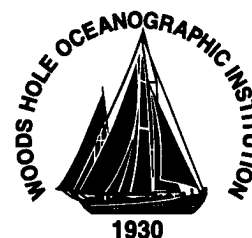


**Massachusetts Institute of Technology
Woods Hole Oceanographic Institution**



**Joint Program
in Oceanography/
Applied Ocean Science
and Engineering**



DOCTORAL DISSERTATION

*Plume-Lithosphere Interaction: Geochemical Evidence
from Upper Mantle and Lower Crustal Xenoliths from
the Kerguelen Islands*

by

Deborah Renee Hassler

June 1999

MIT/WHOI

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Lower Crustal Xenoliths from the Kerguelen Islands**

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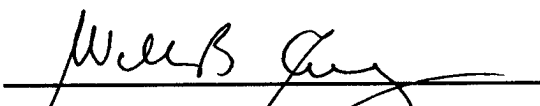
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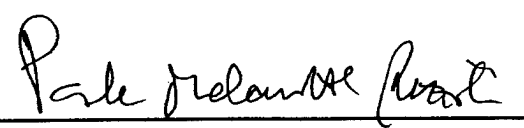
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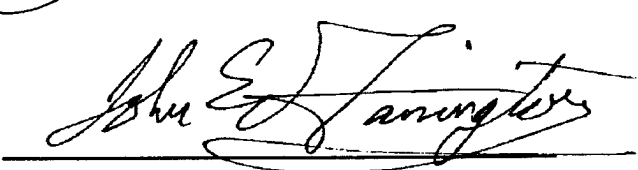
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by

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B.S. University of Kansas, 1989

Submitted to the Joint Program in Oceanography
in partial fulfillment of the requirements for the degree of

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at the

MASSACHUSETTS INSTITUTE OF TECHNOLOGY

and the

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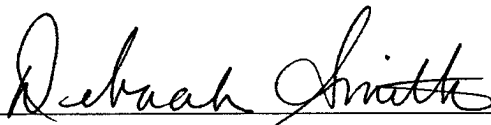
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Plume-Lithosphere Interaction: Geochemical Evidence from Upper Mantle and Lower Crustal Xenoliths from the Kerguelen Islands

by
Deborah Renee Hassler

Submitted in partial fulfillment of the requirements for the degree of
Doctor of Philosophy at the Massachusetts Institute of Technology and the Woods Hole
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Abstract

This study is a geochemical investigation of the evolution of the Kerguelen plume, on the basis of upper mantle and lower crustal xenoliths. Ultramafic xenoliths include harzburgites (predominant), a lherzolite, dunites and pyroxenites, whereas lower crustal xenoliths are cumulate gabbros recrystallized under granulite facies conditions. On the basis of the whole rock major element characteristics and trace element abundance patterns in clinopyroxenes, the harzburgites were found to be residues of extensive melting at high pressures within the Kerguelen plume. These were then recrystallized at low pressures and metasomatized by plume generated melts. Details of the metasomatic process were determined from trace element variations in clinopyroxene in connection to texture. This demonstrated that melt-rock reaction and the precipitation of new clinopyroxenes occurred by metasomatic carbonatitic melts.

It was also found that some of the harzburgites had distinctly unradiogenic Os isotopic compositions and were identified as originating from the sub-Gondwanaland lithosphere. On the basis of major and trace element compositions, the granulite xenoliths were found to be originally gabbroic cumulates formed from plume-derived basaltic melts emplaced at the base of the crust by underplating and subsequently recrystallized isobarically under granulite conditions. The Sr, Nd and Os isotopic compositions of the peridotite and granulite xenoliths demonstrate that the Kerguelen plume is isotopically heterogeneous and displays a temporal progression toward more enriched Sr and Nd isotopic compositions from the Ninetyeast Ridge to granulite xenoliths to Kerguelen basalts and Heard Island basalts.

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Dedication

I dedicate this thesis to my mother Barbara Lyn Heath Hassler,
and to the memory of my father, Carl Samuel Hassler.

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Chapter 1 Introduction

Mantle plumes are a significant means of heat and mass transfer from the earth's deep interior and result in hotspot volcanism on the earth's surface. Since the recognition of their importance in understanding plate tectonics (e.g. Wilson, 1963; Morgan, 1972), geochemical studies of plume-derived ocean island basalts (OIB) have established important insights into mechanisms for creating geochemical heterogeneities in the mantle via recycling (e.g. White and Hofmann, 1982; Allègre, 1982; Zindler and Hart, 1986) and spatial and temporal scales of mantle convection (e.g. Stein and Hofmann, 1994; Hofmann, 1997 and many others). Radiogenic isotopes in OIB reflect time-integrated parent/daughter elemental ratios and characterize the geochemical history of the plume mantle source. On the basis of the isotopic compositions of Sr, Nd and Pb in oceanic basalts, Hart and colleagues identified four "endmember" mantle components and showed that individual plume systems (as manifested by individual islands or island chains) form quasi-linear mixing arrays involving some of the components (e.g. Zindler and Hart, 1986; Hart and Zindler, 1989; Hart et al., 1992). That the asthenospheric mantle is not implicated as a common mixing endmember for the plume mixing arrays led Hart et al. (1992) to postulate that entrainment of the lower mantle could be an important process in determining the isotopic variations of OIB.

As a plume rises to the base of the oceanic lithosphere and plume-derived melts migrate through the lithosphere, interactions between the two could occur and modify the geochemical and isotopic compositions of both the melts and the lithosphere. The precise nature and extent of the interaction can vary on spatial and temporal scales, depending on physical and chemical conditions.

In the case of plume-derived melt interacting with lithospheric peridotites, the isotopic compositions of incompatible trace elements such as Sr, Nd and Pb of the melt would not generally be modified significantly. This is because of the low mineral-melt partition coefficients for these elements and the contrast in their concentrations between melt and mantle (i.e. high in the melt relative to the mantle). In contrast, the isotopic compositions of compatible elements such as Os in a plume-derived melt could easily and significantly be modified even in interactions that occur over a short length scale (for theoretical discussions, see Hauri, 1997). It follows then that isotopic heterogeneities for Sr, Nd and Pb observed in coeval basalts of a given plume system can be attributed to time-integrated heterogeneities with respect to parent/daughter elemental ratios of the plume, but that the Os isotopic composition could represent that of the lithosphere. When a plume is considered to have been positioned beneath a mid-ocean ridge at some time in its evolutionary past, the interpretations of isotopic heterogeneities in plume-related basalts tend to focus on the mixing of plume and MORB source components. Isotopic heterogeneities that may be "indigenous" to the plume itself may thus be overlooked. It is well established that a plume can affect the geochemical and isotopic compositions of nearby ridge melts because the plume mantle flows toward ridges as a result of thermal and rheological gradients (Schilling, 1991; Small, 1995; Ribe, 1996; Yale and Phipps Morgan, 1998). However, when one invokes the effects of the MORB source on plume-related melts, some physical justification is needed, especially for a distal plume-ridge interaction.

A different aspect of the plume-lithosphere interaction can be studied with lithospheric xenoliths that are brought to the surface by plume-derived magmas. Plume-derived melts react with the lithosphere during melt migration, and modify the geochemistry and mineralogy of the lithospheric peridotites. The mechanisms for this process can be diverse: 1.) the mineralogy (the minerals and their modal abundances) remains unchanged, but the

mineral chemistry is modified through elemental change at the interface, and through subsequent diffusion; 2.) new mineral phase(s) crystallize directly from the infiltrating melt; or 3.) the mineralogy and chemistry of the peridotite are modified through dissolution of pre-existing minerals and the precipitation of new ones. These processes represent what is generally termed mantle metasomatism and have been a subject of study for many years. A book was dedicated to the topic (Menzies and Hawkesworth, 1987). A general terminology of "cryptic" and "modal" metasomatism (e.g. Dawson and Kornprobst, 1984; Erlank et al., 1987; Harte et al., 1983) is inadequate in cases where a newly-precipitated mineral is one of the pre-existing species (for instance, the precipitation of diopside in lherzolite). The petrographic appearance of a rock that has been cryptically metasomatized may be the same as a rock that has not been metasomatized at all. If in fact new clinopyroxene was added to the rock, microbeam-based observations could play crucial roles in discerning this.

The understanding of mantle metasomatism has improved considerably over the past decade through theoretical work and by the modeling of trace element variations in rocks and minerals (Navon and Stolper, 1987; Bodinier et al., 1990; Takazawa et al., 1992; Sen et al., 1993; Bedini et al., 1997; Yang et al., 1998b). It is commonly observed that melt-peridotite interaction modifies trace element abundance patterns of minerals (notably clinopyroxene, an important repository of many trace elements in peridotitic systems) to various extents in a given lithosphere-melt system. For instance, Takazawa et al. (1992) and Yang et al. (1998b) demonstrated that smoothly light REE-depleted original patterns in clinopyroxenes were gradually modified toward light REE-enriched "equilibrium" patterns, with sharply inflected (at Nd) or "spoon-shaped" patterns as intermediate stages of interaction. As discussed by Navon and Stolper (1987) and others, modification of REE patterns in peridotitic clinopyroxenes proceeds in a chromatographic manner, beginning with an increase in more incompatible LREE and moving towards less incompatible HREE.

There are many instances where rocks (e.g. a suite of peridotite xenoliths from a single locality) display a wide range of "degrees" of metasomatism, indicating large local-scale variations in melt-rock ratios and perhaps short durations of interaction. Because of low clinopyroxene-melt partition coefficients, Sr and Nd abundances in highly depleted peridotite clinopyroxenes are susceptible to even incipient metasomatism. Hence, the original Sr and Nd isotopic compositions of the clinopyroxene can easily be erased.

1.1 The Kerguelen plume and the scope of the thesis

The Kerguelen plume is unique among mantle plumes because it is known to be especially long-lived. It became active at least 115 Ma ago, and possibly as long as 130 Ma ago—coincident with the breakup of Gondwanaland (Whitechurch et al., 1992; Storey et al., 1989). This plume is responsible for enormous amounts of volcanism in the eastern Indian Ocean, including the Ninetyeast Ridge, Broken Ridge, and the Kerguelen Plateau, the second largest of the oceanic plateaus after the Ontong Java Plateau (Coffin and Eldholm, 1994). The Kerguelen plume may be a carrier of the DUPAL isotopic anomalies that are characteristic of the basalts that are erupted in the southern hemisphere (Hart, 1984), and the interaction of this plume with the Indian Ocean lithosphere has potentially affected the geochemistry of the entire southeastern Indian Ocean (Davies et al., 1989; Storey et al., 1989; Mahoney et al., 1989).

The Kerguelen Plateau, located in the southern Indian Ocean (Figure 1–1), is one of the large igneous provinces described by Coffin and Eldholm (Coffin and Eldholm, 1994). It is the second largest of the oceanic plateaus, covering an area greater than $2 \times 10^6 \text{ km}^2$, and extending over 2000 km in a northwest-southeast direction. The plateau is divided into two structural units on the basis of their separate tectonic and volcanic histories at

approximately 54°S (Houtz et al., 1977; Coffin et al., 1986). In the northern province, the plateau is 4 km above the surrounding ocean floor, is submerged under less than 1 km of water. It decreases to 2 km of relief and 2.5 km water depth southward. Several Ocean Drilling Program (ODP) drill-holes (legs 119, 120 and most recently 183; Barron et al., 1991; Wise et al., 1992) and dredges recovered tholeiitic to transitional basement basalts from the Plateau (Leclaire et al., 1987). The Kerguelen Plateau was emplaced in a subaerial to shallow water environment (Coffin, 1992; Schlich, 1982), but little is known of the rocks that make up its volume.

Plate tectonic reconstructions (e.g. McKenzie and Sclater, 1971; Royer and Sandwell, 1989) indicate the Bunbury basalts and Rajmahal Traps, the Naturaliste Plateau and the southern Kerguelen Plateau all formed during the initial phase of magmatism attributed to the Kerguelen plume, and may have been coincident with the breakup of Gondwanaland (Royer and Sandwell, 1989). From 115 Ma to 43 Ma, the northern Kerguelen Plateau, Broken Ridge and the Ninetyeast Ridge formed while sea-floor spreading between India and Antarctica-Australia began and India moved northward. At approximately 43 Ma, Broken Ridge rifted from the Kerguelen Plateau at the southeast Indian Ridge. Since as long as 43 Ma until the present, the northern Kerguelen Plateau and the Kerguelen Islands formed. The most recent known activity of the plume was 40 ka at Ross Volcano, in the southern Kerguelen Islands (Nougier et al., 1983).

Hypotheses for the origin of the Kerguelen Plateau have included a continental fragment, excessive mid-ocean ridge magmatism, an uplifted block of oceanic crust, or a hotspot (cf. Recq and Charvis, 1986). The most controversial of these has been the suggestion that the plateau is a continental fragment left behind after the breakup of Gondwanaland. Geological and geochemical studies (e.g. Watkins et al., 1974; Dosso et al., 1979; Gautier et al., 1990; Yang et al., 1998a) found no evidence of continental crust

in rocks from the Kerguelen Islands, and seismic studies in the northern Kerguelen Plateau indicated a thickened oceanic crust (approximately 17 km ; Recq and Charvis, 1986). However, basalts from the southern Kerguelen Plateau exhibit geochemical characteristics of continental affinity (Alibert, 1991; Mahoney et al., 1994), and this part of the plateau has a seismic structure that is similar to that of a volcanic passive margin with a thinned continental crust overlain by basaltic lava flows (Recq and Charvis, 1986; Operto and Charvis, 1995). Moreover, Coffin *et al.* (1994) argued that the southern plateau is a mixture of structural elements, including crustal uplifts and continental fragments.

The Kerguelen Islands, located on the northern Kerguelen Plateau, are an archipelago consisting of 400 islands across an area of 4000 km² (Watkins et al., 1974). More than 80% of the exposed rocks are transitional to highly alkaline plateau basalts (Giret and Lameyre, 1983; Gautier et al., 1990). The remaining 20% are intrusive rocks: a tholeiitic to transitional series, and two alkaline differentiation series, including gabbro-syenite-alkaline granite, and gabbro-nepheline syenite. The rocks of this study are from two (Lac Supérieur and Mt. Trapeze) of four xenolith localities on the Courbet Peninsula (Figure 1-1) that were discovered during the 1992-93 austral field season. The xenoliths are included in relatively young alkaline dikes that cross-cut the 24.5 Ma plateau basalts (Nicolaysen et al., 1999) that make up the Courbet Peninsula. The Lac Supérieur xenolith suite contains harzburgites, rare lherzolites, dunites, pyroxenites and mafic granulites, but the Mt. Trapeze locality has only harzburgitic peridotite xenoliths.

The main purpose of this study is to investigate the geochemical evolution and interaction between the Kerguelen plume and the lithospheric mantle. As mentioned above, the Kerguelen plume is unique in that it has encountered various geodynamic conditions, from continental breakup to oceanic rift, and to the present-day intra-plate conditions. This

thesis attempts to decipher how the diverse geodynamic environments have affected the geochemical evolution of the plume.

An approach taken here is an integrated petrological and geochemical study of lower crustal and upper mantle xenoliths brought to the surface by Kerguelen plume magmas. As discussed above, geochemical signatures of plume-lithosphere interaction processes were varied, and details of the interaction can be investigated based on characteristic trace element signatures of the mechanisms and the types of melt involved. Textures of rocks under which melt-rock interaction is recognized are also important in distinguishing between newly-formed metasomatic minerals and those that were present when the interaction occurred.

Given that the Kerguelen plume has encountered diverse geodynamic environments, the upper mantle xenoliths in this study can be any one of the following: 1.) pieces of young oceanic lithosphere upon which the Kerguelen plume impinged; 2.) pieces of the plume mantle itself, either as unmelted source mantle or as residues of melting; 3.) pieces of lithosphere that delaminated from the Gondwanaland subcontinental lithosphere when the Kerguelen plume began its activity. All of these varieties could have pristine geochemistry or be in modified forms due to interaction with plume-derived melts.

In their pristine form, pieces of the oceanic lithosphere can be depleted harzburgites with clinopyroxene trace element characteristics like those in abyssal peridotites (e.g. Johnson et al., 1990) and with Sr and Nd isotopic compositions similar to that of the Indian Ocean MORB. However, these peridotites are extremely susceptible to interaction with plume-derived melts with respect to incompatible trace elements, and Sr and Nd isotopes, and the chances for sampling pristine oceanic lithosphere could be slim.

In the case of plume mantle residues, the end products could be diverse. If melting proceeds beyond the garnet-out or the clinopyroxene-out phase boundaries due to high temperatures, the plume residue could also be highly depleted harzburgite. Since the melting of mantle plumes begin in the presence of garnet, garnet-bearing residues could be produced. However, they may well be recrystallized to a lower pressure assemblage during the adiabatic upwelling. In plume melting at pressures greater than 3 GPa, the clinopyroxenes are subcalcic. Upon recrystallization at lower pressures, high-Ca clinopyroxenes are formed, and the direct application of high-Ca pyroxene melt partition coefficients in such a case can not be made to produce meaningful results. However, these rocks could still contain high-Ca clinopyroxenes upon recrystallization due to the relative large CaO contents of high-temperature orthopyroxenes.

In their pristine form, these type of harzburgites may not be distinguishable from an unmetasomatized oceanic lithosphere, but Sr and Nd isotopic characteristics can be used to identify them. If these high pressure residual harzburgites were later metasomatized, they could be very similar geochemically to metasomatized oceanic mantle lithosphere. The Os isotopic compositions could be the only criterion by which their provenance can be distinguished, provided that Os isotopes in peridotites are insensitive to metasomatic reactions. The Os isotopic compositions would also reveal peridotites derived from sub-Gondwanaland lithosphere by their unradiogenic $^{187}\text{Os}/^{188}\text{Os}$. This is because subcontinental lithospheric roots are known to possess unradiogenic Os compositions due to ancient melt extraction (Re depletion) events (e.g. Walker et al., 1989; Pearson et al., 1995; Carlson and Irving, 1994).

From the point of view of mantle metasomatism, the peridotite xenolith suites of this study provide opportunities for carrying out detailed geochemical studies using in-situ trace element compositions and isotopic data obtained for minerals and whole-rocks. Diversity in

plume-derived melt compositions in terms of their trace elements and isotopes would be imprinted in peridotite xenoliths during melt-lithosphere interaction, and would be revealed.

The lower crustal xenoliths studied here are granulite facies rocks with gabbroic precursors. As noted before, the occurrence of granulite facies xenoliths in OIB was documented for the first time in the Kerguelen samples (Grègoire et al., 1994) including the first occurrence of sapphirine from the oceanic environment. However, their provenance (continental v. oceanic) must be carefully evaluated because of the complex evolutionary history of the Kerguelen plume. This history involves the possibility that delaminated lower crust from Gondwanaland may be present within the Kerguelen Plateau. If the granulite xenoliths are identified as part of the Kerguelen plume magmatism, they also provide an important opportunity for obtaining geochemical data on a part of the magmatism that may not have erupted as basalts on the surface. The origin of the granulites is also connected to the geodynamic evolution of the Kerguelen plume. If their precursors were to be identified as typical ocean-ridge gabbros, then they must have been incorporated into the Kerguelen plume system when the plume was ridge-centered. However, the recrystallization under granulite facies conditions would require significant thickening of crust (Grègoire et al., 1998), which may not be plausible. Alternatively, if their provenance was found to be the Kerguelen plume, the gabbroic precursors could represent pieces of lower crust formed by underplating, providing material evidence in support of geophysical inference based on thickened crust for many oceanic islands (e.g. ten Brink and Brocher, 1987; Caress et al., 1995).

The Kerguelen Islands provide an ideal setting to examine plume interaction with the lithosphere of different tectonic environments. This research presented an opportunity to use a diverse population of xenoliths to attempt to unravel the complexities of materials and processes associated with a mantle plume that had a varied tectonic history.

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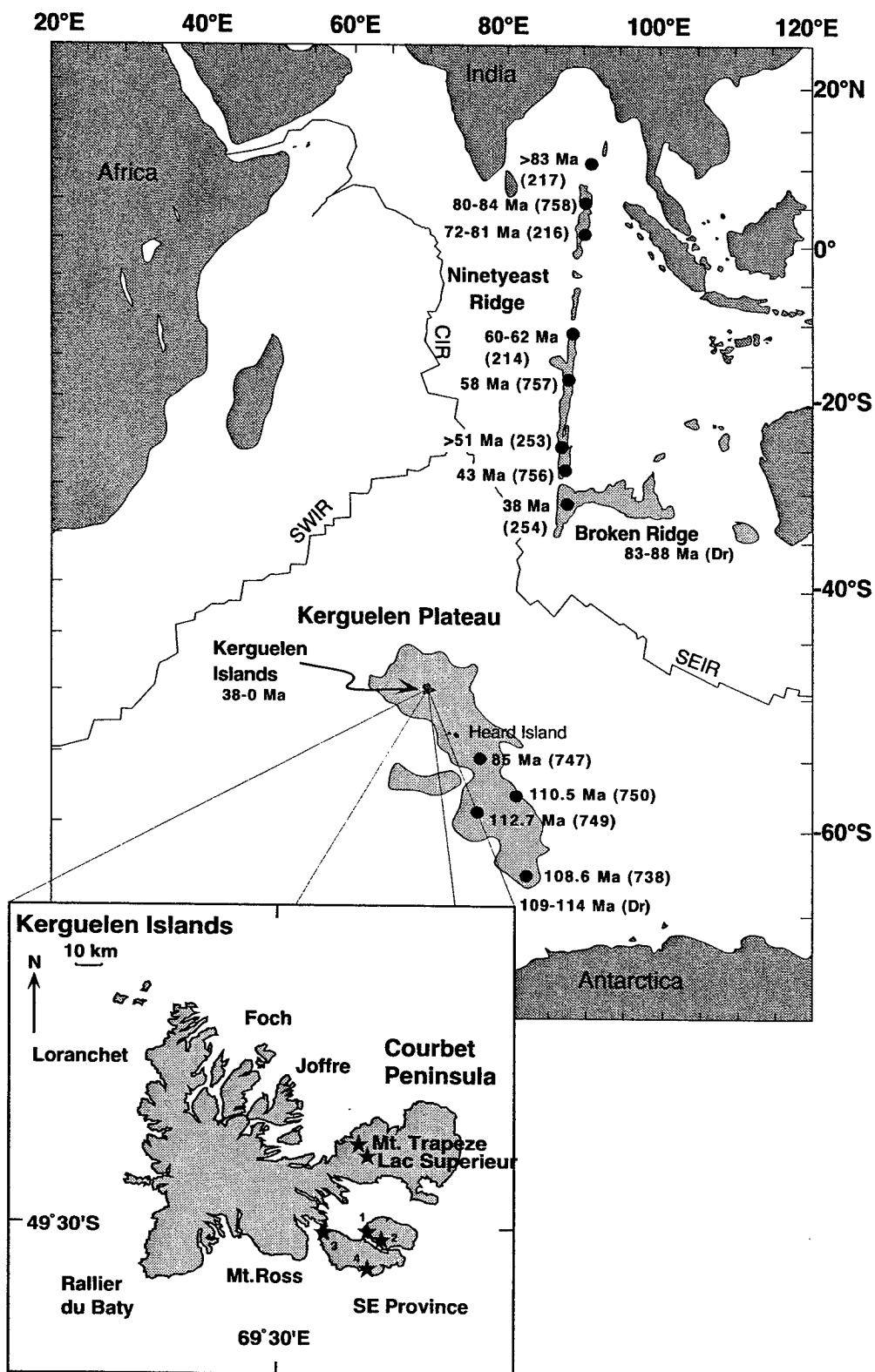
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1.3 Figures

Figure 1–1. Map of the Indian Ocean (after Coffin, 1992). Submarine ridges and plateaus related to the Kerguelen plume are highlighted in light gray shading. Major plate boundaries are labeled (SWIR, SEIR, CIR). Filled circles mark ODP and DSDP drill sites. The ages or range of ages (Duncan, 1978; Duncan, 1991; Duncan and Storey, 1992; Whitechurch et al., 1992; Storey et al., 1996; Pringle et al., 1997) of recovered basaltic basement or oldest fossils (marked with “greater than” symbol) are adjacent to the symbols. Ages of dredged samples are marked with the suffix (Dr). The Kerguelen Islands (inset) are located at the northern end of the Kerguelen Plateau and range from 39-0 Ma (Giret and Lameyre, 1983; Nougier, 1972; Nicolaysen et al., 1999). Xenolith localities are shown as filled stars. The samples from this study are from the Lac Superieur site on the Courbet Peninsula. Southeast Province localities are labeled 1: Pointe de l’Espérance, 2: Val Phonolite, 3: Dome Rouge, and 4: Mont Tizard.



Chapter 2 The Geochemistry of Peridotite Xenoliths from the Courbet Peninsula, Northern Kerguelen Islands

2.1 Introduction

Peridotite xenoliths brought to the earth's surface by erupting magmas are a window through which the lithospheric mantle can be studied directly and in detail. Indeed, much of the present knowledge about the materials that make up the continental lithosphere has been derived from studies of kimberlite-borne peridotite xenoliths. For instance, Boyd (1973) determined the pressure and temperature conditions that were frozen in peridotite minerals and established that suites of peridotite xenoliths from the Kaapvaal craton defined geothermal gradients at the time of kimberlite eruption. This was the first quantitative information extracted from peridotite xenoliths and much progress has been made since in the understanding of evolutionary pathways of continental lithospheres. Applications of the Re-Os isotopic system to the sub-cratonic peridotites have confirmed their antiquity and their long term stabilization (e.g. Walker et al., 1989; Carlson and Irving, 1994; Pearson et al., 1995).

Peridotite xenoliths have also provided direct access to geochemical processes involved in the formation and evolution of the mantle lithosphere. Studies of mantle metasomatism (see for instance articles in *Mantle Metasomatism* (Menzies and Hawkesworth, 1987 and references therein) and the recognition that peridotites from the Kaapvaal and Siberian cratons are enriched in orthopyroxene to the extent that they cannot be simple residues of melt extraction at any pressure (e.g. Boyd, 1989; Kelemen et al., 1998) further stimulated the field in the direction of investigating interactions between

lithospheric peridotites and infiltrating melts and fluids. As discussed in Chapter 1, peridotites can be a sensitive substrate upon which records of melt rock-reactions are imprinted. The combination of trace element abundances and Sr, Nd, and Os isotopic compositions is a useful tool to decipher processes responsible for cratonic mantle formation and evolution. In contrast to peridotite xenoliths from continental localities, where the major objective of study is the continental lithosphere itself, those from ocean islands basalts (OIB) have not attracted as much attention. As the source of OIB is getting better characterized however, geochemical studies of OIB peridotite xenoliths could help to understand the source mantle, melting process and perhaps more importantly, interaction between a plume and the lithosphere. For instance, Sen et al. (1993), Hauri et al. (1993), Hauri and Hart (1994) and Yang et al. (1998b) used trace element abundance patterns in clinopyroxenes from peridotite xenoliths from Hawaii, the Austral Islands, and Samoa to demonstrate that plume mantle melts more extensively than ridges, and noted complex plume-lithosphere interactions involving basaltic and carbonatite melts. Moreover, Schiano et al. (1994) discovered that carbonate-rich melts are present as melt inclusions in oceanic peridotites, including within xenoliths from the Kerguelen Islands.

Peridotites constitute a significant member of the Kerguelen Islands xenolith suites; all known xenolith localities thus far produce significant proportions of peridotites. As discussed in Chapter 1, the Kerguelen plume has evolved under different geodynamic environments and has possibly interacted with different types of lithosphere. If pieces of the different types of mantle lithosphere, such as oceanic mantle lithosphere, continental mantle lithosphere (related to Gondwanaland), or recrystallized plume mantle, were to be present as xenoliths, they would present a complex geochemical spectrum. This is especially true since geochemical and isotopic modification of original geochemical signatures by Kerguelen plume-derived melts would be expected during plume-lithosphere

interaction. Conversely, if any of peridotite xenoliths could be identified as pieces of specific lithospheric types, the dynamics of plume-lithosphere interaction could be constrained.

The major purpose of this study is to present a detailed description of two suites of peridotite xenoliths from the Courbet Peninsula (Lac Superieur and Mont Trapeze), to constrain their provenance, and to examine the geochemical aspects of plume-lithosphere interaction as recorded by the geochemistry of these xenoliths. This study expands on the previous work of Grègoire (1994) on petrography and major and trace element compositions, and on the isotopic studies of Mattielli (1996), and Mattielli et al. (1996). These studies were made on peridotites from the Southeast Province of the Archipelago, whereas samples for the present study are from the Courbet Peninsula, to the northeast (Figure 1-1).

2.2 Sample description

Thirty ultramafic xenoliths from the Lac Superieur locality (samples OB93-51 to OB93-102) and twenty-five from Mt. Trapeze (samples OB93-280 to OB93-318) will be described in this chapter. The Lac Superieur suite comprises mostly spinel harzburgites, but also includes one spinel lherzolite, several wehrlitic dunites and pyroxenites. The Mt. Trapeze suite contains only spinel harzburgites. In situ trace element analyses were made on clinopyroxenes from all of these xenoliths. Representative samples of the various rock types were chosen for further study that included whole rock geochemistry, major element compositions of minerals, and isotopic compositions (Sr, Nd, and Os). As discussed below, the peridotites fall into distinct geochemical groups defined by their petrography and trace element geochemistry.

The xenoliths from Lac Supérieur are generally larger (>6-7 cm, often >10 cm in the largest dimension) than those from Mt. Trapeze (4-5 cm diameter or less). All of the harzburgites comprise olivine, orthopyroxene, less than 5% clinopyroxene and chromite. Two types of harzburgites are present in both of the localities and are grouped on the basis of their clinopyroxene texture. In one type (the REE depleted group, defined below), clinopyroxenes occur predominantly in symplectite intergrowths with chromite and discrete (or individual) clinopyroxene grains are rarely present. In the other type (found in all of the other harzburgite REE groups), clinopyroxene and spinel occur as two textural populations, of which the majority are discrete grains, but which also occur together in symplectite intergrowths. The clinopyroxene and spinel grain sizes are variable; those in symplectites are 1 mm to 3 mm, but discrete grains are as large as 5 mm. Olivine and orthopyroxene can be as large as 1 cm, and often are strained as exhibited by deformation bands and subgrain boundaries. Orthopyroxene occurs in poikilitic texture in one of the samples (OB93-306). Exsolution of clinopyroxene from orthopyroxene is common in all of the harzburgites. Exsolution of orthopyroxene and spinel from clinopyroxene is present in some of the harzburgites. Symplectite textures vary from vermicular spinel contained entirely within ~1-2 mm clinopyroxene grains, to coarsened intergrowths (>2-3 mm). Some of the Mt. Trapeze harzburgites have discrete spinels with an octahedral habit, in addition to the chromites with symplectite texture (e.g. OB93-287, OB93-291, OB93-297, OB93-309). Some samples contain accessory phlogopite (OB93-83, OB93-299) and apatite (OB93-83), a relatively rare assemblage in rocks from oceanic environments. To date, only Hauri et al. (1993) have described apatite-bearing harzburgite xenoliths from an OIB in Samoa.

Of the remaining Lac Supérieur xenolith types, sample OB93-52 is a very rare piece of fertile spinel lherzolite that comprises olivine, orthopyroxene, chromite, and nearly 10%

clinopyroxene. The clinopyroxene in this rock occurs only in the discrete form and the spinel is euhedral. Wehrlitic dunites (OB93-61c, OB93-64, OB93-82), are finer-grained than the rest of the suite (<2-3mm), and comprise olivine, euhedral spinels and less than 10% discrete clinopyroxene. Pyroxenites are megacrystic (~8 cm) to coarse-grained (2-3 cm), and two of the samples (OB93-94, OB93-81b), contain phlogopite and octahedral spinel. Exsolution lamellae of orthopyroxene are up to 0.5mm wide in the phlogopite bearing pyroxenites. The pyroxenites (OB93-75, OB93-87, OB93-100, OB93-102), also have macroscopic exsolved orthopyroxene, but are lacking phlogopite and spinel and contain a small amount (<1%) of partly altered plagioclase.

Some typical petrographic features of these rocks are shown in Figure 2-1, and a list of the samples of this study in Table 2-1.

2.3 Analytical Methods

Surfaces were cut from the xenoliths and saw marks were polished off with 120 and 240 grit silicon-carbide paper. The polished pieces were crushed to smaller fragments between plastic with a hammer. Pieces that did not have any basalt veins or visible alteration were selected under a binocular microscope for grinding. All samples were powdered in an agate shatterbox vial except OB93-52, OB93-56, OB93-80, OB93-83, OB93-85, OB93-99, OB93-299, OB93-305, OB93-306, OB93-310, and OB93-317, which were ground in a tungsten carbide vial at the GeoAnalytical Laboratory at Washington State University (WSU). The samples powdered at WSU may have W, Co, Nb and Ta contamination from the grinding vessel. Whole rock major and trace element compositions were determined on the unleached powders by XRF and ICP-MS at the WSU. Precision for XRF major element analysis is < 0.4% (1 S.D.). The precision of all trace element analyses by ICP-

MS is 1-3%, except for U and Th, which are ~10% (1 S.D.). Detection limits for the trace elements by ICP-MS are 0.1 to 0.5 times chondritic levels.

Major element compositions of the rock-forming minerals in thin sections or mounts of representative mineral grains were determined with the JEOL 733 Superprobe electron microprobe at the Massachusetts Institute of Technology (MIT) with an acceleration potential of 15 kV, a beam current of 10nA and a 1 μ m spot. Raw data were reduced online using the method of Bence and Albee (1968) modified by Albee and Ray (1970). The precision based on counting statistics are 6-25% (1 S.D.) for <0.5 wt.%, 2.5-6% for 0.5-1 wt.%, 1-3% for 1-15 wt.%, <0.5% for >15 wt.%.

The Cameca IMS 3f ion probe at WHOI was used to analyze trace element and rare earth element contents of clinopyroxenes following the methods of Shimizu and Hart (1982). A primary beam of O⁻ ions was focused to ~20 μ m for REE (La, Ce, Nd, Sm, Eu, Dy, Er, Yb) and ~10 μ m for other trace elements (Cr, V, Y, Ti, Zr, Sr). Molecular interferences were eliminated by energy filtering and a secondary voltage offset of -30 to -60V for the REE and -90V for the other trace elements. Uncertainties on the basis of counting statistics for the REE (1 S.D.) were 20-40% for chondrite normalized REE abundances of less than 1, 5-10% for those between 1 and 10, and 1-5% for those greater than 10. Uncertainties (1 S.D.) for the other trace elements were as high as 25% for some of the ultradepleted samples (e.g. Ti <100 ppm), but otherwise <10%. All normalized REE data are reported relative to abundances in C1 chondrites (Anders and Grevesse, 1989).

Strontium and Nd isotopic compositions were determined on hand picked mineral separates of phlogopite and clinopyroxenes. Phlogopites were not leached. Clinopyroxenes were leached for 30 minutes in 100°C, 2.5 N HCl, followed by 5 minutes in cold HF,

followed by another 30 minutes of HCl, to remove minor surface alteration (Taras and Hart, 1987). The chemistry and mass spectrometry of Sr and Nd isotopes were done at WHOI, MIT and Okayama University. At WHOI, isotopes were measured using a VG-354 5-collector mass spectrometer. Cation exchange separation methods were used for Sr and the LREE, and HDEHP extraction was used for Sm and Nd, as described by Zindler (1980) and Richard et al. (1976). Strontium data are normalized to $^{87}\text{Sr}/^{88}\text{Sr} = 0.1194$, and isotopic compositions are reported relative to $^{87}\text{Sr}/^{86}\text{Sr} = 0.710240$ for NBS987 (average $^{87}\text{Sr}/^{86}\text{Sr} = 0.710266 \pm 18$, $n=3$). Neodymium data are normalized to $^{146}\text{Nd}/^{144}\text{Nd} = 0.7219$, and isotopic compositions are reported relative to $^{143}\text{Nd}/^{144}\text{Nd} = 0.511847$ for the La Jolla standard (average $^{143}\text{Nd}/^{144}\text{Nd} = 0.511855 \pm 8$, $n=3$). The procedural blank for Sr was 300 pg/g sample, and 150 pg/g sample for Nd. These were insignificant in relation to the amount of analyte, thus no blank corrections were made on these Sr or Nd isotopic ratios. Counting statistics (2σ) associated with mass spectrometry were <25 ppm for Sr and <10 ppm for Nd. All concentrations of Sr and Nd, were determined by isotope dilution, during isotopic analysis. Analyses of some samples with relatively low Sr concentrations in clinopyroxenes (OB93-94cpx, OB93-94phlog, OB93-64) were made at MIT, using the VG-Sector 54 mass spectrometer and methods optimized for low concentrations. The clinopyroxenes were acid-leached according to the same procedure. Separation of Sr was accomplished using Sr-specTM resin (Horwitz et al., 1991; Pin and Bassin, 1992). Strontium was analyzed using the dynamic multi-collector mode with Sr intensity at $^{88}\text{Sr} = 3\text{V}$. Analytical blanks for Sr were insignificant (4 pg total blank) and no correction was necessary. Strontium data are normalized to $^{86}\text{Sr}/^{88}\text{Sr} = 0.1194$, and replicate analyses of NBS987 at MIT during the times of these analyses yielded $^{87}\text{Sr}/^{86}\text{Sr}$ of 0.710240 ± 10 ($n=147$). Acid-leached clinopyroxene separates from samples with extremely low Sr and

Nd concentrations (samples OB93-78, OB93-80, OB93-64) were analyzed for Sr and Nd isotopes by E. Nakamura at Okayama University, Misasa, Japan. Neodymium isotope analysis of clinopyroxene from OB93-78 failed because only 0.13 ng of analyte was available. Strontium isotopic compositions were normalized to NBS987 $^{87}\text{Sr}/^{86}\text{Sr} = 0.710240$ and Nd isotopic compositions to La Jolla $^{143}\text{Nd}/^{144}\text{Nd} = 0.511847$. Total blanks at the Okayama lab were 45 pg for Sr and 10 pg for Nd. Concentrations of Sr and Nd were determined by isotope dilution on the same aliquots.

Osmium isotopes for the peridotite are discussed in detail in Chapter 3 and the analytical techniques are quoted here for reference (Hassler and Shimizu, 1998; note 10). "Olivine separates and peridotite whole-rock powders were ground in an agate shatterbox vial, prepared for Os analysis by NiS fire-assay and distillation, and measured using negative thermal ionization mass spectrometry techniques at Woods Hole Oceanographic Institution (Ravizza, 1993; Hauri and Hart, 1993). On the basis of counting statistics, the precision associated with $^{187}\text{Os}/^{188}\text{Os}$ range from 0.10 - 0.44% (2σ), although replicate analyses (one was a separate aliquot from the same powder split and another from a separate powder split) agree within $\pm 0.5\%$ and $\pm 0.7\%$ respectively. Osmium standard runs were reproducible within 0.3% (2σ). Osmium concentrations were determined by fire-assay and isotope dilution, and were reproducible at $\pm 7\%$ (2σ). Blank corrections for Os isotopic composition and concentrations were less than the uncertainty of counting statistics in most cases, and only applied when the correction was greater than the counting error. Rhenium concentrations were measured on separate powder splits by inductively coupled plasma-mass spectrometry at the Lamont-Doherty Earth Observatory. Reported Re concentrations are not blank-corrected (procedural blank was approximately 10-20 ppt),

and counting statistics were 0.6 - 3.2% (2σ). Thus, most of the samples had negligible amounts of Re."

2.4 Results and Discussion

2.4.1. Whole rock compositions

Major and trace element abundances were determined on whole rock powders from a subset of samples (Table 2-2). Often, difficulties in obtaining representative whole rock powders for peridotite xenoliths arise from their small size relative to their grain size. Assuming a minimum dimension of ten times the average grain size for a representative sample, a rock with an average grain size of 5 mm would require 125 cm^3 ($\sim 375 \text{ g}$, assuming average mineral density = 3 g/cm^3) to be representative. Because of the small size of xenoliths in this study, whole rock powders were prepared using a maximum of approximately 50 g of sample, and often much less (20-30 g). This is much less than the above requirement, and the representativeness of the bulk rock powder for major element compositions may therefore be suspect (cf. Boyd, 1989; Boyd et al., 1997).

In a general sense, the whole rock major element compositions of the harzburgites reflect their depleted character. For instance, CaO ranges from 0.51 to 1.2 wt.% and Al_2O_3 from 0.47 to 1.3 wt.%. These ranges are similar to harzburgite and lherzolite xenoliths from Samoa (Hauri and Hart, 1994) and Hawaii (Sen, 1988). The Samoan peridotites described by Hauri and Hart (1994) appear (on the basis of rock reconstruction from mineral modes and compositions) to be more depleted than the xenoliths of the present study with CaO and Al_2O_3 ranging from 0.21 to 0.88 wt.% and 0.41 to 1.11 wt.% respectively. The Hawaiian xenolith suites described by Sen (1988) contain lherzolites and harzburgites with similar olivine compositions to the present suites, and with CaO and

Al₂O₃ contents within the range reported here (e.g. 77SL-31 with Fo_{91.7} olivine has bulk CaO = 0.42 wt.% and Al₂O₃ = 1.84 wt.%, whereas 77KASP-4 with Fo_{90.9} olivine contains bulk CaO 2.28 wt.% and Al₂O₃ 2.29 wt.%). In the xenoliths of the present study, the Mg# [atomic Mg/(Mg+Fe)×100] for the harzburgites are greatest, ranging from 89.8 to 91.9, except for OB93-306 (Mg# = 84.3), followed by 88.8 for the lherzolite (OB93-52), and 87.9 and 88.4 for the dunites (OB93-64, OB93-82). No major oxide correlations are present other than slight negative correlations of CaO and Al₂O₃ with MgO.

Whole rock geochemistry and the "oceanic trend"

Modal abundances of clinopyroxene, orthopyroxene, olivine and spinel were calculated by mass balance for the peridotites that have both whole rock and mineral major element data available (Table 2–3). The modes of these rocks lie in the harzburgite field, except for the one lherzolite (OB93-52; Figure 2–2). Apparent negative spinel modes (with large errors) in some of the rocks probably results from the presence of spinel in trace amounts, and the possibility that the whole rock powder may not be wholly representative (i.e. a nugget effect). Figure 2–3 illustrates the relationship between modal abundance of olivine and olivine composition. This diagram is often used to characterize suites of peridotites with respect to their origin. Boyd (1989) was among the first to note that low-temperature peridotite xenoliths from the Kaapvaal and Siberian lithosphere possess lower and variable modal olivine for their high Fo contents relative to abyssal peridotites. Abyssal peridotites are believed to be residues of melting process that occurred at relatively low pressures, and the oceanic trend is constructed on the basis of observations made on these rocks (Boyd, 1989). Kelemen et al. (1998) postulated that the cratonic peridotites initially formed as residues of extensive melting and then reacted with SiO₂-rich slab derived melts to produce an excess of orthopyroxene as well as an olivine deficiency.

Of the xenoliths in this study, four lie on or near the oceanic trend (OB93-52, OB93-287, OB93-289, OB93-297; Figure 2-3). In comparison, the remaining samples have a higher Mg# for a given olivine mode than that defined by the oceanic trend, similar to the low-temperature cratonic xenoliths. Moreover, it is also notable that the Samoan peridotites of Hauri and Hart (1994) as well as some Hawaiian samples of Sen (1988) also lie considerably off the oceanic trend and within the range observed here for the Kerguelen peridotites.

One of the reasons for the occurrence of an oceanic trend is that modal olivine increases with increasing degrees of melting. This happens as long as the melting reactions produce olivine. It has been well established that peridotite melting does produce olivine at pressures below ~ 1.5 GPa (Kinzler and Grove, 1992; Walter, 1998 among others). That the Kerguelen xenoliths (and those from other OIB) have too little olivine for their Fo contents and plot off the oceanic trend, suggests that they are not simple residues of melting at low pressures. Niu et al. (1997) argued that melt-rock reactions could consume orthopyroxene and produce olivine, and that abyssal peridotites show evidence for this process. For the present suite of xenoliths this type of process is clearly not applicable on the basis of the following reasoning. As suggested by Kelemen et al. (1998), reaction between olivine-rich residues and SiO_2 -rich melt could produce orthopyroxene and shift a peridotite off the oceanic trend. Because of the discovery of the ubiquitous occurrence of SiO_2 -rich melt inclusions in peridotite xenoliths, including those from the Kerguelen Islands (Schiano et al., 1992; Schiano et al., 1994; Schiano and Clocchiatti, 1994), the mechanism of Kelemen et al. (1998) must be considered as a potential origin of the major element characteristics of the Kerguelen peridotites. However, as Schiano et al. (1992) noted, these SiO_2 -rich melt inclusions are enriched in LREE, and the proposed reaction would produce LREE-enriched signatures in the minerals of the rocks with which it

interacted. As will be discussed below, the REE abundance patterns in clinopyroxenes from some of the Kerguelen harzburgites that plot off the oceanic trend are not LREE enriched. For instance, OB93-80, with the least modal olivine at 41 wt.%, has clinopyroxene with $\text{La} = 0.28$ times chondrite and $\text{Yb} = 0.67$ times chondrite. Thus, a reaction between olivine and silica-rich melt does not appear to be the major mechanism to produce the observed SiO_2 enrichment that is observed here.

As discussed above, experiments have shown that mantle melting reactions change from olivine-producing at pressures < 1.5 GPa to olivine consuming and orthopyroxene-producing at higher pressures (e.g. Walter et al., 1995; Kinzler, 1997; Walter, 1998 among others). Melting residues that formed at high pressures could therefore plot above and off the oceanic trend and resemble the Kerguelen (and Samoan) peridotites. Indeed, Walter's (1998) experimental results on the melting of garnet-free peridotite showed that after 18% melting at 3.0 GPa, the residual harzburgite has 63.2 wt.% olivine with a composition of $\text{Fo}_{91.1}$, similar to one of the harzburgites from Mt. Trapeze (OB93-291; see Figure 2-3). This experimental residue has a bulk composition of 2.6 wt.% Al_2O_3 and 1.4 wt.% CaO, similar to the bulk compositions of those elements in the harzburgites of this study. If this experimental residue were to recrystallize at a lower pressure (less than approximately 1.0 GPa), mass balance indicates that the olivine mode would not change (63.2 wt.%), and that the new modal mineralogy would be 25 wt.% orthopyroxene, 6.9% calcic-clinopyroxene and 4.9% spinel, near the modes that are observed here.

Low pressure recrystallization of a garnet-bearing harzburgite residue that originally formed at high pressures could also result in modal olivine with a Fo content that lies off of the oceanic trend. Garnet breakdown during recrystallization would further consume olivine and produce orthopyroxene, thereby shifting a point horizontally to the right and off of the oceanic trend shown in Figure 2-3. In fact, Hauri and Hart (1994) interpreted trace

element abundance patterns of clinopyroxenes in some of the Samoan harzburgites as originating from the recrystallization of garnet-bearing precursors, and they, too, plot above the oceanic trend (see Figure 2-3). As will be discussed below, clinopyroxenes of the Kerguelen harzburgites do not possess any garnet-derived signatures; nevertheless their major element characteristics are consistent with an origin as recrystallized residues of extensive melting at high-pressure. Such extensive melting at high pressures can easily be envisioned in mantle plumes.

In contrast to the harzburgites, the lherzolite in this suite (OB93-52) plots on the oceanic trend. Its modal abundances are slightly depleted relative to KLB-1, a primitive peridotite from Kilborne Hole, NM, that is widely used as a starting material for experiments: 61.7 wt.% olivine in OB93-52 v. 58 wt.% in KLB-1, 9.5 wt.% clinopyroxene in OB93-52 v. 15 wt.% in KLB-1 (Takahashi, 1986). Sample OB93-52 has geochemical and isotopic characteristics that indicate it may be unmelted piece of the Kerguelen plume mantle, and these will be described in various sections to follow.

Present-day bulk major element compositions of the Kerguelen peridotites may not be a primary feature if new mineral phases were added during a late melt-rock interaction (or metasomatic) event. Petrographic details will be described in more detail below, but it should be noted here that petrographic evidence exists for the addition of new clinopyroxene to some of these harzburgites. The amounts of added clinopyroxene could be significant in comparison to the original clinopyroxene modes. In such specimens, the CaO contents (as well as most of the incompatible trace element contents) must be treated with caution. This reinforces the importance of microbeam-based trace element petrography like that practiced here, in deciphering the history of peridotites.

Trace element mass balance between whole-rocks and clinopyroxenes

Whole rock and mineral data can be used to examine which mineral phase hosts the trace elements and whether clinopyroxene can account for the trace element budget of the whole rocks. This calculation relies on the clinopyroxene modes that were calculated by mass balance from the whole rock and mineral major element data, and is therefore subjected to the uncertainties previously discussed. In addition, some of the harzburgites contain clinopyroxenes with vastly different trace element abundances, and the whole rock mass balance is influenced by this type of chemical heterogeneity. Although whole rock powders were carefully prepared to avoid minute veins of the host alkaline basalt, contamination may still be present. Given the contrast in trace element abundances between these strongly depleted harzburgites and those of enriched alkaline basalts, the magnitude of undetected contamination could be large.

To illustrate this situation, calculations were made for La and Yb to assess how much of the whole rock concentrations can be accounted for by clinopyroxene. Sample OB93-52 (a lherzolite), is one of the few cases where clinopyroxene is shown to be a predominant host for those elements; 68% of the La, and 82% of the Yb in the bulk rock can be accounted for by the 9.5 wt.% modal clinopyroxene in this sample (Table 2-2, Table 2-3, and see Table 2-8). Among harzburgites, OB93-297 presents another straightforward case; 100% of the La and 81% of the Yb are in clinopyroxene. For samples with heterogeneous clinopyroxenes (e.g. OB93-99, OB93-287), the La mass balance can be achieved with appropriate amounts of La-rich discrete grains. Concentrations of Yb are generally homogeneous among texturally different clinopyroxenes. Clinopyroxenes account for 66% of whole rock Yb in OB93-287, whereas about 40% of Yb is in clinopyroxene for OB93-99. For other depleted harzburgites such as OB93-289, OB93-307 and OB93-310, abundances of clinopyroxene and concentrations of these elements in clinopyroxene are so

low that whole rock abundances cannot be explained. The mass balance for Sr and Nd is particularly important when isotopic compositions of these elements are obtained only for whole rocks. This type of whole rock analysis is normally done because of the technical difficulties associated with the separation of pure clinopyroxene from a rock and the subsequent handling of minute quantities of these elements in clinopyroxene. The points presented above describe the difficulties associated with whole rock data and caution against reliance upon it.

Studies of mantle metasomatism began with the pioneering work of Frey and Green (1974) who recognized for the first time that unaccountable enrichment in incompatible elements occurs in rocks that are depleted in basaltic components. Petrographic manifestations of this phenomenon were not made clear until ion microprobe based trace element data of clinopyroxenes were produced (e.g. Sen, 1988; Takazawa et al., 1992; Hauri et al., 1993; Yang et al., 1998b among others). In these studies however, the connection between whole rock and mineral trace element data was not clearly made. In this sense, this study provides rare unequivocal petrographic evidence for trace element enrichment process in depleted rocks.

Figure 2–4 compares REE patterns for clinopyroxene and whole rocks for selected samples. Mass balance is violated for samples that show greater REE concentrations for whole rocks than clinopyroxenes (e.g. OB93-78, OB93-80, OB93-85). Until residential hosts other than clinopyroxene for incompatible elements in harzburgites can be identified, geochemical discussions based solely on whole rock data could be grossly misleading. In addition, the geochemical characterization of the lithosphere on the basis of whole rock data without that of trace element petrography can be equally misleading. In the sections that follow, most of the geochemical arguments are presented on the basis of clinopyroxene trace element data.

2.4.2. Mineral compositions

Major Elements

Representative major element compositions of minerals in peridotite xenoliths are summarized in Table 2-4, Table 2-5, Table 2-6, and Table 2-7. The entire dataset is tabulated in Appendix 1. Mineral compositions vary consistently among rock types, and the salient features can be summarized as follows. 1.) The Mg# of minerals are lowest in the wehrlitic dunites, followed by the lherzolite, and are highest in the harzburgites. For instance, the wehrlitic dunites have olivines with $\text{Fo}_{88.3}$ to $\text{Fo}_{90.2}$, the lherzolite has olivines with $\text{Fo}_{89.9}$ to $\text{Fo}_{90.2}$, and olivines in harzburgites have $\text{Fo}_{90.0}$ to $\text{Fo}_{92.2}$. 2.) Clinopyroxenes are broadly chromian diopsides in all of the rock types. Their compositions vary from Wo_{47-50} , En_{46-48} , $\text{Fs}_{3.6-4.9}$ for wehrlitic dunites, $\text{Wo}_{48.8}$, En_{47} , $\text{Fs}_{4.2}$ for the lherzolite, and Wo_{44-48} , En_{48-53} , $\text{Fs}_{2.9-5.5}$ for harzburgites. 3.) The TiO_2 and Al_2O_3 contents of clinopyroxenes are generally high in wehrlitic dunites (0.25 - 0.80 wt.% TiO_2 , 2.30 - 4.20 wt.% Al_2O_3), followed by the lherzolite (0.95 wt.% TiO_2 , 5.4 wt.% Al_2O_3) and those in harzburgites are quite low (<0.26 wt.% TiO_2 , < 2.8 wt.% TiO_2). 4.) Orthopyroxenes are low in Ca with Wo contents less than 2.3 for all samples. The orthopyroxene Mg# is the lowest in wehrlitic dunites at 89.5 followed by the lherzolite at 90.5, and 91.9 - 92.7 for the harzburgites. 5.) The Cr# [atomic Cr/(Cr+Al)] of spinels varies from 25 in the lherzolite (OB93-52), to 27 - 46 in OB93-83, 33 - 48 in wehrlitic dunites and 39-60 in harzburgites (Figure 2-5). 6.) Two pyroxene temperatures range from 878 to 1042°C.

The whole rock and mineral major element geochemistry of many of the Kerguelen harzburgites indicate their refractory nature. For instance, the spinel compositions, in comparison to those from abyssal peridotites (Dick and Bullen, 1984), suggest that these rocks are residual from relatively high degrees of partial melting, consistent with the elemental depletions noted above. Dick and Bullen (1984), postulated that the apparent

maximum Cr# of 60 observed in abyssal peridotites is a result of mantle melting to the clinopyroxene-out phase boundary. Similarly, in the case of the harzburgites of this study, the high Cr# of spinels (Figure 2-5) suggests that melting may have consumed all original clinopyroxene.

Major element compositions of minerals not only reflect the varying degrees of depletion of the peridotites, but also demonstrate subtle but significant changes during melt-rock interaction process and are closely related to trace element abundances in clinopyroxenes. For instance, "cpx4" in sample OB93-51 is depleted in its trace element abundances, has a higher Mg# (94.3) than the trace element enriched "cpx2" (Mg# = 93.2; Table 2-4, Table 2-8, and see Figure 2-6). In addition, "cpx4" is more diopsidic (Wo_{46} , En_{51} , $Fs_{3.3}$ v. Wo_{43} , En_{52} , Fs_4), lower in TiO_2 , Al_2O_3 (1.61 wt.% v. 2.05 wt.%) and Na_2O (0.26 wt.% v. 1.15 wt.%) than "cpx2". Similar trends can be detected at different levels of subtlety for OB93-99, OB93-287, and OB93-307, for which significant trace element diversity is also observed. These observations clearly indicate that the process which resulted in trace element enrichment involved silicate melt components with lower Mg#, higher Na_2O , TiO_2 , and Al_2O_3 contents than the precursor peridotite. Sample OB93-83, a phlogopite and apatite-bearing harzburgite that is considered to be most metasomatized sample of this study, has mineral major element compositions that are distinctly lower in Mg# (e.g. olivine with $Fo_{89.7}$), spinel Cr# as low as 27, and clinopyroxene with Wo_{41} , En_{53} , $Fs_{5.5}$, consistent with these other grain-by-grain trends (Figure 2-5).

As previously noted, OB93-52 has a whole rock composition that is slightly depleted relative to KLB-1, a popular primitive lherzolite (e.g. Takahashi, 1986). Mineral compositions of the two further reflect their similarities, with olivine Fo content ($Fo_{90.2}$ v. $Fo_{89.1}$ in KLB -1), TiO_2 in clinopyroxene (0.95 wt.% in OB93-52 v. 0.58 wt.% in KLB-

1), Al_2O_3 in clinopyroxene (5.41 wt.% v. 7.4 wt.%), Al_2O_3 in orthopyroxene (2.78 wt.% v. 4.97 wt.%), and Cr# of spinel (25.2 v. 8.2).

Trace elements

Representative trace element compositions of clinopyroxenes from peridotite xenoliths are summarized in Table 2–8. The entire dataset is tabulated in Appendix 1. Clinopyroxenes in harzburgites from the two localities have variable REE contents both among and within individual xenoliths. The range of REE contents in clinopyroxenes from the harzburgites and the one lherzolite from Lac Supérieur are shown in Figure 2–6, and those from the Mt. Trapeze harzburgites are shown in Figure 2–7.

The relationship between texture and chemistry

Trace element compositions of clinopyroxenes from the xenoliths were extensively examined in this work, and constitute one of the major results of this chapter. Particularly important is the observation that changes of clinopyroxene texture and geochemistry are correlated, indicating their mechanisms of formation. For example, trace element abundance patterns of clinopyroxenes that are highly metasomatized can be used to infer the identities of the metasomatic agents that created them. Figure 2–6 and Figure 2–7 illustrate the ranges of REE abundances observed in the clinopyroxenes from the peridotites of this study. The total range of REE concentrations for the Lac Supérieur specimens (Figure 2–6) are greater than those of any other peridotites observed thus far. For instance, kimberlite-borne peridotite xenoliths are known to be heterogeneous within individual pipes (e.g. Shimizu et al., 1997), but the variability seen in the Lac Supérieur suite is much greater. In these samples, Ce varies by a factor of 5000 (from 0.04 times chondrite to more than 200 times chondrite). The Samoan peridotite xenoliths display trace element variations nearly comparable, but smaller than the Lac Supérieur suite (Hauri and Hart, 1994). It is likely

that a combination of geochemical variability among pre-metasomatic lithosphere pieces and further variations induced by metasomatism create these observed trace element ranges. It is therefore important to note in this context that large variations are also observed within a single sample (OB93-51; Figure 2-6). Samples from Mt. Trapeze show much less variation; nevertheless it is notable that Ce varies by a factor of 500 between OB93-307 and OB93-286, and is accompanied by identical Yb concentrations for the two samples (Figure 2-7). This is reminiscent of the type of single-sample variation shown by OB93-51 (Figure 2-6).

Figure 2-8 summarizes the REE variations in clinopyroxenes from individual harzburgites in which a correlation between texture and chemistry is observed. For instance, in OB93-51, clinopyroxenes occur in two separate textural domains: in symplectite intergrowth with spinel (Figure 2-1), and also as discrete grains. The data for "cpx4s" (a symplectite grain) resembles numerous other harzburgitic clinopyroxenes, which also occur as symplectites, in abundances of Yb through Nd, whereas La and Ce concentrations are significantly elevated in comparison (Table 2-8). Another symplectite clinopyroxene from this same sample, "cpx3s", appears to be more extensively modified, while maintaining an identical Yb concentration as "cpx4s". In contrast, co-existing discrete clinopyroxene grains in this sample ("cpx1" and "cpx2") have smooth, LREE-enriched patterns that are indicative of complete equilibration with a strongly LREE-enriched melt. Thus, it appears that the REE variations in OB93-51 record successive modifications of originally REE depleted clinopyroxenes and precipitation of new grains that are equilibrated with the metasomatic melt. Comparisons of the REE variations shown in Figure 2-8 clearly exhibit a progressive change of REE patterns from La, at the most modified, and diminishing towards Yb. This type of progressive variation in clinopyroxene REE abundances has been observed in numerous peridotite clinopyroxenes, and has been

successfully modeled by chromatographic melt-mantle reaction mechanisms (Navon and Stolper, 1987; Bodinier et al., 1990; Takazawa et al., 1992; Hauri and Hart, 1994; Yang et al., 1998b among others). That the clinopyroxene Wo content for OB93-51 decreases from 45.73 for "cpx4s" to 43.48 for "cpx2", indicates that "cpx2" equilibrated at higher temperatures than "cpx4s". The ubiquity of harzburgitic clinopyroxenes similar to "cpx4s" and that various extents of modification are illustrated in these internally heterogeneous samples suggest: 1.) harzburgites were a common precursor rock to metasomatism, 2.) the melt-rock reaction that resulted in metasomatism was of short duration, and 3.) this process occurred shortly before the eruption that carried these xenoliths to the surface.

It should be noted that this relation does not seem to hold for some of the Mt. Trapeze xenoliths; in some cases the symplectite clinopyroxenes show a greater metasomatic signature than the discrete clinopyroxenes (OB93-299, OB93-318). However, because these analyses were made on clinopyroxenes in grain mounts rather than in thin section, it is possible that the grains we describe as discrete may actually be fragments of symplectite clinopyroxenes. Regardless of an imperfect textural correlation in some of the Mt. Trapeze harzburgites, that it does exist in many of the xenoliths it is the important observation. It demonstrates that the metasomatic event modified some existing grains as well as resulted in the precipitation of new discrete grains.

Carbonatite melts and metasomatism

The majority of the LREE enriched samples are from Mt. Trapeze, and over half of the Mt. Trapeze samples are LREE enriched (Table 2-8, Figure 2-9). The $[La/Yb]_N$ ranges of this type of sample are from 1.3 to 31.8, and the middle to heavy REE are relatively constant at ~2 to 3 times chondritic values. The LREE enrichment is indicative of advanced stages of metasomatic modification of these clinopyroxenes. That the middle to heavy REE are

mostly unfractionated and at abundances much the same as those in the REE depleted harzburgites is evidence that the precursor clinopyroxenes probably had a similar origin as clinopyroxenes from the depleted harzburgites. Some of these clinopyroxenes possess smooth and strongly LREE-enriched REE patterns that are indicative of near-equilibrium with a LREE-enriched melt. In addition, Ti and Zr abundances in these clinopyroxenes are typically less than the neighboring REE in chondrite-normalized trace element diagrams (spidergrams), indicating that melts in equilibrium with these clinopyroxenes are also depleted in Ti and Zr (Figure 2-10)

With known clinopyroxene/melt partitioning it is possible to determine the REE contents of melts that were in REE equilibrium with the clinopyroxenes from these rocks. These hypothetical melts can then be compared to the REE contents of other melts that could be potential metasomatic agents. Many of the samples have clinopyroxenes that are in REE equilibrium with melts that are similar to Kerguelen Islands basalts (Figure 2-11). In contrast, Figure 2-12 illustrates that trace element patterns of equilibrium melts for OB93-51, OB93-83, OB93-99, OB93-280, OB93-286 (cpx1), OB93-291, OB93-297 and OB93-313 plot within the field defined by the trace element abundances of carbonatites (Nelson et al., 1988). This indicates that carbonatitic melts were involved in metasomatic processes associated with the Kerguelen plume system. These characteristics of carbonatite metasomatism are similar to those found in mantle xenoliths from Samoa and the Society Islands (Schiano et al., 1992; Hauri et al., 1993), indicating that carbonatite metasomatism could be a common mantle process even in oceanic environments. Furthermore, Schiano *et al.* (1994), found cogenetic carbonate and silicate melt inclusions in peridotites from the Southeastern Province of the Kerguelen Islands, proof that carbonate is at the least present in the mantle beneath Kerguelen. Sample OB93-83 is an extreme case and was metasomatized to the point of having precipitated new clinopyroxene, as well as phlogopite

and apatite. These minerals are also observed in other examples of carbonatite metasomatism in mantle xenoliths from both continental and oceanic environments (Green and Wallace, 1988; Yaxley et al., 1991; Hauri et al., 1993; Ionov et al., 1993; Rudnick et al., 1993), and thus may be characteristic of carbonatite metasomatism.

The origin of the precursor harzburgites

One of the characteristics of the Kerguelen peridotite suites is the ubiquitous occurrence of depleted harzburgites upon which metasomatic signatures were imprinted. As discussed earlier, their major element compositions are inconsistent with their origin as residues of low pressure melting processes. It is thus unlikely that these xenoliths are pieces of oceanic lithosphere upon which the Kerguelen plume impinged.

The argument was presented that residues of melting at high pressure, where melting reactions produce orthopyroxene and consume olivine, could potentially explain the major element compositions of the harzburgites of this study. It was further noted that low-pressure recrystallization of the experimental residues of Walter (1998) could produce bulk major element compositions similar to the range of those of the Kerguelen harzburgites.

As shown in Figure 2-13, the REE abundance patterns of the clinopyroxenes encountered here are substantially different than those from abyssal peridotites (Johnson et al., 1990) and those from Samoan peridotites (Hauri and Hart, 1994). The strong mid to heavy REE fractionation and characteristic positive Zr anomalies observed by Hauri and Hart (1994) in clinopyroxenes from the Samoan peridotites led these authors to argue for their origin as recrystallized residues of extensive melting in the presence of garnet. In the case of Kerguelen, the clinopyroxenes from the peridotites are characterized by much less fractionation of the REE and low overall concentrations that are devoid of any recognizable precursor signatures. The following arguments are made to supplement the those made

from the major elements, but now using trace element abundance patterns in clinopyroxenes in an attempt to understand the history of the Kerguelen peridotites.

Assuming a starting material that has chondritic REE abundances, the bulk rock trace element abundance patterns of the high pressure residues reported in Walter (1998) can be estimated with the mineral-melt partition coefficients recently reported by Salters and Longhi (1999). It is necessary to use these particular partition coefficients in this calculation because the clinopyroxene in Walter's (1998) high pressure residue is pigeonitic and partition coefficients obtained for high-Ca pyroxenes are not relevant. The parameters used to determine the bulk REE abundances of the residue in this calculation were: run 30.07, at 3.0 GPa and 1530°C, with olivine + orthopyroxene + subcalcic clinopyroxene with 18.5% batch melting (from Walter, 1998). The REE abundances of the calcic clinopyroxenes that would be present after the originally high pressure residual harzburgite recrystallizes at low pressures were then calculated. This was done according to the clinopyroxene/orthopyroxene ratio that is predicted from the major element composition of the experimental charge and using clinopyroxene/orthopyroxene partitioning for the REE (Eggins et al., 1998). The results of this calculation predict this low pressure calcic clinopyroxene to have the following chondrite-normalized concentrations: Ce: 0.067; Nd: 0.18; Sm: 0.29; Er: 0.62; and Y: 0.65. These abundances are remarkably close to the those of the LREE-depleted clinopyroxenes found in OB93-307 (Figure 2-14). Similarly LREE-depleted patterns are observed in OB93-95 (cpx2) and OB93-310 and can easily be interpreted in the same manner. Walter (1998) also reports results on residues from different pressures that are both garnet-free harzburgitic as well as olivine-pigeonite, and they could produce similar results as described above. The important point is that the magnitude of REE fractionation for an olivine + orthopyroxene + subcalcic clinopyroxene melt assemblage is much less than that of a garnet-bearing one.

As shown in Figure 2-14, these LREE-depleted clinopyroxenes coexist with clinopyroxenes that have LREE enriched patterns (OB93-307, "cpx10") or flat patterns (OB93-95) within the same rock. In particular, OB93-307 has clinopyroxenes with extreme LREE abundances, but share identical Yb and Er concentrations. This demonstrates that originally LREE-depleted patterns were modified by metasomatism to various extents and that HREE concentrations in these clinopyroxenes were maintained during that process.

Most of the harzburgites have clinopyroxenes that possess upward-sloping HREE abundances from Dy to Er to Yb that are accompanied by relatively flat LREE (Figure 2-15). The observations just mentioned and those on the metasomatized clinopyroxenes (Figure 2-9) clearly suggest that LREE are most susceptible to reaction with LREE-enriched melts, whereas the HREE such as Dy, Er and Yb are the last of the REE to be modified during metasomatism. Therefore, this suggests that most of the harzburgite clinopyroxenes originally had LREE-depleted patterns such as OB93-307 or OB93-95 ("cpx2") and were modified by metasomatism. Because the LREE concentrations of these clinopyroxenes are so low (<0.01 times chondrite), even incipient metasomatism could easily modify light through heavy REE.

On the basis of the discussion presented here, the conclusion is that the majority of the harzburgites in the Kerguelen peridotite suites were originally residues of extensive melting at high pressures. These residues were subsequently metasomatized to vastly different extents by plume-related melts including those that are carbonatitic. Although the model of low pressure recrystallization of high pressure melting accounts for many of the major and trace element characteristics of these harzburgites, it does not predict the low olivine modes (down to 40 wt.%) that are present in some of the rocks (OB93-80, OB93-85), and an additional process be involved in the petrogenesis of these two particular olivine-poor and orthopyroxene-rich rocks.

Lherzolites

Only one spinel lherzolite (from Lac Supérieur, OB93-52) is present in this suite of xenoliths. It is relatively fertile and contains clinopyroxenes and olivines with low Mg# relative to the harzburgites (Mg# = 92 and 90 respectively; Table 2-4, Table 2-6). The clinopyroxenes in this xenolith also have the highest TiO₂ content of the sample suite (~0.95 wt.%). The spinel Cr# are among the lowest observed in these xenoliths (~25; Table 2-7, Figure 2-5). As previously noted, the major element mineral compositions are similar to but slightly depleted relative to those of the KLB-1 peridotite. Clinopyroxene REE patterns are somewhat LREE depleted ($[La/Yb]_N = 0.65$), at about 10 times chondritic abundances (Figure 2-16). These data indicate that sample OB93-52 is a fertile piece of mantle that could be typical of the source of Kerguelen lavas. As discussed below, the Sr, Nd, and Os isotopic composition of this rock support this interpretation. The REE patterns show no signs of metasomatism. This, combined with the fertile nature of the whole rock and mineral geochemistry, argue against a metasomatic origin for this xenolith. Normalized REE contents of melts in equilibrium with these clinopyroxenes lie within the field of REE abundances of Kerguelen basalts (Figure 2-11). Thus, this xenolith is a good candidate for a Kerguelen basalt source, and may represent a piece of unmelted plume mantle.

Wehrlitic dunites

All minerals from the wehrlitic-dunites (OB93-61c, OB93-64, OB93-82; Lac Supérieur only), have significantly lower Mg# than the other peridotites in this suite (*e.g.* olivine Mg# = 88.3 to 90.2; Table 2-4, Table 2-6, Table 2-7). Clinopyroxenes contain higher TiO₂, Al₂O₃, Na₂O, and Cr₂O₃ contents relative to the Lac Supérieur harzburgites, and are closer to the abundances observed in clinopyroxene from the Mt. Trapeze harzburgites. The spinels have a variable Cr# and are within the range defined by the other peridotites, but are offset to lower Mg# for a given Cr# (Table 2-7, Figure 2-5). The REE compositions of the

clinopyroxenes are all similar at approximately 6-10 times chondritic values, though clinopyroxenes from OB93-82 have heterogeneous HREE (Table 2-8, Figure 2-17).

The geochemistry of these rocks indicate that they are not melting residues, nor are they fertile peridotites. However, these data are consistent with both a cumulate and a replacive origin similar to the clinopyroxene-bearing dunites discussed by Kelemen et al. (1995). The clinopyroxenes are in REE equilibrium with Kerguelen Islands basalts (Figure 2-11), and these rocks could be the product of reaction with or cumulates of melts directly connected to Kerguelen Islands basalts. The clinopyroxene trace element variations especially LREE/HREE ratios appear to be correlated with Mg#, TiO₂, and Al₂O₃ contents, and these features could be attributed to a chromatographic-type reaction with basaltic melt that has equilibrated to different degrees among these samples (Navon and Stolper, 1987; Takazawa et al., 1992; Hauri and Hart, 1994). Although these data do not distinguish between a cumulate or replacive origin for the dunites, they are likely related to the Kerguelen Islands basalts. The isotopic data presented below also support this interpretation.

REE enriched harzburgites

Clinopyroxenes from the REE enriched harzburgites have unfractionated REE patterns at abundances of approximately 10 times chondritic values (Figure 2-18). This type of harzburgite is only present in the Mt. Trapeze location, but is similar in its REE patterns to the dunite xenoliths from Lac Superieur. Clinopyroxenes from these samples have REE equilibrium melts that are similar to the Kerguelen Islands basalts and may have formed by reaction with this type of melt. Sample OB93-297 belongs to the LREE-enriched group, but its normalized HREE abundances are at approximately 10 times chondrites like the

enriched samples. This suggests that it may have been an enriched sample before it was yet again metasomatized by a carbonatite melt.

Pyroxenites

Pyroxenite samples are only present in the Lac Supérieur suite and have approximately 10 times chondritic REE concentrations (Table 2–8). On the basis of the REE and mineralogy, there are two compositional types (Figure 2–19): phlogopite-bearing with LREE enriched clinopyroxenes ($[La/Yb]_N = 3$ to 6; OB93-81b and OB93-94), and those with LREE depleted clinopyroxenes ($[La/Yb]_N = 0.3$ to 1). Overall, the six pyroxenites are close to trace element equilibrium with Kerguelen basalts. This, and that they are megacrystic suggests that they are probably high pressure cumulates of Kerguelen magmatism (e.g. Irving and Frey, 1984).

2.4.3. Isotopic compositions of Sr, Nd and Os

Sr and Nd isotopes

The Sr and Nd isotopic compositions of clinopyroxenes are summarized in Table 2–9 and shown in Figure 2–20. The total ranges of the isotopic compositions observed in this study far exceed those of Kerguelen basalts and are greater than the published datasets on the Kerguelen system (Gautier et al., 1990; Weis and Frey, 1991; Weis et al., 1993; Barling et al., 1994; Mattielli et al., 1996; Weis et al., 1998; Yang et al., 1998a).

The lherzolite sample (OB93-52) falls within the center of the range of Sr and Nd isotopic compositions of Kerguelen basalts ($^{87}Sr/^{86}Sr = 0.705205$, $\epsilon_{Nd} = -0.3$), consistent with its proposed origin as a piece of fertile plume mantle. The isotopic composition of dunite sample OB93-64 is also similar to Kerguelen basalts ($^{87}Sr/^{86}Sr = 0.705238$, $\epsilon_{Nd} =$

-2.5). That this sample is nearly in Sr and Nd isotopic equilibrium with Kerguelen basalts is consistent with either of the proposed origins for the dunites.

At the "isotopically depleted" end of the spectrum is harzburgite OB93-78, with $^{87}\text{Sr}/^{86}\text{Sr} = 0.70330$. The Nd content and the abundance of clinopyroxene in this rock are so low that an attempt to measure the Nd isotopic composition was unsuccessful (Dr. E. Nakamura at Okayama University). This is one of the most depleted samples among the rocks in this study, and despite its slightly LREE-depleted pattern (see Figure 2-15) it is uncertain whether any metasomatic changes have occurred. Because of the $^{87}\text{Sr}/^{86}\text{Sr}$ of clinopyroxene from this rock, which is within the range of SEIR basalts and lower than any known Kerguelen samples, the provenance of this xenolith is important. Although the olivine composition is not available for this rock, the whole-rock major element composition ($\text{Mg\#} = 91.6$, $\text{SiO}_2 = 48.79$ wt.%, Table 2-2) strongly indicates that this sample could well plot off the oceanic trend together with the other harzburgites from Kerguelen (Figure 2-3). If this were the case, and if it were taken to suggest its origin as a high pressure residue that was recrystallized at low pressure, like many of the other harzburgites, the $^{87}\text{Sr}/^{86}\text{Sr}$ data might have to be considered as defining the most depleted member of the Kerguelen plume observed thus far. The most depleted isotopic compositions known were reported by White and Hofmann (1982) on basalts from Foch Island ($^{87}\text{Sr}/^{86}\text{Sr} = 0.70388$, $^{143}\text{Nd}/^{144}\text{Nd} = 0.51291$), northern Kerguelen Islands (Figure 1-1). An alternative scenario for the origin of this rock is that a recrystallized high-pressure residue was originally LREE depleted was metasomatized by MORB melt from the SEIR. This process would have increased the LREE, but not affected the $^{87}\text{Sr}/^{86}\text{Sr}$. The Os isotopic composition (see below) is $^{187}\text{Os}/^{188}\text{Os} = 0.1266$, indistinguishable from many of the other harzburgites and is not definitive for the provenance of this rock.

At the most "isotopically enriched" end of the spectrum is a megacrystic clinopyroxenite, OB93-94, with $^{87}\text{Sr}/^{86}\text{Sr} = 0.70869$ and $^{143}\text{Nd}/^{144}\text{Nd} = 0.51199$ ($\epsilon_{\text{ND}} = -12.7$). This is the most enriched isotopic composition that has been observed in any rocks from the Kerguelen Islands. Basalts from ODP drill-site 738 in the southern Kerguelen Plateau ($62^\circ 42'\text{S}$, $82^\circ 47'\text{E}$, Figure 1-1) have eruption age corrected $^{87}\text{Sr}/^{86}\text{Sr} = 0.7090$ to 0.7130 and $\epsilon_{\text{ND}} = -7$ to -13 (Alibert, 1991; Mahoney et al., 1994). These isotopic signatures were interpreted as being derived from the incorporation of old subcontinental lithosphere beneath Gondwanaland. Mahoney et al. (1994) reported similar isotopic compositions for basalts from the Naturaliste Plateau, and argued that contamination by continental lithosphere produced these isotopic signatures as well. One of the basalts (in the late Pleistocene-Holocene Big Ben series) from Heard Island, also believed to be derived from the Kerguelen plume, has $^{87}\text{Sr}/^{86}\text{Sr} = 0.70792$ and $^{143}\text{Nd}/^{144}\text{Nd} = 0.51239$ (Barling et al., 1994). These authors postulated that recycling of upper crustal material that had previously been subducted beneath a continental precursor of Gondwanaland produced this isotopic signature.

Sample OB93-94, as noted above, is a cumulate from a melt with trace element characteristics similar to those of the Kerguelen Islands basalts, and was subsequently (~ 20 Ma, on the basis of a two-point Rb-Sr isochron for the phlogopite-clinopyroxene pair) metasomatized to produce phlogopite. Incorporating the Heard Island isotopic data with the present dataset significantly enlarges the ranges of Sr and Nd isotopic compositions of the Kerguelen plume as a whole, in comparison to the Kerguelen basalts alone. Overall, these data appear to encompass the isotopic ranges observed in the present study, except for the isotopically depleted OB93-78.

The Sr and Nd isotopic compositions of clinopyroxene from peridotites of this study and clinopyroxene data from Mattielli et al. (1996) and Mattielli et al. (1999) plot near the isotopically enriched end of the Kerguelen basalts and extend into the ranges of the Heard Island basalts (Figure 2–20). The host dike that brought the xenoliths to the surface has $^{87}\text{Sr}/^{86}\text{Sr} = 0.70638$ and $^{143}\text{Nd}/^{144}\text{Nd} = 0.51242$, isotopically more enriched than any basalts that have been analyzed from the Kerguelen Islands. It is, however, well within the ranges for the Heard Island basalts (Barling et al., 1994) and a legitimate member of the Kerguelen plume diversity. Note also that OB93-52, the alleged unmelted Kerguelen plume peridotite, possesses Sr and Nd isotopic compositions typical for the Kerguelen basalts.

Samples OB93-51 and OB93-83 were identified as being metasomatized by carbonatitic melts. In this sense, it is notable that their Sr and Nd isotopic compositions are identical, suggesting a unique carbonatitic component in the plume. However, Mattielli et al. (1999) reported the Sr and Nd isotope data for a harzburgitic clinopyroxene that was evidently metasomatized by carbonatite, but is not as isotopically enriched as these two samples (see Figure 2–20). This data point appears to indicate that the "carbonate components" of the Kerguelen plume are also isotopically heterogeneous.

As discussed earlier, when trace element-enriched plume melts interact with depleted peridotites, the Sr and Nd isotopic compositions of the melts would remain unchanged, whereas the Sr in the peridotites would be equilibrated with the infiltrating melts. A study by Lesher et al. (1990) of isotopic v. chemical equilibration rates during magma mixing demonstrated that isotopic equilibration is achieved much faster than chemical equilibration during the mixing of melts. On the basis of this result, it is possible that during melt-rock reaction processes such as those observed here, isotopic exchange could occur more quickly than trace element equilibrium. Therefore, this suggests that even for samples with signatures of incomplete reactions with infiltrating melt (e.g. clinopyroxenes with highly

heterogeneous trace element abundance patterns in individual rocks), the incompatible element isotopic compositions would be completely dominated by the plume-derived melts. Taking this idea one step further, the diversity of the Sr and Nd isotopic compositions observed here could be evidence for isotopic heterogeneities of the Kerguelen plume that are much larger than that sampled by Kerguelen Islands basalts.

In the two xenoliths where clinopyroxene-phlogopite pairs were analyzed (OB93-83, OB93-94), a two-point Rb-Sr isochron gives an "age" of 18 Ma for OB93-83 and 20 Ma for OB93-94. As described before, OB93-83 is fully equilibrated with the metasomatic agent, and clinopyroxene and phlogopite may have formed under isotopic equilibrium. If this is true, the two-point isochron could have significant meaning, namely the time when metasomatism occurred. Sample OB93-94 however, is a clinopyroxene substrate into which a phlogopite-producing melt infiltrated and isotopic equilibrium between clinopyroxene and phlogopite may not have been achieved. Thus, it is equivocal whether if these ages have significance.

Os isotopes

It has been well documented that the Os isotopic composition of peridotites are insensitive to melt-rock interaction due to the compatible character of Os in the peridotite-basalt system (Morgan et al., 1981; Walker et al., 1989; Pearson et al., 1995; Hart and Ravizza, 1996). It follows that interaction between plume-derived melts and peridotites should not modify the Os isotopes of the peridotites, but those of the melts could be equilibrated with the peridotites. For instance, Hauri (1997) concluded that partial melts of pyroxenites should acquire the Os isotopic signature of the peridotite through which is traveling within a distance of 200 to 1000 m. The data from this study indicates that the major elements and incompatible trace elements are decoupled. Figure 2-21 shows that there is no relation

between Os isotopic composition and bulk Al_2O_3 contents nor with $[\text{La/Yb}]_N$, demonstrating that the metasomatism in this case did not include Os.

On the basis of this discussion, the Os isotope compositions of peridotite xenoliths can be used to decipher their provenance. For the Kerguelen plume, which has evolved through different geodynamic environments, the provenance of peridotite xenoliths could shed new light on the dynamics of mantle plumes. The results of the Os isotope analysis of peridotite xenoliths are one of the major contributions of this study and are presented in Chapter 3.

The $^{187}\text{Os}/^{188}\text{Os}$ of whole-rock peridotites show that these that they have differing provenance. Many of the samples from Mt. Trapeze have an origin in the Gondwanaland subcontinental lithospheric mantle (Hassler and Shimizu, 1998). It is important to note here that significant isotopic heterogeneities exist among peridotites of non-Gondwanaland origin (all samples from Lac Superieur and OB93-306 and OB93-314). Table 2-10 shows that $^{187}\text{Os}/^{188}\text{Os}$ varies from 0.1252 to 0.1383, covering a significant portion of the Os isotopic variations reported for individual plume systems. For instance, this range covers all but one sample of the Loa trend of Hawaii discussed by Hauri (1996), showing that the peridotite "matrix" of a plume system could contain a large indigenous Os isotopic variability. Possible correlations between Os and other isotopes are not discussed here, because as noted earlier, the Sr and Nd compositions in clinopyroxenes are dominated by plume-derived melts, but the Os compositions are primary. Therefore, identifying the less radiogenic end of the spectrum (excluding the extreme unradiogenic samples) with the modern-day oceanic lithosphere is tempting but not straightforward. This is because the major element and mineralogy-based discussion given earlier implies that the bulk of the harzburgites are recrystallized residues of high-pressure melting and thus members of the plume, rather than modern oceanic lithosphere.

