



REPORT DOCUMENTATION PAGE

Form Approved
OAM No. 704 0198

Public reporting burden for this collection of information is estimated to average 1 hour per response, including the time for reviewing instructions, searching existing data sources, gathering and maintaining the data needed, reviewing and collecting the data, and reviewing and editing the information to be included in this report. Send comments regarding this burden estimate or any other aspect of this collection of information, including suggestions for reducing the burden, to Washington Headquarters Service, Directorate for Information Operations and Reports, 1215 Jefferson Davis Highway, Suite 1204, Arlington, VA 22202-4302, and to the Office of Management and Budget, Paperwork Project (0704-0198) Administration, Washington, DC 20503.

AD-A230 810

1 AGENCY USE ONLY (leave blank)

2 REPORT DATE
12/13/90

3 REPORT TYPE AND DATES COVERED

4 TITLE AND SUBTITLE

Integrated Synthesis and Post Processing of Silicon Carbide and Aluminum Nitride

5 FUNDING NUMBERS

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8. PERFORMING ORGANIZATION REPORT NUMBER

9. SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS(ES)

U.S. Army Research Office
P.O. Box 12211
Research Triangle Park, NC 27709-2211

10. SPONSORING/MONITORING AGENCY REPORT NUMBER

11. SUPPLEMENTARY NOTES

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12a. DISTRIBUTION/AVAILABILITY STATEMENT

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12b. DISTRIBUTION CODE

13. ABSTRACT (Maximum 200 words)

Synthesis of nonoxide powders in a nonthermal (microwave) plasma was demonstrated, using SiC, Si₃N₄ and Si as model materials. Particle sizes are ultrafine, 2-5 nm. Compositions and structures have been characterized. Equipment has been developed and characterized.

Creep of polycrystalline AlN has been investigated for the first time. Mechanisms are identified.

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JAN 16 1991

14. SUBJECT TERMS

nonoxide powder synthesis; nonthermal plasma; microwave; superplasticity; creep; SiC, Si, Si₃N₄, AlN

* Silicon Carbide, * Aluminum

15 NUMBER OF PAGES
4

16. PRICE CODE

17. SECURITY CLASSIFICATION OF REPORT
UNCLASSIFIED

18 SECURITY CLASSIFICATION OF THIS PAGE
UNCLASSIFIED

19. SECURITY CLASSIFICATION OF ABSTRACT
UNCLASSIFIED

20. LIMITATION OF ABSTRACT
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**Integrated Synthesis and Post-Processing of Silicon Carbide
and Aluminum Nitride**

**Interim Technical Report for Year 1
Covering the Period September 15, 1989-September 14, 1990**

Dr. A. I. Kingon, Dr. R. F. Davis and Dr. A. K. Singh

**Submitted to U. S. Army Research Office
Research Triangle Park, North Carolina**

on December 14, 1990

Contract No. DAAL03-89-K-0131

**Department of Materials Science and Engineering
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Brief Outline of Research Findings

The objectives of this program are to:

- (1) Determine whether crystalline nonoxide powders can be synthesized in a nonthermal microwave plasma.
- (2) Develop the necessary equipment to undertake the synthesis investigations, and later develop it for completely anaerobic synthesis and post-processing.
- (3) Characterize the properties of ultrafine nonoxide particles, particularly diffusion and creep in assemblies of these particles. The interest is in "superplastic" effects.

In the first year of research, significant progress has been made. It has been shown that nonthermal synthesis is indeed feasible. The demonstrations have been performed on SiC, Si₃N₄ and Si to date. Preparations are being made to handle the precursors for AlN synthesis. Two presentations have been made of this work, at the Fall MRS meeting in Boston. Three manuscripts have been prepared, one of which is in the review process, and two of which are in draft form (attached.)

In addition, the first investigation has been made of the high temperature creep of polycrystalline AlN. Under constant compressive stress the conditions investigated were 100-350 MPa and 1150-1400°C. The activation energy curve shows a knee at ~1300°C. In the low temperature range (activation energy ~400 kcal/mol) the mechanism is grain boundary diffusion controlled creep; at high temperatures (activation energy ~700 kcal/mol) it is lattice diffusion controlled.

Some further details of our results are presented below:

Nonthermal Synthesis of Silicon Carbide

Synthesis was performed at 1 torr using SiH₄, C₂H₂ and Ar carrier. Chemical composition was confirmed by XPS and Auger. Phase analysis was by XRD and TEM. Particles are nanosized, 2-5 nm. Many structural modifications were observed, including 3C, 2H, 4H, 15R, 21R. We showed that the results have implications for existing models of polytype formation.

Non-Thermal Synthesis of Silicon Nitride

Similar procedures were utilized for Si₃N₄, although growth rates were lower. The phase is a (hexagonal symmetry). Particle sizes 2-5 nm. Reaction efficiencies have been measured.

Crystalline Nanoparticles of Si

It was observed that the formation of SiC is via Si intermediates or nuclei. Thus synthesis of Si was investigated. Conditions were optimized, and particles characterized as above. The crystallinity is unusually high, considering the size of the particles.

Plasma Characterization

After design and construction of the apparatus, the microwave plasma was characterized by Langmuir probe.

Manuscripts Submitted

1. Synthesis of SiC Clusters in a Nonthermal Microwave Plasma
A. K. Singh, A. I. Kingon, G14.2, MRS Fall Meeting 1990, 11/26-12/1/90 (in review)
2. Nonthermal Synthesis of SiC and Si₃N₄ Ultrafine Particles
A. K. Singh, A. I. Kingon, to be submitted to J. Mat. Res. (draft attached)
3. Crystalline Nano-particles of Si by Nonthermal Microwave Plasma
A. K. Singh, A. I. Kingon, to be submitted to J. Mat. Res. (draft attached)

Presentations

1. Synthesis of SiC Clusters in a Nonthermal Microwave Plasma
G14.2, MRS Fall Meeting 1990
2. Nonthermal Synthesis of Ultrafine SiC Powders
N2.7, MRS meeting Fall 1990

Scientific Personnel Supported

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4. J. Cirra (M.S. Student)

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