

DTIC FILE COPY

ADF 302006

4

AD-A194 678

TECHNICAL REPORT BRL-TR-2896

BRL

1938 - Serving the Army for Fifty Years - 1988

**IMPACT SENSITIVITY OF HAN-BASED
LIQUID PROPELLANTS AT
ELEVATED TEMPERATURES**

IRVIN C. STOBIE
BRUCE D. BENSINGER
JOHN D. KNAPTON

**DTIC
ELECTE
JUN 13 1988**
S D
H

MARCH 1988

Reproduced From
Best Available Copy

APPROVED FOR PUBLIC RELEASE; DISTRIBUTION UNLIMITED.

U.S. ARMY LABORATORY COMMAND

**BALLISTIC RESEARCH LABORATORY
ABERDEEN PROVING GROUND, MARYLAND**

88 6 13 100

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE

REPORT DOCUMENTATION PAGE				Form Approved OMB No. 0704-0188	
1a. REPORT SECURITY CLASSIFICATION Unclassified			1b. RESTRICTIVE MARKINGS		
2a. SECURITY CLASSIFICATION AUTHORITY			3. DISTRIBUTION / AVAILABILITY OF REPORT		
2b. DECLASSIFICATION / DOWNGRADING SCHEDULE					
4. PERFORMING ORGANIZATION REPORT NUMBER(S) BRL-TR-2896			5. MONITORING ORGANIZATION REPORT NUMBER(S)		
6a. NAME OF PERFORMING ORGANIZATION US Army Ballistic Rsch Lab		6b. OFFICE SYMBOL (if applicable) SLCBR-IB	7a. NAME OF MONITORING ORGANIZATION		
6c. ADDRESS (City, State, and ZIP Code) Aberdeen Proving Ground, MD 21005-5066			7b. ADDRESS (City, State, and ZIP Code)		
8a. NAME OF FUNDING / SPONSORING ORGANIZATION		8b. OFFICE SYMBOL (if applicable)	9. PROCUREMENT INSTRUMENT IDENTIFICATION NUMBER		
8c. ADDRESS (City, State, and ZIP Code)			10. SOURCE OF FUNDING NUMBERS		
		PROGRAM ELEMENT NO.	PROJECT NO.	TASK NO.	WORK UNIT ACCESSION NO.
11. TITLE (Include Security Classification) IMPACT SENSITIVITY OF HAN-BASED LIQUID PROPELLANTS AT ELEVATED TEMPERATURES					
12. PERSONAL AUTHOR(S) Stobie, Irvin C., Bensinger, Bruce D., and Knapton, John D.					
13a. TYPE OF REPORT TR		13b. TIME COVERED FROM _____ TO _____		14. DATE OF REPORT (Year, Month, Day)	15. PAGE COUNT
16. SUPPLEMENTARY NOTATION The work summarized in this report was presented at the 23rd JANNAF Combustion Meeting.					
17. COSATI CODES			18. SUBJECT TERMS (Continue on reverse if necessary and identify by block number)		
FIELD	GROUP	SUB-GROUP			
00	11				
19	01				
19. ABSTRACT (Continue on reverse if necessary and identify by block number)					
<p>A Technoproducts drop weight tester was used to evaluate impact sensitivity of four liquid monopropellants; OTTO II, NOS 365, LP 1845 and LP 1846. The monopropellants were evaluated for impact sensitivity at ambient temperatures and at temperatures in excess of 50°C. The propellants were all relatively insensitive to impact compared to the calibration liquid n-propyl nitrate at ambient temperatures. The impact sensitivity increased slightly for the HAN-based monopropellants at temperatures up to 50 degrees. The impact sensitivity of OTTO II increased dramatically as a function of increased temperature.</p>					
20. DISTRIBUTION / AVAILABILITY OF ABSTRACT <input type="checkbox"/> UNCLASSIFIED/UNLIMITED <input checked="" type="checkbox"/> SAME AS RPT <input type="checkbox"/> DTIC USERS			21. ABSTRACT SECURITY CLASSIFICATION Unclassified		
22a. NAME OF RESPONSIBLE INDIVIDUAL Irvin C. Stobie			22b. TELEPHONE (Include Area Code) (301) 278-6155	22c. OFFICE SYMBOL SLCBR-IB-B	