2.5 Summary

The results of this study of ultramafic xenoliths from the Courbet Peninsula, Kerguelen Islands provide the following conclusions:

1. Harzburgites, a lherzolite, wehrlitic dunites, and pyroxenites comprise the ultramafic xenolith suites. The harzburgites of the suite have the highest whole rock Mg#, followed by the lherzolite, and the wehrlitic dunites and pyroxenites. The harzburgites are depleted in basaltic components. Many of the harzburgites plot off the oceanic trend that is defined by the relationship between modal abundances of olivine and olivine composition. This indicates that these rocks did not form by a simple melting process. However, residues of melting at high pressures that recrystallized at lower pressure, could create the olivine modes and compositions of some of the harzburgites observed here. The composition of the lherzolite is fertile and lies on the oceanic trend. Its composition is similar to those of fertile peridotites such as KLB-1 from Kilborne Hole, NM.
2. The major element compositions of minerals that make up these rocks also reflect their depleted character in the case of the harzburgites, or the degree of fertility for the lherzolite. In situ trace element compositions of clinopyroxenes as determined by ion microprobe, are correlated with the petrographic occurrence of clinopyroxene and indicate their mechanism of formation. The total range of REE content is large, and varies by a factor of 5000 in the Lac Superieur suite. Large REE variations are also present within single samples, and these REE differences correspond to clinopyroxene texture. Discrete clinopyroxenes have significant elevations in their REE contents relative to those that occur in symplectite intergrowths with spinel. Clinopyroxenes from some samples record successive metasomatic modifications of originally depleted

clinopyroxenes as well as the precipitation of new grains that are equilibrated with the metasomatic melt. This type of successive REE variation in clinopyroxenes occurs by a chromatographic melt-mantle reaction mechanism.

3. Trace element abundances of melts in equilibrium with the clinopyroxenes from these rocks indicate carbonatite melts and alkaline basalts as the metasomatic agents. The low pressure recrystallization of high pressure residues model that was developed from major element considerations is also supported by the clinopyroxene trace element compositions. The LREE depleted nature of clinopyroxenes from some of these harzburgites can be recreated by this model. Other REE enriched harzburgites, the lherzolite, the wehrlitic dunites and the pyroxenites all have trace element characteristics indicative of a relation to the Kerguelen Islands basalts.
4. The Sr and Nd isotopic compositions of clinopyroxenes from the peridotites were dominated by the infiltrating melts. The lherzolite lies in the center of the field defined by the basalts, further supporting the idea that it is a piece of unmelted plume mantle. The ranges of isotopic composition of the xenoliths are much larger than those observed thus far for the Kerguelen Islands basalts, and it is possible that all of these compositions are associated with the Kerguelen plume. One sample, a pyroxenite is extreme in its isotopic enrichment, and probably formed as a cumulate from magmas that were contaminated by continental crust.
5. The Osmium isotopic compositions show that subcontinental lithosphere, probably stranded from the rifting of Gondwanaland, are present beneath the Kerguelen Islands. The remaining samples cover a significant portion of the $^{187}\text{Os}/^{188}\text{Os}$ ranges reported on basalts from other plumes, demonstrating that the Kerguelen plume may have a diverse Os isotopic composition.

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2.7 Tables

Table 2-1. List of samples.

Sample List	Lac Superieur	Mt. Trapeze		Lac Superieur	Mt. Trapeze
REE depleted samples	OB93-53	OB93-289	REE enriched samples		OB93-283
	OB93-56	OB93-314			OB93-284
	OB93-58	OB93-314			OB93-297
	OB93-66				OB93-305
	OB93-74				OB93-306
	OB93-77				
	OB93-78		Heterogeneous REE samples	OB93-51	OB93-282
	OB93-79			OB93-99	OB93-299
	OB93-80				OB93-318
	OB93-81c				
	OB93-85		lherzolite	OB93-52	
	OB93-86				
	OB93-89		wehrlitic dunites	OB93-61c	
	OB93-98			OB93-64	
				OB93-82	
LREE depleted samples	OB93-95	OB93-307			
		OB93-310	Pyroxenites	OB93-75	
				OB93-87	
LREE enriched samples	OB93-83	OB93-280		OB93-100	
		OB93-281		OB93-102	
		OB93-286		OB93-81b	
		OB93-287		OB93-94	
		OB93-288			
		OB93-291			
		OB93-298			
		OB93-301			
		OB93-302			
		OB93-308			
		OB93-309			
		OB93-313			

Table 2-2. Bulk rock major and trace element compositions of selected xenoliths.

Lac Superieur								
	Lherzolites	Harzburgites						
	OB93-52	OB93-56	OB93-77	OB93-78	OB93-80	OB93-83	OB93-85	OB93-99
<i>XRF wt. %</i>								
SiO ₂	43.35	48.63	48.03	48.69	49.37	48.54	49.32	48.23
Al ₂ O ₃	3.26	0.91	0.71	0.90	1.31	1.20	0.95	0.52
TiO ₂	0.12	0.01	0.00	0.00	0.02	0.03	0.00	0.02
FeO*	9.24	7.18	7.31	7.14	6.80	8.44	7.12	7.22
MnO	0.14	0.12	0.12	0.12	0.12	0.14	0.12	0.12
CaO	2.31	0.77	0.58	0.68	1.05	0.69	0.89	0.93
MgO	40.89	42.50	44.90	43.82	40.35	41.52	42.07	42.98
K ₂ O	0.01	0.00	0.00	0.00	0.00	0.07	0.00	0.00
Na ₂ O	0.17	0.07	0.11	0.09	0.16	0.12	0.07	0.05
P ₂ O ₅	0.02	0.02	0.01	0.01	0.02	0.04	0.02	0.01
Total	99.51	100.20	101.77	101.45	99.20	100.79	100.55	100.08
Mg#	88.75	91.34	91.63	91.62	91.36	89.76	91.33	91.39
<i>XRF ppm</i>								
Ni	2536	2279	2375	2338	2038	2122	2233	2407
Cr	7831	3419	2597	3334	4646	1516	3192	1443
V	67	d.l.	d.l.	d.l.	45	d.l.	d.l.	d.l.
Sr	20	9	9	d.l.	23	18	25	31
Zr	9	5	6	6	8	32	6	7
Ga	5	1	3	3	1	3	1	3
Cu	41	15	15	12	12	8	14	18
Zn	68	38	43	41	48	59	39	36
Nb	d.l.	d.l.	d.l.	d.l.	d.l.	12	d.l.	d.l.
Rb	d.l.	d.l.	d.l.	d.l.	d.l.	3	d.l.	d.l.
<i>ICP-MS ppm</i>								
La	0.26	0.21	0.02	0.08	0.30	2.31	0.08	0.11
Ce	0.66	0.39	0.01	0.10	0.57	4.80	0.13	0.22
Pr	0.12	0.05	d.l.	0.01	0.07	0.62	0.02	0.03
Nd	0.61	0.18	0.02	0.04	0.28	2.79	0.06	0.12
Sm	0.21	0.04	0.01	0.02	0.06	0.68	0.01	0.03
Eu	0.09	0.01	d.l.	0.01	0.02	0.21	d.l.	0.01
Gd	0.28	0.03	0.01	0.01	0.05	0.59	0.01	0.02
Tb	0.05	d.l.	d.l.	d.l.	0.01	0.10	d.l.	0.01
Dy	0.34	0.04	0.01	0.01	0.04	0.54	0.02	0.04
Ho	0.07	0.01	d.l.	d.l.	0.01	0.11	d.l.	0.01
Er	0.22	0.01	0.01	0.01	0.02	0.27	0.01	0.02
Tm	0.03	d.l.	d.l.	d.l.	d.l.	0.04	d.l.	0.00
Yb	0.22	0.02	0.01	0.02	0.02	0.25	0.01	0.03
Lu	0.04	d.l.	d.l.	0.01	0.01	0.04	d.l.	0.01
Ba	3	1	d.l.	1	13	6	1	1
Th	0.04	0.03	d.l.	0.01	0.05	0.25	0.01	0.02
Nb	0.18	0.42	0.02	0.04	0.35	11.94	0.13	0.27
Y	1.91	0.14	0.05	0.04	0.24	2.69	0.08	0.22
Hf	0.11	0.03	0.01	0.01	0.05	0.99	0.01	0.02
Ta	0.20	0.62	d.l.	d.l.	0.10	2.12	0.17	0.54
U	0.01	0.01	d.l.	d.l.	0.01	0.04	d.l.	0.00
Pb	0.06	0.01	0.06	0.30	0.92	0.07	d.l.	0.01
Rb	0.5	0.4	0.1	0.3	0.4	2.0	0.3	0.6
Cs	0.05	0.05	0.01	0.03	0.04	0.06	0.04	0.09
Sr	17	9	9	2	23	18	25	33
Sc	15.3	9.1	7.8	6.5	12.0	9.0	8.7	9.1

d.l.: below detection limit

Table 2-2. Continued.

Lac Supérieur			Mt. Trapeze					
Dunites			Harzburgites					
	OB93-64	OB93-82	OB93-280	OB93-284	OB93-287	OB93-289	OB93-291	OB93-297
<i>XRF wt. %</i>								
SiO ₂	43.25	44.08	48.58	47.41	44.70	48.52	47.45	45.36
Al ₂ O ₃	1.33	1.12	0.98	0.85	0.55	1.14	0.66	0.60
TiO ₂	0.04	0.05	0.01	0.01	0.01	0.02	0.02	0.02
FeO*	10.45	10.65	6.94	7.51	8.19	7.99	7.69	8.31
MnO	0.16	0.16	0.12	0.14	0.16	0.13	0.14	0.17
CaO	0.43	1.00	0.51	1.21	1.06	0.67	1.05	0.78
MgO	44.51	43.42	44.15	43.19	45.48	42.80	44.24	45.90
K ₂ O	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Na ₂ O	0.10	0.09	0.11	0.15	0.09	0.14	0.16	0.10
P ₂ O ₅	0.02	0.02	0.01	0.02	0.02	0.01	0.02	0.02
Total	100.28	100.59	101.41	100.48	100.26	101.42	101.42	101.26
Mg#	88.36	87.90	91.90	91.11	90.82	90.52	91.12	90.78
<i>XRF ppm</i>								
Ni	2462	2724	2273	2259	2499	2209	2305	2656
Cr	8716	3771	3772	1541	2631	2740	2162	2305
V	37	37	33	d.l.	d.l.	35	d.l.	d.l.
Sr	d.l.	23	d.l.	24	25	4	85	24
Zr	8	8	6	7	10	7	10	10
Ga	2	4	5	2	8	2	4	3
Cu	10	11	13	12	15	16	12	21
Zn	77	61	46	41	45	49	40	48
Nb	d.l.	d.l.	d.l.	d.l.	d.l.	d.l.	d.l.	d.l.
Rb	d.l.	d.l.	3	d.l.	d.l.	d.l.	d.l.	4
<i>ICP-MS ppm</i>								
La	0.31	0.17	0.20	0.19	0.28	0.11	0.41	0.32
Ce	0.09	0.42	0.36	0.40	0.49	0.21	0.64	0.54
Pr	0.02	0.06	0.04	0.05	0.05	0.03	0.06	0.07
Nd	0.08	0.28	0.16	0.21	0.18	0.09	0.19	0.25
Sm	0.02	0.12	0.05	0.06	0.05	0.03	0.05	0.07
Eu	0.02	0.05	0.02	0.02	0.01	0.01	0.02	0.01
Gd	0.03	0.15	0.04	0.04	0.04	0.03	0.04	0.05
Tb	0.01	0.03	0.01	0.01	0.01	0.01	0.01	0.01
Dy	0.04	0.15	0.04	0.06	0.05	0.04	0.04	0.06
Ho	0.01	0.03	0.01	0.01	0.01	0.01	0.01	0.01
Er	0.03	0.09	0.02	0.05	0.03	0.03	0.03	0.04
Tm	0.01	0.01	0.00	0.01	d.l.	0.01	d.l.	d.l.
Yb	0.04	0.10	0.02	0.05	0.03	0.04	0.03	0.03
Lu	0.01	0.02	d.l.	0.01	0.01	0.01	0.01	0.01
Ba	1	2	d.l.	1	1	2	1	d.l.
Th	0.03	0.04	0.04	0.03	0.06	0.03	0.28	0.06
Nb	0.02	0.07	0.19	0.20	0.41	0.14	0.22	0.36
Y	0.35	0.82	0.16	0.31	0.24	0.18	0.20	0.30
Hf	0.01	0.06	0.03	0.02	0.06	0.03	0.03	0.06
Ta	d.l.	0.01	0.01	0.01	0.02	0.01	0.01	0.02
U	0.01	0.01	0.01	0.02	0.06	0.01	0.09	0.02
Pb	0.19	0.36	0.16	0.21	0.15	0.12	0.26	2.04
Rb	0.1	0.2	0.7	0.7	0.4	0.8	0.4	2.2
Cs	d.l.	0.03	d.l.	0.01	d.l.	d.l.	0.01	0.01
Sr	0	24	1	25	24	3	88	23
Sc	4.8	10.0	7.6	9.9	5.2	8.1	7.2	3.6

Table 2-2. Continued.

Mt. Trapeze							
Harzburgites							
	OB93-299	OB93-305	OB93-306	OB93-307	OB93-310	OB93-314	OB93-317
<i>XRF wt. %</i>							
SiO ₂	47.90	46.96	41.55	48.29	46.00	47.60	47.12
Al ₂ O ₃	1.21	0.98	0.47	1.26	1.32	0.80	0.67
TiO ₂	0.00	0.02	0.03	0.00	0.00	0.01	0.01
FeO*	6.96	7.22	13.93	7.11	6.99	7.38	7.65
MnO	0.13	0.17	0.18	0.12	0.13	0.12	0.14
CaO	1.17	1.17	1.06	0.74	1.16	0.82	0.64
MgO	42.31	41.91	41.83	43.31	43.10	44.45	43.86
K ₂ O	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Na ₂ O	0.07	0.06	0.05	0.10	0.05	0.12	0.06
P ₂ O ₅	0.02	0.02	0.02	0.01	0.01	0.01	0.02
Total	99.77	98.51	99.12	100.95	98.76	101.31	100.16
<i>Mg#</i>	91.55	91.19	84.26	91.57	91.66	91.48	91.09
<i>XRF ppm</i>							
Ni	2167	2197	2436	2308	2010	2348	2252
Cr	3948	3208	308	2488	3985	2001	1668
V	44	d.l.	d.l.	d.l.	38	d.l.	d.l.
Sr	39	40	38	d.l.	d.l.	11	4
Zr	7	9	9	7	6	6	6
Ga	1	4	2	3	3	1	1
Cu	12	16	36	20	13	11	11
Zn	45	47	98	40	42	39	38
Nb	d.l.	d.l.	d.l.	d.l.	d.l.	d.l.	d.l.
Rb	d.l.	d.l.	d.l.	3	d.l.	d.l.	d.l.
<i>ICP-MS ppm</i>							
La	0.15	0.17	0.17	0.07	0.12	0.42	0.23
Ce	0.26	0.32	0.37	0.11	0.20	0.79	0.36
Pr	0.03	0.04	0.04	0.02	0.03	0.11	0.05
Nd	0.09	0.19	0.15	0.06	0.09	0.43	0.17
Sm	0.02	0.05	0.03	0.02	0.02	0.20	0.04
Eu	0.01	0.02	0.01	d.l.	d.l.	0.14	0.01
Gd	0.02	0.04	0.04	0.02	0.01	0.25	0.04
Tb	d.l.	0.01	0.01	d.l.	d.l.	0.04	0.01
Dy	0.02	0.06	0.06	0.02	0.02	0.20	0.05
Ho	d.l.	0.01	0.01	0.01	0.01	0.04	0.01
Er	0.01	0.03	0.04	0.02	0.01	0.10	0.02
Tm	d.l.	d.l.	0.01	d.l.	d.l.	0.02	d.l.
Yb	0.03	0.03	0.04	0.02	0.03	0.11	0.03
Lu	0.01	0.01	0.01	0.01	0.01	0.03	0.01
Ba		1	1	1	2	36	2
Th	0.03	0.02	0.02	0.01	0.04	0.27	0.04
Nb	0.18	0.29	0.38	0.12	0.30	0.28	0.24
Y	0.11	0.34	0.34	0.10	0.10	0.33	0.26
Hf	0.02	0.04	0.03	0.02	0.02	0.04	0.04
Ta	0.10	0.14	0.62	d.l.	0.52	0.02	0.16
U	0.14	0.03	0.03	0.01	0.01	0.02	0.01
Pb	0.07	0.02	0.03	0.09	0.06	0.19	0.03
Rb	0.4	0.7	0.7	0.7	0.4	0.5	0.8
Cs	d.l.	0.01	0.01	0.01	0.01	0.01	0.01
Sr	39	40	40	2	1	11	6
Sc	11.1	7.9	5.3	7.9	13.9	6.7	7.5

Table 2-3. Calculated modes.

Sample	Cpx	error	Opx	error	Olivine	error	Spinel	error
<i>wt.% mode</i>								
OB93-52	9.54	1.38	23.46	2.36	61.69	1.65	4.87	0.58
OB93-80	2.66	1.14	54.88	1.96	40.79	1.38	1.14	0.50
OB93-83	1.11	2.45	48.39	3.96	50.15	2.52	-0.29	0.89
OB93-85	1.91	0.85	49.17	1.45	48.25	1.01	0.35	0.36
OB93-99	2.93	0.86	40.21	1.50	56.88	1.04	-0.21	0.35
OB93-284	3.86	0.70	35.51	1.19	60.78	0.81	-0.48	0.30
OB93-287	3.59	0.78	20.71	1.30	75.11	0.89	0.34	0.31
OB93-289	0.38	0.38	44.90	0.65	54.62	0.44	0.03	0.16
OB93-291	3.14	0.78	33.96	1.33	62.75	0.86	-0.03	0.30
OB93-297	2.88	0.79	22.40	1.37	74.18	0.90	0.31	0.28
OB93-305	3.63	1.05	39.31	1.76	56.79	1.18	0.07	0.43
OB93-307	1.05	0.42	42.35	0.71	56.85	0.47	-0.39	0.18
OB93-310	3.30	0.57	32.59	0.95	63.62	0.64	0.62	0.24
OB93-314	2.45	0.56	35.46	0.96	62.16	0.66	-0.25	0.24
OB93-317	1.73	1.34	36.43	2.30	61.52	1.57	-0.07	0.55

Modes (in wt.%) are calculated by mass balance using average composition of minerals.
Error is 1 S.D.

Table 2-4. Clinopyroxene major element compositions

Lac Superieur											
Lherzolites Harzburgites											
Sample	OB93-52	OB93-51		OB93-79		OB93-80		OB93-81c		OB93-83	OB93-99
Grain ID	cpx1	cpx2	cpx4s	cpx1	cpx1s	cpx4	cpx1s	cpx2	cpx1s	cpx1s	cpx2
wt. %											
SiO ₂	52.02	55.28	54.54	54.39	54.32	54.00	54.32	54.53	51.91	54.48	54.68
TiO ₂	0.95	0.02	0.00	0.02	0.00	0.02	0.00	0.02	0.07	0.02	0.13
Al ₂ O ₃	5.41	2.05	1.61	1.85	2.08	1.90	2.08	1.83	5.24	1.91	2.35
FeO*	2.37	2.35	2.02	1.87	1.78	1.88	1.78	1.86	2.86	1.76	1.97
MnO	0.11	0.06	0.14	0.07	0.11	0.08	0.11	0.09	0.11	0.09	0.07
MgO	15.50	17.99	18.55	18.25	18.04	18.31	18.04	18.07	16.22	18.17	17.27
CaO	22.43	20.69	23.16	22.96	23.14	22.90	23.14	23.03	17.46	23.44	21.31
Na ₂ O	1.05	1.15	0.26	0.26	0.23	0.25	0.23	0.33	2.04	0.18	1.42
Cr ₂ O ₃	0.87	0.60	0.55	0.70	0.72	0.70	0.72	0.68	2.24	0.71	1.00
Total	100.71	100.18	100.83	100.37	100.43	100.04	100.43	100.44	98.15	100.75	100.19
Cations/6 Oxygens											
Si	1.877	1.986	1.958	1.959	1.956	1.953	1.956	1.963	1.909	1.956	1.970
Ti	0.026	0.000	0.000	0.000	0.000	0.000	0.000	0.001	0.002	0.001	0.003
Al	0.230	0.087	0.068	0.079	0.088	0.081	0.088	0.078	0.227	0.081	0.100
Fe	0.072	0.070	0.061	0.056	0.054	0.057	0.054	0.056	0.088	0.053	0.059
Mn	0.003	0.002	0.004	0.002	0.003	0.002	0.003	0.003	0.004	0.003	0.002
Mg	0.833	0.963	0.993	0.980	0.968	0.987	0.968	0.969	0.889	0.972	0.927
Ca	0.867	0.797	0.891	0.886	0.893	0.887	0.893	0.888	0.688	0.902	0.823
Na	0.073	0.080	0.018	0.018	0.016	0.017	0.016	0.023	0.145	0.013	0.099
Cr	0.025	0.017	0.015	0.020	0.021	0.020	0.021	0.019	0.065	0.020	0.028
Total	4.007	4.002	4.009	4.000	3.998	4.005	3.998	4.000	4.016	3.999	4.012
Mg#	92.09	93.18	94.25	94.56	94.76	94.54	94.76	94.54	91.01	94.85	93.99
Molar percentage of endmembers											
Wo	48.85	43.48	45.73	46.05	46.55	45.89	46.55	46.35	41.24	46.74	45.42
En	46.94	52.56	50.95	50.92	50.48	51.04	50.48	50.57	53.28	50.39	51.19
Fs	4.21	3.95	3.32	3.03	2.97	3.07	2.97	3.07	5.47	2.88	3.39

Each column represents one analysis. Different analyses of the same grain are noted in "Grain ID". The suffix "s" = symplectite cp.

Table 2-4. Continued.

Mt Trapeze													
Dunites		Pyroxenite				Harzburgites							
Sample	OB93-61c	OB93-64	OB93-82	OB93-94	OB93-281	OB93-282	OB93-284	OB93-287	OB93-289	OB93-291			
Grain ID	cpx1	cpx2	cpx2	cpx1	cpx3	cpx1	cpx1	cpx5s	cpx3	cpx7s	cpx1	cpx1	cpx1
wt. %													
SiO ₂	52.13	53.39	52.41	52.97	54.17	53.92	53.13	54.83	54.16	53.23	54.63		
TiO ₂	0.53	0.35	0.80	0.41	0.12	0.14	0.03	0.09	0.02	0.01	0.11		
Al ₂ O ₃	4.20	3.12	4.07	3.48	2.47	2.50	3.18	2.27	2.55	2.55	2.74		
FeO*	2.27	2.15	2.74	2.61	1.98	1.74	2.13	2.34	2.39	2.16	2.19		
MnO	0.08	0.07	0.08	0.11	0.04	0.05	0.04	0.09	0.06	0.07	0.08		
MgO	15.61	15.92	15.96	17.01	16.50	16.60	17.36	17.13	17.32	17.35	16.87		
CaO	23.48	23.91	21.94	22.41	22.80	22.24	22.33	21.69	22.82	23.28	21.40		
Na ₂ O	0.85	0.66	1.05	0.54	1.03	1.07	0.59	1.22	0.50	0.44	1.33		
Cr ₂ O ₃	1.07	0.86	0.83	1.13	1.19	1.64	1.34	1.19	0.94	1.08	1.42		
Total	100.22	100.43	99.88	100.66	100.31	99.90	100.13	100.84	100.75	100.18	100.76		
Cations/6 Oxygens													
Si	1.896	1.933	1.907	1.912	1.958	1.955	1.924	1.968	1.949	1.931	1.960		
Ti	0.015	0.009	0.022	0.011	0.003	0.004	0.001	0.002	0.000	0.000	0.003		
Al	0.180	0.133	0.175	0.148	0.105	0.107	0.136	0.096	0.108	0.109	0.116		
Fe	0.069	0.065	0.083	0.079	0.060	0.053	0.065	0.070	0.072	0.066	0.066		
Mn	0.002	0.002	0.002	0.003	0.001	0.002	0.001	0.003	0.002	0.002	0.002		
Mg	0.846	0.859	0.865	0.915	0.889	0.897	0.937	0.916	0.929	0.938	0.902		
Ca	0.915	0.928	0.855	0.867	0.883	0.864	0.866	0.834	0.880	0.905	0.823		
Na	0.060	0.046	0.074	0.038	0.072	0.075	0.041	0.085	0.035	0.031	0.092		
Cr	0.031	0.024	0.024	0.032	0.034	0.047	0.038	0.034	0.027	0.031	0.040		
Total	4.014	4.002	4.008	4.006	4.006	4.002	4.009	4.008	4.001	4.014	4.005		
Mg#	92.44	92.95	91.21	92.08	93.68	94.45	93.56	92.87	92.80	93.46	93.20		
Molar percentage of endmembers													
Wo	49.93	50.03	47.35	46.51	48.17	47.60	46.38	45.81	46.79	47.41	45.94		
En	46.16	46.34	47.91	49.10	48.49	49.41	50.16	50.33	49.38	49.15	50.38		
Fs	3.91	3.63	4.75	4.40	3.34	2.99	3.45	3.86	3.83	3.44	3.67		

Table 2-4. Continued.

Sample	OB93-297	OB93-302	OB93-305	OB93-307	OB93-309	OB93-310	OB93-314	OB93-317
Grain ID	cpx1	cpx2	cpx7	cpx3	cpx1	cpx3	cpx1	cpx1
wt. %								
SiO ₂	53.96	54.59	54.44	53.28	54.16	53.30	53.94	54.34
TiO ₂	0.26	0.02	0.01	0.00	0.05	0.00	0.00	0.05
Al ₂ O ₃	3.13	2.71	2.75	2.81	2.36	2.59	2.17	2.65
FeO*	2.33	2.01	2.02	1.85	2.08	1.98	1.87	1.98
MnO	0.10	0.05	0.05	0.07	0.06	0.11	0.09	0.11
MgO	16.70	16.96	16.74	17.26	16.71	17.76	17.68	17.06
CaO	21.68	22.48	22.13	23.70	22.09	23.72	23.25	22.43
Na ₂ O	1.27	1.12	1.10	0.10	1.17	0.00	0.41	0.92
Cr ₂ O ₃	1.60	1.07	1.11	1.05	1.26	0.88	0.98	1.47
Total	101.03	101.00	100.35	100.11	99.94	100.33	100.38	101.02
<i>Cations/6 Oxygens</i>								
Si	1.937	1.956	1.961	1.931	1.963	1.928	1.948	1.949
Ti	0.007	0.001	0.000	0.000	0.001	0.000	0.000	0.001
Al	0.132	0.114	0.117	0.120	0.101	0.110	0.092	0.112
Fe	0.070	0.060	0.061	0.056	0.063	0.060	0.056	0.059
Mn	0.003	0.001	0.002	0.002	0.002	0.003	0.003	0.003
Mg	0.894	0.906	0.899	0.932	0.902	0.958	0.951	0.912
Ca	0.834	0.863	0.854	0.920	0.858	0.919	0.900	0.862
Na	0.089	0.077	0.077	0.007	0.082	0.000	0.028	0.064
Cr	0.045	0.030	0.032	0.030	0.036	0.025	0.028	0.042
Total	4.011	4.010	4.002	3.998	4.009	4.004	4.006	4.005
Mg#	92.75	93.78	93.66	94.33	93.46	94.13	94.40	93.88
<i>Molar percentage of endmembers</i>								
Wo	46.40	47.20	47.09	48.22	47.05	47.47	47.16	47.01
En	49.71	49.52	49.55	48.84	49.48	49.45	49.88	49.74
Fs	3.89	3.29	3.36	2.94	3.46	3.09	2.96	3.24

Table 2-5. Orthopyroxene major element compositions.

Lac Supérieur										Mont Trapeze			
Lherzollites Harzburgites										Pyroxenites			
Sample	OB93-52	OB93-51	OB93-79	OB93-80	OB93-81c	OB93-83	OB93-85	OB93-99		OB93-82	OB93-94	OB93-281	OB98-282
Grain ID	opx 1	opx2	opx4	opx4	opx2	opx1	opx1	opx1		opx6ex	opx2ex	opx4.5	opx5
wt. %													
SiO ₂	56.14	56.92	57.66	57.13	57.60	56.25	57.70	56.77		56.69	56.22	57.47	58.24
TiO ₂	0.21	0.00	0.00	0.00	0.00	0.03	0.00	0.04		0.17	0.12	0.00	0.04
Al ₂ O ₃	2.78	1.77	1.78	1.81	1.67	3.42	1.84	1.51		2.35	2.78	1.96	1.53
FeO*	6.25	5.05	4.89	4.98	5.11	5.39	5.14	4.99		6.98	6.67	5.32	5.01
MnO	0.14	0.14	0.13	0.15	0.16	0.13	0.14	0.14		0.20	0.21	0.12	0.16
MgO	34.10	34.86	34.84	34.67	35.40	33.57	35.25	35.29		33.36	33.32	34.42	34.91
CaO	0.42	0.62	0.71	0.91	0.73	0.99	0.68	0.59		0.59	0.80	0.55	0.50
Na ₂ O	0.04	0.12	0.03	0.00	0.00	0.28	0.00	0.05		0.02	0.05	0.04	0.04
Cr ₂ O ₃	0.36	0.43	0.54	0.49	0.48	0.81	0.51	0.52		0.37	0.78	0.59	0.42
Total	100.43	99.90	100.58	100.14	101.15	100.88	101.25	99.88		100.75	100.95	100.46	100.84
<i>Cations/6 Oxygens</i>													
Si	1.930	1.957	1.965	1.959	1.956	1.923	1.957	1.953		1.948	1.930	1.964	1.978
Ti	0.005	0.000	0.000	0.000	0.000	0.001	0.000	0.001		0.004	0.003	0.000	0.001
Al	0.113	0.072	0.072	0.073	0.067	0.138	0.073	0.061		0.095	0.112	0.079	0.061
Fe	0.180	0.145	0.140	0.143	0.145	0.154	0.146	0.143		0.201	0.192	0.152	0.142
Mn	0.004	0.004	0.004	0.004	0.005	0.004	0.004	0.004		0.006	0.006	0.003	0.005
Mg	1.746	1.786	1.770	1.772	1.792	1.710	1.781	1.809		1.708	1.705	1.753	1.767
Ca	0.016	0.023	0.026	0.033	0.027	0.036	0.025	0.022		0.022	0.029	0.020	0.018
Na	0.003	0.008	0.002	0.000	0.000	0.019	0.000	0.004		0.002	0.003	0.003	0.002
Cr	0.010	0.012	0.015	0.013	0.013	0.022	0.014	0.014		0.010	0.021	0.016	0.011
Total	4.006	4.006	3.992	3.998	4.004	4.006	4.000	4.011		3.996	4.002	3.990	3.986
Mg#	90.68	92.48	92.69	92.55	92.51	91.73	92.44	92.65		89.49	89.90	92.02	92.55
<i>Molar percentage of endmembers</i>													
Wo	0.80	1.17	1.33	1.71	1.36	1.90	1.25	1.10		1.12	1.52	1.04	0.95
En	89.76	91.21	91.28	90.76	91.04	89.81	91.09	91.44		88.22	88.25	90.90	91.45
Fs	9.44	7.62	7.39	7.54	7.61	8.29	7.66	7.46		10.66	10.23	8.06	7.60

Each column represents one analysis. The suffix "ex" = opx exsolution from cpx.

Table 2-5. Continued.

Sample	OB93-284	OB93-287	OB93-289	OB93-291	OB93-297	OB93-302	OB93-305	OB93-307	OB93-309	OB93-310	OB93-314	OB93-317
Grain ID	opx1	opx1	opx2	opx1	opx2	opx1	opx1	opx1	opx1	opx2	opx2	opx1
wt. %												
SiO ₂	56.38	57.25	56.24	56.91	56.86	56.33	56.91	56.86	57.65	57.16	56.95	56.95
TiO ₂	0.00	0.02	0.00	0.00	0.00	0.03	0.00	0.00	0.00	0.00	0.00	0.00
Al ₂ O ₃	2.68	2.40	2.53	2.33	2.46	2.06	3.26	3.53	2.50	3.38	2.86	2.53
FeO*	5.38	5.45	5.93	5.29	5.30	5.19	5.42	5.10	5.34	5.21	5.30	5.17
MnO	0.11	0.12	0.13	0.18	0.15	0.09	0.16	0.17	0.16	0.16	0.13	0.13
MgO	33.87	34.07	33.15	34.05	34.20	34.29	33.82	33.85	34.50	33.81	34.30	34.78
CaO	0.82	0.78	1.18	0.57	0.55	0.59	0.82	0.65	0.84	0.99	0.66	0.67
Na ₂ O	0.03	0.02	0.00	0.07	0.03	0.07	0.05	0.00	0.05	0.00	0.00	0.03
Cr ₂ O ₃	0.67	0.59	0.67	0.55	0.68	0.59	0.72	0.74	0.54	0.65	0.62	0.74
Total	99.92	100.67	99.82	99.95	100.22	99.23	101.14	100.89	101.59	101.35	100.84	101.00
<i>Cations/6 Oxygens</i>												
Si	1.941	1.955	1.945	1.956	1.949	1.951	1.935	1.934	1.951	1.938	1.941	1.939
Ti	0.000	0.000	0.000	0.000	0.000	0.001	0.000	0.000	0.000	0.000	0.000	0.000
Al	0.109	0.097	0.103	0.094	0.099	0.084	0.131	0.141	0.100	0.135	0.115	0.101
Fe	0.155	0.156	0.172	0.152	0.152	0.150	0.154	0.145	0.151	0.148	0.151	0.147
Mn	0.003	0.003	0.004	0.005	0.004	0.003	0.004	0.005	0.005	0.005	0.004	0.004
Mg	1.738	1.734	1.708	1.744	1.747	1.770	1.714	1.716	1.740	1.708	1.742	1.764
Ca	0.030	0.028	0.044	0.021	0.020	0.022	0.030	0.024	0.030	0.036	0.024	0.024
Na	0.002	0.001	0.000	0.005	0.002	0.004	0.003	0.000	0.003	0.000	0.000	0.002
Cr	0.018	0.016	0.018	0.015	0.018	0.016	0.019	0.020	0.015	0.017	0.017	0.020
Total	3.997	3.989	3.994	3.992	3.992	4.001	3.991	3.985	3.994	3.986	3.993	4.002
Mg#	91.82	91.77	90.88	91.98	92.00	92.17	91.75	92.21	92.00	92.04	92.02	92.31
<i>Molar percentage of endmembers</i>												
Wo	1.57	1.49	2.28	1.10	1.05	1.12	1.56	1.26	1.59	1.90	1.26	1.26
En	90.38	90.40	88.80	90.97	91.04	91.14	90.32	91.05	90.55	90.30	90.86	91.14
Fs	8.05	8.11	8.91	7.93	7.91	7.74	8.12	7.70	7.87	7.81	7.88	7.60

Table 2-6. Olivine major element compositions

Lac Supérieur														
		Harzburgites					Dunites							
Sample		OB93-52	OB93-51	OB93-79	OB93-80	OB93-81c	OB93-83		OB93-85		OB93-99	OB93-61c	OB93-64	OB93-82
Grain		ol3	ol8	ol3	ol6	ol1	ol8	ol2	ol1	ol3	ol7	ol3	ol3	ol0
wt.%														
SiO ₂		40.36	41.00	41.43	41.22	41.03	40.91	40.41	40.76	41.55	40.93	40.40	40.30	
FeO		9.57	8.10	8.00	8.05	8.57	8.79	10.12	8.31	8.11	10.01	10.43	11.14	
MnO		0.15	0.09	0.11	0.11	0.06	0.12	0.13	0.13	0.13	0.17	0.14	0.17	
MgO		49.32	50.74	50.70	50.90	51.21	50.59	49.21	51.08	49.98	49.40	49.38	48.25	
CaO		0.03	0.06	0.09	0.07	0.06	0.08	0.10	0.06	0.06	0.07	0.10	0.04	
NiO		0.34	0.33	0.32	0.33	0.36	0.32	0.36	0.35	0.35	0.29	0.30	0.34	
Al ₂ O ₃		0.00	0.00	0.00	0.03	0.00	0.00	0.03	0.03	0.03	0.04	0.00	0.04	
Cr ₂ O ₃		0.01	0.00	0.03	0.01	0.02	0.00	0.01	0.00	0.04	0.02	0.10	0.02	
Total		99.78	100.32	100.68	100.73	101.30	100.81	100.37	100.72	100.23	100.92	100.85	100.29	
Cations/4 Oxygens														
Si		0.992	0.995	1.001	0.996	0.989	0.992	0.991	0.988	1.008	0.996	0.987	0.993	
Fe		0.197	0.164	0.162	0.163	0.173	0.178	0.207	0.168	0.164	0.204	0.213	0.229	
Mn		0.003	0.002	0.002	0.002	0.001	0.002	0.003	0.003	0.003	0.003	0.003	0.004	
Mg		1.807	1.835	1.825	1.833	1.840	1.828	1.797	1.844	1.807	1.791	1.798	1.772	
Ca		0.001	0.002	0.002	0.002	0.001	0.002	0.003	0.002	0.001	0.002	0.002	0.001	
Ni		0.007	0.006	0.006	0.006	0.007	0.006	0.007	0.007	0.007	0.006	0.006	0.007	
Al		0.000	0.000	0.000	0.001	0.000	0.000	0.001	0.001	0.001	0.001	0.000	0.001	
Cr		0.000	0.000	0.001	0.000	0.000	0.000	0.000	0.000	0.001	0.000	0.002	0.000	
Total		3.007	3.005	2.999	3.003	3.011	3.008	3.009	3.012	2.991	3.003	3.012	3.006	
Fo		90.18	91.78	91.86	91.84	91.42	91.11	89.65	91.64	91.66	89.79	89.40	88.53	
Each column represents one analysis.														

Each column represents one analysis.

Table 2-6. Continued.

Mont Trapeze																			
Harzburgites																			
Sample	OB93-281			OB93-282		OB93-284		OB93-287		OB93-289		OB93-291		OB93-297		OB93-302		OB93-305	
Grain	ol1	ol4	ol2	ol1	ol1	ol1	ol1	ol1	ol1	ol2	ol1	ol1	ol1	ol1	ol1	ol1	ol1	ol2	ol2
wt. %																			
SiO ₂	40.61	40.35	40.77	40.97	40.97	40.67	40.67	40.70	40.70	40.72	40.76	40.93	40.93	41.05	41.05	40.95	40.95	40.95	40.95
FeO	8.00	9.69	7.57	8.58	8.58	8.58	8.58	8.93	8.93	10.43	8.74	8.81	8.81	8.41	8.41	8.28	8.28	8.28	8.28
MnO	0.13	0.15	0.11	0.09	0.09	0.08	0.08	0.10	0.10	0.11	0.09	0.10	0.10	0.12	0.12	0.13	0.13	0.13	0.13
MgO	50.24	49.16	50.29	49.66	49.66	49.66	49.66	49.97	49.97	48.67	50.09	49.73	49.73	49.99	49.99	50.10	50.10	50.10	50.10
CaO	0.08	0.08	0.05	0.09	0.09	0.04	0.04	0.03	0.03	0.05	0.06	0.05	0.05	0.03	0.03	0.04	0.04	0.04	0.04
NiO	0.33	0.28	0.34	0.50	0.50	0.45	0.45	0.53	0.53	0.50	0.44	0.57	0.57	0.46	0.46	0.46	0.46	0.46	0.46
Al ₂ O ₃	0.04	0.00	0.00	0.02	0.02	0.02	0.02	0.00	0.00	0.03	0.06	0.02	0.02	0.15	0.15	0.00	0.00	0.00	0.00
Cr ₂ O ₃	0.03	0.12	0.00	0.00	0.00	0.02	0.02	0.00	0.00	0.00	0.00	0.00	0.00	0.02	0.02	0.00	0.00	0.00	0.00
Total	99.46	99.83	99.13	99.90	99.90	99.51	99.51	100.26	100.26	100.51	100.24	100.21	100.21	100.22	100.22	99.96	99.96	99.96	99.96
Cations/4 Oxygens																			
Si	0.994	0.992	0.999	1.001	1.001	0.998	0.998	0.993	0.993	0.997	0.994	0.998	0.998	0.999	0.999	0.999	0.999	0.999	0.999
Fe	0.164	0.199	0.155	0.175	0.175	0.176	0.176	0.182	0.182	0.214	0.178	0.180	0.180	0.171	0.171	0.169	0.169	0.169	0.169
Mn	0.003	0.003	0.002	0.002	0.002	0.002	0.002	0.002	0.002	0.002	0.002	0.002	0.002	0.003	0.003	0.003	0.003	0.003	0.003
Mg	1.833	1.802	1.836	1.808	1.808	1.816	1.816	1.818	1.818	1.777	1.820	1.808	1.808	1.812	1.812	1.821	1.821	1.821	1.821
Ca	0.002	0.002	0.001	0.002	0.002	0.001	0.001	0.001	0.001	0.001	0.002	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001
Ni	0.007	0.006	0.007	0.010	0.010	0.009	0.009	0.010	0.010	0.010	0.009	0.011	0.011	0.009	0.009	0.009	0.009	0.009	0.009
Al	0.001	0.000	0.000	0.001	0.001	0.001	0.001	0.000	0.000	0.001	0.002	0.001	0.001	0.004	0.004	0.000	0.000	0.000	0.000
Cr	0.001	0.002	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Total	3.005	3.006	3.001	2.999	2.999	3.002	3.002	3.007	3.007	3.002	3.005	3.001	3.001	2.999	2.999	3.001	3.001	3.001	3.001
Fo	91.80	90.04	92.21	91.16	91.16	91.16	91.16	90.89	90.89	89.26	91.08	90.95	90.95	91.38	91.38	91.51	91.51	91.51	91.51

Table 2-6. Continued.

Sample	OB93-307	OB93-309	OB93-310	OB93-314	OB93-317
Grain	oil	oil	oil	oil	oil
wt. %					
SiO ₂	40.95	40.72	40.89	40.83	41.06
FeO	8.43	8.61	8.51	8.30	7.98
MnO	0.10	0.08	0.12	0.09	0.09
MgO	49.90	49.65	49.54	50.23	50.35
CaO	0.04	0.06	0.10	0.05	0.02
NiO	0.53	0.50	0.50	0.52	0.57
Al ₂ O ₃	0.05	0.03	0.03	0.02	0.03
Cr ₂ O ₃	0.00	0.00	0.00	0.01	0.00
Total	99.99	99.64	99.69	100.05	100.10
<i>Cations/4 Oxygens</i>					
Si	0.999	0.998	1.001	0.996	0.999
Fe	0.172	0.176	0.174	0.169	0.162
Mn	0.002	0.002	0.002	0.002	0.002
Mg	1.814	1.813	1.807	1.825	1.825
Ca	0.001	0.002	0.003	0.001	0.001
Ni	0.010	0.010	0.010	0.010	0.011
Al	0.001	0.001	0.001	0.000	0.001
Cr	0.000	0.000	0.000	0.000	0.000
Total	3.000	3.002	2.998	3.004	3.001
Fo	91.34	91.13	91.21	91.52	91.84

Table 2-7. Spinel major element compositions.

Lac Supérieur														
Lherzolite		Harzburgites			OB93-78		OB93-79	OB93-80	OB93-81c	OB93-83			OB93-85	
Sample	Grain ID	OB93-52	OB93-51	OB93-78		OB93-79	OB93-80	OB93-81c	OB93-83			OB93-85		
		sp2 n=2	sp5 n=5	sp1 n=3	sp3 n=3	sp1 n=6	sp6 n=3	sp1 pt1	sp1 pt3	sp2 pt2	sp2 pt3	sp10, disc.	sp0 n=2	sp1 n=3
wt. %														
SiO ₂		0.01	0.08	0.06	0.07	0.13	0.06	0.00	0.00	0.00	0.00	0.00	0.00	0.00
TiO ₂		0.48	0.09	0.08	0.06	0.06	0.12	0.12	0.14	0.18	0.10	0.14	0.03	0.05
Al ₂ O ₃		42.99	24.95	27.25	27.37	26.87	26.81	34.72	28.39	28.72	41.49	21.46	26.01	27.55
Cr ₂ O ₃		21.54	42.65	40.78	39.92	40.70	42.31	29.91	36.52	35.24	22.68	42.69	43.73	42.00
FeO*		14.23	14.72	14.02	13.35	13.69	13.40	15.00	16.18	17.29	14.88	18.70	12.05	11.95
MnO		0.09	0.13	0.14	0.16	0.12	0.14	0.23	0.26	0.25	0.24	0.29	0.29	0.26
MgO		18.89	17.18	17.40	17.68	17.30	17.12	19.22	17.76	17.19	19.91	15.98	17.01	17.33
CaO		0.01	0.05	0.02	0.05	0.10	0.04	0.06	0.08	0.00	0.00	0.00	0.00	0.06
NiO		0.26	0.20	0.18	0.16	0.15	0.16	0.31	0.33	0.34	0.38	0.28	0.26	0.26
Total		98.50	100.03	99.92	98.81	99.13	100.16	99.56	99.66	99.20	99.67	99.53	99.40	99.45
FeO		9.63	9.63	9.62	8.80	9.46	10.06	7.62	8.83	9.75	7.65	10.57	9.48	9.23
Fe ₂ O ₃		5.11	5.66	4.90	5.06	4.70	3.71	8.19	8.16	8.37	8.04	9.04	2.86	3.01
New Total		99.02	100.60	100.41	99.32	99.60	100.53	100.38	100.48	100.04	100.47	100.44	99.68	99.75
Cations/4 Oxygens														
Si		0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Ti		0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Al		1.42	0.88	0.95	0.96	0.95	0.94	1.18	1.00	1.01	1.37	0.79	0.92	0.96
Cr		0.48	1.01	0.96	0.94	0.96	0.99	0.68	0.86	0.84	0.50	1.05	1.03	0.98
Fe		0.33	0.37	0.35	0.33	0.34	0.33	0.36	0.40	0.43	0.35	0.49	0.30	0.30
Mn		0.00	0.00	0.00	0.00	0.00	0.00	0.01	0.01	0.01	0.01	0.01	0.01	0.01
Mg		0.79	0.77	0.77	0.79	0.77	0.76	0.83	0.79	0.77	0.83	0.74	0.76	0.77
Ca		0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Ni		0.01	0.00	0.00	0.00	0.00	0.00	0.01	0.01	0.01	0.01	0.01	0.01	0.01
Cation Total		3.04	3.05	3.04	3.04	3.04	3.03	3.07	3.07	3.07	3.06	3.08	3.02	3.03
Oxygen Total		3.95	3.94	3.95	3.94	3.95	3.96	3.91	3.91	3.91	3.92	3.90	3.97	3.97
Fe ²⁺		0.22	0.24	0.24	0.22	0.23	0.25	0.18	0.22	0.24	0.18	0.27	0.24	0.23
Fe ³⁺		0.11	0.13	0.11	0.11	0.10	0.08	0.17	0.18	0.18	0.17	0.21	0.06	0.07
New Cat. Total		3.04	3.04	3.04	3.04	3.04	3.03	3.06	3.06	3.06	3.06	3.07	3.02	3.02
New O Total		4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00
Cr#		25.16	53.41	50.09	49.52	50.39	51.43	36.62	46.32	45.15	26.83	57.16	53.01	50.57
Mg#Fe ²⁺		78.00	76.37	76.57	77.42	76.76	75.41	82.12	78.57	76.28	82.57	73.44	76.33	77.13

Each column represents one analysis, except where average analysis is noted by n=x, and x=number of averaged points.
 Different analyses of the same grain are noted in "Grain ID". The suffix "s" = symplectite spinel; and "disc." = discrete spinel.

Table 2-7. Continued.

Sample Grain ID	Mt. Trappe									
	Dunites					Pyroxenites		Harzburgites		
	OB93-99	OB93-61c	OB93-64	OB93-82	OB93-94	OB93-284	OB93-287	OB93-289		
sp3	sp5	sp5	sp5	sp2	sp3	sp10	sp11, disc.	sp6	sp12	
n=2	n=5		n=2	n=3		n=2				
wt. %										
SiO ₂	0.00	0.03	0.03	0.04	0.00	0.00	0.00	0.00	0.00	
TiO ₂	0.29	0.60	0.58	0.97	0.60	0.57	0.29	0.09	0.11	
Al ₂ O ₃	25.71	30.97	25.11	36.24	29.94	31.32	21.57	27.03	26.52	
Cr ₂ O ₃	42.92	30.20	35.08	27.11	35.73	36.81	46.12	40.84	40.63	
FeO*	12.63	19.97	22.87	17.16	15.64	10.77	15.23	15.48	17.24	
MnO	0.27	0.21	0.21	0.14	0.28	0.24	0.30	0.30	0.27	
MgO	16.68	16.35	14.38	17.30	16.61	20.29	15.39	15.89	15.07	
CaO	0.01	0.01	0.00	0.01	0.00	0.01	0.01	0.01	0.02	
NiO	0.24	0.25	0.24	0.27	0.28	0.19	0.25	0.23	0.30	
Total	98.76	98.94	98.57	99.25	99.08	100.21	99.15	99.86	100.16	
FeO	10.00	11.75	13.83	11.67	11.16	6.19	11.43	11.60	12.84	
Fe ₂ O ₃	2.92	9.14	10.05	6.10	4.98	5.10	4.23	4.31	4.89	
New Total	99.05	99.49	99.49	99.86	99.58	100.72	99.57	100.29	100.65	
Cations/4 Oxygens										
Si	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	
Ti	0.01	0.01	0.01	0.02	0.01	0.01	0.01	0.00	0.00	
Al	0.91	1.10	0.93	1.24	1.05	1.06	0.79	0.96	0.94	
Cr	1.02	0.72	0.87	0.62	0.84	0.83	1.13	0.97	0.97	
Fe	0.32	0.50	0.60	0.42	0.39	0.26	0.39	0.39	0.43	
Mn	0.01	0.01	0.01	0.00	0.01	0.01	0.01	0.01	0.01	
Mg	0.75	0.73	0.67	0.75	0.74	0.87	0.71	0.71	0.68	
Ca	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	
Ni	0.01	0.01	0.01	0.01	0.01	0.00	0.01	0.01	0.01	
Cation Total	3.02	3.08	3.09	3.05	3.04	3.04	3.04	3.04	3.04	
Oxygen Total	3.97	3.90	3.89	3.93	3.94	3.95	3.95	3.95	3.95	
Fe ²⁺	0.25	0.29	0.35	0.28	0.27	0.15	0.29	0.29	0.32	
Fe ³⁺	0.07	0.20	0.23	0.13	0.11	0.11	0.10	0.10	0.11	
New Cat. Total	3.02	3.02	3.07	3.04	3.04	3.04	3.03	3.03	3.04	
New O Total	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	
Cr#	52.82	39.55	48.38	33.42	44.46	44.09	58.92	50.34	50.68	
Mg#Fe ²⁺	74.98	71.78	65.61	72.87	72.90	85.56	70.85	71.18	67.97	

Table 2-7. Continued.

Sample Grain ID	OB93-291		OB93-297		OB93-302		OB93-305		OB93-307		OB93-309		OB93-310		OB93-314		OB93-317	
	sp6	sp11	sp1	sp8	sp1	sp8	sp4	sp1	sp8	sp6	sp5	sp8	sp1	sp5	sp1	sp5	sp11	sp11
	n=2	n=2	n=2	n=2	n=2	n=2	n=2	n=2	n=2	n=2	n=2	n=3	n=2	n=2	n=2	n=4	n=2	n=2
wt %																		
SiO ₂	0.06	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
TiO ₂	0.20	0.23	0.53	0.10	0.14	0.17	0.14	0.17	0.04	0.20	0.25	0.04	0.09	0.08	0.09	0.08	0.15	0.15
Al ₂ O ₃	23.74	25.41	25.69	28.34	26.33	27.71	26.33	27.71	35.08	20.82	25.93	33.70	29.06	31.24	29.06	31.24	24.85	24.85
Cr ₂ O ₃	44.89	42.81	42.59	40.38	41.89	40.84	41.89	40.84	33.78	46.72	41.25	34.96	40.60	38.23	40.60	38.23	44.08	44.08
FeO*	14.64	14.98	14.81	13.29	14.80	14.53	14.80	14.53	11.49	16.36	16.22	11.85	12.83	12.15	12.83	12.15	13.92	13.92
MnO	0.27	0.31	0.28	0.29	0.26	0.28	0.26	0.28	0.25	0.33	0.29	0.23	0.30	0.25	0.30	0.25	0.29	0.29
MgO	16.19	15.93	16.50	17.14	16.08	16.47	16.08	16.47	18.46	14.82	15.88	18.21	17.29	17.82	17.29	17.82	16.26	16.26
CaO	0.07	0.01	0.00	0.03	0.03	0.03	0.03	0.03	0.02	0.02	0.01	0.00	0.03	0.01	0.03	0.01	0.02	0.02
NiO	0.25	0.25	0.25	0.27	0.23	0.23	0.23	0.23	0.30	0.24	0.30	0.32	0.25	0.27	0.25	0.27	0.25	0.25
Total	100.31	99.94	100.65	99.84	99.76	100.27	99.76	100.27	99.42	99.51	100.12	99.31	100.43	100.04	100.43	100.04	99.82	99.82
FeO	10.87	11.38	11.09	9.83	11.21	11.00	11.21	11.00	8.69	12.18	11.62	8.86	9.90	9.35	9.90	9.35	10.69	10.69
Fe ₂ O ₃	4.19	4.00	4.14	3.85	3.99	3.93	3.99	3.93	3.12	4.64	5.12	3.32	3.26	3.11	3.26	3.11	3.59	3.59
New Total	100.73	100.34	101.07	100.22	100.16	100.66	100.16	100.66	99.74	99.97	100.63	99.65	100.76	100.35	100.76	100.35	100.18	100.18
Cations/4 Oxygens																		
Si	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Ti	0.00	0.01	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Al	0.84	0.90	0.90	0.99	0.93	0.97	0.93	0.97	1.18	0.76	0.92	1.15	1.00	1.07	1.00	1.07	0.88	0.88
Cr	1.07	1.02	1.00	0.94	0.99	0.96	0.99	0.96	0.76	1.15	0.98	0.80	0.94	0.88	0.94	0.88	1.05	1.05
Fe	0.37	0.38	0.37	0.33	0.37	0.36	0.37	0.36	0.28	0.42	0.41	0.29	0.31	0.29	0.31	0.29	0.35	0.35
Mn	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
Mg	0.73	0.72	0.73	0.76	0.72	0.73	0.72	0.73	0.79	0.69	0.71	0.78	0.75	0.77	0.75	0.77	0.73	0.73
Ca	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Ni	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
Cation Total	3.04	3.03	3.03	3.03	3.03	3.03	3.03	3.03	3.03	3.04	3.04	3.03	3.03	3.03	3.03	3.03	3.03	3.03
Oxygen Total	3.95	3.96	3.95	3.96	3.96	3.96	3.96	3.96	3.97	3.95	3.94	3.96	3.96	3.97	3.96	3.97	3.96	3.96
Fe ²⁺	0.27	0.28	0.27	0.24	0.28	0.27	0.28	0.27	0.21	0.31	0.29	0.21	0.24	0.22	0.24	0.22	0.27	0.27
Fe ³⁺	0.09	0.09	0.09	0.08	0.09	0.09	0.09	0.09	0.07	0.11	0.11	0.07	0.07	0.07	0.07	0.07	0.08	0.08
New Cat. Total	3.03	3.03	3.03	3.03	3.03	3.03	3.03	3.03	3.02	3.03	3.04	3.02	3.02	3.02	3.02	3.02	3.03	3.03
New O Total	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00	4.00
Cr#	55.92	53.06	52.66	48.87	51.62	49.72	51.62	49.72	39.24	60.09	51.63	41.04	48.38	45.09	48.38	45.09	54.34	54.34
Mg#Fe ²⁺	72.87	71.62	72.85	75.86	72.11	72.97	72.11	72.97	79.25	68.73	71.20	78.70	75.85	77.41	75.85	77.41	73.25	73.25

Table 2-8. Clinopyroxene trace element compositions.

Lac Superieur Lherzolites Harzburgites													
Sample	OB93-52	OB93-51			OB93-53			OB93-56			OB93-58		
Grain ID	cpx1	cpx1	cpx2	cpx3s	cpx4s	avg	I S.D.	avg	I S.D.	avg	avg	I S.D.	OB93-74 cpx3
ppm													
La	2.0	16.4	23.0	9.7	0.72	0.04	0.01	0.12	0.02	0.07	0.07	0.04	0.49
Ce	6.4	31.1	45.9	14.0	0.64	0.05	0.01	0.30	0.06	0.08	0.08	0.03	1.29
Nd	6.1	7.4	11.4	2.1	0.098	0.051	0.009	0.218	0.027	0.085	0.085	0.044	0.801
Sm	2.4	1.0	1.5	0.3	0.030	0.034	0.005	0.095	0.053	0.096	0.096	0.071	0.278
Eu	0.9	0.3	0.5	0.1	0.015	0.010	0.003	0.036	0.012	0.022	0.022	0.006	0.105
Dy	3.2	0.4	0.6	0.1	0.033	0.032	0.005	0.096	0.032	0.063	0.063	0.054	0.360
Er	2.0	0.2	0.3	0.1	0.034	0.039	0.003	0.069	0.019	0.078	0.078	0.011	0.290
Yb	2.1	0.2	0.2	0.1	0.086	0.081	0.017	0.117	0.053	0.087	0.087	0.030	0.233
Ti	5290	32	36	36	23	45	3	33	5	n.d	35	2	254
V	287	114	114	157	98	131	3	114	6	n.d	125	8	145
Cr	6093	4364	4200	4267	3134	4511	312	4003	574	n.d	4010	524	4703
Sr	126	322	413	273	4	1	0	3	0	n.d	1	0	10
Y	17	2	2	1	0	0	0	0	0	n.d	0	0	1
Zr	30	3	9	1	0.4	0.7	0.4	0.9	0.1	n.d	0.6	0.2	3.9
CI Normalized													
La	8.6	69.9	98.0	41.2	3.1	0.2		0.5		0.3	0.1		2.1
Ce	10.6	51.6	76.1	23.2	1.1	0.1		0.5		0.1	0.1		2.1
Nd	13.5	16.3	25.2	4.7	0.2	0.1		0.5		0.2	0.1		1.8
Sm	16.0	6.5	10.4	2.2	0.2	0.2		0.6		0.6	0.3		1.9
Eu	16.0	5.1	8.4	1.7	0.3	0.2		0.6		0.4	0.3		1.9
Dy	13.3	1.5	2.5	0.4	0.1	0.1		0.4		0.3	0.1		1.5
Er	12.4	1.0	1.7	0.7	0.2	0.2		0.4		0.5	0.3		1.8
Yb	12.6	1.1	1.3	0.6	0.5	0.5		0.7		0.5	0.5		1.4
Ti	12.1	0.1	0.1	0.1	0.1	0.1		0.1			0.1		0.6
Sr	16.2	41.3	52.9	35.0	0.5	0.1		0.4			0.2		1.3
Zr	7.6	0.7	2.3	0.3	0.1	0.2		0.2			0.1		1.0
(La/Yb) _N	0.68	66.02	72.71	70.40	5.81	0.38		0.71		0.59	0.28		1.44
(La/Sm) _N	0.54	10.81	9.42	19.04	14.92	0.81		0.78		0.48	0.41		1.10

All columns represent a single analysis, unless noted by the suffix "avg", where n=x is the number of averaged points. The suffix "s" denotes symplectite cpx.

Table 2-8. continued.

Sample Grain ID	OB93-77 cpx4	OB93-78 avg n=3	OB93-78 1 S.D.	OB93-79 cpx	OB93-80 avg 1 S.D.	OB93-81c cpx	OB93-83 cpx7	OB93-83 cpx1s	OB93-85 cpx	OB93-86 avg n=3	OB93-86 1 S.D.
<i>ppm</i>											
La	0.18	0.02	0.01	0.05	0.06	0.02	37.70	36.60	0.07	0.10	0.04
Ce	0.49	0.02	0.01	0.07	0.14	0.04	135.00	129.00	0.11	0.17	0.04
Nd	0.538	0.028	0.014	0.069	0.126	0.041	110.000	92.100	0.133	0.255	0.004
Sm	0.258	0.038	0.016	0.059	0.075	0.035	32.100	25.700	0.052	0.154	0.023
Eu	0.088	0.008	0.001	0.017	0.025	0.005	10.100	8.200	0.015	0.084	0.014
Dy	0.306	0.019	0.007	0.041	0.059	0.023	17.300	15.000	0.041	0.139	0.033
Er	0.155	0.038	0.018	0.052	0.058	0.018	6.950	6.430	0.037	0.195	0.022
Yb	0.239	0.040	0.019	0.133	0.109	0.026	5.610	5.270	0.123	0.178	0.039
Ti	113	n.d.		61	n.d.		646		26	99	5
V	136	n.d.		134	n.d.		126		107	128	7
Cr	7766	n.d.		4064	n.d.		8482		3835	4004	557
Sr	4	n.d.		1	n.d.		655		2	2	0
Y	1	n.d.		0	n.d.		74		0	1	0
Zr	3.2	n.d.		0	n.d.		1372		0.4	2.1	0.2
<i>CI Normalized</i>											
La	0.8	0.1		0.2	0.3		160.6	155.9	0.3	0.4	
Ce	0.8	0.0		0.1	0.2		223.8	213.9	0.2	0.3	
Nd	1.2	0.1		0.2	0.3		243.1	203.6	0.3	0.6	
Sm	1.8	0.3		0.4	0.5		218.2	174.7	0.4	1.0	
Eu	1.6	0.2		0.3	0.4		180.4	146.4	0.3	1.5	
Dy	1.3	0.1		0.2	0.2		71.3	61.8	0.2	0.6	
Er	1.0	0.2		0.3	0.4		43.7	40.5	0.2	1.2	
Yb	1.5	0.2		0.8	0.7		34.5	32.4	0.8	1.1	
Ti	0.3			0.1			1.5		0.1	0.2	
Sr	0.5			0.2			84.0		0.2	0.3	
Zr	0.8						348.2		0.1	0.5	
(La/Yb) _N	0.53	0.35		0.26	0.41		4.65	4.81	0.37	0.39	
(La/Sm) _N	0.45	0.34		0.52	0.54		0.74	0.89	0.78	0.41	

Table 2-8. continued.

Sample Grain ID	OB93-89			OB93-95			OB93-98			OB93-99			Dunites			OB93-82		
	avg n=7	1 S.D.		cpx2	avg other n=6	1 S.D.	avg n=6	1 S.D.		cpx1s	cpx2	cpx1	OB93-61c	cpx1		cpx2	cpx2	pt2
<i>ppm</i>																		
La	0.11	0.03		0.01	0.19	0.03	0.06	0.00		0.68	9.85	1.61	0.84			0.84		4.6
Ce	0.26	0.04		0.02	0.58	0.13	0.16	0.04		1.54	16.50	5.34	2.61			2.61		11.7
Nd	0.238	0.036		0.091	0.446	0.046	0.109	0.034		1.250	3.690	3.720	2.300			2.300		5.2
Sm	0.162	0.043		0.156	0.308	0.154	0.089	0.044		0.483	0.615	1.060	1.010			1.010		1.98
Eu	0.050	0.014		0.026	0.088	0.018	0.027	0.012		0.185	0.250	0.353	0.349			0.349		0.52
Dy	0.156	0.016		0.071	0.288	0.113	0.071	0.021		0.540	0.563	1.500	1.390			1.390		1.99
Er	0.103	0.022		0.052	0.132	0.020	0.058	0.021		0.313	0.381	0.972	0.889			0.889		1.37
Yb	0.156	0.018		0.038	0.209	0.056	0.112	0.042		0.304	0.360	0.978	0.754			0.754		0.96
Ti	75	4			115	1616	30	4		619	733	2881	1764			1764		4390
V	124	7			129	63	101	7		166	168	256	157			157		272
Cr	3848	408			4326	340	3628	319		7035	6045	6884	6996			6996		5520
Sr	3	0			5	21	2	0		16	138	87	35			35		80
Y	1	0			1	5	0	0		3	3	11	8			8		18
Zr	1.4	0.3			2	12	0.9	0.2		6.0	8.5	40.4	22.8			22.8		48.4
<i>CI Normalized</i>																		
La	0.5			0.1	0.8		0.3			2.9	42.0	6.9	3.6			3.6		19.6
Ce	0.4			0.0	1.0		0.3			2.6	27.4	8.9	4.3			4.3		19.5
Nd	0.5			0.2	1.0		0.2			2.8	8.2	8.2	5.1			5.1		11.5
Sm	1.1			1.1	2.1		0.6			3.3	4.2	7.2	6.9			6.9		13.5
Eu	0.9			0.5	1.6		0.5			3.3	4.5	6.3	6.2			6.2		9.4
Dy	0.6			0.3	1.2		0.3			2.2	2.3	6.2	5.7			5.7		8.2
Er	0.6			0.3	0.8		0.4			2.0	2.4	6.1	5.6			5.6		8.6
Yb	1.0			0.2	1.3		0.7			1.9	2.2	6.0	4.6			4.6		5.9
Ti	0.2									1.4	1.7	6.6	4.0			4.0		10.1
Sr	0.4									2.1	17.7	11.1	4.5			4.5		10.2
Zr	0.4									1.5	2.2	10.3	5.8			5.8		12.3
(La/Yb) _N	0.49			0.27	0.62		0.39			1.54	18.94	1.14	0.77			0.77		3.30
(La/Sm) _N	0.43			0.06	0.38		0.45			0.88	10.04	0.95	0.52			0.52		1.45

Table 2-8. continued.

Sample Grain ID	Pyroxenites				Mt Trapeze Harzburgites				OB93-282			
	OB93-75	OB93-81b	OB93-87	OB93-94	OB93-100	OB93-102	OB93-280	OB93-281	OB93-281	OB93-281	OB93-282	OB93-282
	grl	grl	grl	cpx1	cpx3	cpx1	cpx1	cpx1	cpx3	cpx1	cpx1	cpx8s
ppm												
La	1.06	4.21	0.86	2.59	5.97	0.45	0.75	2.53	9.67	1.52	0.41	0.39
Ce	3.26	11.04	2.90	7.22	14.80	1.89	2.90	5.02	17.83	1.39	0.86	0.72
Nd	2.894	7.282	1.895	5.080	7.780	2.025	2.886	2.524	4.584	0.266	0.264	0.350
Sm	1.423	1.656	0.980	1.580	1.620	1.096	1.285	0.844	0.737	0.085	0.119	0.224
Eu	0.496	0.480	0.318	0.488	0.477	0.563	0.647	0.248	0.304	0.021	0.055	0.123
Dy	1.735	1.322	0.986	1.330	1.260	2.112	2.141	0.628	0.658	0.151	0.157	0.621
Er	1.141	1.011	0.725	0.794	0.719	0.989	1.272	0.259	0.295	0.170	0.162	0.451
Yb	0.975	0.626	0.629	0.780	0.671	1.039	1.309	0.211	0.305	0.245	0.223	0.389
Ti	3131	2351	3095	2115	1077	2447	3063	138		n.d.	n.d.	
V	305	231	301	248	213	316	295	180		n.d.	n.d.	
Cr	2574	6939	2756	7246	6388	3357	2471	7072		n.d.	n.d.	
Sr	24	81	21	62	90	20	24	32		n.d.	n.d.	
Y	13	7	14	6	6	11	12	2		n.d.	n.d.	
Zr	23.6	27.1	23.1	19.2	39.2	18.6	22.2	6.8		n.d.	n.d.	
<i>CI Normalized</i>												
La	4.5	17.9	3.7	11.0	25.4	1.9	3.2	10.8	41.2	6.5	1.7	1.7
Ce	5.4	18.3	4.8	12.0	24.5	3.1	4.8	8.3	29.6	2.3	1.4	1.2
Nd	6.4	16.1	4.2	11.2	17.2	4.5	6.4	5.6	10.1	0.6	0.6	0.8
Sm	9.7	11.3	6.7	10.7	11.0	7.5	8.7	5.7	5.0	0.6	0.8	1.5
Eu	8.9	8.6	5.7	8.7	8.5	10.1	11.5	4.4	5.4	0.4	1.0	2.2
Dy	7.2	5.4	4.1	5.5	5.2	8.7	8.8	2.6	2.7	0.6	0.6	2.6
Er	7.2	6.4	4.6	5.0	4.5	6.2	8.0	1.6	1.9	1.1	1.0	2.8
Yb	6.0	3.9	3.9	4.8	4.1	6.4	8.1	1.3	1.9	1.5	1.4	2.4
Ti	7.2	5.5	7.1	4.9	2.5	5.6	7.0	0.3				
Sr	3.1	10.6	2.7	7.9	11.5	2.5	3.0	4.1				
Zr	6.0	6.6	5.9	4.9	9.9	4.7	5.6	1.7				
(La/Yb) _N	0.75	4.66	0.94	2.30	6.16	0.30	0.40	8.29	21.95	4.30	1.27	0.90
(La/Sm) _N	0.47	1.59	0.55	1.03	2.31	0.26	0.37	1.88	8.22	11.23	2.14	0.51
												1.09

Table 2-8. continued.

Sample	OB93-283	OB93-284	OB93-286		OB93-287		OB93-288		OB93-289		OB93-291		OB93-297		OB93-298	
Grain ID	cpx1	cpx2	cpx1	cpx4s	cpx5s	cpx6	cpx1	cpx3	cpx7s	cpx1	cpx1	cpx1	cpx1	cpx1	cpx1	cpx6
ppm																
La	4.93	1.84	1.72	10.65	3.80	16.03	3.15	1.34	0.56	0.42	6.69	13.19	6.23	3.96		
Ce	13.48	4.46	4.02	17.48	5.16	26.17	4.26	2.57	0.94	0.61	7.74	21.92	12.62	10.36		
Nd	9.102	2.308	1.994	3.165	0.728	5.048	1.098	0.902	0.503	0.202	1.182	7.241	3.873	4.314		
Sm	2.792	0.614	0.529	0.383	0.171	0.367	0.657	0.284	0.352	0.145	0.483	2.079	0.606	0.932		
Eu	1.042	0.208	0.248	0.138	0.037	0.208	0.206	0.086	0.111	0.047	0.147	0.568	0.232	0.319		
Dy	2.648	0.618	0.826	0.321	0.176	0.571	0.711	0.232	0.301	0.166	0.649	1.519	0.371	0.531		
Er	1.349	0.443	0.570	0.202	0.200	0.493	0.514	0.151	0.302	0.195	0.401	0.876	0.244	0.310		
Yb	1.328	0.438	0.607	0.232	0.203	0.533	0.651	0.185	0.328	0.277	0.390	0.782	0.335	0.365		
Ti	n.d.		299	245		n.d.		371	n.d.		516	1543	258			
V	n.d.		226	121		n.d.		183	n.d.		199	212	181			
Cr	n.d.		8293	4026		n.d.		7250	n.d.		6903	10826	5589			
Sr	n.d.		28	28		n.d.		30	n.d.		54	128	74			
Y	n.d.		5	1		n.d.		2	n.d.		3	7	2			
Zr	n.d.		4.1	5.1		n.d.		9.7	n.d.		7.3	78.4	9.7			
CI Normalized																
La	21.0	7.8	7.3	45.4	16.2	68.3	13.4	5.7	2.4	1.8	28.5	56.2	26.5	16.9		
Ce	22.3	7.4	6.7	29.0	8.5	43.4	7.1	4.3	1.6	1.0	12.8	36.3	20.9	17.2		
Nd	20.1	5.1	4.4	7.0	1.6	11.2	2.4	2.0	1.1	0.4	2.6	16.0	8.6	9.5		
Sm	19.0	4.2	3.6	2.6	1.2	2.5	4.5	1.9	2.4	1.0	3.3	14.1	4.1	6.3		
Eu	18.6	3.7	4.4	2.5	0.7	3.7	3.7	1.5	2.0	0.8	2.6	10.1	4.1	5.7		
Dy	10.9	2.5	3.4	1.3	0.7	2.4	2.9	1.0	1.2	0.7	2.7	6.3	1.5	2.2		
Er	8.5	2.8	3.6	1.3	1.3	3.1	3.2	0.9	1.9	1.2	2.5	5.5	1.5	1.9		
Yb	8.2	2.7	3.7	1.4	1.2	3.3	4.0	1.1	2.0	1.7	2.4	4.8	2.1	2.2		
Ti			0.7	0.6				0.9			1.2	3.5	0.6			
Sr			3.6	3.5				3.8			6.9	16.5	9.5			
Zr			1.0	1.3				2.5			1.9	19.9	2.5			
(La/Yb) _N	2.57	2.91	1.96	31.84	12.97	20.81	3.35	5.03	1.19	1.06	11.87	11.68	12.87	7.50		
(La/Sm) _N	1.11	1.88	2.03	17.44	13.91	27.39	3.01	2.96	1.00	1.83	8.68	3.98	6.44	2.66		

Table 2-8. continued.

Sample	OB93-299			OB93-301		OB93-302		OB93-305		OB93-306	
Grain ID	avg cpx1 n=2	1 S.D. avg cpx3s n=2	1 S.D. avg cpx4s n=2	1 S.D.	cpx1	cpx2	cpx7	cpx5	cpx2	cpx1, pt2	cpx2
<i>ppm</i>											
La	2.28	0.59	1.36	0.02	0.08	6.23	2.70	3.74	3.21	2.40	0.22
Ce	2.40	0.79	2.05	0.05	0.21	11.51	4.40	9.78	9.64	6.45	0.58
Nd	0.330	0.044	0.589	0.040	0.135	2.829	1.131	4.805	6.827	5.131	0.594
Sm	0.035	0.010	0.123	0.049	0.023	0.463	0.272	1.254	2.359	1.750	0.344
Eu	0.022	0.001	0.043	0.005	0.024	0.120	0.073	0.394	0.813	0.800	0.109
Dy	0.103	0.019	0.108	0.004	0.038	0.251	0.154	0.612	1.498	2.566	0.414
Er	0.168	0.029	0.153	0.008	0.026	0.241	0.264	0.404	0.665	1.356	0.253
Yb	0.240	0.029	0.204	0.054	0.033	0.193	0.235	0.377	0.872	1.497	0.266
<i>CI Normalized</i>											
Ti	68	63	63		69	169	168	166	781	4655	4071
V	156	155	169		151	192	222	150	155	348	350
Cr	5287	6245	5764		4467	6229	6457	6357	5526	3934	4864
Sr	12	4	24		19	74	36	91	129	92	81
Y	1	1	1		1	1	1	2	7	15	13
Zr	0.7	0.4	6.2		5.3	8.3	3.9	11.2	20.9	43.5	40.3
La	9.7	5.8	15.7		14.3	26.6	11.5	15.9	13.7	10.2	1.0
Ce	4.0	3.4	12.5		11.7	19.1	7.3	16.2	16.0	10.7	1.0
Nd	0.7	1.3	6.3		6.4	6.3	2.5	10.6	15.1	11.3	1.3
Sm	0.2	0.8	4.1		3.2	3.1	1.8	8.5	16.0	11.9	2.3
Eu	0.4	0.8	3.4		3.3	2.2	1.3	7.0	14.5	14.3	1.9
Dy	0.4	0.4	0.9		0.8	1.0	0.6	2.5	6.2	10.6	1.7
Er	1.1	1.0	0.9		1.9	1.5	1.7	2.5	4.2	8.5	1.6
Yb	1.5	1.3	1.7		1.8	1.2	1.4	2.3	5.4	9.2	1.6
Ti	0.2	0.1	0.1		0.2	0.4	0.4	0.4	1.8	10.7	9.3
Sr	1.6	0.5	3.1		2.5	9.5	4.6	11.7	16.5	11.8	10.4
Zr	0.2	0.1	1.6		1.3	2.1	1.0	2.8	5.3	11.0	10.2
(La/Yb) _N	6.57	4.61	9.08		8.14	22.34	7.95	6.87	2.55	1.11	0.58
(La/Sm) _N	40.99	6.95	3.85		4.42	8.44	6.22	1.87	0.85	0.86	0.41

Table 2-8. continued.

Sample Grain ID	OB93-307	OB93-308	OB93-309	OB93-310	OB93-313	1 S.D.
	cpx3 avg n=2	1 S.D.	cpx5s cpx10s avg n=2	1 S.D.	cpx4	1 S.D.
ppm						
La	0.16	0.08	0.03	5.05	0.39	0.39
Ce	0.14	0.07	0.03	9.73	1.22	1.22
Nd	0.072	0.016	0.053	3.218	0.154	0.154
Sm	0.050	0.022	0.021	0.496	0.019	0.019
Eu	0.015	0.000	0.020	0.175	0.028	0.028
Dy	0.092	0.017	0.086	0.235	0.037	0.037
Er	0.172	0.038	0.111	0.183	0.043	0.043
Yb	0.246	0.036	0.173	0.215	0.010	0.010
Ti	68			3187		
V	152			125		
Cr	5712			60		
Sr	1			48		
Y	1			27		
Zr	0.7			1.1		
<i>Cl Normalized</i>						
La	0.7			21.5		
Ce	0.2			16.1		
Nd	0.2			7.1		
Sm	0.3			3.4		
Eu	0.3			3.1		
Dy	0.4			1.0		
Er	1.1			1.2		
Yb	1.5			1.3		
Ti	0.2			7.3		
Sr	0.1			6.2		
Zr	0.2			0.3		
(La/Yb) _N	0.45			16.30		
(La/Sm) _N	2.04			6.38		

Table 2-8. continued.

Sample	OB93-314		OB93-317		OB93-318	
Grain ID	avg cpx	1 S.D.	cpx1	cpx2	cpx4s avg	1 S.D.
	n=4				n=2	
<i>ppm</i>						
La	0.30	0.01	0.87	0.43	1.91	0.15
Ce	0.81	0.05	1.87	0.63	4.15	0.01
Nd	0.687	0.076	1.748	0.178	1.206	0.008
Sm	0.354	0.149	0.685	0.079	0.298	0.156
Eu	0.097	0.036	0.235	0.036	0.079	0.003
Dy	0.296	0.113	0.783	0.188	0.248	0.013
Er	0.245	0.081	0.682	0.242	0.275	0.001
Yb	0.303	0.102	0.389	0.317	0.298	0.015
Ti	167		359	268		
V	167		191	187		
Cr	7436		8126	6888		
Sr	7		25	22		
Y	1		3	2		
Zr	2.2		16.3	2.8		
<i>CI Normalized</i>						
La	1.3		3.7	1.8	8.1	
Ce	1.3		3.1	1.0	6.9	
Nd	1.5		3.9	0.4	2.7	
Sm	2.4		4.7	0.5	2.0	
Eu	1.7		4.2	0.6	1.4	
Dy	1.2		3.2	0.8	1.0	
Er	1.5		4.3	1.5	1.7	
Yb	1.9		2.4	1.9	1.8	
Ti	0.4		0.8	0.6		
Sr	0.9		3.2	2.8		
Zr	0.5		4.1	0.7		
(La/Yb) _N	0.70		1.55	0.93	4.43	
(La/Sm) _N	0.54		0.79	3.39	4.02	

Table 2-9. Sr and Nd isotopic compositions.

Sample	$^{87}\text{Sr}/^{86}\text{Sr}$	[Sr] ppm	$^{143}\text{Nd}/^{144}\text{Nd}$	Epsilon Nd	[Nd] ppm	[Rb] ppm	[Sm] ppm
OB93-52*	0.705205	115	0.512598	-0.78	5.1	0.007	n.d.
OB93-51*	0.706450	177	0.512428	-4.11	6.5	n.d.	n.d.
OB93-78†	0.703296	0.2	n.d.		n.d.	0.006	0.0004
OB93-80†	0.705272	1.8	0.512451	-3.6	0.1	0.005	0.001
OB93-83 cpx*	0.706427	557	0.512413	-4.4	119.5	0.069	n.d.
OB93-83 phlog*	0.706814	286	0.512434	-4.0	n.d.	155.4	n.d.
OB93-94 cpx§	0.708686	63.9	0.511988	-12.7	6.9	0.2	n.d.
OB93-94 phlog§	0.712744	43.1	n.d.		n.d.	209	n.d.
OB93-64†	0.705239	29.3	0.512510	-2.5	2.3	n.d.	0.78
OB93-64§	0.705393	24.6	n.d.	n.d.	n.d.	n.d.	n.d.
OB93-59 dike*	0.706380	889	0.512419	-4.3	23.04	n.d.	n.d.

n.d. not determined

* Sr and Nd analyzed at WHOI; † Sr, Nd, Rb and Sm analyzed at Okayama University; § Sr analyzed at MIT, Nd at WHOI.

All analyses are on clinopyroxene separates, except those noted "phlog" (phlogopite), and the whole rock dike.

All concentrations were determined by isotope dilution and TIMS on the same sample aliquot as the isotopic composition.

Precision of isotopic compositions (2s) for Sr measured at WHOI = 21-25 ppm, and 5-6 ppm for those measured at MIT.

Precision of isotopic compositions (2s) for Nd measured at WHOI are <10 ppm, except OB93-94 phlog = 72 ppm.

Precision of isotopic compositions (2s) measured at Okayama for Sr = 8-22 ppm,

and 16 ppm (OB93-64) and 46 ppm (OB93-80) for Nd.

Table 2-10. Re-Os isotopic compositions (from Hassler and Shimizu, 1998).

Sample	$^{187}\text{Os}/^{188}\text{Os}$	Os (ppt)	Re (ppt)	Re/Os ($\times 10^{-3}$)	T_{RD} (Ga)
<i>Lac Superieur</i>					
OB93-51*	0.1262 $\pm$ 2	87	n.d.		
OB93-52†	0.1286 $\pm$ 3	4218	n.d.		
OB93-64‡	0.1300 $\pm$ 2	968	n.d.		
OB93-77	0.1263 $\pm$ 3	3199	5	2	
OB93-78	0.1266 $\pm$ 6	2888	n.d.		
OB93-80	0.1252 $\pm$ 2	1727	n.d.		
OB93-82‡	0.1287 $\pm$ 2	6999	n.d.		
OB93-83*§	0.1257 $\pm$ 4	44	n.d.		
<i>Mont Trapeze</i>					
OB93-280	0.1189 $\pm$ 2	3067	6	2	1.36
OB93-284	0.1236 $\pm$ 1	6652	29	4	0.63
OB93284b	0.1229 $\pm$ 3	6645	n.d.		0.74
OB93-287	0.1239 $\pm$ 3	1267	49	39	0.58
OB93-289	0.1224 $\pm$ 3	3411	12	4	0.81
OB93-291	0.1205 $\pm$ 4	2157	11	5	1.11
OB93-297	0.1228 $\pm$ 3	2409	31	13	0.75
OB93-305	0.1211 $\pm$ 3	2012	72	36	1.02
OB93-305r	0.1202 $\pm$ 4	2179	n.d.		1.16
OB93-306	0.1276 $\pm$ 5	516	41	80	
OB93-307	0.1229 $\pm$ 3	5605	49	9	0.74
OB93-314	0.1383 $\pm$ 6	3281	9	3	
OB93-317	0.1196 $\pm$ 4	2652	11	4	1.25

*Olivine Separate. †Spinel lherzolite. ‡Wehrlitic dunite.

§Phlogopite-bearing spinel lherzolite. ||Replicate.

2.8 Figures

Figure 2-1. Examples of peridotite textures and mineralogy. Reflected light photomicrograph of spinel and cpx symplectite intergrowth from sample OB93-79. The light colored mineral with vermicular intergrowth is spinel and the darker gray is clinopyroxene. The scale is 1mm across the grain.

