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**REPLACEMENT OF FIRST FIRE COMPOSITION IN M127A1
GROUND ILLUMINATION SIGNAL**

Russell N. Broad

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US ARMY
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ARMAMENTS COMMAND
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13. ABSTRACT (Maximum 200 words) A study was conducted to replace the first fire composition in the M127A1 ground illumination signal. The original first fire composition contained tetranitrocarbazole, a sole source material, and barium nitrate, a toxic material. Program costs were minimized by choosing a presently used first fire composition as a replacement. A data base of such compositions was created. It was used to pick candidate replacement compositions. The compositions were loaded into illuminant assemblies and tested statically. Results from this test showed that Starter Mix (SM) XXV was the best candidate composition. It was loaded into complete signals that underwent ballistic testing. Signals with SM-XXV met all applicable Military specification requirements. The success of the program justified the approach of choosing a currently used first fire composition.			
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INTRODUCTION

The pyrotechnic first fire compositions covered by MIL-P-48240 are used on a wide range of colored signals and illuminating projectiles. There are three of these compositions, each formulated to give a distinctive color. Table 1 give the formulations for these compositions. TNC is a common constituent of these compositions. TNC was first synthesized in the late 1800s and found use in pyrotechnics in World War II (ref 1). Picatinny Arsenal investigated its use as a first fire constituent in the early 1950s (ref 2). It is considered an explosive material. Upon ignition it produces considerable energy and essentially gaseous products. These properties enable the first fire compositions to ignite illuminating and signal compositions. The first fire compositions are either pressed on top of, or brushed onto, the surfaces of illuminating and signal compositions pressed into canisters or sleeves. The first fire compositions are initiated by expelling charges or ignition compositions. Through the years, the first fire compositions have given reliable, consistent performance.

In the early 1990s, the availability of TNC became uncertain because its sole producer did not want to manufacture it anymore. Although another company picked up production, availability could not be guaranteed. TNC has no commercial market and the military market has declined dramatically. The resulting low demand make its production marginally profitable. Thus, there is little motivation to continue its production. This situation led to execution of an engineering study to eliminate use of TNC. A secondary purpose of the engineering study was the replacement of barium nitrate in the igniter compositions. This material is in two of the pyrotechnic first fire compositions. Like all water soluble barium salts, it has high acute toxicity. Further, disposal of waste containing it is relatively expensive because it cannot be landfilled. This report describes the results of the engineering study.

TECHNICAL APPROACH

One approach to remove TNC could have been reformulating the first fire compositions without it. This would have resulted in novel compositions. The approach taken, however, was replacement of the three first fire compositions with a pyrotechnic composition that was presently being used on other items. The advantages of this approach follow. First, it minimized testing required for qualification since a presently used composition had a history associated with it. This included functioning performance in various end items, environmental considerations, cost and data on storage stability, safety, thermodynamics, and kinetics. Second, manufacturing procedures and drawings already existed for the composition. Third, use of one first fire composition would simplify manufacturing since one batch of composition could be used for various items, regardless of the items' signal or illumination colors.

One possible drawback to this approach was the elimination of first fire compositions that were tailored to the illuminating or signal composition color. Several individuals knowledgeable in the field use the colored signals revealed that the color of the first fire flash was insignificant in affecting an observer's ability to distinguish illuminant or signal colors. The flash of the first fire is of very quick duration (<500 ms) compared to the burn time of the illuminants or signals (>25 sec). Thus, elimination of the color requirement for the first fire compositions was of no concern and this approach was pursued.

The program was conducted in three phases. First, other presently used pyrotechnic compositions were identified and pertinent data was collected on them. For maximum flexibility, we did not confine our search to only igniter and first fire compositions (henceforth we will use the term first fire to apply to igniter compositions as well). A data base was created from which the best candidate compositions were selected. Second, static functioning tests were performed on illuminant assemblies which contained the candidate first fire compositions and the standard first fire compositions. These tests determined if the candidate replacement compositions would have any adverse affect on the static requirements for the illuminating assemblies. Third, items were loaded with both the candidate replacement composition and the standard first fire composition, and underwent first article ballistic testing.

