

MTL TR 89-41

AD

2

AD-A211 308

EFFECT OF ANNEALING CONDITIONS ON THE STRUCTURAL AND SUPERCONDUCTING PROPERTIES OF Y-Ba-Cu-O FILMS

JAE RYU, YIZHOU HUANG, and CARMINE VITTORIA
NORTHEASTERN UNIVERSITY

DANIEL F. RYDER, Jr.
TUFTS UNIVERSITY

JAMES V. MARZIK
U.S. ARMY MATERIALS TECHNOLOGY LABORATORY
MATERIALS SCIENCE BRANCH

RICHARD BENFER and WILLIAM A. SPURGEON
U.S. ARMY MATERIALS TECHNOLOGY LABORATORY
CERAMICS RESEARCH BRANCH

DTIC
ELECTE
AUG 16 1989
S D & D

May 1989

Approved for public release; distribution unlimited.



US ARMY
LABORATORY COMMAND
MATERIALS TECHNOLOGY LABORATORY



U.S. ARMY MATERIALS TECHNOLOGY LABORATORY
Watertown, Massachusetts 02172-0001

89

The findings in this report are not to be construed as an official Department of the Army position, unless so designated by other authorized documents.

Mention of any trade names or manufacturers in this report shall not be construed as advertising nor as an official indorsement or approval of such products or companies by the United States Government.

DISPOSITION INSTRUCTIONS

Destroy this report when it is no longer needed.
Do not return it to the originator.

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

REPORT DOCUMENTATION PAGE		READ INSTRUCTIONS BEFORE COMPLETING FORM
1. REPORT NUMBER MTL TR 89-41	2. GOVT ACCESSION NO.	3. RECIPIENT'S CATALOG NUMBER
4. TITLE (and Subtitle) EFFECT OF AN ... CONDITIONS ON THE STRUCTURAL ... SUPERCONDUCTING PROPERTIES OF Y-Ba-C ... FILMS		5. TYPE OF REPORT & PERIOD COVERED
7. AUTHOR Jae Rye,* Yizhou Huang,* Carmine Vittoria,* Daniel F. Ryder, Jr.,† James V. Marzik, Richard Benfer, and William A. Spurgeon		6. PERFORMING ORG. REPORT NUMBER
9. PERFORMING ORGANIZATION NAME AND ADDRESS U.S. Army Materials Technology Laboratory Woburn, Massachusetts 02172-0001 ATTN: SLCMT-EMS		8. CONTRACT OR GRANT NUMBER(s)
11. CONTROLLING OFFICE NAME AND ADDRESS U.S. Army Laboratory Command 100 Powder Mill Road Aberdeen, Maryland 20783-1145		10. PROGRAM ELEMENT, PROJECT, TASK AREA & WORK UNIT NUMBERS D/A Project: 1L161102AH42
14. MONITORING AGENCY NAME & ADDRESS (if different from Controlling Office)		12. REPORT DATE May 1989
		13. NUMBER OF PAGES 11
		15. SECURITY CLASS. (of this report) Unclassified
		15a. DECLASSIFICATION/DOWNGRADING SCHEDULE
16. DISTRIBUTION STATEMENT (of this Report) Approved for public release; distribution unlimited.		
17. DISTRIBUTION STATEMENT (of the abstract entered in Block 20, if different from Report)		
18. SUPPLEMENTARY NOTES *Northeastern University †Tufts University (See Reverse - Block 18. Continued)		
19. KEY WORDS (Continue on reverse side if necessary and identify by block number) Superconductors Electrical properties Ceramic oxide Ceramic materials Oxides Thin films Electroceramics		
20. ABSTRACT (Continue on reverse side if necessary and identify by block number) (SEE REVERSE SIDE)		

UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

Block 18. Continued

To be published in Proceedings of the Conference on the Science and Technology of Thin Film Superconductors, Colorado Springs, CO, Nov. 14-18, 1988.

Block 20.

ABSTRACT

The effect of various annealing treatments on ion beam sputter-deposited $Y_1Ba_2Cu_yO_{7-x}$ ($y=3$ and 5) amorphous thin films has been studied using scanning electron microscopy (SEM), X-ray diffraction and dc resistivity measurements. SEM analysis showed that the morphology of the films were clearly divided into three categories: an equiaxial grain structure, a well-oriented grid-like microstructure and a textured platelet structure. The equiaxial grain structure (random orientation according to X-ray diffraction measurements) was obtained when the film was annealed at slow heating and cooling rates in an oxygen atmosphere. The grid-like microstructure (b/c-axes oriented normal to the plane) was obtained in Cu-rich composition ($y=5$) by rapid thermal annealing. A textured platelet microstructure (c-axis oriented normal to the plane) was obtained using slow heating rates and by carefully controlling the gas environment at the temperature region near the orthorhombic-tetragonal transition. Thus, we have demonstrated the ability to control the morphology and orientation of the film by varying the annealing process.

Accession For	
NTIS CRA&I	☒
DTIC TAB	☐
Unannounced	☐
Justification	
By	
Distribution /	
Availability Codes	
Dist	Avail and/or Special
A-1	



UNCLASSIFIED

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

EFFECT OF ANNEALING CONDITIONS ON THE STRUCTURAL AND SUPERCONDUCTING PROPERTIES OF Y-Ba-Cu-O FILMS

J. Ryu^a, Y. Huang^a, C. Vittoria^a, D. F. Ryder, Jr.^b, J. Marzik^c, R. Benfer^c, and W. Spurgeon^c

a: Northeastern University, Center for Electromagn. Research, Boston, MA 02115

b: Tufts University, Laboratory for Materials and Interfaces, Medford, MA 02155

c: Army Materials Technology Laboratory, Watertown, MA 02912

ABSTRACT

The effect of various annealing treatments on ion beam sputter-deposited $Y_1Ba_2Cu_yO_{7-x}$ ($y=3$ and 5) amorphous thin films has been studied using scanning electron microscopy (SEM), x-ray diffraction and dc resistivity measurements. SEM analysis showed that the morphology of the films were clearly divided into three categories: an equiaxial grain structure, a well oriented grid-like microstructure and a textured platelet structure. The equiaxial grain structure (random orientation according to x-ray diffraction measurements) was obtained when the film was annealed at slow heating and cooling rates in an oxygen atmosphere. The grid-like microstructure (b/c-axes oriented normal to the plane) was obtained in Cu-rich composition ($y=5$) by rapid thermal annealing. A textured platelet microstructure (c-axis oriented normal to the plane) was obtained using slow heating rates and by carefully controlling the gas environment at the temperature region near the orthorhombic-tetragonal transition. Thus, we have demonstrated the ability to control the morphology and orientation of the film by varying the annealing process.