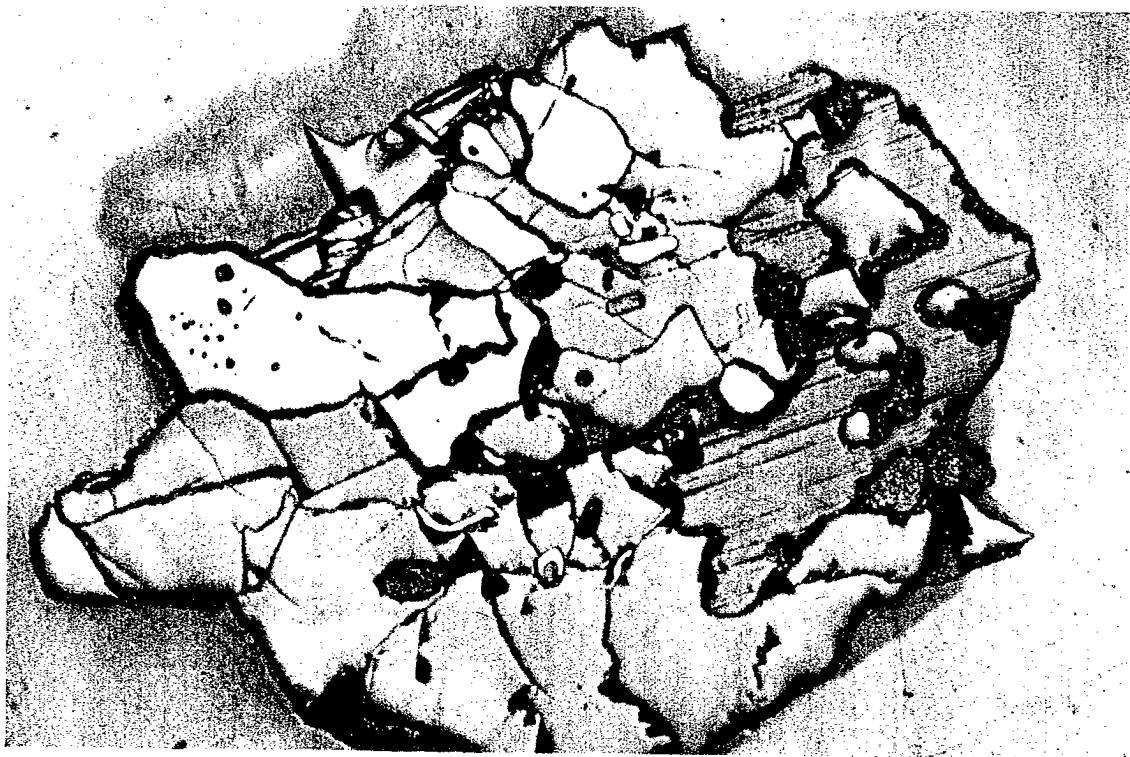


Figure 2-1. continued. Reflected light photomicrograph showing the metasomatic texture of sample OB93-83. Remnants of symplectite spinel (light colored, central lower part of the grain), can be seen in the clinopyroxene (darker gray color). Phlogopite (central small euhedral grain and the striated area on the right) coexists with apatite (stippled area to lower right of phlogopite). The scale is 1.5 mm across the grain.

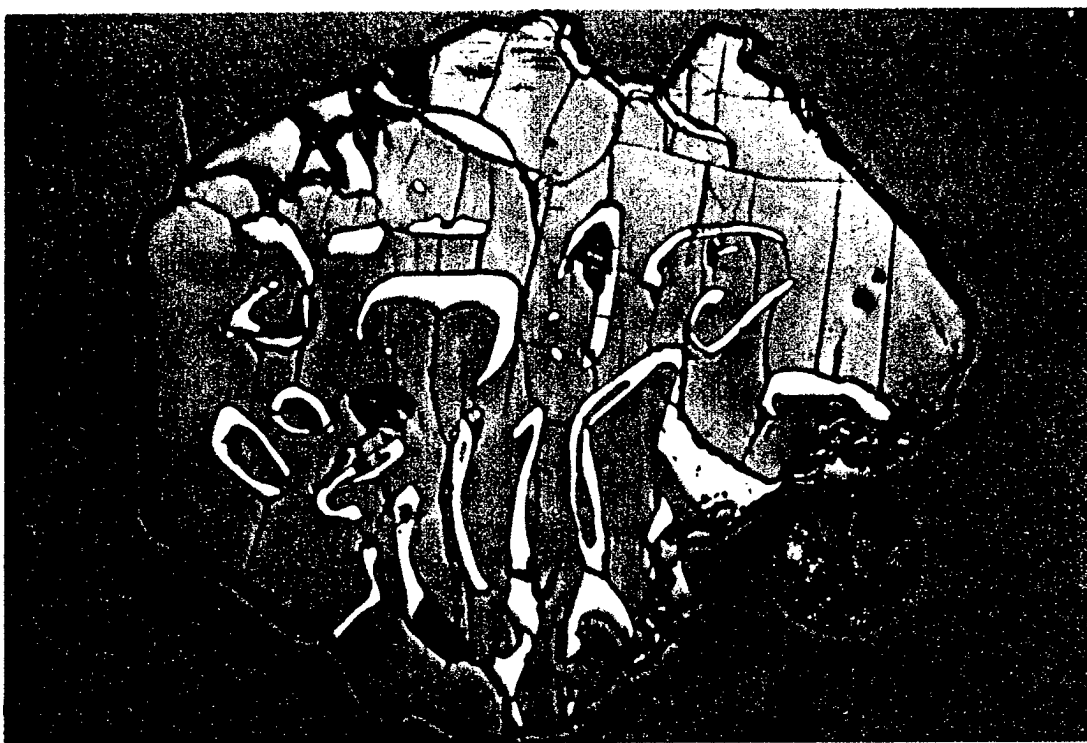


Figure 2-1. continued. Plane polarized light photomicrograph of sample OB93-301, exhibiting a typical coarse symplectite texture. Chromite spinel is the central dark vermicular mineral. One small clinopyroxene is present next to the leftmost clinopyroxene. Other minerals are orthopyroxene and olivine. The field of view is 10 mm across.

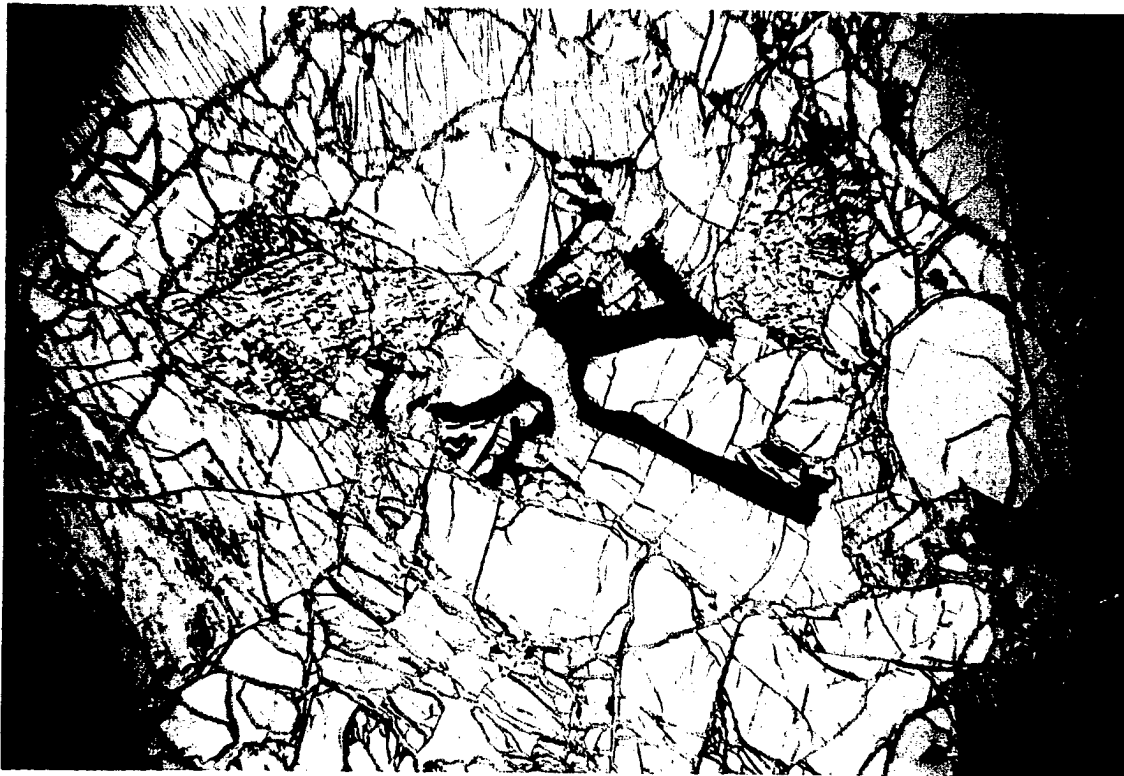


Figure 2-1. continued. Cross polarized light photomicrograph of sample OB93-22, showing the typical deformation texture that is present in these peridotite. Orthopyroxenes are the minerals with the lower birefringence, those with the higher order colors are olivine. Note the deformation banding and the subgrains (smaller rounded olivines) along larger olivine boundaries. The field of view is 10 mm across.



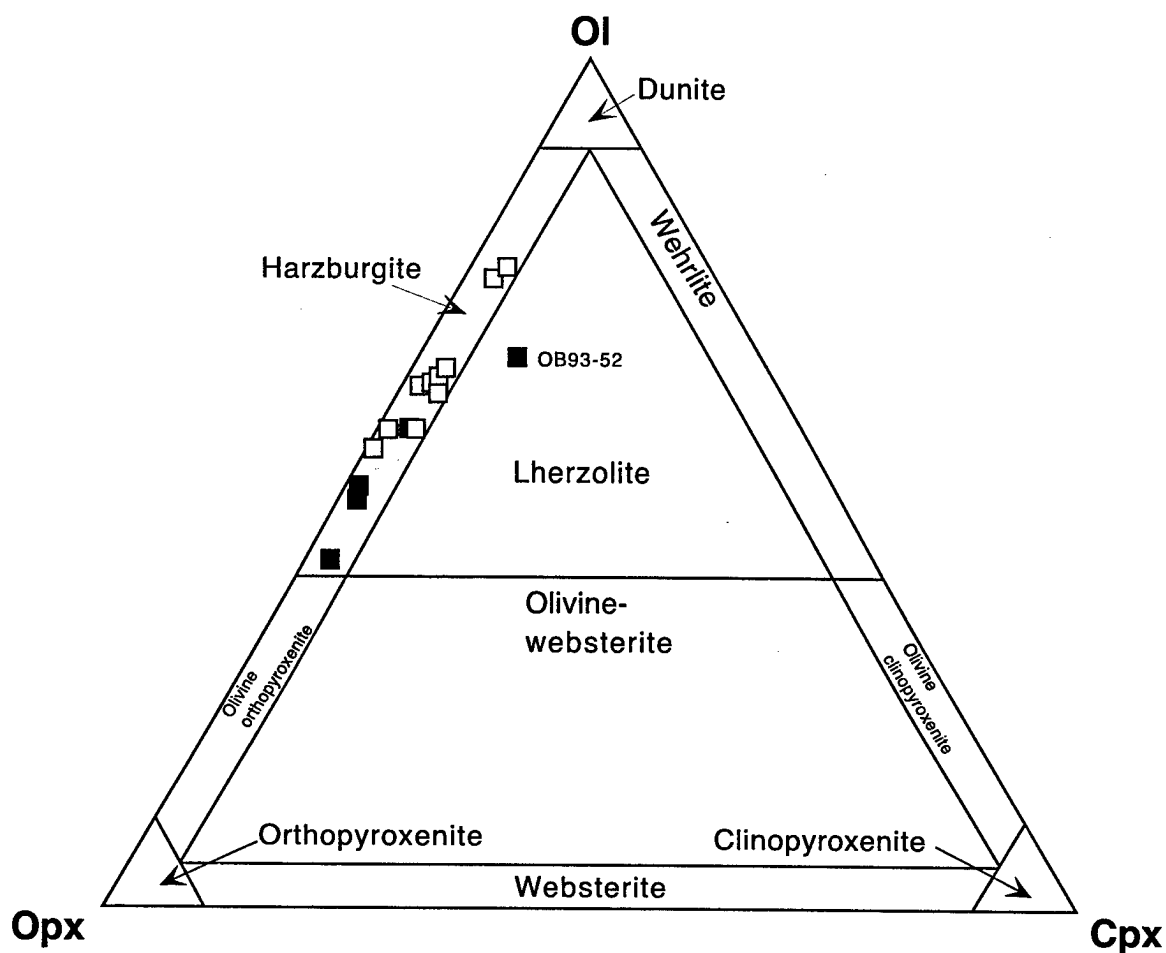


Figure 2-2. Whole rock geochemistry of Kerguelen peridotites classified according to the IUGS system (Streckheisen, 1973). The calculated mineral modal abundances are shown as filled squares. All samples are harzburgites, except sample OB93-52, which plots in the lherzolite field. Note that many of the samples are extremely rich in their orthopyroxene (Opx) contents. Mt. Trapeze: open squares; Lac Superieur: closed squares.

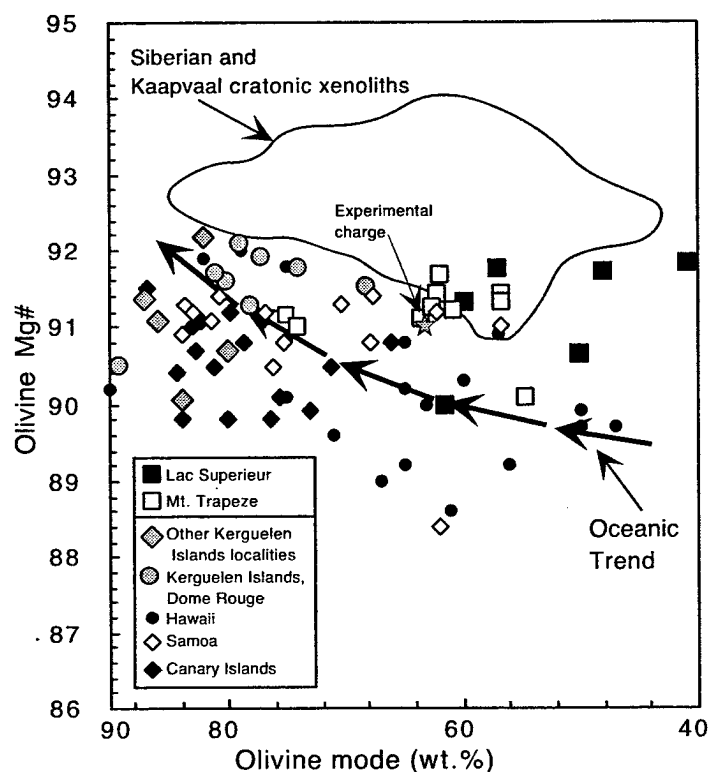
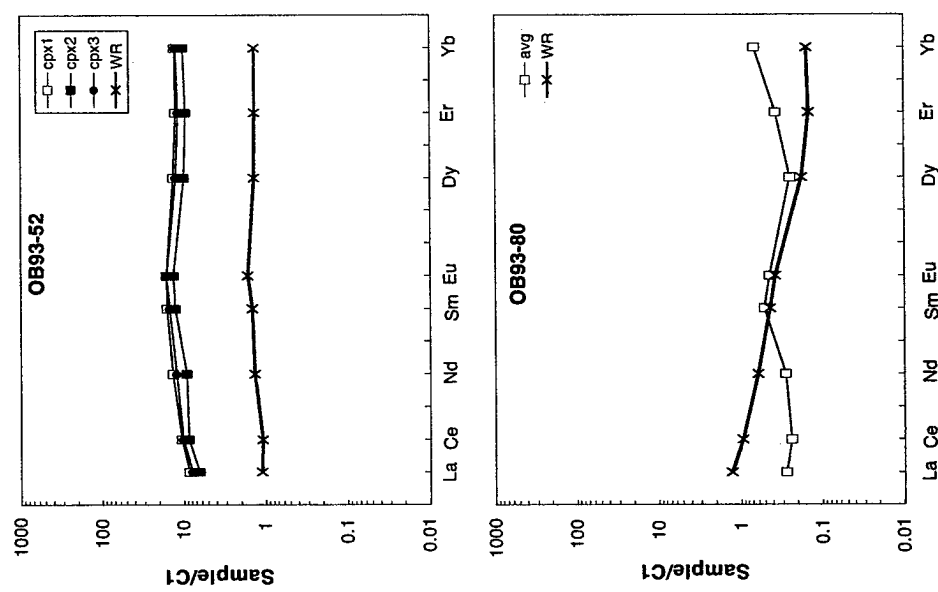
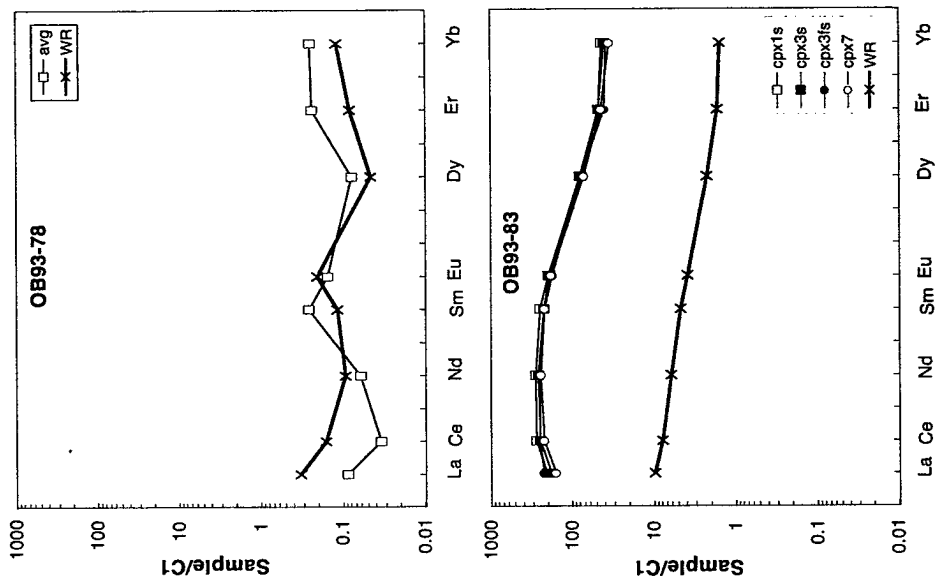
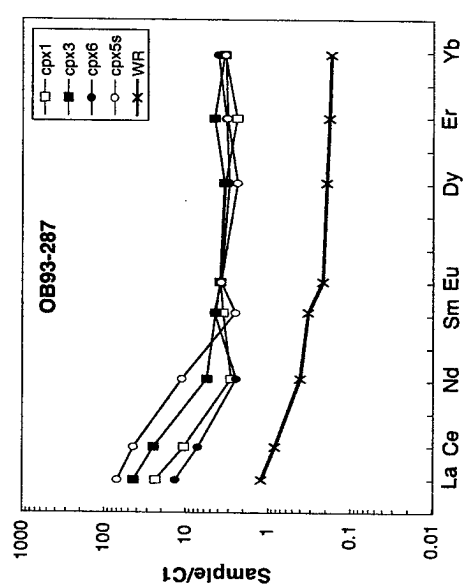
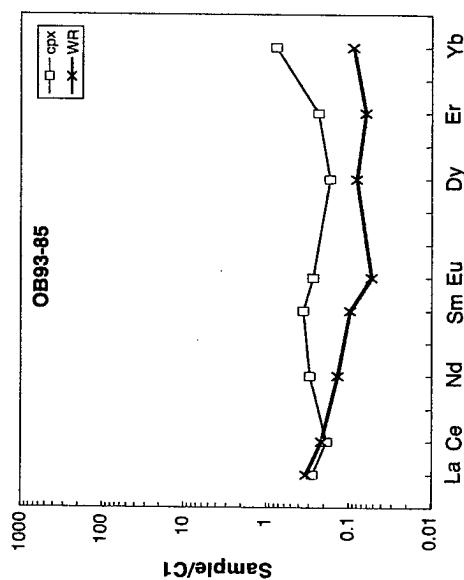
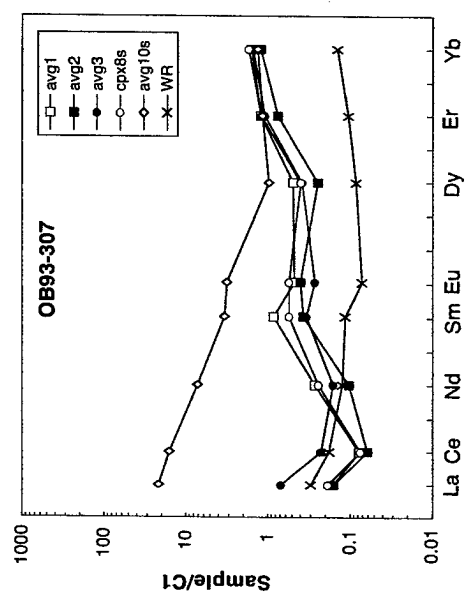
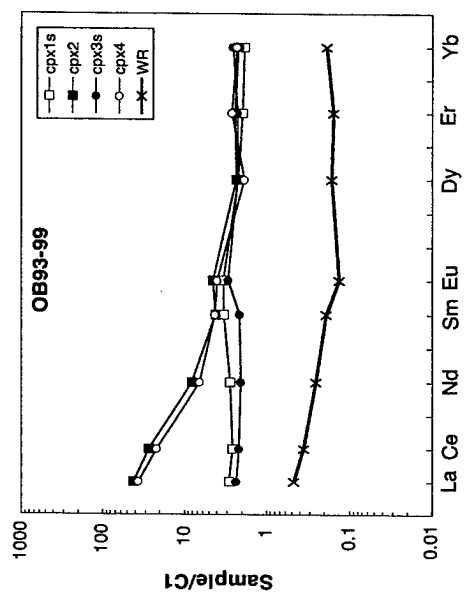


Figure 2-3. Oceanic trend diagram (after Boyd, 1989). Kerguelen samples of this study (Lac Superieur locality: filled squares, Mt. Trapeze open squares, and other Southeastern Province samples (Dome Rouge: gray circles and others: gray diamonds) are shown along with other oceanic samples (Hawaii: small filled circles, Samoa, small open diamonds, and the Canary Islands: small filled diamonds). The lherzolite (Lac Superieur sample OB93-52) plots on the oceanic trend, as do three of the Mt. Trapeze samples (OB93-287, OB93-289, OB93-297). The remaining samples of this study, and some samples from the Southeast Province, Hawaii, Samoa, and one from the Canary Islands lie above the oceanic trend, some even extending into the field defined by Siberian and Kaapvaal cratonic xenoliths. Fields and data from Boyd (1989), Boyd et al. (1997), Kelemen et al. (1998), Grègoire (1994), Hauri and Hart (1994), Sen (1988), and Wulff-Pedersen et al. (1996).

Figure 2-4. Comparison of REE clinopyroxenes and whole rocks for selected samples, illustrating the trace element mass balance. See the text for more discussion.





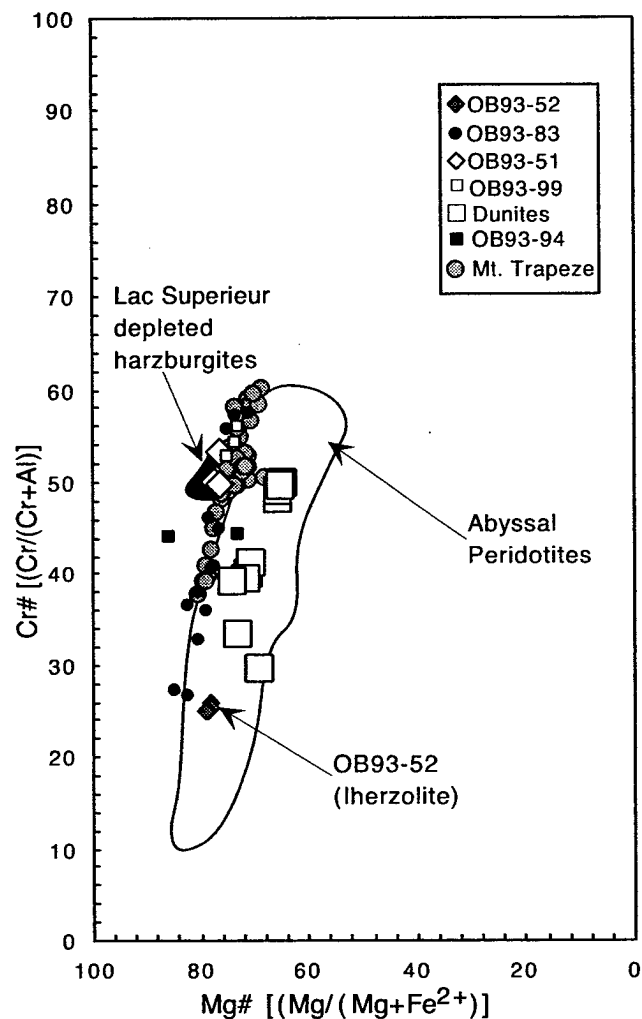


Figure 2-5. Spinel Cr# v. Mg#. Spinel from the Lac Superior REE depleted samples (gray field) have a restricted range of Cr-rich spinels indicating their extreme depletion. The Iherzolite (OB93-52) plots towards the other end of the array, consistent with its fertile nature. The remaining samples fall between the two extremes, some extending to even more Cr-rich compositions. This indicates the variation among residual and metasomatic signatures in these samples. The dunites have lower Mg# for a given Cr#. The abyssal peridotite field shown for reference (from Dick and Bullen, 1984)

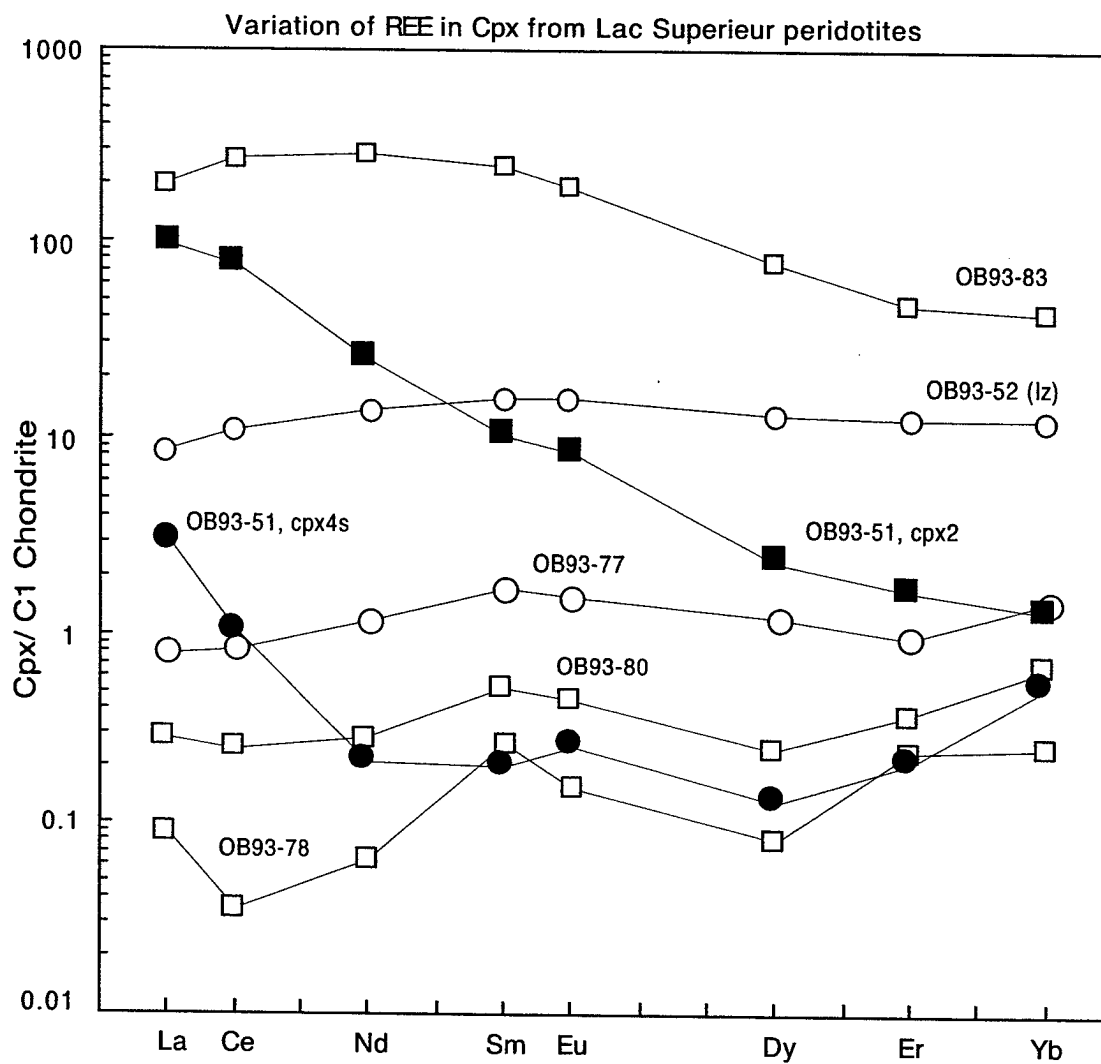


Figure 2-6. Range of clinopyroxene REE compositions from the Lac Superieur locality. Open symbols represent grains from individual rocks, closed symbols are from a single xenolith (OB93-51).

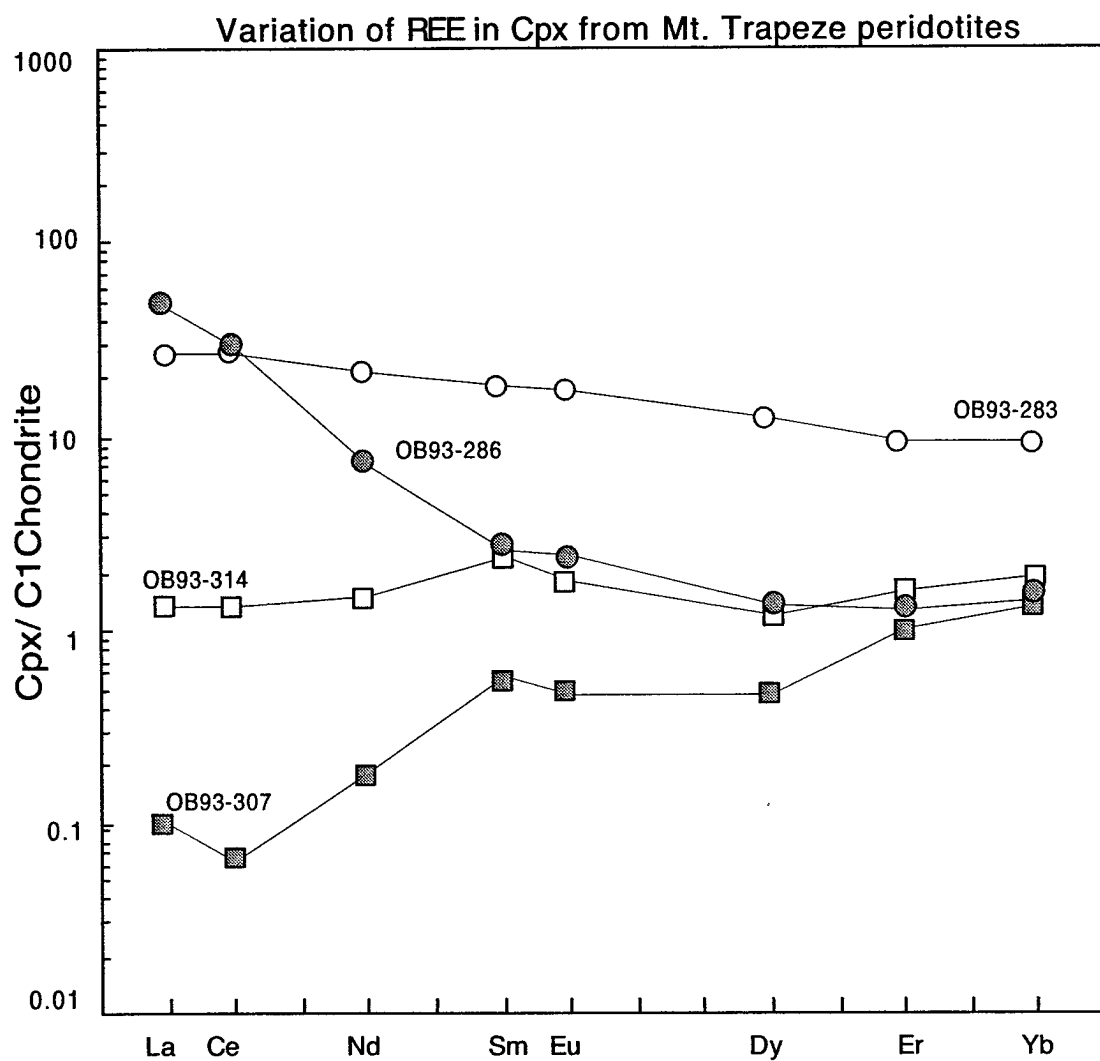
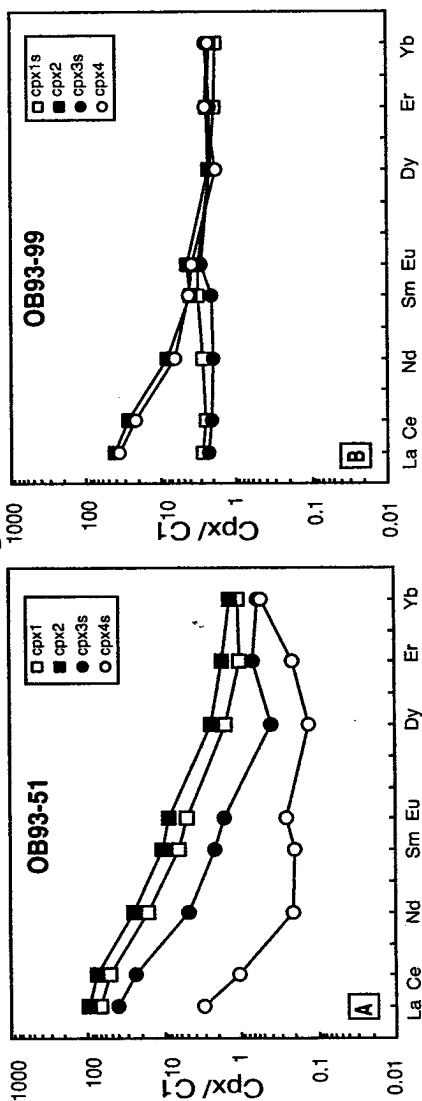


Figure 2-7. Range of clinopyroxene REE compositions from the Mt. Trapeze locality.

Figure 2-8. Normalized REE patterns of harzburgites with heterogeneous REE abundances.

Heterogeneous REEharzburgites (Lac Superieur)



Heterogeneous REEharzburgites (Mont Trapeze)

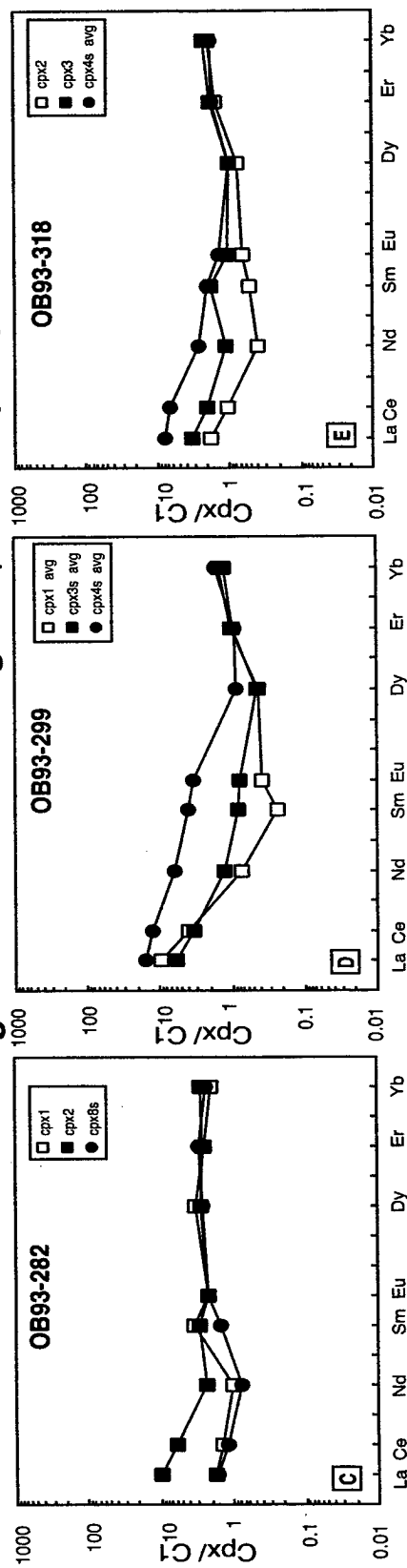
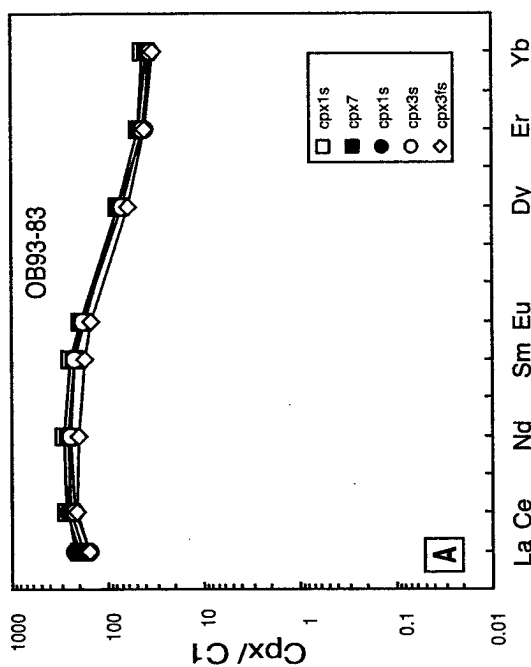
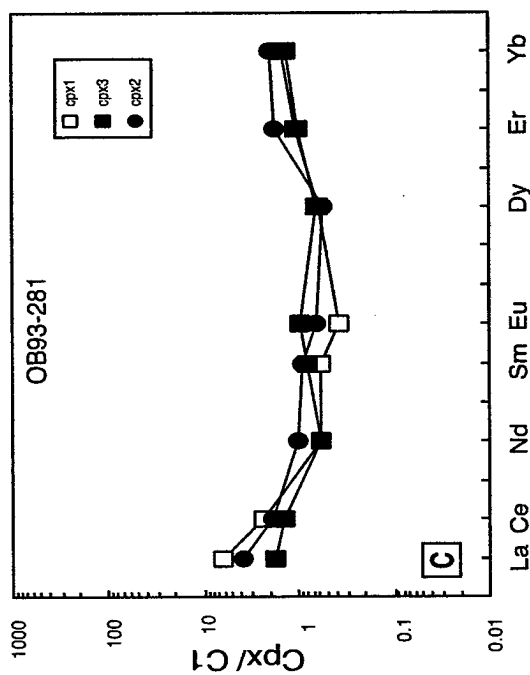
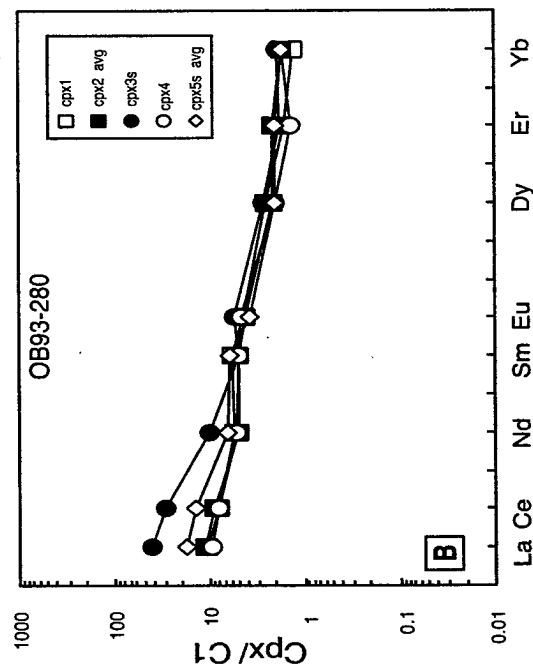


Figure 2-9. Normalized REE patterns of LREE enriched harzburgites.

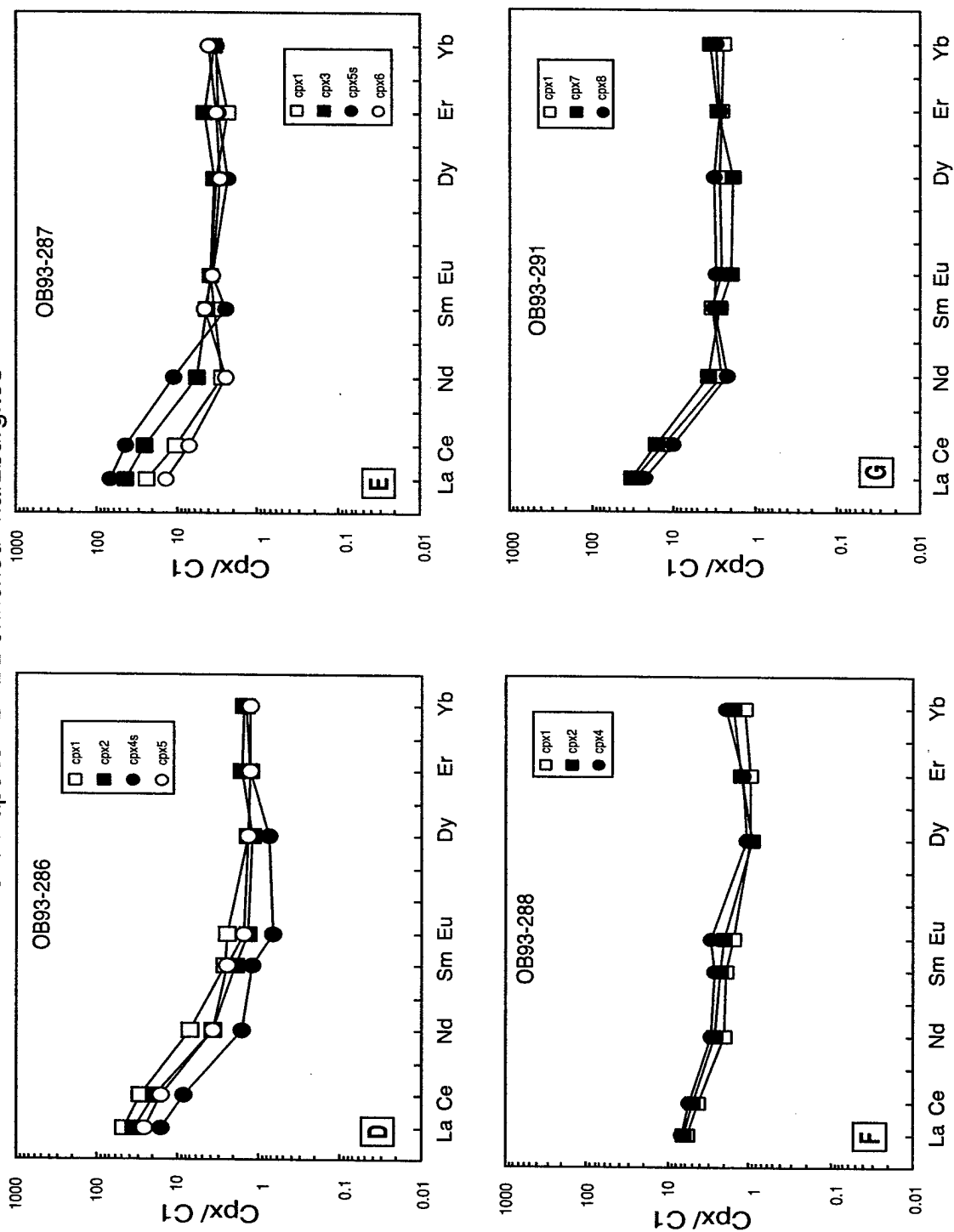
Lac Superior LREE enriched harzburgites



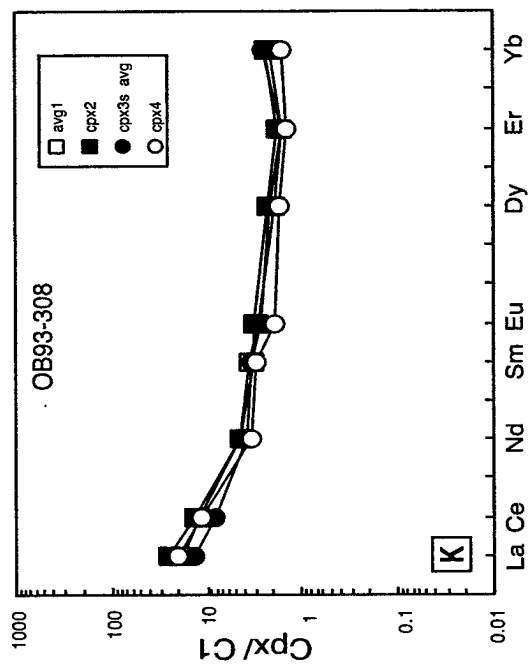
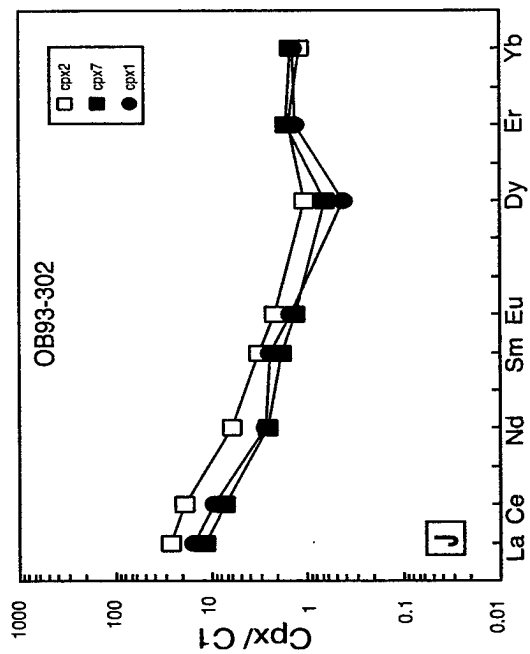
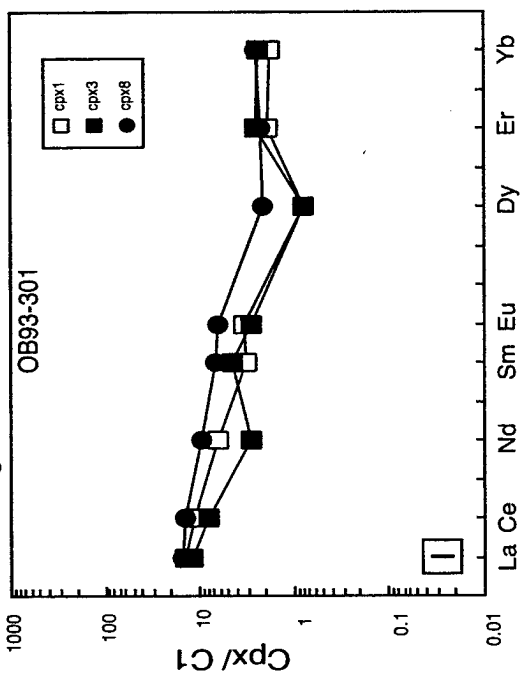
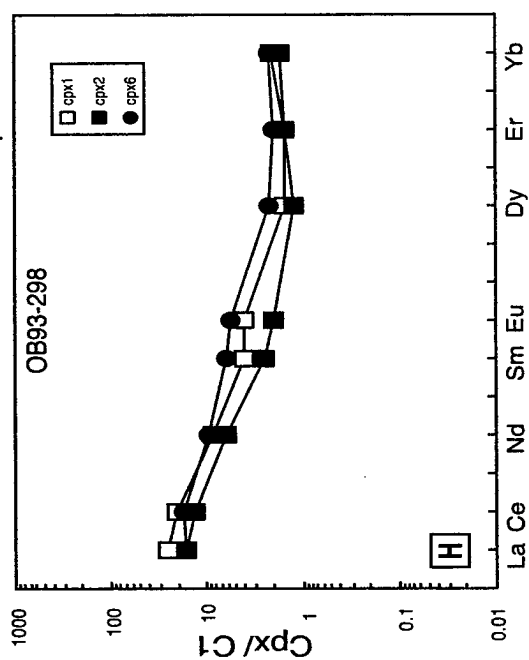
Mont Trapeze LREE enriched harzburgites



Mont Trapeze LREE enriched harzburgites



Mont Trapeze LREE enriched harzburgites



Mont Trapeze LREE enriched harzburgites

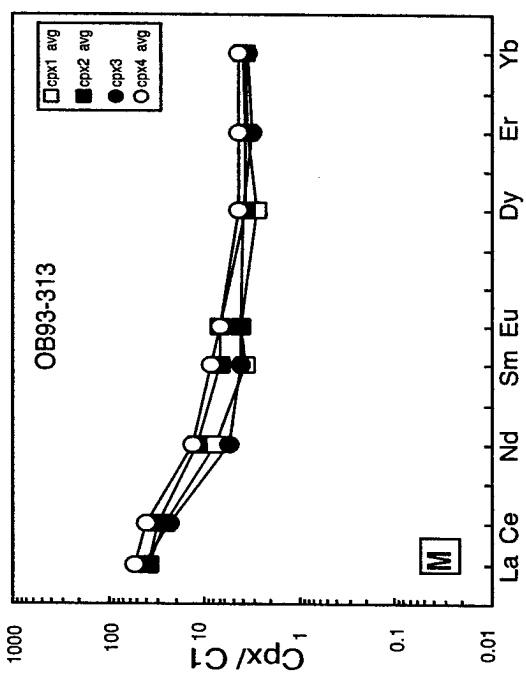
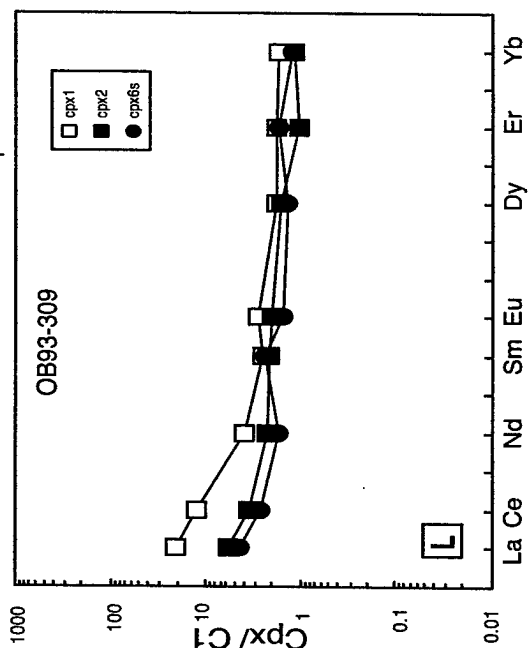
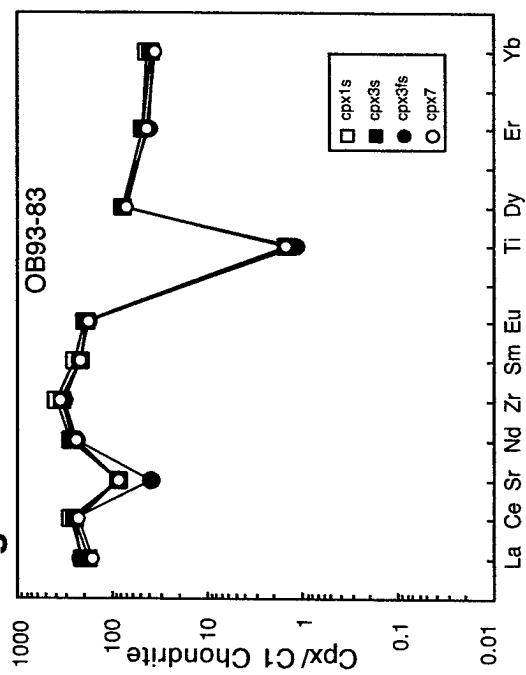
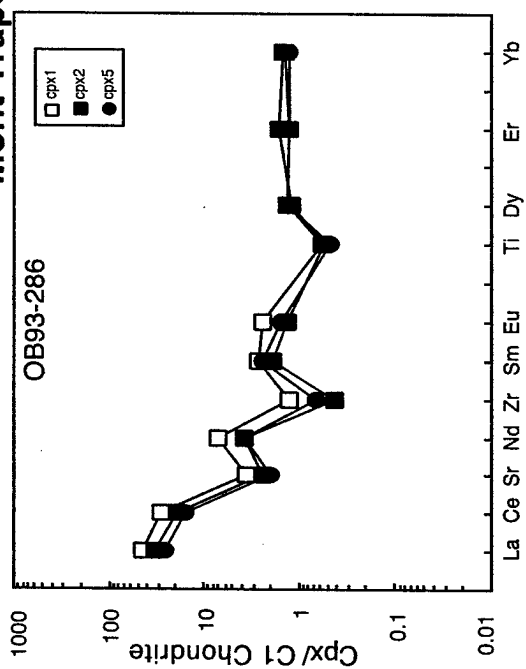
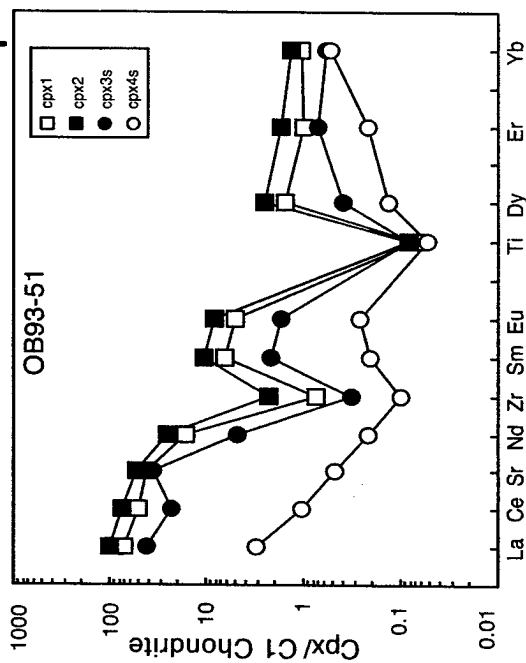
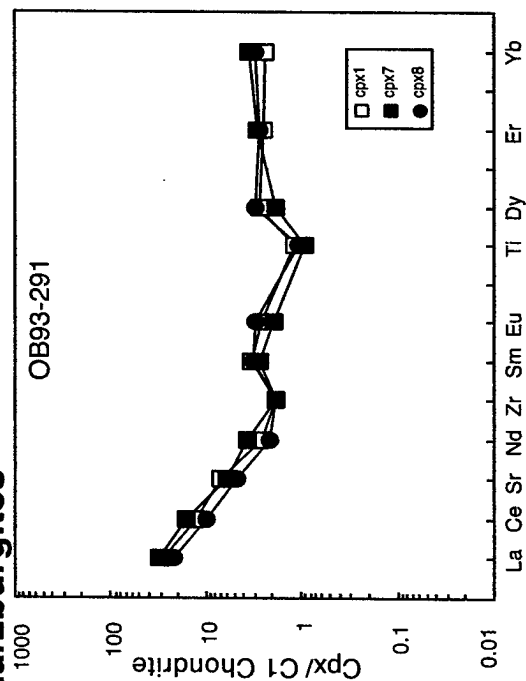


Figure 2-10. Spidergrams of samples with high REE abundances and negative relative Ti and Zr anomalies.

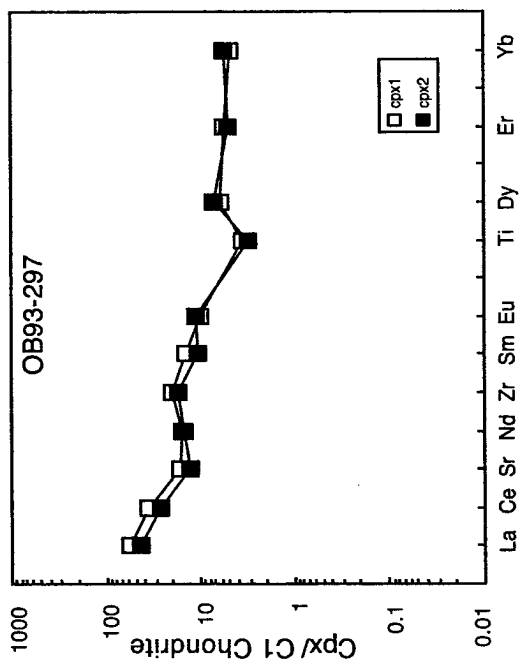
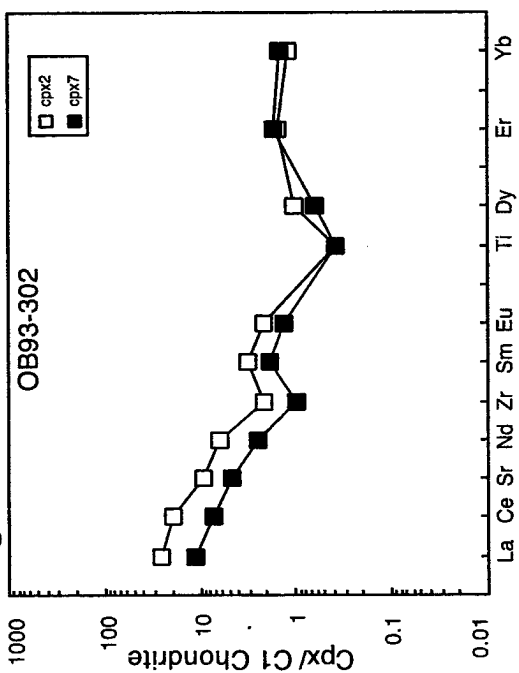
Lac Supérieur harzburgites



Mont Trapeze harzburgites



Mont Trapeze harzburgites



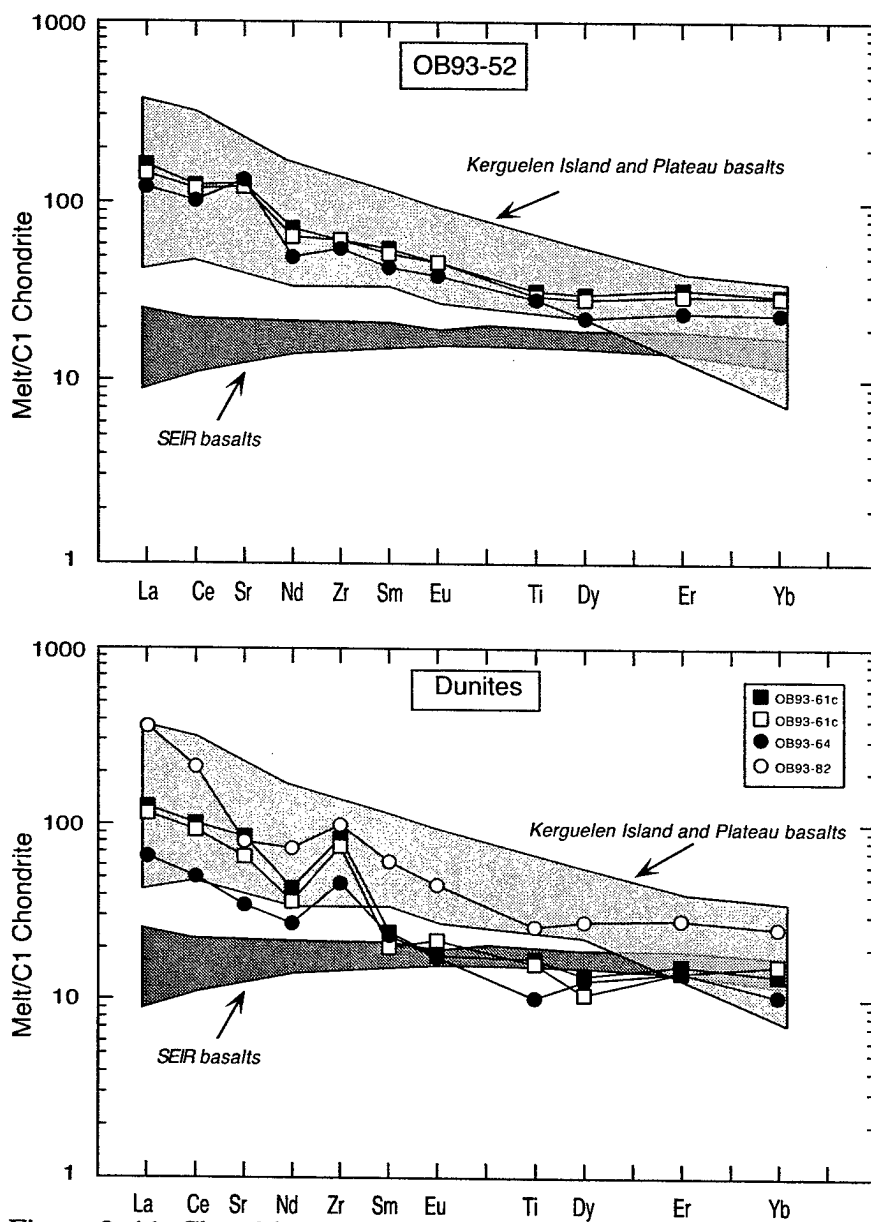


Figure 2-11. Chondrite-normalized REE abundances of melts in equilibrium with clinopyroxenes from lherzolite OB93-52 and from the wehrlitic-dunites. Abundances in Kerguelen basalts are shaded light gray, those from the SEIR are darker gray. Fields based on data from: (Davies et al., 1989; Alibert, 1991; Gautier et al., 1990; Weis et al., 1993; Yang et al., 1998a; Michard et al., 1986; Dosso et al., 1988).

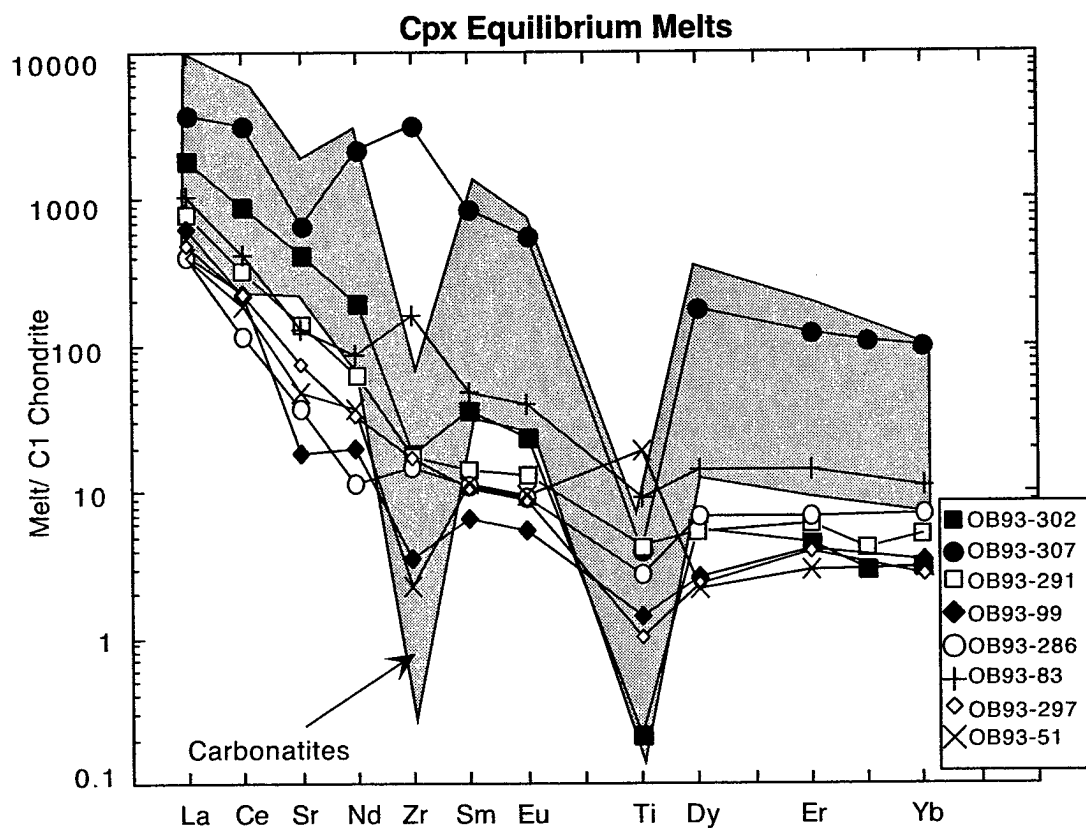


Figure 2-12. REE patterns of hypothetical melts that are similar to REE abundances observed in carbonatites (gray field; Nelson et al., 1988).

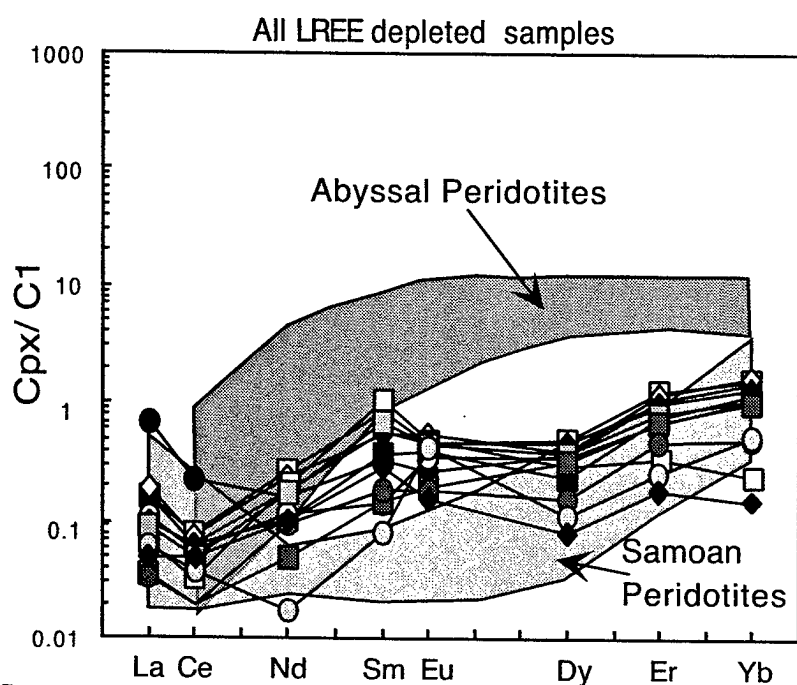


Figure 2-13. Summary of trace element abundances of LREE depleted samples. Fields for Samoan peridotites (Hauri and Hart, 1994) and abyssal peridotites (Johnson et al., 1990) are shown for comparison.

Figure 2-14. Trace element abundances of clinopyroxenes from LREE depleted harzburgites.

LREE depleted harzburgites

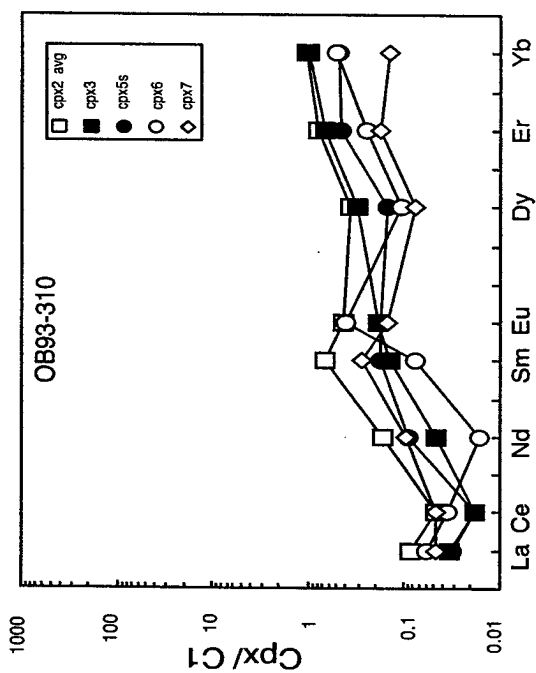
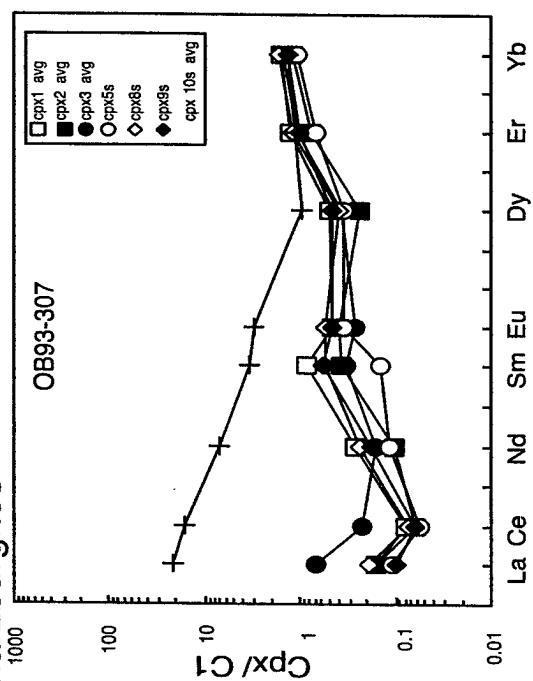
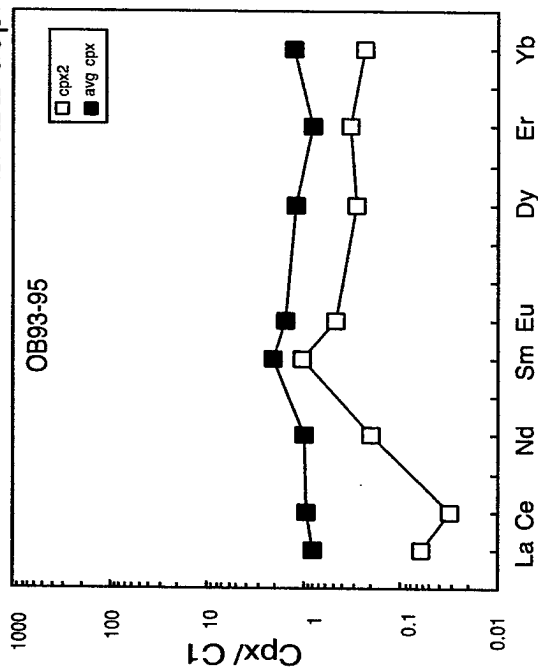
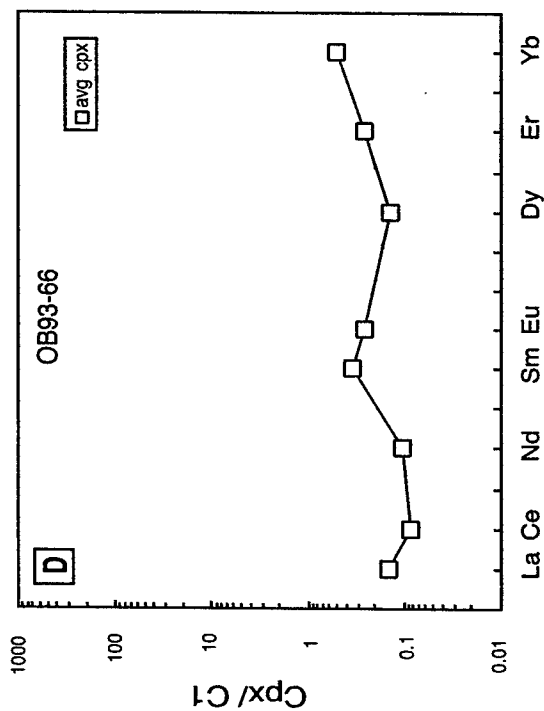
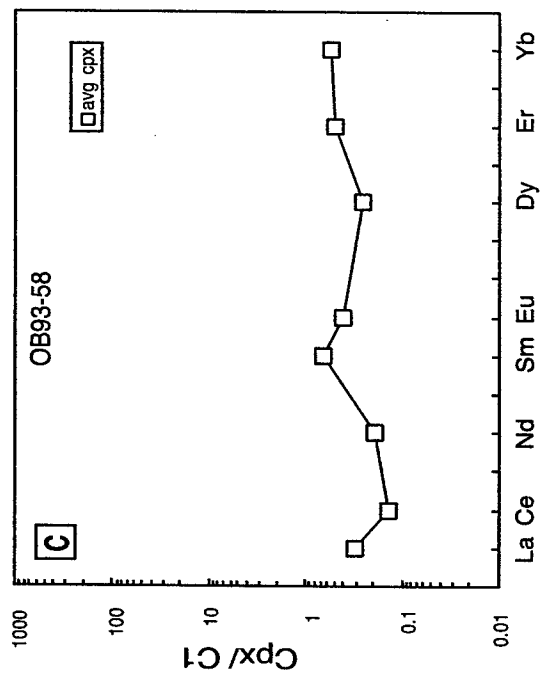
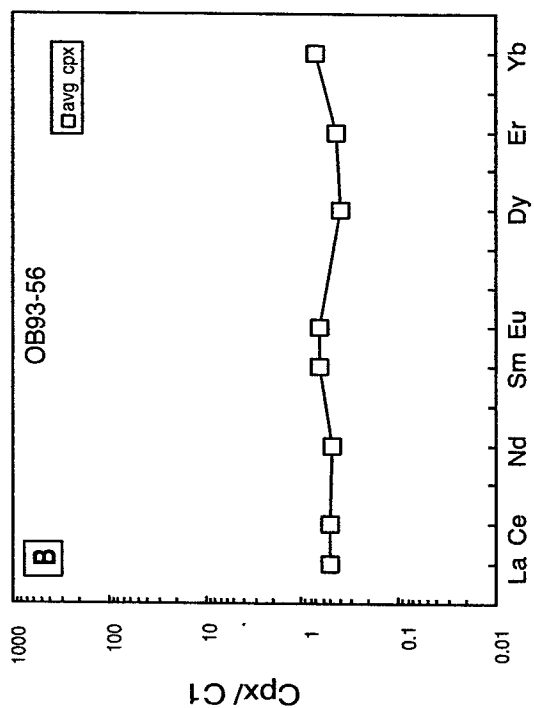
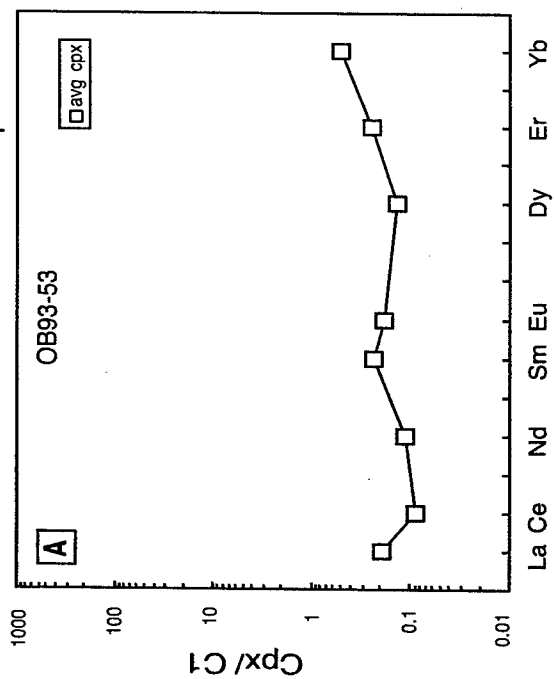
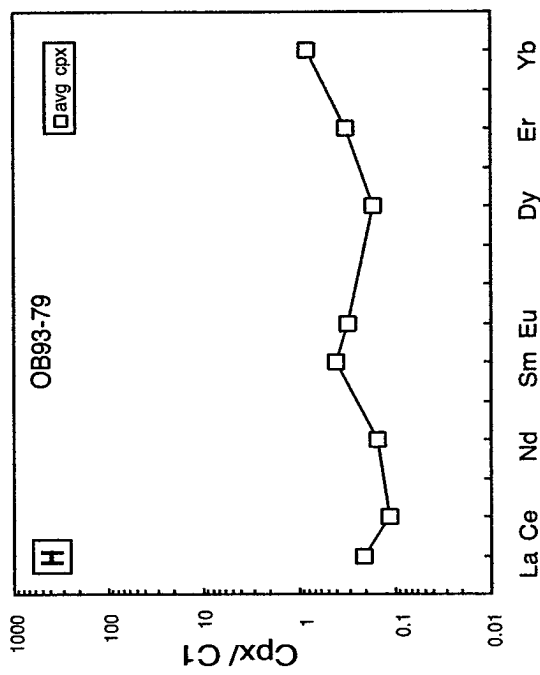
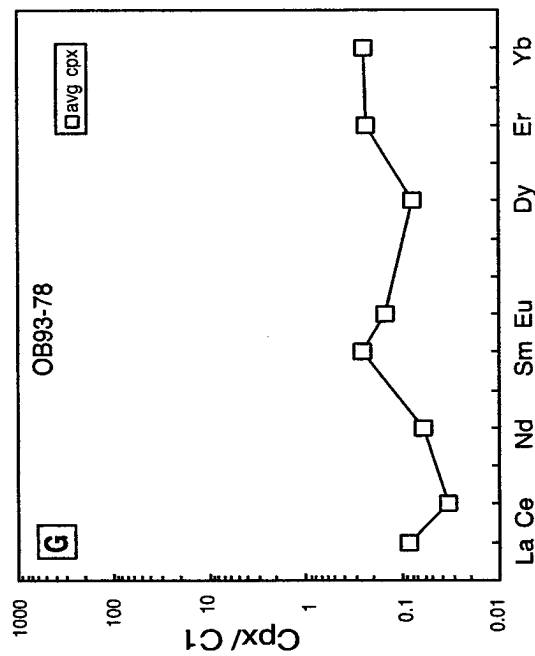
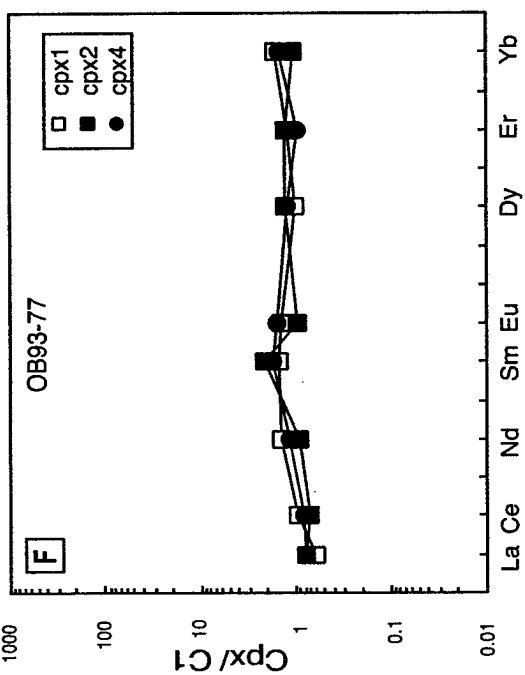
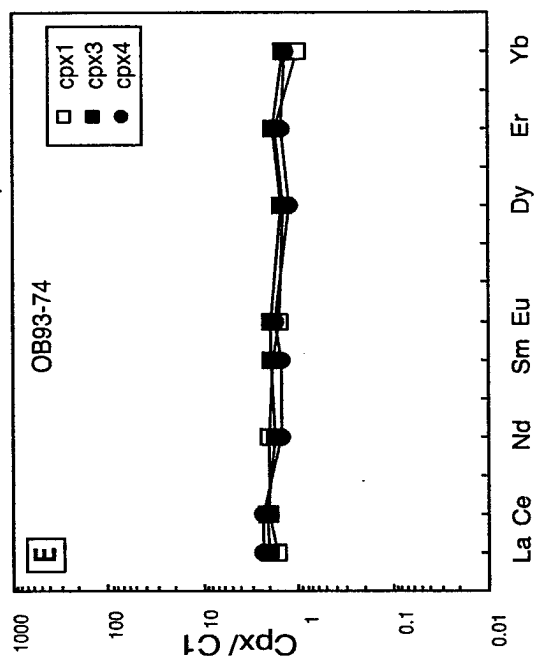


Figure 2-15. Normalized REE patterns of clinopyroxenes from depleted harzburgites.