TABLE OF CONTENTS

	<u>Page</u>
LIST OF TABLES	v
I. INTRODUCTION	1
II. APPROACH	1
III. EXPERIMENTAL	2
IV. AMBIENT TEMPERATURE TESTS	2
V. ELEVATED TEMPERATURE TESTS	3
VI. CONCLUSIONS	5
REFERENCES	7
DISTRIBUTION LIST	9

Accession For	
NTIS GRA&I	<input checked="" type="checkbox"/>
DTIC TAB	<input type="checkbox"/>
Unannounced	<input type="checkbox"/>
Justification	
By _____	
Distribution/	
Availability Codes	
Dist	Avail and/or Special
A-1	

LIST OF TABLES

<u>Table</u>		<u>Page</u>
1	Ambient Temperature Monopropellant Drop Weight Results	3
2	Ambient Results and Temperature Ranges and Results for Elevated Temperature Drop Weight Tests.	3
3	Sensitivity Results of OTTO II Conditioned at Temperatures Less Than 70°C	4
4	LP 1846 Drop Weight Sensitivity After Being Conditioned at Temperatures Lee Than 82°C.	5

I. INTRODUCTION

Impact energy tests have been used extensively to evaluate sensitivity of energetic liquids.¹ These tests have been well defined in procedure, and have been performed on a variety of energetic liquids. The drop weight tests are useful as a sensitivity rating device for combustion type environments as well as an indication of behavior under rough handling storage conditions. The HAN-based liquid propellants have been evaluated in other drop weight tests^{2 3} at ambient temperatures, however the tests in the present study were expanded to include tests above the maximum temperature requirements for military ammunition.

II. APPROACH

Impact tests are designed to test compression ignition of propellants in a partially filled chamber. The test procedure is defined in ASTM standard tests D2540-70.⁴ The tests are performed by placing 0.03 ml of the test liquid in a sample cup containing an "O" ring to determine the initial chamber. A 0.41 mm thick stainless steel diaphragm and a piston with a vent orifice are placed on the O ring in the sample cup. The sample chamber is completed by placing the sample cup in a body and assembling with a steel ball and cap as shown in Figure 1. A weight of two kilograms is dropped from a measured height and a positive or negative reaction is recorded. A positive test result is determined by the rupture of the diaphragm and the consumption of the liquid monopropellant. A series of tests are performed on the liquid propellant to determine a height that yields 50% positive results with the 2 kg weight. The ASTM standard states that monopropellants with E50 (defined as the 50% ignition condition) values of 2 kg-cm or less, such as nitroglycerin and diethylene glycol dinitrate, are to be considered sensitive explosives and should be handled with extreme caution.⁴