RESULT

Composition Data Base/Selection of Candidate Compositions

Identification of presently used first fire compositions was begun by generating a list of drawings that contained words igniter, ignition, and first fire in their titles. These drawings were then obtained. A second source for such compositions was reference 3. This procedure ensured that the vast majority of, if not all, presently used first fire compositions were identified. With this information, a minimal data base that included first fire formulations and common names was created. Further information was then added. This included impact, electrostatic and friction sensitivity data, hazard class, heat of reaction, autoignition temperature, burn rate, cost per pound, and qualitative ranking of toxicity. Not all of his data could be obtained for every composition. The data base is shown in table 2. The references for the various data are cited in the table.

Criteria were developed to select the most promising candidate replacement compositions. Criteria included toxicity, cost, sensitivity, availability of constituents, history of problems, and burning characteristics. It was decided to eliminate those that would be least likely selected. The first criterion for elimination was presence of acutely toxic or carcinogenic constituents. This included barium, lead salts, and chromates. This consideration eliminated many of the compositions shown in table 2 as candidates. The remaining compositions were eliminated because of safety or processing issues, autoignition temperatures exceeding 500°C (ref 4), or other considerations such as cost. Since the first fire compositions are in contact with the flash from expelling charges for short duration's, they must reach their autoignition temperatures quickly. This is more easily accomplished if the autoignition temperature is relatively low. The remaining compositions, which were chosen as candidates, were Starter Mix XXV, IM-6, and I-548 (no. 10, 34, and 40, respectively in table 2). Table 3 shows further detail on these compositions.

Static Tests of Candidate Compositions

Composition I-548 was dropped from consideration because the required grade of calcium resonate could not be obtained easily. Only starter Mix XXV and IM-6 were evaluated statically. Table 4 is a matrix of assembly types and quantities statically tested. Assemblies with the standard first fire compositions served as controls. The choice of assemblies was based upon the items planned for the qualification tests. In turn, the choice of items was based upon what items would be in production during the duration of the program.

The required parts for the assemblies were ordered and tooling was fabricated for loading the assemblies. The Pyro Systems Branch of the U.S. Army Armament Research, Development and Engineering Center (ARDEC), Picatinny Arsenal, New Jersey mixed all the pyrotechnic compositions and loaded all the assemblies used in the static tests. The Branch also conducted all the static tests in its own flare tunnel. Table 5 is a roll up of averaged static test parameters for the assembly/first fire combinations tested.

The efficiency data shows that SM-XXV offers significantly better performance over IM-6 for the M125A1 and M127A1, while the IM-6 is much better for the M158. They are nearly equal for the M583A1. SM-XXV was better than the standard first fire for all items except the M583A1. Here, the standard was approximately 13% higher. Based on this static data, as well as SM-XXV's lower cost, we decided to test SM-XXV in the ballistic testing.

Ballistic Testing for Prove Out of New First Fire Composition

In consultation with ARDEC's Product Assurance and Test Directorate, a test plan was drafted for ballistic testing. The ballistic test consisted of functioning tests at conditioning temperatures specified in the military specifications. Additionally, some illuminant assemblies from the lot used for the ballistic test were statically tested. Quantities were per first article requirements. This minimal amount of testing was justified since changing the first fire would in no way affect the hardware and other energetic material fills in the item.

The ballistic tests were conducted by Thiokol Corporation at Longhorn Army Ammunition Plant (LHAAP). During the time period in which static testing at ARDEC concluded and ballistic tests begun, numerous production lines were closed at LHAAP. Startup of these lines exclusively for this program would have been prohibitively expensive. Consequently, only the M127A1 signal experienced ballistic testing.

Thiokol loaded the items per the technical data package requirements and performed testing per the Scope of Work. Table 6 shows their static burn data. The candlepowers measured at LHAAP were higher than ours; this was due to differences in the tunnels and test procedures. As expected, the candlepower achieved with SM-XXV was higher than FF-I. Since exact illuminant weight data was unavailable, efficiencies are not reported. Table 7 presents the ballistic (flight) data. The data shows that aside from two failures at 70°F, all signals met the requirements. One failure was non-expulsion of the signal; the other, failure of the round to expel. Neither was related to first fire function. The only ballistic parameter that could have been possibly affected by the change in first fire was the burn time. The differences between the signals with FF-I and SM-XXV for this parameter were small and within the standard deviations at all temperatures.