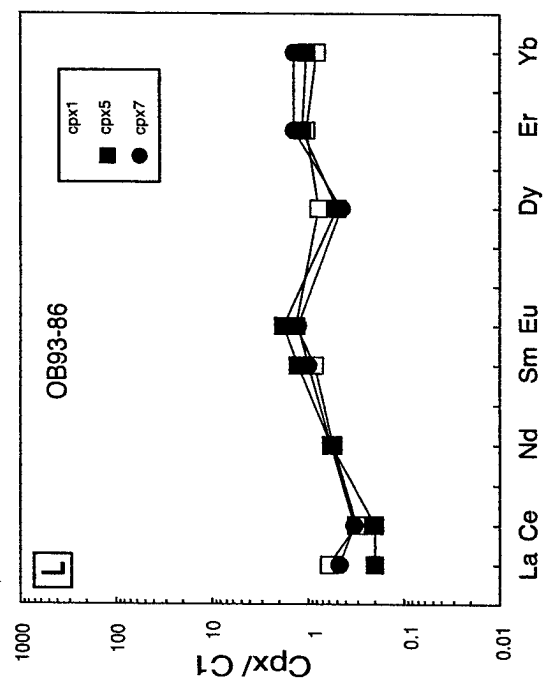
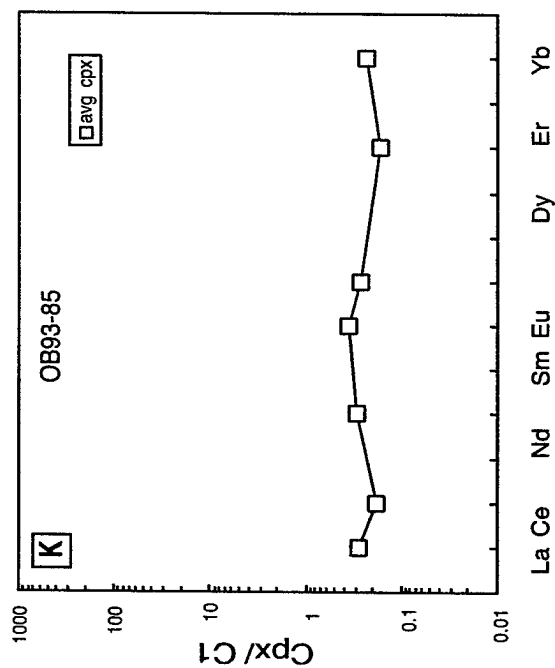
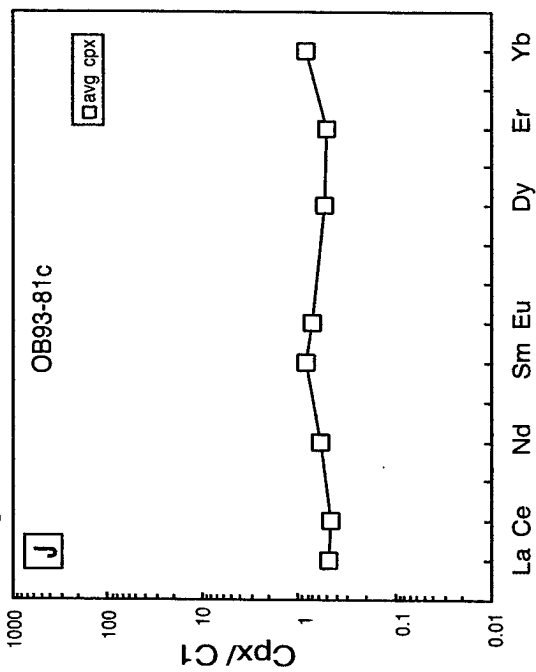
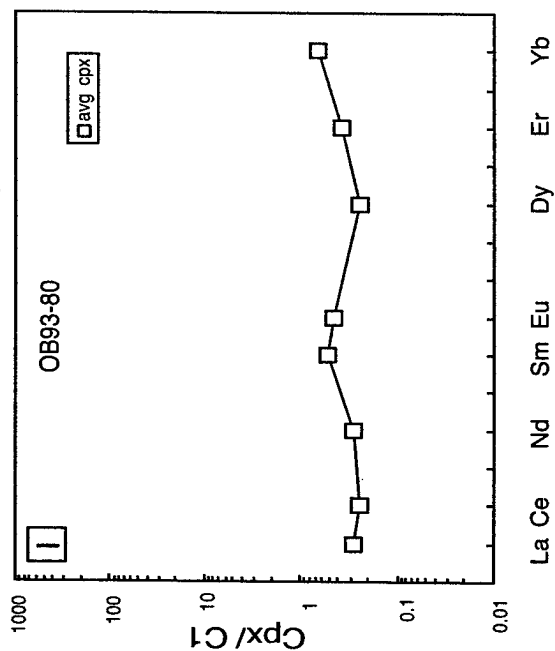
Lac Superior REE depleted harzburgites



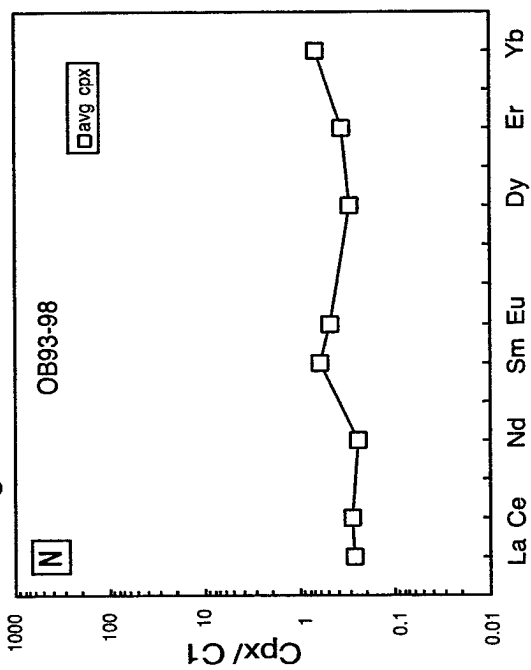
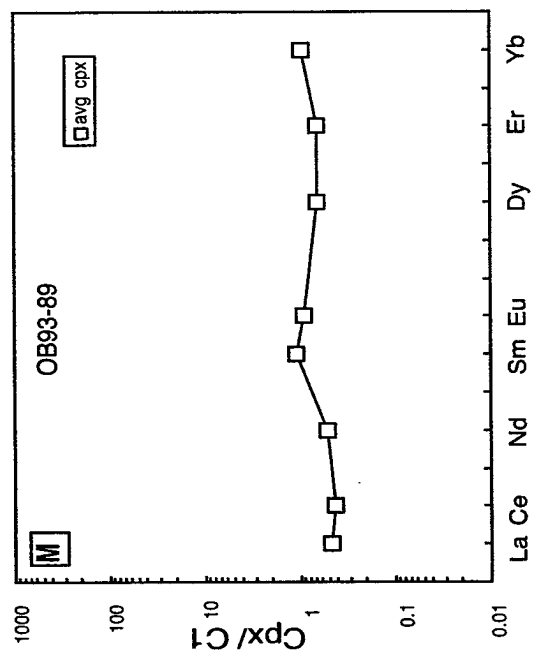
Lac Superior REE depleted harzburgites



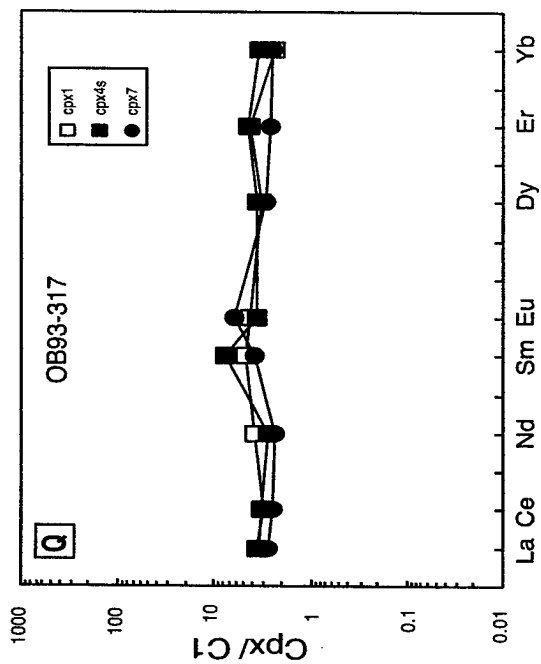
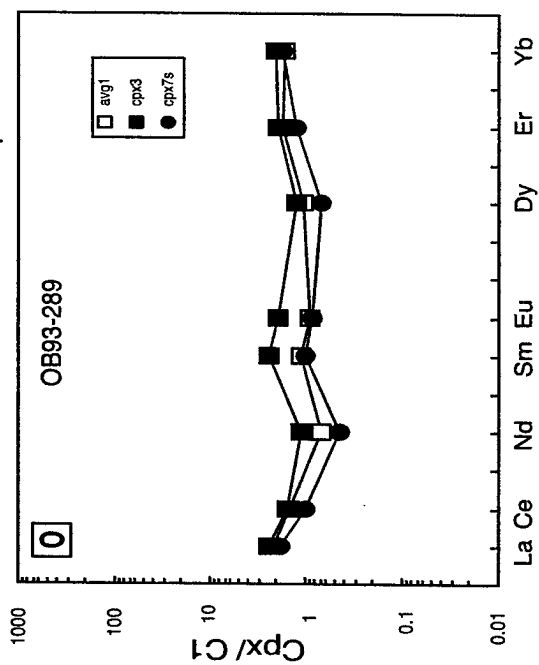
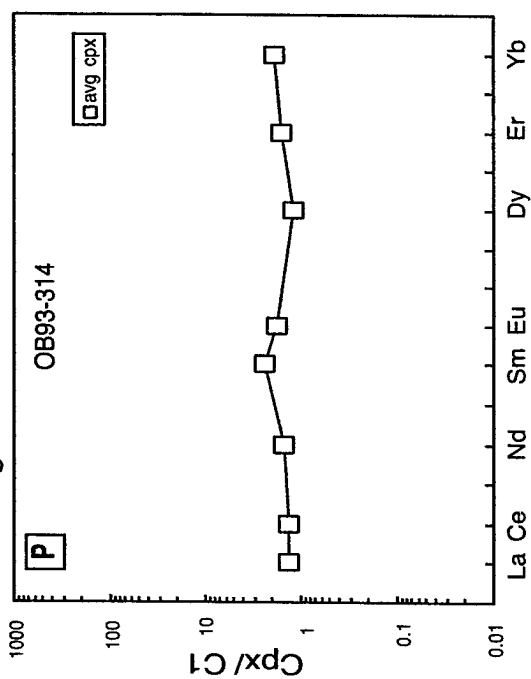
Lac Superior REE depleted harzburgites



Lac Superior REE depleted harzburgites



Mont Trapeze REE depleted harzburgites



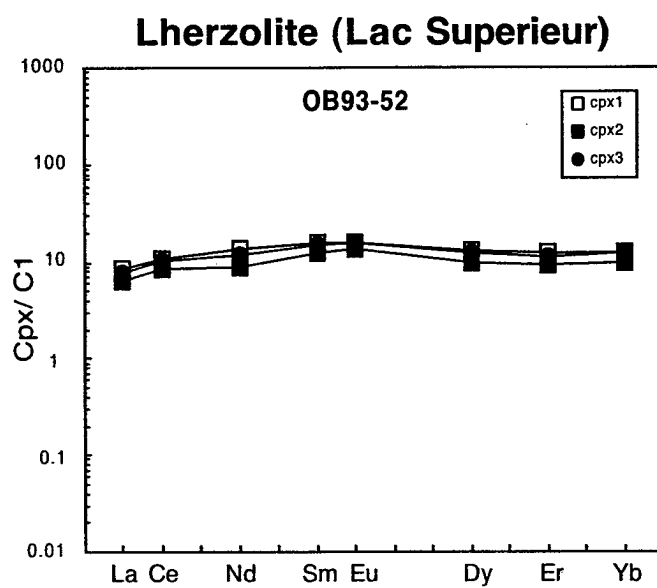


Figure 2-16. Normalized REE patterns of the lherzolite.

Figure 2-17. Normalized REE abundances of the wehrlitic-dunitites.

Dunites (Lac Superieur)

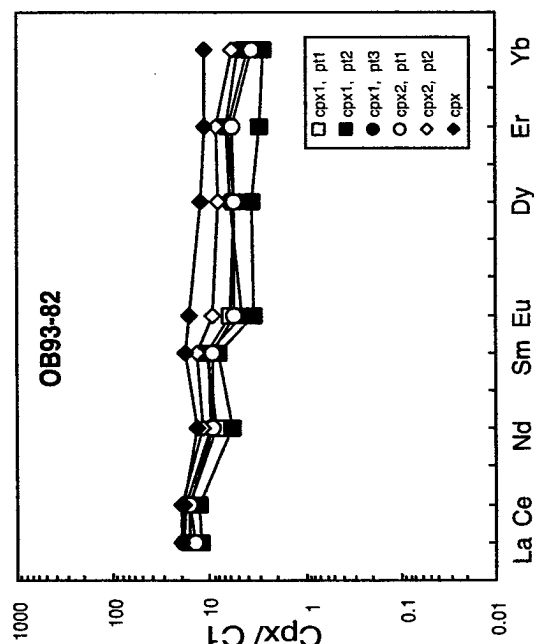
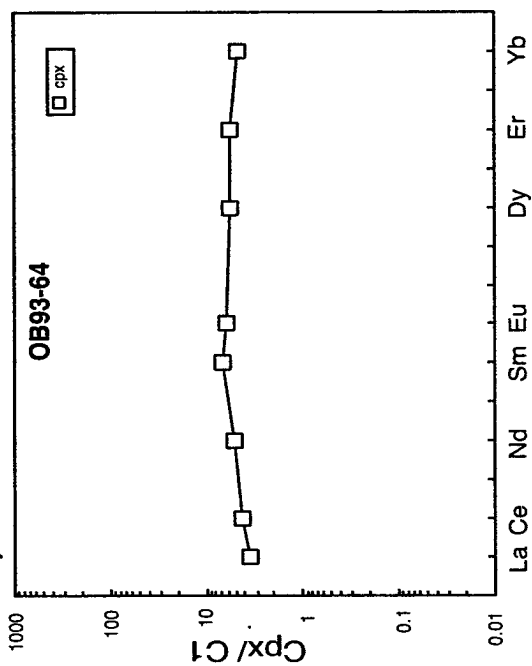
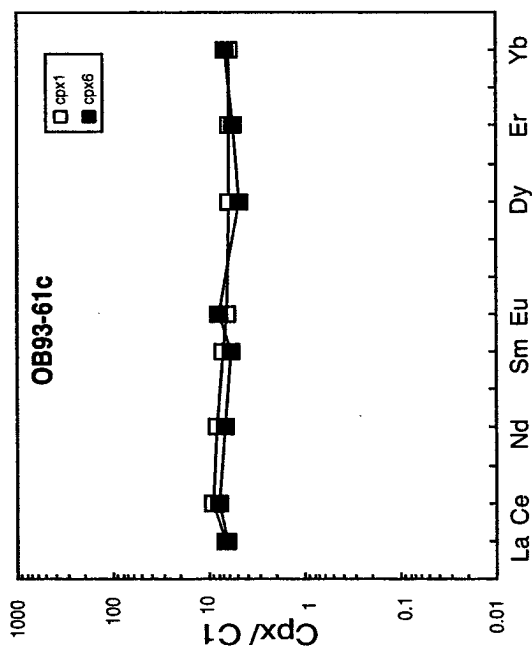
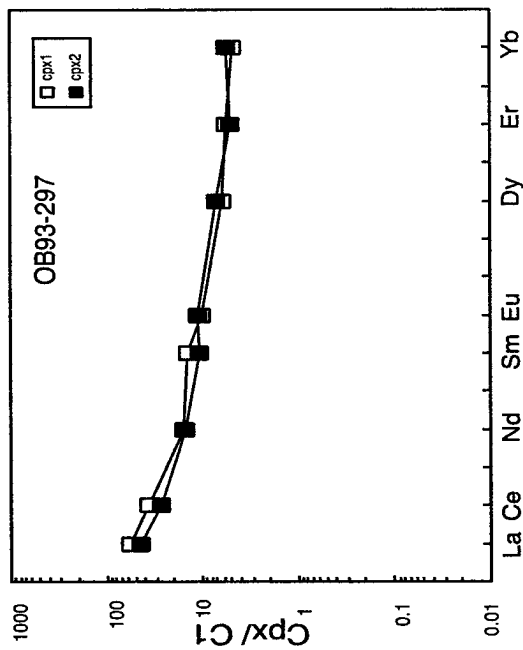
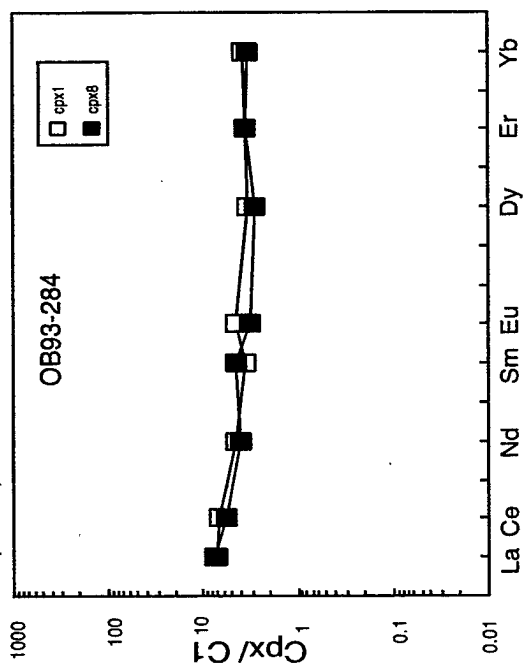
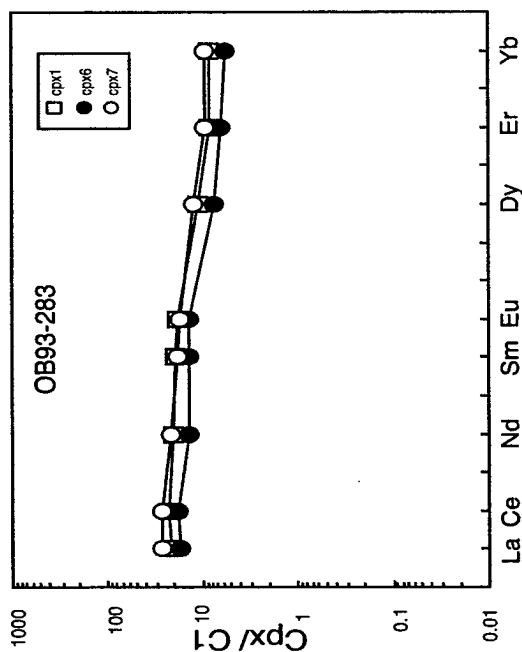


Figure 2-18. Normalized REE patterns of clinopyroxenes from the REE enriched harzburgites.

REE enriched harzburgites (Mont Trapeze)



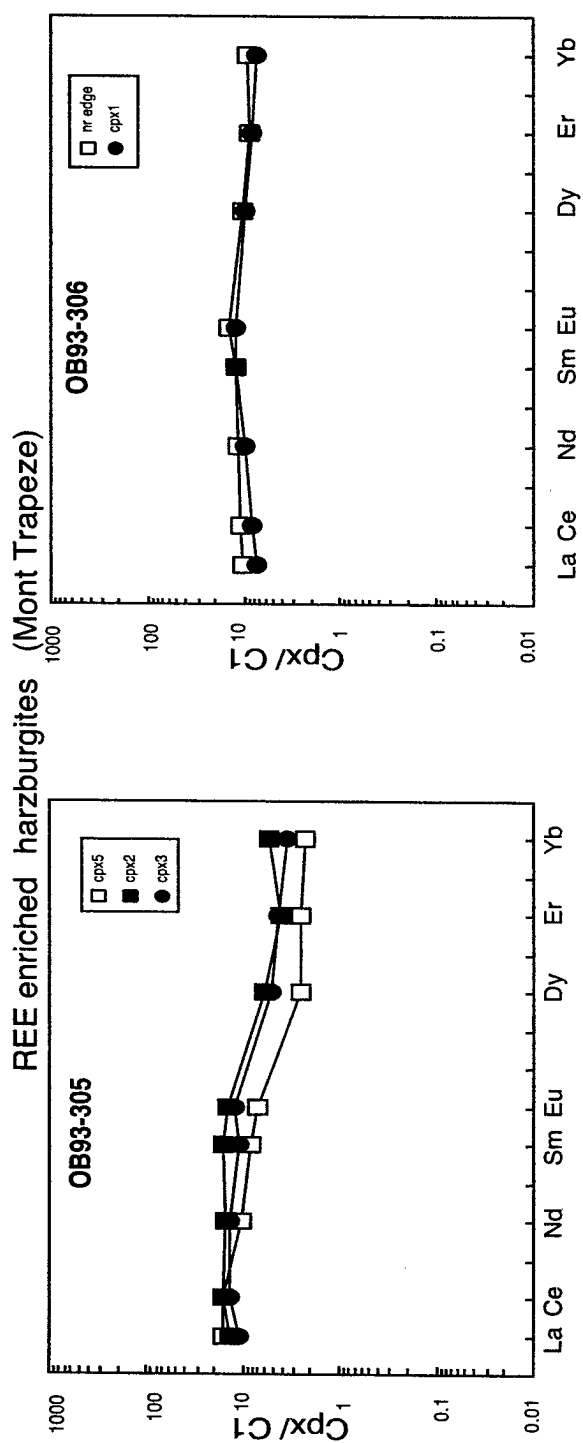
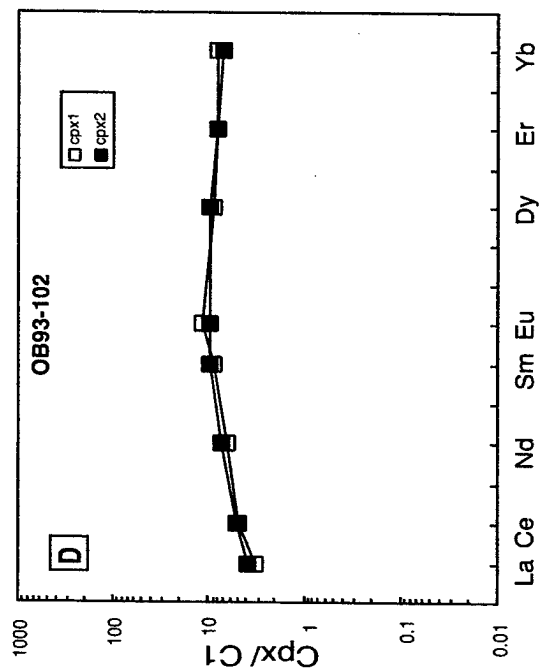
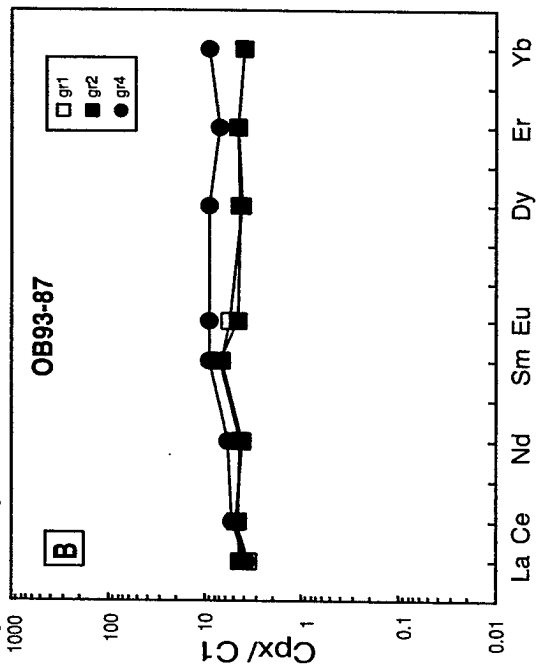
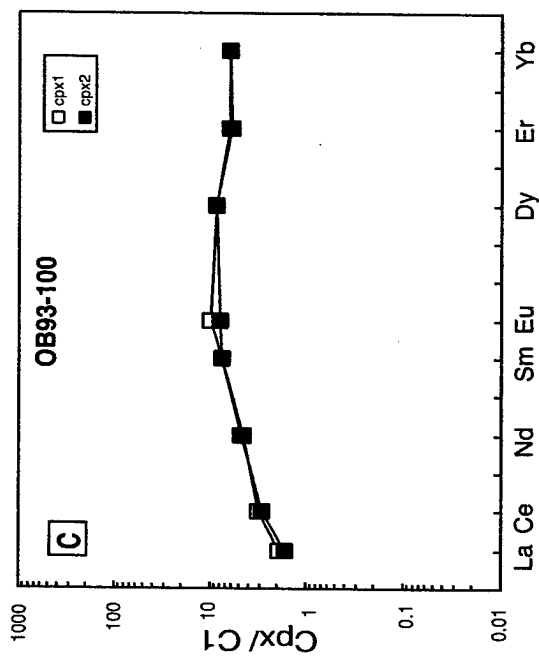
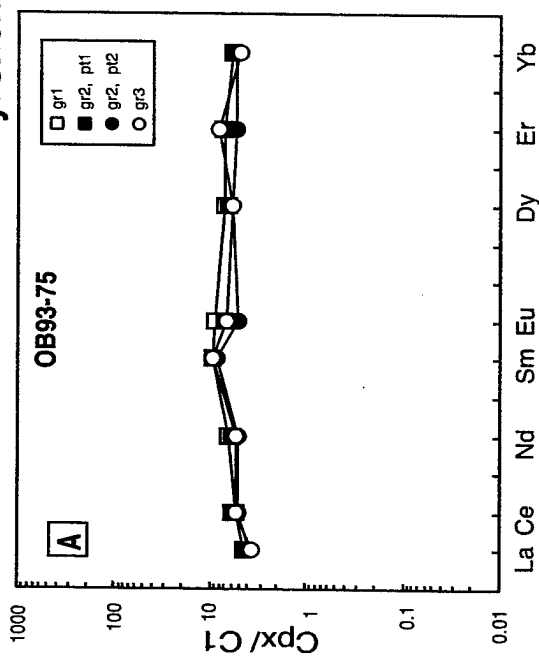


Figure 2-19. Normalized REE abundance patterns of clinopyroxenes from pyroxenites.

Pyroxenites (Lac Supérieur)



Phlogopite-bearing pyroxenites (Lac Superieur)

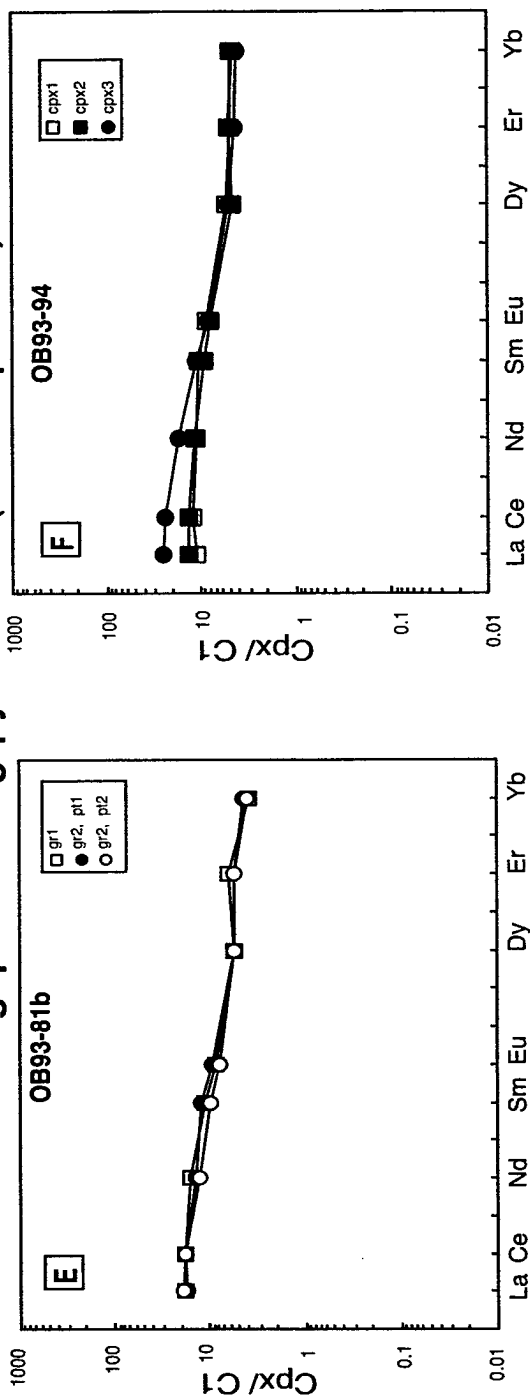
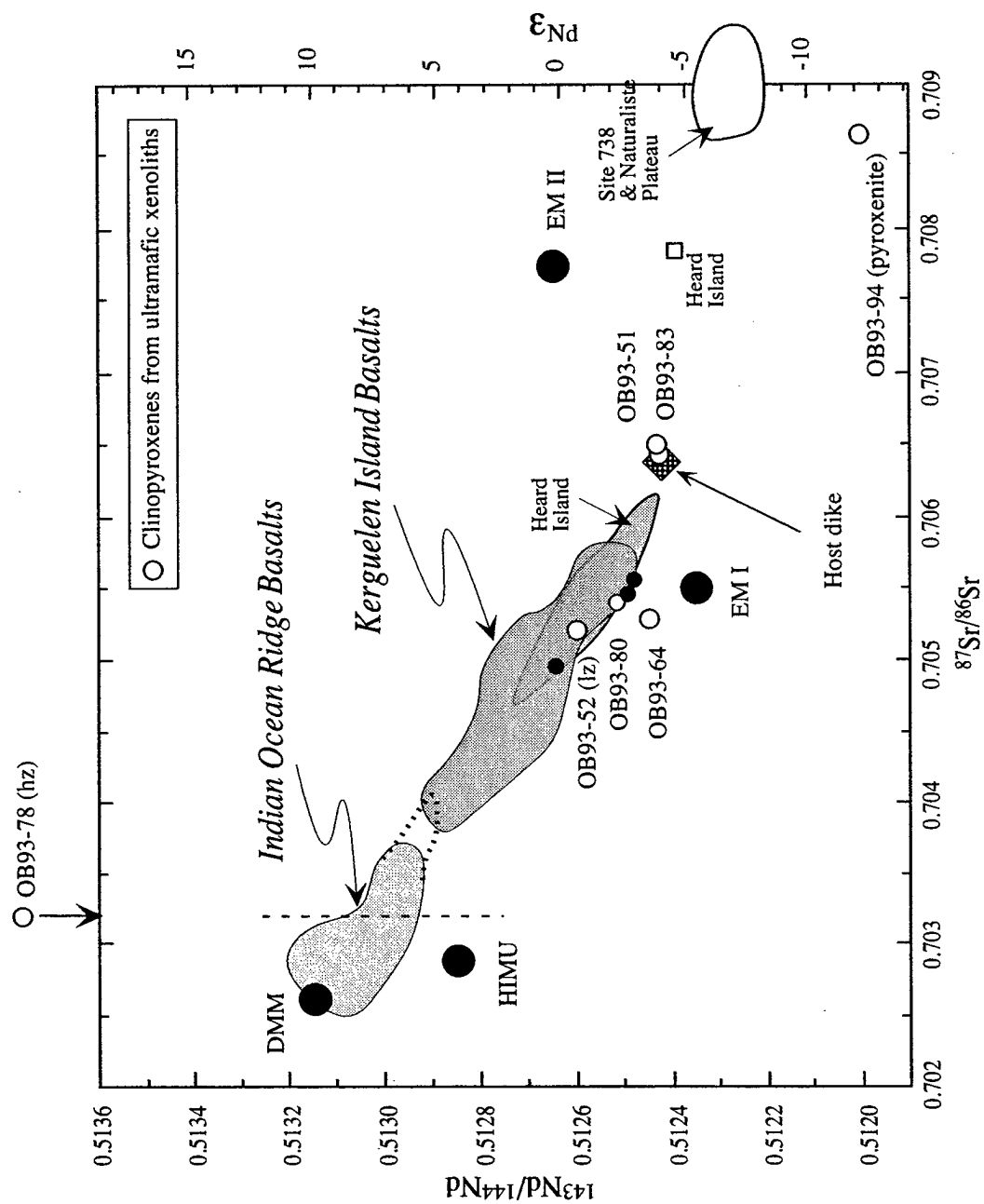


Figure 2–20. Nd v. Sr isotopic diagram of clinopyroxenes and host basalt from the Kerguelen xenolith suite. Clinopyroxenes from xenoliths from the Southeast Province are also shown for reference (small closed circles). Fields for the Indian Ocean MORB, Kerguelen Islands basalts, Heard Islands basalts and Site 735B and the Naturaliste Plateau are shown as shaded fields (from Alibert, 1991; Dosso et al., 1988; Gautier et al., 1990; Mahoney et al., 1994; Mattielli et al., 1996; Mattielli et al., 1999; Weis et al., 1993; Yang et al., 1998a).



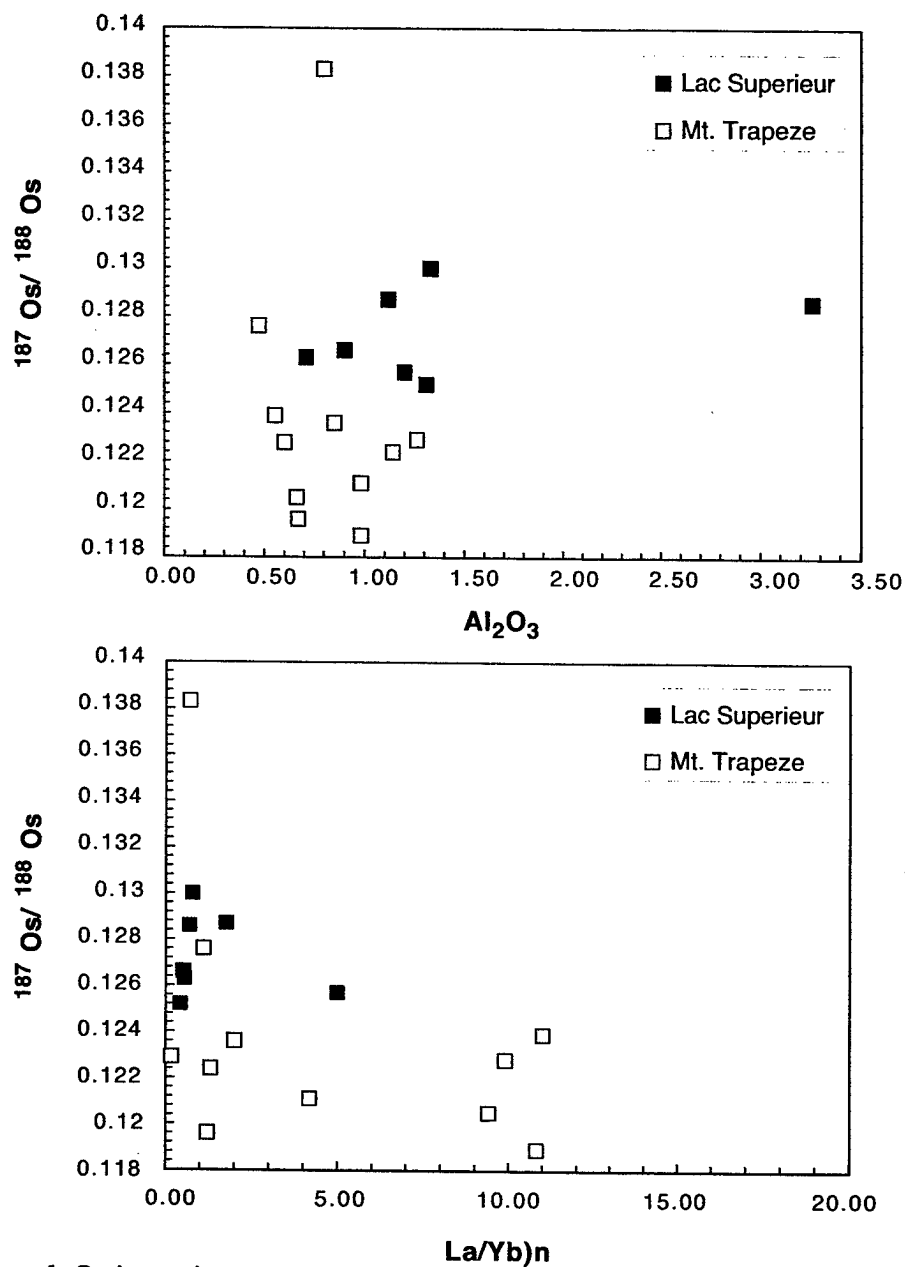


Figure 2-21. Whole rock Os isotopic compositions v. whole rock Al_2O_3 (upper panel) and clinopyroxene $[\text{La}/\text{Yb}]_N$ (lower panel). No correlations are present among these parameters, indicating that metasomatism did not affect the Os isotopes.

Chapter 3 Osmium Isotopic Evidence for Ancient Subcontinental Lithospheric Mantle Beneath the Kerguelen Islands, Southern Indian Ocean

3.1 Abstract

Upper mantle xenoliths found in ocean island basalts are an important window through which the oceanic mantle lithosphere may be viewed directly. Osmium (Os) isotopic data on peridotite xenoliths from the Kerguelen Islands, an archipelago that is located on the northern Kerguelen Plateau in the southern Indian Ocean, demonstrate that pieces of mantle of diverse provenance are present beneath the Islands. In particular, peridotites with unradiogenic Os and ancient rhenium (Re)-depletion ages (to 1.36 Ga old) may be pieces of the Gondwanaland subcontinental lithosphere that were incorporated into the Indian Ocean lithosphere as a result of the rifting process.

3.2 Os evidence for subcontinental lithosphere

The Kerguelen hotspot became active at least 115 Ma (1), and possibly as long as 130 Ma, when India separated from Australia-Antarctica (2). The Kerguelen mantle plume is considered to be responsible for extensive hotspot volcanism in the Indian Ocean basin (3), including the Ninetyeast Ridge, Broken Ridge, as well as the Kerguelen Plateau, the second largest of the oceanic plateaus (Fig. 1). Moreover, the Kerguelen plume is a carrier of the DUPAL isotopic anomaly (4), a characteristic of basalts erupted in the Southern Hemisphere. The interaction of this plume with the Indian Ocean lithosphere has potentially affected the geochemical and tectonic evolution of the entire Indian Ocean basin (2, 5).

One notion for the origin of the Kerguelen Plateau and the Kerguelen Islands is that their lithospheric basement contains a continental fragment left behind from the breakup of

Gondwanaland (6). Basalts from the southern Kerguelen Plateau (ODP drill-site 738) do have trace element and isotopic signatures indicative of a continental component in their source (7) and recent seismic experiments showed that the southern Kerguelen Plateau has a structure similar to a continental passive margin (8). However, a seismic refraction study, as well as early isotopic work on basalts from the archipelago, established an apparent oceanic origin (6, 9). Here we present Os isotopic data that establishes the presence of continental lithosphere in the northern Kerguelen Plateau and the involvement of continental materials in the formation of the Kerguelen Islands.

We examined the Re-Os isotopic systematics in peridotite xenoliths (10) from two sample localities that are less than 10 km apart (Lac Superieur and Mont Trapeze) in the Courbet Peninsula, northeastern Kerguelen Islands (Fig. 1 inset). The xenoliths are abundant in alkaline basaltic dikes that cross-cut "plateau basalt" lavas that comprise the Courbet Peninsula and formed 23 to 25 Ma (11).

The samples are predominantly refractory coarse granular harzburgites [olivine: ~60 % (forsterite content = 90-92%), orthopyroxene: ~35 %, clinopyroxene: 1-5 %, and trace amounts of spinel], but also include more clinopyroxene-rich lherzolites and wehrlitic-dunites. The rocks are fresh and do not contain any secondary alteration minerals. Many of the peridotites (from both localities) reacted with melts that percolated through them. For example, some harzburgites contain newly formed clinopyroxene crystals that are enriched in incompatible trace elements (for example, chondrite normalized $La_N = 100$), whereas coexisting original clinopyroxene grains occur as symplectite intergrowths with both orthopyroxene and spinel, and have depleted incompatible trace element abundances (for example, $La_N = 3$). Some peridotites at more advanced stages of reaction have trace element enriched clinopyroxenes (for example, $La_N = 300$) that are intergrown with phlogopite (12).

During interactions with percolating metasomatic melts, the Sr and Nd isotopic characteristics of the original peridotites are likely to be replaced by those of the melt, because the concentrations of these incompatible elements in the melt are much higher than in the peridotite. For instance, Sr isotopic compositions of clinopyroxenes from the few unmetasomatized peridotite xenoliths (for example OB93-78: $^{87}\text{Sr}/^{86}\text{Sr} = 0.70329 \pm 2$) are similar to those of Indian mid-ocean ridge basalts [MORB, (13)], whereas metasomatic clinopyroxenes have Sr isotopic compositions (for example OB93-83: $^{87}\text{Sr}/^{86}\text{Sr} = 0.706427 \pm 10$) similar to those of alkaline basalts from the Kerguelen Islands (6), reflecting their origin from Kerguelen plume melts (12, 14). Thus, the incompatible geochemical tracers record mantle processes associated with plume-lithosphere interaction, but information about xenolith provenance is lost in the xenoliths that have undergone metasomatic reaction.

In contrast, the isotopic compositions of elements that are compatible in peridotites are less likely to be modified during metasomatism because their concentrations are much higher in peridotites than in the percolating melts. Osmium is a compatible element during partial melting of peridotite (15); therefore, ^{187}Os , the radiogenic daughter of ^{187}Re , is an isotopic tracer that is relatively insensitive to subsequent mantle metasomatic processes. For these reasons, the Re-Os system is particularly suitable for determining the provenance of peridotites and for dating primary melting events of peridotite formation (16).

The Os isotopic compositions of peridotite xenoliths from the Kerguelen Islands (Table 1, Fig. 2) vary over a wide range ($^{187}\text{Os}/^{188}\text{Os} = 0.1189 \pm 2$ to 0.1383 ± 6) and extend to compositions that are less radiogenic than have been observed in rocks from oceanic settings, including xenoliths from other ocean islands [for example, Samoan peridotites have $^{187}\text{Os}/^{188}\text{Os} = 0.1225$ to 0.1309 , (17)]. Re/Os ratios in the xenoliths are low (2×10^{-3} to 80×10^{-3} ; Table 1), and age correction for isotopic ingrowth since the time of the eruption

of the host basalts is negligible. The more radiogenic Os isotopic compositions of this xenolith suite cannot be explained by contamination with Kerguelen Islands basalt because unrealistic additions of basalt would be required [$>20\%$; with $^{187}\text{Os}/^{188}\text{Os} = 0.1565$, and $[\text{Os}] = 100 \text{ ppt}$ (18)]. Thus, we interpret these variations as primary signatures that reflect the diversity of the xenolith provenance.

Within the sample suite, two spinel harzburgites from Mont Trapeze, and three comparatively clinopyroxene-rich lherzolites from Lac Supérieur, have relatively radiogenic Os isotopic compositions ($^{187}\text{Os}/^{188}\text{Os}$ of 0.1276 ± 5 to 0.1383 ± 6). These values are typical of ocean island basalts (19-23), and may reflect a part of the spectrum of Os isotopic compositions of the Kerguelen plume.

The remaining xenoliths from Lac Supérieur are grouped at $^{187}\text{Os}/^{188}\text{Os} \approx 0.1260 \pm 7$, within the range of compositions estimated for the modern oceanic lithosphere from abyssal peridotites (19, 24). Indeed, some clinopyroxenes in samples of this group have rare earth element abundance patterns comparable to those in abyssal peridotites, and Sr isotopic compositions of clinopyroxenes from these rocks are similar to Indian Ocean MORB values as noted above (12). These characteristics are evidence that these xenoliths originated from the Indian Ocean lithospheric mantle.

In contrast, the harzburgites from Mont Trapeze exhibit a greater range in Os isotopic composition, and most are significantly less radiogenic (to $^{187}\text{Os}/^{188}\text{Os} = 0.1189 \pm 2$) than the xenoliths from Lac Supérieur. Such unradiogenic Os isotopic compositions require reservoirs that had extremely low Re/Os ratios for geologically long periods of time. The Re-depletion model ages (16, 25) for the Mont Trapeze xenoliths are from 0.58 to 1.36 Ga (T_{RD} ; Table 1). Such unradiogenic Os isotopic compositions have never been observed in oceanic environments, and are not expected to be part of the young

(~40 Ma beneath the Kerguelen Islands) Indian Ocean mantle lithosphere. Comparably unradiogenic Os isotopic compositions have only been found in ancient subcontinental lithospheric mantle peridotites such as orogenic massifs (26) and xenoliths from cratonic areas (16, 27).

We suggest that these anomalous unradiogenic xenoliths are pieces of the Gondwanaland subcontinental lithosphere. That the unradiogenic xenoliths are found in only one of the two localities suggests that the pieces of subcontinental lithosphere could be small and localized and thus undetectable in geophysical observations of the northern Kerguelen Plateau (9).

A possible mechanism for emplacing small discrete fragments of subcontinental lithosphere into the middle of a young ocean basin is that subcontinental lithospheric mantle was delaminated and incorporated into the newly-forming Indian Ocean lithosphere during the rifting of the eastern Gondwanaland continents. A similar process may explain the unusual geochemical characteristics of some basalts from the Southwest Indian Ridge (SWIR, 39°E to 41°E), the Afanasy-Nikitin Rise in the northern Indian Ocean, the Oceanographer fracture zone on the Mid-Atlantic Ridge, the geochemical signatures of basalts from the Azores, as well as the origin of the DUPAL anomaly (21, 28). This process is consistent with the seismic structure of the southern Kerguelen Plateau (8). If delamination is common in continental rifting (29), it could contribute to mantle heterogeneities that are manifested in basalt geochemistry (21, 28).

Another possibility is that eastern Gondwanaland subcontinental lithospheric mantle was transported to the Kerguelen Plateau by the Kerguelen plume. In this scenario, the convecting incipient plume-head eroded the base of the Gondwanaland lithosphere, pieces of which were then integrated into the Indian Ocean lithosphere. The earliest volcanism

that can be related to the Kerguelen plume with certainty is the southern Kerguelen Plateau, which formed at 115 Ma (1, 30). At 115 Ma, the Kerguelen Plateau was forming within the young Indian Ocean basin more than 500 km from the Gondwanaland continental margins (*see* 2, 31). In this case, an unusually large plume head (diameter = 1000 km) would be required to erode and entrain the Gondwanaland lithosphere into the Kerguelen plume. If the 130 Ma Bunbury Basalts (West Australia) and Rajmahal Traps (India) were identifiable as early Kerguelen plume activities, the requirement for such a large plume head may not be necessary. However, a genetic relationship of these rocks to the plume is still controversial (30). On the basis of these arguments, we consider that this model is unlikely.

Alternatively, these unradiogenic xenoliths could be pieces of Precambrian oceanic mantle lithosphere recycled through a subduction zone and returned to the lithosphere by way of the Kerguelen plume. That peridotites with unradiogenic Os isotopic compositions have not yet been observed from any other oceanic islands suggests that this hypothesis, too, is unlikely.

Our present data suggest that mantle of diverse provenance is present beneath the Kerguelen Islands. The results also indicate that the northern part of the Kerguelen Plateau is partially composed of lithospheric mantle derived from the Gondwanaland continents and is not completely oceanic in origin.

3.3 References and Notes

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$$T_{MA} = \frac{1}{\lambda} \cdot \ln \left[\frac{\left(\frac{{}^{187}\text{Os}}{{}^{188}\text{Os}} \right)_{\text{Mantle}} - \left(\frac{{}^{187}\text{Os}}{{}^{188}\text{Os}} \right)_{\text{Sample}}}{\left(\frac{{}^{187}\text{Re}}{{}^{188}\text{Os}} \right)_{\text{Mantle}} - \left(\frac{{}^{187}\text{Re}}{{}^{188}\text{Os}} \right)_{\text{Sample}}} + 1 \right]$$

where λ = the decay constant for ${}^{187}\text{Re}$ [1.59×10^{-11} ; M. Lindner, et al., *Nature* **320**, 246 (1986)] and $({}^{187}\text{Os}/{}^{188}\text{Os})_{\text{Mantle}} = 0.12757$ and $({}^{187}\text{Re}/{}^{188}\text{Os})_{\text{Mantle}} = 0.39716$ from the mean of carbonaceous chondrites as originally defined by Walker et al., (16). This age is a minimum because it assumes that Re is completely removed during the melting event that formed these rocks. There is some uncertainty as to the exact Re/Os ratio of Earth's mantle. Present day $({}^{187}\text{Os}/{}^{188}\text{Os})_{\text{Mantle}}$ may be as unradiogenic as some abyssal peridotites [~ 0.122 (24)] in which case the model ages decrease, or more radiogenic as estimated from fertile xenoliths [~ 0.1290 ; T. Meisel, R. J. Walker, J. W. Morgan, *Nature* **383**, 517 (1996)], where model ages would be even older. Most estimates of $({}^{187}\text{Os}/{}^{188}\text{Os})_{\text{Mantle}}$ are between 0.124 and 0.130 (19, 24), and the use of either extreme does not affect our conclusions. Ancient $({}^{187}\text{Os}/{}^{188}\text{Os})_{\text{Mantle}}$ is better defined because of iron meteorite initial ratios [M. F. Horan, J. W. Morgan, R. J. Walker, J. N. Grossman, *Science* **255**, 1118 (1992)], and because ingrowth of ${}^{187}\text{Os}$ was still small. The T_{MA} for most samples are within the estimates for T_{RD} because of the samples' low Re contents. Ages were omitted for those samples giving future ages, or samples whose ${}^{187}\text{Os}/{}^{188}\text{Os}$ are similar to the estimated range of $({}^{187}\text{Os}/{}^{188}\text{Os})_{\text{Mantle}}$.

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33. Our samples were collected during the 1992 to 1993 austral field season. We thank A. Giret, F. Frey, D. Weis, and J.-Y. Cotin for assistance in sampling. We also thank G. Ravizza for discussions, S. Hart for sharing of laboratory facilities, J. Blusztajn for advice, K. Burrhus for technical support, E. Nakamura for Sr isotopic analysis of OB93-78 cpx, and F. Frey and M.F. Roden for comments. NS acknowledges support from the National Geographic Society (4629-91), and the National Science Foundation (EAR-9219158 and OPP -9417806). This research is also supported by a Cecil and Ida Green Fellowship to DRH.

3.4 Tables

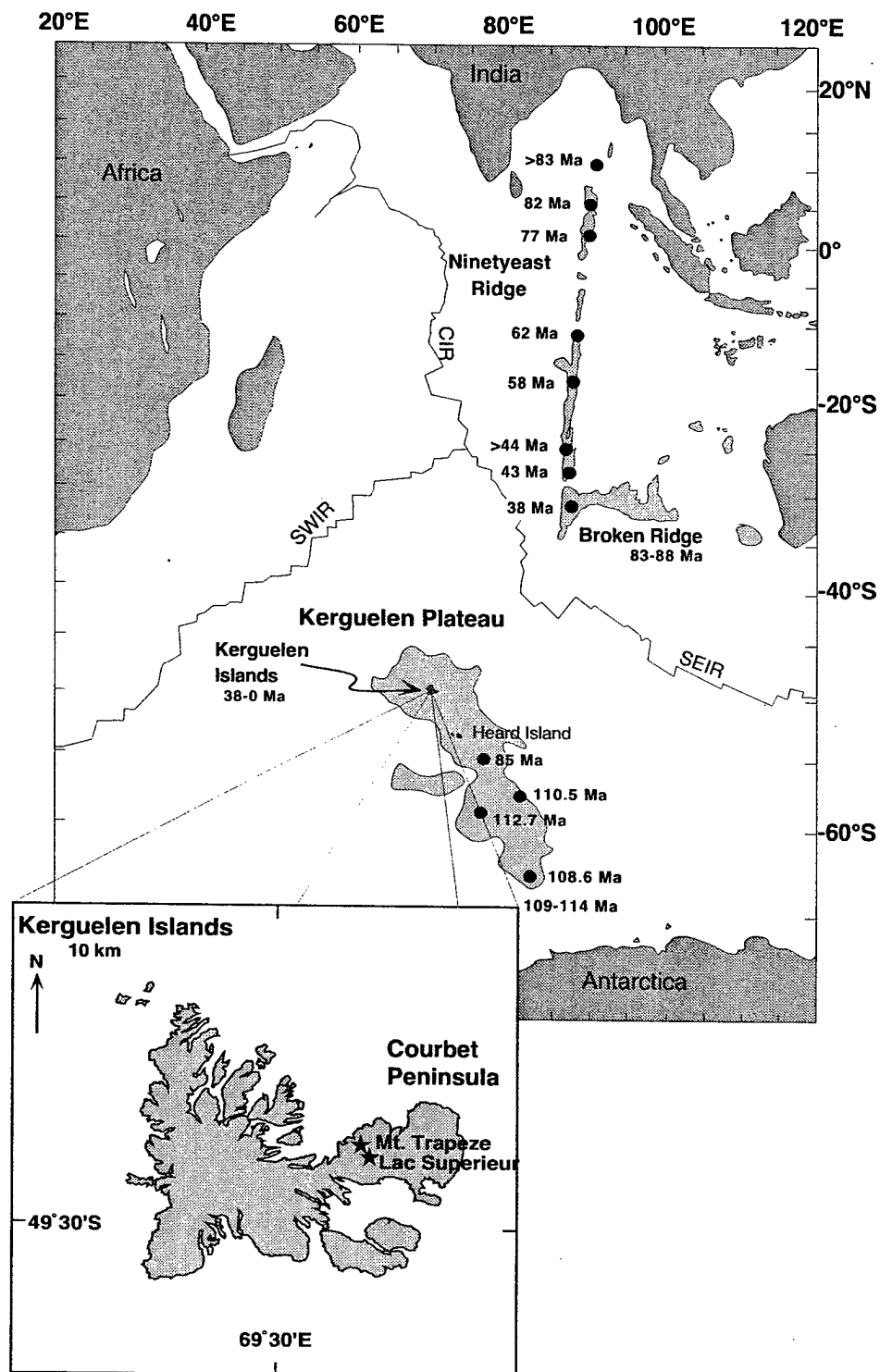
Table 3-1. Rhenium and osmium isotopic results from Kerguelen Islands peridotite xenoliths. All samples are spinel harzburgites except as noted.

Sample	$^{187}\text{Os}/^{188}\text{Os}$	Os (32) (ppt)	Re (ppt)	Re/Os ($\times 10^{-3}$)	T_{RD} (25) (Ga)
Lac Superieur					
OB93-51*	0.1262 $\pm$ 2	87	n.d.		
OB93-52†	0.1286 $\pm$ 3	4218	n.d.		
OB93-64†	0.1300 $\pm$ 2	968	n.d.		
OB93-77	0.1263 $\pm$ 3	3199	5	2	
OB93-78	0.1266 $\pm$ 6	2888	n.d.		
OB93-80	0.1252 $\pm$ 2	1727	n.d.		
OB93-82†	0.1287 $\pm$ 2	6999	n.d.		
OB93-83*§	0.1257 $\pm$ 4	44	n.d.		
Mt. Trapeze					
OB93-280	0.1189 $\pm$ 2	3067	6	2	1.36
OB93-284	0.1236 $\pm$ 1	6652	29	4	0.63
OB93-284bII	0.1229 $\pm$ 3	6645	n.d.		0.74
OB93-287	0.1239 $\pm$ 3	1267	49	39	0.58
OB93-289	0.1224 $\pm$ 3	3411	12	4	0.81
OB93-291	0.1205 $\pm$ 4	2157	11	5	1.11
OB93-297	0.1228 $\pm$ 3	2409	31	13	0.75
OB93-305	0.1211 $\pm$ 3	2012	72	36	1.02
OB93-305rII	0.1202 $\pm$ 4	2179	n.d.		1.16
OB93-306	0.1276 $\pm$ 5	516	41	80	
OB93-307	0.1229 $\pm$ 3	5605	49	9	0.74
OB93-314	0.1383 $\pm$ 6	3281	9	3	
OB93-317	0.1196 $\pm$ 4	2652	11	4	1.25

* olivine separate; † wehrlitic dunite; ‡ spinel lherzolite; § phlogopite-bearing spinel lherzolite; II replicate; n.d. not determined.

3.5 Figures

Figure 3-1. Map of the Indian Ocean and Kerguelen Islands. Submarine plateaus and ridges related to the Kerguelen Plume are highlighted in light gray shading, and the locations of the SWIR and SEIR are shown. Filled circles correspond to Deep Sea Drilling Project (DSDP) and Ocean Drilling Project (ODP) drill sites. The ages of recovered basaltic basement for each drill-hole are adjacent to the symbols. The Kerguelen Islands (inset), are located at the northern end of the Kerguelen Plateau. On the Courbet Peninsula, filled stars mark the two xenolith localities of this study, Lac Superieur and Mont Trapeze.



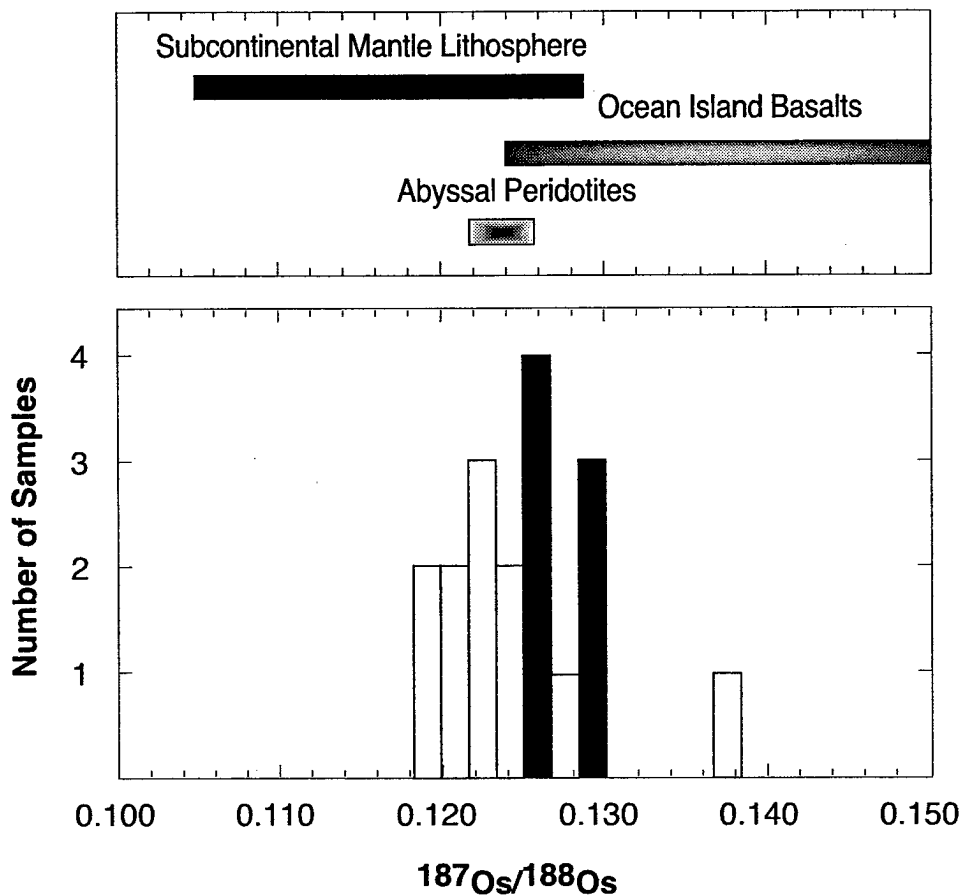


Figure 3-2. Histogram of the Os isotopic composition of the Kerguelen Islands peridotite xenoliths by location. Most samples from the Lac Superieur locality (black) are more radiogenic than most of those from the Mont Trapeze locality (white). The typical error is less than the width of a bar. Shown for comparison, is the range of initial Os isotopic compositions of other rock types from around the world (16, 19-24, 26, 27). Only data for abyssal peridotites that are not contaminated with seawater Os are shown.

Chapter 4 Trace Element and Sr, Nd and Os Isotopic Composition of Mafic Granulite Xenoliths from the Courbet Peninsula, Kerguelen Islands

4.1 Introduction

The Kerguelen Islands are situated in the southern Indian Ocean atop the Kerguelen Plateau, the second largest of the giant oceanic plateaus. The Kerguelen Plateau and Islands, along with other large bathymetric features of the Indian Ocean (the Ninetyeast Ridge and Broken Ridge, Figure 1-1) are results of basaltic hotspot volcanism that is attributed to the long-lived Kerguelen mantle plume (Mahoney et al., 1983; Storey et al., 1989). Plume-related basalts (and included xenoliths) have been the subject of many comprehensive studies (Watkins et al., 1974; Dosso et al., 1979; Gautier et al., 1990; Weis et al., 1993; Yang et al., 1998; Grègoire, 1994; Mattielli, 1996; Hassler and Shimizu, 1998), two Deep Sea Drilling Project (DSDP) legs (Davies et al., 1974; von der Borch et al., 1974), and three Ocean Drilling Program (ODP) drilling legs [Leg 119 (Barron et al., 1991), Leg 120 (Wise et al., 1992), and Leg 183 (Dec., 1998 – Feb., 1999)] have sampled the basaltic basement of the plateau. But, mostly because of its huge size and remote location, large areas of the 2×10^6 km² Kerguelen Plateau remain unsampled and unsurveyed.

It is not completely known how these large igneous provinces (LIP) form, but it is clear that they are not a result of normal seafloor spreading and require a mantle plume to

produce such voluminous magmatism (Coffin and Eldholm, 1994). However, what role plume-ridge or plume-lithosphere interaction plays in oceanic plateau formation is also not understood. For example, the Ontong Java Plateau was near an oceanic spreading center early in its history, but a large part of it was emplaced off-axis (Neal et al., 1997).

Similarly, the history of the Kerguelen LIP is believed to have included a spreading ridge-centered location for the Kerguelen plume, but its timing and details of tectonic history are not clearly delineated.

Mafic granulites are part of the Kerguelen Islands xenolith suite (Grègoire, 1994). This occurrence is quite unusual; pressure-temperature conditions for the oceanic crust are generally outside those for granulite facies metamorphism. In fact, the Kerguelen xenolith suites represent the first and only occurrence of granulite facies rocks in the oceanic environment (Grègoire et al., 1994). These granulite xenoliths are pieces of the plateau lower crust (Grègoire, 1994), and provide an excellent opportunity to study an inaccessible part of the plateau. The geochemistry of these rocks could contain information about the geodynamic evolution of the Kerguelen plume inaccessible through erupted lavas and its relationship to the formation of the Kerguelen Plateau. The crustal thickness of the Kerguelen Plateau generally increases from approximately 17 km in the immediate vicinity of the Archipelago to as much as 26 km in the south (Recq and Charvis, 1986). Charvis et al. (1995) however, report that the crustal thickness could be variable between 17 and 20 km beneath the Kerguelen Islands and could be as much as 25 km in the Plateau between Kerguelen and Heard Islands. The southern Kerguelen Plateau has a seismic structure similar to that of a volcanic passive margin (Recq and Charvis, 1986; Operto and Charvis, 1995), and though geochemical evidence exists for discrete pieces of subcontinental lithosphere in the northern plateau (Chapter 2; Chapter 3; Hassler and Shimizu, 1998), the thickened crust here was interpreted to have an oceanic origin (e.g. Houtz et al., 1977;

Recq and Charvis, 1986). The Ontong Java Plateau, and other oceanic islands such as Hawaii and the Marquesas also lie above a thickened oceanic layer 3 (e.g. Furumoto et al., 1976; Gladchenko et al., 1997; ten Brink and Brocher, 1987; Caress et al., 1995). In the case of the Marquesas, Caress et al. (1995) estimated that up to twice the volume erupted at the surface was present in the lower crust. Hence, this thickening of the lower oceanic crust is an important process in ocean island volcanism as well as in LIP formation. A study of such lower crustal rocks can help to answer questions about the origin of the Kerguelen Plateau in particular, and oceanic plateaus in general.

This study expands on the previous major and trace element and petrographic work of Grègoire et al. (1994; 1998) and the isotopic study of Mattielli et al. (1996) and Mattielli (1996), all of which concentrated on the xenoliths from localities in the southeastern province of the Islands. The xenoliths of this study are from the Lac Superieur location to the north on the Courbet Peninsula (Figure 1–1). Some of the granulites studied by Grègoire et al. (1994; 1998), Mattielli et al. (1996) and Mattielli (1996) contain garnet in a typical mafic garnet granulite mineral assemblage (clinopyroxene + orthopyroxene + plagioclase + garnet $\pm$ ilmenite). A smaller number of samples contain sapphirine $[(\text{Mg,Fe})_2\text{Al}_4\text{O}_6\text{SiO}_4]$; Grègoire et al. (1994) reported the first occurrence of this mineral in an oceanic setting. The presence of these minerals helps constrain pressure-temperature conditions of metamorphism. The samples of the present study are of a typical mafic pyroxene granulite assemblage (clinopyroxene + orthopyroxene + plagioclase $\pm$ ilmenite) without garnet or sapphirine. One of the goals of this project is to compare the lower crust of the Islands and the Plateau as represented by the xenoliths from these two areas.

A major objective of this work is to determine the provenance of the granulites. Because of the tectonic history of the Indian Ocean and the presence of subcontinental mantle lithosphere beneath the Kerguelen Islands (Chapter 2; Chapter 3; Hassler and

Shimizu, 1998), the possibility that these xenoliths have an origin in the continental lower crust cannot be precluded. In this chapter, the major element, trace element and Sr, Nd and Os isotopic compositions of the Kerguelen granulites will be compared to those of mafic continental granulites in order to discuss their possible origin as pieces of continental lower crust. As it turns out, a simple comparison of elemental abundances is inadequate to constrain the origin of the Kerguelen granulites. It will be shown below that Os isotopes are the best indicators that these xenoliths have an oceanic rather than continental origin. Given that these granulites have an oceanic origin, the question remains: are the granulites recrystallized cumulates of melts that are related to the Southeast Indian Ridge (SEIR) mid-ocean ridge basalts (MORB), the Kerguelen plume, or some mixture of these two? Further, is there a geochemical signature of contamination of their parent melts by continental crust?

These rocks could also serve as a geochemical record of plume-ridge and plume-lithosphere interaction that one may use in an attempt to unravel part of the history of the Kerguelen plume. The argument presented here is that the origin of these xenoliths is most likely from an isotopically heterogeneous Kerguelen plume, and that a continental origin is unlikely. Though an origin as a plume-MORB mixture cannot be completely discounted, it is clear that none of the rocks formed solely by mid-ocean ridge processes.

4.2 Tectonic Setting

Earlier studies divided the Kerguelen Plateau into two distinct structures, the Mesozoic southern plateau and Cenozoic northern plateau, with the break just south of the Kerguelen Islands (Figure 1-1) (Munschy and Schlich, 1987; Coffin and Eldholm, 1994). Broken Ridge and the conjugate southern and central Kerguelen Plateau erupted beginning in the Cretaceous and were later rifted from each other by seafloor spreading at the SEIR (Royer and Sandwell, 1989). Magnetic anomaly 18 (~42.7 Ma, Harland et al., 1989) is the oldest

identified lineation along the SEIR side of the south to central parts of the Kerguelen Plateau and the southern side of Broken Ridge (Munschy et al., 1992). Steeply dipping slopes, remnants of SEIR rifting, face the SEIR on both of these features, but there is a gentle slope towards the abyssal seafloor that looks like volcanic flows at the northernmost Kerguelen Plateau. Anomaly 9 is the oldest that is clearly present along this part of the Plateau. Anomaly 13 is tentatively identified in this area, but anomalies 18-13 (42.7-35 Ma; Harland et al., 1989) may actually underlie the northern part of the plateau (Munschy et al., 1992). Moreover, plate tectonic reconstructions give an overlap between the northern part of the Kerguelen Plateau and the southern extension of Broken Ridge (Royer and Sandwell, 1989). Thus, the northern Kerguelen Plateau formed at least after anomaly 18 time, and portions of it must be no older than 35 Ma. The ages of submarine rocks from the northern Kerguelen Plateau are not known.

The oldest rocks on the Kerguelen Islands are from a 39 ± 3 Ma gabbroic pluton, the Val Gabbro, on the southeastern province (Giret and Lameyre, 1983). These rocks are somewhat zeolitized and age determinations were made using the K-Ar technique. However, K-Ar is not a preferred method for dating pyroxene-bearing plutonic rocks, especially those that contain these secondary minerals (Faure, 1986), and these ages may not be accurate. The next oldest known rocks from the archipelago are ~29 Ma plateau flood basalts (Nicolaysen et al., 1999). It is possible then, that the Kerguelen Islands did not start to form until well after 35 Ma, after the SEIR had moved away from the plume. At a spreading rate of 25-45 km/Ma (Munschy et al., 1992), the Kerguelen Islands (and the northern Kerguelen Plateau) could have been anywhere from 200-400 km away from the SEIR. This is complicated though, in that Kerguelen plume related basalts from DSDP site 254 on Broken Ridge give ages of 38 Ma (Duncan, 1978).

4.3 Sample Description

The granulite xenoliths of this study are from the Lac Supérieur locality on the Courbet Peninsula (Figure 1-1), and are part of a suite of xenoliths that also contains lherzolites, harzburgites, and pyroxenites. The xenoliths are included in a dike of unknown age that cross-cuts the 24.5 Ma plateau basalts (Nicolaysen et al., 1999) that make up the Courbet Peninsula.

The granulite xenoliths range in size from ~5 to 20 cm in the largest dimension. Most of the 11 rocks of this study are free of alteration, and are medium-grained and equigranular, with no relict igneous textures. They contain clinopyroxene (to 3.5 mm), orthopyroxene (to 1.4 mm), anorthitic plagioclase (to 1.2 mm), $\pm$ green spinel, $\pm$ ilmenite, and $\pm$ sulfides. Only one relict olivine grain was found, in sample OB93-71. Figure 4-1 shows photomicrographs of typical granulites. Some of the rocks contain strained pyroxene and plagioclase. Many have pyroxene exsolution, both orthopyroxene from clinopyroxene, and vice versa. Two samples, OB93-72 and OB93-90, have poikilitic orthopyroxene. Spinel and coarse symplectite textures (~400 μ m spinels) are present in the samples with higher whole rock Mg# (OB93-50, OB93-60, OB93-71, OB93-72, OB93-90, OB93-96), and very fine symplectites (2-10 μ m spinels), ilmenite and Ti-rich magnetite occur only in the lowest Mg# rocks (OB93-54, OB93-57, OB93-68, and OB93-69 –no thin section is available for OB93-61A). Symplectite textures are present as clinopyroxene + spinel and orthopyroxene + spinel intergrowths. Sulfides occur as rounded blebs included in clinopyroxene and along grain boundaries. They are generally Fe-rich, but Fe-Ni, Fe-Cu, Ni-Cu, and Cu-Zn-rich sulfides are also present. A summary of the petrographic characteristics of the xenolith suite are shown in Table 4-1.

4.4 Analytical Methods

Surfaces were cut from the xenoliths and saw marks were polished off with 120 and 240 grit silicon-carbide paper. The polished pieces were crushed to smaller fragments between plastic with a hammer. Pieces that did not have any basalt veins or visible alteration were selected under a binocular microscope for grinding in an agate shatterbox vial. Whole rock major and trace element compositions were determined on unleached powders by XRF and ICP-MS at the GeoAnalytical Laboratory at Washington State University. Precision for XRF major element analysis is $<0.4\%$ (1 S.D.). The precision of all trace element analyses by ICP-MS is 1-3%, except for U and Th, which are $\sim 10\%$ (1 S.D.).

Major element compositions of the rock-forming minerals in thin sections or grain mounts were determined with the JEOL 733 Superprobe electron microprobe at the Massachusetts Institute of Technology with an acceleration potential of 15 kV, a beam current of 10 nA and a 1 μm spot. Raw data were reduced online using the method of Bence and Albee (1968) modified by Albee and Ray (1970). The precision based on counting statistics are 6-25% (1 S.D.) for <0.5 wt.%, 2.5-6% for 0.5-1 wt.%, 1-3% for 1-15 wt.%, $<0.5\%$ for >15 wt.%.

The Cameca IMS 3f ion probe at WHOI was used to analyze trace element and rare earth element contents of clinopyroxenes following the methods of Shimizu and Hart (1982). A primary beam of O^- ions was focused to ~ 20 μm for REE (La, Ce, Nd, Sm, Eu, Dy, Er, Yb) and ~ 10 μm for other trace elements (Cr, V, Y, Ti, Zr, Sr). Molecular interferences were eliminated by energy filtering and a secondary voltage offset of -30 to -60V for the REE and -90V for the other trace elements. Uncertainties on the basis of

counting statistics were 5-10% (1 S.D.) for the REE and 1-5% (1 S.D.) for the other trace elements. All normalized REE data are reported relative to abundances in C1 chondrites (Anders and Grevesse, 1989).

The chemistry and mass spectrometry of Sr and Nd isotopes were done at WHOI. Isotopes were measured using a VG-354 5-collector mass spectrometer. Cation exchange separation methods were used for Sr and the LREE, and HDEHP extraction was used for Sm and Nd, as described by Zindler (1980) and Richard et al. (1976). For some samples (OB93-54, OB93-60, OB93-61A, OB93-68, OB93-69, OB93-71, OB93-72, OB93-90, OB93-96), Sr was separated with Sr-spec™ resin (Horwitz et al., 1991; Pin and Bassin, 1992). Isotopic compositions of Sr and Nd were measured on whole rock powders that were leached in cold 2.5N HCl for 15 minutes. Some samples were leached in 6.2N HCl at 100°C for 1.5 hours, but this dissolved a large fraction, if not all, of the plagioclase from plagioclase-rich samples, and most of the Sr was lost. Strontium data are normalized to $^{87}\text{Sr}/^{88}\text{Sr} = 0.1194$, and isotopic compositions are reported relative to $^{87}\text{Sr}/^{86}\text{Sr} = 0.710240$ for NBS987 (average $^{87}\text{Sr}/^{86}\text{Sr} = 0.710255 \pm 14$, $n=7$). Nd data are normalized to $^{146}\text{Nd}/^{144}\text{Nd} = 0.7219$, and isotopic compositions are reported relative to $^{143}\text{Nd}/^{144}\text{Nd} = 0.511847$ for the La Jolla standard (average $^{143}\text{Nd}/^{144}\text{Nd} = 0.511849 \pm 5$, $n=11$). No blank corrections were made on Sr or Nd isotopic ratios. The procedural blank for Sr separation by cation column was 304 pg/g sample, and as much as 12 ng/g sample for Sr separated by Sr-spec resin. The relatively high blank for the Sr-spec method was most likely a result of inadequate cleaning of the resin prior to use. Assuming all of the Sr blank came from the Sr-spec resin, corrections on $^{87}\text{Sr}/^{86}\text{Sr}$ would be only between 2 and 10 ppm for these samples, because 3 to 6.5 μg of analyte was used. Procedural blanks for Nd were 665 pg/g sample and 255 pg/g sample for the two blanks analyzed during this study. Counting

statistics (2σ) associated with mass spectrometry were less than 30 ppm for Sr, except OB93-54* (an unleached sample, Table 4–10), which was 186 ppm. The 2σ counting statistics for Nd were less than 12 ppm, except OB93-60 and OB93-96, which were 56 and 36 ppm respectively. In the figures where these data are shown, the errors are less than the width of the symbols. All concentrations of Sr and Nd, were determined by isotope dilution, during isotopic analysis. Samarium concentrations were determined on the same sample dissolutions and were measured by isotope dilution ICP-MS at WHOI. Errors based on counting statistics varied from 0.23 to 0.47% (2σ), but were 0.70% and 0.77% (2σ) for OB93-54rep and OB93-57 respectively.

Unleached whole rock powders were prepared for Os analysis by NiS fire-assay and distillation, and measured using negative thermal ionization mass spectrometry techniques at WHOI (Hauri and Hart, 1993; Ravizza, 1993). Between 2.5 to 10 g of sample were fused. All Os analyses are corrected for the analytical blank (1.4 pg/g flux with $^{187}\text{Os}/^{188}\text{Os} = 0.2648$). Correction for the Os fusion procedure blank can be significant for low concentration samples, so it is important that the blank $^{187}\text{Os}/^{188}\text{Os}$ and the Os concentration are well known. This is difficult because the blank has such a low concentration. Measurements of blank concentration vary by 5.5% (1 S.D) and the blank isotopic composition, by 8% (1 S.D.). For the samples of this study, that translates as up to 0.6% uncertainty in the accuracy of the isotopic composition due to blank correction, depending on the sample concentration. The other major source of uncertainty is sample reproducibility due to nugget effects. In this study, replicate analyses agree within <0.5%. The precisions associated with the $^{187}\text{Os}/^{188}\text{Os}$ counting statistics range from 0.10 to 0.44% (2σ) for the rocks of this study. Osmium standard runs had $2\sigma = 0.59\%$ ($n=7$). Osmium

concentrations were determined on the same sample splits by fire-assay and isotope dilution, and were within 4.6% and 8.5% of each other for the two replicated samples. Rhenium concentrations were measured on separate powder splits by acid digestion and isotope dilution ICP-MS at WHOI. All Re concentrations are blank-corrected, except OB93-60, OB93-61A, OB93-90, where the blank chemistry for this group of samples failed. Procedural blanks were 0.8 and 10 pg/g sample for the other samples of this study. Uncertainties based on counting statistics were ± 0.13 to 0.38% (2σ). Replicate analyses were within 5 and 20% for the two repeated samples (OB93-71 and OB93-54 respectively).

4.5 Results and Discussion

4.5.1. Major Element Compositions

The bulk compositions of the granulites are broadly basaltic; $\text{SiO}_2 = 45.60$ to 48.99 wt. %, and $\text{Mg\#}$ [molar $\text{Mg}/(\text{Mg}+\text{Fe}) \times 100$] = 55.3 to 82.8 (with Fe as total Fe, Table 4-2). These xenoliths can be divided into two geochemical groups, those that have bulk rock $\text{Mg\#} < 80$ (samples OB93-54, OB93-57, OB93-61A, OB93-68, OB93-69), and those where the $\text{Mg\#} > 80$ (samples OB93-50, OB93-60, OB93-71, OB93-72, OB93-90, OB93-96). Other geochemical characteristics tend to follow this division.

The major element compositions of the granulites are shown on MgO variation diagrams in Figure 4-2. There are negative correlations of Al_2O_3 and Na_2O with MgO , and FeO^* (total Fe) is positively correlated with MgO . No other obvious major element relationships exist. The two outlier samples (OB93-71: highest MgO , and OB93-54: least MgO) are labeled in many figures, as they are extreme in some of their other geochemical characteristics (e.g. [Os] and $^{187}\text{Os}/^{188}\text{Os}$). The granulite xenoliths of this study show a

similar compositional range as those from the Southeastern province of the archipelago (Figure 4-3) (Grègoire, 1994; Grègoire et al., 1998). Thus, the eleven xenoliths that comprise this study are geochemically representative of the lower crustal xenoliths from the Kerguelen Islands.

Table 4-3 through Table 4-7, and Table 4-9 summarize representative compositions of the constituent minerals. Where significant variations are present, more than one analysis is shown. In some cases, all of the data are plotted in the figures. The complete data can be found in Appendix 2.

Clinopyroxenes from the granulites are aluminous diopside, except for one augitic pyroxene in sample OB93-57 (Table 4-3). The Mg# of these minerals range from 68.1 to 85.0. For the lower Mg# samples, cation totals tend to be greater than 4.01 (up to 4.03 in OB93-61A). Stoichiometric corrections using Fe^{3+} bring the cation totals closer to 4.00, indicating the presence of up to 2 wt.% of Fe_2O_3 in clinopyroxenes in the low Mg# samples. The major element compositions of clinopyroxenes are relatively homogenous within a particular sample, though some samples that contain co-existing symplectite clinopyroxenes have compositional differences between the two types (Appendix 2). In many of these granulites (OB93-60, OB93-72, OB93-90, OB93-96), symplectite clinopyroxenes have less Al_2O_3 relative to discrete clinopyroxenes, but clinopyroxene from some (OB93-57 and OB93-69) have higher relative Al_2O_3 contents. Symplectite clinopyroxenes may also be different from discrete clinopyroxenes in SiO_2 , MgO , TiO_2 , Cr_2O_3 , and Na_2O abundances, but not consistently more or less.

Orthopyroxenes are enstatite, with Mg# that range from 62.3 to 81.8 (Table 4-4). As with the clinopyroxene, orthopyroxene from the lower Mg# samples may have significant Fe^{3+} ($\text{Fe}_2\text{O}_3 = 1\text{-}2$ wt%). Discrete orthopyroxenes are generally homogenous within

samples, but symplectite orthopyroxenes tend to have higher Al_2O_3 and lower SiO_2 and MgO contents than co-existing discrete orthopyroxenes.

Anorthite contents of plagioclase are $\text{An}_{77-82.4}$ for rocks from the high Mg# group, and An_{66-72} for the low Mg# samples (Table 4-5). Orthoclase contents are generally less than Or_1 , except for some grains in samples 61A and 68 that are up to $\text{Or}_{1.9}$.

Only Sample OB93-71 contains rare relict olivine; the composition of the one analyzed point is $\text{Fo}_{79.8}$ (Table 4-6).

The oxide minerals in these rocks are Mg-rich spinels, Ti-magnetite, and ilmenite (Table 4-7). In the high Mg# samples, the Mg# of spinels is approximately 70. Of the low Mg# xenoliths, samples OB93-54, OB93-68, and OB93-69 contain ilmenite as well as spinel. The Mg# of spinels from samples OB93-57 and OB93-68 is 58.5 and 52 respectively. One spinel from OB93-68 has a Ti-rich rim with $\text{Mg\#} = 22$. Some spinels from OB93-69 are Cr-bearing; the Mg# of spinels from this rock range from 38 to 60. Spinel in OB93-54 are the most Fe-rich, with $\text{Mg\#} = 27$. Both magnetite and ilmenite co-exist with the spinels in this sample, and both are rimmed with Ti-magnetite with a Mg# of 8 to 10. For all of the granulites, the Cr# of the spinels is generally low (<10) and variable for a given Mg#.

In summary, the most magnesian pyroxenes and spinels, and the most calcic plagioclase are from the high Mg# xenoliths (samples OB93-50, OB93-60, OB93-71, OB93-72, OB93-90 and OB93-96), and the least from the lower Mg# samples (OB93-54, OB93-57, OB93-61A, OB93-68, and OB93-69).

4.5.2. These xenoliths are metamorphic rocks

Evidence for metamorphism

Gabbroic cumulates have been recovered from all ocean basins, but none of these rocks were considered to be metamorphic (e.g. Tiezzi and Scott, 1980; Elthon, 1987; Meyer et al., 1989; Hébert et al., 1991; Dick et al., 1992; Ross and Elthon, 1997). As will be shown below, the xenoliths of the present study possess major and trace element characteristics of gabbroic cumulates, and are determined to be mafic granulites for the following reasons: 1) the equigranular metamorphic texture, 2) the lack of olivine, 3) the relatively high modal abundance of orthopyroxene, and 4) garnet-bearing granulite xenoliths occur on the southeastern province of the archipelago.

Determining the original characteristics of these rocks, including the protolith mineralogy, and ultimately their provenance and the nature of their parent melts, is complicated by the metamorphic recrystallization. An approach taken here to estimate the original igneous mineralogy of these rocks is by their CIPW normative compositions. Unlike volcanic rocks, CIPW norm compositions of gabbroic cumulates are good approximations for true mineralogy. This same approach was used by Rogers and Hawkesworth (1982) and Rudnick et al. (1986) for mafic granulites from continental localities. On the basis of this calculation, the granulites were originally igneous assemblages of olivine, clinopyroxene and plagioclase (Table 4-2). In ternary classification diagrams (Streckheisen, 1973), the normative mineral assemblages of the majority of the granulites fall in the field for olivine gabbros, whereas OB93-68 and OB93-69 lie just inside the orthopyroxene gabbro field (Figure 4-4). Nine of the samples are hypersthene normative, two (OB93-72 and OB93-90) are Si-undersaturated with nepheline as a normative phase.

Modal abundances of minerals in the xenoliths were calculated by mass balance of the whole rock and mineral compositions (Table 4-2). A comparison of the CIPW norms (the postulated protolith mineralogy) and the calculated modes (the current mineralogy) reveals some notable differences. Namely, abundances of orthopyroxene, clinopyroxene and oxides are higher in the current than the postulated protolith mineralogy, whereas those of olivine and plagioclase are lower (Figure 4-4). Particularly conspicuous is a large difference in abundances of orthopyroxene. For instance, OB93-50 contains 4.4 wt.% normative orthopyroxene, whereas its modal abundance is 18.4 wt.% (Table 4-2). Another significant difference is the nearly complete disappearance of modal olivine. Sample OB93-50 contains no modal olivine, in contrast to 15 wt.% of CIPW normative olivine. It should also be noted that clinopyroxene modes are 1.5 to 1.9 times their normative abundances.

The CIPW norms of the granulites are consistent with the mineral modes of oceanic olivine gabbros, but the calculated modes of the granulites are not. In particular, the modal orthopyroxene contents of the granulites are too high for their olivine gabbro bulk composition. Orthopyroxene only occurs in low modal abundances (usually <1.5 %) in oceanic olivine gabbros with Mg# comparable to these xenoliths (Tiezzi and Scott, 1980; Elthon, 1987; Meyer et al., 1989; Hèbert et al., 1991; Dick et al., 1992; Ross and Elthon, 1997; Schminke et al., 1998). For orthopyroxene to be an igneous phase, the protoliths should be more evolved than these granulites. Thus, during metamorphism primary olivine was almost completely consumed, clinopyroxene and orthopyroxene were produced, and excess plagioclase remains a modal constituent. This is indicative of the reaction $\text{olivine} + \text{plagioclase} \rightarrow \text{orthopyroxene} + \text{clinopyroxene} + \text{spinel}$, supporting the idea that was previously proposed by McBirney and Aoki (1973) to explain the formation of similar Kerguelen xenoliths that they termed spinel gabbros. This reaction can be driven in two

ways, either by an increase in pressure or a temperature decrease, both of which will be discussed as mechanisms of metamorphism in more detail below.

Further evidence for a metamorphic origin lies in the compositions of the rock forming minerals. For a given Mg#, pyroxenes from these xenoliths are more aluminous than pyroxenes from other oceanic cumulates, and clinopyroxenes are more sodic (Figure 4-5, Figure 4-6). These features are typical of pyroxenes that are metamorphic products of the olivine + plagioclase reaction (Kushiro and Yoder, 1966). Additionally, oceanic olivine gabbros usually have chromite as their oxide phase, but these rocks contain Cr-poor spinels (Figure 4-7).

In summary, the Kerguelen gabbroic xenoliths contain essentially no olivine, too much modal orthopyroxene to be of primary igneous origin, orthopyroxenes and clinopyroxenes that are too rich in Al_2O_3 , clinopyroxenes that are too sodic, and Cr-poor spinels rather than chromite. Therefore, the conclusion is that these rocks must have been produced by metamorphism. The metamorphism was probably isochemical. Only one small amphibole was observed in sample OB93-54, otherwise no hydrous minerals or other evidence for geochemical changes are present in these rocks. The likelihood that metamorphism was isochemical will be discussed in more detail below in the context of trace elements.

Finally, that mafic granulites occur in the southeastern province of the Kerguelen Islands (Grègoire, 1994; Grègoire et al., 1998) is good indication that mafic granulites might also exist in some of the other xenolith localities. Some of the granulites from the Southeastern Province have both garnet and sapphirine, sometimes co-existing, indicating equilibration pressures and temperatures of at least 900 °C and 1.0 GPa (Figure 4-8) (Christy, 1989; Grègoire et al., 1994).

Conditions of Metamorphism

It is not possible to determine the exact pressure of metamorphism for this suite of xenoliths because they do not contain garnet, although some constraints are possible. Given that garnet or sapphirine are not present in these granulites, the metamorphic pressures must have been between 0.6 and 1.0 GPa at lower temperatures ($\sim 800^{\circ}\text{C}$) and 0.8-1.2 GPa at near solidus temperatures (Figure 4-8) (Kushiro and Yoder, 1966; Ringwood, 1975; Christy, 1989). These temperature and pressure ranges are defined by the reaction boundaries of the appearance of orthopyroxene and garnet $\pm$ sapphirine as determined on the CMAS experimental system (e.g. Christy, 1989). The range of metamorphic conditions can be further refined with the temperatures of equilibration for these granulites as estimated with a two-pyroxene thermometer (Wells, 1977). These temperatures vary from 844 to 961 $^{\circ}\text{C}$ (Table 4-8) and are consistent with the equilibration of these rocks in the granulite facies. The field of possible metamorphic pressure and temperature conditions for the rocks of this study are shown in Figure 4-8. This field is at significantly lower pressures and higher temperatures than the normal thermal structure of 40 Ma oceanic crust. This characteristic will be discussed in more detail below in relation to possible mechanisms of metamorphism.

There are no significant differences between temperatures calculated with co-existing discrete pyroxene pairs and symplectite pairs. Major element compositional differences do exist between discrete and symplectite pyroxenes, and are most pronounced for orthopyroxene pairs. If all orthopyroxenes in these rocks are products of a single metamorphic reaction, they should all have the same composition, regardless of their texture. The observed compositional differences in pyroxenes as a function of texture, suggest there may have been more than one stage of orthopyroxene formation.

Symplectite textures can be a result of several different processes: cooling from igneous temperatures (Griffin and O'Reilly, 1987), reheating of the xenolith by the host magma on the way to the surface (McBirney and Aoki, 1973), garnet breakdown (Kushiro and Yoder, 1966), or a kinetic effect. No differences were seen in trace elements between symplectite and discrete clinopyroxenes, arguing against a kinetic origin. Moreover, the symplectite clinopyroxene trace element signatures rule out a garnet precursor. The origin of these symplectites is puzzling and will require further investigation beyond this study.