INTRODUCTION

The discovery of superconductivity above liquid nitrogen temperature in oxide ceramics has ignited intense research on these materials throughout the world. For planar device applications, such as microwave and millimeter wave components, high power electromagnetics, and high speed electronic devices, it is essential to produce thin superconducting films. Furthermore, it is now well known that the superconducting properties of these materials are highly anisotropic.¹ Therefore, it is important to be able to control the orientation and morphology of these films. Although preferentially oriented polycrystalline films^{2,3} and single crystalline films^{4,5} have been reported by several groups, it is not yet clear what the determining factors are in controlling film orientation. In the present study, we report the annealing processes by which we have been able to produce thin films having several unique morphologies and orientations.

Futhermore, we report here the effect of microstructure on the superconducting, properties, wet chemical etching rate, and the stability of this film to the laboratory environment.

EXPERIMENTAL PROCEDURE

Films of YBaCuO were deposited by using an ion beam sputtering technique with sintered single oxide targets. The nominal compositions of the targets were Y=1, Ba=2.3 and Cu=3.8 or 6. The compositions of films measured by electron probe microanalysis (EPMA) were $Y_1Ba_2Cu_3O_{7-x}$ and $Y_1Ba_2Cu_5O_{7-x}$ using $YBa_{2.3}Cu_{3.8}O_{7-x}$ and $Y_1Ba_{2.3}Cu_6O_{7-x}$ targets, respectively. Usually, a 1mm thick film was deposited on single crystalline (100) oriented cubic zirconia (YSZ), $SrTiO_3$ or MgO substrates in 1hr. All films were deposited without externally heating the substrate. As-deposited films had an amorphous structure. The films were then annealed using various heating and cooling cycles, as shown in Fig. 1. In process A, the films were heated under flowing oxygen (20sccm) in a tube furnace to 1093K-1143K in 3hrs and held at this temperature for 1-4hrs and subsequently cooled to room temperature in 10-18hrs.

Process B involved three steps. In step one, the films were rapidly heated using a quartz lamp to temperatures ranging from 923K to 973K for 20 min in a 200-500 millitorr oxygen atmosphere and then cooled to room temperature. In step two, the films were heated from ambient temperature to 1173K-1223K in 30sec in a helium atmosphere, held at this temperature for 5-10min. In step 3, the films were cooled in an oxygen atmosphere to 473K over 4 hr. The temperature cycle of process C was identical to that of process A. However, in process C, the heating cycle was performed under helium gas in order to control the orthorhombic-tetragonal transition temperature. The atmosphere was then switched to oxygen for the high temperature hold and subsequent cool down. Surface profilometry, SEM, and thin film x-ray diffraction were used to determine surface roughness, morphology, and orientation of the film, respectively. The dc-resistance of the film as a function of temperature was measured by using a standard four probe technique with pressed indium contacts. The current was 0.1mA in this measurement. For the stability study, the resistance of the film was remeasured after exposure to air for one month. In the chemical etching rate study, the films were etched using a 1:60 mixture of phosphoric acid and DI water.

RESULTS AND DISCUSSIONS

The SEM micrographs and x-ray diffraction patterns of the films are shown in Fig. 2 and Fig. 3, respectively. A random oriented equiaxial grain structure with grain size around 1mm was observed in films annealed using process A (slow heating and cooling in O_2). Films annealed in this process always exhibited a smooth, shiny surface (surface roughness less than 40nm). The texture was independent of the substrate material and slight variations in the Cu concentration. The changes in the film structure during the annealing cycle are complex with several factors to be considered including the amorphous-crystalline transition, the orthorhombic-tetragonal transition, and crystallite nucleation and growth. For process A in which the heating and cooling rates are relatively slow, it is likely that the nucleation rate is much higher than the crystallite growth rate, resulting in a fine grain, randomly oriented structure.

A well oriented grid-like structure with grain size around 1-2mm was observed from the film annealed using process B. The initial rapid heating (process B, step 1) was found to be critical to the formation of this microstructure. In addition, a successful deposition was only achieved using a $SrTiO_3$ substrate.

Films deposited on YSZ and MgO tended to peel away from the substrate during the annealing cycle. Generally, process B was more successful for the films with higher Cu concentration. The surface roughness was determined to be less than 180nm. X-ray diffraction results (Fig. 3b) indicated that the films showed preferred orientation with strong indications of c-axis oriented normal to the plane and weaker orientation normal to the b-axis. Further work is required to determine whether there are two types of grains oriented normal to the (001) and (010) directions, respectively or there exists a single preferred orientation. In the Cu rich film, the existence of a secondary phase, such as CuO and BaCuO₂ was observed (as marked (*) in Fig. 3b). Further, in some films annealed by this process, we noticed a presence of an amorphous phase, even after heating at 1193K for 5min. While the average composition of the film was YBa₂Cu₅O_{7-x}, as determined by EPMA, the composition of grid area only was YBa₂Cu₃O_{7-x}. This result indicated that Cu rich films with the grid-like structure consisted of two phases - stoichiometric composition grid (see G in Fig. 2b), and Cu-rich matrix (see M in Fig. 2c). It has been reported elsewhere⁵ that a c-axis oriented epitaxial film of YBaCuO has been produced on SrTiO₃ by heating the substrate during deposition at a temperature of 923K. Based on our results and others^{4,5}, we believe that the well oriented tetragonal crystallites are nucleated at the film-substrate interface during the first step rapid thermal heating at 923K-973K for 20min. These fine crystallites at the interface rapidly grow during the second rapid heating at 1173K-1223K. As a result, stoichiometric composition (as determined by EPMA) grid structure and Cu rich composition islands are formed. In addition, taking into consideration that an excessive amount of Cu enhances the formation of the grid-like structure, the growth rate of the grid may be controlled by the diffusion rate of Cu ion during rapid thermal annealing.

A well oriented platelet structure was obtained in films on the three substrates investigated using process C. At temperatures around 973K-1073K, the annealing gas was switched abruptly from helium to oxygen. The helium gas was introduced during heating in order to control the orthorhombic-tetragonal transition temperature. The surface roughness of the film was less than 300nm, and the platelet-like grains were approximately 20nm in size (Fig. 2c). The film showed strong preferred orientation with the plane of the film normal to the c-axis as shown by x-ray diffraction (Fig. 3C). It was found that the helium atmosphere during the heating step was critical to the formation of the platelet microstructure. Study on the film-substrate interface using cross section transmission electron microscopy is underway in order to understand the mechanism of nucleation of the crystallite and its effect on the film orientation.

Temperature dependence of the resistance for the films prepared by aforementioned annealing processes are shown in Fig. 4. The resistance (y-axis) is normalized to the room temperature resistance of each film. A T_c onset for film B (grid-like microstructure) and film C (platelet) is the same at 94K, but the zero resistance is observed at 74K and 84K, respectively. A T_c onset and T_c zero for film A (equiaxial grains) is observed at 80K and at 60K, respectively. It is noted that the slope in the resistance curve for film A changes at 180 K, and a slight drop in resistance is observed at 105 K. The broad transition and changes in slope (dr/dK) in the normal state may imply that cation (Cu) ordering is not uniform in the randomly oriented film.⁶

In films with oriented microstructures, there was no degradation in T_c onset, but room temperature resistance increased 5-10% after exposure to air for one month. However, the chemical etching rate (in dilute phosphoric acid) of the oriented

film is 2-3 times higher than that of random orientation film. Chemical etching rate, microstructural, and superconducting properties of the films are summarized in Table 1.

