eaton Corporation, 26201 Northwestern Highway, P.O. Box 766, Southfield, MI 48037 ATTN: Kenneth A. Hatcher
1	Mr. Thomas J. Wissing, Manager, Government Contract Administration, Engineering and Research Center
2	EG&G, Inc., Materials Technology Division, Idaho National Laboratories, P.O. Box 1625, Idaho Falls, ID 83415 ATTN: Dr. Victor Kelsey, and Dr. Gary Korth
1	EATON Operations and Technical Center, 4201 North 27th St. Milwaukee, WI 53216 ATTN: Dr. T. Kuczkowski, Materials Research
2	EPRI, P.O. Box 10412, 3412 Hillview Avenue, Palo Alto, CA 94304 ATTN: Dr. A. Cohn and Mr. K. Kinsman
1	Fiber Materials, Inc., Biddeford Industrial Park, Biddeford, ME 04005 ATTN: Dr. H. D. Batha
1	Ford Motor Company, P.O. Box 2053, Dearborn, MI 48121 ATTN: Dr. A. McLean, Ceramics Materials Department
1	Lewis R. Swank, Rm. E-3172, SRI
1	Dr. R. Govila
1	The Garrett Corporation, 9851 Sepulveda Boulevard, P.O. Box 92248, Los Angeles, CA 90009 ATTN: Dr. John Mason, Vice President - Engineering

No. of Copies	To
	Systems Research Laboratories, 2800 Indian Ripple Road, Dayton, OH 45440
1	ATTN: Dr. M. G. Mandiratta
	SRI International, 333 Ravenswood Avenue, Menlo Park, CA 94025
2	ATTN: Dr. David Rowcliffe and Dr. Phil Umhold
	Tam Ceramics, Inc., Box C, Bridge Station, Niagara Falls, NY 14305
1	ATTN: Roy C. Holmes, Technical Product Manager
	Teledyne-CAE, P.O. Box 6971, Toledo, OH 43612
1	ATTN: Mr. R. Beck
	Teledyne Continental Motors, 76 Getty Street, Muskegon, MI 49442
1	ATTN: William H. Schimanski, and Michael Moynihan
	Teresco, Inc., P.O. Box 2269, Station A, Champaign, IL 61820
1	ATTN: Morris Berg
	TRW/Martin Rockwell Division, 402 Chandler Street, Jamestown, NY 14701
1	ATTN: Michael Franklin
	United Technologies Research Center, Silver Lane, East Hartford, CT 06108
1	ATTN: Dr. F. Galasso
1	Karl M. Prewé, and Dr. J. Brennan, MS-24

No. of Copies	To
	Varian/EIMAC, 301 Industrial Way, San Carlos, CA 94070
1	ATTN: Mr. S. William Kessler, Materials and Processing Laboratory
	W. R. Grace and Company, Washington Research Center 7379 Route 32, Columbia, MD 21044
1	ATTN: Dr. Roy Rice, Director, Materials Research
1	Kevin E. Bennet
	Warner-Ishi, 707 Southside Drive, Decatur, IL 62525
1	ATTN: Hideaki Matsuoka, Director of Advanced Technology
	Westinghouse Research and Development Center, 1310 Beulah Road, Pittsburgh, PA 15235
3	ATTN: Dr. D. Wilson, and Dr. S. C. Singhal, Research Laboratory
1	Dr. Raymond J. Bratton, Manager, Ceramic Science
	Williams International, 2280 West Maple Road, P.O. Box 200, Walled Lake, MI 48088
3	ATTN: Mr. William Chappment, Mike Egan, and Warren J. Pollington
	Director, U.S. Army Materials Technology Laboratory, Watertown, MA 02172-0001
2	ATTN: SLCMT-IML
1	Author

☆ U.S. GOVERNMENT PRINTING OFFICE: 1987-701-587/80018

U.S. Army Materials Technology Laboratory
Watertown, Massachusetts 02172-0001
STATIC FATIGUE OF A SILICONIZED
SILICON CARBIDE - George D. Quinn

Technical Report MTL TR 87-20, March 1987,
14 pp, illus-table, Interagency Agreement
DE-A105-840R-21411

AD

UNCLASSIFIED
UNLIMITED DISTRIBUTION

Key Words

Mechanical properties
Ceramic materials
Silicon carbides

The static fatigue resistance of a novel siliconized silicon carbide has been evaluated. The material is fabricated by the in-situ reaction of silicon with fine, uniform carbon skeletons produced from liquid polymer solutions. The process has potential to make net shape parts without final machining. A very fine microstructure results which, in turn, gives high strength. Flexural specimens tested in this study had high strength at room and elevated temperatures (up to 1300°C), and had good static fatigue and creep resistance.

U.S. Army Materials Technology Laboratory
Watertown, Massachusetts 02172-0001
STATIC FATIGUE OF A SILICONIZED
SILICON CARBIDE - George D. Quinn

Technical Report MTL TR 87-20, March 1987,
14 pp, illus-table, Interagency Agreement
DE-A105-840R-21411

AD

UNCLASSIFIED
UNLIMITED DISTRIBUTION

Key Words

Mechanical properties
Ceramic materials
Silicon carbides

The static fatigue resistance of a novel siliconized silicon carbide has been evaluated. The material is fabricated by the in-situ reaction of silicon with fine, uniform carbon skeletons produced from liquid polymer solutions. The process has potential to make net shape parts without final machining. A very fine microstructure results which, in turn, gives high strength. Flexural specimens tested in this study had high strength at room and elevated temperatures (up to 1300°C), and had good static fatigue and creep resistance.

U.S. Army Materials Technology Laboratory
Watertown, Massachusetts 02172-0001
STATIC FATIGUE OF A SILICONIZED
SILICON CARBIDE - George D. Quinn

Technical Report MTL TR 87-20, March 1987,
14 pp, illus-table, Interagency Agreement
DE-A105-840R-21411

AD

UNCLASSIFIED
UNLIMITED DISTRIBUTION

Key Words

Mechanical properties
Ceramic materials
Silicon carbides

The static fatigue resistance of a novel siliconized silicon carbide has been evaluated. The material is fabricated by the in-situ reaction of silicon with fine, uniform carbon skeletons produced from liquid polymer solutions. The process has potential to make net shape parts without final machining. A very fine microstructure results which, in turn, gives high strength. Flexural specimens tested in this study had high strength at room and elevated temperatures (up to 1300°C), and had good static fatigue and creep resistance.

U.S. Army Materials Technology Laboratory
Watertown, Massachusetts 02172-0001
STATIC FATIGUE OF A SILICONIZED
SILICON CARBIDE - George D. Quinn

Technical Report MTL TR 87-20, March 1987,
14 pp, illus-table, Interagency Agreement
DE-A105-840R-21411

AD

UNCLASSIFIED
UNLIMITED DISTRIBUTION

Key Words

Mechanical properties
Ceramic materials
Silicon carbides

The static fatigue resistance of a novel siliconized silicon carbide has been evaluated. The material is fabricated by the in-situ reaction of silicon with fine, uniform carbon skeletons produced from liquid polymer solutions. The process has potential to make net shape parts without final machining. A very fine microstructure results which, in turn, gives high strength. Flexural specimens tested in this study had high strength at room and elevated temperatures (up to 1300°C), and had good static fatigue and creep resistance.