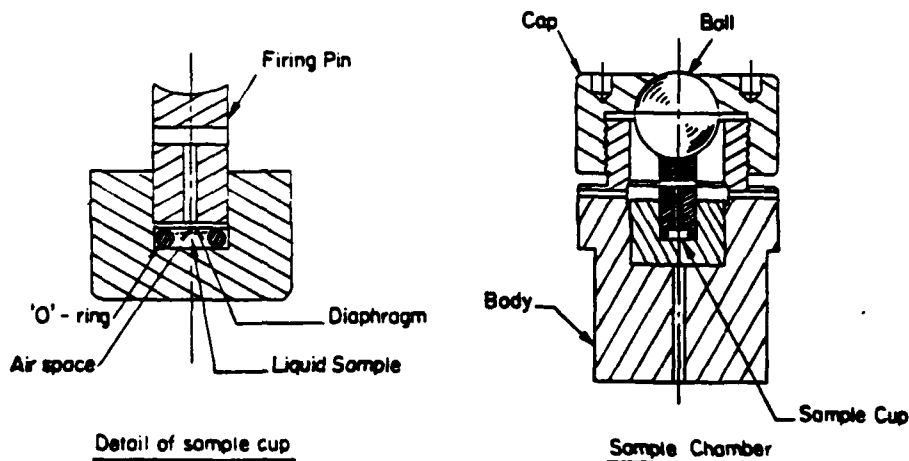


Figure 1. Drop Weight Assembly

III. EXPERIMENTAL

Four monopropellants were chosen for evaluation in this study. OTTO II was evaluated because it has been used extensively in bulk loaded and regenerative liquid propellant guns by many groups. Although Mason et al. report a wide range of the E50 values for OTTO II,⁵ the monopropellant was considered an excellent material for calibration of the drop weight tester and an example of a shock insensitive material. The other three monopropellants were NOS-365, LP 1845, and LP 1846 which were all hydroxyl ammonium nitrate (HAN) based. The NOS-365 monopropellant has been used extensively in bulk loaded firing programs^{6,7} in medium and small caliber guns. The LP 1845 and LP 1846 are considered leading candidates for liquid propellant guns.

In addition to the monopropellant drop weight tests, a series of tests were performed on the liquid n-propyl nitrate, which is used in ASTM D2540 as a reference test liquid. The initial tests of ambient HAN based propellants and the n-propyl nitrate were performed in 1982. An E50 value of 12.8 kg-cm for n-propyl nitrate was measured as compared to the ASTM value of 8.4 kg-cm. This value was considered reasonably consistent with the ASTM value and other reported sensitivities. Subsequent sensitivity tests performed in 1985 and 1986 were also preceded by tests with n-propyl nitrate. No consistent results were noted with three samples of n-propyl nitrate. Subsequent gas chromatography tests indicated the presence of contamination of isopropyl nitrate.⁸

The drop weight tester was standardized by comparing ambient sensitivities of the liquid monopropellants to the earlier 1982 tests where uncontaminated n-propyl nitrate samples were available. The same lots of liquid monopropellants were used throughout the tests. The results at ambient temperatures for all the liquid propellants agreed well with the 1982 ambient tests.

IV. AMBIENT TEMPERATURE TESTS

The liquid propellants were evaluated at 20 degrees plus or minus one degree Centigrade. These results were compared to ambient tests performed by other investigators. Havens reported in Reference 1 that some caution should be used in comparing results from different drop weight tests devices. Cruice (Reference 2) reported sensitivity values above the 100 kg-cm maximum achievable with the instrument we used with a 2 kg drop weight. Attempts to duplicate sensitivity values near 150 kg-cm would have resulted in a positive result, (perforated diaphragm), with only water as the test liquid. The results of the ambient tests along with results from other sources are shown in Table 1. The monopropellant LP 1846 had not been previously evaluated in a drop weight tester. The 50% ignition values of 98 kg-cm for OTTO II and NOS-365 indicate highly shock insensitive materials. Only 33% positive tests were noted for LP 1845 at the maximum drop weight, and LP 1846 sensitivity was beyond the capabilities of the instrument.

TABLE 1. Ambient Temperature Monopropellant Drop Weight Results

PROPELLANT	EARLIER TESTS		PRESENT STUDY
	50% IGNITION kg-cm	REFERENCE	% IGNITION %, kg-cm
OTTO II	8.5-70	5	50%, 98
NOS-365	>100	3	50%, 98
LP 1845	152	2	33%, 100
LP 1846	---		>100

V. ELEVATED TEMPERATURE TESTS

The elevated temperature tests were performed by keeping the sample cup, piston and liquid propellant in a temperature conditioning oven prior to firing. The temperature was recorded immediately before the weight was dropped by means of a thermocouple on the sample cup. Table 2 summarizes the sensitivity results and the temperature ranges for the four monopropellants.

TABLE 2. Ambient Results and Temperature Ranges and Results for Elevated Temperature Drop Weight Tests.

PROPELLANT	TEMPERATURE °C	50% IGNITION kg-cm
OTTO II	ambient 34-54	98 <52
NOS-365	ambient 34-49	98 93
LP 1845	ambient 37-57	(33%-100) 88
LP 1846	ambient 31-53	>100 96

The results with OTTO II at elevated temperatures were somewhat confusing. The initial temperature conditioned tests yielded results that were borderline positive reactions such as liquid residue in the cup and minimum noise from the reaction. As more temperature conditioned tests were performed, the reactions became more complete even at shorter drop heights. It was speculated that the temperature conditioning had an effect on the propellant. A new sample of OTTO II was therefore tested. The ambient sensitivity was checked with previous

tests and the elevated temperature tests were performed. For this sample of propellant the conditioning temperatures were monitored to assure that the sample was never above 70°C. Once again the propellant apparently became more sensitive as a function of time in the conditioning over. The results of the monitored temperature conditioned tests are shown in Table 3.