CONCLUSION

High quality thin films of the YBaCuO superconductor have been produced on YSZ, SrTiO₃ and MgO substrates via ion beam sputter-deposition and annealing. The morphology and orientation of the film have been successfully controlled by employing unique annealing processes. Preliminary results indicate that the orientation and morphology of the films are determined primarily by the nucleation rate of crystallites at the film-substrate interface during the initial stage of heating, and the growth rate (or temperature) of these crystallites during high temperature heating. In any case, the lattice matching between the film and the substrate is a key factor in this solid state phase transition. Additional work is required to define the mechanism of nucleation and crystallite growth, and its effect on film orientation. The observed T_c zero resistance is 60K, 74K, and 84K for the film with equiaxial grains, b/c-axis oriented grid-like structure and c-axis oriented platelet microstructure, respectively.

Table 1. Chemical etching rate, microstructural and superconducting properties of the films with various orientations.

films	structure	room temp. resistivity (mW-cm)	on set (K)	T _c zero (K)	etch rate ^a (mm/min)
A ^b	equiaxial grains, random oriented	0.63	80	60	0.154
B ^c	grid-like, b/c-axes oriented	1.07	94	74	0.909
C ^b	platelet, c-axis oriented	0.21	94	84	0.309

a: Film was etched in a 1:60 mixture of phosphoric acid and DI water at room temperature.

b: On YSZ substrate.

c: On SrTiO₃ substrate

1. S. W. Tozer, A. W. Kleinsasser, T. Penney, D. Kaiser, and F. Holtzberg, Measurement of Anisotropic resistivity and Hall Constant for Single-Crystal $\text{YBa}_2\text{Cu}_3\text{O}_{7-x}$, Phys. Rev. Lett. 59:1768 (1987).
2. J. Kwo, T. C. Hsieh, R. M. Fleming, M. Hong, S. H. Liou, B. A. Davidson, and L. C. Feldman, Structure and Superconducting Properties of Orientation-Ordered $\text{YBa}_2\text{Cu}_3\text{O}_{7-x}$ Films Prepared by Molecular-beam Epitaxy, Phys. Rev. B 36:4039 (1987).
3. K. Setsune, T. Kamada, H. Adachi, and K. Wasa, Epitaxial Y-Ba-Cu-O thin films prepared by rf-magnetron sputtering, J. Appl. Phys. 64:1318 (1988).
4. T. Terashima, K. Iijima, K. Yamamoto, Y. Bando, and H. Mazaki, Single-Crystal $\text{YBa}_2\text{Cu}_3\text{O}_{7-x}$ Thin Films by Activated Reactive Evaporation, Jpn. J. Appl. Phys. 27:L91 (1988).
5. J. Fujita, T. Yoshitake, A. Kamijo, T. Satoh, and H. Igarashi, Preferentially oriented epitaxial Y-Ba-Cu-O films prepared by the ion beam sputtering method, J. Appl. Phys. 64:1292 (1988).
6. G. F. Dionne, Transition-Metal Oxide Superconductivity, Technical Report 802, ESD-TR-277, Lincoln Laboratory, Lexington, Ma (1987)

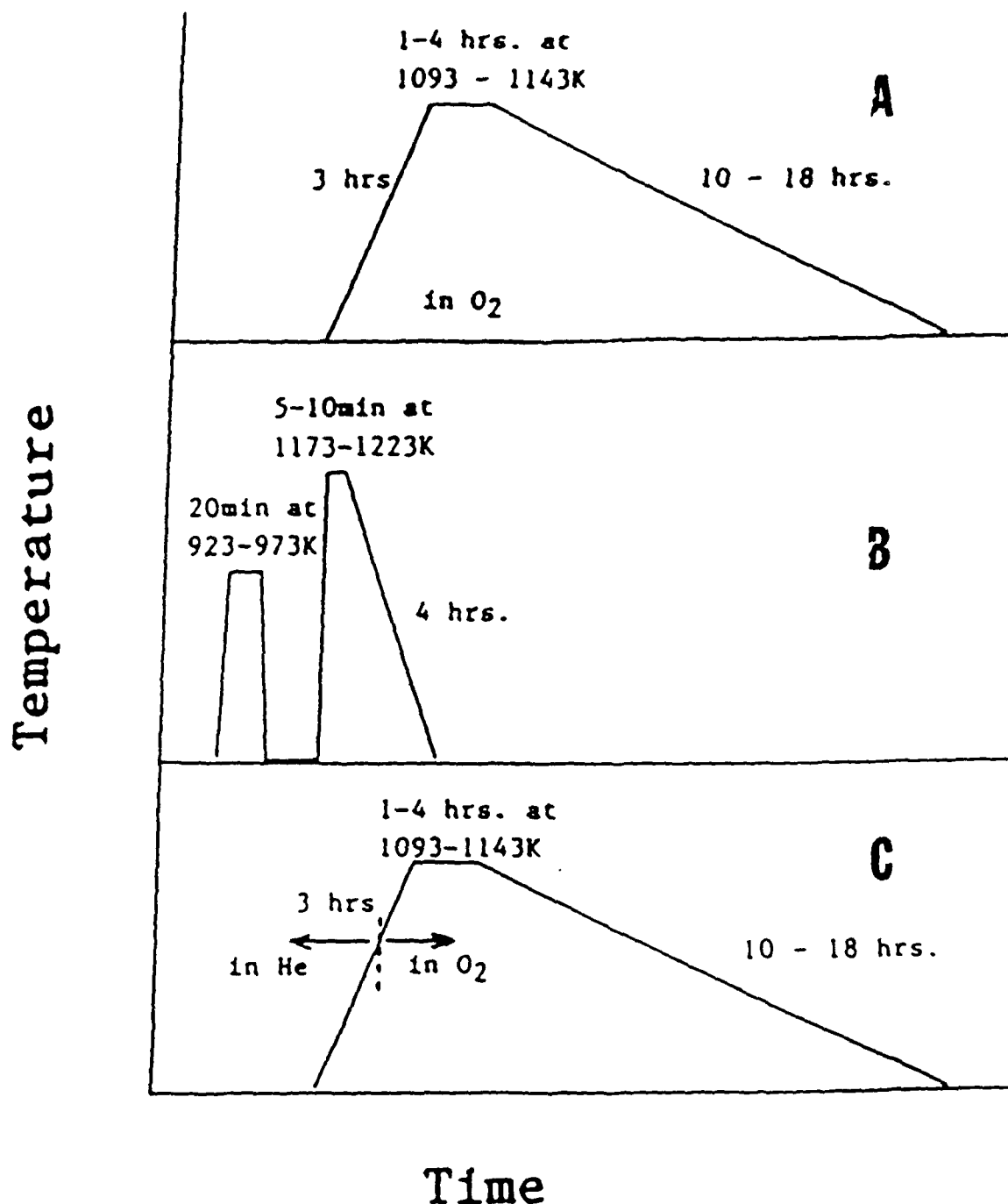
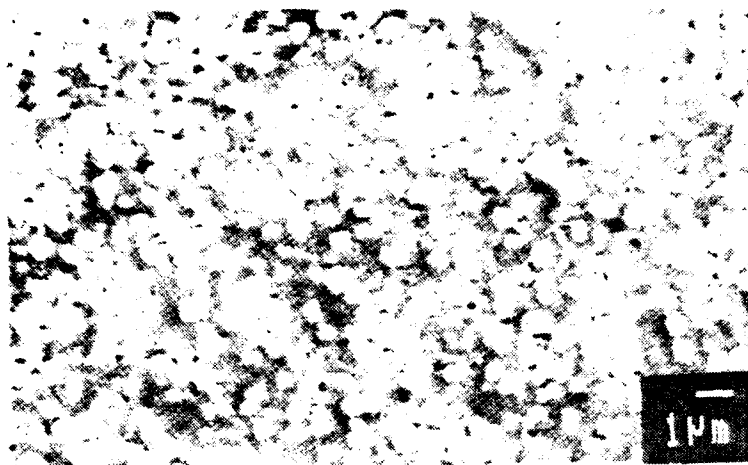
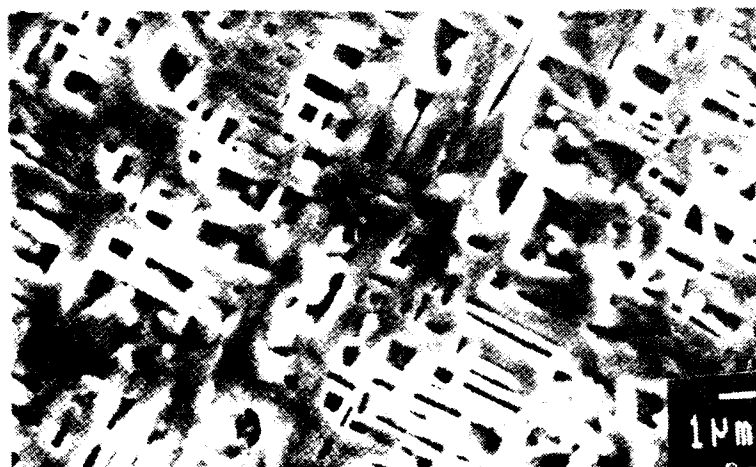