4.5.3. Comparison to other oceanic cumulates

Because the granulite xenoliths were originally oceanic olivine gabbros that underwent metamorphism, it is important to compare them to other oceanic gabbroic cumulates, especially for the purpose of determining provenance. Samples of oceanic gabbros come from several lower crustal drill sites: DSDP site 334, just west of the FAMOUS locality on the Mid-Atlantic Ridge (Aumento et al., 1977), ODP Leg 118, SWIR (Von Herzen et al., 1991; Hébert et al., 1991; Dick et al., 1992), ODP Leg 147, Hess Deep (Mével et al., 1996; Natland and Dick, 1996), ODP Leg 153, MARK (Karson et al., 1997; Ross and Elthon, 1997), dredged from fracture zones: MAR, 26°N (Tiezzi and Scott, 1980), the Mid-Cayman Rise (Elthon, 1987), SWIR and CIR (Meyer et al., 1989), and xenoliths that are included in MORB (Dixon and Clague, 1986; Davis and Clague, 1990) and ocean island basalts (Clague and Chen, 1986; Clague and Bohrsen, 1991; Grègoire et al., 1994; Grègoire et al., 1998; Mattielli et al., 1996; Hassler and Shimizu, 1998; Hoernle, 1998; Lassiter and Hauri, 1998; Schminke et al., 1998). Cumulate xenoliths are only rarely found in MORB, are small in size and often have a significant amount of interstitial melt (e.g. Dixon and Clague, 1986; Davis and Clague, 1990). They, and rocks dredged from fracture zones, are often quite altered, both by seafloor weathering and hydrothermal alteration. Thus, very little whole rock data exist for these rocks, and analyses of mineral

compositions comprise the majority of the data. More whole rock data exist for the drill cores, though these rocks too can be altered. Xenoliths included in ocean island basalts (the Hawaiian Islands, the Canary Islands, and the Kerguelen Islands) are by far the least altered samples, but very few geochemical data are available.

Comparisons with other oceanic cumulate rocks are further complicated because the various rocks can represent different parts of complex magmatic systems. A spectrum of cumulate types exist, from pure adcumulates to frozen liquids, and determining which particular type a rock may be, is not always straightforward. The amounts of trapped melt are variable among cumulates, and diametrically opposed interpretations that were made even on the same rock samples (e.g. Hess Deep, Natland and Dick, 1996; Pederson et al., 1996) illustrate the difficulties inherent in these interpretations. Often, as in hole 735B, the transitions through the different cumulate types can occur over length scales of less than one meter (Dick et al., 1992).

Overall, the major element compositions of the granulites of this study are similar to other oceanic gabbros, as well as mafic granulite xenoliths from continents (Figure 4–9). A comparison of these rocks to olivine gabbros, gabbros, and troctolites with similar Mg# from the MARK area, the SWIR, and the Canary Islands show no significant differences. The compositional fields for continental mafic granulites completely encompass all of the data from oceanic cumulates and the granulites of this study. The oceanic cumulates shown in Figure 4–9 were chosen for comparison not only because of geochemical similarities of the rocks to the granulites, but also because they are considered to have little to no trapped melt (<7%, by definition of classic adcumulate).

The major element variations of cumulates are determined by the nature of the parent melt, fractional crystallization and the relative amounts of cumulate minerals and trapped

melt. As noted above, the granulites in this study exhibit no obvious trends in major element space. They do not follow a MORB fractionation trend (liquid line of descent) and are therefore not related to each other by simple crystal fractionation (Figure 4–9). On the basis of the major element systematics of these rocks, and that they are most similar to other oceanic gabbros that are interpreted as nearly pure adcumulates, leads to the initial assertion that the granulites are recrystallized adcumulates that had negligible amounts of trapped melt. However, the major element compositions cannot discriminate among a ridge, a plume-ridge interaction, or a plume-only origin for these rocks. For this, the trace element and isotopic compositions of these rocks are used, and are described in the following sections.

4.5.4. Trace Element compositions

The granulite xenoliths have whole rock REE contents that are between 1 and 12 times C1 chondrite normalized values, except OB93-54, which has C1 normalized REE abundances that range from 15 to 25 (Figure 4–10). All of the granulites, but OB93-61A and OB93-54, have positive whole rock Sr and Eu anomalies. The REE abundances of these rocks also tend to correlate with Mg#; the lower Mg# samples have the highest REE contents, and vice versa. Of the lower Mg# samples, OB93-54, OB93-61A, and OB93-68 are the most REE enriched, whereas samples OB93-57 and OB93-69 are transitional between these and the high Mg# samples. Sample OB93-68 is LREE enriched with a normalized La/Sm ratio ($\text{La/Sm}_N = 1.1$), OB93-61A has a flat REE pattern ($\text{La/Sm}_N \sim 1$), and the remaining samples of this group are LREE depleted ($\text{La/Sm}_N < 0.75$). The high Mg# group of rocks also tend to be LREE depleted ($\text{La/Sm}_N < 0.8$). Sample OB93-60 is an exception with $\text{La/Sm}_N = 1.1$. The normalized whole rock HREE patterns of all the granulites are parallel to each other at values between 2 and 6.

Trace element data for clinopyroxenes, determined by ion microprobe, are shown in Table 4-9 and Figure 4-11. Again, representative analyses are shown in the table, whereas all of the data are plotted in Figure 4-11. The entire data set is tabulated in Appendix 2.

The highest REE abundances occur in clinopyroxenes from the lower Mg# granulite sample group and vice versa. Trace element contents of clinopyroxenes generally display minor grain-to-grain variations within a particular sample and the symplectite clinopyroxenes possess very similar abundance patterns relative to co-existing discrete grains. A symplectite clinopyroxene in OB93-72 is an exception in that it has a significant positive Eu anomaly, suggesting that plagioclase was involved in a reaction that produced this symplectite intergrowth of clinopyroxene and spinel. A comparison of REE patterns between clinopyroxene and whole rocks, together with the clinopyroxene modes tabulated in Table 4-2, demonstrates that the bulk of the REE reside in clinopyroxene and calculations show that >80% of most of the REE are in fact hosted by clinopyroxene. The LREE are not always consistent with this generalization; ion probe analyses made on plagioclase using clinopyroxene working curves indicate that the LREE are strongly partitioned into plagioclase.

Due to the overall mass balance, clinopyroxene REE patterns mimic those of whole rocks. In some samples (such as OB93-60), the LREE depletion in clinopyroxene relative to the whole rocks is accentuated because of the plagioclase effects mentioned above. The REE abundance patterns in OB93-54, the most Fe-rich sample, require special consideration because of their unique shape. While the whole rock pattern of this sample is slightly LREE depleted ($[La/Yb]_n = 0.99$; $[La/Sm]_n = 0.59$; Table 4-9), the clinopyroxene patterns are characterized by significant depletion in the HREE. For instance, the whole rock Yb concentration (2.60 ppm, Table 4-2) is 3.7 times that in clinopyroxene (0.71 ppm, Table 4-9), completely opposite of the general observation made above. That is, less than

30% of the HREE are present in the clinopyroxene. It follows that mass balance requires the existence of a phase that could account for the HREE. One obvious candidate is garnet. Although garnet is not observed in this particular rock, many areas of fine-grained, partly opaque alteration or oxidation products are present. However, Grègoire (1994) describes granulites in which garnet is surrounded by dark-colored, opaque alteration coronas. They closely resemble the texture in which the granular fine-grained alteration areas occur in OB93-54.

Even though the original igneous textures of the granulites were obliterated by metamorphism, there are several lines of evidence that these rocks are cumulates. Whole rock Cr and Ni abundances have positive correlations with MgO that are usually associated with accumulations of olivine and clinopyroxene (Figure 4-12). Correlations of other compatible elements, such as Sc and V, with MgO are less clear (Figure 4-12). The abundances of Ni, Cr and Sc are higher, and V lower for a given MgO content, than those in Kerguelen Islands basalts (e.g. Yang et al., 1998).

It is notable that the bulk rock trace element patterns lie within the field defined by gabbros from the Oman ophiolite, and have positive Sr and Eu anomalies of similar magnitudes as the Oman gabbros (Figure 4-10). The exception to this is sample OB93-54. In a plot of Sr/Sr^* v. Eu/Eu^* all of the xenoliths but OB93-54 are greater than one in both dimensions (Figure 4-13). These characteristics are indicative of plagioclase accumulation. The samples that are outliers in some of the panels of Figure 4-2 (OB93-54 and OB93-71) are also noted in Figure 4-10, Figure 4-12, and Figure 4-13. For example, sample OB93-71, the most MgO-rich sample, is extreme in all of the panels of Figure 4-12, clearly indicating its cumulate origin. In contrast, sample OB93-54 does not lie in the trace element field of Oman gabbros, is an outlier in its V content, and has no Eu anomaly. Thus, the possibility that it is a cumulate is not as certain. However, it does have a small positive Sr

anomaly (Figure 4-13), and is not significantly different than the other xenoliths in its Cr, Ni, and Sc contents. Moreover, the Ba/Th of this rock is high, like that of the other xenoliths and the Oman gabbros (Figure 4-10), and has low abundances of other incompatible elements (e.g. Rb and K₂O, Table 4-2).

In a gabbro-basalt system, the overall concentration levels of the REE in both the liquid and the cumulate would continue to increase as crystallization proceeds, on the basis of partitioning arguments. At the same time, as plagioclase continues to crystallize, the liquid would develop greater negative Sr and Eu anomalies and the positive anomalies of these elements in the cumulates would decrease. Hence, it would not be surprising if a late stage cumulate displayed no or even small negative Sr and Eu anomalies, and OB93-54 could be an example of such a process. An important point to note in this context is the low Th content of this sample (0.03 ppm, Table 4-2), which argues strongly against a contribution from melt in the origin of this rock, either as an interstitial melt or as a frozen liquid. Therefore, the major and trace element data indicate that the protoliths of these xenoliths were formed as olivine-clinopyroxene-plagioclase cumulates from basaltic melts.

The whole of the data, including trace elements, seem to indicate that metamorphism has not significantly affected the geochemistry of these rocks. As noted above, the whole rock major element compositions are no different in any obvious ways from other oceanic cumulates. The similarity in trace element patterns between the granulites and the Oman gabbros are striking, and the REE part of these patterns are smooth (Figure 4-10). These characteristics argue against any significant elemental gain or loss during metamorphism. Therefore, it seems that metamorphism was isochemical, though there are no protoliths for which to compare these rocks for an unequivocal interpretation.

4.5.5. Did the granulite protoliths contain an intercumulus melt?

If one accepts the assertion made above that the present day chemical compositions of the granulites represent those of their igneous precursors, the next step toward determining their provenance is to estimate the compositions of the melts from which they had accumulated. Because of the metamorphism however, this task is not straightforward.

For example, the trace elements in any melts that may have been trapped when the protoliths accumulated as olivine gabbros would be incorporated into the clinopyroxenes during recrystallization, and the compositions of equilibrium melts estimated from these clinopyroxenes would be in error. A similar situation would be true for equilibrium melts estimated from the whole rock trace element abundances. Therefore, it is necessary to attempt a quantitative assessment of the original assertion that the protoliths of these granulites contain a low trapped melt fraction before attempting to determine their provenance on the basis of trace element abundances.

There are at least two possible sources of inter-cumulus melt in a cumulate rock. In one, the melt is related to the parent melt; that is, the melt would have been trapped as the cumulus minerals precipitated. The second possibility is that a pre-existing rock is impregnated by a melt that is percolating through it. The percolating melt may be genetically related to the rock through which it is traveling, or not, as in the case of a plume related melt percolating through oceanic lithosphere.

Classically, abundances of trapped melt in cumulates were estimated by the amount of minerals in intercumulus texture and zoned overgrowths of cumulus minerals (Wager and Brown, 1967). However, because of the metamorphic recrystallization, this method is not possible in the present case. Others have estimated trapped melt abundances on the basis of

the amount of P_2O_5 and Zr in the bulk rock (Wager and Brown, 1967; Natland et al., 1991). This method assumes that these two elements are perfectly incompatible, and that any trapped melt would contribute the entire bulk rock budget of these elements. In reality however, some minerals, particularly clinopyroxene could contain some small amount of P_2O_5 and more significant quantities of Zr. Even so, the concentrations of both P_2O_5 and Zr are quite low in these rocks (Table 4-2), and by this reasoning the abundances of trapped melt should be relatively low.

A quantitative assessment of possible interstitial melt would require a prior knowledge of abundances of the trace elements in the melt. For unmetamorphosed cumulates the mass balance method described by Bédard (1994) can be applicable. This approach apportions the whole rock REE abundances among the cumulus minerals, but also includes a melt phase with a mineral/melt partition coefficient of 1. He successfully used this model to calculate the trace element abundances of clinopyroxenes and feldspars of gabbro-norites from the Bay of Islands ophiolite. Given this, using the bulk rock and the CIPW norms, it should be possible to calculate the REE abundances of granulite clinopyroxenes, and compare these to what is actually measured in situ to get an indication of trapped melt effects. However, because the cumulates of the present study are metamorphosed, this approach is not applicable. In this case, the modal abundances of clinopyroxene increased by a factor of at least 1.5 during the metamorphic reaction, thereby diluting the original clinopyroxene REE abundances to their present concentrations.

Applying the incompatible element method mentioned above, the whole rock abundances of very incompatible elements, such as Th, can provide the most sensitive method to assess the amount of trapped melt in these rocks. If one assumes that all of the Th in the granulites came from a trapped Kerguelen basalt, the percentage of melt trapped in the granulites can be calculated. Thorium contents in Kerguelen basalts from the

Raboulliere and Bureau sections vary from 0.19 to 7.4 ppm with averages of 3.76 ($n=20$, 1 S.D. = 1.58) and 2.74 ($n=29$, 1 S.D. = 1.33) respectively (Yang et al., 1998). The Kerguelen basalt with the least Th content (0.19 ppm) gives unrealistic trapped melt abundances (up to 108%), but the average Th content of the Raboulliere basalt gives melt contents that are less than 1% for six of the samples (OB93-54, OB93-57, OB93-69, OB93-71, OB93-72, OB93-96), less than 4% for four (OB93-50, OB93-60, OB93-68, OB93-90), and one with 6.3% (OB93-61A). These results argue against significant amounts of trapped melt.

If, as noted above, a plume melt impregnated a MORB cumulate, a mass balance calculation would not be valid. To determine if this type of process might have been involved in the petrogenesis of these rocks, calculations were made for the $^{87}\text{Sr}/^{86}\text{Sr}$ and Sr contents in two-component mixing of a gabbro in equilibrium with an SEIR MORB and Kerguelen Islands basalts. Figure 4-14 illustrates how bulk $^{87}\text{Sr}/^{86}\text{Sr}$ and Sr contents change in a gabbro as impregnation proceeds. The gabbro starting material was assumed to have formed from a SEIR melt with $^{87}\text{Sr}/^{86}\text{Sr} = 0.70288$ and $[\text{Sr}] = 145$ ppm (Dosso et al., 1988). The abundance of Sr in the gabbro was calculated from the SEIR melt using bulk partitioning based on 13.5% olivine, 31.5% clinopyroxene, and 55% plagioclase, and mineral partition coefficients from Dunn and Sen (1994) and Hart and Dunn (1993). Simple mixing of two different Kerguelen basaltic melts (Yang et al., 1998) with this gabbro increases both $^{87}\text{Sr}/^{86}\text{Sr}$ and Sr content. It is evident that the mixing curves do not match the observations: the granulites are more radiogenic and have less Sr than any of the mixtures. Even if the original starting gabbro was displaced to match the lowest Sr content observed in the granulites, the mixing curves are still away from the granulite data points. Thus, these calculations also argue against impregnation of Kerguelen basalt into SEIR

gabbros in the formation of the granulite protoliths. The overall conclusion is that the amount of intercumulus melt present in the protoliths of these rocks 6% or less.

4.5.6. Granulite Protolith Parent Melts

Given that the granulite protoliths were gabbroic cumulates with only small quantities of trapped melt, the geochemical characteristics of their parental melts can be estimated on the basis of their trace element abundances.

To estimate parent melt geochemistry from bulk rock data, the bulk partition coefficient for each of the REE was calculated from the abundances of the CIPW normative minerals (representing the protolith minerals and their proportions) and partition coefficients from the literature (Schnetzer and Philpotts, 1970; Hart and Dunn, 1993; Dunn and Sen, 1994; Kelemen et al., 1993). Figure 4-15 demonstrates that the normalized REE patterns of the equilibrium melts that result from this calculation lie in the field defined by basalts from the Kerguelen Islands rather than that defined by "non-plume" SEIR MORB. The slopes of the melt REE patterns are within the range of those for the Kerguelen Islands basalts (e.g. $La/Yb_N = 2-13$ for basalts and 3-10 for granulite equilibrium melts). In detail however, the melts calculated from the whole rock data have slight negative Eu anomalies that are not present in most of the basalts. This is most likely an artifact of the D value chosen for Eu, which is dependent on the composition and the oxidation state of the magmatic system (e.g. Schnetzler and Philpotts, 1970; Dunn and Sen, 1994). The values used for this calculation may not be quite appropriate. Observed plagioclase/clinopyroxene REE partitioning from Oman ophiolite gabbros (Kelemen et al., 1997) are within 1 standard deviation of the plagioclase/clinopyroxene partition coefficients used here (Schnetzer and Philpotts, 1970), except for Dy and Yb. For these two elements the experimental data are both higher than the that observed at Oman, illustrating the variation of partition coefficients. Other uncertainties in the equilibrium melt calculation may be

introduced from how accurately the CIPW normative minerals recreate the original mineralogy.

Trace element abundances in clinopyroxene can also be used for estimating the geochemistry of the parent melt. However, caution must be exercised because recrystallization has changed the abundances of clinopyroxene in these rocks. As discussed above, nearly all of the REE reside in the clinopyroxene after metamorphism and the modal abundance of clinopyroxene has increased during recrystallization, thereby diluting the pre-metamorphic REE abundances in this mineral. It follows that equilibrium melts calculated from recrystallized clinopyroxene are minimum estimates of the REE abundances of the equilibrium melt. In addition, La abundances are further underestimated due to its partitioning into plagioclase.

Figure 4-16 illustrates the results of the REE abundances calculated from clinopyroxene. The overall shape and slope of the REE patterns of the equilibrium melts calculated from clinopyroxene are also similar to the Kerguelen basalts but have a larger variation than the equilibrium melts calculated from the whole rock. The clinopyroxene equilibrium melts have $La/Yb_N = 2-10$, and the patterns extend to the more enriched of the Kerguelen Islands basalts. Though some of these equilibrium patterns overlap the SEIR basalts field, they have greater slopes than the SEIR basalts: $[La/Yb]_N = 2-10$ for equilibrium melts v. $[La/Yb]_N = 0.4-1.3$ for the SEIR basalts (Michard et al., 1986). In summary, equilibrium melts calculated from both whole rocks and clinopyroxenes indicate that the granulite parent melts have stronger affinities to the Kerguelen basalts than the SEIR MORB. This conclusion is consistent with the isotopic compositions of the granulites, as will be discussed below.

4.5.7. Sr, Nd, and Os isotopic compositions

Rubidium-Sr, Sm-Nd, and Re-Os isotopic compositions of the granulite xenoliths are listed in Table 4-10 and Table 4-11. The $^{87}\text{Sr}/^{86}\text{Sr}$ for bulk rocks ranges from 0.704453 to 0.706048, and $^{143}\text{Nd}/^{144}\text{Nd}$ varies from 0.512731 to 0.512838 ($\epsilon_{\text{Nd}} = 1.81$ to 3.90).

Samples OB93-57 and OB93-61A were not analyzed for Nd. The isotopic composition of the host dike is $^{87}\text{Sr}/^{86}\text{Sr} = 0.706380$ and $^{143}\text{Nd}/^{144}\text{Nd} = 0.512419$ ($\epsilon_{\text{Nd}} = -4.27$). The blank corrected $^{187}\text{Os}/^{188}\text{Os}$ of the granulites ranges from 0.1286 to 0.1792.

The Sr and Nd isotopic compositions of all the granulites (both Courbet Peninsula and Southeast Province) are within the range observed for Kerguelen basalts (Figure 4-17). The Sr and Nd isotopic compositions for six granulite xenoliths from the Mt. Tizard location (Southeast Province) reported by Mattielli et al. (1996) plot on the isotopically “depleted” end of the spectrum, ranging in $^{87}\text{Sr}/^{86}\text{Sr} = 0.70398$ to 0.70467 and in $^{143}\text{Nd}/^{144}\text{Nd}$ from 0.51277 to 0.51288. Two granulites from the Dome Rouge and one from the Val Phonolite locations (Mattielli, 1996) plot at the most enriched part of the spectrum ($^{87}\text{Sr}/^{86}\text{Sr} = 0.70507$ to 0.70523, and $^{143}\text{Nd}/^{144}\text{Nd} = 0.51254$ to 0.51258). The results of this study fill the gap of Sr and Nd isotopic compositions and extend the isotopic range of the granulites toward a more isotopically enriched direction, with $^{87}\text{Sr}/^{86}\text{Sr}$ up to 0.70601 ($^{143}\text{Nd}/^{144}\text{Nd}$ for this sample, OB93-61A, was not determined). The Sr and Nd isotopic compositions of the granulite xenoliths of this study, combined with those of Mattielli et al. (1996) and Mattielli (1996), cover practically the entire range of Sr and Nd isotopes observed for Kerguelen Islands basalts. Furthermore, the isotopic distinction between peridotite and granulite xenoliths discussed by Mattielli et al. (1996) has virtually disappeared. Before further discussion on the origin of these isotopic signatures, it is important to assess the magnitude of alteration effects, age correction, and secondary processes, to establish that the observed isotopic characteristics are primary features.

For Sr and Nd analyses, the majority of the samples were leached in a weak HCl solution. Some were leached in stronger hot HCl, and three were not leached. Some of the rocks (OB93-54, OB93-68, OB93-71, OB93-72) were analyzed more than once and different aliquots of these were treated with the different leaching procedures. The Sr isotopic compositions of the samples that were most strongly leached are less radiogenic than those that were only weakly leached or weren't leached at all. The results indicate the presence of a leachable radiogenic component, but its quantity appears to be small. For instance, the $^{87}\text{Sr}/^{86}\text{Sr}$ for OB93-71 was reduced from 0.704617 in the unleached powder to 0.704453 in the strongly leached one, a total reduction of 1.6×10^{-4} . The Nd isotopic compositions of leached and unleached samples are identical. Rhenium and Os analyses were made on unleached whole rock powders, but given that the abundances of these elements in the granulites are relatively high, it seems unlikely that alteration much affected the Re-Os systematics. However, there is some evidence from ODP site 735B that a complex history of fluid and melt interaction had different effects on the different isotopic and trace element signatures of these rocks, including Os (Hart et al., 1999). A similar situation does not seem likely in this case, mainly because the co-variation of $^{87}\text{Sr}/^{86}\text{Sr}$ with $^{143}\text{Nd}/^{144}\text{Nd}$ is not an alteration signal (Figure 4-17). Furthermore, neither Sr nor Nd correlates with any of the trace element indices of alteration (e.g. Cs, Rb, Ce).

The host basalt dike has high concentrations of isotopically enriched Sr and Nd and could potentially alter the isotopic signature of the granulites through interaction. Although the Sr and Nd isotopic signatures of the host dike could be affected by small disseminated metasomatic peridotites, this is highly unlikely. Mixing calculations between the most isotopically enriched basalt from the Islands ($^{87}\text{Sr}/^{86}\text{Sr} = 0.7055$ and 339 ppm Sr; Yang et al., 1998), and peridotite clinopyroxene like that of sample OB93-83 (Chapter 2; $^{87}\text{Sr}/^{86}\text{Sr} = 0.706427$ and 558 ppm Sr) indicate that for this to happen, the basalt must consist of

80% clinopyroxene to explain the Sr isotopic composition of the dike. No thin section is available for this sample, but there is no evidence for this much clinopyroxene in the hand sample. Thus, the measured isotopic composition of the host dike is real.

To change the granulite sample with the least radiogenic Sr to that with the most radiogenic Sr by addition of the host basalt, the most radiogenic granulite must have had approximately 10% basalt physically added. There is no evidence that this much basalt is incorporated into any of these xenoliths. Along with the lack of petrographic signs of alteration or interaction with the host basalt, these data indicate that the geochemistry of the granulite xenoliths was not affected by alteration or the host basalt at the level of the interpretations, and that the isotopic compositions of these xenoliths are therefore considered primary.

Although the age of the xenolith host dike is unknown, it must be younger than approximately 24.5 Ma, the age of basalts in the area (Nicolaysen et al., 1999). The absolute ages of the granulites are also not known, but as will be discussed below, they could be relatively young (<70 Ma). Though the absolute initial isotopic ratios of the granulites cannot be determined, corrections for isotopic ingrowth since the time of eruption of their host basalt can be estimated to compare their isotopic compositions to those of the Kerguelen Islands basalts. Corrections for 24.5 Ma on the granulites, the maximum age of the host dike, are only 7-19 ppm for $^{87}\text{Sr}/^{86}\text{Sr}$, and 0.5-0.9 epsilon units for $^{143}\text{Nd}/^{144}\text{Nd}$. Thus, no correction is applied for Sr and Nd. Corrections for 24.5 Ma of Os isotopic ingrowth are less than 1.4% for all samples (most are less than 1%) except OB93-54, for which the correction was 9.8% (Table 4-11).

Figure 4-18 shows that the $^{187}\text{Os}/^{188}\text{Os}$ of the granulites are independent of their Os concentrations, and no obvious mixing relationships are present. The most radiogenic

granulite (OB93-54) is among the lowest Os content samples, but its Os concentration (100 ppt; Table 4-11) is well above the 20 ppt cut-off for potentially significant contamination effects that have been observed in many basalts (e.g. Shirey and Walker, 1998). Indeed, several of the Kerguelen plume related basalts that have been analyzed exhibit elevated $^{187}\text{Os}/^{188}\text{Os}$ at low concentrations (Figure 4-18) (Reisberg et al., 1993; Yang et al., 1998). After eruption age correction, the $^{187}\text{Os}/^{188}\text{Os}$ of these basalts are quite similar to the 24.5 Ma corrected $^{187}\text{Os}/^{188}\text{Os}$ of OB93-54 (see Figure 4-21). The two analyzed basalts from Heard Island also have relatively low Os contents and have $^{187}\text{Os}/^{188}\text{Os}$ similar to the remaining granulites (Barling et al., 1997). On the basis that Os concentrations in the granulites are beyond the "susceptibility" limit of 20 ppt and that the $^{187}\text{Os}/^{188}\text{Os}$ can be considered primary, the Os isotope similarity between granulites and basalts provides credibility to the basalt data. It is noteworthy that the Os isotope heterogeneity observed in the granulites appears to be reproduced in the difference between the Kerguelen and Heard Islands basalts, although the numbers of analyzed basalt samples are currently limited.

No isochron relationship exists among these xenoliths, though the most radiogenic sample OB93-54 does have the highest $^{187}\text{Re}/^{188}\text{Os}$ (Figure 4-18). As shown in the figure, this sample would have roughly the same $^{187}\text{Os}/^{188}\text{Os}$ as the other granulites at 69 Ma. A best fit line gives an initial $^{187}\text{Os}/^{188}\text{Os} = 0.1290$.

Comparison to isotopic compositions of other lower crustal rocks

In comparison to other oceanic cumulates, the isotopic composition of these rocks are unique. The Sr and Nd isotopic compositions of unaltered gabbros from the SWIR, MARK, Hawaii and the Canary Islands are all isotopically less enriched than any of the Kerguelen granulites and basalts. Gabbros from the SWIR are isotopically similar to Indian Ocean MORB (Kempton et al., 1991; Hart et al., 1999), and the MARK gabbros are

isotopically like the North Atlantic MORB (Kempton and Hunter, 1997). The gabbro xenoliths from the Canary Islands have primary Sr and Nd isotopic signatures of ca. 170 Ma Atlantic N-MORB (Hoernle, 1998), and Hawaiian gabbro xenoliths have Sr and Nd isotopic compositions of modern East Pacific Rise MORB (Lassiter and Hauri, 1998). In short, no other oceanic cumulates, not even the mafic xenoliths included in plume-related basalts, come near to exhibiting the isotopic enrichment seen in the Kerguelen granulites.

Mafic granulites from continental settings have quite variable geochemistry. In general, those from granulite terrains are richer in SiO_2 (>55 wt.%) than granulite xenoliths erupted through continental crust ($\text{SiO}_2 < 55$ wt.%; Rudnick and Presper, 1990), thus the comparison presented here is restricted to mafic granulite xenoliths from continental localities. As noted above, the major element compositions of the continental granulites vary widely and encompass the entire ranges shown by the granulites of this study, and other oceanic cumulates (Figure 4–9). Trace element and Sr and Nd isotopic compositions of continental granulite xenoliths are similarly diverse (see reviews by Rudnick and Presper, 1990; Rudnick, 1992; Rudnick and Fountain, 1995 and references therein) and a simple conclusion cannot be drawn for a continental v. oceanic origin for the xenoliths of this study. Figure 4–19 shows that the Sr and Nd isotopic compositions of the mafic continental granulites are highly variable and completely enclose the isotopic compositions displayed by the Kerguelen granulites.

In contrast, Re-Os systematics within the continental and oceanic lower crust are quite different and inferences can be made from these data on the provenance of the Kerguelen granulite xenoliths. Figure 4–20 compares the Os isotopic compositions of the granulites of this study with those of both oceanic and continental lower crust. It is evident that the Kerguelen granulites are less radiogenic than gabbros from site 735B and the Hawaiian Islands (Hart et al., 1999; Lassiter and Hauri, 1998), and lower crustal samples from

Udachnaya (Pearson et al., 1995), Australia (Saal et al., 1998), and the western USA (Esperança et al., 1997). All of the Australian samples and one of the western USA samples are granulites; all the remaining samples from the western USA and Udachnaya are eclogites. The Os concentrations in the Kerguelen granulites are an order of magnitude greater than the gabbros from site 735B and the Hawaiian Islands (<25 ppt, Lassiter and Hauri, 1998; Hart et al., 1999), and have Os concentrations that are more similar to those of the high Os concentration end of the spectrum for the continental lower crust (3-1043 ppt; Pearson et al., 1995; Esperança et al., 1997; Saal et al., 1998). Figure 4-20 also shows that the $^{187}\text{Re}/^{188}\text{Os}$ of the granulites of this study have a much more limited range and overall lower value than all of the other samples. Rhenium concentrations for most of the Kerguelen granulites are substantially less than the 2 ppb average for 735B samples (Hart et al., 1999), though a few (OB93-54, OB93-68, and OB93-71) overlap the ranges of the other samples.

Osmium model ages of the continental samples are obtained by tracing back isotopic growth on the basis of the present-day Re/Os to the time where an intersection with mantle growth curve occurs. These model ages are >250 Ma for Australia (Saal et al., 1998), >1 Ga for Siberia (Pearson et al., 1995), and >160 Ma for the western USA (Esperança et al., 1997). The Os model ages of the Kerguelen granulites range from -8 Ga to 14 Ga (Table 4-11) (calculated with parameters from Lindner et al., 1986; and Walker and Morgan, 1989), and it is clear that these numbers must be carefully examined. The $^{187}\text{Os}/^{188}\text{Os}$ of many of the Kerguelen granulites are near, or slightly more radiogenic than the present-day value for primitive mantle (0.12757, based on carbonaceous chondrites; Walker and Morgan, 1989), and are much less radiogenic than the majority of the continental samples. In this case, small variations in the Re/Os of a sample will magnify the error in model age and negative ages result from sub-chondritic Re/Os. The cumulate character of the

Kerguelen granulite protoliths contributes to this effect. Note in Table 4-11 that sub- or near chondritic Re/Os occur in many samples with high Mg# and low incompatible element abundances (e.g. OB93-71). Given that Re and Os concentrations were reproducible at 4.6 to 20% on the basis of replicate analyses (mainly due to nugget effects; see Analytical Methods section), the Os model ages for samples with $^{187}\text{Re}/^{188}\text{Os}$ close to (or below) chondritic values would not have significance. For those with $^{187}\text{Re}/^{188}\text{Os}$ greater than 1.0 however (OB93-50, OB93-54, OB93-60, OB93-61A, OB93-68 and OB93-90), the closeness of the measured $^{187}\text{Os}/^{188}\text{Os}$ to the present-day primitive mantle indicates their relatively young ages. The $^{187}\text{Os}/^{188}\text{Os}$ of the granulites are also consistent with the idea that these rocks are related to the Kerguelen plume rather than having an origin in the continental crust. This is because the Os isotopic composition of mantle plumes were postulated to have a primitive mantle composition (e.g. Martin, 1991) or even a radiogenic core contribution. Although not completely unequivocal, the Re-Os characteristics of the granulites argue for their provenance connected to the Kerguelen plume rather than the lower continental crust of Gondwanaland.

Origin of the isotopic signatures

As discussed above, the Sr and Nd isotope variations in the granulites cover practically the entire range observed for the Kerguelen Islands basalts. This, together with the trace element arguments presented above for a genetic relation of the granulites to the Kerguelen Islands basalts, indicates that the isotopic signatures of the two should have similar origins. Previous models for basalt petrogenesis invoked variations on the theme of two-component mixing (Storey et al., 1988; Gautier et al., 1990; Weis et al., 1993; Yang et al., 1998). For example, Gautier et al. (1990) found a correlation of isotopic composition and sample age: the youngest basalts were the most alkaline and isotopically enriched. They postulated that the input of the DMM component declined and that of the plume component increased

during the ~40 Ma that the SEIR has been moving northeastward, away from the Kerguelen plume. However, no basalt from Kerguelen that is as isotopically depleted as the normal Indian MORB has yet been found (although some Indian Ocean MORB with hotspot influence do extend into field of Kerguelen basalts, Figure 4-17). Though the proposed plume endmember is commonly seen in the Kerguelen Islands basalts, the DMM endmember has not yet been observed in any of the mafic rocks. More recent results from Yang et al. (1998) found the entire range of isotopic composition of the Kerguelen Islands basalts in one stratigraphic sequence. These basalts had erupted over a ~1 Ma time span between 29-30 Ma (Nicolaysen et al., 1999) in an oscillatory fashion between two isotopic extremes (e.g. $^{87}\text{Sr}/^{86}\text{Sr} = 0.7040$ and 0.7053). Consequently, the origin of those basalts could not be explained by simple DMM-plume mixing where the proportions changed with time. These authors proposed an origin for these rocks as the tapping of distinct melts from an isotopically enriched plume and a relatively isotopically depleted plateau lower crust, as represented by the granulites reported by Mattielli (1996). That even the most isotopically depleted granulites still possess the Sr and Nd characteristics of Kerguelen basalts, suggests that a DMM-plume mixing origin for the granulites encounters the same difficulty as that for the basalts.

Another dilemma exists for the DMM-plume mixing hypothesis. It is well known that mantle plumes can affect the chemistry of MORB (e.g. Schilling, 1985), and geophysical studies of plume-ridge interaction indicate that plume mantle flows towards mid-ocean ridges even when they are separated by up to 1000 km (Schilling, 1991; Small, 1995; Ribe, 1996; Yale and Phipps Morgan, 1998). Given that plume mantle flows toward ridges because of thermal and rheological gradients, it appears that the invoked DMM-plume geochemical mixing could occur only when a plume is located directly beneath a ridge. For the case of the Kerguelen Islands basalts, if one assumes endmember compositions for the

SEIR melt (pure DMM with $^{87}\text{Sr}/^{86}\text{Sr} = 0.70275$, $^{143}\text{Nd}/^{144}\text{Nd} = 0.51315$; and depleted MORB concentrations with 100 ppm Sr, 4.5 ppm Nd) and the plume melt ($^{87}\text{Sr}/^{86}\text{Sr} = 0.70550$, $^{143}\text{Nd}/^{144}\text{Nd} = 0.51255$, 339 ppm Sr, and 20 ppm Nd: GM-59 of Yang et al, 1998), the most depleted Kerguelen Islands basalt ($^{87}\text{Sr}/^{86}\text{Sr} = .7040$ and $^{143}\text{Nd}/^{144}\text{Nd} = 0.5219$) requires a mix of 80% SEIR melt and 20% plume melt. Changing the SEIR endmember to an isotopically less depleted composition causes the mixing proportions to become even greater for the SEIR endmember. Eruption of such a ridge dominated melt as a part of plume volcanism appears unlikely even for a ridge-centered plume, considering the greater melt production rate for a plume at a depth where ridge melting is occurring. For basalts erupted after SEIR moved away from the Kerguelen plume, the physical difficulties for mixing distal ridge melts into plume magmas appear insurmountable and require that the geophysical reasoning behind the popular model be re-invented.

The Os data also do not support a simple plume-DMM mixing model. On an Os v. Sr isotopic diagram (Figure 4-21) the bulk of the granulite data lie halfway between DMM and EM II, but in Os v. Nd space (Figure 4-21), the granulites nearly overlap the proposed EM II component. If the isotopic compositions of these rocks were a result of DMM-plume mixing it would be DMM-EM II mixing which contradicts a general view that the Kerguelen plume is strongly EM I, and the relative proportions would need to be different for Sr and Nd. Even if the Os isotopic characteristics of EM I were similar to the Kerguelen granulites, the mismatch in mixing proportions for Sr and Nd remain. Overall, these isotopic data argue against a MORB-plume mixing for the observed isotopic characteristics for the granulites.

Whole rock powders from six granulite xenoliths from the Southeastern Province of the Kerguelen Islands were analyzed for Pb by Mattielli et al. (1996). These data plot near the unradiogenic end of the range defined by Kerguelen Islands basalts and within that of

Indian Ocean ridge basalts. Thus, these granulites were interpreted to represent “deep cumulates of tholeiitic magmas” that were DMM-plume (both EM I and EM II) mixtures with high proportions of the DMM component. Given the body of the available isotopic data for both basalts and granulites, no coherent mixing model can explain the observed isotopic variations.

Contamination of the granulites’ parental basalt magmas with differing amounts of continental material could also result in the observed Sr and Nd variation. This idea is not unreasonable because evidence for continental material does exist in various parts of the Kerguelen Plateau. Basalts drilled from the southern Kerguelen Plateau have trace element and isotopic compositions indicative of continental contamination (Alibert, 1991; Salters et al., 1992; Mahoney et al., 1994), some peridotite xenoliths from the archipelago have an Os signature of subcontinental lithosphere, and a clinopyroxene megacryst has $^{87}\text{Sr}/^{86}\text{Sr}$ and $^{143}\text{Nd}/^{144}\text{Nd}$ characteristic of continental crust contamination (Chapter 2; Chapter 3; Hassler and Shimizu, 1998). However, no definitive continental signatures have been discovered in the basalts from the islands, and all of the studies on basalts ruled out a continental source of isotopic variation (Gautier et al., 1990; Weis et al., 1993; Yang et al., 1998).

The data shown in Figure 4–19 imply that crustal contamination could significantly change the Sr and Nd isotopic characteristics of mafic magmas, and the parental magmas that precipitated the granulite precursor gabbros might have been contaminated by Gondwanaland lower crust. The granulites of this study and those of Mattielli et al. (1996) plot on the “mantle array”, and if their Sr and Nd variations were produced by crustal contamination, the amounts of contaminant must have been very small. Such crustal contamination could alter the Os isotopic composition toward much more radiogenic

values. The closeness of the Os isotopic data of the granulites of the present study to the present-day primitive mantle value, strongly argues further against crustal contamination.

The favored alternative to explain the isotopic diversity of the granulites, and by their genetic association, the basalts as well, is plume heterogeneity. It is still unclear whether the Kerguelen plume consists of mantle material with two different isotopic compositions that melt and mix in various proportions to eventually form the granulites and basalts, or if more endmembers within the plume are required. In the proposed scenario, a depleted component of the plume would be similar isotopically to the most depleted Kerguelen basalt or granulite observed, and the enriched end member could be isotopically like the host dike or the metasomatic melts described in Chapter 2.

It is notable that although the granulites have virtually the same Sr and Nd isotopic composition as the Kerguelen basalts, the eruption age corrected $^{187}\text{Os}/^{188}\text{Os}$ are much less radiogenic (Reisberg et al., 1993; Yang et al., 1998). The exception is sample OB93-54; the $^{187}\text{Os}/^{188}\text{Os}$ is as radiogenic as the basalts, but its concentration is much higher (Figure 4-18). Conversely, the Sr, Nd, and Os isotopes of two basalts from Heard Island (Barling et al., 1994; Barling et al., 1997), which also have a Kerguelen plume origin, are like the majority of the Kerguelen granulites. Thus, the granulites cover the known $^{187}\text{Os}/^{188}\text{Os}$ range of Kerguelen plume basalts. This implies that the plume is heterogeneous in its Os isotopic composition as well as the other isotopes. But because Sr, Nd, and Os data exist only for a few Kerguelen Islands basalts and the granulites of this study, the full Os isotopic spectrum of the Kerguelen plume may not yet be known.

Yang et al. (1998) discounted the heterogeneous plume hypothesis for the Kerguelen Islands basalts. One reason was because the samples with the most isotopic variation have a mafic, plagioclase-rich component in their source (group D basalts). They named the

Kerguelen Plateau lower crust as the source of this mafic component on the basis of the geochemistry of the granulites measured by Mattielli et al. (1996). These authors postulated that plume heat induced the melting of the lower crust which was inferred to be made of SEIR-derived cumulates by Mattielli et al. (1996). Yang et al. (Yang et al., 1998) further noted that the most isotopically enriched basalts had the least isotopic variation and no mafic source component, and postulated that this isotopic signature was intrinsic to the plume. The major difficulty with this model stems from the REE patterns of the group D melts, and from the fact that they are undersaturated with plagioclase. Figure 8 of Yang et al. (1998) shows that the group D basalts have $[Sm/Yb]_N > 3$ for all but one (GM92-46), indicating their origin of melting in the presence of garnet. This is not consistent with the melting of plagioclase-bearing, SEIR-derived cumulates at low pressures by plume heat. The absence of plagioclase phenocrysts in these basalts indicates that the group D melts are undersaturated with plagioclase. Again this inconsistent with the hypothesis of melting of plagioclase bearing lower crustal rocks, and points to a plume origin instead.

Sobolev and Hofmann (1998) recently discussed a similar situation found in Mauna Loa melt inclusions. These authors found basaltic melt inclusions with large positive Sr and smaller positive Eu anomalies included in high-Mg olivines. Despite these obvious plagioclase signatures, these melts were found to be undersaturated with plagioclase. Their explanation was that a recycled plagioclase-bearing cumulate component was present within the Hawaiian plume and melted at depths where plagioclase is absent. The same explanation could apply to the group D basalts of the Kerguelen Islands, and suggests that these basalts are products of the Kerguelen plume, rather than melts of the plateau lower crust.

This further implies that the Kerguelen plume is heterogeneous with respect to its Sr and Nd isotopic compositions and that distinct parts of the plume were tapped in succession

to produce the basalts of the Mt. Bureau and Rabouillere without mixing. The co-existence of different melts in close proximity in space and time occur in nature and has been well documented by melt inclusion studies (Sobolev and Shimizu, 1994; Shimizu, 1998; Shimizu and Grove, 1998) and the lack of mixing in the Kerguelen basalts need not be surprising.

4.5.8. Cumulate emplacement in time and space

It has been discussed above that the trace element and isotopic compositions of the granulites indicate a strong genetic link to the Kerguelen plume. Their mineralogical constitution and major element compositions indicate their origin as gabbroic cumulates that were recrystallized under granulite facies conditions as shown by Figure 4–8. The metamorphic reaction responsible for granulite formation is $\text{olivine} + \text{plagioclase} \rightarrow \text{clinopyroxene} + \text{orthopyroxene} + \text{spinel}$. For this reaction to occur, the gabbro protoliths must have either experienced an increase in pressure or decrease in temperature (Figure 4–8). An increase in depth, and thereby an increase in pressure, could have occurred as the originally subaerial Kerguelen Plateau subsided and the repeated volcanic eruptions that built the Plateau and Islands were piled onto the surface. This scenario seems unlikely however, in that subsidence rates are too small. The estimated maximum subsidence of the Kerguelen Plateau during a 50 Ma time span is only 1–2 km (Munschy and Schlich, 1987; Coffin, 1992). Thus, it appears more likely that the granulites formed by crossing the reaction boundary as a result of isobaric cooling. It follows that a thick northern Kerguelen Plateau already existed when cumulate emplacement occurred. Clearly, cumulate formation at the SEIR, where crustal thickness is approximately 7 km, does not satisfy the pressure conditions. In addition to the problems for pressure increase due to subsidence, the idea that the Kerguelen granulites are “deep cumulates from tholeiitic magma” (Mattielli et al., 1996; Grègoire et al., 1998) must be reconsidered.

Seismic experiments showed that the crustal thickness beneath the Kerguelen Islands and the Kerguelen Plateau ranges from 15 to 23 km (Recq, 1990, 1994, Charvis, 1993, 1995, Operto and Charvis, 1995, 1996). Because of the presence of both pyroxene and garnet granulite assemblages, and of sapphirine-bearing assemblages (Christy, 1989), the pressure of metamorphism must have been at least 0.6-0.7 GPa, or about 21 km deep (Figure 4-8). This observation suggests that the granulite protoliths formed by magmatic underplating at the base of the crust. Mechanisms by which the lower oceanic crust forms have been a subject of debate and recent development based on studies of ODP site 735B (Dick et al., 1992; Hart et al., 1999) and the Oman ophiolite (Kelemen et al., 1997) can be used to infer processes of underplating for the present case. Particularly interesting are the observations of Kelemen et al. (1997) for the emplacement of gabbro sills in the Moho transition zone of the Oman ophiolite. Numerous small (<1 to 50 m thick by >10 to 200 m long) gabbro bodies are formed by the crystallization of melts at the Moho. In this model, ascending melts are trapped at the Moho and crystallize because of permeability and density barriers (Stolper and Walker, 1980; Parsons et al., 1992; Kelemen et al., 1997). The more evolved residual melts are effectively extracted by a mechanism of periodic hydrofracture of the overlying brittle crust, leaving behind a nearly pure cumulate. In addition to being observed in Oman (Kelemen et al., 1997), these types of gabbros are inferred by geophysical methods to be currently in the process of forming at the East Pacific Rise (Crawford et al., 1999), and their formation was successfully modeled in hotspot settings using the MELTS code (Farnetani et al., 1996). Moreover, the presence of underplated lower crustal material was observed seismically beneath Hawaii and the Marquesas (ten Brink and Brocher, 1987; Caress et al., 1995). Thus, underplating seems to be a common process of lower crustal growth in oceanic settings, both ridge and hotspot.

If the metamorphism occurred by isobaric cooling, then why and when did the cooling begin? This is somewhat problematic in that cooling of the lithosphere is not easily explained in a plume setting. The absolute motion of Antarctic plate is in a slow northeasterly direction, and it is possible that the Islands have moved away from the plume, although the current location of the Kerguelen plume is not known (e.g. Duncan, 1991; Duncan and Storey, 1992; Royer et al., 1991). It is clear however, that plume activity is now a fraction of what it was at times during the last 30 Ma, i.e. the plume may be waning. Together, the plate movement and the waning plume may have allowed the temperature of the lithosphere to decrease enough to drive the metamorphism. If the high temperature anomaly is completely removed, conductive cooling can happen relatively quickly. Ten km of lithosphere can cool to the ambient geothermal gradient within 300 ka after the heat source is removed.

Why are granulites present beneath the Kerguelen Islands and not other oceanic Islands? The answer is that they may well be present beneath other islands, but they have just not been sampled by plume volcanism. As noted earlier, other oceanic island platforms, including Hawaii, the Marquesas and the Ontong Java Plateau, have indications that large igneous bodies are underplated below (ten Brink and Brocher, 1987; Caress et al., 1995). Thus, granulites may potentially exist beneath these islands as well.

It has also been discussed in this chapter that the Sr and Nd isotope variations in the granulites are due to inherent heterogeneities of the Kerguelen plume. The fact that the small isotopically and geochemically heterogeneous pieces of recrystallized gabbro cumulates are sampled by the Kerguelen Islands dikes suggests that the original emplacement of gabbros might have occurred in numerous small units, similar to those observed in Oman (Kelemen et al., 1997). In this regard, it is notable that granulites from individual localities appear to form distinct groups of Sr and Nd isotopic compositions.

Figure 4–19 shows that granulites from Mt Tizard have the lowest $^{87}\text{Sr}/^{86}\text{Sr}$ and highest $^{143}\text{Nd}/^{144}\text{Nd}$ (Mattielli et al., 1996), followed by the Lac Superieur granulites of this study, and those from Dome Rouge have the most isotopically enriched signatures (Mattielli, 1996). The combination of these isotopic characteristic with the locality-dependent differences in the relative abundances of the different xenolith types and granulite mineralogy implies that plume-derived magmas with distinct isotopic characteristics were emplaced at different depths in the crust-mantle transition zone and under varied thermal conditions. That is, because garnet and sapphirine-bearing granulites are present only in the Southeastern Province, and two-pyroxene granulites in the Courbet Peninsula, the crustal thickness is probably different between these two localities.

4.5.9. What is the $^{187}\text{Os}/^{188}\text{Os}$ of the EM I component?

The Os composition of the EM I component is not yet well defined, though there is some indication that the EM I end member has radiogenic $^{187}\text{Os}/^{188}\text{Os}$ relative to EM II (Hauri and Hart, 1993; Reisberg et al., 1993; Yang et al., 1998; Shirey and Walker, 1998). Basalts from the Kerguelen plume have a strong EM I component (e.g. Gautier et al., 1990; Weis et al., 1993; Yang et al., 1998), so the Os data from the Kerguelen plume related rocks may give some indication of the Os isotopic composition of EM I. The eruption age corrected basalts and granulites fall into two Os isotopic groups, a relatively radiogenic group ($^{187}\text{Os}/^{188}\text{Os} = 0.1600$) and unradiogenic group ($^{187}\text{Os}/^{188}\text{Os} = 0.1300$; Figure 4–21). As noted above, that basalts and granulites are present in both isotopic groups is evidence for a heterogeneous plume. It follows that the Os isotopic signature of the EM I component of the Kerguelen plume is heterogeneous as well. This is consistent with the origin of the EM I component in the subcontinental lithosphere (Hart, 1988).

One caveat to this idea, is that the absolute ages of the granulites are not known.

Corrections for a 70 Ma crystallization age for the granulite protoliths gives a range in

$^{187}\text{Os}/^{188}\text{Os}$ of 0.1266 to 0.1337, and OB93-54, the most radiogenic sample, has $^{187}\text{Os}/^{188}\text{Os} = 0.1285$ (see Figure 4-18). Hence, at 70 Ma the Os isotopic heterogeneity of these rocks significantly decreases. At this time, rocks from the DSDP site 216 indicate that the Ninetyeast Ridge was being constructed by the Kerguelen plume at $\sim 50^\circ\text{S}$ (as determined by paleomagnetic studies; e.g. Royer et al., 1991). Therefore, it is possible that these granulites' precursor could have been underplated during the Ninetyeast Ridge formation. Preliminary seismic data indicate the crust is about 25 km thick beneath the ridge near ODP site 757, and also exhibits evidence for crustal underplating (Flueh et al., 1999). Although these conditions satisfy those necessary for granulite metamorphism, the geography is more difficult to reconcile. For the protolith of these granulites to be related to Ninetyeast Ridge formation, a sizeable piece of the Ninetyeast Ridge should be present beneath the northern Kerguelen Plateau. That is, a piece of the ridge would have been rifted from the rest of the structure and subsequently covered by the volcanism that created the northern plateau and the Kerguelen Islands. A determination of the absolute ages of the granulites and a better understanding of the structure and tectonic history of the northern Kerguelen Plateau will be necessary to solve this problem.

4.5.10. What is the mineral host for Re and Os?

Because Re and Os are chalcophile elements, it is generally assumed that they are controlled by sulfides during processes such as melting and fractional crystallization (Jagoutz et al., 1979; Stockman, 1982; Morgan and Baedeker, 1983; Hart and Ravizza, 1996). Sulfides can account for as much as 80% of the Re and Os budget of peridotites, and therefore do have a significant role in their geochemistry (Jagoutz et al., 1979; Morgan and Baedeker, 1983; Hart and Ravizza, 1996). It seems clear that if sulfides are present in the magmatic system, they should dominate Re and Os geochemistry

The granulites of this may hold some clues to the geochemistry of Re and Os during crystallization of basaltic magmas, because as noted above, many of the granulites contain sulfides. Figure 4-22 shows a series of variation diagrams of whole rock abundances of chalcophile elements and Re and Os. No correlation exists between Re and Os, possibly indicating different mineral hosts for these elements. Correlations are present however, with Re and Os and other chalcophile elements, though not the same ones. Osmium has a good positive correlation with Zn and Pb, but not as good with Cu. In contrast, Re correlates positively with Cu and less well with Zn, but shows no correlation with Pb. These data indicate that both Re and Os are controlled by sulfide segregation, but not necessarily the same sulfide phase. If this is true, then Re tends to occur in the same sulfides as Cu but not Pb-rich sulfides, whereas Os will be present in the same sulfides as Pb rather than Cu-rich sulfides. Zinc can be present in sulfides that contain both Re and Os. Moreover, Os does not correlate with Ni, but rather remains relatively constant for a given Ni content and does not follow an olivine control line. This demonstrates that Os contents of these rocks are buffered by sulfide segregation (Hart and Ravizza, 1996). That many of the granulites contain sulfides supports this idea.

4.6 Summary

The results of this geochemical study of granulite xenoliths from the Courbet Peninsula, Kerguelen Islands provide the following conclusions:

The xenoliths of this study are two-pyroxene granulites, with olivine gabbro protoliths. Their equigranular texture, absence of olivine and high modal abundance of orthopyroxene provide evidence for a metamorphic origin. Differences in CIPW normative mineralogy (olivine + clinopyroxene + plagioclase) and modal mineralogy (clinopyroxene +

orthopyroxene + plagioclase) indicate that the protoliths were olivine gabbro and the reaction that formed the granulites was olivine + plagioclase $\rightarrow$ orthopyroxene + clinopyroxene + spinel. Although it is not possible to determine the absolute pressure of metamorphism for these rocks, on the basis of two-pyroxene thermometry and the CMAS experimental system reaction boundaries, pressure-temperature conditions must have been between 0.6 and 1.0 GPa, and from 845 to 960°C respectively.

The bulk rock major element compositions of these rocks are similar to both oceanic gabbros and mafic granulite xenoliths from continents. Thus, these type of data are not sufficient to determine the provenance of the parent melts of the granulites. Furthermore, these granulites are not related to each other by simple crystal fractionation of a single parent magma. Whole rock trace element abundances indicate that these rocks are cumulates. Positive correlations of compatible elements with MgO, and positive Sr and Eu anomalies of a similar magnitude as gabbro cumulates from the Oman ophiolite support this interpretation. The body of whole rock data also suggest that metamorphism was isochemical. Two line of evidence rule out a significant interstitial melt in the granulites: a.) The low level of incompatible elements present in the granulites argues for a minimal melt fraction that may have been trapped during mineral accumulation, and b.) Sr isotopes rule out impregnation of a lower oceanic crustal gabbro by a plume melt.

Rare earth element abundances of bulk granulites and of clinopyroxenes are used to infer the trace element compositions of the parent melts of the granulite protoliths. Equilibrium melts of both whole rocks and clinopyroxenes give similar REE patterns that are comparable to those of Kerguelen Islands basalts rather than SEIR MORB. This characteristics are show that the granulite protoliths are genetically related to the magmatism that produced the Kerguelen Islands basalts rather than the SEIR MORB.

The Sr and Nd isotopes of the granulites are in equilibrium with the Kerguelen Islands basalts, further supporting an origin of the precursor's parent melt from the Kerguelen plume. Combining the data of the present study with the isotope data of granulites from the Southeastern Province shows that the granulites as a whole span the isotopic range exhibited by the Kerguelen basalts. This implies a similar origin for the isotopic variations of the two.

The Re-Os systematics of the granulites are very similar to that of the present-day primitive mantle and show that the rocks cannot be very old, probably less than ~100 Ma. The Os isotopic compositions of the granulites are less radiogenic than samples of both the oceanic and continental lower crust. Together, these characteristics also point to a plume rather than mid-ocean ridge or lower continental crustal origin for the granulites.

The proposed origin of the diversity of isotopic signatures is a heterogeneous plume rather than a result of DMM-plume mixing, as was previously proposed. At 30 Ma, the age of the oldest basalts on the Kerguelen Islands, the Kerguelen plume would have been up to 400 km from the SEIR. Given that plume mantle flows towards ridges, it is unlikely that the mixing of 80% SEIR melt and 20% plume melt that is required by the Sr and Nd isotopes could have occurred. The Sr, Nd, and Os isotopic systematics also argue against any significant contamination of the granulite precursor parent melt by continental crust. That none of the isotopic systems (Rb-Sr, Sm-Nd, Re-Os and U-Pb) allow a coherent mixing model among the mantle endmembers is further evidence that the isotopic variations originated within the Kerguelen plume.

The parent melts of the granulite protoliths were most likely emplaced at the Moho (i.e. underplated) by Kerguelen plume related magmatism as a series of small bodies, similar to the gabbro sills observed at the Oman ophiolite. Metamorphism occurred by isobaric

cooling of the olivine gabbro protoliths, perhaps caused by migration of the Kerguelen Islands away from the plume, or a waning of the plume itself. The differences in the mineralogy of the granulites from the Southeast Province (garnet and sapphirine-bearing) and those of this study (two-pyroxenes, plagioclase and spinel), though the bulk chemistry is generally similar, indicates a variable depth of emplacement for the protoliths, and hence a variable crustal thickness beneath the Kerguelen Islands.

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4.8 Tables:

Table 4-1. Petrographic characteristics of the granulite xenolith suite.

Sample	olivine	opx	cpx	spinel	ilmenite	Notes
<i>High Mg# samples</i>						
OB93-50	-	+	+	+	-	Coarse symplectites, strained opx, cpx, and plag, cpx and opx exsolution
OB93-60	-	+	+	+	-	Coarse symplectites, strained plag, cpx and opx exsolution
OB93-71	trace	+	+	?	?	No thin section available
OB93-72	-	+	+	+	-	Coarse symplectite, poikilitic opx
OB93-90	-	+	+	+	-	Coarse symplectite, poikilitic opx, cpx and opx exsolution
OB93-96	-	+	+	+	-	Coarse symplectites, cpx and opx exsolution
<i>Low Mg# samples</i>						
OB93-54	-	+	+	-	+	Fine symplectites, strained plag, ilmenite rimmed by Ti-magnetite
OB93-57	-	+	+	-	+	Fine symplectites, cpx and opx exsolution
OB93-61A	-	+	+	-	?	No thin section available
OB93-68	-	+	+	trace	+	Fine symplectites, spinels rimmed by ilmenite and Ti-magnetite
OB93-69	-	+	+	trace	+	Fine symplectites, strained opx, spinels rimmed by ilmenite

Table 4-2. Major element and trace element compositions, CIPW normative compositions and modes of granulite whole rocks.

Sample	OB93-50	OB93-54	OB93-57	OB93-60	OB93-61A	OB93-68
Rock Type	H	L	L	H	L	L
<i>XRF wt. %</i>						
SiO ₂	48.51	46.27	48.93	48.31	48.99	47.36
Al ₂ O ₃	14.47	14.22	17.03	16.95	13.96	17.04
TiO ₂	0.25	2.22	0.37	0.36	0.74	0.62
FeO*	5.34	12.91	6.19	5.02	5.78	6.73
MnO	0.11	0.19	0.12	0.10	0.13	0.12
CaO	16.09	12.16	14.56	15.85	15.75	13.62
MgO	14.04	8.97	11.00	12.17	12.52	12.42
K ₂ O	0.02	0.05	0.04	0.05	0.16	0.08
Na ₂ O	0.93	2.18	1.78	1.15	1.40	1.32
P ₂ O ₅	0.02	0.14	0.02	0.03	0.02	0.03
Total	99.77	99.31	100.03	99.99	99.45	99.33

Mg# 82.42 55.33 76.01 81.21 79.43 76.69

Rock type: L = Low Mg# sample; H = High Mg# sample; *Total Fe as FeO

CIPW norms

fo	12.28	9.19	10.78	11.13	10.63	12.33
fa	2.84	6.90	3.66	2.71	2.72	3.86
en	3.62	1.97	2.15	2.84	1.71	5.31
fs	0.76	1.34	0.66	0.63	0.40	1.51
di	29.85	15.65	21.28	25.00	30.86	17.32
hd	5.46	9.29	5.72	4.82	6.25	4.29
wo	0.00	0.00	0.00	0.00	0.00	0.00
an	35.25	28.87	38.36	40.94	31.33	40.33
ab	7.87	18.45	15.06	9.73	11.85	11.17
or	0.12	0.30	0.24	0.30	0.95	0.47
ilm	0.47	4.21	0.69	0.68	1.41	1.17
ap	0.04	0.36	0.05	0.07	0.06	0.07
q	0.00	0.00	0.00	0.00	0.00	0.00
co	0.00	0.00	0.00	0.00	0.00	0.00
mt	1.16	2.81	1.35	1.09	1.26	1.46
ne	0.00	0.00	0.00	0.00	0.00	0.00
Cr ₂ O ₃	0.14	0.04	0.07	0.13	0.08	0.11

CIPW norms calculated assuming Fe₂O₃/FeO* = 0.15.

Modes

Cpx	57.85	n.d.	46.45	49.54	57.11	35.72
error	1.04	n.d.	1.44	0.75	3.24	5.87
Opx	18.42	n.d.	16.07	16.58	15.18	23.30
error	0.60	n.d.	0.87	0.44	2.28	4.18
Plag	20.53	n.d.	34.29	30.51	26.68	37.80
error	0.73	n.d.	0.96	0.53	2.02	3.57
Sp	3.45	n.d.	3.15	3.20	n.d.	1.29
error	0.33	n.d.	0.43	0.24	-	1.96

Modes (in wt.%) are calculated by mass balance using average composition of non-symplectite minerals.

Error is 1 S.D.; n.d.: not determined

Table 4-2. Whole rock chemistry continued.

Sample	OB93-69	OB93-71	OB93-72	OB93-90	OB93-96
Rock Type	L	H	H	H	H
<i>XRF wt. %</i>					
SiO ₂	48.57	45.60	48.18	46.49	48.41
Al ₂ O ₃	19.54	11.68	20.29	18.48	15.72
TiO ₂	0.47	0.33	0.23	0.23	0.24
FeO*	6.02	8.65	3.58	4.23	4.96
MnO	0.11	0.15	0.07	0.08	0.10
CaO	14.10	11.53	16.44	15.48	16.09
MgO	9.48	21.17	9.57	10.86	13.42
K ₂ O	0.06	0.02	0.04	0.06	0.01
Na ₂ O	1.76	0.83	1.51	1.32	0.98
P ₂ O ₅	0.02	0.02	0.02	0.02	0.01
Total	100.12	99.98	99.93	97.25	99.94

Mg# 73.73 81.35 82.66 82.07 82.83

Rock type: L = Low Mg# sample; H = High Mg# sample;

*Total Fe as FeO

CIPW norms

fo	8.42	29.36	9.47	11.91	12.09
fa	3.13	7.33	2.10	2.78	2.71
en	4.54	1.91	0.00	0.00	3.28
fs	1.54	0.43	0.00	0.00	0.67
di	15.23	19.25	22.26	21.70	27.81
hd	4.49	3.80	3.91	4.01	4.94
wo	0.00	0.00	0.00	0.00	0.00
an	45.24	28.08	48.47	44.32	38.46
ab	14.89	7.02	11.47	10.20	8.29
or	0.35	0.12	0.24	0.35	0.06
ilm	0.89	0.62	0.44	0.44	0.45
ap	0.04	0.05	0.05	0.04	0.04
q	0.00	0.00	0.00	0.00	0.00
co	0.00	0.00	0.00	0.00	0.00
mt	1.31	1.88	0.78	0.92	1.08
ne	0.00	0.00	0.71	0.53	0.00
Cr ₂ O ₃	0.11	0.26	0.11	0.14	0.13

CIPW norms calculated assuming Fe₂O₃/FeO* = 0.15.

Modes

Cpx	33.26	n.d.	46.29	47.69	55.37
error	2.78	n.d.	1.63	0.94	1.18
Opx	17.17	n.d.	8.95	12.84	16.76
error	1.78	n.d.	0.95	0.56	0.68
Plag	46.64	n.d.	41.59	35.28	24.23
error	1.77	n.d.	1.13	0.66	0.82
Sp	2.92	n.d.	3.38	4.30	3.89
error	0.97	n.d.	0.50	0.30	0.37

Modes (in wt.%) are calculated by mass balance using average composition of non-symplectite minerals.

Error is 1 S.D.; n.d.: not determined

Table 4-2. Whole rock chemistry continued.

Sample	OB93-50	OB93-54	OB93-57	OB93-60	OB93-61A	OB93-68
Rock Type	H	L	L	H	L	L
<i>XRF ppm</i>						
Ni	247	83	186	217	209	231
Cr	960	263	465	871	525	747
Sc	45	45	43	44	53	38
V	212	402	170	196	268	207
Ba	d.l.	61	d.l.	d.l.	143	30
Sr	104	184	156	135	248	179
Zr	13	66	18	19	36	28
Y	5	27	7	6	10	7
Ga	14	18	15	14	12	15
Cu	4	245	5	11	13	41
Zn	22	94	33	24	16	43
<i>ICP-MS ppm</i>						
La	0.52	3.71	0.63	1.25	2.67	2.07
Ce	0.98	11.51	1.52	2.50	6.85	4.84
Pr	0.19	1.92	0.27	0.38	1.06	0.72
Nd	1.04	10.97	1.56	1.87	5.42	3.45
Sm	0.49	3.96	0.72	0.69	1.74	1.14
Eu	0.30	1.52	0.44	0.38	0.68	0.56
Gd	0.74	5.01	1.01	0.98	2.04	1.34
Tb	0.15	0.90	0.19	0.18	0.36	0.24
Dy	0.95	5.78	1.23	1.10	2.23	1.52
Ho	0.19	1.17	0.25	0.23	0.45	0.30
Er	0.52	3.09	0.67	0.57	1.14	0.80
Tm	0.08	0.43	0.09	0.08	0.16	0.11
Yb	0.44	2.60	0.59	0.49	0.97	0.66
Lu	0.07	0.40	0.09	0.08	0.14	0.10
Ba	10.7	99.2	11.2	23.3	135.6	30.0
Th	0.07	0.03	0.02	0.12	0.21	0.09
Nb	0.36	6.86	0.22	1.18	1.65	0.58
Y	5.08	29.41	6.47	5.77	11.16	7.72
Hf	0.19	1.96	0.29	0.32	0.91	0.60
Ta	0.02	0.48	0.02	0.08	0.11	0.06
U	0.02	0.03	0.00	0.03	0.04	0.03
Pb	1.23	0.26	2.67	1.03	3.79	2.65
Rb	0.81	1.04	0.54	1.17	2.98	1.21
Cs	0.13	0.10	0.04	0.08	0.69	0.07
Sr	105	186	157	138	258	181
Sc	54.0	53.3	45.1	46.2	61.3	39.9

d.l.: below detection limit

Table 4-2. Whole rock chemistry continued.

Sample	OB93-69	OB93-71	OB93-72	OB93-90	OB93-96
Rock Type	L	H	H	H	H
<i>XRF ppm</i>					
Ni	167	701	189	241	238
Cr	762	1772	782	926	903
Sc	40	30	40	40	43
V	159	130	103	103	214
Ba	34	d.l.	41	125	d.l.
Sr	179	77	199	169	126
Zr	18	18	18	17	14
Y	6	5	4	3	5
Ga	17	9	13	11	14
Cu	39	55	16	15	6
Zn	35	72	16	22	16
<i>ICP-MS ppm</i>					
La	0.87	0.55	0.57	0.53	0.39
Ce	1.91	1.56	1.13	1.09	0.76
Pr	0.31	0.27	0.20	0.20	0.15
Nd	1.74	1.50	1.03	0.99	0.88
Sm	0.74	0.63	0.46	0.43	0.44
Eu	0.54	0.31	0.29	0.29	0.28
Gd	1.04	0.87	0.68	0.68	0.66
Tb	0.20	0.17	0.13	0.12	0.13
Dy	1.28	1.09	0.83	0.81	0.88
Ho	0.27	0.23	0.17	0.18	0.19
Er	0.71	0.59	0.45	0.45	0.47
Tm	0.10	0.08	0.06	0.06	0.07
Yb	0.61	0.52	0.38	0.37	0.41
Lu	0.09	0.08	0.06	0.05	0.06
Ba	22.1	3.2	37.7	116.7	27.6
Th	0.03	0.01	0.02	0.04	0.01
Nb	0.21	0.06	0.18	0.24	0.15
Y	6.65	5.51	4.25	4.11	4.65
Hf	0.29	0.36	0.21	0.21	0.16
Ta	0.01	0.02	0.02	0.02	0.01
U	0.01	0.00	0.00	0.01	0.00
Pb	1.31	18.74	0.21	0.62	3.23
Rb	0.93	0.51	0.71	1.93	0.33
Cs	0.05	0.08	0.29	0.35	0.07
Sr	183	78	195	166	125
Sc	36.9	35.9	35.7	36.1	51.1

d.l.: below detection limit

Table 4-3. Representative clinopyroxene major element compositions.

Sample	OB93-50	OB93-54	OB93-57		OB93-60	OB93-61A
<i>wt. %</i>						
SiO ₂	50.11	50.30	50.76	49.55	50.05	48.29
TiO ₂	0.47	0.61	0.72	0.79	0.43	1.20
Al ₂ O ₃	7.34	5.17	6.31	6.67	7.00	7.23
Cr ₂ O ₃	0.28	0.03	0.16	0.18	0.23	0.09
FeO*	4.76	9.85	8.74	6.53	4.75	6.34
MnO	0.10	0.10	0.16	0.11	0.12	0.13
MgO	13.82	11.78	15.53	13.16	14.05	12.89
CaO	21.74	20.19	16.98	20.61	21.64	21.58
Na ₂ O	0.74	1.35	0.87	1.02	0.77	0.99
Total	99.37	99.39	100.24	98.62	99.04	98.75
<i>Cations/6 Oxygens</i>						
Si	1.845	1.892	1.862	1.852	1.850	1.811
Ti	0.013	0.017	0.020	0.022	0.012	0.034
Al	0.319	0.229	0.273	0.294	0.305	0.319
Cr	0.008	0.001	0.005	0.005	0.007	0.003
Fe	0.147	0.310	0.268	0.204	0.147	0.199
Mn	0.003	0.003	0.005	0.003	0.004	0.004
Mg	0.759	0.660	0.849	0.733	0.774	0.721
Ca	0.858	0.814	0.667	0.825	0.857	0.867
Na	0.053	0.099	0.062	0.074	0.056	0.072
Total	4.005	4.025	4.011	4.013	4.010	4.030
Mg#	83.80	68.06	76.00	78.21	84.05	78.36
Eq Liq Mg#	56.93	35.26	44.74	47.85	57.39	48.06
<i>Molar percentage of endmembers</i>						
Wo	48.64	45.60	37.39	46.82	48.19	48.53
En	43.04	37.02	47.58	41.60	43.54	40.33
Fs	8.32	17.37	15.02	11.59	8.26	11.14

Eq Liq Mg#: Mg# of cpx equilibrium melt, calculated from

[(Fe²⁺/Mg)cpx/(Fe²⁺/Mg)liq]=0.23 (Grove, and Bryan, 1983); and Fe²⁺/Fe³⁺=0.9.

Table 4-3. Clinopyroxene major element compositions continued.

Sample	OB93-68	OB93-69	OB93-71	OB93-72	OB93-90	OB93-96
<i>wt. %</i>						
SiO ₂	48.73	49.87	50.52	50.36	51.01	50.50
TiO ₂	1.50	1.20	0.61	0.60	0.52	0.44
Al ₂ O ₃	7.49	6.17	7.14	7.59	7.19	7.29
Cr ₂ O ₃	0.13	0.28	0.20	0.20	0.21	0.26
FeO*	6.84	7.07	4.39	4.87	4.73	4.77
MnO	0.11	0.14	0.13	0.11	0.07	0.12
MgO	12.66	12.71	13.89	13.52	14.19	14.22
CaO	21.41	21.37	20.92	20.99	21.53	21.78
Na ₂ O	0.93	0.95	1.29	0.85	0.84	0.74
Total	99.81	99.76	99.09	99.10	100.28	100.12
<i>Cations/6 Oxygens</i>						
Si	1.809	1.852	1.860	1.855	1.857	1.845
Ti	0.042	0.034	0.017	0.017	0.014	0.012
Al	0.328	0.270	0.310	0.329	0.308	0.314
Cr	0.004	0.008	0.006	0.006	0.006	0.008
Fe	0.212	0.220	0.135	0.150	0.144	0.146
Mn	0.004	0.004	0.004	0.003	0.002	0.004
Mg	0.700	0.703	0.762	0.742	0.770	0.775
Ca	0.852	0.850	0.825	0.828	0.840	0.853
Na	0.067	0.068	0.092	0.061	0.059	0.052
Total	4.017	4.010	4.012	3.991	4.001	4.008
Mg#	76.73	76.20	84.95	83.18	84.24	84.15
Eq Liq Mg#	45.73	45.00	59.05	55.82	57.74	57.57
<i>Molar percentage of endmembers</i>						
Wo	48.27	47.96	47.90	48.14	47.89	48.09
En	39.69	39.66	44.25	43.13	43.90	43.68
Fs	12.04	12.39	7.84	8.72	8.21	8.23

Eq Liq Mg#: Mg# of cpx equilibrium melt, calculated from

[(Fe²⁺/Mg)_{cpx}]/[(Fe²⁺/Mg)_{liq}]=0.23 (Grove, and Bryan, 1983); and Fe²⁺/Fe³⁺=0.9.

Table 4-4. Representative orthopyroxene major element compositions.

Sample	OB93-50	OB93-54	OB93-57	OB93-60	OB93-61A	OB93-68
wt. %						
SiO ₂	52.70	51.43	52.35	52.42	51.37	53.40
TiO ₂	0.08	0.15	0.09	0.05	0.13	0.72
Al ₂ O ₃	6.01	2.32	2.24	7.25	5.23	1.76
Cr ₂ O ₃	0.01	0.03	0.03	0.06	0.07	0.07
FeO*	12.65	23.69	22.78	12.37	15.59	16.46
MnO	0.20	0.31	0.24	0.17	0.29	0.33
MgO	28.33	21.97	22.22	27.95	25.85	25.65
CaO	0.53	0.60	0.60	0.54	0.50	2.02
Na ₂ O	0.06	0.01	0.00	0.06	0.02	0.05
Total	100.56	100.51	100.54	100.87	99.07	100.44
<i>Cations/6 Oxygens</i>						
Si	1.863	1.920	1.941	1.844	1.872	1.935
Ti	0.002	0.004	0.002	0.001	0.004	0.020
Al	0.250	0.102	0.098	0.301	0.225	0.075
Cr	0.000	0.001	0.001	0.002	0.002	0.002
Fe	0.374	0.740	0.706	0.364	0.475	0.499
Mn	0.006	0.010	0.008	0.005	0.009	0.010
Mg	1.493	1.223	1.228	1.465	1.404	1.385
Ca	0.020	0.024	0.024	0.020	0.020	0.078
Na	0.004	0.001	0.000	0.004	0.002	0.004
Total	4.012	4.024	4.007	4.006	4.012	4.008
Mg#	79.97	62.31	63.48	80.10	74.72	73.53
<i>Molar percentage of endmembers</i>						
Wo	1.07	1.21	1.21	1.10	1.04	3.99
En	79.11	61.55	62.71	79.22	73.94	70.59
Fs	19.82	37.24	36.08	19.68	25.02	25.41

Table 4-4. Orthopyroxene major element compositions continued.

Sample	OB93-69	OB93-71	OB93-72	OB93-90	OB93-96
wt. %					
SiO ₂	51.57	51.66	51.43	52.15	52.81
TiO ₂	0.10	0.15	0.09	0.08	0.06
Al ₂ O ₃	4.32	4.35	6.98	6.29	6.17
Cr ₂ O ₃	0.14	0.18	0.10	0.11	0.02
FeO*	18.95	19.83	12.90	12.50	12.40
MnO	0.31	0.28	0.20	0.18	0.15
MgO	23.85	23.44	27.01	28.01	28.41
CaO	0.46	0.79	0.60	0.55	0.51
Na ₂ O	0.03	0.01	0.07	0.08	0.06
Total	99.71	100.69	99.37	99.93	100.57
<i>Cations/6 Oxygens</i>					
Si	1.895	1.889	1.907	1.855	1.863
Ti	0.003	0.004	0.002	0.002	0.002
Al	0.187	0.188	0.188	0.264	0.256
Cr	0.004	0.005	0.002	0.003	0.001
Fe	0.582	0.607	0.340	0.372	0.366
Mn	0.010	0.009	0.006	0.005	0.005
Mg	1.306	1.278	1.443	1.485	1.494
Ca	0.018	0.031	0.023	0.021	0.019
Na	0.002	0.001	0.005	0.006	0.004
Total	4.007	4.010	4.007	4.012	4.009
Mg#	69.16	67.81	78.87	79.98	80.32
<i>Molar percentage of endmembers</i>					
Wo	0.95	1.61	1.25	1.11	1.02
En	68.51	66.72	77.89	79.09	79.50
Fs	30.54	31.67	20.87	19.80	19.48

Table 4-5. Representative plagioclase major element compositions.

Sample	OB93-50	OB93-54	OB93-57	OB93-60	OB93-61A
<i>wt. %</i>					
SiO ₂	47.80	47.18	49.73	47.41	50.26
Al ₂ O ₃	34.07	33.82	32.02	33.71	31.61
FeO*	0.08	0.15	0.06	0.10	0.15
MgO	0.03	0.05	0.02	0.04	0.02
CaO	16.68	15.59	14.29	16.48	13.94
Na ₂ O	2.00	2.46	3.02	1.91	3.41
K ₂ O	0.06	0.04	0.10	0.04	0.32
Total	100.71	99.28	99.24	99.68	99.71
<i>Cations/8 Oxygens</i>					
Si	2.175	2.176	2.281	2.179	2.298
Al	1.827	1.838	1.731	1.826	1.703
Fe	0.003	0.006	0.002	0.004	0.006
Mg	0.002	0.004	0.002	0.002	0.002
Ca	0.813	0.771	0.702	0.811	0.683
Na	0.177	0.220	0.269	0.170	0.302
K	0.003	0.002	0.006	0.002	0.019
Total	5.001	5.016	4.991	4.995	5.011
<i>Mole percentage of endmembers</i>					
Ab	17.79	22.12	27.53	17.32	30.10
An	81.89	77.64	71.90	82.46	68.03
Or	0.32	0.24	0.57	0.22	1.86
n.d.: not determined					

Table 4-5. Plagioclase major element compositions continued.

Sample	OB93-68	OB93-69	OB93-71	OB93-72	OB93-90	OB93-96
wt. %						
SiO ₂	50.86	49.40	n.d.	47.60	47.95	47.48
Al ₂ O ₃	31.36	32.41	n.d.	33.32	33.40	33.88
FeO*	0.16	0.17	n.d.	0.08	0.06	0.11
MgO	0.03	0.05	n.d.	0.03	0.03	0.03
CaO	13.60	14.81	n.d.	15.97	15.63	16.40
Na ₂ O	3.69	3.02	n.d.	2.21	2.52	1.98
K ₂ O	0.13	0.10	n.d.	0.07	0.06	0.04
Total	99.82	99.95		99.27	99.64	99.91
<i>Cations/8 Oxygens</i>						
Si	2.318	2.256		2.195	2.201	2.177
Al	1.684	1.744		1.811	1.807	1.831
Fe	0.006	0.006		0.003	0.002	0.004
Mg	0.002	0.003		0.002	0.002	0.002
Ca	0.664	0.725		0.789	0.769	0.806
Na	0.326	0.269		0.198	0.224	0.176
K	0.008	0.006		0.004	0.003	0.002
Total	5.007	5.008		5.001	5.009	4.997
<i>Mole percentage of endmembers</i>						
Ab	32.66	26.82		19.98	22.49	17.85
An	66.56	72.61		79.62	77.17	81.91
Or	0.78	0.57		0.39	0.34	0.23
n.d.: not determined						

Table 4-6. Olivine major element compositions.

Sample	OB93-71
<i>wt. %</i>	
SiO ₂	39.44
TiO ₂	0.02
Al ₂ O ₃	0.00
Cr ₂ O ₃	0.01
FeO*	18.86
MnO	0.22
MgO	41.84
CaO	0.04
Na ₂ O	0.00
Total	100.42
<i>Cations/4 Oxygens</i>	
Si	1.003
Ti	0.000
Al	0.000
Cr	0.000
Fe	0.401
Mn	0.005
Mg	1.586
Ca	0.001
Na	0.000
Cation Total	2.997
Fo content	79.81

Table 4-7. Representative spinel major element compositions.

Sample	OB93-50				OB93-54				OB93-57				OB93-60				OB93-68			
wt. %																				
SiO ₂	0.06	0.02	0.22	0.00	0.07	0.11	0.06	0.06	0.11	0.06	0.00	0.06	0.11	0.11	0.10	0.10	0.11	0.11	0.10	
TiO ₂	0.02	0.01	0.48	49.95	20.45	0.12	0.12	0.12	0.12	0.12	0.04	0.12	0.04	0.04	0.00	1.76	50.75	50.75	1.76	
Al ₂ O ₃	65.05	64.72	44.06	0.18	3.55	62.18	60.55	60.55	62.18	60.55	64.64	60.55	64.64	65.77	0.00	47.93	0.00	0.00	47.93	
Cr ₂ O ₃	0.34	0.99	0.06	0.08	0.43	0.17	1.04	1.04	0.17	1.04	1.22	1.04	1.22	0.44	0.19	5.84	0.19	0.19	5.84	
FeO*	15.37	16.12	46.61	45.54	70.65	21.52	23.26	23.26	21.52	23.26	16.22	15.93	16.22	15.93	45.31	31.15	45.31	45.31	31.15	
MnO	0.10	0.08	0.56	0.27	0.33	0.13	0.10	0.10	0.13	0.10	0.11	0.08	0.11	0.08	0.28	0.18	0.28	0.28	0.18	
MgO	18.76	18.00	5.66	3.09	2.22	15.20	14.68	14.68	15.20	14.68	17.75	18.05	17.75	18.05	2.88	12.65	2.88	2.88	12.65	
CaO	0.12	0.02	0.12	0.00	0.01	0.18	0.03	0.03	0.18	0.03	0.01	0.01	0.01	0.01	0.09	0.08	0.09	0.09	0.08	
NiO	0.31	0.32	0.00	0.02	0.05	0.23	0.29	0.29	0.23	0.29	0.30	0.33	0.30	0.33	0.38	0.28	0.38	0.38	0.28	
Total	100.12	100.27	97.77	99.11	97.76	99.83	100.13	100.13	99.83	100.13	100.30	100.65	100.30	100.65	100.00	99.97	100.00	100.00	99.97	
FeO	13.25	14.47	29.98	39.14	47.08	18.29	19.04	19.04	18.29	19.04	14.85	14.71	14.85	14.71	39.87	21.61	39.87	39.87	21.61	
Fe ₂ O ₃	2.36	1.83	18.49	7.12	26.19	3.58	4.69	4.69	3.58	4.69	1.53	1.35	1.53	1.35	6.04	10.60	6.04	6.04	10.60	
New Total	100.36	100.45	99.63	99.82	100.39	100.19	100.60	100.60	100.19	100.60	100.45	100.79	100.45	100.79	100.44	101.03	100.44	100.60	101.03	
Cations/3 or 4 Oxygens																				
Si	0.002	0.001	0.007	0.000	0.003	0.003	0.002	0.002	0.003	0.002	0.000	0.001	0.000	0.001	0.003	0.003	0.003	0.003	0.003	
Ti	0.000	0.000	0.011	0.951	0.613	0.002	0.002	0.002	0.002	0.002	0.001	0.000	0.001	0.000	0.957	0.038	0.957	0.957	0.038	
Al	1.955	1.952	1.634	0.005	0.167	1.933	1.900	1.900	1.933	1.900	1.951	1.969	1.951	1.969	0.000	1.615	0.007	0.000	1.615	
Cr	0.007	0.020	0.002	0.002	0.014	0.003	0.022	0.022	0.003	0.022	0.025	0.009	0.025	0.009	0.004	0.132	0.020	0.004	0.132	
Fe	0.328	0.345	1.226	0.964	2.354	0.475	0.518	0.518	0.475	0.518	0.347	0.338	0.347	0.338	0.950	0.745	0.814	0.950	0.745	
Mn	0.002	0.002	0.015	0.006	0.011	0.003	0.002	0.002	0.003	0.002	0.002	0.002	0.002	0.002	0.006	0.004	0.009	0.006	0.004	
Mg	0.713	0.687	0.265	0.117	0.132	0.598	0.583	0.583	0.598	0.583	0.678	0.684	0.678	0.684	0.108	0.539	0.202	0.108	0.539	
Ca	0.003	0.000	0.004	0.000	0.000	0.005	0.001	0.001	0.005	0.001	0.000	0.000	0.000	0.000	0.002	0.002	0.002	0.002	0.002	
Ni	0.006	0.006	0.000	0.000	0.002	0.005	0.006	0.006	0.005	0.006	0.006	0.007	0.006	0.007	0.008	0.006	0.004	0.008	0.006	
Cation Total	3.017	3.013	3.164	2.045	3.294	3.027	3.035	3.035	3.027	3.035	3.011	3.010	3.011	3.010	2.038	3.086	2.023	2.038	3.086	
Oxygen Total	3.978	3.982	3.793	2.934	3.643	3.965	3.954	3.954	3.965	3.954	3.985	3.987	3.985	3.987	2.944	3.889	2.967	2.944	3.889	
Fe ²⁺	0.281	0.308	0.748	0.810	1.428	0.400	0.419	0.419	0.400	0.419	0.317	0.312	0.317	0.312	0.821	0.502	0.739	0.821	0.502	
Fe ³⁺	0.045	0.035	0.415	0.133	0.715	0.071	0.093	0.093	0.071	0.093	0.029	0.026	0.029	0.026	0.112	0.222	0.067	0.112	0.222	
New Cat. Total	3.015	3.012	3.101	2.024	3.084	3.022	3.029	3.029	3.022	3.029	3.010	3.009	3.010	3.009	2.020	3.065	2.013	2.020	3.065	
New O Total	4.000	4.000	4.010	3.009	4.034	4.001	4.001	4.001	4.001	4.001	4.000	4.000	4.000	4.000	3.008	4.004	3.004	3.008	4.004	
Cr#	0.35	1.01	0.10	22.35	7.56	0.18	1.14	1.14	0.18	1.14	1.25	0.44	1.25	0.44	100.00	7.56	74.51	100.00	7.56	
Mg#Fe ²⁺	71.74	69.01	26.20	12.58	8.46	59.91	58.17	58.17	59.91	58.17	68.14	68.69	68.14	68.69	11.62	51.77	21.48	11.62	51.77	
No. of Cations	3	3	3	2	3	3	3	3	3	3	3	3	3	3	2	3	2	2	3	
No of Oxygens	4	4	4	3	4	4	4	4	4	4	4	4	4	4	3	4	3	3	4	
Notes	cpx sympl.		in plag react zone	ilm core	Ti-mit rim	cpx sympl					coarse cpx sympl.	fine cpx sympl.			ilm	ilm, included in Fe-S			sp core	

Table 4-7. Spinel major element compositions continued.

