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AFRL-SR-BL-TR-98-

0624

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<b>1. AGENCY USE ONLY (Leave blank)</b>	<b>2. REPORT DATE</b> 8/21/98	<b>3. REPORT TYPE AND DATES COVERED</b> Annual Technical 8/31/97 - 9/1/98	
<b>4. TITLE AND SUBTITLE</b> (U) Theoretical and Computational Studies of Nucleation in Supercritical Fuels		<b>5. FUNDING NUMBERS</b> PE - 61102F PR - 2308 SA - BS G - F49620-96-1-0169	
<b>6. AUTHOR(S)</b> Pablo G. Debenedetti		<b>8. PERFORMING ORGANIZATION REPORT NUMBER</b>	
<b>7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES)</b> Princeton University Department of Chemical Engineering Princeton, NJ 08544-5263		<b>10. SPONSORING / MONITORING AGENCY REPORT NUMBER</b>	
<b>9. SPONSORING / MONITORING AGENCY NAME(S) AND ADDRESS(ES)</b> AFOSR/NA 110 Duncan Avenue, Suite B115 Bolling AFB DC 20332-0001		<b>11. SUPPLEMENTARY NOTES</b>	
<b>12a. DISTRIBUTION / AVAILABILITY STATEMENT</b> Approved for public release; distribution is unlimited  DATA QUALITY INSPECTED 1		<b>12b. DISTRIBUTION CODE</b>	
<b>13. ABSTRACT (Maximum 200 words)</b> The goal of this research is to improve basic understanding of how thermodynamics and nucleation kinetics influence the deposit of pyrolytic products from supercritical fuels. The precipitation of picene from supercritical methylcyclohexane was investigated theoretically, under the assumptions of both thermodynamic and kinetic control. The kinetic calculations indicated that a negligible amount of deposits were formed as a result of friction-driven depressurization in tubular flow. However, the presence of small amounts of particulate impurities dramatically enhanced the rate of deposit formation. For isobaric cooling conditions, the rate of deposit formation was found to depend sensitively on whether the mixture entered the retrograde region or not. In the retrograde region where solubility decreases with increasing temperature, isobaric cooling was found to suppress deposit formation. The time evolution of isolated quiescent droplets exposed to an excess of partially miscible compressed solvent was investigated under sub-critical conditions. Over a wide range of conditions the calculations showed appreciable droplet swelling, followed by evaporation. An aerosol dynamics code was combined with mass, momentum, energy balances and equation of state calculations to yield the evolution of the particle size during steady, one-dimensional friction-driven expansions of a supercritical fluid containing a dissolved solute.			
<b>14. SUBJECT TERMS</b> Homogeneous and heterogeneous nucleation; supercritical fuels; deposit formation		<b>15. NUMBER OF PAGES</b> 8	<b>16. PRICE CODE</b>
<b>17. SECURITY CLASSIFICATION OF REPORT</b> Unclassified	<b>18. SECURITY CLASSIFICATION OF THIS PAGE</b> Unclassified	<b>19. SECURITY CLASSIFICATION OF ABSTRACT</b> Unclassified	<b>20. LIMITATION OF ABSTRACT</b> UL

# THEORETICAL AND COMPUTATIONAL STUDIES OF NUCLEATION IN SUPERCRITICAL FUELS

(AFOSR GRANT F49620-96-1-0169)

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## SUMMARY/OVERVIEW

The objective of this research is to improve the basic understanding of how thermodynamics and nucleation kinetics influence the deposit of pyrolytic products from supercritical fuels. We investigate the precipitation of picene from supercritical methylcyclohexane under both thermodynamic and kinetic control. The kinetic calculations are useful for identifying operating conditions that minimize deposit formation. We also model the time evolution of single organic droplets exposed to an excess of miscible supercritical fluid, the typical configuration used in the supercritical anti-solvent (SAS) process for the production of microparticulate polymeric powders. Finally, we compute the particle size distribution during the adiabatic expansion of a supercritical fluid containing dissolved solutes, accounting for nucleation, condensation, and coagulation.