Fig. 1. Annealing procedures used in this research to produce films having a unique morphology and orientation. Note that the temperature-time cycle of process C is identical to that of process A. However, the heating cycle was performed under He gas in process C. Process B involved three steps: in-situ rapid thermal annealing in a low pressure O₂; rapid heating in a He atmosphere; and followed by furnace cooling in O₂.



(a)



(b)



(c)

Fig. 2. SEM micrographs showing typical microstructures of films with equiaxial grains (a); grid-like structure (b); and platelet (c).

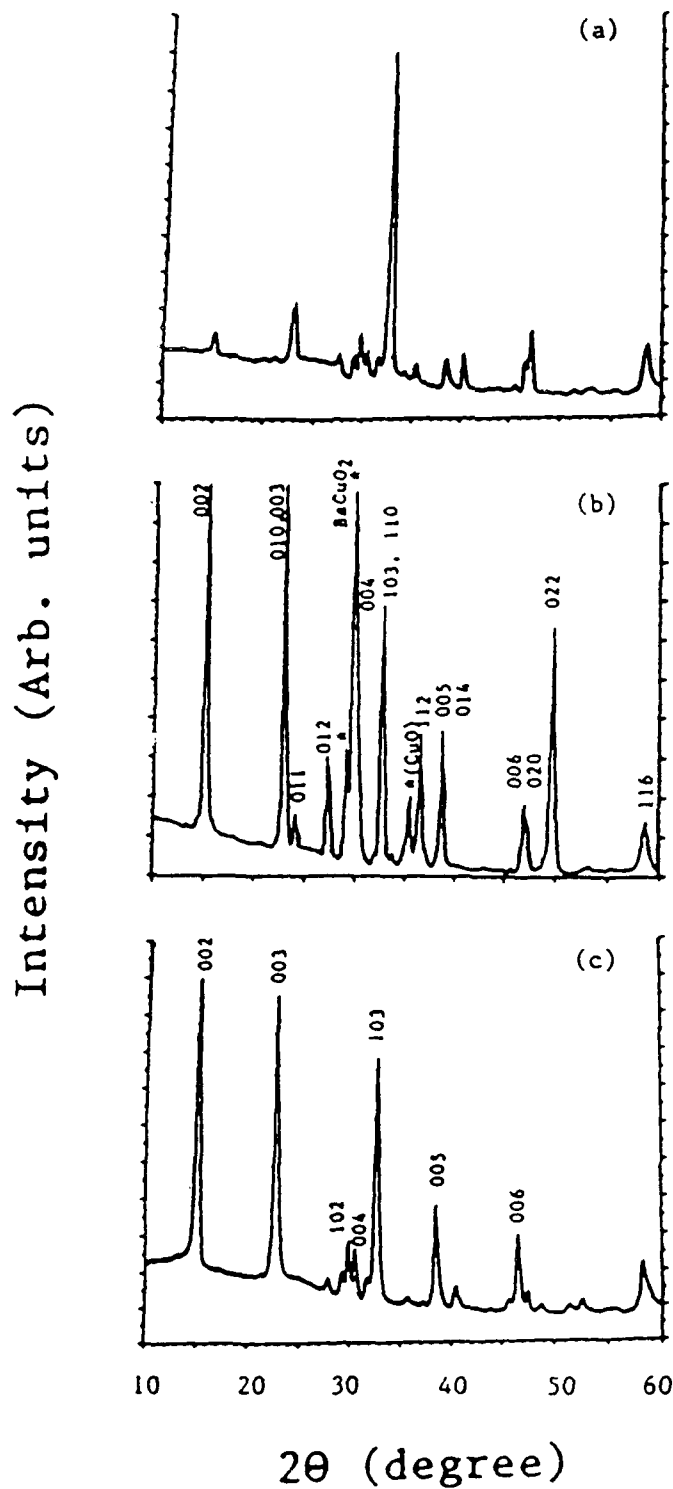


Fig. 3. Thin film x-ray diffraction pattern of the films with various microstructures: (a), equiaxial grain structure - random orientation; (b) grid-like structure - b/c-axes normal to the plane. The peaks from secondary phase are marked (*); (c) platelet microstructure - c-axis oriented normal to the plane.