CONCLUSIONS

The success of the project vindicated the selected approach. Choosing a presently used igniter composition kept the cost and technical risks of the program as low as expected. The need for extensive testing above first article requirements was eliminated.

RECOMMENDATIONS

Other items incorporating igniter compositions with sole source, toxic, or environmental/ objectionable constituents should be evaluated for igniter replacement. Programs to achieve this objective would be of minimal scope since they would have the data base of igniters generated in this program as a starting point.

Table 1
Formulations for pyrotechnic first fire compositions

<u>Constituent</u>	<u>Requirements</u>	<u>Nominal weight percent</u>		
		<u>Type I</u>	<u>Type II</u>	<u>Type III</u>
Barium nitrate	MIL-B-162 Average particle size $\leq 20 \mu$	50	---	50
Strontium nitrate	MIL-S-20322 Grade A or B	---	50	--
Tetranitrocarbazole	MIL-T-13723	10	10	10
Silicon	MIL-S-230 Average particle size $\leq 10 \mu$	20	16	13
Zirconium hydride	Commercial	15	15	20
Polyvinyl chloride	MIL-P-20307	---	5	3
Laminac 4116 + 1% Lupersol DDM catalyst Color requirement	Commercial	5 Yellow	4 Red	4 Green

Table 2
Database for first fire and pyrotechnic compositions

Table 2 Database for first fire and pyrotechnic compositions

No Fuel (Percent):	Oxidant (Percent):	Additive (Percent):	Binder (Percent):	Solvent (Percent):	Type of Comp. (BM)	Cost E.Sta.(J) Uncon/Con Steel/F PA, (*) (BM)	Fric.P Steel/F PA, (*)	Impact BR	Heat of Rec Cal/g Auto/DTA	Ig.Temp Degrees C	Compat Toxicity	
1 Lead Azide					Exp	0.007	Exp/Exp 4-5					
2 Nitrocellulose					Exp	0.049J	Burns/ 3-5		160	D /1.1	Not Toxic	
3 PETN					Exp	.06 / .21	Cr./NR 6				/1.1	
4 TNT					Exp	.06 / 4.4	NR/NR 14	10.60		/0475	/1.1	
5 RDX					Exp	>11.03	Exp/NR 8	1240	260	(5s)	/1.1	
6 Black Powder					Exp	3.00	12.5/0.8 Snp/NR 16	684	288	(5s)	/1.1	
7 98.5-99 RDX		(1.5 St.ACid)			Exp.	11.025	Cr./NR 8 -12		190		/1.1	
8 30 Charcoal	70 KN03				St.MIXII	3.24	0.75 BM NR/NR 310 NR	1.9s/mg	980	418	G /1.3 MOD.	
9 83.3 Dry Mix Dry Mix (#9): 26/4/13/51/C/Al Binder Solu(#9):			16.7 B.Solu		StartMix	2.35	1.25 NR/NR 15 BOE	0.3s/mg	401		G /1.3 MOD	
10 26/13/4/51/Al/C	35/22/KN03/Fe203		6 NC	94Acetone		0.59					D /1.1 Slight	
11 60 Dry Mix Dry Mix (#11): 16.8/10/5/C/Starch Binder Solu(#11):	35/22/KN03/Fe203		40 B.Solu.		St.MIXIXV	2.71	1.5 BM NR/NR	5.0s/in	1186	421	/1.3 SLIGHT	
12 35 pts Sb	35/30pts;CaS12/KCl04		4 NC	96Acetone		0.42		3.2BOE	25 s/in	946	216	
13 40 Al	30/30;Ba(N03)2/KCl04		6pts(B NC	92 Ace.)		0.51		15 BOE	13s/in	1812	446	/1.3
14 23/15; Al / S	62 Ba(N03)2				PFp-555			26		2147	700 (5s)	
15 91 Bk.Powder	9 Al				Flash	NR 3.25	Cr/ 22 ERL					D /1.1 SL-MOD.
16 34/26; Mg/Al	40 KCl04				Flash	3.09	0.225 NR/NR 14	0.2s/mg		344	D /1.1 MOD.	
17 25.5/12.8; Mg/Al	25.5/28.1;BaCr04/KCl				PFp-I	7.53	3.2/0.73 Snp/NR 27 -36			762	D /1.1 TDx	
18 50/50; FF/Flare	90 BaCr04				FF	22.4	0.0125 CO/CD 6			380	A /1.1 TOXIC	
FFI 10 B	28 Teflon		7 Hycar	MEK or Acetone	FF,Flare	8.89	0.051 CB/CB 12			550	G /1.3 MOD	
Flare: 65 Mg					FF-906	10.1	0.0034 CB/CB 16	0.6s/in	480	615	/1.1 CAR.SUS.	
19 DryMix & B.Solu. Dry Mix (#16): 25/25; Ti/Si Binder Solu:	25/25; Fe304/Pb304		(10.9 NC	89.1 Ace- tone)	FF,VI1	0.875	NR/NR	3.0s/in	360	3762	G /1.3 MOD	
20 30 W	55/10;BaCr04/KCl04	5 D.earth	Uiton A		FFMIX			15 BOE	3.0s/in	360	762	
21 25/25; Si/Ti	50 Pb304	(Graphite	NC		FF,MIxX	1.29	1.625 NR/NR NR 915	0.3s/mg	275	780	G /1.1 MOD	
22 10 B	90 Pb02				E.Acetate FF,PComp	7.90	0.0125 CB/CB 2			327	A /1.1 MOD	