## TECHNICAL DISCUSSION

The Statistical Associating Fluid Theory (SAFT) was used to model the fluid-phase equilibrium of picene in methylcyclohexane (Huang and Radosz, 1990). An expression based on classical nucleation theory with due account for fluid-phase nonideality was used to calculate the deposition rate of picene from methylcyclohexane (Debenedetti, 1990). Steady, one-dimensional, friction-driven, adiabatic expansions were described by coupling mass, momentum, and energy balances with the SAFT equation of state (Lele and Shine, 1992).

Figures 1 and 2 contrast predictions for thermodynamically and kinetically-controlled precipitation of picene from supercritical methylcyclohexane solutions during partial adiabatic expansion while flowing through cylindrical tubes. Figure 1 illustrates the effects of varying inlet temperature and pressure of methylcyclohexane solutions. The assumption of instantaneous precipitation (thermodynamic control) always leads to substantial accumulation of solid-phase picene. The kinetic calculation, however, suggests that to minimize deposits along fuel transfer lines it is advantageous to operate with low inlet temperatures and high inlet pressures. Figure 2 depicts the effects of varying inlet pressure and transfer line length. Longer transfer lines give rise to larger fuel expansions,

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which result in greater solute precipitation. Results which account for nucleation kinetics indicate that there exists a critical length for each inlet pressure above which deposit formation occurs suddenly and rapidly approaches the limit of instantaneous precipitation. Higher inlet pressures enable longer transfer lines before the burst of nucleation occurs.

The supercritical antisolvent (SAS) process has been used in several experimental studies for the precipitation of polymers with interesting morphologies (Yeo *et al.*, 1993) and biologically active protein particles (Winters *et al.*, 1996). However, the complex interaction of jet hydrodynamics, phase behavior, and mass transfer that governs the process has not been investigated theoretically. Theoretical work on fuel droplet combustion (Hsieh *et al.*, 1991; Jia and Gogos, 1993) was used as a basis for developing a model which predicts the evolution in time of a single droplet of an organic solvent exposed to an excess of miscible supercritical fluid. Mass transfer occurs in two directions: from the solvent droplet into the supercritical fluid and from the supercritical fluid into the droplet.

Figure 3A shows the effect of pressure on the initial interfacial molar flux for several isotherms. The temperatures and pressures shown are typical for SAS operation. It can be seen that the initial molar flux is always into the droplet. Each isotherm passes through a minimum, which corresponds to the pressure for maximum droplet swelling. The sensitivity to pressure is greater near the critical temperature of the supercritical fluid. Figure 3B shows the change in the droplet radius as a function of time. Droplet lifetimes are characterized by rapid diffusion of supercritical fluid into the droplet, followed by gradual shrinking as the organic solvent diffuses into the excess supercritical fluid. These results suggest that the model can be used to gain insight into the relationship between operating conditions (temperature, pressure, droplet size) and droplet behavior as a function of time.

We have developed an aerosol dynamics code that tracks the evolution of the particle size distribution during steady, one-dimensional expansion of a supercritical fluid containing dissolved solutes. The calculation takes into account fluid-phase nonideality, nucleation, condensation, and inter-particle coagulation. Figure 4 shows the evolution of the particle size distribution during a friction-driven adiabatic expansion. This is the first time that this type of calculation has been implemented to study particle formation in supercritical fluids.

#### REFERENCES

- Debenedetti, P. G. 1990. *AIChE J.* **36**: 1289
- Hsieh, K. C., Shuen, J. S., Yang, V. 1991. *Combust. Sci. Tech.* **76**: 111.
- Huang, S. H., Radosz, M. 1990. *Ind. Eng. Chem. Res.* **29**: 2284.
- Jia, H., Gogos, G. 1993. *Int. J. Heat Mass Transf.* **36**: 4419.
- Lele, A., Shine, A. D. 1992. *AIChE J.* **38**: 742.
- Winters, M. A., Knutson, B. L., Debenedetti, P. G., Sparks, H. G., Przybycien, T. M., Stevenson, C. L., Prestrelski, S. J. 1996. *J. Pharm. Sci.* **85**: 586.
- Yeo, S. D., Debenedetti, P. G., Radosz, M., Schmidt, H. W. 1993. *Macromolecules.* **26**: 6207.