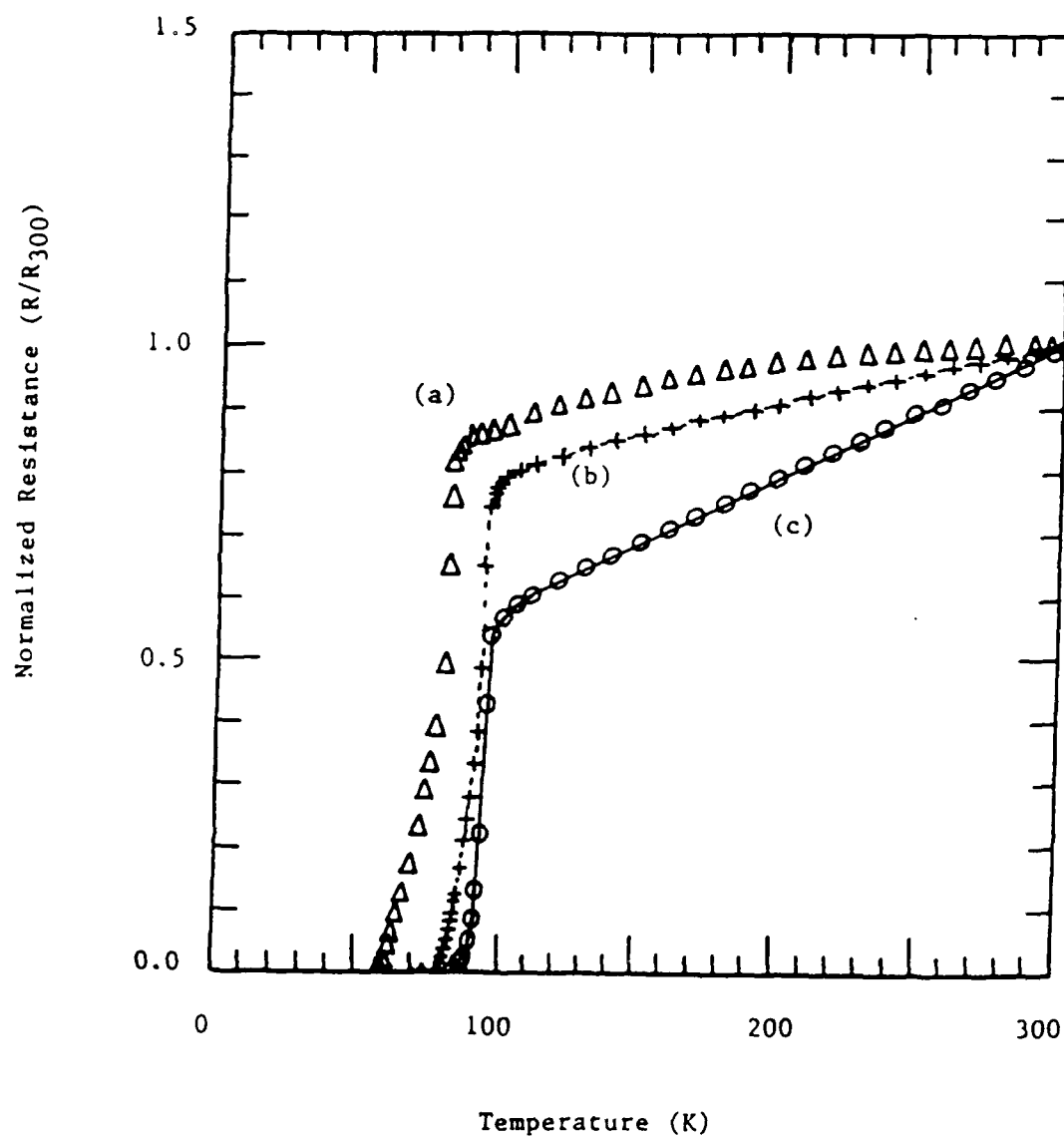


Fig. 4. Temperature dependence of the resistance for the film with various morphology and orientations. The resistance is normalized to room temperature resistance (r/r_{300}) of each film: (a) randomly oriented film; (b) b/c-axes normal to the plane; (c) c-axis normal to the plane.