Table 2
(cont)

Table 2 Database for first fire and pyrotechnic compositions

No Fuel	Oxidant	Additive	Blinder	Solvent	Type	Cost	E.Sta.	Fric.P	Impact	BR	Heat	Ig.Temp	Compat	Toxicity
(Percent):	(Percent):	(Percent):	(Percent):	(Percent):	of Comp.	\$/Lb	Uncom/Con	Steel/F	PA ₁ (*)	**As	of Rec	Degrees C	Gr/Haz	*****Ref
					(BH)				Shown		Cal/g	Auto/DTA	(DOD)	
23	33 Bk Powder Red Pyro Comp. 1 2/17 Mg, Gran 1 Mg, Gran 4	67 Red Pyro. Comp. 13/21.4 NaNO ₃ /KClO ₄ 7.5 Glisonite 2.8 Graphite		FFfM131S									D /1.1 MOD CAR.SUS	
24	30 Ti	70 Fe2O3		FF30M1XV	1.09 N/R			Cr/NR	10 BOE	6.5s/in	659		9 /1.3 SLIGHT	
25	FF: I, II & III, TNC Pyro. Comp. 1			FF: I, II, III	9.76			Cr-a/NR	26	680	476		0 /1.1 MOD	
	I:20/15; Si/ZrH2 II:16/15; Si/ZrH2 III:30/20; Si/ZrH2	10TNC 10TNC; 5PvCl 10TNC; 3PvCl	5 Lam.4116 4 Lam.4116 4 Lam.4116	FFII	3.78					8.6s/in			CAR.SUS	
	50 Ba(NO3)2 50 Sr(NO3)2 50 Ba(NO3)2			FFIII	3.67 3.64					12 s/in 12 s/in			TOXIC CAR.SUS	
26	TNC, FFIII(Slurry)	Pluronic F68	Cell.Nitrate	FFM125A1										
27	TNC, FFII(Slurry)	Pluronic F68	Cell.Nitrate	FF127A1										
28	TNC, FFII(Slurry) or 16 B Powder, Cl-8	Pluronic F68 Pluronic F68 Pluronic F68	Cell.Nitrate Cell.Nitrate Cell.Nitrate	FF158A1 FF158A1 FF158A1										
29	TNC, FFII (Not A Slurry) or 19 B powder 19 B Pwd.; 19 B 18/58; Ty IV Tef/KNO3			FF583A1	3.78									
30	20 Si	80 Pb304		FFM201A1	0.82			NR/NR	>15BOE	4.4s/in	395		/1.3 TOXIC	
31	10 Si	90 Pb304		FFMIX	0.65			NR/NR		1.5s/in	256		/1.3 TOXIC	
32	25 B VAAAR is no longer being produced, An old quotation	75 KNO3	I UAAAR	I-SpMix	19.3	0.124		CB/NR	10	2.3s/in	1594		414/431.2 /1.1 MOD.	
33	50 Si	20/30; PbO2/CuO	9 \$19.00 / lb. was	Igniter	1.67				15		380		476	TOXIC
34	40 Si	54 KNO3	6 Viton A	IM-6	3.82	> 1			> 2477	> 49.6	11.6in/ ft. lbs		673.5	6 MOD
35	65 Zr	25 Fe2O3	10 D.Earth	I-A1A	>69	0.0024		CB/NR	24	.06s/mg	550		427	6/1.3 Slight
36	21 Zr	79 BaCrO4	VAAAR added	DP-162	25.6	0.0013		PB/NR	23	1.0s/in	376		418	/1.1 CAR.SUS
37	19 B	58/18; KNO3/TFE	5 Lam4116	SI-282	0.283			Spk/NR	9				NR(Ssec)	D/1.1 MOD.
38	23.1 B	70.7 KNO3	0.5PluronicF68	Igniter	1.08M			NR	13		1600		400/565	D /1.1 MOD
39	16.5 Mg	80.5 BaO2	2/1; CaRes/Gr- phite	I-527	1.95	1.25		SpS/NR	23				375	6 /1.3 CAR.SUS
40	15.0 Ty-III	Mg(Gran 65.0 SrO2	7/13; TyI/TyII Ca Resinate	I-548	0.05	8M		SpkS/NR	8				239	MOD
41	6 Mg (Gr.12) 1-136; 10 Ca Res.	94 1-136 90 SrO2		I-194 I-136	5.97 0.05	0.25 0.05		Spns/NR Spns/NR	35ERL 16	.5s/mg			287 280	6 /1.1 Slight 6/1.1 Slight