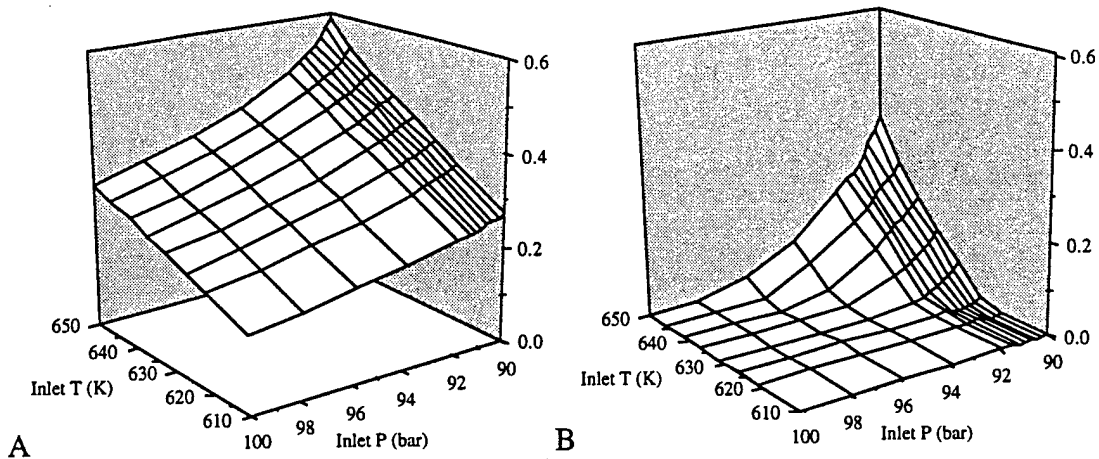


Figure 1: (A) Instantaneous and (B) kinetically-constrained precipitation of picene from adiabatically-expanding supercritical methylcyclohexane flowing through a cylindrical tube [length ( $L$ ) = 0.40 m, diameter ( $D$ ) = 2.0 mm]. The vertical axis gives the cumulative amount precipitated as a fraction of the total inlet flow. The assumption of instantaneous precipitation always leads to a substantial accumulation of solid-phase picene. The more realistic, kinetically-controlled calculations show that it is possible to suppress precipitation by operating at low enough inlet temperatures and high enough inlet pressures.

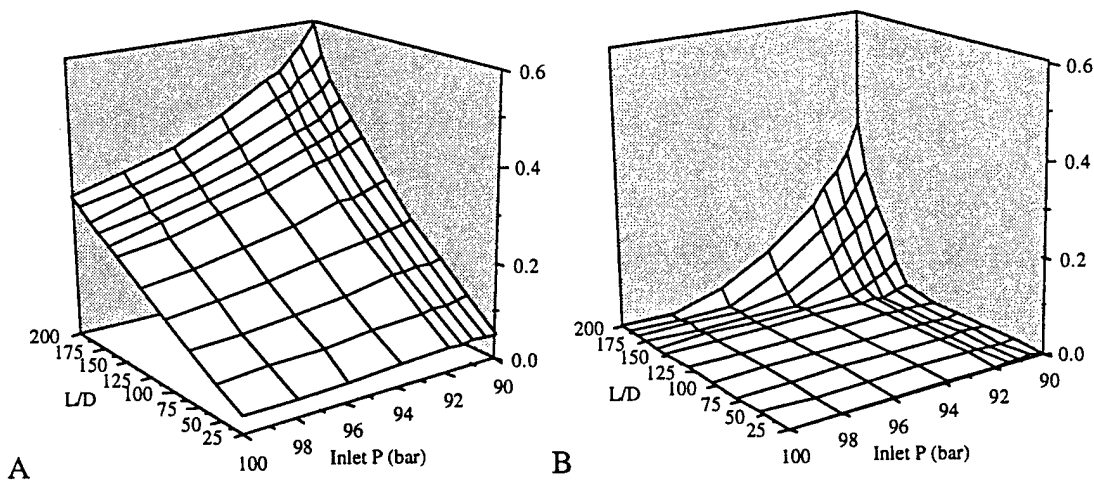


Figure 2: (A) Instantaneous and (B) kinetically-constrained precipitation of picene from adiabatically-expanding supercritical methylcyclohexane flowing through a cylindrical tubing [length ( $L$ ) = 0.05 to 0.40 m, diameter ( $D$ ) = 2.0 mm]. Same vertical scale as in Figure 1. Inlet temperature maintained at 650 K. There is a critical length for each inlet pressure at which accumulation rapidly increases, approaching the limit of instantaneous precipitation.