Sample	OB93-68	OB93-69		OB93-71		OB93-72		OB93-90		OB93-96
wt. %										
SiO ₂	0.18	0.04	0.11	0.04	0.19	0.02	0.00	0.02	0.00	0.11
TiO ₂	19.12	0.28	53.38	0.20	0.54	0.13	0.11	0.02	0.06	0.03
Al ₂ O ₃	7.86	52.50	0.70	56.43	60.83	63.47	61.58	64.96	65.09	63.73
Cr ₂ O ₃	4.33	7.23	0.60	4.86	0.72	2.75	2.63	1.22	1.29	0.96
FeO*	60.05	31.54	38.29	28.22	22.32	16.70	16.86	16.71	16.69	16.19
MnO	0.43	0.22	0.30	0.19	0.09	0.10	0.13	0.07	0.09	0.09
MgO	6.19	8.94	7.44	10.97	15.60	18.10	17.70	17.04	17.25	18.93
CaO	0.11	0.02	0.06	0.02	0.03	0.00	0.00	0.01	0.10	0.02
NiO	0.26	0.28	0.21	0.32	0.24	0.28	0.28	0.30	0.28	0.31
Total	98.53	101.04	101.09	101.26	100.56	101.56	99.01	100.35	100.85	100.37
FeO	40.39	26.87	34.29	24.37	18.42	14.69	14.38	16.02	15.79	13.08
Fe ₂ O ₃	21.85	5.19	4.45	4.28	4.34	2.24	2.75	0.77	1.00	3.46
New Total	100.72	101.56	101.53	101.68	100.99	101.78	99.28	100.42	100.95	100.72
<i>Cations/3 or 4 Oxygens</i>										
Si	0.007	0.001	0.003	0.001	0.005	0.001	0.000	0.000	0.000	0.003
Ti	0.531	0.006	0.956	0.004	0.011	0.003	0.002	0.000	0.001	0.001
Al	0.342	1.743	0.020	1.818	1.889	1.906	1.901	1.963	1.957	1.954
Cr	0.126	0.161	0.011	0.105	0.015	0.055	0.054	0.025	0.026	0.033
Fe	1.855	0.743	0.762	0.645	0.492	0.356	0.369	0.358	0.356	0.347
Mn	0.013	0.005	0.006	0.004	0.002	0.002	0.003	0.001	0.002	0.002
Mg	0.341	0.375	0.264	0.447	0.613	0.688	0.691	0.651	0.656	0.723
Ca	0.004	0.001	0.002	0.001	0.001	0.000	0.000	0.000	0.003	0.001
Ni	0.008	0.006	0.004	0.007	0.005	0.006	0.000	0.006	0.006	0.006
Cation Total	3.228	3.041	2.027	3.033	3.032	3.016	3.020	3.006	3.007	3.005
Oxygen Total	3.718	3.946	2.961	3.956	3.957	3.979	3.973	3.993	3.990	3.967
Fe ²⁺	1.160	0.624	0.673	0.551	0.401	0.311	0.313	0.343	0.336	0.278
Fe ³⁺	0.564	0.108	0.079	0.087	0.085	0.043	0.054	0.015	0.019	0.066
New Cat. Total	3.097	3.031	2.017	3.026	3.027	3.014	3.018	3.005	3.006	3.022
New O Total	4.022	4.003	3.004	4.002	4.001	4.001	4.001	4.000	4.000	4.001
Cr#	26.98	8.46	36.46	5.46	0.79	2.82	2.79	1.24	1.31	1.00
Mg#Fe ²⁺	22.71	37.55	28.16	44.79	60.42	68.84	68.83	65.51	66.12	72.23
No. of Cations	3	3	2	3	3	3	3	3	3	3
No. of Oxygens	4	4	3	4	4	4	4	4	4	4
Notes	Ti-mt. rim of sp core	embayed and rimmed by opx	ilm		symp.	cpx sympl.	fine cpx sympl.	coarse cpx sympl.	fine cpx sympl.	

Table 4-8. Metamorphic equilibration temperatures estimated with the two pyroxene thermometer of Wells (1977).

Sample	Mineral Combinations	Average Temperature (°C)	Temperature Range (°C)
<i>Mg# <90</i>			
OB93-54	3 Cpx & 2 Opx, both discrete	841	830-850
OB93-57	3 Cpx & 4 Opx, both discrete	902	898-907
	1 Cpx & 2 Opx, both symplectites	901	898-903
OB93-61A	2 Cpx & 2 Opx, both discrete	864	808-907
OB93-68	3 Cpx & 3 Opx, both discrete	848	844-849
	3 discrete Cpx & 1 symplectite Opx	844	
OB93-69	2 Cpx & 3 Opx, both discrete	867	851-882
	1 Cpx & 1 Opx, both symplectites	870	
<i>Mg# >90</i>			
OB93-50	2 Cpx & 1 Opx, both discrete	899	896-902
	2 discrete Cpx & 3 symplectite Opx	896	894-898
OB93-60	4 Cpx & 2 Opx, both discrete	912	903-917
	2 Cpx & 2 Opx, both symplectite	922	902-933
OB93-71	3 Cpx & 2 Opx, both discrete	888	876-905
OB93-72	1 Cpx & 2 Opx, both discrete	945	942-947
	1 Cpx & 1 Opx, both symplectites	961	961
OB93-90	3 Cpx & 2 Opx, both discrete	939	937-941
	1 Cpx & 1 Opx, both symplectites	933	
OB93-96	2 Cpx & 2 Opx, both discrete	919	915-922
	1 Cpx & 1 Opx, both symplectites	943	942-943

Equilibrium temperatures determined with the two-pyroxene thermometer of Wells (1977).

Table 4-9. Representative trace element compositions of clinopyroxenes.

Sample	OB93-50	OB93-54	OB93-57	OB93-60	OB93-61A	OB93-68
<i>ppm</i>						
La	0.26	2.00	2.41	0.41	0.30	2.38
Ce	1.13	11.82	15.90	2.22	1.36	10.01
Nd	1.96	16.28	22.54	2.88	1.87	7.97
Sm	1.32	4.73	6.56	1.51	0.87	2.47
Eu	0.47	1.82	2.84	0.65	0.44	0.84
Dy	1.54	2.14	6.32	2.08	1.68	2.82
Er	0.94	0.70	2.20	1.25	0.85	1.50
Yb	0.78	0.71	2.25	1.16	0.99	1.48
Ti	2555	5464	2804	4001	2274	n.d.
V	444	414	517	396	323	n.d.
Cr	1861	219	251	1246	434	n.d.
Sr	17	23	25	16	15	n.d.
Y	8	21	16	12	7	n.d.
Zr	9	141	120	17	8	n.d.
<i>CI Normalized</i>						
La	1.12	8.51	10.28	1.75	1.29	10.15
Ce	1.87	19.60	26.36	3.68	2.26	16.60
Nd	4.34	35.98	49.83	6.36	4.13	17.62
Sm	8.94	32.17	44.61	10.25	5.93	16.80
Eu	8.45	32.47	50.66	11.67	7.80	14.93
Dy	6.33	8.81	26.04	8.58	6.93	11.63
Er	5.93	4.38	13.83	7.88	5.36	9.43
Yb	4.79	4.38	13.87	7.15	6.11	9.11
La/Yb _N	0.23	1.94	0.74	0.24	0.21	1.11
La/Sm _N	0.12	0.26	0.23	0.17	0.22	0.60
symp.						

symp.: symplectite texture; n.d.: not determined

Table 4-9. Trace element compositions of clinopyroxenes continued.

Sample	OB93-69		OB93-71		OB93-72		OB93-90		OB93-96	
ppm										
La	0.79	0.33	1.16	1.11	0.76	0.31	0.52	0.40	0.25	0.25
Ce	3.57	1.55	3.63	3.01	2.05	1.42	2.39	1.63	1.22	1.22
Nd	4.87	2.78	3.06	2.18	1.31	1.48	2.46	2.10	1.49	1.49
Sm	1.86	1.82	1.24	0.83	0.62	0.74	1.47	1.33	1.33	1.33
Eu	0.74	0.76	0.57	0.47	0.92	0.32	0.67	0.40	0.49	0.49
Dy	3.03	2.72	1.83	1.13	0.92	1.13	2.56	1.44	1.55	1.55
Er	2.04	1.63	0.95	0.66	0.47	0.63	1.52	0.75	0.80	0.80
Yb	1.67	1.67	1.00	0.66	0.44	0.60	1.27	0.77	0.88	0.88
Ti	n.d.	7213	n.d.	n.d.	n.d.	2692	2758	2658	2409	2409
V	n.d.	546	n.d.	n.d.	n.d.	230	282	235	466	466
Cr	n.d.	2536	n.d.	n.d.	n.d.	1537	1794	1640	2036	2036
Sr	n.d.	17	n.d.	n.d.	n.d.	14	14	15	20	20
Y	n.d.	19	n.d.	n.d.	n.d.	8	10	8	8	8
Zr	n.d.	25	n.d.	n.d.	n.d.	16	17	14	10	10
<i>CI Normalized</i>										
La	3.37	1.40	4.92	4.73	3.24	1.33	2.22	1.70	1.07	1.07
Ce	5.92	2.57	6.01	4.99	3.40	2.35	3.96	2.70	2.01	2.01
Nd	10.76	6.15	6.77	4.81	2.89	3.27	5.45	4.63	3.30	3.30
Sm	12.66	12.37	8.44	5.65	4.23	5.00	9.98	9.04	9.02	9.02
Eu	13.26	13.57	10.15	8.33	16.38	5.71	11.92	7.06	8.68	8.68
Dy	12.48	11.20	7.52	4.64	3.80	4.67	10.54	5.93	6.38	6.38
Er	12.84	10.27	6.00	4.16	2.94	3.99	9.55	4.70	5.05	5.05
Yb	10.25	10.30	6.16	4.09	2.73	3.67	7.79	4.75	5.39	5.39
La/Yb _N	0.33	0.14	0.80	1.16	1.19	0.36	0.29	0.36	0.20	0.20
La/Sm _N	0.27	0.11	0.58	0.84	0.77	0.27	0.22	0.19	0.12	0.12
		symp.		symp.		symp.				

symp.: symplectite texture; n.d.: not determined

Table 4-10. Sr and Nd isotopic compositions.

Sample	$^{87}\text{Sr}/^{86}\text{Sr}$	[Sr] ppm	$^{143}\text{Nd}/^{144}\text{Nd}$	ϵ_{Nd}	[Nd] ppm	[Sm] ppm	$^{147}\text{Sm}/^{144}\text{Nd}$
OB93-50	0.704656	80	0.512791	2.98	0.91	0.41	0.2735
OB93-50 rep	0.704686	80	n.d.	n.d.	n.d.	n.d.	n.d.
OB93-54*	0.705056	185	0.512745	2.09	11.28	3.71	0.1983
OB93-54	0.704937	187	0.512749	2.17	8.49	n.d.	n.d.
OB93-54rep	n.d.	n.d.	0.512749	2.17	8.43	2.82	0.2016
OB93-57†	0.704709	81	n.d.	n.d.	n.d.	0.40	n.d.
OB93-60	0.704739	108	0.512732	1.83	1.34	0.54	0.2451
OB93-61A	0.706006	184	n.d.	n.d.	3.66	1.50	n.d.
OB93-61A rep	0.706048	185	n.d.	n.d.	n.d.	n.d.	n.d.
OB93-68	0.705201	169	0.512754	2.26	1.36	0.55	0.2456
OB93-68†	0.705137	84	n.d.	n.d.	n.d.	n.d.	n.d.
OB93-69	0.704788	173	0.512755	2.28	1.72	0.66	0.2318
OB93-71*	0.704617	56	0.512839	3.92	1.39	0.59	0.2535
OB93-71	0.704504	71	0.512838	3.90	1.31	0.49	0.2241
OB93-71†	0.704453	53	n.d.	n.d.	n.d.	n.d.	n.d.
OB93-72*	0.704829	196	0.512734	1.87	1.06	0.45	0.2467
OB93-72	0.704604	154	n.d.	n.d.	n.d.	n.d.	n.d.
OB93-90	0.704541	107	0.512731	1.81	1.03	0.41	0.2474
OB93-96	0.704658	100	0.512796	3.08	0.88	0.39	0.2680
OB93-59 dike†	0.706380	889	0.512419	-4.27	23.04	2.42	0.0634

*no leach

†strong leach

n.d.: not determined

Counting uncertainties in isotopic compositions (2σ): Sr: <32 ppm, except OB93-54* = 186 ppm, and Nd: <12 ppm except OB93-60 = 56 ppm and OB93-90 = 36 ppm.

Table 4-11. Re-Os isotopic compositions.

Sample	$^{187}\text{Os}/^{188}\text{Os}$	$\pm 2\sigma$	[Os] ppt	$^{187}\text{Os}/^{188}\text{Os}$ @24.5Ma	maximum eruption age correction (%)	[Re] ppt	$^{187}\text{Re}/^{188}\text{Os}$	Model Age (Ma)
OB93-50	0.1300	0.0002	282	0.1294	0.46	88	1.5053	138
OB93-54	0.1787	0.0004	113	0.1611	9.85	934	§45.1078	72
OB93-54 rep	0.1792	0.0003	108	0.1616	9.82	1165	-	-
OB93-57	0.1299	0.0001	360	0.1298	0.08	20	0.2680	-1100
OB93-60	0.1304	0.0002	291	0.1300	0.31	†65	1.0775	261
OB93-61A	0.1292	0.0002	197	0.1286	0.46	58	1.4201	100
OB93-68	0.1340	0.0002	428	0.1321	1.42	425	4.7925	92
OB93-69	0.1293	0.0002	910	0.1291	0.15	80	0.4240	3900
OB93-71	0.1286	0.0003	1731	0.1285	0.08	128	0.3566	-1600
OB93-71 rep*	0.1289	0.0003	1583	0.1287	0.16	127	0.3869	-8700
OB93-72	0.1344	0.0005	167	0.1342	0.15	21	0.6069	2000
OB93-72 run 2	0.1338	0.0003	170	0.1336	0.15	-	-	-
OB93-90	0.1298	0.0002	160	0.1287	0.85	†94	2.8339	56
OB93-96	0.1303	0.0005	130	0.1301	0.15	†11	0.4082	13900

* replicated from new powder

rep: replicated from same powder

run 2: rerun of same bead

†not blank corrected for Re

§calculated from average

Model age calculated using $\lambda = 1.59 \cdot 10^{-11} \text{ a}^{-1}$, $^{187}\text{Os}/^{188}\text{Os}$ mantle = 0.12757,and $^{187}\text{Re}/^{188}\text{Os}$ mantle = 0.39716

(Lindner, 1986; Walker & Morgan, 1989).

4.9 Figures

Figure 4-1. Examples of granulite textures and mineralogy. Crossed polarized light photomicrograph of the typical granulite texture (Sample OB93-50). Note the relatively equigranular texture. Plagioclase (low birefringent and twinned grains), orthopyroxene (low birefringence with no twinning), and clinopyroxene (high order colors) are present. Pyroxene exsolution is present in both orthopyroxene and clinopyroxene. The field of view is 10 mm across.

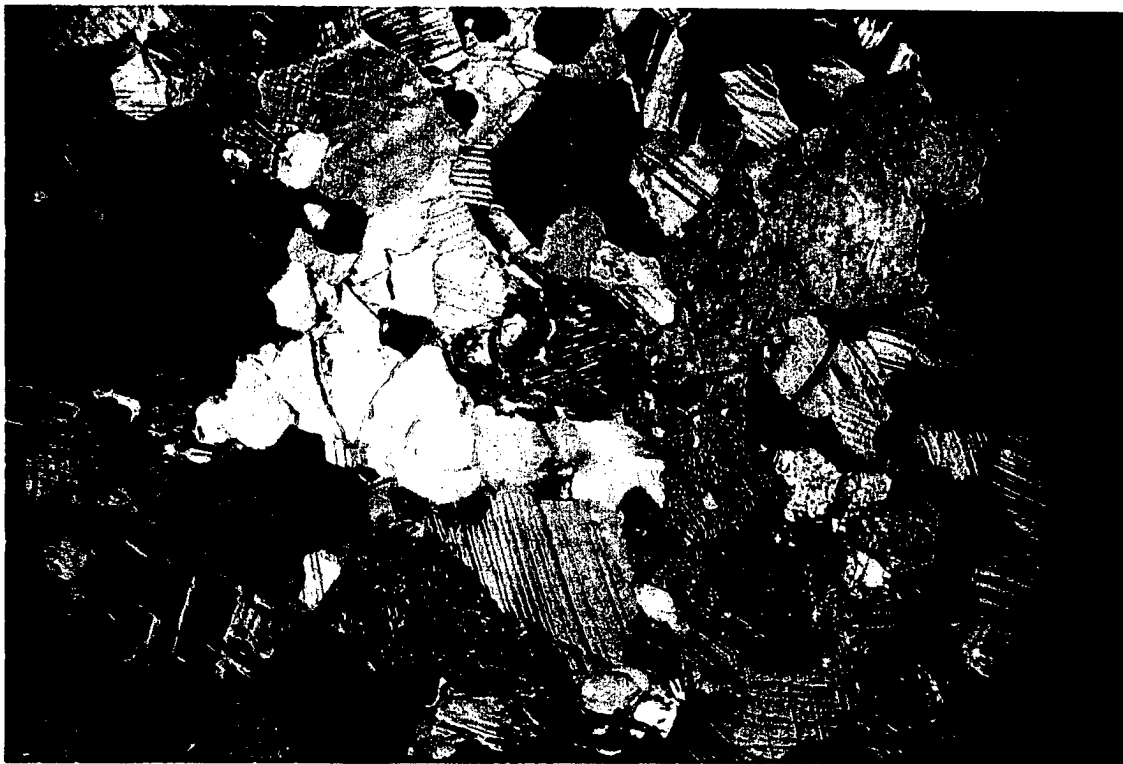


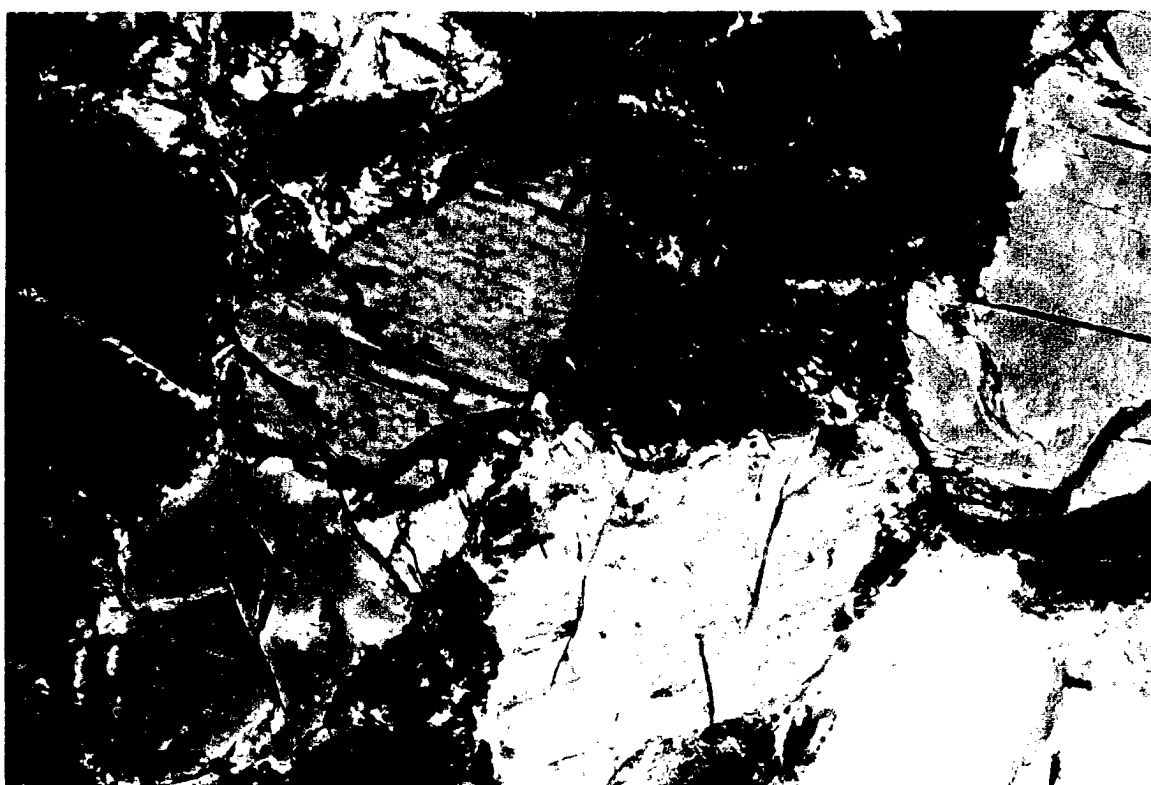
Figure 4-1 continued. Close up crossed polarized light photomicrograph of the typical granulite texture (Sample OB93-90). Note the equigranular texture and 120° grain boundaries among plagioclase and orthopyroxene grains (central). A spinel (black) is visible in the top central part of the frame. The field of view is 2.5 mm across.



Figure 4-1 continued. Crossed polarized light photomicrograph of an example of the coarse symplectite texture (Sample OB93-57). Again, note the equigranular texture with 120° grain boundaries. Symplectite spinel (black, lower left) is within an orthopyroxene grain. The field of view is 2.5 mm across.



Figure 4-1 continued. Crossed polarized light photomicrograph of an example of the fine symplectite texture (Sample OB93-68). Symplectite spinel (black, upper central) is within an orthopyroxene grain. Plagioclase (lower central) and clinopyroxene are present. The field of view is 1.25 mm across.



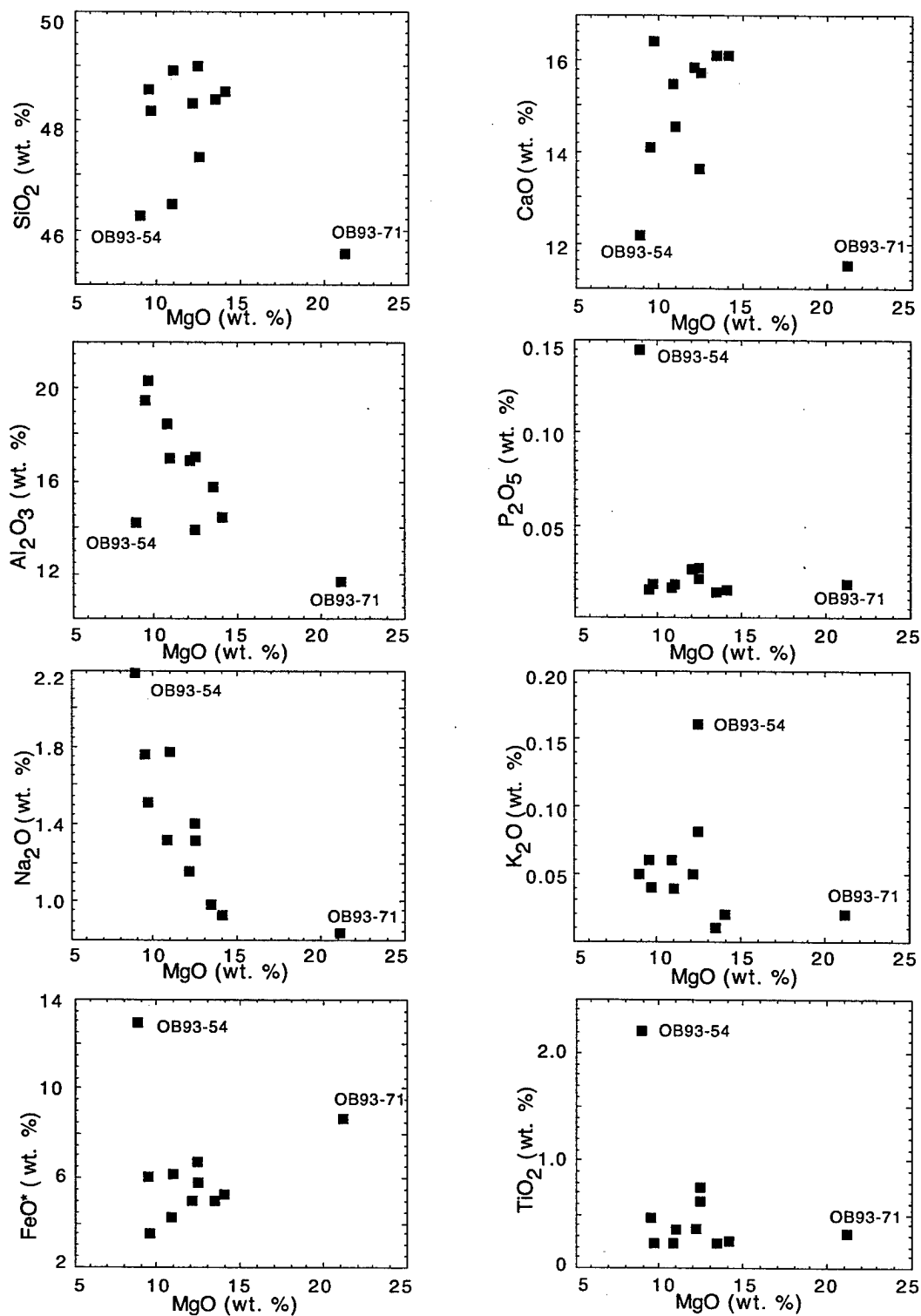


Figure 4-2. Major element bulk-rock variation diagrams. Granulite compositions are shown as filled squares. Each symbol represents a single sample.

Figure 4-3. Bulk rock major element compositions (Mg# v. SiO₂ wt.%) of Courbet Peninsula granulites compared to those from the Southeastern Province. The granulites of this study (filled squares) show a similar variation as those from the Southeastern Province (Grègoire et al., 1994; Grègoire, 1994; Grègoire et al., 1998). Granulite types from the Southeastern Province are two pyroxene (open squares), ilmenite-bearing (gray squares), ilmenite and spinel-bearing (x), sapphirine-bearing (open circles), sapphirine and garnet-bearing (closed circles) and garnet-bearing (+).

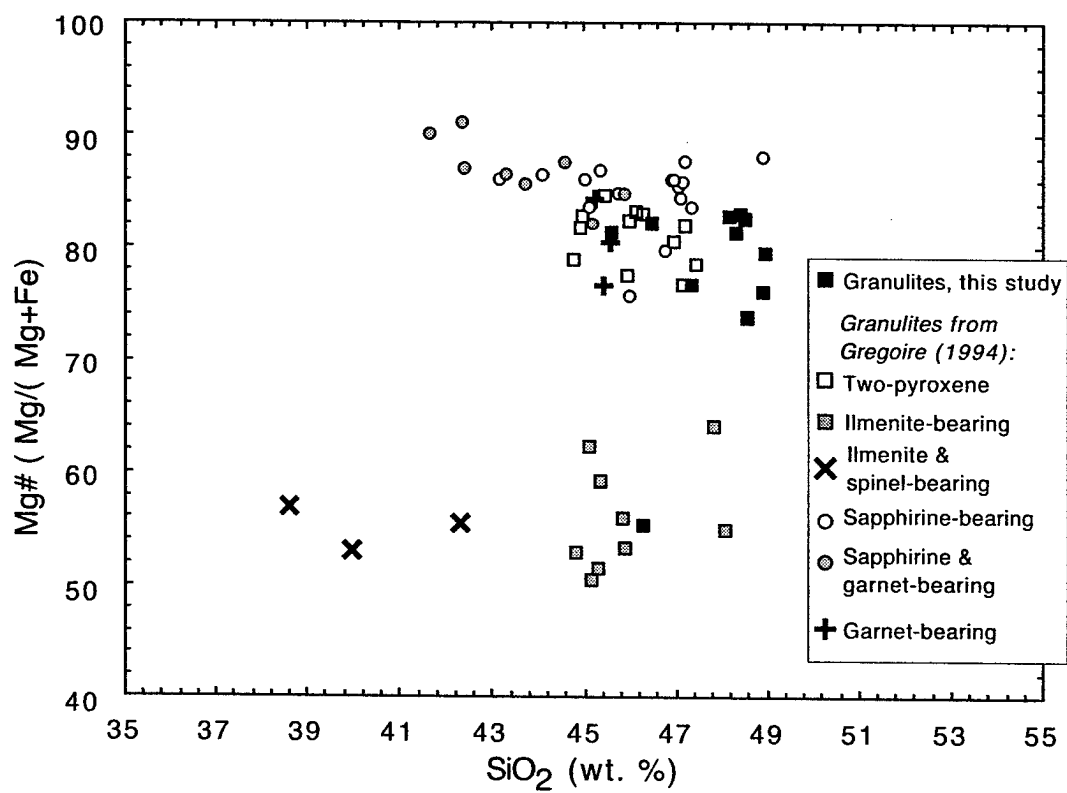


Figure 4-4. Whole rock geochemistry of Kerguelen granulites classified according to the IUGS system (Streckheisen, 1973). Calculated mineral modal abundances (gray circles) and CIPW normative minerals (filled squares) are shown.

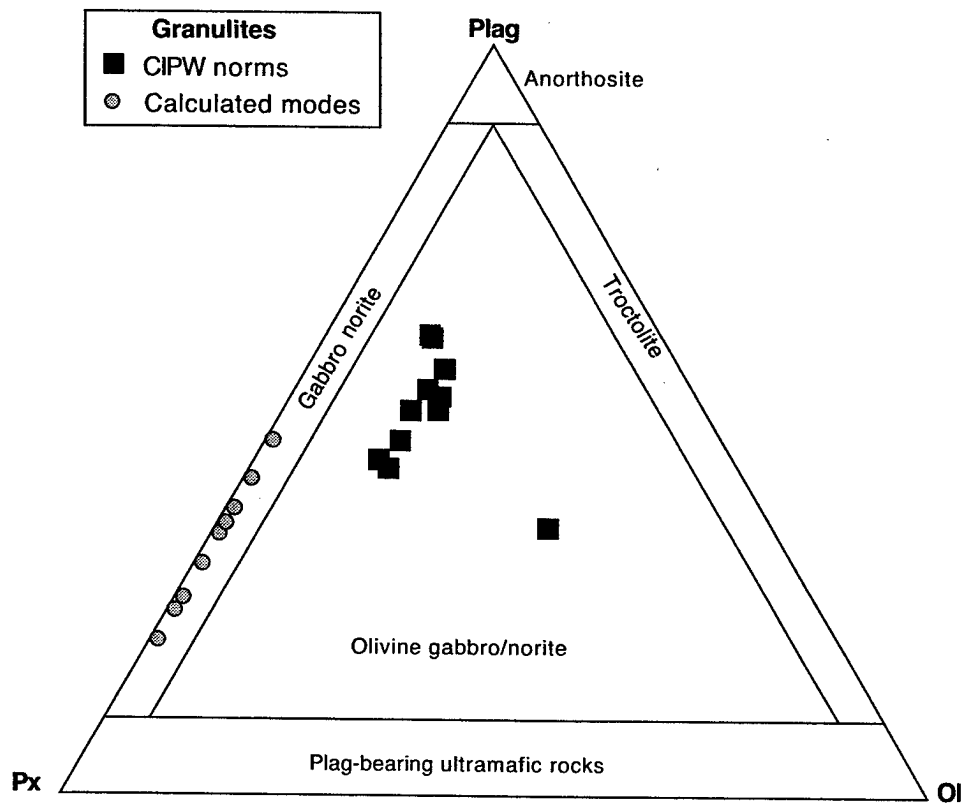
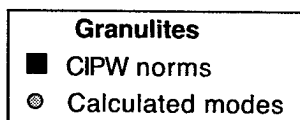
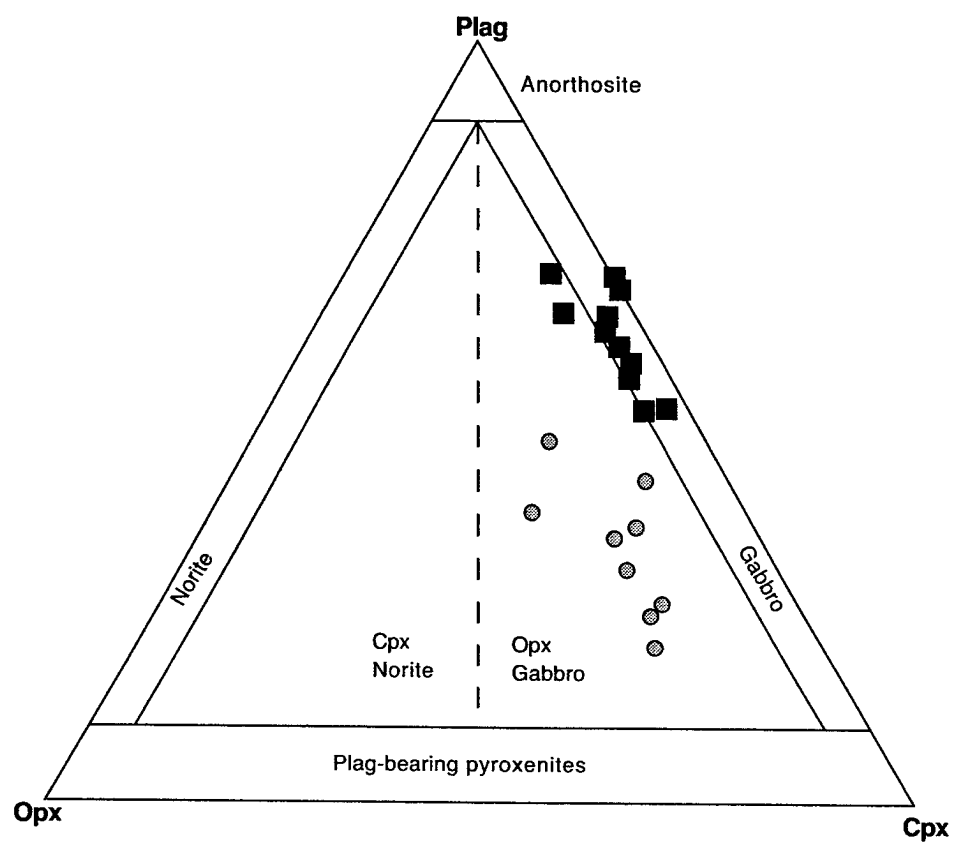


Figure 4–5. Granulite clinopyroxene major element compositions compared to those from other oceanic cumulate rocks. Granulite clinopyroxenes (filled diamonds) are more aluminous and sodic than other oceanic gabbros (panels A and D), but have similar TiO_2 and Cr_2O_3 for a given Mg# (panels B and C). Fields for oceanic cumulates and abyssal peridotite clinopyroxene compositions from a review paper by Elthon et al. (1992).

Clinopyroxenes from Granulites

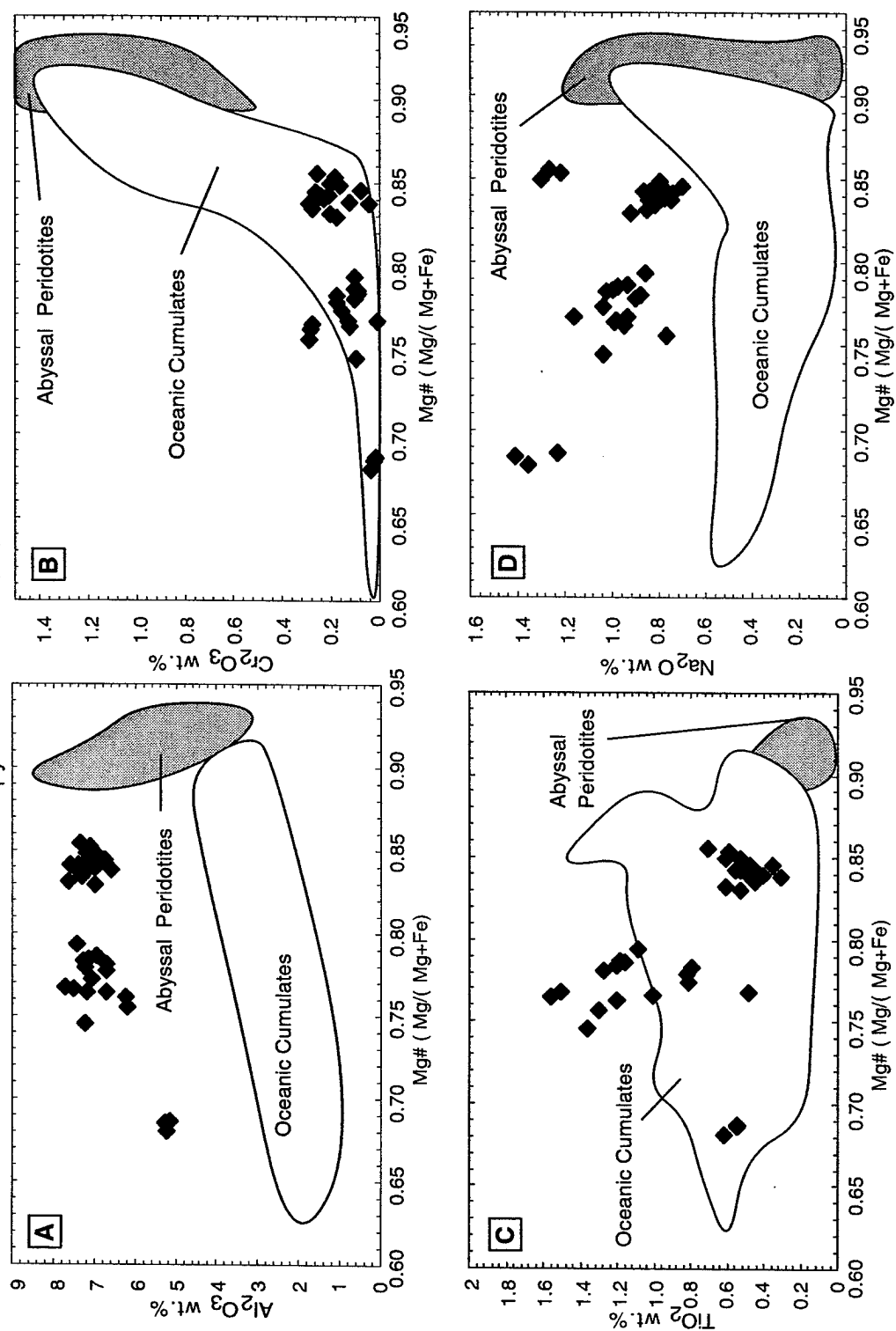
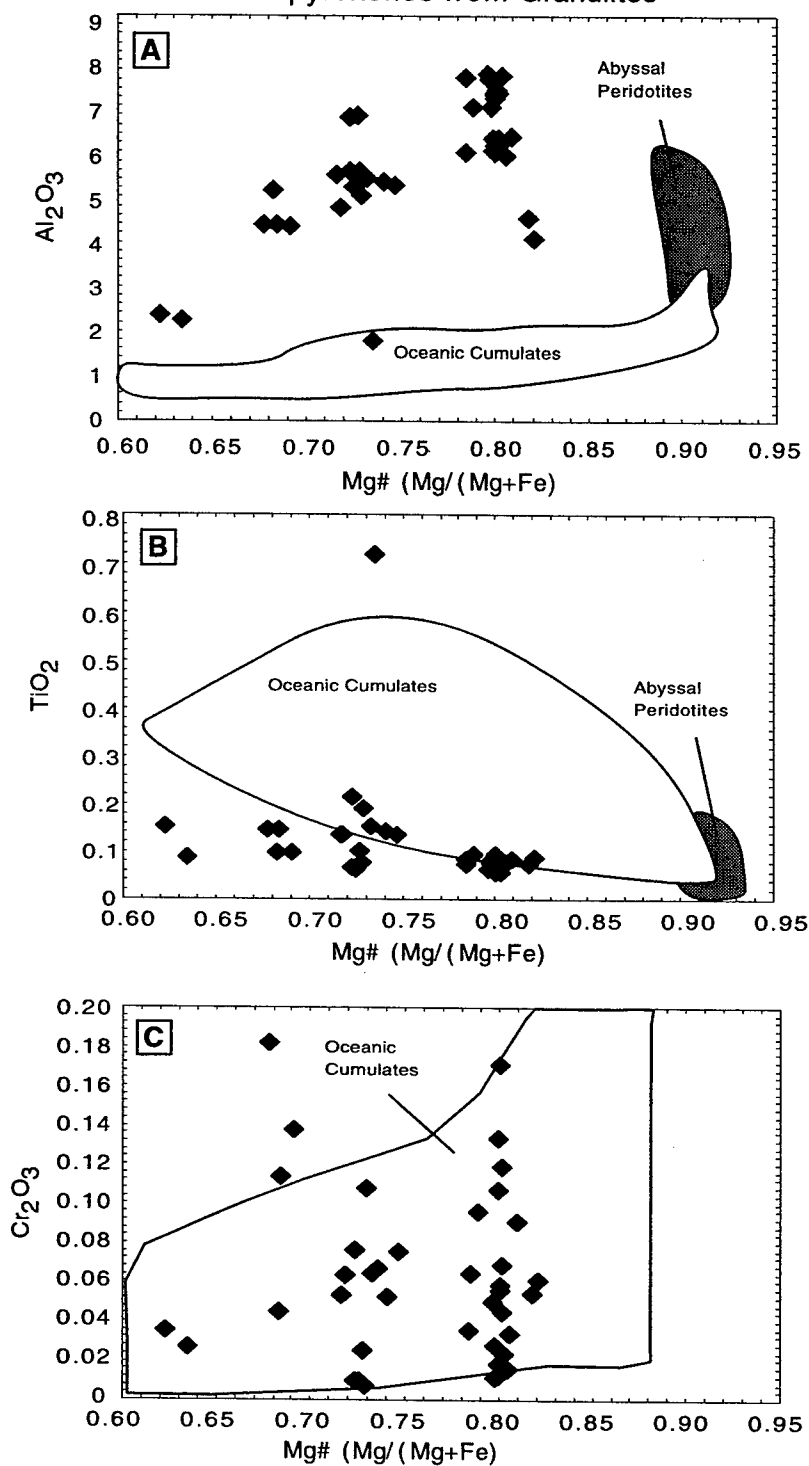


Figure 4-6. Granulite orthopyroxene major element compositions compared to orthopyroxene compositions from other oceanic cumulate rocks. Granulite orthopyroxenes (filled diamonds) are more aluminous than other oceanic cumulate rocks (panel A), but have similar TiO_2 and Cr_2O_3 contents for a given Mg# (panels B and C). Oceanic cumulates and abyssal peridotite fields from a review by Elthon et al. (1992).

Orthopyroxenes from Granulites



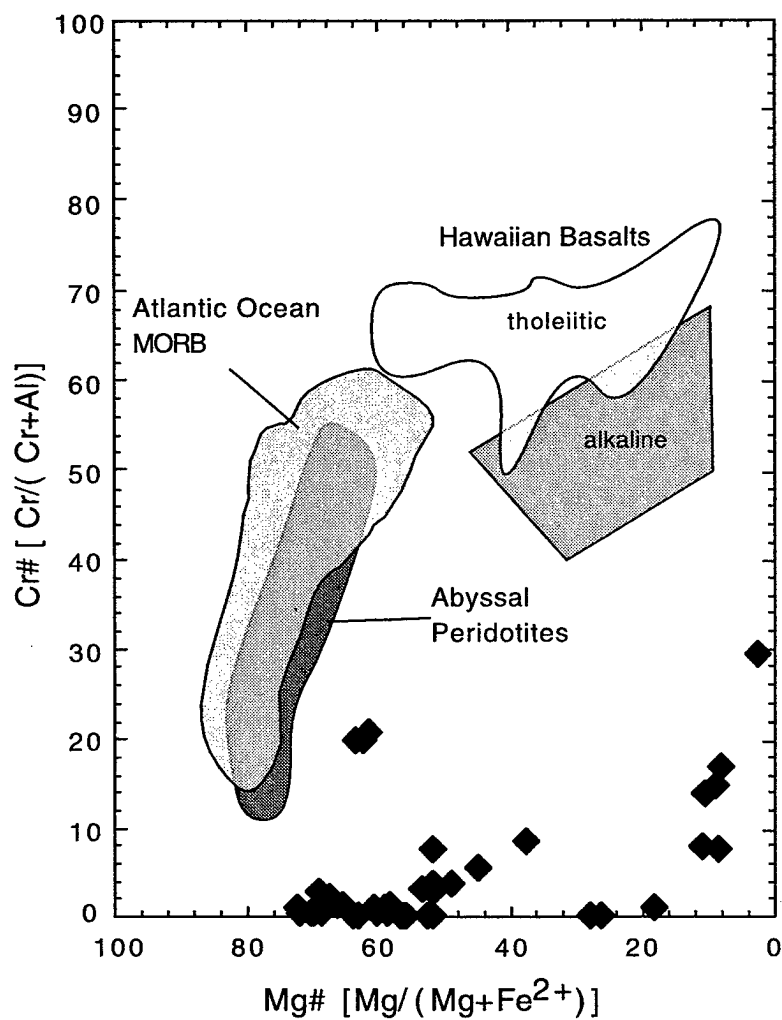


Figure 4-7. Cr# v. Mg# diagram of spinels from granulites. Granulites (filled diamonds) are significantly less chromian in comparison to other oceanic rocks, signifying their metamorphic origin. Compositional fields of spinels from abyssal peridotites (Dick and Bullen, 1984), Hawaiian basalts (Sen, 1987), and Atlantic MORB (Dick and Bullen, 1984) are shown for comparison.

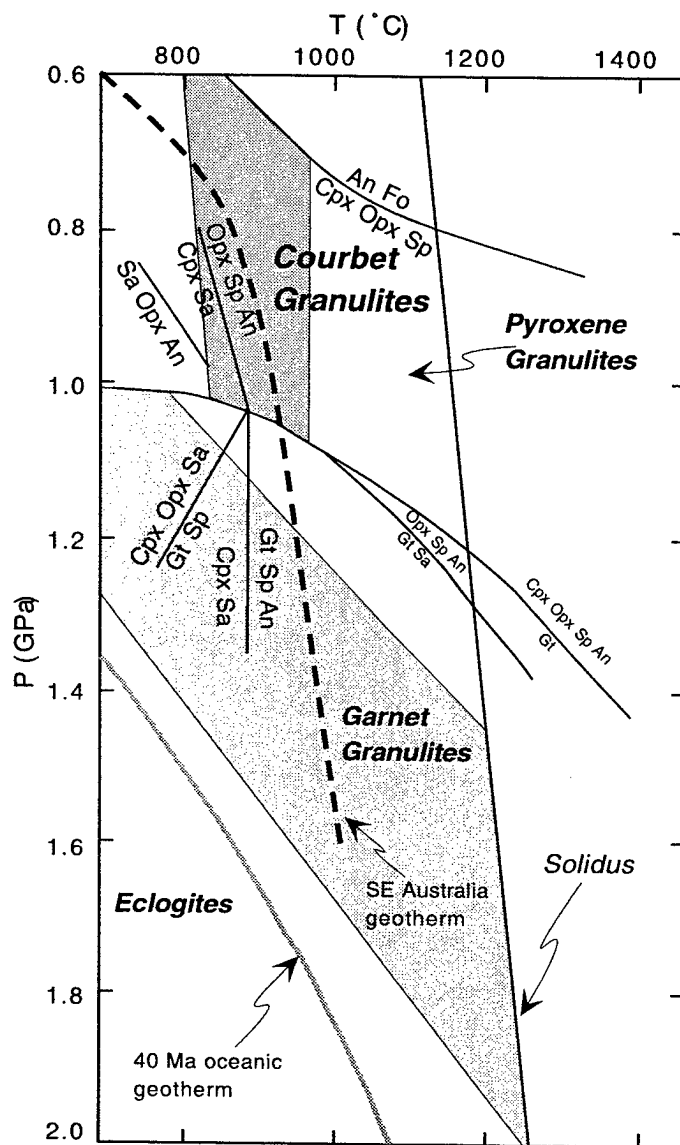


Figure 4-8. Pressure -temperature diagram for mafic composition granulite-facies metamorphism (after Ringwood, 1975; Christy, 1989). Reaction boundaries are labeled with the reactant and product mineral phases. These, and the basalt solidus define the fields for garnet granulites (light shaded field) and pyroxene granulites (labeled). The conditions of metamorphism for the xenoliths of this study are shown as the darker gray field that is labeled "Courbet Granulites". Shown for reference are an oceanic geotherm calculated for 40 Ma (solid gray line; after Fowler, 1990), and a geotherm for southeastern Australia (after Rudnick et al., 1986).

Figure 4-9. Whole rock major element compositions in comparison to oceanic gabbros and continental granulite xenoliths. Granulites from this study are shown as filled squares. Fields represent the ranges of compositions of oceanic gabbros from ODP site 735B, SWIR (dark gray; Dick et al., 1992), ODP Leg 153, MARK (medium gray; Casey, 1997), dredged gabbros from the SWIR (light gray; Meyer et al., 1989), and gabbro xenoliths from the Canary Islands (white; Schminke et al., 1998). The black arrow represents a liquid line of descent (after Grove et al., 1992). The fields for granulite xenoliths from continents are also outlined in gray. Continental data from Rudnick and Presper (1990).

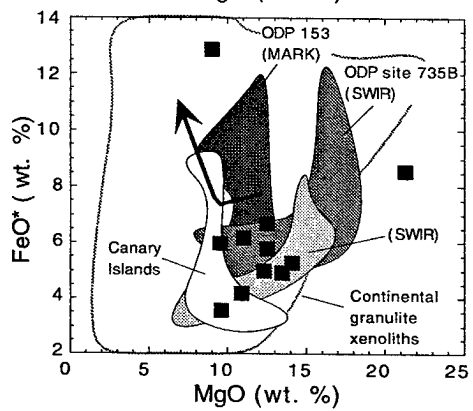
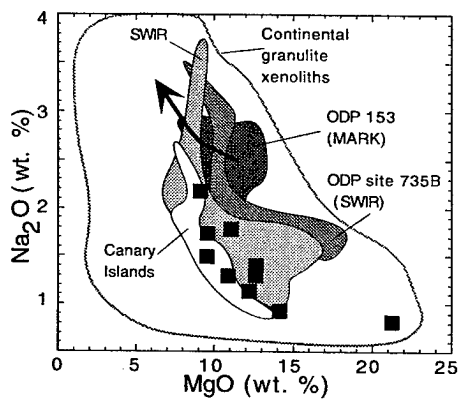
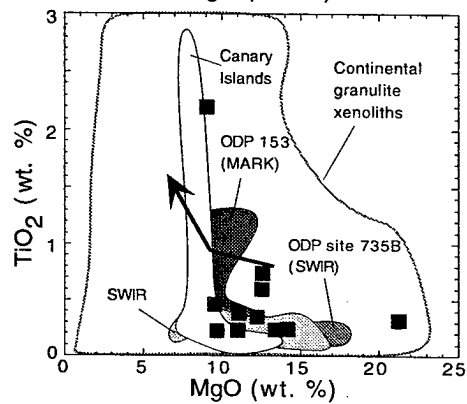
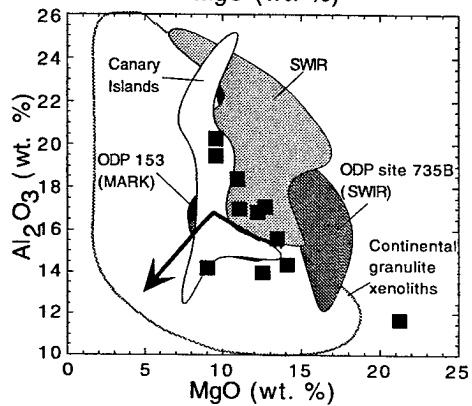
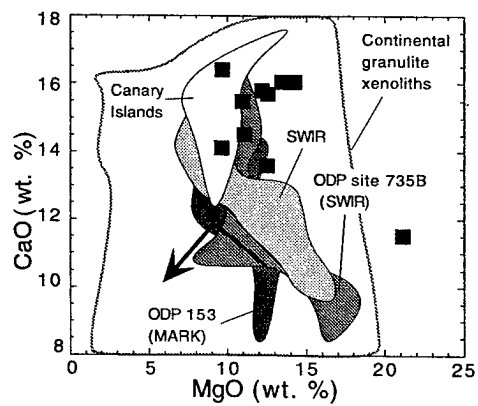
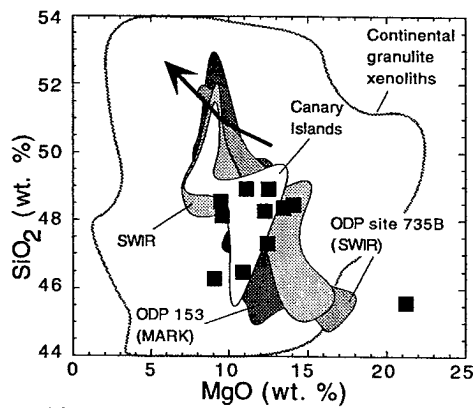


Figure 4-10. Semi-log plot of granulite whole rock trace element contents normalized to C1 chondrites. Each pattern represents the elemental abundances of a single xenolith. The upper panel shows the trace element patterns of the lower Mg# samples; the lower panel, the higher Mg# samples. Also shown for comparison is the field (gray shade) defined by whole rock trace element abundances of gabbros from the Oman ophiolite (Garrido and Kelemen, 1999). The Kerguelen granulites have similar abundances, as well as positive Sr and Eu anomalies, as the Oman ophiolites.

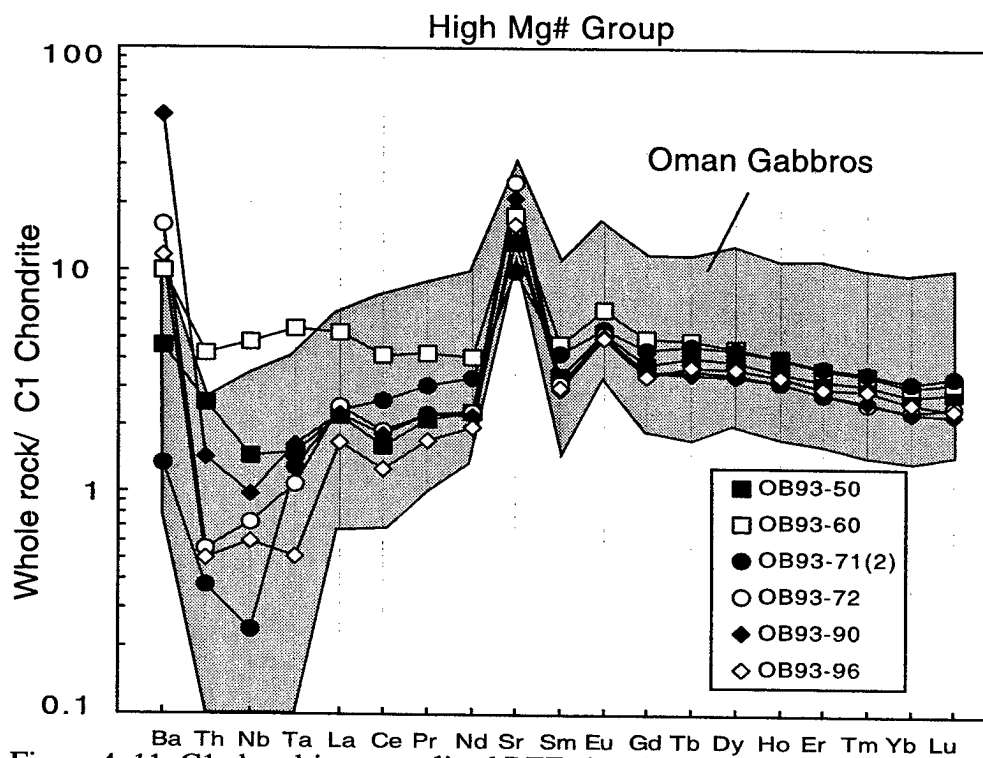
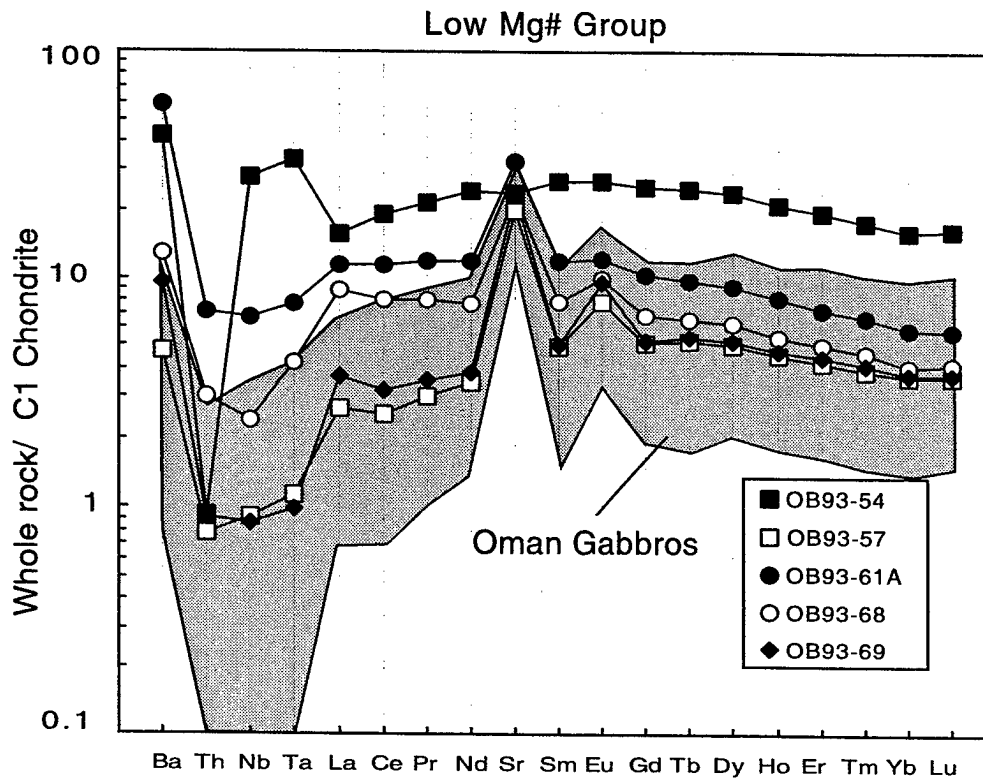
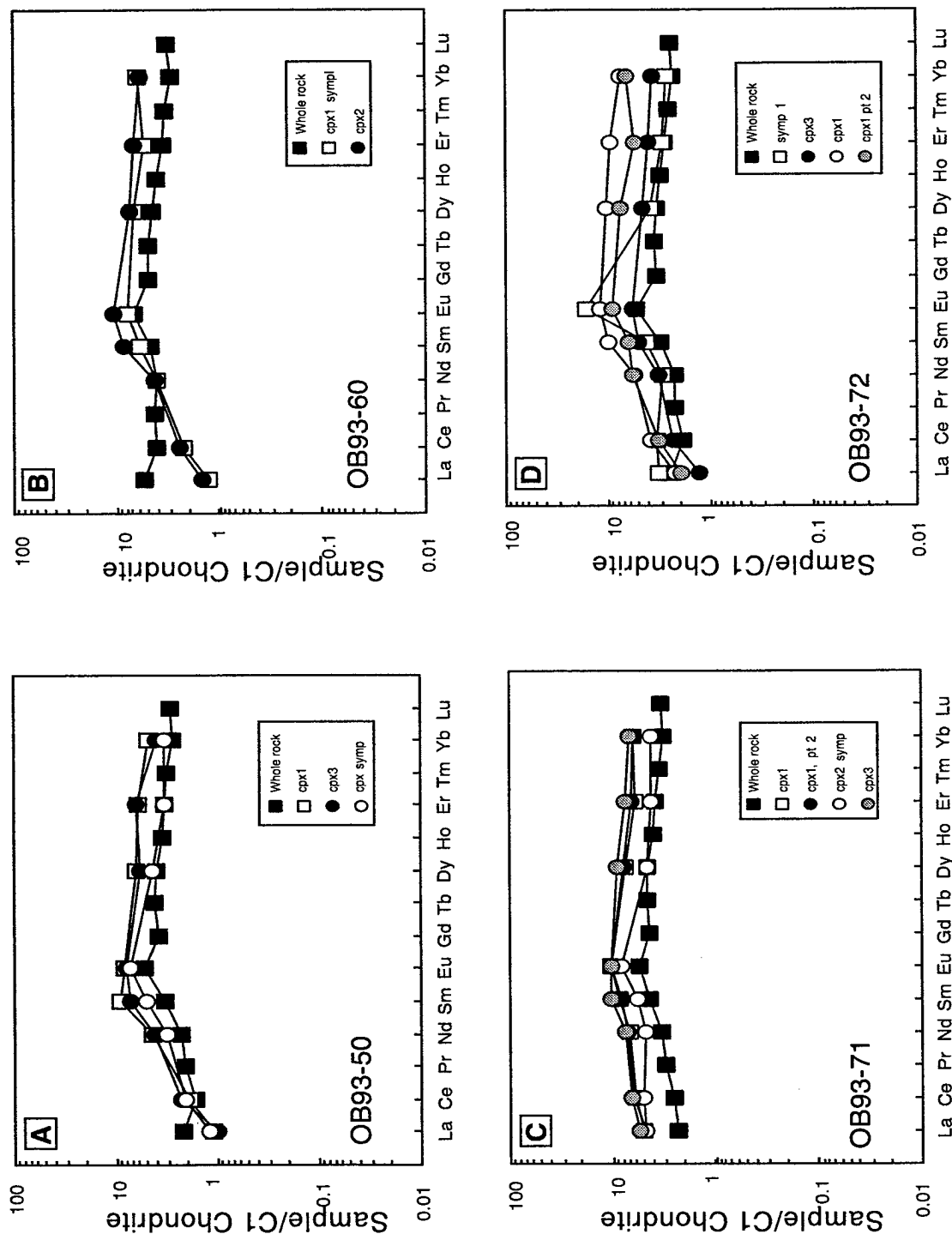
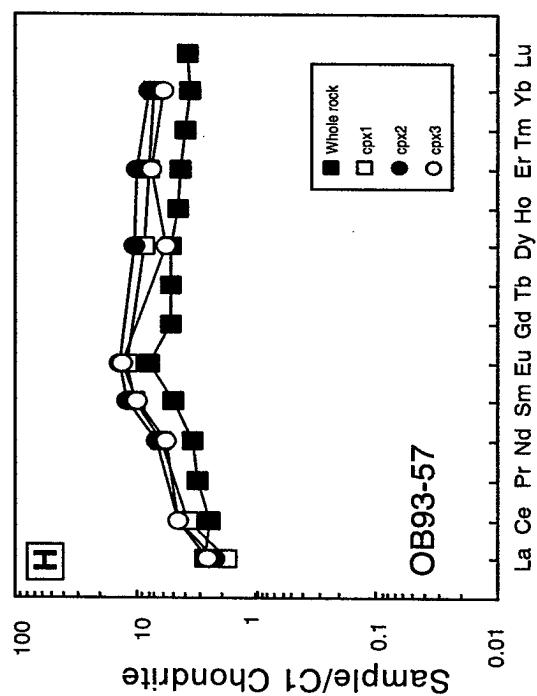
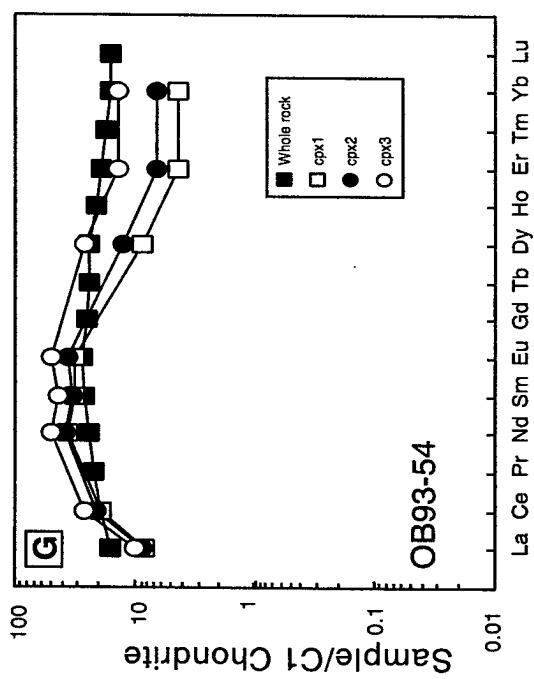
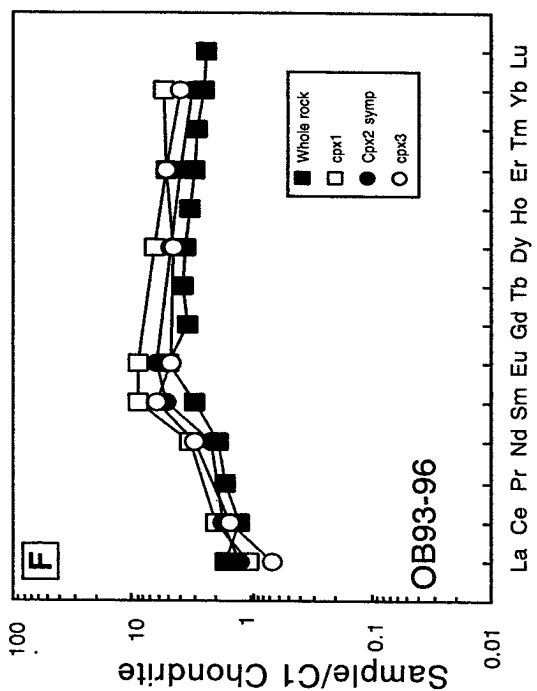
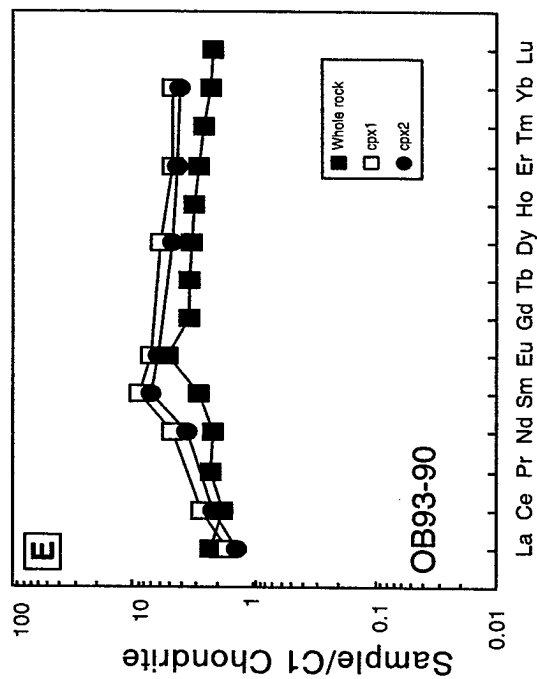


Figure 4-11. C1 chondrite normalized REE abundances of clinopyroxenes and bulk rocks. Whole rock patterns (filled squares) are shown along with clinopyroxene patterns. Each

panel represents one sample, and each clinopyroxene pattern corresponds to a single analysis. Repeat analyses are noted as "pt 1, pt 2" etc, all other analyses are from different clinopyroxene grains. Grains with symplectite textures are marked as "symp". Higher Mg# samples are shown in panels A to F, lower Mg# samples in panels G to K.





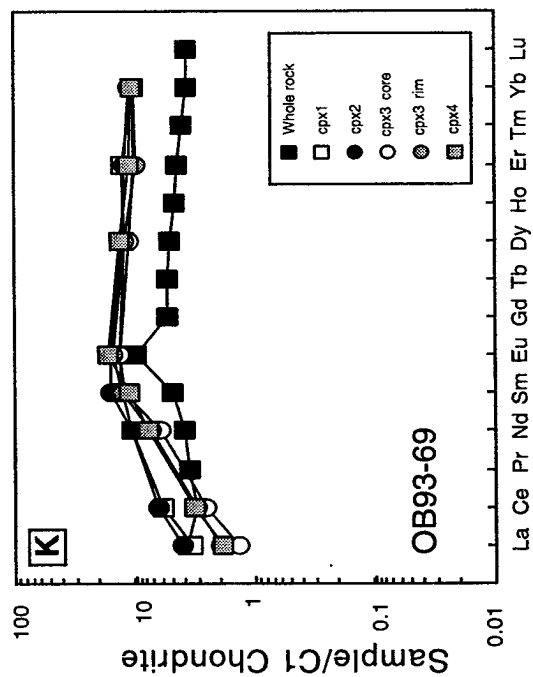
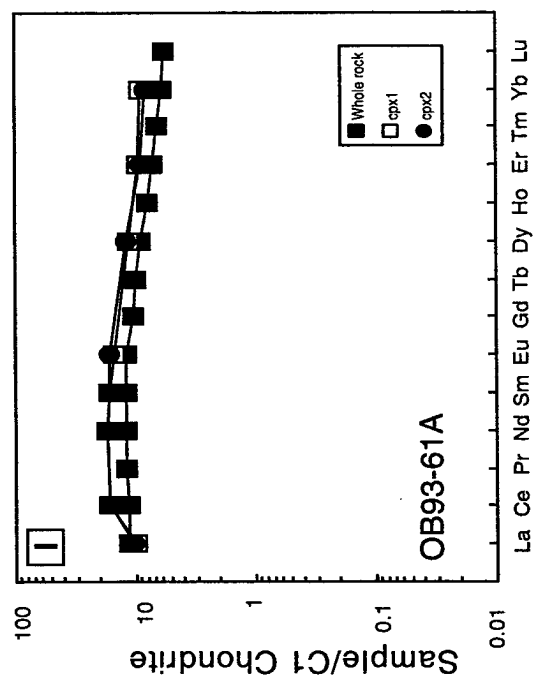
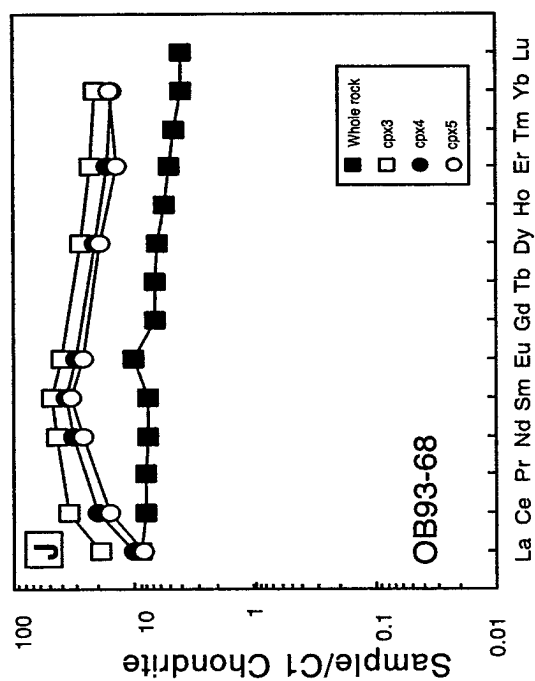
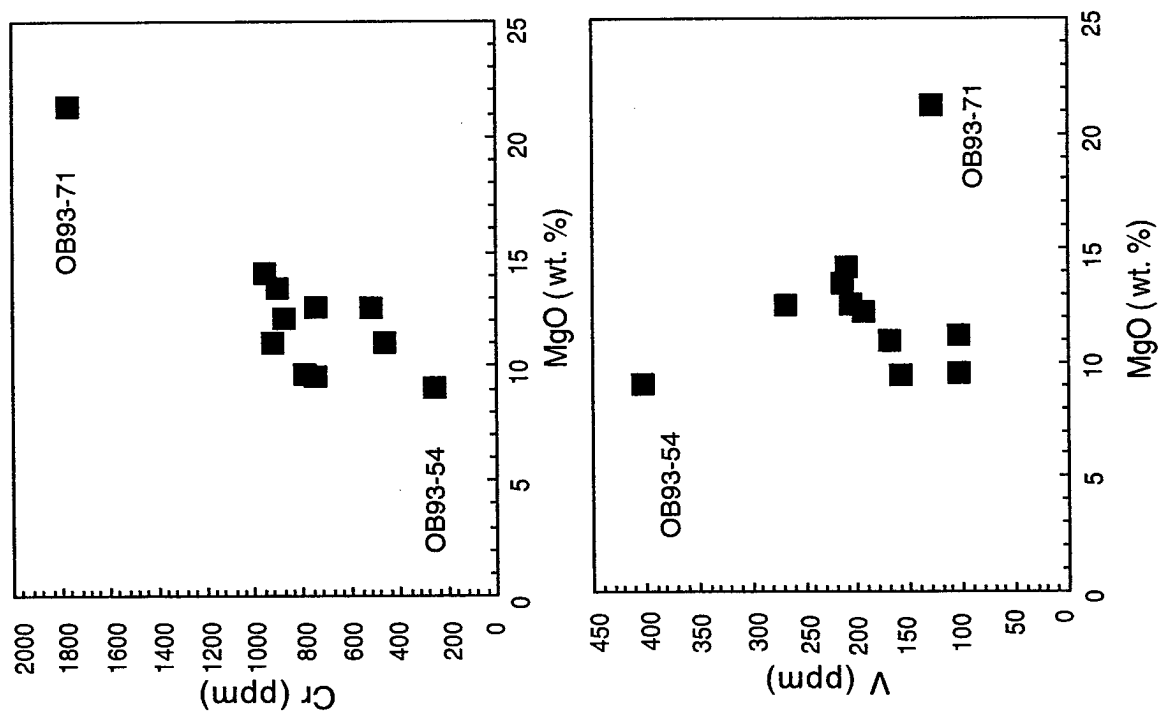
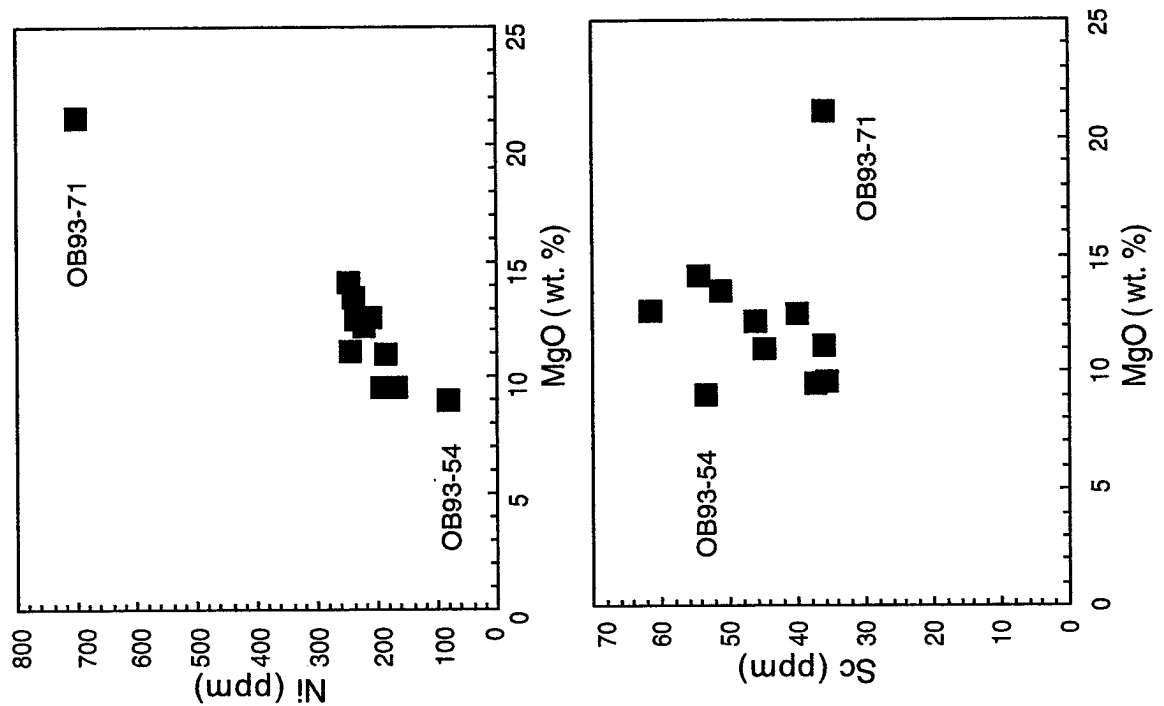


Figure 4-12. Whole rock abundances of transition metals (Cr, Ni, V, Sc) v. MgO wt.%. Granulites are shown as filled squares. Each point represents a single sample.



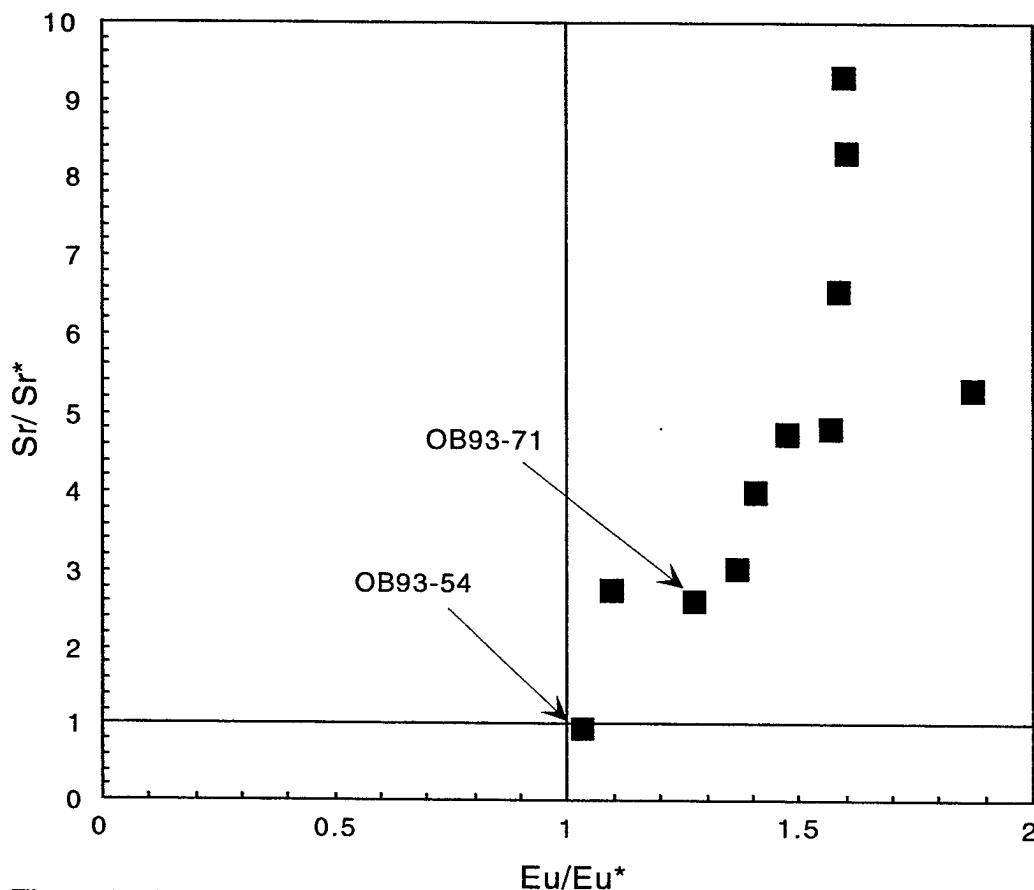


Figure 4-13. Granulite whole rock Sr/Sr^* v Eu/Eu^* . Sr^* and Eu^* are interpolated from the concentrations (in ppm) of the neighboring trace elements ($Sr^* = (Nd + Sm)/2$ and $Eu^* = (Sm + Gd)/2$). All samples, except OB93-54 have both Sr^* and $Eu^* > 1$, indicating a cumulate protolith. Sample OB93-54 is also interpreted as a cumulate, as discussed in the text.

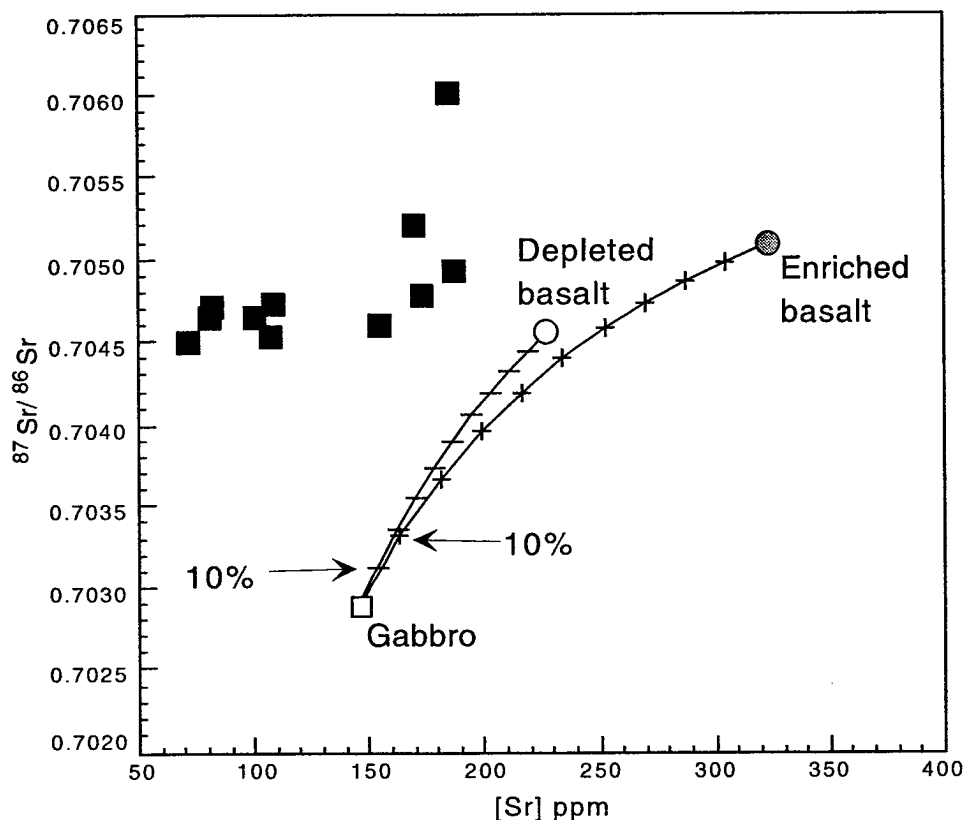


Figure 4-14. Strontium isotopic composition v. $[\text{Sr}]$ (ppm). Bulk rock compositions of granulites are compared to those of two component mixtures of a calculated SEIR basalt equilibrium gabbro and two Kerguelen Islands basalts. The possibility of an oceanic olivine gabbro accumulate (open square) impregnated with Kerguelen OIB (open and gray circles) as a protolith for the granulites (filled squares) is illustrated in this figure. Mixtures of the gabbro with a REE depleted basalt (–) and REE enriched basalt (+) are shown. Symbols mark the melt fraction in 10% increments. On the basis of this calculation, the granulites could not have originally been a mixture of lower oceanic crust and Kerguelen OIB. The basalt data is from Yang et al. (1998). The gabbro data is calculated from a SEIR basalt from Dosso et al. (1988) using gabbro modes of 13.5% olivine, 31.5% clinopyroxene and 55% plagioclase, and Sr partition coefficients from Dunn and Sen (1994) and Hart and Dunn (Hart and Dunn, 1993).