Table 2
(cont)

Table 2 Database for first fire and pyrotechnic compositions

No Fuel (Percent):	Oxidant (Percent):	Additive (Percent):	Binder (Percent):	Solvent (Percent):	Type of Comp.	Cost \$/Lb Uncon/Con	Fric.P Steel/F	Impact BR Pa ₁ (^h)	Heat of Rec Cal/g	Ig.Temp Degrees C Auto/DTA	Compat Gr/Haz (DDD)	Toxicity
42 5 B	95 BaCrO4				DP-T-10	7.15	CB/NR	>40	1.9s/in 265	553/675	/1.1	CAR.SUS
43 10 B	90 BaCrO4				DP-479	10.1	CB/CB	12	0.7s/in 480	615/705	/1.1	CAR.SUS
44 10 B	90 BaCrO4		1 VAAR (See record #32)		JP-879	9.70	CB/NR	24	1.5s/in 463	560	/1.1	CAR.SUS
45 15 B	85 BaCrO4				DP-523	13.2	CB/NR	26	1.5s/in 502	/645	/1.1	CAR.SUS
46 19 B	81 BaCrO4				DP-T-10	15.7	CB/NR	10	2.0s/in 276	656	/1.1	CAR.SUS
47 15 B	44/41; BaCrO4/Cr2O3				DelayMix	13.9			4.5s/in			CAR.SUS
48 14 B	44/42; BaCrO2/Cr2O3				DelayMix	13.2			6.5s/in			CAR.SUS
49 13 B	41/44; BaCrO2/Cr2O3				DelayMix	12.5			8.5s/in			CAR.SUS
50 50 W	40/10; BaCrO4/KClO4				DelayMix	7.27	NR/NR	22 BOE	12s/in 233	270	/1.3	CAR.SUS
51 20 W	70/10; BaCrO4/KClO4				DelayMix	5.68			41s/in			CAR.SUS
52 9 70/30Ni alloy & 17 30/70Ni alloy	60/14; BaCrO4/KClO4				DP-1415; Tyll	17.9	CB/NR	>40	6.0s/in 521	325	/1.3	CAR.CUS TOXIC
53 3 70/30 ZnNi alloy & 23 30/70ZnNi alloy	60/14; BaCrO4/KClO4				DP-1415; Tyll	16.6	CB/	>40	11s/in 521	325	/1.3	CAR.SUS TOXIC
54 55 Mn	45 PbCrO4				DP-D16	1.07			2.2s/in 230			CAR.SUS
55 93 Mn	30/37; BaCrO4/PbCrO4				DP-D16B	1.99	NR/NR	15 BOE	8.4s/in 256	460		CAR.SUS
56 32.8 Mn	37/30.2; BaCrO4/PbCrO				DP-D16C		NR/NR	15 BOE	13s/in 262		/1.3	CAR.SUS
57 28 Zr	72 PbO2				DelayMix	22.5			4.5s/in			Toxic
58 5/31; Zr/Ni	42/22; BaCrO4/KClO4				DP-T-2	10.5			6.5s/in			CAR.SUS
59 5/17; Zr/Ni	70/8; BaCrO4/KClO4				DP-HP-25	9.71			18 s/in			CAR.SUS
60 32-58 W	32-56 BaCrO4 & 10-14 KClO4		VAAR (See record #32)		Delay P.	7.04	NR/	36	11.03	270		MOD
61 53 Zr	21/26; KClO4/McO3				SI-98					1174	372	
62 48.7 Zr	31.3/20; MoO3/Cr2O3				SI-113	54.1	CB/CB	34	0.8s/in 605	400	G/1.3	Low
63 40 Zn; 20 Al	20 KClO4; 20 KNO3				PPF-600	3.82		14		700	G /1.3	MOD
64 60-67 Al	33-40 KClO4				PPF-600	5.9	CB/NR	24		2284	D /1.1	
65 22.5/10; Al(F1.)/S	64/3.5; KClO4/SbS2				M-80							