DISTRIBUTION LIST

No. of Copies	To	No. of Copies	To
1	Office of the Under Secretary of Defense for Research and Engineering, The Pentagon, Washington, DC 20301	1	Commander, U.S. Army Missile Command, Redstone Scientific Information Center, Redstone Arsenal, AL 35898-5241
1	ATTN: Mr. J. Persh	1	ATTN: AMSMI-RD-CS-R/Doc
1	Dr. L. Young	1	AMSMI-R, Dr. W. C. McCorkle
1	Mr. K. R. Foster		
	Commander, U.S. Army Laboratory Command, 2800 Powder Mill Road, Adelphi, MD 20783-1145		Commander, U.S. Army Aviation Systems Command, P.O. Box 209, St. Louis, MO 63120-1798
2	ATTN: AMSLC-IM-TL	1	ATTN: AMSAV-NS, Mr. M. L. Bauccio
1	AMSLC-TD	1	Technical Library
1	AMSLC-TD-A		
1	AMSLC-PA		Commander, U.S. Army Natick Research, Development, and Engineering Center, Natick, MA 01760
1	AMSLC-TP	1	ATTN: Technical Library
	Commander, Defense Technical Information Center, Cameron Station, Building 5, 5010 Duke Street, Alexandria, VA 22304-6145	1	Dr. J. A. Sousa
2	ATTN: DTIC-FDAC	1	Dr. R. J. Byrne
		1	Dr. R. Lewis
1	National Technical Information Service, 5285 Port Royal Road, Springfield, VA 22161		Commander, U.S. Army Satellite Communications Agency, Fort Monmouth, NJ 07703
	Director, Defense Advanced Research Projects Agency, 1400 Wilson Boulevard, Arlington, VA 22209	1	ATTN: Technical Document Center
1	ATTN: Dr. P. Parrish		Commander, U.S. Army Science and Technology Center Far East Office, APO San Francisco, CA 96328
1	Dr. B. Wilcox	1	ATTN: Terry L. McAfee
1	Dr. K. Hardmann-Rhyne		Commander, U.S. Army Communications and Electronics Command, Fort Monmouth, NJ 07703
	Battelle Columbus Laboratories, Metals and Ceramics Information Center, 505 King Avenue, Columbus, OH 43201	1	ATTN: AMSEL-TDD, Mr. T. A. Pfeiffer, Technical Dir.
1	ATTN: Mr. W. Duckworth		Director, Electronic Technology and Devices Lab, Fort Monmouth, NJ 07703
1	Dr. O. Niesz	1	ATTN: DELET-D, Dr. C. G. Thornton
	Department of the Army, Office of the Assistant Secretary of the Army (RDA), Washington, DC 20310		Commander, U.S. Army Tank-Automotive Command, Warren, MI 48397-5000
1	ATTN: Dr. J. G. Prather, Dep for Sci & Tech	1	ATTN: Dr. W. Bryzik
1	Dr. J. R. Sculley, SARD	1	D. Rose
	Deputy Chief of Staff, Research, Development, and Acquisition, Headquarters, Department of the Army, Washington, DC 20310	1	AMSTA-RKA
1	ATTN: DAMA-ZE, Mr. C. M. Church	1	AMSTA-UL, Technical Library
	Commander, U.S. Army Research and Development Office, Chief Research and Development, Washington, DC 20315	1	AMSTA-R
1	ATTN: Physical and Engineering Sciences Division	1	AMSTA-NS, Dr. H. H. Dobbs
	Commander, Army Research Office, P.O. Box 12211, Research Triangle Park, NC 27709-2211		Commander, U.S. Army Armament, Munitions and Chemical Command, Dover, NJ 07801
1	ATTN: Information Processing Office	1	ATTN: Mr. J. Lannon
1	Dr. J. Hurt	1	Mr. H. E. Peibly, Jr., PLASTEC, Director
1	Dr. A. Crowson	1	Technical Library
1	Dr. R. Reeber	1	Dr. T. Davidson
1	Dr. R. Shaw	1	Dr. B. Ebihara
	Commander, U.S. Army Materiel Command, 5001 Eisenhower Avenue, Alexandria, VA 22333		Commander, U.S. Army Armament, Munitions and Chemical Command, Rock Island, IL 61299
1	ATTN: AMCQA-EQ, Mr. H. L. Light	1	ATTN: Technical Library
1	AMCQA, Mr. S. J. Lorber		Commander, Aberdeen Proving Ground, MD 21005
	Commander, U.S. Army Electronics Research and Development Command, Fort Monmouth, NJ 07703-5000		ATTN: AMDAR-CLB-PS, Mr. J. Vervier
1	ATTN: AMDET-ES, Dr. A. Tauber		U.S. Army Corps of Engineers, Construction Engineering Research Lab, P.O. Box 4005, Champaign, IL 61820
	Director, Electronics Warfare Laboratory, Fort Monmouth, NJ 07703	1	ATTN: Dr. Robert Quattrone
1	ATTN: AMDEW-D, Mr. M. Adler		Commander, U.S. Army Belvoir R&DE Center, Fort Belvoir, VA 22060-5606
	Commander, U.S. Army Materiel Systems Analysis Activity, Aberdeen Proving Ground, MD 21005	1	ATTN: STRBE-FS, Mr. W. McGovern, Fuel & Wtr Sup Div
1	ATTN: AMXSY-MP, H. Cohen	1	AMOME-V, Mr. E. York
	Commander, U.S. Army Night Vision Electro-Optics Laboratory, Fort Belvoir, VA 22060	1	STRBE-ZTS, Dr. K. H. Steinbach, Office of the Chief Scientist
1	ATTN: DELNV-S, Mr. P. Travesky	1	AMOME-ZT, Mr. T. W. Lovelace, Tech Dir
1	DELNV-L-D, Dr. R. Buser	1	Mr. M. Lepera
1	DELNV-D, Dr. L. Cameron		Director, U.S. Army Ballistic Research Laboratory, Aberdeen Proving Ground, MD 21005
	Commander, Harry Diamond Laboratories, 2800 Powder Mill Road, Adelphi, MD 20783	1	ATTN: SLCBR-BLT, Dr. A. M. Dietrich
1	ATTN: Technical Information Office	1	SLCBR-BLF, Dr. A. Niller
1	SLCHD-RAE		Commander, Rock Island Arsenal, Rock Island, IL 61299
	Director, U.S. Army Research & Technology Labs, Ames Research Center, Moffet Field, CA 94035	1	ATTN: SARRI-EN
1	ATTN: DAVOL-D, Dr. R. Carlson		Director, U.S. Army Industrial Base Engineering Activity, Rock Island, IL 61299
1	DAVOL-AL-D, Dr. I. C. Statler, MS215-1, Aeromechanics Laboratory	1	ATTN: AMXIB-MT, Mr. G. B. Ney
			Chemical Research and Development Center, Aberdeen Proving Ground, MD 21010
		1	ATTN: AMSMC-CLD(A), Dr. B. Richardson