TABLE 3. Sensitivity Results of OTTO II Conditioned at Temperatures Less Than 70°C

TESTS	ENERGY kg-cm	TEMPERATURE °C	POSITIVE RESULTS
1-5	96-100	38-40	3
6-10	80-92	38-45	4
11-15	68-74	35-45	4

LP 1846 was chosen for further evaluation at the monitored temperature conditions. This propellant, along with LP 1845, is considered a likely candidate for future weapons testing. LP 1845 differs from LP 1846 by three percent less water. It is interesting to note that the sensitivity for both the ambient and elevated tests indicated slightly more sensitivity for LP 1845 as compared to LP 1846.

A sample of LP 1846, along with the sample cup and piston, was conditioned for at least 30 minutes to temperatures up to 58°C. Fourteen tests were performed at drop weight heights of 100 kg-cm at temperatures between 32° and 42°C. Three of the fourteen tests were recorded as positive results. The conditioning temperature was raised to a value less than 78°C for the next series of tests. There were five positive tests noted at 100 kg-cm values and no positive reactions in four attempts at 98 kg-cm. A new sample of LP 1846 was conditioned for at least 30 minutes at temperatures up to 82°C. The results of this conditioning are shown in Table 4. The obvious change in sensitivity in the 18 June tests are noted. The sample was sealed and stored and the 20 June ambient tests were performed with no apparent change in sensitivity. The sample was then conditioned to a temperature that had little effect on the previous sample. The 20 June tests 6-15 revealed the unusual sensitivity shown in Table 4. A Carl Fisher titration of the conditioned sample and a sample from the same lot of propellant was performed. The conditioned sample contained 15.05% water, while the unconditioned sample from the same lot contained 19.32% water.⁹ The dramatic shift in sensitivity between the 78° and 82°C was apparently caused by a change in the chemical composition of the propellant.

TABLE 4. LP 1846 Drop Weight Sensitivity After Being Conditioned at Temperatures Less Than 82°C.

TEST NOS.	TEMPERATURE °C	ENERGY kg-cm	RESULTS
18 June 51-57	54-62	100	2/4 Positive
		98	1/1 Positive
		96	1/1 Positive
		80	1/1 Positive
20 June 1-5	22	100	0/5 Positive
20 June 6-15	44-48	100	1/1 Positive
		94	2/2 Positive
		92	2/4 Positive
		90	0/2 Positive

VI. CONCLUSIONS

1. The four liquid monopropellants are insensitive to the maximum drop weight condition at ambient temperatures.
2. The drop weight sensitivity increased dramatically for the elevated temperature tests with OTTO II. Even when conditioning temperatures were held under 58°C, the sensitivity remained quite high.
3. The drop weight sensitivity increased slightly for the three HAN based monopropellants at temperatures up to 57°C.
4. The sensitivity of LP 1846 increased dramatically at conditioning temperatures between 78 and 82°C. This shift in drop weight sensitivity is believed to be due to the loss of water in the liquid propellant.

REFERENCES

1. "A Review of Hazard Assessment Procedures for Liquid Gun Propellants," BRL-CR-537, U. of Arkansas, November 1984.
2. Cruice, W.J., "Classification of Liquid Gun Propellants and Raw Materials for Transportation and Storage," ARBRL-CR-00454, Hazards Research Corp., May 1981.
3. Smith, B., Harrison, J., Gibbs, R., and Garrison, J., "Binary Explosives," Naval Surface Weapons Center Report NSWC/DL TR-3214, October 1974.
4. "Standard Test Method for Drop-Weight Sensitivity of Liquid Monopropellants," ASTM D2540-70 (Reapproved 1976), pp. 562-573.
5. Mason, C.M., Ribovich, J., and Weiss, M.L., "Safety and Combustion Characteristics of Homogeneous and Heterogeneous Monopropellant Systems," Bureau of Mines Semi-Annual Summary Report No. 3811, July 1, 1960 to December 31, 1960.
6. Knapton, J.D., et al., "Analysis of Ignition and Combustion of Hydroxylammonium Nitrate Based Liquid Monopropellants in a Medium Caliber Bulk Loaded Gun," ARBRL-TR-02487, (C) March 1983.
7. "Small Caliber, Multipurpose, Automatic Liquid Propellant Gun," Pulsepower Systems Incorporated, Report No. TR-65, November 1973.
8. Private communication with Dr. P. Duff, Ballistic Research Laboratory, 1986.
9. Private communication with Ms. M. Decker, Ballistic Research Laboratory, 1986.