Table 2
(cont)

Table 2 Database for first fire and pyrotechnic compositions

No Fuel (Percent):	Oxidant (Percent):	Additive (Percent):	Binder (Percent):	Solvent (Percent):	Type of Comp.	Cost E.Sta.(J) Uncon/Con (BM)	Fric.P Steel/F FA, (T)	Impact BR ***A Shown	Heat offic Cal/g Au/DTA	Ig.Temp Degrees C	Compat G/Hiz 0000	Toxicity *****Ref
66 40.0 Zr	60 BaCrO4								502			
67 M1 Propellant				Propel't		11.03	NR/NR	4 -6			C /1.1	
68 M9 Propellant				Propel't		5.2	Sp/NR	2 -3			C /1.1	
69 M30 Propellant				Propel't		>12.5	NR/NR	4			C /1.1	
70 42 30/50 Mg	44 Sr(NO3)2	7 Declarane (See record #32)		111,RT		>50	Cr/NR	19		344	D /1.1	High
71 48-65 20/50 Mg	31-47.5 NaN03	4-4.5 VAAR (See record #32)		FY-1450		11.82	NR/NR	21		/518	D /1.1	MOD
72 75 30/50 Mg	10 Teflon		15 Viton A	Acetone	111,F,IR	9.00	NR/NR	18		400	A /1.3	SL-MOD.
73 50 30/50 Mg	38 NaN03		5 Lam4116	Acetone	FY-1444	4.28	NR/NR	19		1456	/1.1	SL-MOD
74 45.9 30/50 Mg	34 NaN03		9 Lam4116	Acetone	FY-1192		NR/NR	20		610	A/	
75 46/ 20/50 Mg	45 NaN03		7 Hycar	MEK or Ac	111,F,IR	7.60	NR/NR	16		400	G /1.3	SL-MOD
76 85 Mg	29 TFE		5 Lam4116		111,M125	11.02						
77 Opt. Fuel: 33 Mg 30/50 of: a.Ty-1(Sp.14067)or b.Ty-111(Sp.392C)or c.Ty-IV, E1, Sp.14067	46 Ba(NO3)2, Cl-2 (30um)	16 Pvc1										
78 Opt. Composition: 66 Mg30/50Ty1(14067 or 65 Mg30/50Ty1(14067	29 NaN03 31 NaN03		5 Lam4116 4 VAAR		111,M127	11.02						
79 33 Mg30/50Ty1 14067	48 Sr(NO3)2,GrB	15 Pvc1	4 VAAR		111,M158	11.02						
80 28 Mg30/50Ty1 14067 & 20 Mg50/100E1 14067	41 NaN03,Gr-B,Cl-2		11Lam4116		111,583	11.02						