Figure 4–15. Normalized REE compositions of equilibrium melts calculated from whole rock trace element contents. Granulite equilibrium melt patterns (white field, open squares) are more similar in shape and abundance to the field defined by the Kerguelen Islands and Plateau basalts (dark gray; Storey et al., 1988; Davies et al., 1989; Gautier et al., 1990; Alibert, 1991; Salters et al., 1992; Weis et al., 1993; Yang et al., 1998) than that defined by the SEIR MORB (medium gray; Michard et al., 1986; Dosso et al., 1988).

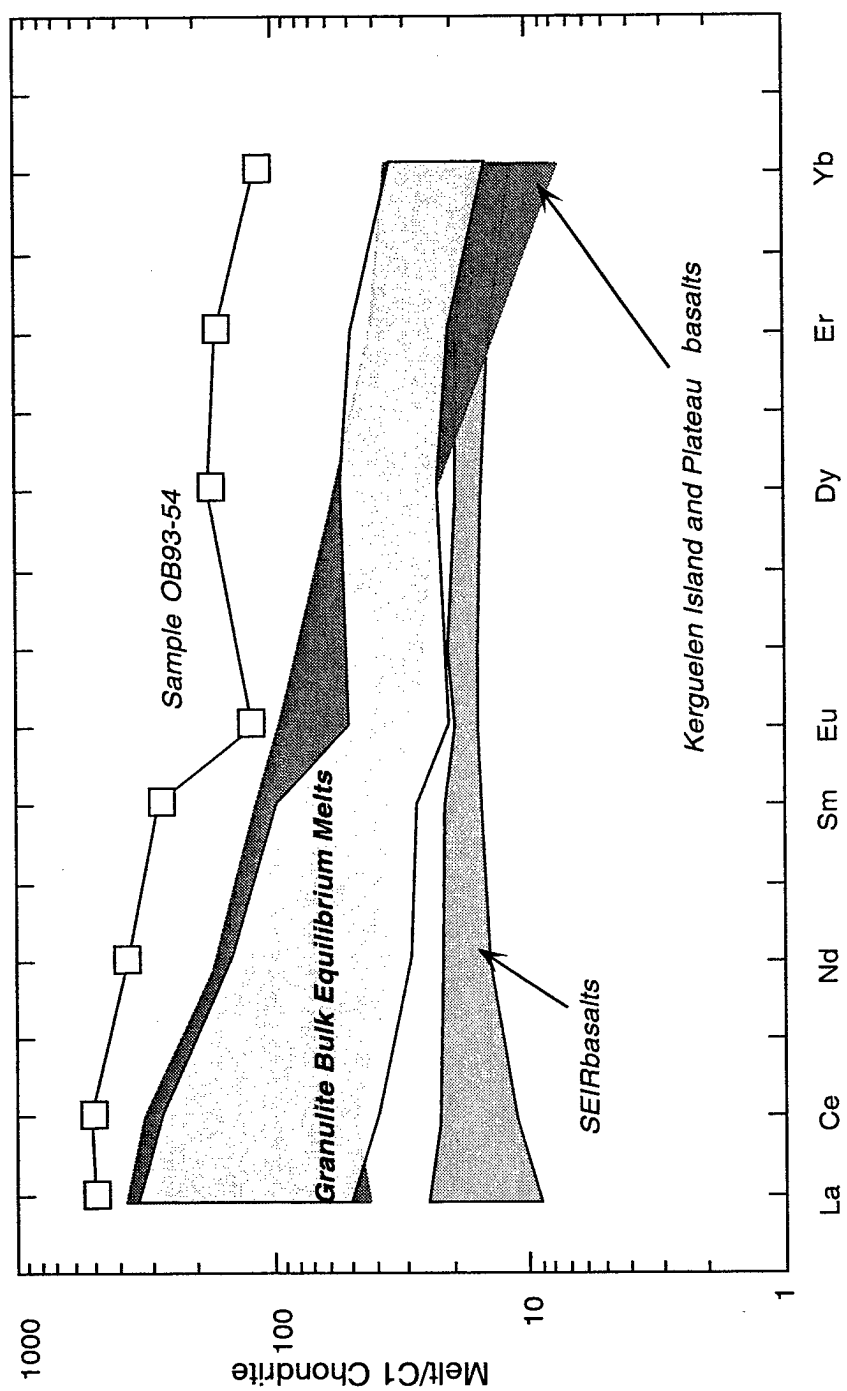


Figure 4–16. Normalized REE content of melts in equilibrium with granulite clinopyroxenes. Granulite clinopyroxene equilibrium melt patterns (white field) are similar in shape and abundance to the field defined by the Kerguelen Islands and Plateau basalts (dark gray; Storey et al., 1988; Davies et al., 1989; Gautier et al., 1990; Alibert, 1991; Salters et al., 1992; Weis et al., 1993; Yang et al., 1998). The field defined by the SEIR MORB (medium gray; Michard et al., 1986; Dosso et al., 1988) has lower REE abundances and is less REE enriched than the Kerguelen basalts and the clinopyroxene equilibrium melts. The clinopyroxene equilibrium melts lack the negative Eu anomalies that are present in the whole rock equilibrium melts and have higher overall REE abundances (Figure 4–15).

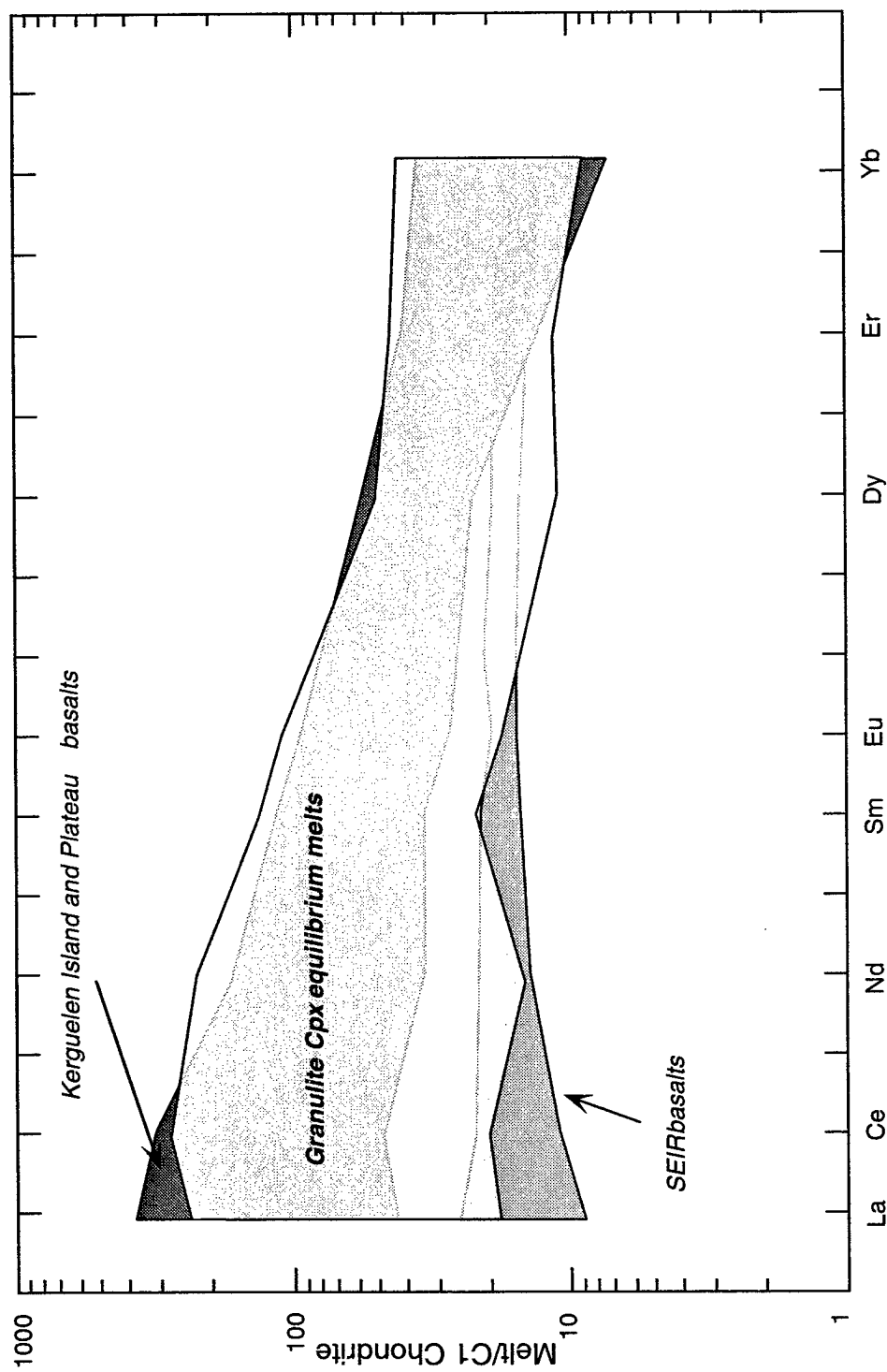


Figure 4–17. Sr v. Nd isotopic diagram of bulk rock granulite compositions. Measured isotopic compositions of granulites from this study (open squares), granulites from the Southeastern Province (Mt. Tizard: gray diamonds, Dome Rouge and Val Phonolite: white diamonds; Mattielli et al., 1996; Mattielli, 1996), clinopyroxenes from peridotites (open circles; Chapter 2), and the host dike are shown in comparison to initial isotopic compositions of basalts from the Indian Ocean ridges (MORB with no plume influence: light gray, all Indian Ocean MORB: light gray + dashed line extension) and the Kerguelen Islands (dark gray). The Sr isotopic composition of OB93-61A is shown as a broken line. Granulites lie nearly completely within the field defined by Kerguelen Islands basalts (White and Hofmann, 1982; Gautier et al., 1990; Weis et al., 1993; Weis et al., 1998; Yang et al., 1998), and none overlap the compositions of Indian Ocean MORB (Hamelin et al., 1986; Michard et al., 1986; Dosso et al., 1988). Mantle isotopic endmembers are shown for reference (filled circles; Hart, 1988).

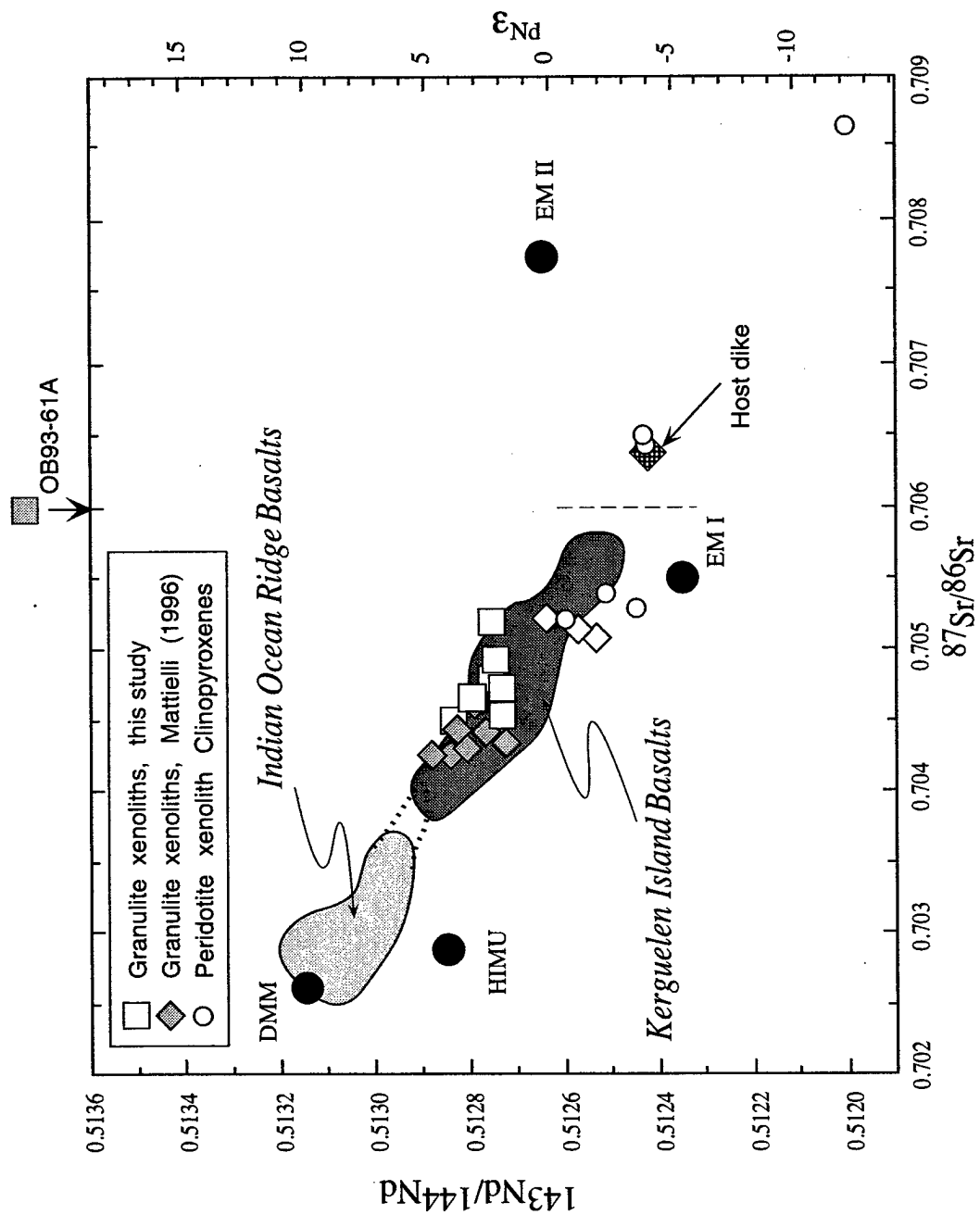


Figure 4–18. $^{187}\text{Os}/^{188}\text{Os}$ v. $[\text{Os}]$ (ppt) upper panel, and $^{187}\text{Os}/^{188}\text{Os}$ v. $^{187}\text{Re}/^{188}\text{Os}$, lower panel, of granulite whole rocks (filled squares) and Kerguelen plume related basalts (open and gray circles; Reisberg et al., 1993; Barling et al., 1997; Yang et al., 1998). Data are blank corrected, but not age corrected. Samples show the general trend of more radiogenic $^{187}\text{Os}/^{188}\text{Os}$ at lower concentrations, though the most radiogenic granulite (OB93-54) has a higher concentration than that thought to be easily susceptible to contamination. In the isochron diagram of the lower panel, a 69 Ma isochron is plotted for reference. Although there is no isochron relationship among these samples, this indicates that OB93-54 would have had a similar $^{187}\text{Os}/^{188}\text{Os}$ as the other samples at ~69 Ma.

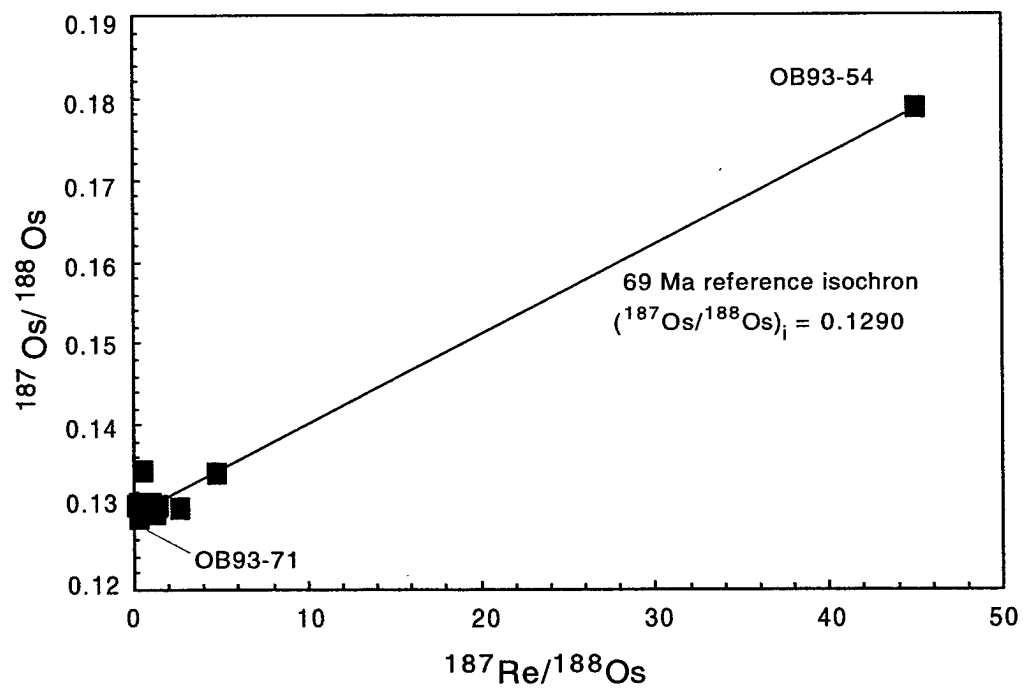
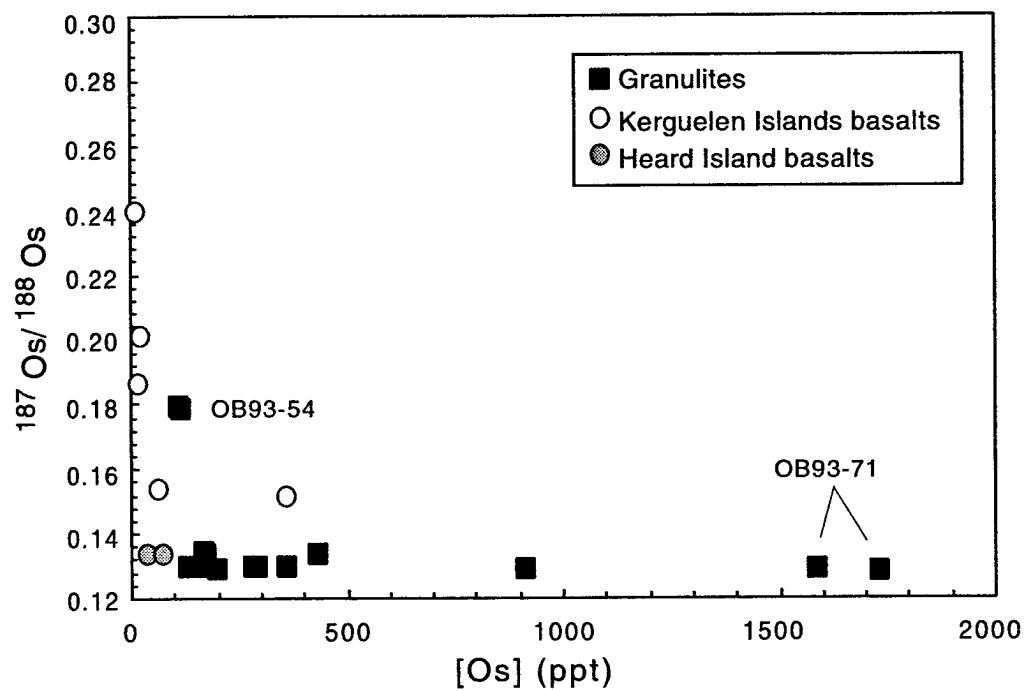


Figure 4–19. Nd v. Sr isotopic compositions of continental granulites (after Rudnick, 1992). Labeled fields are for mafic granulites, other mafic granulite xenoliths are designated as '+'. Other granulites are also shown for comparison. The white field delineates the Kerguelen granulite xenoliths. Continental samples have a large isotopic variation that encompasses the relatively limited isotopic compositions of the Kerguelen granulites.

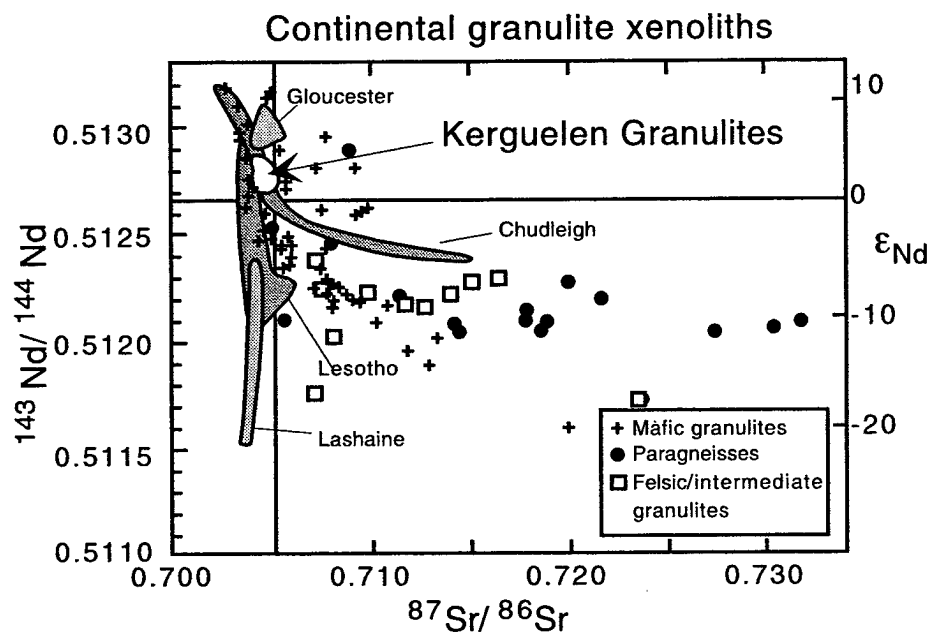


Figure 4–20. Comparison of the present day values of the $^{187}\text{Os}/^{188}\text{Os}$ and $^{187}\text{Re}/^{188}\text{Os}$ of all lower crustal xenoliths analyzed to date. The location and number of samples in each dataset are noted next to each bar. Present-day primitive mantle values are shown as a broken line in each panel. The Kerguelen granulites have a limited range of these values, especially $^{187}\text{Re}/^{188}\text{Os}$, relative to the other analyzed samples. Comparison data are from Hart et al. (1999), Lassiter et al. (1998), Pearson et al. (1995), Saal et al. (1998), and Esperança et al. (1997).

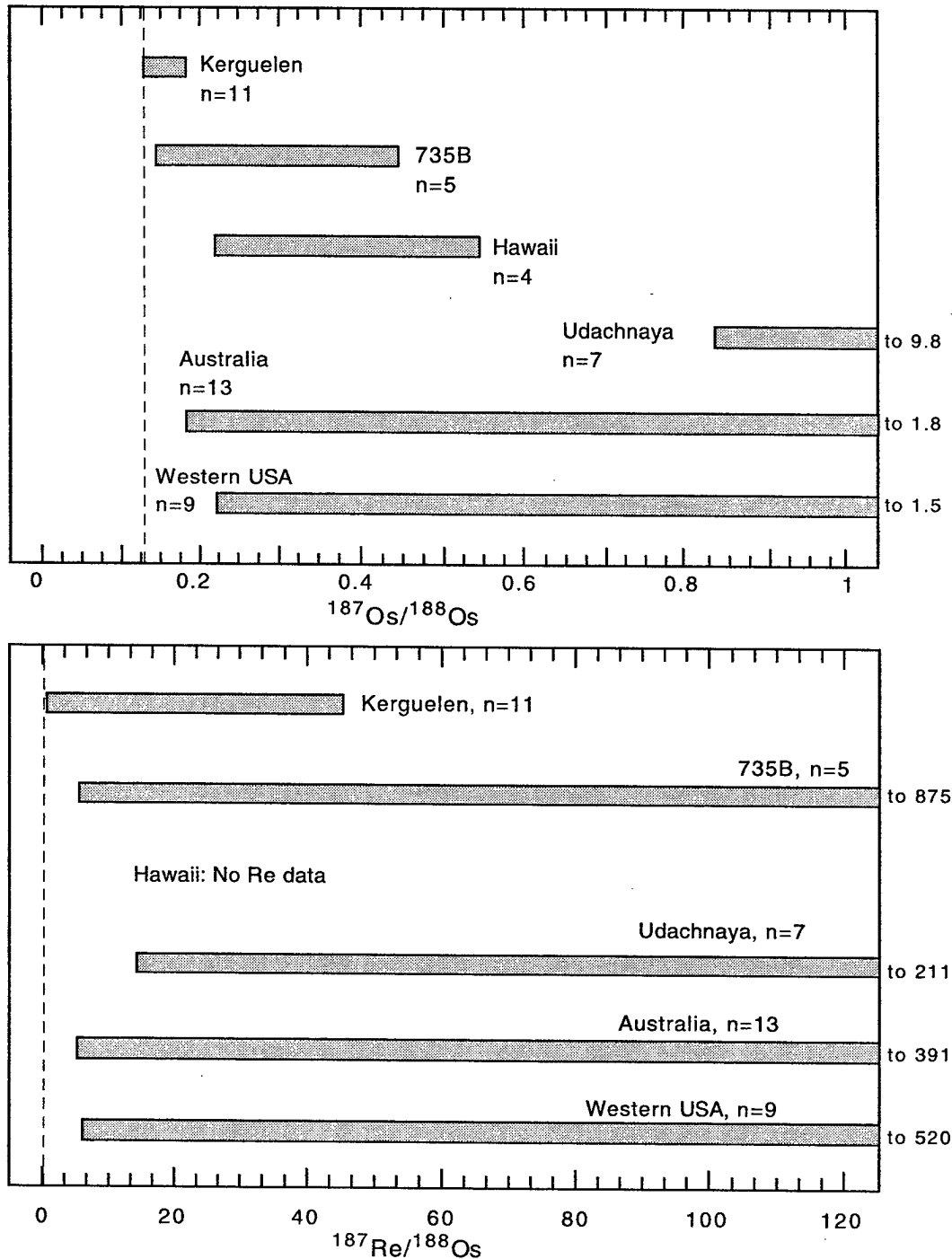


Figure 4–21. Os v. Sr (upper panel) and v. Nd (lower panel) isotopic compositions of Kerguelen plume rocks. Isotopic compositions of Courbet Peninsula granulites (filled squares), Kerguelen Islands basalts (open circles; Reisberg et al., 1993; Yang et al., 1998), and Heard Island basalts (gray circles; Barling et al., 1994; Barling et al., 1997) have a relatively large variation. Isotopic compositions shown for Kerguelen basalts are initial values and those for Heard basalts are measured values (~zero age). Osmium compositions for the granulites are corrected for a maximum eruption age of 24.5 Ma (Nicolaysen et al., 1999); Sr and Nd isotopic compositions are as measured (correction for 24.5 Ma eruption age is negligible). Compositions of mantle endmembers (Hart, 1988; Martin, 1991; Hauri and Hart, 1993; Reisberg et al., 1993; Marcantonio et al., 1995; Widom and Shirey, 1996) are labeled and displayed as open diamonds. Compositional fields for endmember ocean islands are also shown; EM I: Kerguelen Islands and Pitcairn Islands; EM II: Samoa and Tahaa; HIMU: St. Helena, Rurutu, Tubuai, and Mangaia. Data sources for these fields are as noted in the figure explanation. A vertical line through the symbol marks the reported range of the $^{187}\text{Os}/^{188}\text{Os}$ EM II component. No one mixing relation among the isotopic endmember components can account for all of the Kerguelen plume related rocks.

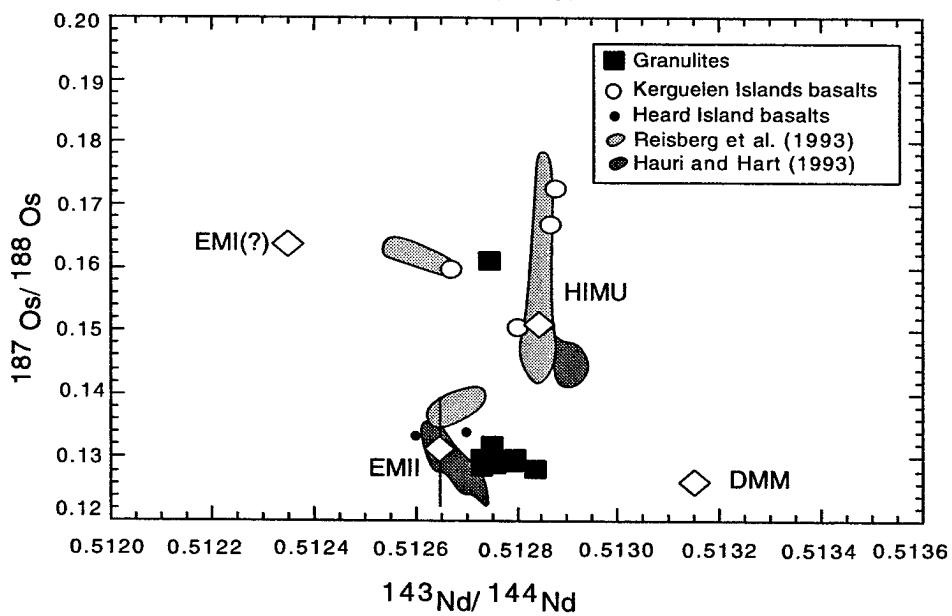
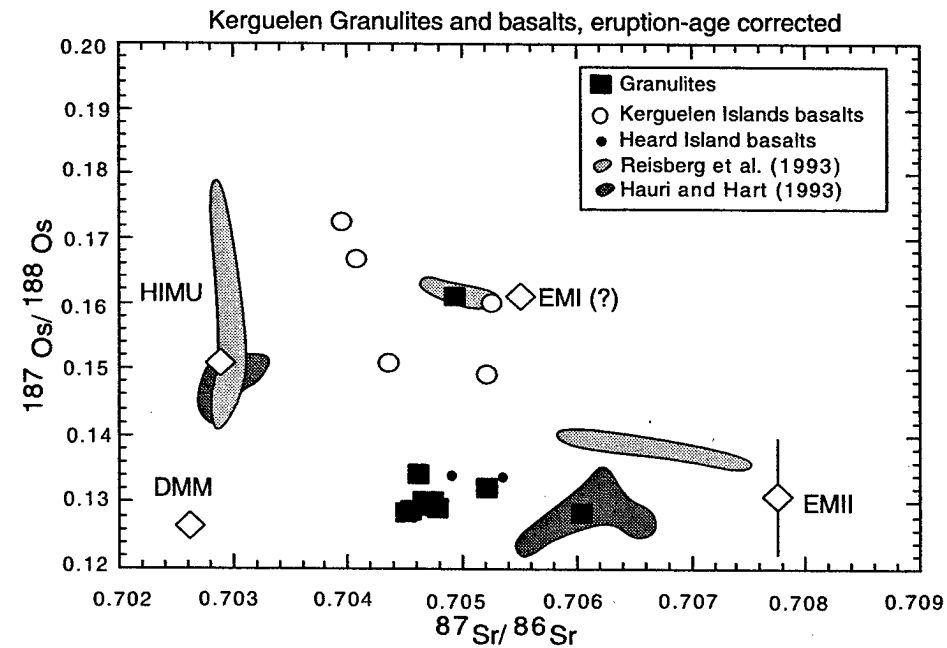
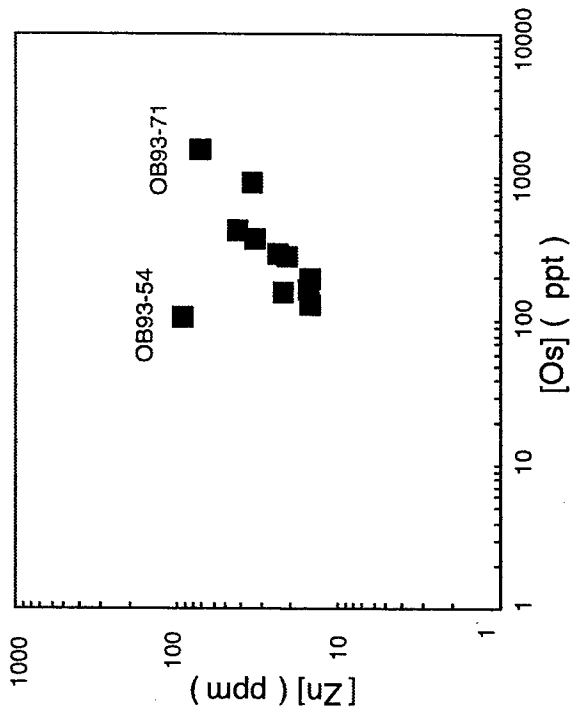
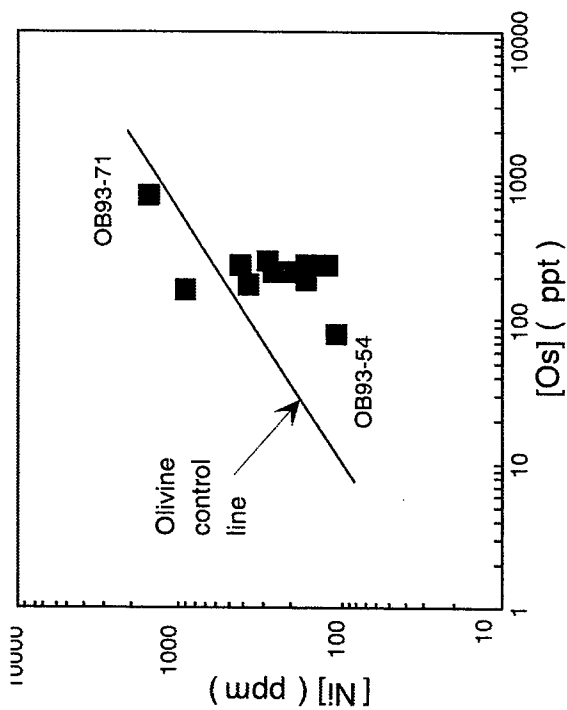
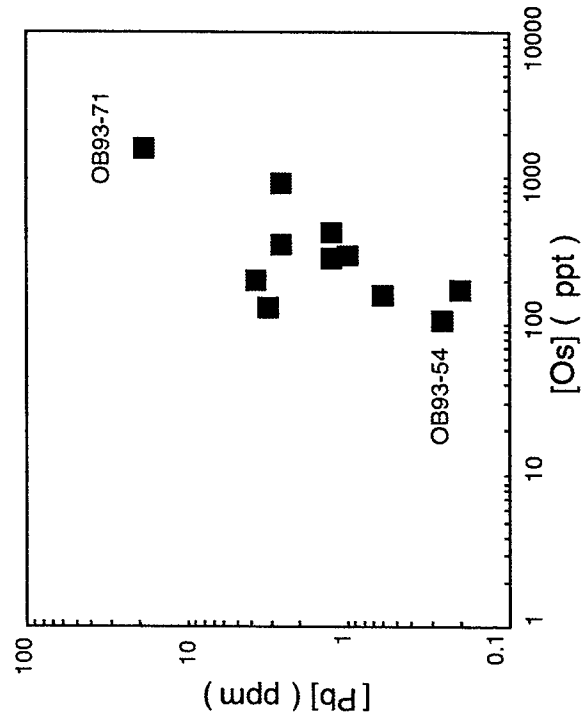
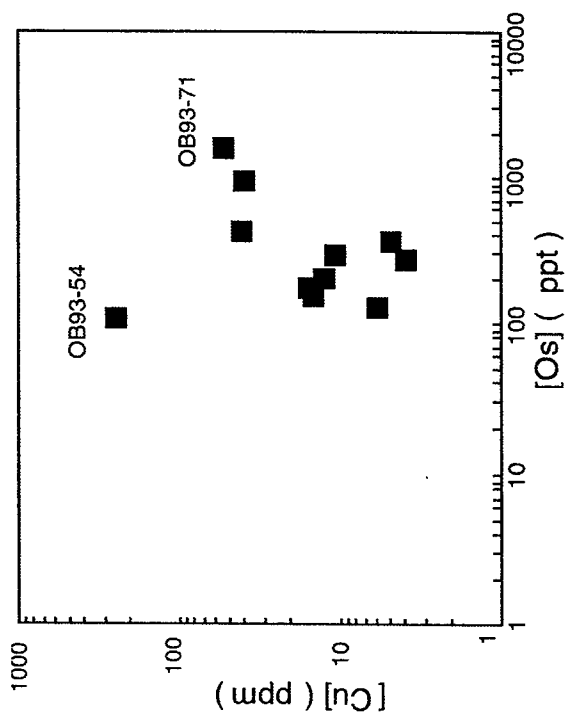
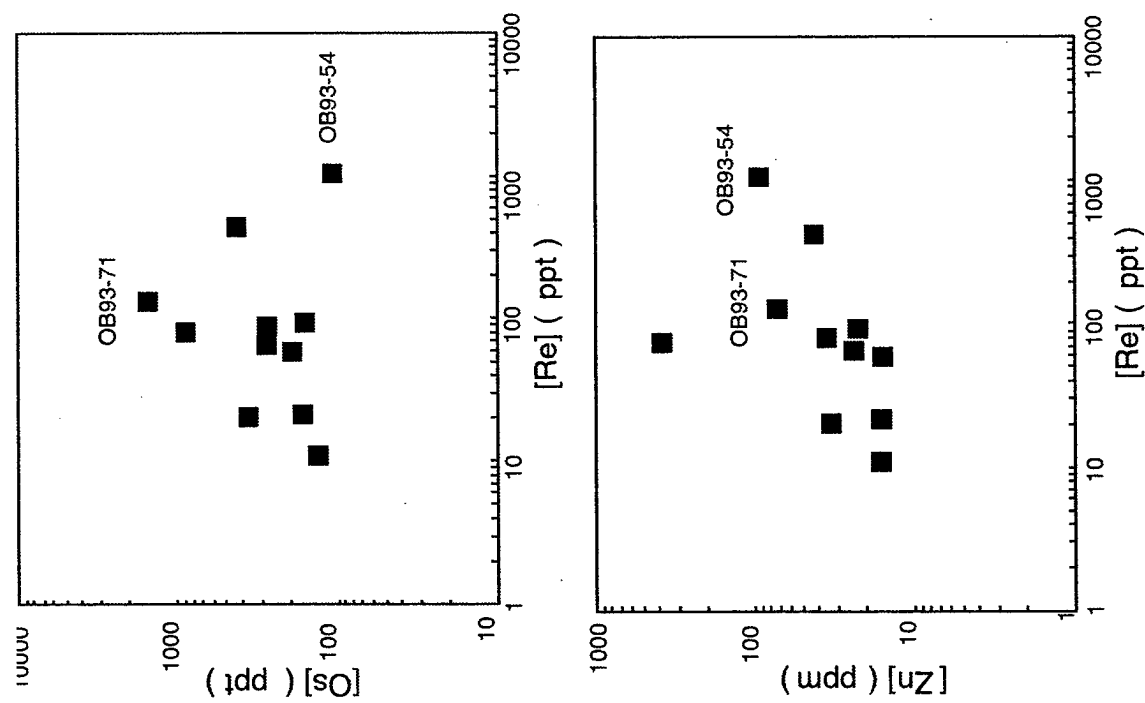
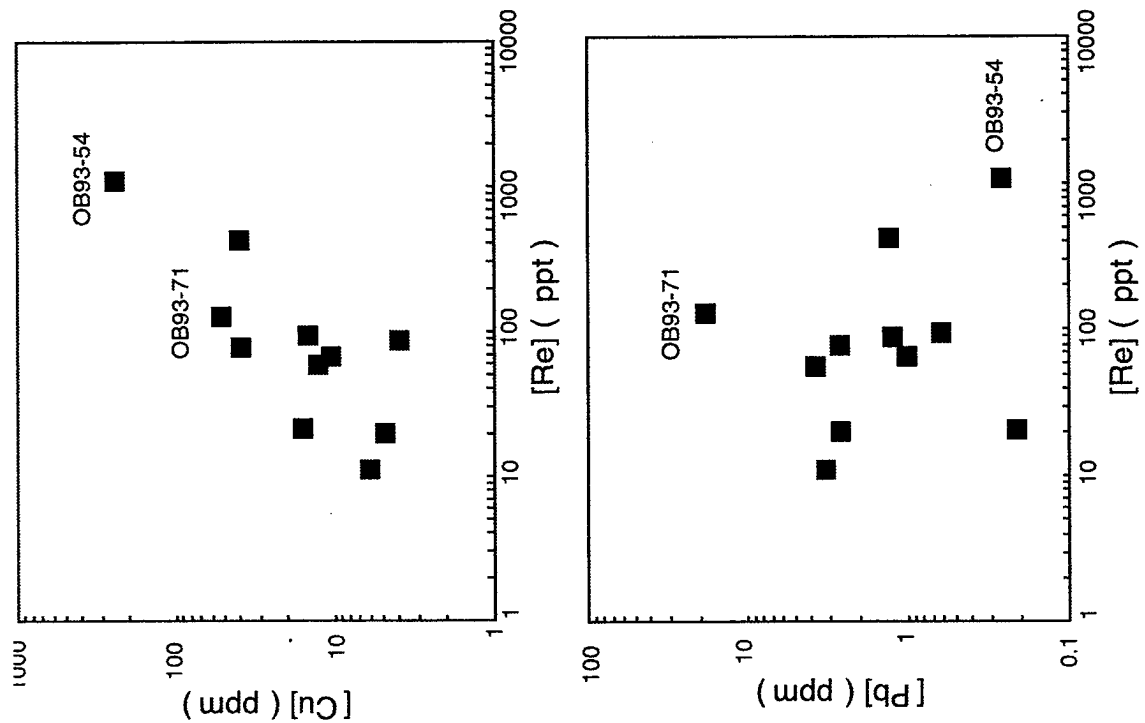


Figure 4-22. Granulite whole rock abundances (filled squares) of the chalcophile element (Ni, Zn, Pb, Cu) v. Re and Os concentrations. Rhenium and Osmium co-vary with different chalcophile elements, indicating their tendency to occur in different mineral phases. The most likely mineral hosts for Re and Os are sulfides. Further evidence for this is that the granulite Ni and Os compositions do not follow an olivine control line, but plot as a vertical trend, suggesting buffering of the Os content by sulfides (Hart and Ravizza, 1996).





Chapter 5 Summary of Results and Conclusions

The major focus of this thesis is a study of the geochemistry of lower crustal and upper mantle xenoliths included in Kerguelen magmas. The objective was to gain insights into the materials and processes that were involved in the evolution of the Kerguelen plume. Of particular interest was the examination of whether the xenoliths carry information pertinent to evaluating the effects of diverse geodynamic environments on the evolution of the Kerguelen plume. For this reason, the determination of the provenance of the different xenoliths was essential. Specifically, the thesis addresses whether pieces of Gondwanaland lithosphere are present beneath the Kerguelen Islands and whether lower crustal xenoliths originated from the Kerguelen plume-derived magmas or the SEIR-derived ones. The process of determining the answers to these questions, whole-rock and mineral major element compositions, trace-element abundances and isotopic compositions of Sr, Nd and Os were used in close connection so that an internally consistent interpretation could be made.

5.1 Upper mantle xenoliths

The major element compositions and modal abundances of minerals were found to place important constraints on the origins of the majority of the harzburgites. Their olivine compositions and modal abundances were found to be off the "oceanic trend" of Boyd (1989). The oceanic trend is interpreted to define a trend of melt extraction at low pressures (e.g. at mid-ocean ridges). The Kerguelen harzburgites generally have too little olivine for a given olivine composition to be residues of low-pressure melting. Many of the Samoan

xenoliths studied by Hauri & Hart (1994), and some of the Hawaiian xenoliths (Sen, 1988) share these same characteristics but have never been discussed in this context.

On the basis that melting reactions change from olivine-producing at low pressures to orthopyroxene-producing and olivine-consuming at higher pressures (Walter et al., 1995; Walter, 1998; Kinzler, 1997), it was determined that the major element characteristics of the Kerguelen harzburgites are consistent with their origin as residues of extensive melt extraction at high pressures. This therefore suggests that the harzburgites are not pieces of oceanic lithosphere upon which the Kerguelen plume impinged, but rather pieces of plume mantle residues.

Abundance patterns of incompatible trace elements in clinopyroxenes from peridotite xenoliths were then used to evaluate whether the interpretation of a high pressure origin on the basis of the major elements is consistent with the trace element abundances. Moreover, the clinopyroxene trace element abundances were used to determine the magnitude of chemical modifications that occurred to the peridotite xenoliths during reactions with plume-derived melt. Rare earth element patterns were calculated for the high-Ca clinopyroxenes that are predicted to be present after the low pressure recrystallization of high pressure melting residues. These calculated patterns produced a close match with the LREE-depleted clinopyroxenes that are present in some of the harzburgites. This trace element evidence further supports the model for the formation of many of the Kerguelen harzburgites by low pressure recrystallization of high pressure melting residues.

Furthermore, it was found that these depleted patterns were modified, sometimes significantly during metasomatism, resulting in ranges of trace element abundances in clinopyroxenes that are staggeringly large (a factor of 5000!). Cerium concentration varies from 0.04 times chondrite in depleted "unmetasomatized" harzburgites to 200 times

chondrite in a fully metasomatized harzburgite that contain phlogopite and apatite. In many samples, a correlation was found between texture and chemistry. Clinopyroxenes in symplectite are depleted in trace elements, whereas discrete grains were are enriched. Additionally, symplectite clinopyroxenes were found to show various stages of modification within a single sample. These observations show that metasomatism began with a reaction between harzburgite and infiltrating melts, with a chromatographic mechanism that has previously been well described (e.g. Navon and Stolper, 1987; Bodinier et al., 1990; Takazawa et al., 1992; Hauri and Hart, 1994) and ended with the precipitation of new clinopyroxene crystals, particularly when the infiltrating melt was carbonatitic.

It is evident that the Sr and Nd isotopic compositions of clinopyroxenes were completely dominated by those of infiltrating melts, i.e. the plume melts. The ranges of the Sr and Nd isotopic compositions were found to be much larger than those found thus far for the Kerguelen basalts. If a harzburgite with the most depleted $^{87}\text{Sr}/^{86}\text{Sr}$ were to be considered as a plume residue on the basis of its major element composition, the $^{87}\text{Sr}/^{86}\text{Sr}$ of the plume ranges from 0.7033 to 0.70681. The enriched end of the spectrum is much more enriched than any of the Kerguelen Islands basalts, but are within those of the Heard Island basalts.

Because of the compatible nature of Os, its isotopes provide a different perspective. Osmium isotopic compositions in peridotites are "indigenous" characteristics, reflecting their provenance. One of the most significant contributions of this study is the discovery of a group of harzburgites with unradiogenic Os. Their Re depletion model ages go up to 1.36 Ga, strongly indicating their origin in the sub-continental lithosphere of Gondwanaland. Their major element characteristics that indicate their origin as high pressure residues are consistent with this interpretation. It is suggested that they represent

pieces of delaminated Gondwanaland lithosphere during continental rifting which coincided with an early part of the Kerguelen plume activation.

The Os isotopic compositions of the non-Gondwanaland peridotites reveal a large heterogeneity within the Kerguelen plume. The range of $^{187}\text{Os}/^{188}\text{Os}$ from 0.1224 to 0.1383 covers practically the entire range observed for the Hawaiian basalts (Hauri, 1996) but is only for peridotites. This, together with the Os data for the granulite xenoliths represents one of the most comprehensive datasets for a plume system, and is particularly important for defining the Os isotopic character of the EM-I endmember.

5.2 Lower crustal xenoliths

Lower crustal xenoliths in ocean island basalts are very rare and the Kerguelen suites are the first granulite facies rocks reported from the oceanic environment (Grègoire et al., 1994). It has been unclear, however, whether their magmatic lineage is connected to the SEIR or the Kerguelen plume. It appears that Grègoire et al. (1998) and Mattielli et al. (1996) linked them to the SEIR because their Sr and Nd isotopic compositions are intermediate between the SEIR basalts and an enriched plume component, represented by basalts from the Southeast Province.

A comparison of the whole rock major element compositions and modal abundances of minerals clearly shows that these rocks were originally olivine-rich gabbros and recrystallized under the reaction: olivine + plagioclase $\rightarrow$ orthopyroxene + clinopyroxene + spinel. Practically all olivine was lost and orthopyroxene was gained as a result of this reaction. The whole rock trace element compositions are indicative that these rocks were gabbroic cumulates, exhibiting large positive Sr anomalies and moderate positive Eu

anomalies. These characteristics, and that Th contents are extremely low argue that the amounts of interstitial melt were very small (<6%).

Abundance patterns of trace elements in granulite whole-rocks and clinopyroxenes were then used to determine their magmatic provenance. This is clearly akin to the Kerguelen basalts rather than to the SEIR mid-ocean ridge basalts. Their Sr and Nd isotopic compositions ($^{87}\text{Sr}/^{86}\text{Sr} = 0.70450$ to 0.70520 ; $\epsilon_{\text{Nd}} = 1.81$ to 3.9) are within the ranges observed for the Kerguelen Islands basalts and are remarkably similar to basalts from the Ninetyeast Ridge (particularly the 58-88 Ma samples from ODP sites 757 and 758 ($^{87}\text{Sr}/^{86}\text{Sr} = 0.70433$ to 0.70501 ; $\epsilon_{\text{Nd}} = 2.5$ to 4.9) (Weis and Frey, 1991). Thus the genetic link of granulites to the Kerguelen plume is well established.

The Os isotopic compositions were close to the present-day bulk earth and are similar to those reported for Heard Island basalts (Barling et al., 1997), but much less radiogenic than values reported for some Kerguelen basalts by Reisburg (1993) and Yang et al. (1998a). Only one sample (OB93-54) has radiogenic $^{187}\text{Os}/^{188}\text{Os}$ similar to the Kerguelen basalts, but because the age of this granulite is unknown precludes any significant inferences to be made. If however, it is assumed that the granulites had uniform initial Os isotopic composition, ~70 Ma of radiogenic ingrowth for the measured $^{187}\text{Re}/^{188}\text{Os}$ could account for the observed $^{187}\text{Os}/^{188}\text{Os}$ in OB93-54. If this is correct, a genetic link between the granulites and the Ninetyeast Ridge could be supported.

Accepting the geochemical conclusion of this study, the granulites were cumulates from Kerguelen plume-derived magmas and subsequently recrystallized under granulite facies conditions by isobaric cooling. On the basis of the mineral assemblages containing sapphirine and garnet, Grègoire et al. (1994) estimated a pressure of recrystallization of ~1 GPa. The granulites studied here are almost garnet-free and may have recrystallized at ~0.6

GPa. This requires cumulate formation at depths of approximately 20 km. By analogy to the formation of the lower crust at ocean ridges, as viewed at the Moho transition zone of the Oman ophiolite (e.g. Kelemen et al., 1997), it could be visualized that magmatic underplating for plume systems occurs as emplacement of series' of gabbroic bodies in much the same way as the Oman Gabbro sills are emplaced.

5.3 Evolution of the Kerguelen plume: a brief synthesis

The results of this study can be placed with previous geochemical and geophysical studies in a context of the evolution of the Kerguelen plume. Although a genetic link with the Bunbury basalts (105-136 Ma, W. Australia) and the Rajmahal traps (108-128 Ma, northeastern India) is uncertain (Storey et al., 1989; Frey et al., 1996), the first certain inception of the Kerguelen Plateau is placed around 115 Ma (Weis et al., 1989). The delamination of Gondwanaland lithosphere and subsequent entrainment into the plume must have occurred by this time. The isotopic data of basalts from ODP site 738 (~115 Ma; Alibert, 1991; Mahoney et al., 1994) appear to require contamination by continental crust in a situation where plume magmatism is not distal to sources of the required contaminants. A pyroxenite from the Kerguelen Islands may have been produced in this period (OB93-94).

The record of magmatic activity between 90 and 40 Ma appears to be on the Ninetyeast Ridge, which is interpreted as a hotspot track. A recent seismological study reveals that the crustal thickness beneath the Ninetyeast Ridge exceeds 25 km (Flueh et al., 1999). This provides an ideal ground for postulating that underplating of plume-derived melts as gabbroic cumulates and subsequent recrystallization as granulites may have occurred during this time.

After the SEIR drifted away from the plume, the Kerguelen Islands were constructed. Temporal variations of the Sr and Nd isotopic compositions from the northwest to the southeast areas of the Kerguelen Islands were previously interpreted as due to a temporal decrease in the mixing proportions of the DMM components (e.g. Gautier et al., 1990; Weis et al., 1993) but in light of Yang et al. (1998a), the mixing hypothesis should be discounted. The position of this study however, is that the observed isotopic variations are due to the heterogeneity of the plume source, rather than the Yang et al. (1998a) hypothesis, which calls for plume heat-induced melting of granulites as a source component. This position is made on petrological and geochemical grounds, in that the basalts are undersaturated in plagioclase and the basalts lack positive Sr anomalies.

It appears that young volcanic rocks (Southeast Province, Mt. Ross and the host dike for the xenoliths of this study), as well as melts that caused young metasomatic events have the Sr and Nd isotopic characteristics of enriched mantle components. When the isotopic compositions of Heard Island (one of the youngest magmatic expressions of the plume; Barling et al., 1994) are added to the picture, temporal variations of the Sr and Nd compositions from the Ninetyeast Ridge stage to the present appear to emerge (Figure 5-1) summarizes this progression. This cannot be due to the aging of a uniform plume as advocated for the Ninetyeast Ridge U-Pb system by Class et al. (1993), because the $^{143}\text{Nd}/^{144}\text{Nd}$ decreases with the temporal progression.

As for plume-lithosphere interaction, the geochemical signatures for peridotite-melt reactions are abundant. It is important that the reactions leave behind rocks that are very heterogeneous on small spatial scales and that melts and peridotites were not equilibrated. That heterogeneities on such small scales under mantle conditions can survive only for limited lengths of time suggests that the metasomatism was a young process that occurred shortly before eruption.

Rocks upon which metasomatic records are printed, are thought to be residues of melting at high pressures that are later recrystallized prior to metasomatism. Although this interpretation was formulated for the Kerguelen harzburgites, it is clearly applicable to the Samoan and some of the Hawaiian xenoliths. For the Samoan xenoliths, Hauri & Hart (1994) reached the same conclusion from the garnet signatures in the trace elements; the interpretation of this study now suggests that their conclusion is consistent with the major element compositions.

Sen et al. (1988) and Yang et al. (1998b) analyzed selected peridotite xenoliths from Oahu which contained clinopyroxenes with original REE patterns resembling those of abyssal peridotites and discussed chemical interactions between the Hawaiian plume and oceanic lithosphere. As discussed above, not a single piece of peridotite which can unequivocally be considered as oceanic lithosphere was found among the Kerguelen peridotites. In terms of $^{87}\text{Sr}/^{86}\text{Sr}$, OB93-78 (= 0.7033) could potentially be, but its major element composition classified this rock as a piece of the plume residue together with the other harzburgites.

The Os isotope data assembled in this study make up one of the most comprehensive datasets for a plume system. The range of $^{187}\text{Os}/^{188}\text{Os}$ obtained for the non-Gondwanaland peridotites is from 0.1252 to 0.1383 with a mean of 0.1283. The range of present-day $^{187}\text{Os}/^{188}\text{Os}$ for the granulites (except for OB93-54) is from 0.1286 to 0.1344, and one for 70 Ma corrected values (including OB93-54) is from 0.1270 to 0.1337 with a mean of 0.1293. Although the magnitude of the age correction is uncertain, the range for the granulites completely overlaps with that for the peridotites. It is also notable that the Os isotope data for the Heard Island basalts (Barling et al., 1997) also overlaps this range.

Figure 5-2 illustrates the relationships between Os, Sr and Nd isotopic compositions of some ocean island basalts and mantle endmember components. Based on the discussions presented earlier for identifying non-Gondwanaland peridotites and granulites with the Kerguelen plume system, the present results place strong constraints for the Os isotope compositions for EM-I, and perhaps more importantly, suggest that the xenolith-based approach taken here provides much better constraints than the conventional basalt-based approach in which low Os concentrations in basalts are extremely susceptible to alteration in surficial processes and present an insurmountable barrier for obtaining reliable answers.

5.4 References Cited

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5.5 Figures

Figure 5-1. Nd v. Sr isotopic diagram of peridotite clinopyroxenes, bulk rock granulite compositions, and the host basalt from the Kerguelen xenolith suite. Clinopyroxenes from xenoliths of this study are small open circles, and those from the Southeast Province are also shown for reference (small closed circles). Measured isotopic compositions of granulites from this study (open squares), granulites from the Southeastern Province (Mt. Tizard: gray diamonds, Dome Rouge and Val Phonolite: white diamonds; Mattielli et al., 1996; Mattielli, 1996), are shown in comparison to initial isotopic compositions of basalts from the Indian Ocean ridges (MORB with no plume influence: light gray, all Indian Ocean MORB: light gray + dashed line extension), the Kerguelen Islands (dark gray), Heard Island (medium gray, Barling, 1994) and the Ninetyeast Ridge (field "hidden" behind the Kerguelen Islands field; Weis and Frey, 1991). The Sr isotopic composition of OB93-61A is shown as a broken line. Granulites lie nearly completely within the field defined by Kerguelen Islands basalts (White and Hofmann, 1982; Gautier et al., 1990; Weis et al., 1993; Weis et al., 1998; Yang et al., 1998a), and none overlap the compositions of Indian Ocean MORB (Hamelin et al., 1986; Michard et al., 1986; Dosso et al., 1988). Mantle isotopic endmembers are shown for reference (filled circles; Hart, 1988).

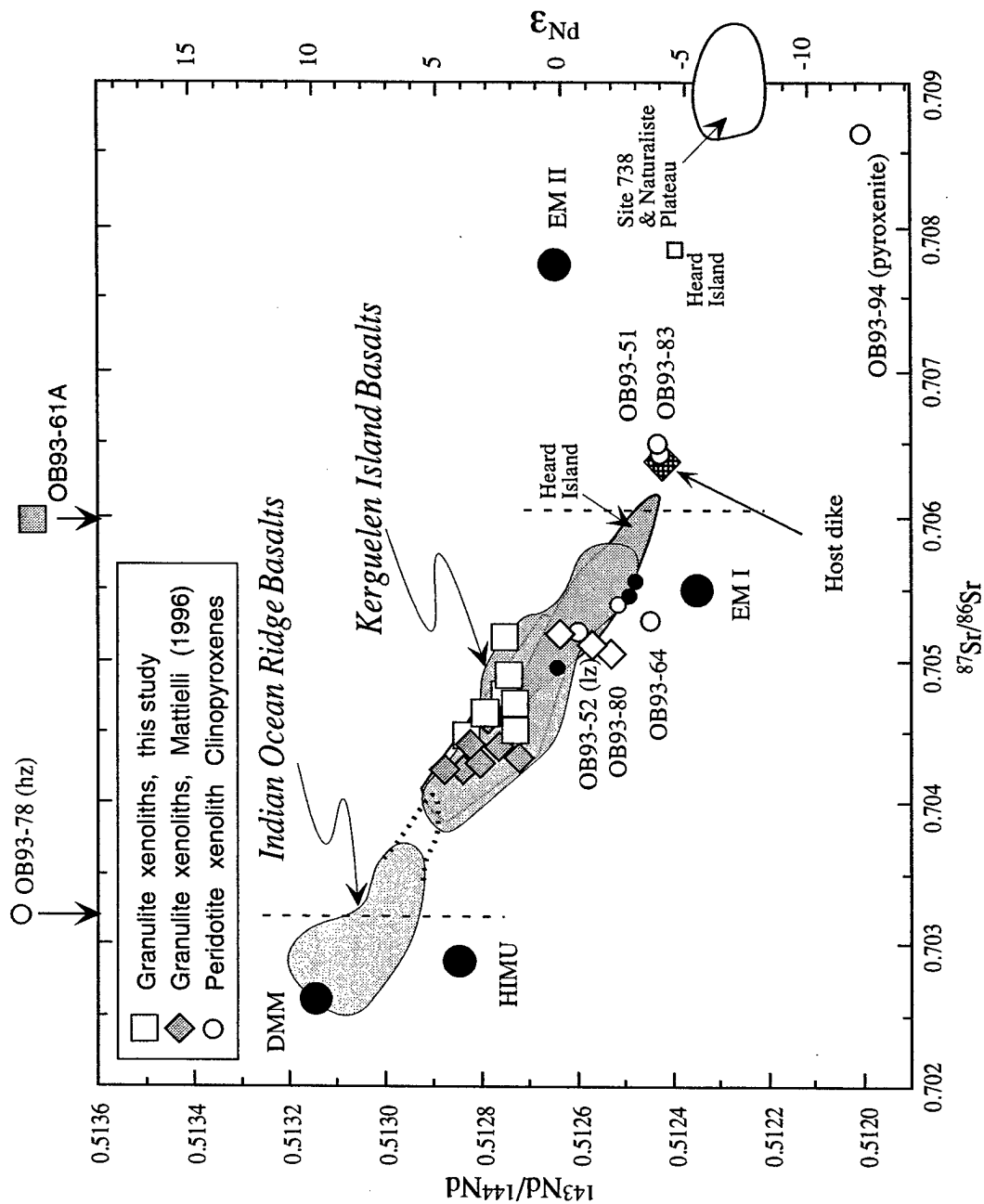
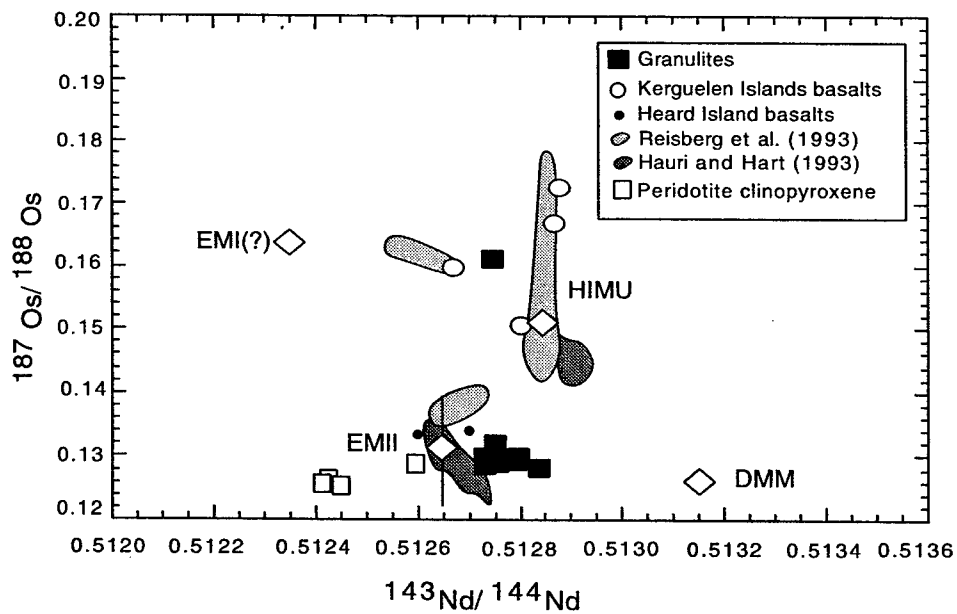
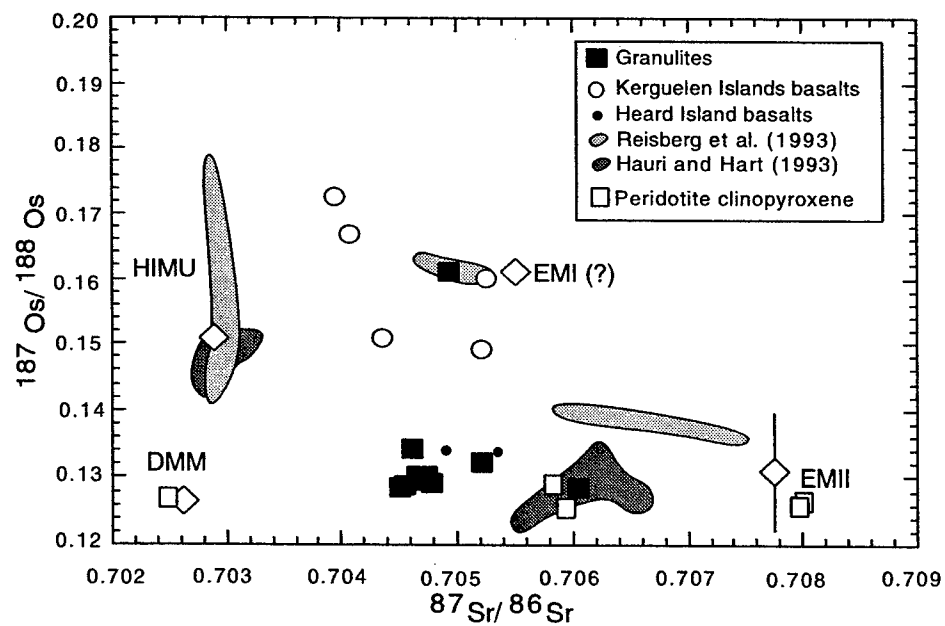


Figure 5–2. Os v. Sr (upper panel) and v. Nd (lower panel) isotopic compositions of Kerguelen plume rocks. Isotopic compositions of Courbet Peninsula granulites (filled squares), peridotite clinopyroxenes (open squares) Kerguelen Islands basalts (open circles; Reisberg et al., 1993; Yang et al., 1998a), and Heard Island basalts (gray circles: Barling et al., 1994; Barling et al., 1997) have a relatively large variation. Isotopic compositions shown for Kerguelen basalts are initial values and those for Heard basalts are measured values (~zero age). Osmium compositions for the granulites are corrected for a maximum eruption age of 24.5 Ma (Nicolaysen et al., 1999); Sr and Nd isotopic compositions are as measured (correction for 24.5 Ma eruption age is negligible). Compositions of mantle endmembers (Hart, 1988; Martin, 1991; Hauri and Hart, 1993; Reisberg et al., 1993; Marcantonio et al., 1995; Widom and Shirey, 1996) are labeled and displayed as open diamonds. Compositional fields for endmember ocean islands are also shown; EM I: Kerguelen Islands and Pitcairn Islands; EM II: Samoa and Tahaa; HIMU: St. Helena, Rurutu, Tubuai, and Mangaia. Data sources for these fields are as noted in the figure explanation. A vertical line through the symbol marks the reported range of the $^{187}\text{Os}/^{188}\text{Os}$ EM II component. No one mixing relation among the isotopic endmember components can account for all of the Kerguelen plume related rocks.



**Appendix 1. Complete major and trace element compositions of
minerals in the peridotites.**

Table A1.1. Major element compositions of ultramafic xenolith clinopyroxenes. Each analysis is a separate grain, unless noted (e.g. pt1, pt2, etc).

Lac Superieur		Harzburgites										OB93-79		
Lherzolitc		OB93-52					OB93-51					OB93-79		
Sample	Grain ID	cpx1	cpx2	cpx3	by opx2	cpx5	cpx1	cpx2	cpx3s	cpx3s	cpx4s	cpx5s	cpx1	cpx2s
wt. %														
SiO ₂	52.02	51.78	51.94	51.83	51.83	52.13	54.98	55.28	54.51	55.01	54.54	54.41	54.39	54.10
TiO ₂	0.95	0.97	0.91	1.00	1.00	0.91	0.00	0.02	0.02	0.00	0.00	0.00	0.02	0.00
Al ₂ O ₃	5.41	5.68	5.48	5.18	5.27	5.27	2.27	2.05	2.51	1.79	1.61	1.74	1.85	2.00
FeO*	2.37	2.43	2.38	2.39	2.40	2.40	2.24	2.35	2.13	2.00	2.02	2.04	1.87	1.77
MnO	0.11	0.09	0.10	0.13	0.10	0.10	0.05	0.06	0.11	0.07	0.14	0.08	0.07	0.11
MgO	15.50	15.42	15.56	15.53	15.53	15.50	17.67	17.99	17.56	17.83	18.55	18.70	18.25	18.26
CaO	22.43	22.15	22.32	22.56	22.54	22.54	21.00	20.69	21.76	22.59	23.16	23.02	22.96	23.19
Na ₂ O	1.05	1.02	0.93	0.98	0.97	0.97	1.21	1.15	0.98	0.78	0.26	0.17	0.26	0.29
Cr ₂ O ₃	0.87	0.98	0.93	0.87	0.89	0.89	0.80	0.60	0.62	0.61	0.55	0.50	0.70	0.75
Total	100.71	100.52	100.54	100.48	100.48	100.70	100.22	100.18	100.19	100.68	100.83	100.66	100.37	100.46
Cations/6 Oxygens														
Si	1.877	1.871	1.876	1.876	1.876	1.881	1.977	1.986	1.964	1.974	1.958	1.955	1.959	1.949
Ti	0.026	0.026	0.025	0.027	0.027	0.025	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Al	0.230	0.242	0.233	0.221	0.221	0.224	0.096	0.087	0.106	0.076	0.068	0.074	0.079	0.085
Fe	0.072	0.073	0.072	0.072	0.072	0.072	0.067	0.070	0.064	0.060	0.061	0.061	0.056	0.053
Mn	0.003	0.003	0.003	0.004	0.004	0.003	0.002	0.002	0.003	0.002	0.004	0.002	0.002	0.003
Mg	0.833	0.830	0.837	0.838	0.838	0.834	0.947	0.963	0.943	0.954	0.993	1.001	0.980	0.981
Ca	0.867	0.858	0.864	0.875	0.875	0.871	0.809	0.797	0.840	0.869	0.891	0.886	0.886	0.895
Na	0.073	0.072	0.065	0.069	0.069	0.068	0.084	0.080	0.069	0.054	0.018	0.012	0.018	0.020
Cr	0.025	0.028	0.026	0.025	0.025	0.026	0.023	0.017	0.018	0.017	0.015	0.014	0.020	0.021
Total	4.007	4.003	4.002	4.008	4.008	4.003	4.006	4.002	4.008	4.006	4.009	4.007	4.000	4.008
Mg#	92.09	91.88	92.08	92.04	92.04	92.01	93.37	93.18	93.62	94.08	94.25	94.23	94.56	94.86
Molar percentage of endmembers														
Wo	48.85	48.61	48.63	48.91	48.91	48.94	44.33	43.48	45.39	46.09	45.73	45.42	46.05	46.33
En	46.94	47.07	47.14	46.82	46.82	46.82	51.89	52.56	50.95	50.61	50.95	51.31	50.92	50.75
Fs	4.21	4.33	4.23	4.27	4.27	4.23	3.77	3.95	3.65	3.30	3.32	3.26	3.03	2.92
Each column represents one analysis. Different analyses of the same grain are noted in "Grain ID". The suffix "s" = symplectite cpx.														

Each column represents one analysis. Different analyses of the same grain are noted in "Grain ID". The suffix "s" = symplectite cpx.

Table A1.1. Major element compositions of ultramafic xenolith clinopyroxenes continued.

Sample	OB93-79			OB93-80			OB93-81c					
	Grain ID	cpx4s	cpx3s	cpx5	cpx4	cpx5s	cpx6s	cpx1s	cpx2	cpx1	cpx3	cpx6s
wt. %												
SiO ₂		54.40	54.29	54.10	54.00	54.27	54.56	54.32	54.53	54.55	54.14	54.21
TiO ₂		0.03	0.02	0.02	0.02	0.02	0.01	0.00	0.02	0.03	0.03	0.05
Al ₂ O ₃		1.77	1.78	2.02	1.90	1.94	1.82	2.08	1.83	1.81	1.97	2.00
FeO*		1.92	1.81	1.85	1.88	1.87	1.88	1.78	1.86	1.79	1.76	1.77
MnO		0.08	0.08	0.09	0.08	0.09	0.10	0.11	0.09	0.09	0.09	0.08
MgO		18.21	18.31	18.03	18.31	18.41	18.30	18.04	18.07	17.96	17.99	17.94
CaO		23.14	23.13	23.07	22.90	23.08	23.28	23.14	23.03	23.35	23.21	23.19
Na ₂ O		0.25	0.26	0.28	0.25	0.22	0.16	0.23	0.33	0.30	0.36	0.31
Cr ₂ O ₃		0.73	0.62	0.74	0.70	0.63	0.67	0.72	0.68	0.73	0.88	0.83
Total		100.53	100.29	100.19	100.04	100.52	100.79	100.43	100.44	100.61	100.43	100.37
Cations/6 Oxygens												
Si		1.958	1.958	1.954	1.953	1.953	1.958	1.956	1.963	1.961	1.952	1.954
Ti		0.001	0.001	0.001	0.000	0.000	0.000	0.000	0.001	0.001	0.001	0.001
Al		0.075	0.075	0.086	0.081	0.082	0.077	0.088	0.078	0.077	0.084	0.085
Fe		0.058	0.055	0.056	0.057	0.056	0.056	0.054	0.056	0.054	0.053	0.053
Mn		0.002	0.003	0.003	0.002	0.003	0.003	0.003	0.003	0.003	0.003	0.002
Mg		0.977	0.984	0.970	0.987	0.987	0.979	0.968	0.969	0.963	0.967	0.964
Ca		0.892	0.894	0.893	0.887	0.890	0.895	0.893	0.888	0.900	0.897	0.896
Na		0.018	0.018	0.020	0.017	0.015	0.011	0.016	0.023	0.021	0.025	0.022
Cr		0.021	0.018	0.021	0.020	0.018	0.019	0.021	0.019	0.021	0.025	0.024
Total		4.002	4.004	4.002	4.005	4.004	3.999	3.998	4.000	4.000	4.006	4.001
Mg#		94.42	94.75	94.57	94.54	94.60	94.56	94.76	94.54	94.70	94.80	94.74
Molar percentage of endmembers												
Wo		46.25	46.19	46.46	45.89	45.96	46.30	46.55	46.35	46.88	46.72	46.77
En		50.63	50.87	50.49	51.04	50.99	50.63	50.48	50.57	50.17	50.37	50.32
Fs		3.11	2.95	3.05	3.07	3.05	3.08	2.97	3.07	2.95	2.90	2.91

Table A1.1. Major element compositions of ultramafic xenolith clinopyroxenes continued.

Sample Grain ID	wt. %	OB93-83				OB93-85				OB93-99			
		by cpx1s	cpx1s	cpx3fs	cpx4	cpx7	cpx1s	cpx2s	cpx3s	cpx1s	cpx2	cpx3s	cpx4
SiO ₂	52.15	52.43	51.91	52.69	52.69	52.51	54.48	54.34	54.19	54.29	54.68	54.40	53.83
TiO ₂	0.15	0.09	0.07	0.13	0.17	0.12	0.02	0.00	0.00	0.16	0.13	0.13	0.16
Al ₂ O ₃	6.20	5.11	5.24	6.02	6.04	5.99	1.91	1.61	1.73	2.45	2.35	2.44	2.45
FeO*	2.82	2.86	2.86	2.82	2.93	2.78	1.76	1.88	1.86	1.80	1.97	1.77	1.70
MnO	0.10	0.11	0.11	0.13	0.12	0.11	0.09	0.12	0.09	0.13	0.07	0.07	0.08
MgO	15.55	16.05	16.22	16.10	16.18	16.22	18.17	18.35	18.34	17.62	17.27	17.35	17.03
CaO	18.89	18.74	17.46	18.84	18.95	19.02	23.44	23.15	23.11	22.52	21.31	22.64	22.47
Na ₂ O	1.79	1.76	2.04	1.75	1.72	1.53	0.18	0.15	0.19	0.85	1.42	0.81	1.18
Cr ₂ O ₃	2.07	2.78	2.24	1.84	1.37	1.33	0.71	0.52	0.64	1.22	1.00	1.12	1.24
Total	99.72	99.92	98.15	100.31	100.17	99.60	100.75	100.12	100.14	101.02	100.19	100.73	100.12
<i>Cations/6 Oxygens</i>													
Si	1.890	1.901	1.909	1.896	1.898	1.900	1.956	1.963	1.958	1.946	1.970	1.953	1.948
Ti	0.004	0.002	0.002	0.003	0.005	0.003	0.001	0.000	0.000	0.004	0.003	0.004	0.004
Al	0.265	0.218	0.227	0.255	0.256	0.256	0.081	0.069	0.074	0.103	0.100	0.103	0.104
Fe	0.085	0.087	0.088	0.085	0.088	0.084	0.053	0.057	0.056	0.054	0.059	0.053	0.051
Mn	0.003	0.003	0.004	0.004	0.004	0.003	0.003	0.004	0.003	0.004	0.002	0.002	0.002
Mg	0.840	0.867	0.889	0.863	0.869	0.874	0.972	0.988	0.987	0.941	0.927	0.929	0.919
Ca	0.734	0.728	0.688	0.726	0.731	0.738	0.902	0.896	0.894	0.865	0.823	0.871	0.871
Na	0.126	0.124	0.145	0.122	0.120	0.107	0.013	0.011	0.013	0.059	0.099	0.056	0.083
Cr	0.059	0.080	0.065	0.052	0.039	0.038	0.020	0.015	0.018	0.034	0.028	0.032	0.035
Total	4.007	4.010	4.016	4.008	4.010	4.004	3.999	4.001	4.003	4.010	4.012	4.004	4.019
Mg#	90.77	90.91	91.01	91.06	90.78	91.22	94.85	94.56	94.61	94.59	93.99	94.59	94.70
<i>Molar percentage of endmembers</i>													
Wo	44.14	43.20	41.24	43.28	43.23	43.40	46.74	46.08	46.09	46.41	45.42	46.96	47.26
En	50.53	51.45	53.28	51.44	51.34	51.45	50.39	50.81	50.87	50.50	51.19	50.06	49.83
Fs	5.33	5.35	5.47	5.29	5.43	5.15	2.88	3.11	3.04	3.09	3.39	2.98	2.92

Table A1.1. Major element compositions of ultramafic xenolith clinopyroxenes continued.

Mt Trapez																
Pyroxenites				Harzburgites												
Sample	OB93-94			OB93-281											OB93-282	
Grain ID	cpx1	cpx2	cpx3	cpx1	cpx2	cpx3.pt1	cpx3.pt2	cpx4.pt1	cpx4.pt2	cpx5	cpx6	cpx5	cpx1	cpx2		
wt. %																
SiO ₂	52.97	53.00	53.56	53.49	53.57	54.04	54.17	53.71	53.61	53.57	53.95	53.57	53.92	53.66		
TiO ₂	0.41	0.32	0.21	0.07	0.11	0.12	0.12	0.10	0.12	0.12	0.12	0.12	0.14	0.15		
Al ₂ O ₃	3.48	3.21	3.31	2.45	2.43	2.49	2.47	2.31	1.98	2.43	2.26	2.43	2.50	2.50		
FeO*	2.61	2.59	2.42	1.91	1.97	2.07	1.98	1.92	1.91	1.98	2.06	1.98	1.74	1.75		
MnO	0.11	0.10	0.09	0.11	0.13	0.00	0.04	0.08	0.03	0.03	0.06	0.03	0.05	0.06		
MgO	17.01	17.26	17.14	17.00	16.78	16.62	16.50	16.84	17.30	16.95	17.24	16.95	16.60	16.20		
CaO	22.41	22.06	22.34	23.35	23.00	22.65	22.80	23.03	23.41	22.84	23.09	22.84	22.24	22.08		
Na ₂ O	0.54	0.58	0.64	0.70	0.60	1.02	1.03	0.56	0.57	0.67	0.55	0.67	1.07	1.29		
Cr ₂ O ₃	1.13	0.91	1.00	1.14	1.21	1.06	1.19	1.21	1.01	1.12	1.05	1.12	1.64	1.50		
Total	100.66	100.01	100.70	100.22	99.80	100.07	100.31	99.76	99.93	99.69	100.38	99.69	99.90	99.18		
Cations/6 Oxygens																
Si	1.912	1.923	1.928	1.939	1.948	1.957	1.958	1.952	1.948	1.948	1.949	1.948	1.955	1.960		
Ti	0.011	0.009	0.006	0.002	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.004	0.004		
Al	0.148	0.137	0.141	0.105	0.104	0.106	0.105	0.099	0.085	0.104	0.096	0.104	0.107	0.108		
Fe	0.079	0.079	0.073	0.058	0.060	0.063	0.060	0.058	0.058	0.060	0.062	0.060	0.053	0.053		
Mn	0.003	0.003	0.003	0.003	0.004	0.000	0.001	0.002	0.001	0.001	0.002	0.001	0.002	0.002		
Mg	0.915	0.933	0.919	0.919	0.909	0.897	0.889	0.912	0.937	0.919	0.928	0.919	0.897	0.882		
Ca	0.867	0.857	0.862	0.907	0.896	0.879	0.883	0.897	0.911	0.894	0.894	0.890	0.864	0.864		
Na	0.038	0.041	0.045	0.049	0.042	0.072	0.072	0.040	0.040	0.047	0.039	0.047	0.075	0.091		
Cr	0.032	0.026	0.028	0.033	0.035	0.030	0.034	0.035	0.029	0.032	0.030	0.032	0.047	0.043		
Total	4.006	4.007	4.004	4.015	4.001	4.007	4.006	3.998	4.012	4.004	4.004	4.004	4.002	4.007		
Mg#	92.08	92.24	92.66	94.07	93.81	93.46	93.68	93.98	94.16	93.85	93.72	93.85	94.45	94.30		
Molar percentage of endmembers																
Wo	46.51	45.80	46.41	48.07	47.93	47.80	48.17	47.97	47.78	47.39	47.39	47.60	47.60	47.97		
En	49.10	49.84	49.52	48.68	48.64	48.79	48.49	48.79	49.12	49.21	49.21	49.14	49.41	48.96		
Fs	4.40	4.35	4.07	3.25	3.43	3.41	3.34	3.25	3.10	3.40	3.40	3.26	2.99	3.07		

Table A1.1. Major element compositions of ultramafic xenolith clinopyroxenes continued.