DISTRIBUTION LIST

<u>No. of Copies</u>	<u>Organization</u>	<u>No. of Copies</u>	<u>Organization</u>
12	Commander Defense Technical Info Center ATTN: DTIC-DDA Cameron Station Alexandria, VA 22304-6145	3	Director Benet Weapons Laboratory Armament R&D Center US Army AMCCOM ATTN: SMCAR-LCB-TL E. Conroy A. Graham Watervliet, NY 12189
1	Director Defense Advanced Research Projects Agency ATTN: H. Fair 1400 Wilson Boulevard Arlington, VA 22209	1	Commander US Army Armament, Munitions and Chemical Command ATTN: SMCAR-ESP-L Rock Island, IL 61299-7300
1	HQDA DAMA-ART-M Washington, DC 20310	1	Commander US Army Aviation Research and Development Command ATTN: AMSAV-E 4300 Goodfellow Blvd St. Louis, MO 63120
1	Commander US Army Materiel Command ATTN: AMCDRA-ST 5001 Eisenhower Avenue Alexandria, VA 22333-0001	1	Commander Materials Technology Lab US Army Laboratory Cmd ATTN: SLCMT-MCM-SB M. Levy Watertown, MA 02172-0001
13	Commander Armament R&D Center US Army AMCCOM ATTN: SMCAR-TSS SMCAR-TDC SMCAR-SCA, B. Brodman R. Yalamanchili SMCAR-AEE-B, D. Downs A. Beardell SMCAR-LCE, N. Slagg SMCAR-AEE-B, W. Quine A. Bracuti J. Lannon SMCAR-CCH, R. Price SMCAR-FSS-A, L. Frauen SMCAR-FSA-S, H. Liberman Picatinny Arsenal, NJ 07806-5000	1	Director US Army Air Mobility Rsch and Development Lab Ames Research Center Moffett Field, CA 94035
		1	Commander US Army Communications Electronics Command ATTN: AMSEL-ED Fort Monmouth, NJ 07703

DISTRIBUTION LIST

<u>No. of</u> <u>Copies</u>	<u>Organization</u>	<u>No. of</u> <u>Copies</u>	<u>Organization</u>
1	Commander ERADCOM Technical Library ATTN: STET-L Ft. Monmouth, NJ 07703-5301	1	Director US Army TRADOC Systems Analysis Activity ATTN: ATAA-SL White Sands Missile Range NM 88002
1	Commander US Army Harry Diamond Labs ATTN: SLCHD-TA-L 2800 Powder Mill Rd Adelphi, MD 20783	1	Commandant US Army Infantry School ATTN: ATSH-CD-CSO-OR Fort Benning, GA 31905
1	Commander US Army Missile Command Rsch, Dev, & Engr Ctr ATTN: AMSMI-RD Redstone Arsenal, AL 35898	1	Commander Armament Rsch & Dev Ctr US Army Armament, Munitions and Chemical Command ATTN: SMCAR-CCS-C, T Hung Picatinny Arsenal, NJ 07806-5000
1	Commander US Army Missile & Space Intelligence Center ATTN: AIAMS-YDL Redstone Arsenal, AL 35898-5500	1	Commandant US Army Field Artillery School ATTN: ATSF-CMW Ft Sill, OK 73503
1	Commander US Army Belvoir R&D Ctr ATTN: STRBE-WC Tech Library (Vault) B-315 Fort Belvoir, VA 22060-5606	1	Commandant US Army Armor Center ATTN: ATSB-CD-MLD Ft Knox, KY 40121
1	Commander US Army Tank Automotive Cmd ATTN: AMSTA-TSL Warren, MI 48397-5000	1	Commander US Army Development and Employment Agency ATTN: MODE-TED-SAB Fort Lewis, WA 98433
1	Commander US Army Research Office ATTN: Tech Library PO Box 12211 Research Triangle Park, NC 27709-2211	1	Commander Naval Surface Weapons Center ATTN: D.A. Wilson, Code G31 Dahlgren, VA 22448-5000
		1	Commander Naval Surface Weapons Center ATTN: Code G33, J. East Dahlgren, VA 22448-5000