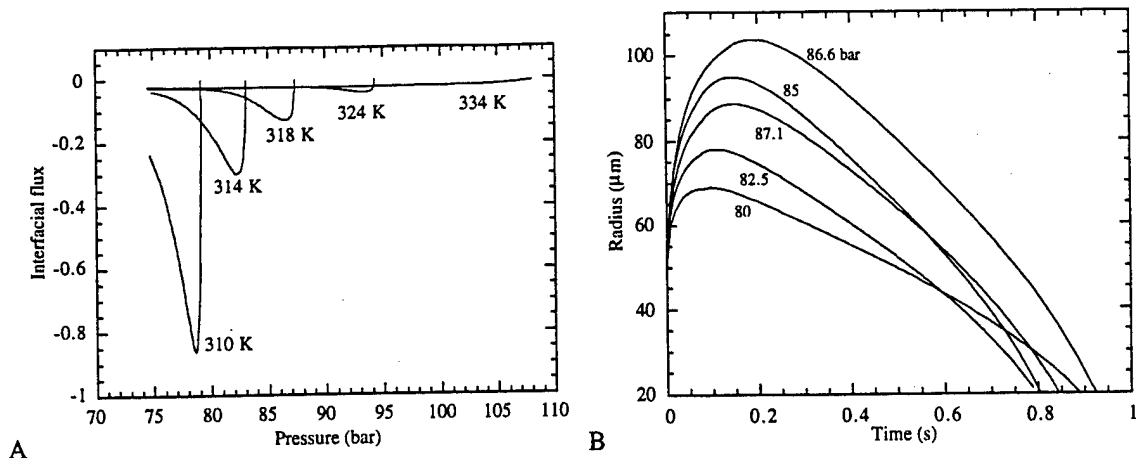


Figure 3: (A) Effect of temperature and pressure on the initial interfacial molar flux for toluene droplets exposed to an excess of supercritical carbon dioxide. Negative values indicate net flux into droplet. The molar flux has been scaled with  $[4000 \cdot \rho^* \cdot D^* / r_0]$  where  $\rho^*$  and  $D^*$  are, respectively, the saturated molar density and diffusivity of the liquid phase, and  $r_0$  is the initial droplet radius. (B) Evolution of toluene droplet radius at 318 K and different carbon dioxide pressures.