DATA COLLECTION

The thermal and stability data were collected from IC808 statements and from a Noves Publication Handbook of Toxic and Hazardous Chemicals, 1991. A compilation of Hazard and Test Data for Laboratory reports on entitled "Thermal Stability of Pyrotechnic Compositions" by G. Nepp et al., and a second entitled "Compilation and Evaluation of Lignite Temperatures, Particle Size, and Thermal Stability of Pyrotechnic Compositions" by G. Nepp et al., G. Gordon and S. O. Crane.

In order to expedite the collection of burning rate data, the test was performed on a different set of test cells. Some of these rates were calculated from IC808 statements which presented burning times for 30 mg of loose composition. The remainder of these rates were made from the test cells. The common practice is to load data at pressures between 20,000 and 40,000 psi.

The toxicity and compatibility data were collected from IC808 statements and from a Noves Publication Handbook of Toxic and Hazardous Chemicals, 1991.

Cost data was collected from the latest editions of the monthly Chemical Marketing Reports published by the Chemical Publishing Company of New York and by direct contact with users and Manufacturing Companies. Finally, a few propellants and explosives were used for comparison.

Table 2
(cont)

Abbreviations

E. Sta.	= Electrostatic	Pyro	= Pyrotechnic
J	= Joules	Hexachl'b	= Hexachlorobenzene
Fric. P	= Friction	TNC	= tetranitrocarbazole
BR	= Burning rate	PvCl	= Polyvinyl chloride
Ig. Temp	= Ignition temperature	LAM	= Laminac
Compat	= Compatibility	Cel	= Cellulose
Comp	= Composition	Pwd.	= Powder
Uncon	= Unconfined	Ty	= Type
Con	= Confined	Tef	= Teflon
F	= Fiber	P.ester	= Polyester
PA	= Picatinny Arsenal	CaRes	= Calcium resinate
Rec	= Recation	Sps, Spks, Spk	= Sparks
DTA	= Differential thermal analysis	Pb	= Partial burn
Gr	= Group	Propel't	= Propellant
Haz	= Hazard	TFE	= tetraflouroethylene
BM	= Bureau of Mines	Cl	= Class
Exp	= Explodes	Sp	= Specification
Cra, Cr	= Crackles	EI	= Ellipsoidal
NR	= No reaction	Opt	= Optimal
St.	= Stearic		
Snp, Snps	= Snaps		
St.	= Starter		
Mod	= Moderate		
B	= Binder		
BOE	= Bureau of Explosives		
Solu	= Solution		
Nc	= Nitrocellulose		
pts	= parts		
Ace	= Acetone		
Tox	= Toxic		
SI-Mod	= Slight to moderate		
ERL	= Energetics Research Laboratory		
Bk.	= Black		
B.Acetate	= Butyl Acetate		
CD	= Complete detonation		
CAR. SUS	= Carcinogen suspect		
MEK	= Methyl Ethyl Ketone		
CB	= Complete burning		
D.earth	= Diatomaceous Earth		
E. Acetate	= Ethyl acetate		

Table 3
Candidate first fire compositions

<u>Constituent</u>	<u>Requirements</u>	<u>Nominal weight percent</u>		
		<u>SM-XXV</u>	<u>IM-6</u>	<u>I-548</u>
Silicon	MIL-S-230 Grade II, Class C	25.7	40.0	---
Potassium nitrate	MIL-P-156 Class I	34.6	54.0	---
Charcoal	JAN-C-178 Class D	4.0	---	--
Aluminum powder	MIL-A-512 Type II, Grade C, Class 4	12.8	---	---
Red iron oxide	MIL-I-275 Grade D	21.7	---	---
Nitrocellulose	MIL-N-244 Grade D	1.2	---	---
Viton A	Commerical	---	6.0	---
Strontium peroxide	MIL-S-612 Grade B	---	---	65.0
Calcium resinate	MIL-C-20470 Type II	---	---	7.0
Calcium resinate	MIL-C-20470 Type I	---	---	13.0
Magnesium powder	MIL-M-382 Type III, Granulation 12	---	---	15.0