DISTRIBUTION LIST

<u>No. of Copies</u>	<u>Organization</u>	<u>No. of Copies</u>	<u>Organization</u>
2	Commander US Naval Surface Weapons Ctr ATTN: O. Dengel K. Thorsted Silver Spring, MD 20902-5000	1	Director Jet Propulsion Lab ATTN: Tech Library 4800 Oak Grove Drive Pasadena, CA 91109
1	Commander Naval Weapons Center China Lake, CA 93555-6001	2	Director National Aeronautics and Space Administration ATTN: MS-603, Tech Lib MS-86, Dr. Povinelli 21000 Brookpark Road Lewis Research Center Cleveland, OH 44135
1	Commander Naval Ordnance Station ATTN: C. Dale Code 5251 Indian Head, MD 20640	1	Director National Aeronautics and Space Administration Manned Spacecraft Center Houston, TX 77058
1	Superintendent Naval Postgraduate School Dept of Mechanical Engr ATTN: Code 1424, Library Monterey, CA 93943	10	Central Intelligence Agency Office of Central Reference Dissemination Branch Room GE-47 HQS Washington, DC 20502
1	AFWL/SUL Kirtland AFB, NM 87117	1	Central Intelligence Agency ATTN: Joseph E. Backofen HQ Room 5F22 Washington, DC 20505
1	Air Force Armament Lab ATTN: AFATL/DLODL Eglin AFB, FL 32542-5000	3	Bell Aerospace Textron ATTN: F. Boorady F. Picirillo A.J. Friona PO Box One Buffalo, NY 14240
1	Commandant USAFAS ATTN: ATSF-TSM-CN Ft Sill, OK 73503-5600	1	Calspan Corporation ATTN: Tech Library PO Box 400 Buffalo, NY 14225
1	US Bureau of Mines ATTN: R.A. Watson 4800 Forbes Street Pittsburgh, PA 15213		

DISTRIBUTION LIST

<u>No. of Copies</u>	<u>Organization</u>	<u>No. of Copies</u>	<u>Organization</u>
7	General Electric Ord Sys Div ATTN: J. Mandzy, OP43-220 R.E. Mayer H. West M. Bulman R. Pate I. Magoon J. Scudiere 100 Plastics Avenue Pittsfield, MA 01201-3698	1	Science Applications, Inc. ATTN: R. Edelman 23146 Cumorah Crest Woodland Hills, CA 91364
1	General Electric Company Armament Systems Department ATTN: D. Maher Burlington, VT 05401	1	Sundstrand Aviation Operations ATTN: Mr. Owen Briles PO Box 7202 Rockford, IL 61125
1	IITRI ATTN: Library 10 W. 35th St Chicago, IL 60616	1	Veritay Technology, Inc. ATTN: E.B. Fisher 4845 Millersport Highway PO Box 305 East Amherst, NY 14051-0305
1	Olin Chemicals Research ATTN: David Gavin PO Box 586 Cheshire, CT 06410-0586	1	Director Applied Physics Laboratory The Johns Hopkins Univ. Johns Hopkins Road Laurel, MD 20707
2	Olin Corporation ATTN: Victor A. Corso Dr. Ronald L. Dotson PO Box 30-9644 New Haven, CT 06536	2	Director CPIA The Johns Hopkins Univ. ATTN: T. Christian Tech Library Johns Hopkins Road Laurel, MD 20707
1	Paul Gough Associates ATTN: Paul Gough PO Box 1614 Portsmouth, NH 03801	1	U. of Illinois at Chicago ATTN: Professor Sohail Murad Dept of Chemical Engr Box 4348 Chicago, IL 60680
1	Safety Consulting Engr ATTN: Mr. C. James Dahn 5240 Pearl St Rosemont, IL 60018	1	U. of MD at College Park ATTN: Professor Franz Kasler Department of Chemistry College Park, MD 20742