No. of Copies	To	No. of Copies	To
1	Commander, U.S. Army Test and Evaluation Command, Aberdeen Proving Ground, MD 21005	1	National Aeronautics and Space Administration, Langley Research Center, Hampton, VA 23665
1	ATTN: AMSTE-ME	1	ATTN: Mr. J. Buckley, MS 387
1	AMSTE-TD, Mr. H. J. Peters	1	Dr. J. Heyman, MS 231
		1	Mr. R. L. Long, MS 266
1	Commander, U.S. Army Foreign Science and Technology Center, 220 7th Street, N.E., Charlottesville, VA 22901	1	Commander, White Sands Missile Range, Electronic Warfare Laboratory, OMEW, ERADCOM, White Sands, NM 88002
1	ATTN: Military Tech	1	ATTN: Mr. Thomas Reader, AMSEL-WLM-ME
1	Mr. J. Crider		
1	Ms. P. Durrer		
1	Mr. P. Greenbawm		
1	Chief, Benet Weapons Laboratory, Watervliet, NY 12189		
1	ATTN: AMDAR-LCB-TL		
1	Dr. G. D'Andrea	1	Department of Energy, Division of Transportation, 20 Massachusetts Avenue, N.W., Washington, DC 20545
1	AMDAR-LCB, Dr. F. Sautter	1	ATTN: Dr. R. J. Gottschall, ER-131, GTN
1	Director, Eustis Directorate, U.S. Army Mobility Research and Development Laboratory, Fort Eustis, VA 23604-5577	1	Mechanical Properties Data Center, Belfour Stulen Inc., 13917 W. Bay Shore Drive, Traverse City, MI 49684
1	ATTN: SAVOL-E-MOS (AMCCOM)		
1	Commander, U.S. Army Engineer Waterways Experiment Station, Vicksburg, MS 39180	1	National Institute of Standards and Technology, Washington, DC 20234
1	ATTN: Research Center Library	1	ATTN: E. S. Etz, Bldg. 222, Rm A-121
		1	D. L. Hunston, Bldg. 224, Rm A-209
		1	Dr. D. H. Reneker, Dep. Dir., Ctr for Mat'l Sci.
		1	Dr. Lyle Schwartz
		1	Dr. Stephen Hsu
		1	Dr. Allan Draggoo
1	Project Manager, Munitions Production Base, Modernization and Expansion, Dover, NJ 07801		
1	ATTN: AMCPM-PBM-P	1	U.S. Bureau of Mines, Mineral Resources Technology, 2401 E. Street, N.W., Washington, DC 20241
		1	ATTN: Mr. M. A. Schwartz
1	Technical Director, Human Engineering Laboratories, Aberdeen Proving Ground, MD 21005-5001		
1	ATTN: SLCHE-D, Dr. J. D. Weisz	1	National Institute of Standards and Technology, Gaithersburg, MD 20899
1	Chief of Naval Research Arlington, VA 22217	1	ATTN: Dr. S. Wiederhorn
1	ATTN: Code 471	1	Dr. N. Tighe
1	Dr. A. Diness		
1	Dr. R. Pohanka	1	National Research Council, National Materials Advisory Board, 2101 Constitution Avenue, Washington, DC 20418
1	Naval Research Laboratory, Washington, DC 20375	1	ATTN: Dr. K. Zwilsky
1	ATTN: Code 5830	1	D. Groves
		1	J. Lane
1	Headquarters, Naval Air Systems Command, Washington, DC 20360		
1	ATTN: Code 5203	1	National Science Foundation, Materials Division, 1800 G Street, N.W., Washington, DC 20006
		1	ATTN: Dr. L. Toth
		1	Dr. J. Hurt
1	Headquarters, Naval Sea Systems Command, 1941 Jefferson Davis Highway, Arlington, VA 22376		
1	ATTN: Code 035	1	AVCO Corporation, Applied Technology Division, Lowell Industrial Park, Lowell, MA 01887
		1	ATTN: Dr. T. Vasilos
1	Headquarters, Naval Electronics Systems Command, Washington, DC 20360		
1	ATTN: Code 504	1	Case Western Reserve University, Department of Metallurgy, Cleveland, OH 60605
		1	ATTN: Prof. A. H. Heuer
1	Commander, Naval Ordnance Station, Louisville, KY 40214		
1	ATTN: Code 85	1	Defence Research Establishment Pacific, FMO, Victoria, B.C., VOS 1B0, Canada
		1	ATTN: R. D. Barer
1	Commander, Naval Material Industrial Resources Office, Building 537-2, Philadelphia Naval Base, Philadelphia, PA 19112		
1	ATTN: Technical Director	1	Ford Motor Company, Turbine Research Department, 20000 Rotunda Drive, Dearborn, MI 48121
		1	ATTN: Mr. A. F. McLean
1	Commander, Naval Weapons Center, China Lake CA 93555	1	Mr. J. A. Mangels
1	ATTN: Mr. F. Markarian		
1	Commander, U.S. Air Force Wright Aeronautical Labs, Wright- Patterson Air Force Base, OH 45433	1	Ford Motor Company, P.O. Box 2053, Dearborn, MI 48121
1	ATTN: Dr. N. Tallan	1	ATTN: Dr. D. Compton, Vice President Research
1	Dr. H. Graham		
1	Dr. R. Ruh	1	General Electric Company, Research and Development Center, Box 8, Schenectady, NY 12345
1	Aero Propulsion Labs, Mr. R. Marsh	1	ATTN: Dr. R. J. Charles
1	Dr. H. M. Burte	1	Dr. C. D. Greskovich
1	AFWAL/MLLP, Mr. D. Forney	1	Dr. S. Prochazka
1	AFML/MLLM, Mr. H. L. Gege		
1	AFSC/MLLM, Dr. A. Katz	1	Georgia Institute of Technology, EES, Atlanta, GA 30332
		1	ATTN: Mr. J. D. Walton
1	Commander, Air Force Armament Center, Eglin Air Force Base, FL 32542		
1	ATTN: Technical Library	1	GTE Sylvania, Waltham Research Center, 40 Sylvania Road, Waltham, MA 02154
		1	ATTN: Dr. W. H. Rhodes
1	National Aeronautics and Space Administration, Lewis Research Center, 21000 Brookpark Road, Cleveland, OH 44135		
1	ATTN: J. Accurio, USAMRDL	1	Martin Marietta Laboratories, 1450 South Rolling Road, Baltimore, MD 21227
1	Dr. H. B. Probst, MS 49-1	1	ATTN: Dr. J. Venables
1	Dr. S. Dutta		
1	NASA - Scientific and Technical Information Facility, P.O. Box 8757, Baltimore/Washington International Airport, Maryland 21240	1	Massachusetts Institute of Technology, Department of Metallurgy and Materials Science, Cambridge, MA 02139
		1	ATTN: Prof. R. L. Coble
		1	Prof. H. K. Bowen
		1	Prof. W. D. Kingery
		1	Prof. J. Vander Sande

No. of Copies	To
1	Materials Research Laboratories, P.O. Box 50, Ascot Vale, VIC 3032, Australia ATTN: Dr. C. W. Weaver
1	Midwest Research Institute, 425 Volker Boulevard, Kansas City, MO 64110 ATTN: Mr. G. W. Gross, Head, Physics Station
1	Pennsylvania State University, Materials Research Laboratory, Materials Science Department, University Park, PA 16802 ATTN: Prof. R. Roy
1	Prof. R. E. Newnham
1	Prof. R. E. Tressler
1	Dr. C. Pantano
1	Mr. C. O. Ruud
1	State University of New York at Albany, Department of Physics, Albany, NY 12222 ATTN: Prof. W. A. Lanford
1	State University of New York at Stony Brook, Department of Materials Science, Long Island, NY 11790 ATTN: Prof. F. F. Y. Wang
1	Stanford Research International, 333 Ravenswood Avenue, Menlo Park, CA 94025d ATTN: Dr. P. Jorgensen
1	Dr. D. Rowcliffe
1	United Technologies Research Center, East Hartford, CT 06108 ATTN: Dr. J. Brennan
1	Dr. K. Prewé
1	University of California, Lawrence Livermore Laboratory, P.O. Box 808, Livermore, CA 94550 ATTN: Mr. R. Landingham
1	Dr. C. F. Cline
1	Dr. J. Birch Holt
1	University of Florida, Department of Materials Science and Engineering, Gainesville, FL 32611 ATTN: Dr. L. Hench
1	University of Washington, Ceramic Engineering Division, FB-10, Seattle, WA 98195 ATTN: Prof. R. Bradt
1	Westinghouse Electric Corporation, Research Laboratories, Pittsburgh, PA 15235 ATTN: Dr. R. J. Bratton
1	Battelle Pacific Northwest Lab, NDT Section, Richland, WA 99353 ATTN: Mr. A. Birks, Associate Manager
1	Rensselaer Polytechnic Institute, Department of Materials Engineering, Troy, NY 12181 ATTN: R. J. Diefendorf
1	Oak Ridge National Laboratory, P.O. Box X Oak Ridge, TN 37830 ATTN: P. F. Becher
1	V. J. Tennery
1	R. Johnson
1	Sandia Laboratories, Albuquerque, NM 87185 ATTN: Dr. F. Gerstle, Div 5814
1	The John Hopkins University, Department of Civil Engineering/ Materials Science and Engineering, Baltimore, MD 28218 ATTN: Dr. R. E. Green, Jr.
1	Director, Office of Science and Technology Policy, Old Executive Office Building, Washington, DC 20223