Table 4
Static tunnel (ARDEC) data for various illuminant assemblies

<u>Assembly for</u>	<u>First Fire</u>	<u># Assemblies</u>	<u>Burn Time, sec</u>	<u>Average Candlepower</u>	<u>Average Efficiency candle-gram/sec</u>	<u>Average Color Value</u>
M125A1 Green Star	FF-III	10	5.1±0.3	5516±1040	2347±430	0.42±0.02
Cluster Ground	IM-6	10	5.6±0.5	4196±1165	1924±471	0.41±0.01
Illumination Signal	SM-XXV	10	5.4±0.5	5962±991	2677±282	0.41±0.00
M158 Red Star	FF-II	10	4.2±0.4	19349±3696	6811±1257	0.54±0.01
Cluster Ground	IM-6	10	4.8±0.4	21177±2897	8511±1533	0.54±0.01
Illumination Signal	SM-XXV	10	4.6±0.5	20323±1193	7717±717	0.53±0.00
M127A1 White Star	FF-I	10	37.9±1.5	76788±4367	34302±2084	0.05±0.00
Parachute Ground	IM-6	10	39.0±1.2	56962±1734	26275±8204	0.04±0.00
Illumination Signal	SM-XXV	10	38.6±0.8	80780±5359	36784±2806	0.05±0.00
M583A1 White Star	FF-I	10	26.0±2.9	110631±4513	36827±3874	N. A.
40mm Parachute	IM-6	10	21.0±4.6	112867±16744	32251±3358	N. A.
Cartridge	SM-XXV	10	23.2±2.4	106327±11498	32369±2655	N. A.

Table 5
 Static tunnel (LHAAP) data for standard and candidate illuminant assemblies

<u>Assembly</u>	<u>First fire</u>	<u>No. assemblies</u>	<u>Average burn time, sec</u>	<u>Average candlepower</u>
M127A1	FF-I	20	34.0 ± 0.8	114,100 ± 5,890
M127A1	SM-XXV	20	31.0 ± 0.9	135,100 ± 9,010

Table 6
Ballistic data for signals conditioned at -65°F

	M127 with FF-I	M127 with SM-XXV	Requirements
Altitude, feet	# Fired	16	16
	# Functioned	16	16
	Average	715±38	700±76
	Maximum	805	784
	Minimum	652	433
Angle, degrees	Average	6.0±3.2	11.0±9.1
	Maximum	14	44
	Minimum	2	3
Chute Delay, seconds	Average	0.82±0.08	0.82±0.15
	Maximum	0.92	1.29
	Minimum	0.66	0.63
Burn Time, seconds	Average	37.0±1.3	37.0±1.2
	Maximum	39.1	38.7
	Minimum	34.1	35.1

Table 7
Ballistic data for signals conditioned at 70°F

	M127 with FF-I	M127 with SM-XXV	Requirements
# Fired	32	32	32
	32	30	32
	815±34	821±42	>725
Altitude, feet	889	916	None
	719	743	500
	4.0±2.4	4.0±2.4	≤12
Angle, degrees	10	9	30
	1	1	None
	0.72±0.06	0.74±0.09	None
Chute Delay, seconds	0.91	0.95	5
	0.64	0.59	None
	32.9±1.2	32.5±0.8	None
Burn Time, seconds	34.7	34.9	None
	30.6	30.5	25

Table 8
Ballistic data for signals conditioned at 165°F

	M127 with FF-I	M127 with SM-XXV	Requirements
# Fired	32	32	32
# Functioned	32	32	32
Average	828±34	845±25	None
Maximum	886	916	None
Minimum	684	803	None
Average	4.0±2.5	4.0±2.4	None
Maximum	10	8	None
Minimum	1	0	None
Average	0.67±0.08	0.69±0.10	None
Maximum	0.83	1.06	None
Minimum	0.53	0.56	None
Average	31.2±1.3	30.6±1.3	None
Maximum	35.1	33.8	None
Minimum	28.8	28.4	None

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