DISTRIBUTION LIST

<u>No. of Copies</u>	<u>Organization</u>	<u>No. of Copies</u>	<u>Organization</u>
1	U. of Missouri at Columbia ATTN: Professor R. Thompson Department of Chemistry Columbia, MO 65211	3	University of Delaware Department of Chemistry ATTN: Mr. James Cronin Professor Thomas Brill Mr. Peter Spohn Newark, DE 19711
1	U. of Michigan ATTN: Prof. Gerard M. Faeth Dept of Aerospace Engr Ann Arbor, MI 48109-3796		<u>Aberdeen Proving Ground</u>
1	U. of Missouri at Columbia ATTN: Professor F.K. Ross Research Reactor Columbia, MO 65211		Dir, USAMSAA ATTN: AMXSY-D AMXSY-MP, H. Cohen
1	U. of Missouri at Kansas City Department of Physics ATTN: Prof. R.D. Murphy 1110 East 48th Street Kansas City, MO 64110-2499		Cdr, USATECOM ATTN: AMSTE-TO-F
1	Pennsylvania State University Dept of Mechanical Engr ATTN: Prof. K. Kuo University Park, PA 16802		Cdr, CRDEC, AMCCOM ATTN: SMCCR-RSP-A SMCCR-MU SMCCR-SPS-IL
2	Princeton Combustion Rsch Laboratories, Inc. ATTN: N.A. Messina M. Summerfield 475 US Highway One North Monmouth Junction, NJ 08852		
1	University of Arkansas Dept of Chemical Engr ATTN: J. Havens 227 Engineering Building Fayetteville, AR 72701		

USER EVALUATION SHEET/CHANGE OF ADDRESS

This Laboratory undertakes a continuing effort to improve the quality of the reports it publishes. Your comments/answers to the items/questions below will aid us in our efforts.

1. BRL Report Number _____ Date of Report _____

2. Date Report Received _____

3. Does this report satisfy a need? (Comment on purpose, related project, or other area of interest for which the report will be used.) _____

4. How specifically, is the report being used? (Information source, design data, procedure, source of ideas, etc.) _____

5. Has the information in this report led to any quantitative savings as far as man-hours or dollars saved, operating costs avoided or efficiencies achieved, etc? If so, please elaborate. _____

6. General Comments. What do you think should be changed to improve future reports? (Indicate changes to organization, technical content, format, etc.) _____

CURRENT ADDRESS

Name

Organization

Address

City, State, Zip

7. If indicating a Change of Address or Address Correction, please provide the New or Correct Address in Block 6 above and the Old or Incorrect address below.

OLD ADDRESS

Name

Organization

Address

City, State, Zip

(Remove this sheet, fold as indicated, staple or tape closed, and mail.)

FOLD HERE

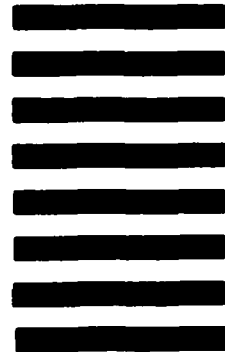
Director
US Army Ballistic Research Laboratory
ATTN: DRXBR-OD-ST
Aberdeen Proving Ground, MD 21005-5066



NO POSTAGE
NECESSARY
IF MAILED
IN THE
UNITED STATES

OFFICIAL BUSINESS
PENALTY FOR PRIVATE USE, \$300

BUSINESS REPLY MAIL
FIRST CLASS PERMIT NO 12062 WASHINGTON, DC
POSTAGE WILL BE PAID BY DEPARTMENT OF THE ARMY



Director
US Army Ballistic Research Laboratory
ATTN: DRXBR-OD-ST
Aberdeen Proving Ground, MD 21005-9989

FOLD HERE