No. of Copies	To
1	Subcommittee on Science, 2319 Rayburn House Office Building, Washington, DC 20515 ATTN: Mr. P. C. Maxwell
1	Aerospace Corporation, Materials Science Laboratory, 2350 East El Segundo Boulevard, El Segundo, CA 90245 ATTN: Dr. L. R. McCreight
1	IBM Corporation, Thomas B. Watson Research Center, Yorktown Heights, NY 10598 ATTN: Dr. G. Onoda
1	Corning Glass Works, Research and Development Division, Corning, NY 14830 ATTN: Dr. W. R. Prindle
1	3M Company, New Products Department, 219-01-01, 3M Center, St. Paul, MN 55144 ATTN: R. E. Richards
1	Technology Strategies, Inc., 10722 Shingle Oak Ct., Burke, VA 22015 ATTN: Dr. E. C. Van Reuth
1	Rutgers University, Center for Ceramics, Rm A274, P.O. Box 909, Piscataway, NJ 08854 ATTN: Prof. J. B. Wachtman, Jr., Director
1	Syracuse University, 304 Administration Building, Syracuse, NY 13210 ATTN: Dr. V. Weiss
1	Lehigh University, Materials Research Center #32, Bethlehem, PA 18015 ATTN: Dr. D. M. Smyth
1	Alfred University, New York State College of Ceramics, Alfred, NY 14802 ATTN: Dr. R. L. Snyder
1	Alfred University, Center for Advanced Ceramic Technology, Alfred, NY 14802 ATTN: R. M. Spriggs
1	University of California, Center for Advanced Materials, 058, Hildebrand Hall, Berkeley, CA 94720 ATTN: Prof. G. Somorjai
1	Boeing Aerospace Company, 11029 Southeast 291, Auburn, MA 98002 ATTN: W. E. Strobelt
1	University of California, Materials Science and Mineral Engineering, Heart Mining Building, Rm 284, Berkeley, CA 94720 ATTN: Prof. G. Thomas
2	Director, U.S. Army Materials Technology Laboratory, Watertown, MA 02172-0001 ATTN: SLCMT-TML
7	Authors

U.S. Army Materials Technology Laboratory, Watertown, Massachusetts 02172-0001 EFFECT OF ANNEALING CONDITIONS ON THE STRUCTURAL AND SUPERCONDUCTING PROPERTIES OF Y-Ba-Cu-O FILMS - Jae Ryu, Yizhou Huang, Carmine Vittoria, Daniel F. Ryder, Jr., James V. Marzik, Richard Benfer, and William A. Spurgeon Technical Report MTL TR 89-41, May 1989, 11 pp - illus-tables, D/A Project IL161102AH42 The effect of various annealing treatments on ion beam sputter-deposited YBa₂Cu₃O_{7-x} (y=3 and 5) amorphous thin films has been studied using scanning electron microscopy (SEM), X-ray diffraction and dc resistivity measurements. SEM analysis showed that the morphology of the films were clearly divided into three categories: an equiaxial grain structure, a well-oriented grid-like microstructure and a textured platelet structure. The equiaxial grain structure (random orientation according to X-ray diffraction measurements) was obtained when the film was annealed at slow heating and cooling rates in an oxygen atmosphere. The grid-like microstructure (b/c-axes oriented normal to the plane) was obtained in Cu-rich composition (y=5) by rapid thermal annealing. A textured platelet microstructure (c-axis oriented normal to the plane) was obtained using slow heating rates and by carefully controlling the gas environment at the temperature region near the orthorhombic-tetragonal transition. Thus, we have demonstrated the ability to control the morphology and orientation of the film by varying the annealing process.	AD UNCLASSIFIED UNLIMITED DISTRIBUTION Key Words Superconductors Ceramic materials Thin films
U.S. Army Materials Technology Laboratory, Watertown, Massachusetts 02172-0001 EFFECT OF ANNEALING CONDITIONS ON THE STRUCTURAL AND SUPERCONDUCTING PROPERTIES OF Y-Ba-Cu-O FILMS - Jae Ryu, Yizhou Huang, Carmine Vittoria, Daniel F. Ryder, Jr., James V. Marzik, Richard Benfer, and William A. Spurgeon Technical Report MTL TR 89-41, May 1989, 11 pp - illus-tables, D/A Project IL161102AH42 The effect of various annealing treatments on ion beam sputter-deposited YBa₂Cu₃O_{7-x} (y=3 and 5) amorphous thin films has been studied using scanning electron microscopy (SEM), X-ray diffraction and dc resistivity measurements. SEM analysis showed that the morphology of the films were clearly divided into three categories: an equiaxial grain structure, a well-oriented grid-like microstructure and a textured platelet structure. The equiaxial grain structure (random orientation according to X-ray diffraction measurements) was obtained when the film was annealed at slow heating and cooling rates in an oxygen atmosphere. The grid-like microstructure (b/c-axes oriented normal to the plane) was obtained in Cu-rich composition (y=5) by rapid thermal annealing. A textured platelet microstructure (c-axis oriented normal to the plane) was obtained using slow heating rates and by carefully controlling the gas environment at the temperature region near the orthorhombic-tetragonal transition. Thus, we have demonstrated the ability to control the morphology and orientation of the film by varying the annealing process.	AD UNCLASSIFIED UNLIMITED DISTRIBUTION Key Words Superconductors Ceramic materials Thin films
U.S. Army Materials Technology Laboratory, Watertown, Massachusetts 02172-0001 EFFECT OF ANNEALING CONDITIONS ON THE STRUCTURAL AND SUPERCONDUCTING PROPERTIES OF Y-Ba-Cu-O FILMS - Jae Ryu, Yizhou Huang, Carmine Vittoria, Daniel F. Ryder, Jr., James V. Marzik, Richard Benfer, and William A. Spurgeon Technical Report MTL TR 89-41, May 1989, 11 pp - illus-tables, D/A Project IL161102AH42 The effect of various annealing treatments on ion beam sputter-deposited YBa₂Cu₃O_{7-x} (y=3 and 5) amorphous thin films has been studied using scanning electron microscopy (SEM), X-ray diffraction and dc resistivity measurements. SEM analysis showed that the morphology of the films were clearly divided into three categories: an equiaxial grain structure, a well-oriented grid-like microstructure and a textured platelet structure. The equiaxial grain structure (random orientation according to X-ray diffraction measurements) was obtained when the film was annealed at slow heating and cooling rates in an oxygen atmosphere. The grid-like microstructure (b/c-axes oriented normal to the plane) was obtained in Cu-rich composition (y=5) by rapid thermal annealing. A textured platelet microstructure (c-axis oriented normal to the plane) was obtained using slow heating rates and by carefully controlling the gas environment at the temperature region near the orthorhombic-tetragonal transition. Thus, we have demonstrated the ability to control the morphology and orientation of the film by varying the annealing process.	AD UNCLASSIFIED UNLIMITED DISTRIBUTION Key Words Superconductors Ceramic materials Thin films