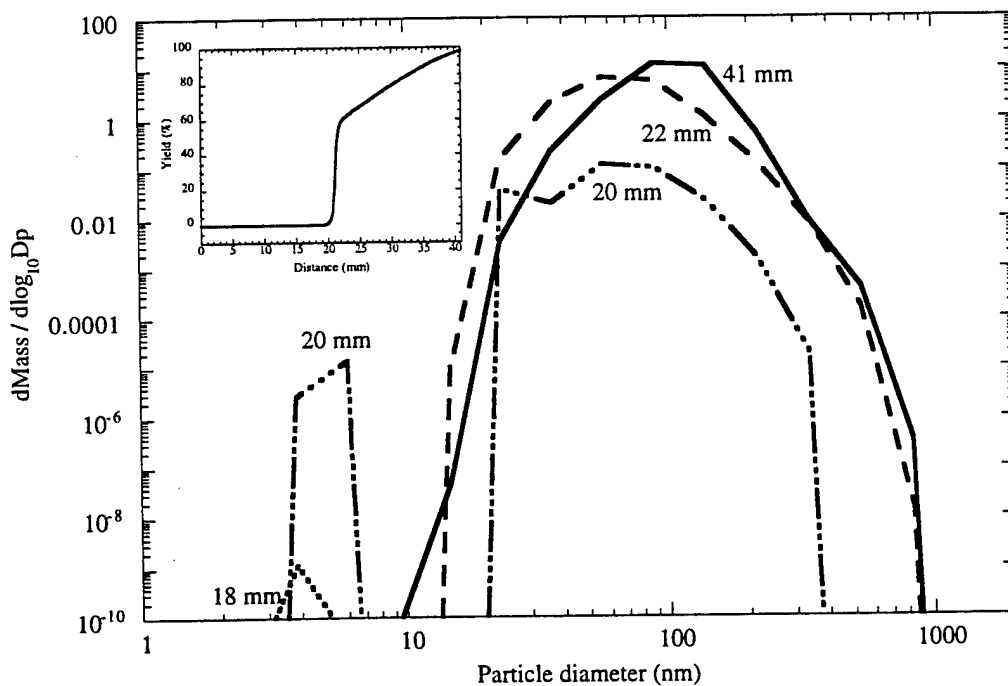


Figure 4: Evolution of the particle size distribution for phenanthrene precipitating from supercritical carbon dioxide expanding in a (50 micron x 41 mm) capillary. Inlet conditions: 200 bar and 325K. The inset shows the total precipitated mass, expressed as a percentage of the total amount of solute present initially, along the capillary's axis.

### 3. PERSONNEL SUPPORTED

Michael Winters; graduate student

Pablo G. Debenedetti; principal investigator

### 4. PUBLICATIONS

M.A. Winters, D.Z. Frankel, P.G. Debenedetti, J. Carey, M. Devaney, and T.M. Przybycien, "Protein Purification with Vapor-Phase Carbon Dioxide." **Biotech. & Bioeng.**, in press, (1998).

### 5. PRESENTATIONS

"Materials Processing with Supercritical Fluids." Engineering Foundation Conference on Separation Technology: Separations for Clean Production. Davos, Switzerland, October 28, 1997.

"Phase Separation by Nucleation and Growth, and by Spinodal Decomposition: Fundamentals." II NATO Advanced Study Institute on Supercritical Fluids. Kemer, Turkey, July 20, 1998.

### 6. HONORS/AWARDS

Presidential Young Investigator, NSF, 1987

Camille and Henry Dreyfus Teacher-Scholar, Dreyfus Foundation, 1989

John Simon Guggenheim Memorial Foundation Fellowship, 1991

Best Professional Book in Chemistry to *Metastable Liquids*, Assoc.Amer.Pubs., 1996

Professional Progress Award, American Institute of Chemical Engineers, 1997

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PI DATA

Name (Last, First, MI) Debenedetti, Pablo G.  
Institution Princeton University  
Contract/Grant No F49620 - 96 - 1 - 0169

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Name of Journal, Book, etc: Biotechnology and Bioengineering

Title of Article Protein Purification with Vapor-Phase Carbon Dioxide

Winters, M.A.; Frankel, D.Z.; Debenedetti, P.G.; Carey, J.;

Author(s) Devaney, M.; Przybycien, T.M.

Publisher (if applicable) \_\_\_\_\_

Volume: N/A Page(s): \_\_\_\_\_ Month Published: \_\_\_\_\_ Year Published: 1998

Name of Journal, Book, etc: \_\_\_\_\_

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Do Not Include: Honors and awards unrelated to the scientific field covered by the contract/grant.

Honor/Awards: Professional Progress Award Year Received: 1997

Honor/Award Recipient(s): Pablo G. Debenedetti

Awarding Organization: American Institute of Chemical Engineers

Honor/Awards: Best Professional/Scholarly Book in Chemistry "Metastable Liquids" Year Received: 1996

Honor/Award Recipient(s): Princeton University Press

Awarding Organization: Association of American Publishers, Inc.

Honor/Awards: Guggenheim Fellow Year Received: 1991

Honor/Award Recipient(s): Pablo G. Debenedetti

Awarding Organization: John Simon Guggenheim Memorial Foundation

Honor/Awards: Camille & Henry Dreyfus Teacher-Scholar Year Received: 1989

Honor/Award Recipient(s): Pablo G. Debenedetti

Awarding Organization: Camille & Henry Dreyfus Foundation

Honor/Awards: Presidential Young Investigator Year Received: 1987

Honor/Award Recipient(s): Pablo G. Debenedetti

Awarding Organization: National Science Foundation