Sample	OB93-282					OB93-284				OB93-287			
	Grain ID	cpx3s,pt1	cpx3s,pt2	cpx4	cpx6,pt1	cpx6,pt2	cpx5	cpx7	cpx1,pt1	cpx1,pt2	cpx8s	cpx1,pt1	cpx1,pt2
wt. %													
SiO ₂	54.11	54.34	54.44	54.21	53.56	54.31	54.50	53.13	53.79	54.24	54.21	53.89	54.23
TiO ₂	0.17	0.17	0.17	0.14	0.16	0.14	0.15	0.03	0.06	0.06	0.04	0.15	0.12
Al ₂ O ₃	2.48	3.06	2.44	2.37	2.47	2.32	2.33	3.18	3.25	3.03	2.77	2.39	2.37
FeO*	1.90	2.10	1.81	1.89	1.75	1.76	1.81	2.13	2.12	2.30	2.47	1.99	2.03
MnO	0.05	0.07	0.07	0.12	0.09	0.08	0.10	0.04	0.10	0.06	0.09	0.05	0.03
MgO	16.38	16.18	16.37	16.64	16.61	16.57	16.53	17.36	17.37	17.60	17.66	16.97	16.86
CaO	21.66	20.40	22.10	22.12	22.28	22.37	22.16	22.33	22.70	22.79	22.48	22.59	22.77
Na ₂ O	1.48	1.77	1.31	1.19	1.12	1.07	1.08	0.59	0.54	0.53	0.47	0.89	0.82
Cr ₂ O ₃	1.44	1.34	1.54	1.26	1.63	1.44	1.32	1.34	1.38	1.28	1.23	1.34	1.31
Total	99.67	99.42	100.26	99.93	99.67	100.05	99.97	100.13	101.31	101.88	101.41	100.26	100.55
<i>Cations/6 Oxygens</i>													
Si	1.965	1.971	1.965	1.964	1.949	1.964	1.971	1.924	1.925	1.930	1.938	1.949	1.955
Ti	0.005	0.005	0.005	0.004	0.004	0.004	0.004	0.001	0.002	0.002	0.001	0.004	0.003
Al	0.106	0.131	0.104	0.101	0.106	0.099	0.100	0.136	0.137	0.127	0.117	0.102	0.101
Fe	0.058	0.064	0.055	0.057	0.053	0.053	0.055	0.065	0.063	0.068	0.074	0.060	0.061
Mn	0.002	0.002	0.002	0.004	0.003	0.002	0.003	0.001	0.003	0.002	0.003	0.002	0.001
Mg	0.886	0.874	0.881	0.898	0.901	0.893	0.891	0.937	0.927	0.933	0.941	0.915	0.906
Ca	0.843	0.793	0.855	0.858	0.869	0.867	0.859	0.866	0.871	0.869	0.861	0.875	0.880
Na	0.104	0.124	0.091	0.084	0.079	0.075	0.076	0.041	0.038	0.037	0.033	0.062	0.058
Cr	0.041	0.038	0.044	0.036	0.047	0.041	0.038	0.038	0.039	0.036	0.035	0.038	0.037
Total	4.009	4.002	4.002	4.006	4.010	3.999	3.994	4.009	4.004	4.005	4.002	4.008	4.002
Mg#	93.88	93.22	94.15	94.00	94.41	94.38	94.23	93.56	93.59	93.18	92.73	93.84	93.68
<i>Molar percentage of endmembers</i>													
Wo	47.12	45.75	47.68	47.23	47.59	47.74	47.52	46.38	46.78	46.46	45.90	47.31	47.63
En	49.56	50.45	49.14	49.43	49.34	49.20	49.29	50.16	49.80	49.89	50.16	49.44	49.06
Fs	3.32	3.79	3.18	3.35	3.06	3.06	3.19	3.45	3.41	3.65	3.94	3.25	3.31

Table A1.1. Major element compositions of ultramafic xenolith clinopyroxenes continued.

Sample	OB93-287			OB93-289			OB93-291		
	Grain ID	cpx3	cpx5s	cpx6	cpx1	cpx3	cpx7s,pt1	cpx7s,pt2	cpx8
<i>wt. %</i>									
SiO ₂	54.56	54.83	54.02	54.16	53.97	54.16	53.83	53.23	53.89
TiO ₂	0.14	0.09	0.12	0.02	0.00	0.02	0.02	0.01	0.07
Al ₂ O ₃	2.35	2.27	2.31	2.55	2.55	2.55	2.65	2.55	2.66
FeO*	1.97	2.34	2.11	2.31	2.31	2.39	2.34	2.16	2.08
MnO	0.02	0.09	0.06	0.06	0.08	0.06	0.08	0.07	0.08
MgO	16.90	17.13	16.94	17.32	17.27	17.32	17.20	17.35	16.78
CaO	22.63	21.69	22.58	22.82	23.04	22.82	23.48	23.28	21.67
Na ₂ O	1.00	1.22	0.88	0.50	0.43	0.50	0.45	0.44	1.26
Cr ₂ O ₃	1.26	1.19	1.27	0.94	1.09	0.94	1.07	1.08	1.24
Total	100.81	100.84	100.28	100.75	100.75	100.75	101.11	100.18	99.74
<i>Cations/6 Oxygens</i>									
Si	1.960	1.968	1.954	1.949	1.944	1.949	1.935	1.931	1.956
Ti	0.004	0.002	0.003	0.000	0.000	0.000	0.001	0.000	0.002
Al	0.099	0.096	0.098	0.108	0.108	0.108	0.112	0.109	0.114
Fe	0.059	0.070	0.064	0.072	0.070	0.072	0.070	0.066	0.063
Mn	0.000	0.003	0.002	0.002	0.002	0.002	0.002	0.002	0.003
Mg	0.905	0.916	0.913	0.929	0.927	0.929	0.921	0.938	0.908
Ca	0.871	0.834	0.875	0.880	0.889	0.880	0.904	0.905	0.843
Na	0.069	0.085	0.062	0.035	0.030	0.035	0.031	0.031	0.089
Cr	0.036	0.034	0.036	0.027	0.031	0.027	0.030	0.031	0.035
Total	4.004	4.008	4.007	4.001	4.002	4.001	4.008	4.014	4.012
Mg#	93.88	92.87	93.48	92.80	93.02	92.80	92.91	93.46	93.51
<i>Molar percentage of endmembers</i>									
Wo	47.47	45.81	47.26	46.79	47.15	46.79	47.69	47.41	46.47
En	49.31	50.33	49.31	49.38	49.16	49.38	48.60	49.15	50.06
Fs	3.22	3.86	3.44	3.83	3.69	3.83	3.71	3.44	3.47

Table A1.1. Major element compositions of ultramafic xenolith clinopyroxenes continued.

Sample Grain ID	OB93-297			OB93-302				OB93-305					
	cpx1.ptl	cpx1.pt2	cpx2	cpx2.pt1	cpx2.pt2	cpx7.pt1	cpx7.pt1	cpx3.pt1	cpx3.pt2	cpx2.pt1	cpx2.pt2	cpx5.pt1	cpx5.pt2
wt. %													
SiO ₂	53.96	53.80	53.53	54.79	54.59	54.44	54.22	53.87	54.07	53.61	54.07	53.47	54.05
TiO ₂	0.26	0.32	0.30	0.04	0.02	0.01	0.04	0.04	0.04	0.18	0.10	0.00	0.00
Al ₂ O ₃	3.13	3.38	3.17	2.63	2.71	2.75	2.60	2.82	2.81	2.94	2.86	2.47	2.56
FeO*	2.33	2.10	2.09	2.08	2.01	2.02	2.00	2.05	2.04	2.04	2.06	2.20	2.19
MnO	0.10	0.06	0.08	0.06	0.05	0.05	0.04	0.05	0.08	0.05	0.04	0.06	0.06
MgO	16.70	16.24	16.50	16.87	16.96	16.74	16.70	16.79	16.93	16.96	16.91	16.89	16.89
CaO	21.68	22.10	21.99	22.09	22.48	22.13	22.72	22.59	22.53	22.46	21.98	22.60	22.85
Na ₂ O	1.27	1.26	1.14	1.13	1.12	1.10	0.96	0.89	0.98	0.87	1.04	0.93	0.89
Cr ₂ O ₃	1.60	1.78	1.51	1.13	1.07	1.11	1.09	1.11	1.12	1.08	1.10	1.19	1.25
Total	101.03	101.04	100.30	100.81	101.00	100.35	100.38	100.20	100.58	100.17	100.16	99.81	100.75
Cations/6 Oxygens													
Si	1.937	1.932	1.935	1.965	1.956	1.961	1.957	1.948	1.948	1.939	1.952	1.946	1.948
Ti	0.007	0.009	0.008	0.001	0.001	0.000	0.001	0.001	0.001	0.005	0.003	0.000	0.000
Al	0.132	0.143	0.135	0.111	0.114	0.117	0.111	0.120	0.119	0.125	0.122	0.106	0.109
Fe	0.070	0.063	0.063	0.062	0.060	0.061	0.060	0.062	0.062	0.062	0.062	0.067	0.066
Mn	0.003	0.002	0.003	0.002	0.001	0.002	0.001	0.001	0.002	0.001	0.001	0.002	0.002
Mg	0.894	0.869	0.889	0.902	0.906	0.899	0.898	0.905	0.909	0.914	0.910	0.916	0.907
Ca	0.834	0.850	0.852	0.849	0.863	0.854	0.879	0.875	0.869	0.870	0.850	0.881	0.882
Na	0.089	0.088	0.080	0.078	0.077	0.077	0.067	0.062	0.068	0.061	0.073	0.065	0.062
Cr	0.045	0.050	0.043	0.032	0.030	0.032	0.031	0.032	0.032	0.031	0.031	0.034	0.036
Total	4.011	4.006	4.007	4.002	4.010	4.002	4.005	4.006	4.010	4.009	4.005	4.017	4.011
Mg#	92.75	93.24	93.38	93.52	93.78	93.66	93.71	93.59	93.66	93.67	93.59	93.20	93.21
Molar percentage of endmembers													
Wo	46.40	47.71	47.22	46.82	47.20	47.09	47.82	47.51	47.26	47.14	46.66	47.27	47.54
En	49.71	48.76	49.29	49.74	49.52	49.55	48.89	49.13	49.40	49.51	49.92	49.14	48.89
Fs	3.89	3.53	3.50	3.44	3.29	3.36	3.28	3.36	3.35	3.35	3.42	3.58	3.56

Table A1.1. Major element compositions of ultramafic xenolith clinopyroxenes continued.

Sample	OB93-307					OB93-309			OB93-310				
Grain ID	cpx1	cpx2	cpx3	cpx5s	cpx9s	cpx10s	cpx1	cpx2	cpx6s	cpx2	cpx3	cpx5s.pt1	cpx5s.pt2
wt. %													
SiO ₂	53.22	53.39	53.28	53.31	53.71	54.36	54.16	54.09	53.63	53.32	53.30	53.65	53.41
TiO ₂	0.00	0.00	0.00	0.00	0.01	0.00	0.05	0.08	0.06	0.00	0.00	0.00	0.00
Al ₂ O ₃	2.38	2.77	2.81	2.44	2.23	1.95	2.36	2.26	2.28	2.28	2.59	1.99	2.71
FeO*	1.97	1.84	1.85	1.95	2.01	2.07	2.08	2.11	2.03	2.07	1.98	2.39	2.48
MnO	0.08	0.06	0.07	0.06	0.06	0.09	0.06	0.05	0.08	0.11	0.11	0.07	0.09
MgO	17.84	17.48	17.26	17.60	17.81	18.00	16.71	17.08	17.25	17.84	17.76	18.75	18.47
CaO	23.72	23.62	23.70	23.55	23.89	22.92	22.09	22.69	22.77	23.42	23.72	22.50	22.38
Na ₂ O	0.10	0.19	0.10	0.08	0.07	0.45	1.17	0.93	0.82	0.00	0.00	0.00	0.00
Cr ₂ O ₃	0.89	1.01	1.05	0.88	0.84	0.68	1.26	1.28	1.21	0.77	0.88	0.55	1.06
Total	100.20	100.36	100.11	99.87	100.63	100.50	99.94	100.57	100.12	99.81	100.33	99.89	100.60
Cations/6 Oxygens													
Si	1.929	1.930	1.931	1.936	1.938	1.958	1.963	1.952	1.945	1.938	1.928	1.944	1.924
Ti	0.000	0.000	0.000	0.000	0.000	0.000	0.001	0.002	0.002	0.000	0.000	0.000	0.000
Al	0.102	0.118	0.120	0.105	0.095	0.083	0.101	0.096	0.098	0.098	0.110	0.085	0.115
Fe	0.060	0.056	0.056	0.059	0.061	0.062	0.063	0.064	0.061	0.063	0.060	0.072	0.075
Mn	0.002	0.002	0.002	0.002	0.002	0.003	0.002	0.002	0.002	0.003	0.003	0.002	0.003
Mg	0.964	0.941	0.932	0.952	0.957	0.966	0.902	0.918	0.932	0.966	0.958	1.013	0.992
Ca	0.921	0.915	0.920	0.917	0.923	0.885	0.858	0.877	0.885	0.912	0.919	0.874	0.864
Na	0.007	0.014	0.007	0.006	0.005	0.031	0.082	0.065	0.057	0.000	0.000	0.000	0.000
Cr	0.025	0.029	0.030	0.025	0.024	0.019	0.036	0.037	0.035	0.022	0.025	0.016	0.030
Total	4.011	4.004	3.998	4.002	4.005	4.007	4.009	4.012	4.016	4.002	4.004	4.005	4.003
Mg#	94.15	94.42	94.33	94.15	94.04	93.95	93.46	93.51	93.82	93.90	94.13	93.33	92.99
Molar percentage of endmembers													
Wo	47.36	47.85	48.22	47.54	47.56	46.23	47.05	47.17	47.10	46.98	47.47	44.61	44.76
En	49.56	49.24	48.84	49.40	49.32	50.51	49.48	49.40	49.63	49.78	49.45	51.70	51.37
Fs	3.08	2.91	2.94	3.07	3.12	3.25	3.46	3.43	3.27	3.23	3.09	3.69	3.87

Table A1.1. Major element compositions of ultramafic xenolith clinopyroxenes continued.

Sample	OB93-314					OB93-317			
Grain ID	cpx1.pt1	cpx1.pt2	cpx5s.pt1	cpx5s.pt2		cpx1	cpx4s	cpx7.pt1	cpx7.pt2
wt. %									
SiO ₂	54.31	53.94	54.04	54.34		54.34	53.85	53.37	53.21
TiO ₂	0.03	0.00	0.02	0.01		0.05	0.07	0.03	0.03
Al ₂ O ₃	2.43	2.17	2.44	2.44		2.65	2.76	2.59	2.62
FeO*	1.98	1.87	1.83	1.95		1.98	1.95	2.11	2.12
MnO	0.06	0.09	0.04	0.10		0.11	0.10	0.06	0.06
MgO	17.61	17.68	17.45	17.72		17.06	16.91	16.93	17.00
CaO	23.31	23.25	23.65	22.07		22.43	22.48	22.28	21.80
Na ₂ O	0.37	0.41	0.44	0.34		0.92	0.73	0.99	0.96
Cr ₂ O ₃	1.02	0.98	1.15	1.13		1.47	1.27	1.44	1.52
Total	101.11	100.38	101.05	100.10		101.02	100.12	99.81	99.30
Cations/6 Oxygens									
Si	1.946	1.948	1.940	1.959		1.949	1.948	1.941	1.943
Ti	0.001	0.000	0.000	0.000		0.001	0.002	0.001	0.001
Al	0.103	0.092	0.103	0.104		0.112	0.118	0.111	0.113
Fe	0.059	0.056	0.055	0.059		0.059	0.059	0.064	0.065
Mn	0.002	0.003	0.001	0.003		0.003	0.003	0.002	0.002
Mg	0.940	0.951	0.934	0.952		0.912	0.911	0.918	0.925
Ca	0.895	0.900	0.910	0.853		0.862	0.871	0.868	0.853
Na	0.025	0.028	0.030	0.024		0.064	0.051	0.070	0.068
Cr	0.029	0.028	0.033	0.032		0.042	0.036	0.041	0.044
Total	4.000	4.006	4.007	3.985		4.005	3.999	4.017	4.012
Mg#	94.07	94.40	94.45	94.20		93.88	93.91	93.46	93.46
Molar percentage of endmembers									
Wo	47.24	47.16	47.93	45.76		47.01	47.30	46.93	46.28
En	49.63	49.88	49.18	51.09		49.74	49.49	49.61	50.20
Fs	3.13	2.96	2.89	3.15		3.24	3.21	3.47	3.51

Table A1.2. Major element compositions of ultramafic xenolith orthopyroxenes. Each analysis is a separate grain, unless noted (e.g. pt1, pt2, etc).

Lac Supérieur														
Lherzolites														
Sample		OB93-52			OB93-51			OB93-79						
Grain ID		opx1	opx1.5	opx2	cpx3 ex	opx8	opx2	opx6	opx5	by cpx4	by cpx5	opx4	opx5	opx5
wt. %														
SiO ₂		56.14	56.17	55.68	56.34	56.90	56.92	57.01	57.43	57.53	57.46	57.66	57.61	57.61
TiO ₂		0.21	0.22	0.27	0.17	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Al ₂ O ₃		2.78	2.96	3.40	3.07	1.92	1.77	1.94	1.69	1.68	1.93	1.78	1.78	1.74
FeO*		6.25	6.32	6.40	6.11	5.11	5.05	5.02	5.04	4.91	5.02	4.89	5.11	5.11
MnO		0.14	0.20	0.17	0.16	0.16	0.14	0.14	0.13	0.13	0.13	0.13	0.13	0.14
MgO		34.10	33.77	33.59	33.04	34.70	34.86	34.99	35.13	35.39	35.11	34.84	35.28	35.28
CaO		0.42	0.44	0.50	1.04	0.64	0.62	0.69	0.60	0.63	6641.00	0.71	0.63	0.63
Na ₂ O		0.04	0.00	0.04	0.10	0.10	0.12	0.00	0.08	0.03	0.01	0.03	0.03	0.03
Cr ₂ O ₃		0.36	0.37	0.51	0.44	0.51	0.43	0.61	0.52	0.45	0.51	0.54	0.46	0.46
Total		100.43	100.45	100.55	100.47	100.03	99.90	100.39	100.62	100.75	6741.16	100.58	101.00	101.00
Cations/6 Oxygens														
Si		1.930	1.930	1.914	1.936	1.954	1.957	1.950	1.959	1.959	0.047	1.965	1.958	1.958
Ti		0.005	0.006	0.007	0.004	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Al		0.113	0.120	0.138	0.125	0.078	0.072	0.078	0.068	0.068	0.002	0.072	0.070	0.070
Fe		0.180	0.181	0.184	0.176	0.147	0.145	0.144	0.144	0.140	0.003	0.140	0.145	0.145
Mn		0.004	0.006	0.005	0.005	0.005	0.004	0.004	0.004	0.004	0.000	0.004	0.004	0.004
Mg		1.746	1.730	1.721	1.692	1.776	1.786	1.784	1.786	1.795	0.043	1.770	1.787	1.787
Ca		0.016	0.016	0.018	0.038	0.024	0.023	0.025	0.022	0.023	5.856	0.026	0.023	0.023
Na		0.003	0.000	0.002	0.006	0.007	0.008	0.000	0.005	0.002	0.000	0.002	0.002	0.002
Cr		0.010	0.010	0.014	0.012	0.014	0.012	0.017	0.014	0.012	0.000	0.015	0.012	0.012
Total		4.006	3.999	4.004	3.994	4.003	4.006	4.002	4.002	4.002	5.952	3.992	4.002	4.002
Mg#		90.68	90.50	90.34	90.60	92.37	92.48	92.56	92.55	92.78	92.57	92.69	92.48	92.48
Molar percentage of endmembers														
Wo		0.80	0.85	0.95	2.00	1.21	1.17	1.29	1.12	1.17	99.21	1.33	1.17	1.17
En		89.76	89.47	89.26	88.57	91.02	91.21	91.17	91.33	91.51	0.73	91.28	91.21	91.21
Fs		9.44	9.68	9.79	9.43	7.76	7.62	7.54	7.55	7.32	0.06	7.39	7.62	7.62
Each column represents one analysis. Different analyses of the same grain are noted in "Grain ID". The suffix "ex" = opx exsolution from cpx														

Each column represents one analysis. Different analyses of the same grain are noted in "Grain ID". The suffix "ex" = opx exsolution from cpx

Table A1.2. Major element compositions of ultramafic xenolith orthopyroxenes continued.

Sample	OB93-79	OB93-80						OB93-81c				
Grain ID	opx6	opx5	opx3, pt1	opx3, pt2	opx4, pt1	opx4, pt2	by cpx5	opx6	opx3	opx4	opx6	opx1
wt. %												
SiO ₂	57.51	56.79	57.04	57.20	57.13	57.19	56.95	56.69	57.32	57.61	57.70	57.27
TiO ₂	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Al ₂ O ₃	1.75	1.70	1.79	1.82	1.81	1.83	2.05	1.82	1.72	1.73	1.62	1.64
FeO*	4.90	4.98	4.95	4.91	4.98	4.81	4.86	4.99	4.98	5.04	5.06	5.04
MnO	0.17	0.13	0.16	0.12	0.15	0.13	0.16	0.13	0.15	0.14	0.17	0.16
MgO	35.22	35.02	35.13	35.31	34.67	34.77	34.96	35.14	35.24	35.60	35.53	35.43
CaO	0.70	0.68	0.61	0.68	0.91	0.86	0.68	0.72	0.77	0.64	0.66	0.62
Na ₂ O	0.03	0.00	0.00	0.00	0.00	0.02	0.03	0.03	0.04	0.02	0.05	0.00
Cr ₂ O ₃	0.47	0.47	0.53	0.55	0.49	0.52	0.62	0.53	0.50	0.50	0.47	0.46
Total	100.74	99.77	100.22	100.58	100.14	100.14	100.31	100.03	100.72	101.28	101.26	100.62
Cations/6 Oxygens												
Si	1.959	1.955	1.954	1.952	1.959	1.959	1.949	1.947	1.955	1.953	1.957	1.955
Ti	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Al	0.070	0.069	0.072	0.073	0.073	0.074	0.083	0.074	0.069	0.069	0.065	0.066
Fe	0.139	0.143	0.142	0.140	0.143	0.138	0.139	0.143	0.142	0.143	0.144	0.144
Mn	0.005	0.004	0.005	0.004	0.004	0.004	0.005	0.004	0.004	0.004	0.005	0.005
Mg	1.788	1.796	1.793	1.796	1.772	1.775	1.783	1.799	1.791	1.799	1.796	1.802
Ca	0.026	0.025	0.022	0.025	0.033	0.032	0.025	0.027	0.028	0.023	0.024	0.023
Na	0.002	0.000	0.000	0.000	0.000	0.001	0.002	0.002	0.002	0.001	0.004	0.000
Cr	0.013	0.013	0.014	0.015	0.013	0.014	0.017	0.014	0.014	0.013	0.013	0.013
Total	4.001	4.005	4.003	4.004	3.998	3.997	4.002	4.009	4.005	4.006	4.006	4.006
Mg#	92.77	92.62	92.67	92.76	92.55	92.79	92.76	92.62	92.65	92.64	92.60	92.61
Molar percentage of endmembers												
Wo	1.31	1.27	1.14	1.26	1.71	1.62	1.27	1.34	1.42	1.18	1.21	1.15
En	91.32	91.26	91.39	91.42	90.76	91.11	91.36	91.20	91.13	91.35	91.25	91.32
Fs	7.37	7.47	7.46	7.32	7.54	7.27	7.37	7.45	7.45	7.46	7.53	7.53

Table A1.2. Major element compositions of ultramafic xenolith orthopyroxenes continued.

Sample	OB93-81c			OB93-83							OB93-85		
Grain ID	opx2	opx7		opx1_pt1	opx1_pt2	opx2	opx3	opx5	opx7	opx9ex	opx1	opx2	opx3
wt. %													
SiO ₂	57.60	57.07		56.25	55.96	55.75	55.37	56.80	56.63	55.55	57.70	57.52	57.86
TiO ₂	0.00	0.00		0.03	0.00	0.03	0.05	0.01	0.00	0.02	0.00	0.00	0.00
Al ₂ O ₃	1.67	1.97		3.42	3.07	4.15	3.79	2.12	2.74	3.98	1.84	1.77	1.66
FeO*	5.11	5.11		5.39	5.54	6.05	5.37	5.39	5.46	5.60	5.14	4.89	5.03
MnO	0.16	0.11		0.13	0.15	0.17	0.14	0.17	0.15	0.18	0.14	0.10	0.14
MgO	35.40	35.12		33.57	33.41	32.72	33.67	34.60	33.86	32.79	35.25	35.37	35.50
CaO	0.73	0.70		0.99	0.86	0.93	0.97	0.80	0.80	1.02	0.68	0.96	0.60
Na ₂ O	0.00	0.03		0.28	0.14	0.10	0.15	0.13	0.12	0.12	0.00	0.03	0.00
Cr ₂ O ₃	0.48	0.63		0.81	0.73	0.75	0.73	0.53	0.51	1.02	0.51	0.49	0.64
Total	101.15	100.75		100.88	99.85	100.64	100.24	100.54	100.26	100.28	101.25	101.12	101.43
Cations/6 Oxygens													
Si	1.956	1.947		1.923	1.932	1.913	1.905	1.945	1.943	1.912	1.957	1.953	1.958
Ti	0.000	0.000		0.001	0.000	0.001	0.001	0.000	0.000	0.000	0.000	0.000	0.000
Al	0.067	0.079		0.138	0.125	0.168	0.154	0.085	0.111	0.162	0.073	0.071	0.066
Fe	0.145	0.146		0.154	0.160	0.174	0.154	0.154	0.157	0.161	0.146	0.139	0.142
Mn	0.005	0.003		0.004	0.004	0.005	0.004	0.005	0.004	0.005	0.004	0.003	0.004
Mg	1.792	1.786		1.710	1.719	1.674	1.727	1.766	1.732	1.682	1.781	1.790	1.791
Ca	0.027	0.026		0.036	0.032	0.034	0.036	0.029	0.029	0.038	0.025	0.035	0.022
Na	0.000	0.002		0.019	0.009	0.007	0.010	0.008	0.008	0.008	0.000	0.002	0.000
Cr	0.013	0.017		0.022	0.020	0.020	0.020	0.014	0.014	0.028	0.014	0.013	0.017
Total	4.004	4.006		4.006	4.001	3.995	4.012	4.009	3.998	3.997	4.000	4.006	4.000
Mg#	92.51	92.45		91.73	91.49	90.60	91.79	91.97	91.71	91.26	92.44	92.80	92.63
Molar percentage of endmembers													
Wo	1.36	1.31		1.90	1.66	1.81	1.86	1.50	1.53	2.00	1.25	1.77	1.10
En	91.04	91.10		89.81	89.76	88.72	89.88	90.35	90.10	89.19	91.09	91.02	91.43
Fs	7.61	7.60		8.29	8.58	9.47	8.26	8.15	8.37	8.81	7.66	7.20	7.47

Table A1.2. Major element compositions of ultramafic xenolith orthopyroxenes continued.

Sample Grain ID	OB93-85		OB93-99				Dunites OB93-82		Pyroxenites OB93-94		
	opx5	opx6	opx1	opx13	opx14	opx3	opx4	opx8	opx2ex	opx4ex	opx3ex
wt. %											
SiO ₂	57.41	56.35	56.77	56.35	57.18	56.39	56.59	56.90	56.22	56.32	56.84
TiO ₂	0.00	0.00	0.04	0.06	0.07	0.09	0.05	0.06	0.12	0.10	0.07
Al ₂ O ₃	1.73	2.22	1.51	1.93	1.46	1.96	1.81	1.72	2.78	2.88	2.64
FeO*	5.02	5.09	4.99	5.07	4.94	4.96	4.95	5.06	6.67	6.86	6.66
MnO	0.12	0.14	0.14	0.10	0.16	0.12	0.12	0.14	0.21	0.19	0.18
MgO	35.26	34.61	35.29	34.96	35.27	34.72	35.04	34.98	33.32	33.42	33.42
CaO	0.61	1.09	0.59	0.58	0.57	0.73	0.58	0.59	0.80	0.70	0.70
Na ₂ O	0.00	0.03	0.05	0.04	0.03	0.05	0.08	0.04	0.05	0.05	0.02
Cr ₂ O ₃	0.50	0.72	0.52	0.63	0.46	0.62	0.62	0.58	0.78	0.67	0.67
Total	100.65	100.25	99.88	99.71	100.13	99.62	99.84	100.06	100.95	101.17	101.19
<i>Cations/6 Oxygens</i>											
Si	1.957	1.936	1.953	1.943	1.960	1.945	1.947	1.954	1.930	1.929	1.943
Ti	0.000	0.000	0.001	0.001	0.002	0.002	0.001	0.001	0.003	0.002	0.002
Al	0.070	0.090	0.061	0.078	0.059	0.080	0.074	0.070	0.112	0.116	0.106
Fe	0.143	0.146	0.143	0.146	0.141	0.143	0.142	0.145	0.192	0.196	0.190
Mn	0.003	0.004	0.004	0.003	0.005	0.003	0.003	0.004	0.006	0.005	0.005
Mg	1.792	1.772	1.809	1.796	1.801	1.785	1.797	1.790	1.705	1.706	1.702
Ca	0.022	0.040	0.022	0.022	0.021	0.027	0.021	0.022	0.029	0.026	0.026
Na	0.000	0.002	0.004	0.002	0.002	0.003	0.006	0.002	0.003	0.003	0.002
Cr	0.013	0.020	0.014	0.017	0.012	0.017	0.017	0.016	0.021	0.018	0.018
Total	4.001	4.010	4.011	4.009	4.004	4.006	4.009	4.004	4.002	4.003	3.994
Mg#	92.60	92.38	92.65	92.48	92.72	92.58	92.66	92.49	89.90	89.68	89.94
<i>Molar percentage of endmembers</i>											
Wo	1.14	2.04	1.10	1.09	1.07	1.39	1.09	1.10	1.52	1.32	1.33
En	91.38	90.30	91.44	91.32	91.52	91.14	91.49	91.28	88.25	88.24	88.51
Fs	7.48	7.66	7.46	7.59	7.42	7.48	7.42	7.62	10.23	10.44	10.16

Table A1.2. Major element compositions of ultramafic xenolith orthopyroxenes continued.

Mont Trapeze													
Sample	OB93-94	OB93-281					OB98-282						
Grain ID	opx3ex	opx1ex	opx4.5	opx4.5	opx5	opx6.pt1	opx6.pt2	opx6ex	opx3ex	opx2.pt1	opx2.pt2	opx3.pt1	
wt. %													
SiO ₂	55.50	58.02	57.47	57.55	57.79	57.42	57.59	57.40	57.49	57.72	57.59	57.85	57.85
TiO ₂	0.12	0.00	0.00	0.00	0.00	0.00	0.02	0.00	0.00	0.03	0.03	0.00	0.00
Al ₂ O ₃	2.84	1.95	1.96	1.91	2.08	1.93	1.94	2.07	2.09	1.69	1.77	1.52	1.52
FeO*	6.70	5.11	5.32	5.40	5.39	5.47	5.45	5.19	5.40	5.01	5.05	5.10	5.10
MnO	0.19	0.15	0.12	0.15	0.12	0.14	0.11	0.13	0.13	0.17	0.11	0.17	0.17
MgO	33.30	34.75	34.42	34.58	34.32	34.64	34.46	34.41	34.50	34.68	34.84	34.85	34.85
CaO	0.71	0.72	0.55	0.53	0.69	0.53	0.58	0.72	0.62	0.60	0.59	0.56	0.56
Na ₂ O	0.04	0.08	0.04	0.00	0.04	0.03	0.02	0.02	0.02	0.06	0.05	0.17	0.17
Cr ₂ O ₃	0.73	0.51	0.59	0.51	0.58	0.49	0.54	0.59	0.60	0.50	0.51	0.46	0.46
Total	100.13	101.30	100.46	100.63	101.02	100.63	100.70	100.54	100.84	100.46	100.52	100.69	100.69
Cations/6 Oxygens													
Si	1.922	1.965	1.964	1.964	1.965	1.960	1.964	1.960	1.959	1.970	1.965	1.972	1.972
Ti	0.003	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.001	0.001	0.000	0.000
Al	0.116	0.078	0.079	0.077	0.084	0.078	0.078	0.083	0.084	0.068	0.071	0.061	0.061
Fe	0.194	0.145	0.152	0.154	0.153	0.156	0.155	0.148	0.154	0.143	0.144	0.145	0.145
Mn	0.006	0.004	0.003	0.004	0.003	0.004	0.003	0.004	0.004	0.005	0.003	0.005	0.005
Mg	1.718	1.754	1.753	1.759	1.739	1.762	1.751	1.751	1.752	1.764	1.772	1.770	1.770
Ca	0.026	0.026	0.020	0.019	0.025	0.019	0.021	0.026	0.023	0.022	0.021	0.020	0.020
Na	0.002	0.005	0.003	0.000	0.003	0.002	0.001	0.001	0.001	0.004	0.003	0.011	0.011
Cr	0.020	0.014	0.016	0.014	0.016	0.013	0.015	0.016	0.016	0.014	0.014	0.012	0.012
Total	4.008	3.992	3.990	3.991	3.987	3.995	3.990	3.991	3.992	3.990	3.994	3.997	3.997
Mg#	89.85	92.38	92.02	91.94	91.90	91.87	91.85	92.19	91.93	92.50	92.48	92.41	92.41
Molar percentage of endmembers													
Wo	1.36	1.36	1.04	0.99	1.32	0.99	1.10	1.37	1.17	1.14	1.10	1.05	1.05
En	88.37	90.92	90.90	90.82	90.53	90.77	90.69	90.75	90.68	91.21	91.31	91.20	91.20
Fs	10.27	7.72	8.06	8.19	8.16	8.24	8.21	7.88	8.16	7.65	7.58	7.74	7.74

Table A1.2. Major element compositions of ultramafic xenolith orthopyroxenes continued.

Sample	OB98-282						OB93-284						OB93-287					
	Grain ID	opx3.pl2	opx5	opx6	opx7.pl1	opx7.pl2	opx1.pl1	opx1.pl2	opx2.pl1	opx2.pl2	opx1.pl1	opx1.pl2	opx1.pl1	opx1.pl2	opx1.pl1	opx1.pl2	opx2	opx2
wt. %																		
SiO ₂		58.22	58.24	58.36	58.07	41.99	56.38	56.64	56.72	56.50	57.25	57.17	57.25	57.17	57.25	57.17	55.75	55.75
TiO ₂		0.03	0.04	0.03	0.05	0.00	0.00	0.00	0.00	0.00	0.02	0.01	0.02	0.01	0.02	0.01	0.04	0.04
Al ₂ O ₃		1.55	1.53	1.54	1.51	0.00	2.68	2.40	3.42	3.25	2.40	2.38	2.40	2.38	2.40	2.38	2.98	2.98
FeO*		5.14	5.01	5.17	5.09	8.18	5.38	5.30	5.45	5.32	5.45	5.43	5.45	5.43	5.45	5.43	5.58	5.58
MnO		0.11	0.16	0.12	0.15	0.18	0.11	0.13	0.18	0.18	0.12	0.13	0.12	0.13	0.12	0.13	0.14	0.14
MgO		34.75	34.91	34.95	34.77	50.29	33.87	33.92	34.06	33.96	34.07	34.07	34.07	34.07	34.07	34.07	33.52	33.52
CaO		0.53	0.50	0.54	0.50	0.05	0.82	0.91	0.85	0.91	0.78	0.96	0.78	0.96	0.78	0.96	0.87	0.87
Na ₂ O		0.11	0.04	0.13	0.09	0.00	0.03	0.03	0.03	0.05	0.02	0.04	0.02	0.04	0.02	0.04	0.06	0.06
Cr ₂ O ₃		0.49	0.42	0.38	0.47	0.03	0.67	0.65	0.77	0.74	0.59	0.66	0.59	0.66	0.59	0.66	1.89	1.89
Total		100.92	100.84	101.21	100.69	100.72	99.92	99.98	101.48	100.91	100.67	100.84	100.67	100.84	100.67	100.84	100.84	100.84
<i>Cations/6 Oxygens</i>																		
Si		1.977	1.978	1.977	1.977	1.518	1.941	1.949	1.925	1.928	1.955	1.951	1.955	1.951	1.955	1.951	1.913	1.913
Ti		0.001	0.001	0.001	0.001	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.001	0.001
Al		0.062	0.061	0.061	0.061	0.000	0.109	0.097	0.137	0.131	0.097	0.096	0.097	0.096	0.097	0.096	0.121	0.121
Fe		0.146	0.142	0.146	0.145	0.247	0.155	0.152	0.155	0.152	0.156	0.155	0.156	0.155	0.156	0.155	0.160	0.160
Mn		0.003	0.005	0.004	0.004	0.006	0.003	0.004	0.005	0.005	0.003	0.004	0.003	0.004	0.003	0.004	0.004	0.004
Mg		1.759	1.767	1.764	1.764	2.709	1.738	1.739	1.722	1.727	1.734	1.732	1.734	1.732	1.734	1.732	1.714	1.714
Ca		0.019	0.018	0.020	0.018	0.002	0.030	0.034	0.031	0.033	0.028	0.035	0.028	0.035	0.028	0.035	0.032	0.032
Na		0.007	0.002	0.008	0.006	0.000	0.002	0.002	0.002	0.003	0.001	0.002	0.001	0.002	0.001	0.002	0.004	0.004
Cr		0.013	0.011	0.010	0.013	0.001	0.018	0.018	0.021	0.020	0.016	0.018	0.016	0.018	0.016	0.018	0.051	0.051
Total		3.988	3.986	3.991	3.988	4.482	3.997	3.995	3.997	3.999	3.989	3.993	3.989	3.993	3.989	3.993	4.001	4.001
Mg#		92.34	92.55	92.34	92.41	91.64	91.82	91.94	91.76	91.91	91.77	91.79	91.77	91.79	91.77	91.79	91.45	91.45
<i>Molar percentage of endmembers</i>																		
Wo		1.00	0.95	1.01	0.94	0.06	1.57	1.75	1.61	1.74	1.49	1.82	1.49	1.82	1.49	1.82	1.68	1.68
En		91.26	91.45	91.24	91.34	91.41	90.38	90.33	90.28	90.32	90.40	90.12	90.40	90.12	90.40	90.12	89.91	89.91
Fs		7.73	7.60	7.75	7.72	8.53	8.05	7.92	8.10	7.95	8.11	8.06	8.11	8.06	8.11	8.06	8.40	8.40

Table A1.2. Major element compositions of ultramafic xenolith orthopyroxenes continued.

Sample	OB93-289		OB93-291		OB93-297			OB93-302		OB93-305	
	Grain ID	wt. %	opx1	opx2	opx1,pt1	opx1,pt2	opx2,pt2	opx1	opx2	opx1,pt1	opx1,pt2
SiO ₂	56.24	56.24	56.91	56.92	57.10	56.79	56.86	56.33	56.09	56.91	57.01
TiO ₂	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.03	0.01	0.00	0.00
Al ₂ O ₃	2.53	2.58	2.33	2.33	2.29	2.25	2.46	2.06	2.35	3.26	3.22
FeO*	5.93	5.96	5.29	5.36	5.43	5.31	5.30	5.19	5.27	5.42	5.11
MnO	0.13	0.17	0.18	0.18	0.12	0.16	0.15	0.09	0.11	0.16	0.14
MgO	33.15	33.31	34.05	33.61	33.47	33.98	34.20	34.29	33.90	33.82	34.24
CaO	1.18	1.31	0.57	1.06	1.18	0.90	0.55	0.59	0.76	0.82	0.75
Na ₂ O	0.00	0.00	0.07	0.09	0.06	0.05	0.03	0.07	0.07	0.05	0.05
Cr ₂ O ₃	0.67	0.68	0.55	0.67	0.78	0.73	0.68	0.59	0.62	0.72	0.71
Total	99.82	100.23	99.95	100.23	100.43	100.18	100.22	99.23	99.17	101.14	101.24
<i>Cations/6 Oxygens</i>											
Si	1.945	1.939	1.956	1.955	1.957	1.951	1.949	1.951	1.946	1.935	1.934
Ti	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.001	0.000	0.000	0.000
Al	0.103	0.105	0.094	0.094	0.093	0.091	0.099	0.084	0.096	0.131	0.129
Fe	0.172	0.172	0.152	0.154	0.156	0.152	0.152	0.150	0.153	0.154	0.145
Mn	0.004	0.005	0.005	0.005	0.004	0.005	0.004	0.003	0.003	0.004	0.004
Mg	1.708	1.712	1.744	1.720	1.710	1.740	1.747	1.770	1.752	1.714	1.732
Ca	0.044	0.048	0.021	0.039	0.043	0.033	0.020	0.022	0.028	0.030	0.027
Na	0.000	0.000	0.005	0.006	0.004	0.003	0.002	0.004	0.005	0.003	0.003
Cr	0.018	0.018	0.015	0.018	0.021	0.020	0.018	0.016	0.017	0.019	0.019
Total	3.994	3.999	3.992	3.992	3.988	3.995	3.992	4.001	4.000	3.991	3.994
Mg#	90.88	90.88	91.98	91.78	91.65	91.94	92.00	92.17	91.98	91.75	92.27
<i>Molar percentage of endmembers</i>											
Wo	2.28	2.50	1.10	2.04	2.28	1.72	1.05	1.12	1.46	1.56	1.43
En	88.80	88.61	90.97	89.91	89.56	90.36	91.04	91.14	90.64	90.32	90.95
Fs	8.91	8.89	7.93	8.05	8.16	7.92	7.91	7.74	7.91	8.12	7.62

Table A1.2. Major element compositions of ultramafic xenolith orthopyroxenes continued.

Sample	OB93-305		OB93-307		OB93-309		OB93-310				
	Grain ID	opx2,pt1	opx2,pt2	opx1,pt1	opx1,pt2	opx1,pt2	opx2,pt1	opx2,pt2	opx1	opx2,pt1	opx2,pt2
wt. %											
SiO ₂		57.24	57.08	56.86	56.81	56.72	57.67	57.40	57.08	56.95	56.91
TiO ₂		0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Al ₂ O ₃		3.06	3.10	3.53	3.53	3.62	2.60	2.62	2.56	3.43	3.24
FeO*		5.08	5.11	5.10	5.27	5.19	5.34	5.31	5.14	5.34	5.11
MnO		0.14	0.11	0.17	0.09	0.17	0.12	0.12	0.16	0.17	0.13
MgO		33.87	33.95	33.85	33.38	33.40	34.50	34.33	34.43	34.05	33.81
CaO		1.17	1.04	0.65	1.42	1.29	0.84	0.72	0.69	0.97	0.82
Na ₂ O		0.04	0.00	0.00	0.00	0.00	0.05	0.02	0.03	0.00	0.00
Cr ₂ O ₃		0.71	0.68	0.74	0.74	0.79	0.54	0.60	0.59	0.68	0.63
Total		101.32	101.07	100.89	101.23	101.18	101.59	101.11	100.68	101.60	100.65
Cations/6 Oxygens											
Si		1.942	1.940	1.934	1.931	1.929	1.951	1.950	1.947	1.929	1.941
Ti		0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Al		0.122	0.124	0.141	0.141	0.145	0.100	0.103	0.103	0.137	0.130
Fe		0.144	0.145	0.145	0.150	0.148	0.151	0.151	0.147	0.151	0.146
Mn		0.004	0.003	0.005	0.003	0.005	0.005	0.003	0.005	0.005	0.004
Mg		1.712	1.720	1.716	1.691	1.693	1.740	1.738	1.750	1.718	1.718
Ca		0.043	0.038	0.024	0.052	0.047	0.030	0.026	0.025	0.035	0.030
Na		0.003	0.000	0.000	0.000	0.000	0.003	0.001	0.002	0.000	0.000
Cr		0.019	0.018	0.020	0.020	0.021	0.015	0.016	0.016	0.018	0.017
Total		3.989	3.989	3.985	3.988	3.988	3.994	3.990	3.995	3.994	3.986
Mg#		92.24	92.21	92.21	91.86	91.98	92.00	91.99	92.02	91.91	92.18
Molar percentage of endmembers											
Wo		2.24	1.99	1.26	2.74	2.49	1.59	1.39	1.36	1.86	1.58
En		90.17	90.37	91.05	89.35	89.69	90.55	90.70	90.77	90.20	90.72
Fs		7.59	7.64	7.70	7.91	7.82	7.87	7.90	7.87	7.94	7.70

Table A1.2. Orthopyroxene major elements continued.

Sample	OB93-314		OB93-317	
	Grain ID	opx2.pt1	opx2.pt2	opx1
wt. %				
SiO ₂	56.95	57.43	56.95	57.15
TiO ₂	0.00	0.00	0.00	0.00
Al ₂ O ₃	2.86	2.88	2.53	2.46
FeO*	5.30	5.27	5.17	5.24
MnO	0.13	0.15	0.13	0.15
MgO	34.30	34.55	34.78	34.34
CaO	0.66	0.67	0.67	0.66
Na ₂ O	0.00	0.00	0.03	0.05
Cr ₂ O ₃	0.62	0.67	0.74	0.71
Total	100.84	101.62	101.00	100.75
<i>Cations/6 Oxygens</i>				
Si	1.941	1.941	1.939	1.949
Ti	0.000	0.000	0.000	0.000
Al	0.115	0.115	0.101	0.099
Fe	0.151	0.149	0.147	0.149
Mn	0.004	0.004	0.004	0.004
Mg	1.742	1.740	1.764	1.745
Ca	0.024	0.024	0.024	0.024
Na	0.000	0.000	0.002	0.003
Cr	0.017	0.018	0.020	0.019
Total	3.993	3.992	4.002	3.993
Mg#	92.02	92.12	92.31	92.11
<i>Molar percentage of endmembers</i>				
Wo	1.26	1.27	1.26	1.25
En	90.86	90.94	91.14	90.96
Fs	7.88	7.78	7.60	7.79

Table A1.3. Major element compositions of ultramafic xenolith olivines. Each analysis is a separate grain, unless noted (e.g. pt1, pt2, etc).

Harzburgites											
Sample		OB93-52					OB93-51				
Grain ID		ol3	ol2	ol6	ol7	ol5	ol8,pt1	ol8,pt2	ol2	ol4	ol5
wt. %											
SiO ₂		40.36	40.50	40.65	40.57	40.62	41.00	40.69	40.96	41.06	41.04
FeO		9.57	9.76	9.95	9.93	9.75	8.10	8.28	8.25	8.14	8.14
MnO		0.15	0.11	0.10	0.14	0.13	0.09	0.11	0.13	0.12	0.11
MgO		49.32	49.69	49.44	49.43	49.42	50.74	51.08	50.93	50.74	50.77
CaO		0.03	0.03	0.03	0.03	0.03	0.06	0.06	0.07	0.07	0.07
NiO		0.34	0.31	0.33	0.33	0.35	0.33	0.32	0.35	0.33	0.31
Al ₂ O ₃		0.00	0.02	0.00	0.02	0.02	0.00	0.03	0.03	0.00	0.02
Cr ₂ O ₃		0.01	0.00	0.00	0.00	0.00	0.00	0.04	0.03	0.00	0.00
Total		99.78	100.42	100.50	100.44	100.31	100.32	100.61	100.75	100.44	100.46
Cations/4 Oxygens											
Si		0.992	0.990	0.993	0.992	0.994	0.995	0.987	0.991	0.996	0.995
Fe		0.197	0.200	0.203	0.203	0.200	0.164	0.168	0.167	0.165	0.165
Mn		0.003	0.002	0.002	0.003	0.003	0.002	0.002	0.003	0.002	0.002
Mg		1.807	1.810	1.800	1.802	1.802	1.835	1.846	1.837	1.833	1.834
Ca		0.001	0.001	0.001	0.001	0.001	0.002	0.002	0.002	0.002	0.002
Ni		0.007	0.006	0.006	0.007	0.007	0.006	0.006	0.007	0.006	0.006
Al		0.000	0.001	0.000	0.000	0.000	0.000	0.001	0.001	0.000	0.001
Cr		0.000	0.000	0.000	0.000	0.000	0.000	0.001	0.001	0.000	0.000
Total		3.007	3.010	3.007	3.008	3.006	3.005	3.012	3.008	3.004	3.005
Fo		90.18	90.07	89.85	89.87	90.03	91.78	91.67	91.67	91.74	91.75

Table A1.3. Major element compositions of ultramafic xenolith olivines continued.

Sample	OB93-79				OB93-80				OB93-81c			
	ol3	ol2	ol5	ol4	ol6	ol7	ol1	ol4	ol1.pt1	ol1.pt2	ol5	ol6
wt. %												
SiO ₂	41.43	41.20	40.97	40.96	41.22	41.15	41.27	40.88	41.03	40.90	40.91	40.83
FeO	8.00	8.23	8.28	8.02	8.05	7.96	7.96	8.24	8.57	8.29	8.37	8.21
MnO	0.11	0.14	0.11	0.12	0.11	0.13	0.09	0.11	0.06	0.12	0.07	0.11
MgO	50.70	51.22	51.20	50.94	50.90	50.82	50.25	50.96	51.21	51.41	51.52	51.25
CaO	0.09	0.06	0.06	0.07	0.07	0.07	0.13	0.08	0.06	0.05	0.05	0.05
NiO	0.32	0.33	0.34	0.31	0.33	0.35	0.30	0.30	0.36	0.33	0.36	0.35
Al ₂ O ₃	0.00	0.00	0.00	0.02	0.03	0.03	0.04	0.02	0.00	0.00	0.00	0.00
Cr ₂ O ₃	0.03	0.01	0.00	0.00	0.01	0.00	0.22	0.00	0.02	0.00	0.00	0.00
Total	100.68	101.20	100.97	100.44	100.73	100.51	100.25	100.59	101.30	101.10	101.27	100.80
<i>Cations/4 Oxygens</i>												
Si	1.001	0.992	0.990	0.993	0.996	0.996	1.001	0.991	0.989	0.987	0.986	0.988
Fe	0.162	0.166	0.167	0.163	0.163	0.161	0.161	0.167	0.173	0.167	0.169	0.166
Mn	0.002	0.003	0.002	0.002	0.002	0.003	0.002	0.002	0.001	0.003	0.002	0.002
Mg	1.825	1.838	1.843	1.840	1.833	1.834	1.817	1.841	1.840	1.849	1.850	1.848
Ca	0.002	0.002	0.002	0.002	0.002	0.002	0.003	0.002	0.001	0.001	0.001	0.001
Ni	0.006	0.006	0.007	0.006	0.006	0.007	0.006	0.006	0.007	0.006	0.007	0.007
Al	0.000	0.000	0.000	0.001	0.001	0.001	0.001	0.000	0.000	0.000	0.000	0.000
Cr	0.001	0.000	0.000	0.000	0.000	0.000	0.004	0.000	0.000	0.000	0.000	0.000
Total	2.999	3.008	3.010	3.007	3.003	3.003	2.996	3.009	3.011	3.013	3.014	3.012
Fo	91.86	91.73	91.68	91.88	91.84	91.92	91.84	91.68	91.42	91.70	91.64	91.75

Table A1.3. Major element compositions of ultramafic xenolith olivines continued.

Sample Grain ID	OB93-83				OB93-85				OB93-99			
	ol7	ol8	ol2	ol9	ol1	ol2	ol4	ol3	ol3	ol10	ol9	ol7
wt. %												
SiO ₂	40.93	40.91	40.41	40.66	40.76	40.72	40.86	40.94	41.55	40.99	40.77	41.12
FeO	8.80	8.79	10.12	9.10	8.31	8.12	8.41	8.14	8.11	7.96	8.25	8.17
MnO	0.13	0.12	0.13	0.11	0.13	0.12	0.13	0.13	0.13	0.14	0.08	0.09
MgO	50.35	50.59	49.21	50.36	51.08	51.03	50.95	51.38	49.98	50.61	50.91	50.86
CaO	0.09	0.08	0.10	0.09	0.06	0.10	0.09	0.07	0.06	0.04	0.04	0.03
NiO	0.33	0.32	0.36	0.35	0.35	0.31	0.34	0.32	0.35	0.31	0.34	0.34
Al ₂ O ₃	0.02	0.00	0.03	0.03	0.03	0.03	0.02	0.03	0.03	0.03	0.00	0.02
Cr ₂ O ₃	0.00	0.00	0.01	0.03	0.00	0.06	0.02	0.05	0.04	0.00	0.00	0.00
Total	100.66	100.81	100.37	100.73	100.72	100.48	100.80	101.06	100.23	100.07	100.39	100.63
<i>Cations/4 Oxygens</i>												
Si	0.994	0.992	0.991	0.988	0.988	0.988	0.989	0.988	1.008	0.997	0.990	0.995
Fe	0.179	0.178	0.207	0.185	0.168	0.165	0.170	0.164	0.164	0.162	0.168	0.165
Mn	0.003	0.002	0.003	0.002	0.003	0.002	0.003	0.003	0.003	0.003	0.002	0.002
Mg	1.822	1.828	1.797	1.825	1.844	1.845	1.838	1.847	1.807	1.834	1.843	1.834
Ca	0.002	0.002	0.003	0.002	0.002	0.002	0.002	0.002	0.001	0.001	0.001	0.001
Ni	0.006	0.006	0.007	0.007	0.007	0.006	0.007	0.006	0.007	0.006	0.007	0.007
Al	0.001	0.000	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.000	0.000
Cr	0.000	0.000	0.000	0.001	0.000	0.001	0.000	0.001	0.001	0.000	0.000	0.000
Total	3.006	3.008	3.009	3.011	3.012	3.011	3.010	3.012	2.991	3.003	3.010	3.005
Fo	91.07	91.11	89.65	90.79	91.64	91.80	91.52	91.83	91.66	91.89	91.67	91.73

Table A1.3. Major element compositions of ultramafic xenolith olivines continued.

Dunites												
Sample	OB93-61c				OB93-64				OB93-82			
Grain ID	ol5	ol7	ol8	ol9	ol3	ol4	ol2	ol5	ol0.pt1	ol0.pt2	ol3	ol1
wt. %												
SiO ₂	40.55	40.93	40.81	40.46	40.40	40.55	40.50	40.47	40.30	40.49	40.35	40.41
FeO	9.61	10.01	9.93	9.93	10.43	10.21	10.48	10.54	11.14	11.19	11.43	11.21
MnO	0.16	0.17	0.17	0.17	0.14	0.12	0.14	0.15	0.17	0.20	0.13	0.17
MgO	49.36	49.40	49.55	49.50	49.38	49.31	49.61	49.37	48.25	48.63	48.60	48.68
CaO	0.08	0.07	0.06	0.07	0.10	0.09	0.09	0.08	0.04	0.04	0.04	0.04
NiO	0.29	0.29	0.33	0.29	0.30	0.31	0.33	0.26	0.34	0.32	0.36	0.37
Al ₂ O ₃	0.00	0.04	0.00	0.00	0.00	0.00	0.02	0.00	0.04	0.02	0.00	0.00
Cr ₂ O ₃	0.00	0.02	0.02	0.03	0.10	0.00	0.00	0.02	0.02	0.00	0.00	0.00
Total	100.04	100.92	100.88	100.44	100.85	100.58	101.17	100.88	100.29	100.89	100.92	100.88
<i>Cations/4 Oxygens</i>												
Si	0.994	0.996	0.994	0.990	0.987	0.992	0.987	0.988	0.993	0.992	0.989	0.990
Fe	0.197	0.204	0.202	0.203	0.213	0.209	0.213	0.215	0.229	0.229	0.234	0.230
Mn	0.003	0.003	0.004	0.004	0.003	0.002	0.003	0.003	0.004	0.004	0.003	0.003
Mg	1.803	1.791	1.798	1.805	1.798	1.797	1.801	1.797	1.772	1.775	1.776	1.778
Ca	0.002	0.002	0.002	0.002	0.002	0.002	0.002	0.002	0.001	0.001	0.001	0.001
Ni	0.006	0.006	0.007	0.006	0.006	0.006	0.006	0.005	0.007	0.006	0.007	0.007
Al	0.000	0.001	0.000	0.000	0.000	0.000	0.001	0.000	0.001	0.001	0.000	0.000
Cr	0.000	0.000	0.000	0.001	0.002	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Total	3.006	3.003	3.006	3.010	3.012	3.008	3.013	3.011	3.006	3.008	3.011	3.010
Fo	90.15	89.79	89.89	89.88	89.40	89.59	89.41	89.30	88.53	88.56	88.34	88.56

Table A1.3. Major element compositions of ultramafic xenolith olivines continued.

Mont Trapeze											
Harzburgites											
Sample Grain ID	OB93-281					OB93-282					
	ol1	ol2,pt1	ol2,pt2	ol4	ol6	ol2	ol2.5	ol3,pt1	ol3,pt2	ol4	ol6
wt. %											
SiO ₂	40.61	40.17	40.26	40.35	40.64	40.77	40.89	40.80	40.50	40.67	40.78
FeO	8.00	8.64	8.77	9.69	8.11	7.57	7.81	7.84	7.95	7.53	7.76
MnO	0.13	0.17	0.15	0.15	0.13	0.11	0.09	0.13	0.15	0.09	0.12
MgO	50.24	50.30	50.36	49.16	50.36	50.29	50.83	50.58	50.35	50.56	50.32
CaO	0.08	0.04	0.05	0.08	0.03	0.05	0.06	0.04	0.05	0.05	0.06
NiO	0.33	0.30	0.30	0.28	0.31	0.34	0.29	0.31	0.30	0.32	0.29
Al ₂ O ₃	0.04	0.02	0.02	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Cr ₂ O ₃	0.03	0.01	0.01	0.12	0.02	0.00	0.03	0.08	0.12	0.02	0.02
Total	99.46	99.65	99.92	99.83	99.60	99.13	99.99	99.77	99.43	99.24	99.35
Cations/4 Oxygens											
Si	0.994	0.986	0.986	0.992	0.994	0.999	0.994	0.995	0.992	0.996	0.998
Fe	0.164	0.177	0.180	0.199	0.166	0.155	0.159	0.160	0.163	0.154	0.159
Mn	0.003	0.004	0.003	0.003	0.003	0.002	0.002	0.003	0.003	0.002	0.003
Mg	1.833	1.840	1.838	1.802	1.836	1.836	1.843	1.838	1.839	1.845	1.835
Ca	0.002	0.001	0.001	0.002	0.001	0.001	0.001	0.001	0.001	0.001	0.002
Ni	0.007	0.006	0.006	0.006	0.006	0.007	0.006	0.006	0.006	0.006	0.006
Al	0.001	0.001	0.001	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Cr	0.001	0.000	0.000	0.002	0.000	0.000	0.000	0.002	0.002	0.000	0.000
Total	3.005	3.014	3.014	3.006	3.006	3.001	3.005	3.004	3.007	3.004	3.002
Fo	91.80	91.21	91.09	90.04	91.71	92.21	92.07	91.99	91.86	92.28	92.04

Table A1.3. Major element compositions of ultramafic xenolith olivines continued.

Sample Grain ID	OB93-284				OB93-287				OB93-289			
	ol1.pt1	ol1.pt2	ol2.pt2	ol2.pt2	ol1.pt1	ol1.pt2	ol2.pt2	ol2.pt2	ol1.pt1	ol1.pt2	ol2.pt2	ol2.pt2
wt. %												
SiO ₂	40.97	40.89	40.89	41.04	40.67	40.55	40.55	40.70	40.70	40.61	39.96	40.72
FeO	8.58	8.57	8.58	8.44	8.58	8.48	8.57	8.70	8.93	9.05	10.27	10.43
MnO	0.09	0.16	0.10	0.09	0.08	0.10	0.14	0.10	0.10	0.11	0.12	0.11
MgO	49.66	49.75	49.68	49.77	49.66	49.58	49.90	49.74	49.97	50.15	48.38	48.67
CaO	0.09	0.06	0.09	0.06	0.04	0.06	0.04	0.06	0.03	0.03	0.06	0.05
NiO	0.50	0.50	0.45	0.52	0.45	0.52	0.45	0.50	0.53	0.49	0.49	0.50
Al ₂ O ₃	0.02	0.03	0.05	0.07	0.02	0.02	0.03	0.03	0.00	0.02	0.02	0.03
Cr ₂ O ₃	0.00	0.00	0.00	0.00	0.02	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Total	99.90	99.94	99.85	99.99	99.51	99.31	99.67	99.84	100.26	100.47	99.30	100.51
<i>Cations/4 Oxygens</i>												
Si	1.001	0.999	1.000	1.001	0.998	0.997	0.994	0.996	0.993	0.990	0.992	0.997
Fe	0.175	0.175	0.175	0.172	0.176	0.174	0.176	0.178	0.182	0.185	0.213	0.214
Mn	0.002	0.003	0.002	0.002	0.002	0.002	0.003	0.002	0.002	0.002	0.003	0.002
Mg	1.808	1.812	1.810	1.809	1.816	1.817	1.823	1.814	1.818	1.822	1.789	1.777
Ca	0.002	0.002	0.002	0.002	0.001	0.002	0.001	0.001	0.001	0.001	0.002	0.001
Ni	0.010	0.010	0.009	0.010	0.009	0.010	0.009	0.010	0.010	0.010	0.010	0.010
Al	0.001	0.001	0.002	0.002	0.001	0.001	0.001	0.001	0.000	0.001	0.001	0.001
Cr	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Total	2.999	3.001	3.000	2.998	3.002	3.003	3.006	3.003	3.007	3.010	3.008	3.002
Fo	91.16	91.19	91.16	91.31	91.16	91.24	91.21	91.06	90.89	90.81	89.36	89.26

Table A1.3. Major element compositions of ultramafic xenolith olivines continued.

Sample Grain ID	OB93-291				OB93-297				OB93-302			
	ol1,pt1	ol1,pt2	ol2,pt2	ol2,pt2	ol1,pt1	ol1,pt2	ol2,pt2	ol2,pt2	ol1,pt1	ol1,pt2	ol2,pt2	ol2,pt2
wt. %												
SiO ₂	40.76	40.69	40.83	40.17	40.93	40.97	40.38	40.70	41.05	41.04	40.97	41.02
FeO	8.74	8.39	8.57	8.60	8.81	8.72	9.04	8.57	8.41	8.57	8.27	8.35
MnO	0.09	0.13	0.12	0.13	0.10	0.12	0.13	0.12	0.12	0.09	0.11	0.14
MgO	50.09	50.12	50.10	50.16	49.73	49.90	50.20	49.81	49.99	50.25	49.98	50.14
CaO	0.06	0.07	0.04	0.03	0.05	0.04	0.03	0.04	0.03	0.03	0.04	0.03
NiO	0.44	0.51	0.47	0.53	0.57	0.47	0.57	0.55	0.46	0.50	0.51	0.47
Al ₂ O ₃	0.06	0.02	0.05	0.03	0.02	0.02	0.04	0.00	0.15	0.21	0.09	0.14
Cr ₂ O ₃	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.02	0.02	0.00	0.00
Total	100.24	99.93	100.17	99.64	100.21	100.23	100.39	99.79	100.22	100.70	99.96	100.27
<i>Cations/4 Oxygens</i>												
Si	0.994	0.994	0.995	0.986	0.998	0.998	0.986	0.996	0.999	0.995	0.999	0.997
Fe	0.178	0.171	0.175	0.177	0.180	0.178	0.185	0.175	0.171	0.174	0.169	0.170
Mn	0.002	0.003	0.002	0.003	0.002	0.002	0.003	0.002	0.003	0.002	0.002	0.003
Mg	1.820	1.825	1.820	1.836	1.808	1.812	1.827	1.817	1.812	1.815	1.816	1.817
Ca	0.002	0.002	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001
Ni	0.009	0.010	0.009	0.010	0.011	0.009	0.011	0.011	0.009	0.010	0.010	0.009
Al	0.002	0.001	0.001	0.001	0.001	0.001	0.001	0.000	0.004	0.006	0.003	0.004
Cr	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Total	3.005	3.006	3.004	3.013	3.001	3.001	3.013	3.004	2.999	3.002	3.000	3.001
Fo	91.08	91.41	91.24	91.22	90.95	91.07	90.82	91.20	91.38	91.27	91.50	91.46

Table A1.3. Major element compositions of ultramafic xenolith olivines continued.

Sample Grain ID	OB93-310				OB93-314				OB93-317			
	ol1.pt1	ol1.pt2	ol2.pt2	ol2.pt2	ol1.pt1	ol1.pt2	ol2.pt2	ol2.pt2	ol1.pt1	ol1.pt2	ol2.pt2	ol2.pt2
wt. %												
SiO ₂	40.89	40.60	40.85	40.81	40.83	40.63	40.93	40.89	41.06	41.05	40.74	41.01
FeO	8.51	8.61	8.49	8.75	8.30	8.40	8.35	8.46	7.98	8.24	8.18	8.04
MnO	0.12	0.12	0.12	0.14	0.09	0.12	0.10	0.11	0.09	0.10	0.08	0.10
MgO	49.54	49.69	49.26	49.23	50.23	50.27	50.43	50.02	50.35	50.58	49.75	49.99
CaO	0.10	0.10	0.07	0.09	0.05	0.05	0.04	0.03	0.02	0.05	0.02	0.03
NiO	0.50	0.56	0.49	0.49	0.52	0.50	0.49	0.51	0.57	0.59	0.50	0.52
Al ₂ O ₃	0.03	0.02	0.02	0.02	0.02	0.03	0.03	0.03	0.03	0.00	0.00	0.02
Cr ₂ O ₃	0.00	0.00	0.00	0.00	0.01	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Total	99.69	99.68	99.30	99.53	100.05	99.99	100.37	100.05	100.10	100.60	99.26	99.71
<i>Cations/4 Oxygens</i>												
Si	1.001	0.995	1.004	1.002	0.996	0.992	0.995	0.997	0.999	0.995	1.000	1.001
Fe	0.174	0.176	0.174	0.180	0.169	0.172	0.170	0.173	0.162	0.167	0.168	0.164
Mn	0.002	0.002	0.002	0.003	0.002	0.002	0.002	0.002	0.002	0.002	0.002	0.002
Mg	1.807	1.816	1.804	1.801	1.825	1.829	1.827	1.818	1.825	1.828	1.820	1.819
Ca	0.003	0.002	0.002	0.002	0.001	0.001	0.001	0.001	0.001	0.001	0.000	0.001
Ni	0.010	0.011	0.010	0.010	0.010	0.010	0.010	0.010	0.011	0.011	0.010	0.010
Al	0.001	0.001	0.001	0.001	0.000	0.001	0.001	0.001	0.001	0.000	0.000	0.001
Cr	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Total	2.998	3.004	2.996	2.998	3.004	3.007	3.005	3.002	3.001	3.005	3.000	2.998
Fo	91.21	91.14	91.18	90.93	91.52	91.43	91.50	91.33	91.84	91.62	91.55	91.72

Table A1.4. Major element compositions of ultramafic xenolith spinels. Each analysis is a separate grain, unless noted (e.g. pt1, pt2, etc).

Lac Superfeur														
Lherzolite		Harzburgites												
		OB93-52				OB93-51				OB93-78		OB93-79		
Grain ID		sp0	sp2 n=2	sp6 n=2	sp1.5	sp5 n=4	sp5	sp8	sp4 n=5	sp3 n=5	sp1	sp3 n=3	sp4 n=2	sp2 n=2
wt. %														
SiO ₂		0.04	0.01	0.00	0.02	0.01	0.08	0.08	0.06	0.07	0.00	0.07	0.08	0.07
TiO ₂		0.59	0.48	0.52	0.55	0.53	0.09	0.07	0.08	0.08	0.05	0.06	0.06	0.05
Al ₂ O ₃		42.54	42.99	43.31	42.86	42.41	24.95	26.71	27.25	26.52	26.79	27.37	27.37	27.08
Cr ₂ O ₃		22.20	21.54	21.48	21.27	22.05	42.65	40.41	40.78	40.62	39.16	39.92	39.87	40.09
FeO*		14.56	14.23	14.20	14.01	14.21	14.72	13.98	14.02	14.32	15.20	13.35	13.49	13.42
MnO		0.12	0.09	0.07	0.08	0.09	0.13	0.11	0.14	0.13	0.26	0.16	0.16	0.17
MgO		18.99	18.89	19.20	19.13	18.88	17.18	17.48	17.40	17.34	17.29	17.68	17.86	17.64
CaO		0.00	0.01	0.01	0.00	0.01	0.05	0.03	0.02	0.06	0.00	0.05	0.01	0.03
NiO		0.25	0.26	0.25	0.28	0.26	0.20	0.17	0.18	0.18	0.27	0.16	0.22	0.18
Total		99.29	98.50	99.04	98.19	98.44	100.03	99.03	99.92	99.31	99.02	98.81	99.12	98.73
FeO		9.78	9.63	9.45	9.24	9.58	9.63	9.17	9.62	9.37	9.16	8.80	8.63	8.79
Fe ₂ O ₃		5.32	5.11	5.29	5.30	5.14	5.66	5.35	4.90	5.50	6.71	5.06	5.40	5.15
New Total		99.82	99.02	99.57	98.72	98.96	100.60	99.57	100.41	99.86	99.69	99.32	99.66	99.25
Cations/4 Oxygens														
Si		0.001	0.000	0.000	0.001	0.000	0.002	0.002	0.002	0.002	0.000	0.002	0.002	0.002
Ti		0.012	0.010	0.011	0.012	0.011	0.002	0.002	0.002	0.002	0.001	0.001	0.001	0.001
Al		1.400	1.421	1.422	1.419	1.405	0.883	0.944	0.954	0.937	0.951	0.965	0.962	0.956
Cr		0.490	0.478	0.473	0.472	0.490	1.012	0.958	0.957	0.963	0.933	0.944	0.940	0.950
Fe		0.340	0.334	0.331	0.329	0.334	0.370	0.351	0.348	0.359	0.383	0.334	0.336	0.336
Mn		0.003	0.002	0.002	0.002	0.002	0.003	0.003	0.003	0.003	0.007	0.004	0.004	0.004
Mg		0.790	0.790	0.797	0.801	0.791	0.769	0.781	0.770	0.775	0.776	0.788	0.794	0.788
Ca		0.000	0.000	0.000	0.000	0.000	0.001	0.001	0.001	0.002	0.000	0.002	0.000	0.001
Ni		0.006	0.006	0.006	0.006	0.006	0.005	0.004	0.004	0.004	0.007	0.004	0.005	0.004
Cat. Tot.		3.042	3.040	3.042	3.042	3.041	3.048	3.045	3.041	3.046	3.057	3.043	3.045	3.044
O Tot.		3.945	3.947	3.945	3.945	3.946	3.937	3.941	3.946	3.939	3.925	3.944	3.940	3.943
Fe ²⁺		0.225	0.223	0.217	0.214	0.222	0.238	0.226	0.236	0.231	0.226	0.217	0.212	0.217
Fe ³⁺		0.110	0.106	0.109	0.110	0.107	0.126	0.119	0.108	0.122	0.149	0.112	0.119	0.114
New Cat. Tot.		3.037	3.036	3.037	3.037	3.036	3.042	3.040	3.036	3.041	3.050	3.038	3.040	3.039
New O Tot.		4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001
Cr#		25.93	25.16	24.96	24.98	25.86	53.41	50.38	50.09	50.68	49.52	49.46	49.43	49.83
Mg#Fe ²⁺		77.83	78.00	78.60	78.90	78.06	76.37	77.53	76.57	77.01	77.42	78.40	78.93	78.40

Each column represents one analysis, except where average analysis is noted by n=x, and x=number of averaged points.
 Different analyses of the same grain are noted in "Grain ID". The suffix "s" = symplectite spinel; and "disc." = discrete spinel.

Table A1.4. Major element compositions of spinels continued.

Sample Grain ID	OB93-79			OB93-80			OB93-81c			OB93-83		
	sp5 n=2	sp7 n=2	sp6 n=2	sp1 n=6	sp3 n=4	sp5 n=2	sp6 n=2	sp8	sp6 n=3	sp1.pt1	sp1.pt2	sp1.pt3 sp2.pt1
wt. %												
SiO ₂	0.07	0.07	0.06	0.13	0.06	0.10	0.06	0.08	0.06	0.00	0.00	0.00
TiO ₂	0.07	0.06	0.05	0.06	0.07	0.08	0.10	0.08	0.12	0.12	0.11	0.15
Al ₂ O ₃	27.15	27.04	27.42	26.87	27.32	26.94	27.23	27.56	26.81	34.72	33.21	28.39
Cr ₂ O ₃	39.97	40.36	39.83	40.70	41.11	40.68	40.70	40.76	42.31	29.91	30.62	36.52
FeO*	13.29	13.47	13.33	13.69	13.12	13.92	13.72	13.83	13.40	15.00	15.23	16.18
MnO	0.16	0.17	0.13	0.12	0.15	0.16	0.13	0.13	0.14	0.23	0.22	0.26
MgO	17.70	17.63	17.85	17.30	17.57	17.27	17.56	17.10	17.12	19.22	18.44	17.76
CaO	0.01	0.05	0.00	0.10	0.02	0.16	0.05	0.10	0.04	0.06	0.04	0.00
NiO	0.17	0.19	0.20	0.15	0.19	0.19	0.20	0.19	0.16	0.31	0.34	0.37
Total	98.59		98.87	99.13	99.62	99.50	99.74	99.84	100.16	99.56	98.20	99.66
FeO	8.71	8.85	8.61	9.46	9.22	9.47	9.29	10.00	10.06	7.62	8.15	8.83
Fe ₂ O ₃	5.09	5.13	5.25	4.70	4.33	4.95	4.93	4.25	3.71	8.19	7.87	8.16
New Total	99.10	99.55	99.40	99.60	100.05	99.99	100.24	100.26	100.53	100.38	98.99	100.48
<i>Cations/4 Oxygens</i>												
Si	0.002	0.002	0.002	0.004	0.002	0.003	0.002	0.002	0.002	0.000	0.000	0.000
Ti	0.002	0.001	0.001	0.001	0.002	0.002	0.002	0.002	0.003	0.003	0.002	0.003
Al	0.959	0.953	0.965	0.948	0.956	0.948	0.953	0.964	0.937	1.180	1.152	0.997
Cr	0.947	0.954	0.940	0.963	0.965	0.960	0.956	0.957	0.992	0.682	0.712	0.860
Fe	0.333	0.337	0.333	0.343	0.326	0.348	0.341	0.343	0.332	0.362	0.375	0.403
Mn	0.004	0.004	0.003	0.003	0.004	0.004	0.003	0.003	0.004	0.006	0.006	0.007
Mg	0.791	0.786	0.795	0.772	0.778	0.768	0.778	0.757	0.757	0.826	0.809	0.789
Ca	0.000	0.002	0.000	0.003	0.001	0.005	0.002	0.003	0.001	0.002	0.001	0.003
Ni	0.004	0.005	0.005	0.004	0.004	0.005	0.005	0.005	0.004	0.007	0.008	0.009
Cat. Tot.	3.043	3.043	3.044	3.040	3.036	3.042	3.041	3.036	3.031	3.067	3.065	3.073
O Tot.	3.943	3.943	3.942	3.948	3.952	3.945	3.946	3.953	3.959	3.913	3.915	3.904
Fe ²⁺	0.215	0.218	0.212	0.234	0.226	0.233	0.228	0.245	0.247	0.180	0.196	0.220
Fe ³⁺	0.113	0.114	0.116	0.104	0.096	0.110	0.109	0.094	0.082	0.174	0.171	0.191
New Cat. Tot.	3.038	3.038	3.039	3.035	3.032	3.037	3.037	3.032	3.028	3.059	3.057	3.063
New O Tot.	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001
Cr#	49.69	50.03	49.35	50.39	50.24	50.32	50.07	49.80	51.43	36.62	38.21	46.32
Mg#Fe ²⁺	78.61	78.26	78.95	76.76	77.46	76.72	77.36	75.51	75.41	82.12	80.48	78.57
												78.13

Table A1.4. Major element compositions of spinels continued.

Sample Grain ID	OB93-83						OB93-85		OB93-99				
	sp2,pt2	sp2,pt3	sp3 n=3	sp10, disc.	sp6,pt1	sp6,pt2	sp6,pt3	sp11,disc.	sp12,disc.	sp0 n=2	sp3 n=2	sp1 n=3	sp3 n=2
wt. %													
SiO ₂	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
TiO ₂	0.18	0.10	0.08	0.14	0.12	0.14	0.14	0.16	0.16	0.03	0.05	0.05	0.29
Al ₂ O ₃	28.72	41.49	41.52	21.46	37.21	31.35	35.04	22.53	21.22	26.01	26.84	27.55	25.71
Cr ₂ O ₃	35.24	22.68	23.42	42.69	27.30	32.64	29.26	42.22	42.69	43.73	42.43	42.00	42.92
FeO*	17.29	14.88	13.33	18.70	15.87	17.11	16.32	17.43	19.91	12.05	11.93	11.95	12.63
MnO	0.25	0.24	0.20	0.29	0.24	0.24	0.22	0.29	0.34	0.29	0.30	0.26	0.27
MgO	17.19	19.91	20.46	15.98	19.08	17.76	18.39	16.44	15.38	17.01	17.19	17.33	16.68
CaO	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.06	0.01
NiO	0.34	0.38	0.38	0.28	0.36	0.35	0.37	0.30	0.29	0.26	0.28	0.26	0.24
Total	99.20	99.67	99.39	99.53	100.18	99.60	99.73	99.37	99.99	99.40	99.03	99.45	98.76
FeO	9.75	7.65	6.73	10.57	8.46	9.39	9.04	9.97	11.54	9.48	9.21	9.23	10.00
Fe ₂ O ₃	8.37	8.04	7.33	9.04	8.24	8.59	8.09	8.29	9.29	2.86	3.03	3.01	2.92
New Total	100.04	100.47	100.12	100.44	101.00	100.46	100.54	100.20	100.92	99.68	99.33	99.75	99.05
Cations/4 Oxygens													
Si	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Ti	0.004	0.002	0.002	0.003	0.003	0.003	0.003	0.004	0.004	0.001	0.001	0.001	0.007
Al	1.015	1.368	1.365	0.786	1.248	1.090	1.193	0.819	0.778	0.916	0.945	0.963	0.914
Cr	0.835	0.502	0.517	1.049	0.614	0.761	0.669	1.029	1.051	1.034	1.002	0.985	1.023
Fe	0.434	0.348	0.311	0.486	0.378	0.422	0.394	0.449	0.518	0.301	0.298	0.296	0.319
Mn	0.006	0.006	0.005	0.008	0.006	0.006	0.005	0.008	0.009	0.007	0.008	0.007	0.007
Mg	0.768	0.830	0.851	0.740	0.809	0.781	0.792	0.756	0.714	0.758	0.765	0.766	0.750
Ca	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.002	0.000
Ni	0.008	0.009	0.009	0.007	0.008	0.008	0.009	0.007	0.007	0.006	0.007	0.006	0.006
Cat. Tot.	3.071	3.063	3.058	3.079	3.066	3.071	3.066	3.072	3.082	3.024	3.026	3.025	3.025
O Tot.	3.908	3.917	3.925	3.897	3.914	3.907	3.914	3.906	3.894	3.968	3.966	3.967	3.967
Fe ²⁺	0.239	0.175	0.154	0.268	0.197	0.226	0.214	0.251	0.293	0.235	0.228	0.227	0.250
Fe ³⁺	0.185	0.166	0.151	0.206	0.173	0.186	0.172	0.188	0.212	0.064	0.068	0.067	0.066
New Cat. Tot.	3.061	3.056	3.052	3.067	3.058	3.062	3.057	3.062	3.068	3.022	3.023	3.023	3.022
New O Tot.	4.001	4.001	4.000	4.001	4.001	4.001	4.001	4.001	4.002	4.001	4.001	4.001	4.001
Cr#	45.15	26.83	27.46	57.16	32.98	41.12	35.91	55.70	57.45	53.01	51.47	50.57	52.82
Mg#Fe ²⁺	76.28	82.57	84.67	73.44	80.43	77.55	78.75	75.06	70.93	76.33	77.05	77.13	74.98

Table A1.4. Major element compositions of spinels continued.

Sample Grain ID	Dunites					OB93-61c					OB93-64			OB93-82		
	OB93-99		OB93-99		wt. %	sp1	sp4	sp5	sp10	sp5	sp2	sp3	sp4	sp5	sp3ex	n=2
	sp1	sp0, disc.	n=2	n=2				n=5				n=2		n=2		
SiO ₂	0.00	0.00				0.03	0.01	0.03	0.02	0.03	0.04	0.03	0.03	0.04	3.97	
TiO ₂	0.31	0.30				0.60	0.65	0.60	0.57	0.58	0.60	0.61	0.59	0.97	0.59	
Al ₂ O ₃	24.74	23.59				31.08	30.05	30.97	31.38	25.11	24.31	24.30	24.89	36.24	36.27	
Cr ₂ O ₃	43.77	45.02				31.02	31.59	30.20	30.22	35.08	35.82	36.30	35.86	27.11	22.80	
FeO*	13.41	13.37				19.10	20.22	19.97	19.29	22.87	23.42	23.26	23.09	17.16	16.04	
MnO	0.29	0.31				0.18	0.17	0.21	0.17	0.21	0.25	0.24	0.24	0.14	0.13	
MgO	16.27	16.11				16.21	16.03	16.35	17.06	14.38	14.30	14.26	14.53	17.30	17.43	
CaO	0.01	0.01				0.03	0.01	0.01	0.00	0.00	0.00	0.00	0.01	0.01	2.09	
NiO	0.22	0.23				0.20	0.22	0.25	0.19	0.24	0.20	0.21	0.21	0.27	0.24	
Total	99.02	98.94				98.45	98.95	98.57	98.90	98.48	98.93	99.21	100.47	99.25	99.56	
FeO	10.57	10.57				11.97	12.29	11.75	10.91	13.83	14.00	14.14	13.87	11.67	13.87	
Fe ₂ O ₃	3.15	3.11				7.92	8.81	9.14	9.31	10.05	10.47	10.13	10.25	6.10	2.41	
New Total	99.33	99.25				99.24	99.83	99.49	99.83	99.49	99.98	100.22	100.47	99.86	99.80	
<i>Cations/4 Oxygens</i>																
Si	0.000	0.000				0.001	0.000	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.112	
Ti	0.007	0.007				0.013	0.015	0.014	0.013	0.014	0.014	0.014	0.014	0.021	0.012	
Al	0.883	0.848				1.100	1.067	1.098	1.103	0.926	0.897	0.895	0.911	1.236	1.205	
Cr	1.049	1.085				0.737	0.753	0.718	0.713	0.868	0.887	0.896	0.881	0.620	0.508	
Fe	0.340	0.341				0.480	0.510	0.502	0.481	0.598	0.614	0.607	0.600	0.415	0.378	
Mn	0.008	0.008				0.005	0.004	0.005	0.004	0.006	0.006	0.006	0.006	0.003	0.003	
Mg	0.735	0.732				0.726	0.720	0.733	0.759	0.671	0.668	0.664	0.673	0.746	0.732	
Ca	0.000	0.000				0.001	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.063	
Ni	0.005	0.006				0.005	0.005	0.006	0.005	0.006	0.005	0.005	0.005	0.006	0.005	
Cat. Tot.	3.027	3.027				3.067	3.075	3.078	3.078	3.089	3.093	3.089	3.090	3.050	3.019	
O Tot.	3.964	3.965				3.912	3.903	3.899	3.898	3.885	3.880	3.884	3.884	3.935	3.975	
Fe ²⁺	0.266	0.267				0.294	0.302	0.288	0.265	0.351	0.356	0.359	0.350	0.278	0.325	
Fe ³⁺	0.071	0.071				0.175	0.195	0.202	0.204	0.230	0.239	0.231	0.233	0.131	0.051	
New Cat. Tot.	3.024	3.024				3.057	3.063	3.065	3.066	3.072	3.074	3.072	3.072	3.043	3.017	
New O Tot.	4.001	4.001				4.002	4.002	4.002	4.001	4.002	4.002	4.002	4.002	4.001	4.001	
Cr#	54.27	56.15				40.12	41.36	39.55	39.25	48.38	49.71	50.05	49.15	33.42	29.66	
Mg#Fe ²⁺	73.46	73.27				71.15	70.45	71.78	74.09	65.61	65.25	64.92	65.80	72.87	69.26	

Table A1.4. Major element compositions of spinels continued.

Mt. Trapze												
Pyroxenites			Harzburgites									
Sample	OB93-94			OB93-284			OB93-287			OB93-289		
Grain ID	gr2 n=3	grn 3	sp10 n=2	sp0 n=2	sp7 n=2	sp11, disc.	sp12, disc.	sp4	sp8	sp6	sp11	sp12
wt. %												
SiO ₂	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
TiO ₂	0.60	0.57	0.13	0.11	0.11	0.29	0.30	0.31	0.33	0.09	0.10	0.11
Al ₂ O ₃	29.94	31.32	34.66	34.35	35.47	21.57	21.98	21.57	23.02	27.03	26.12	26.52
Cr ₂ O ₃	35.73	36.81	33.44	34.09	32.95	46.12	46.16	44.80	44.93	40.84	41.80	40.63
FeO*	15.64	10.77	12.71	12.75	12.44	15.23	15.61	15.81	15.81	15.48	15.36	17.24
MnO	0.28	0.24	0.24	0.20	0.24	0.30	0.34	0.28	0.30	0.30	0.29	0.27
MgO	16.61	20.29	18.55	18.42	18.64	15.39	15.45	16.09	15.60	15.89	15.96	15.07
CaO	0.00	0.01	0.00	0.00	0.00	0.01	0.00	0.03	0.05	0.01	0.02	0.02
NiO	0.28	0.19	0.34	0.32	0.35	0.25	0.26	0.23	0.27	0.23	0.28	0.30
Total	99.08	100.21	100.07	100.23	100.19	99.15	100.09	99.12	100.30	99.86	99.92	100.16
FeO	11.16	6.19	8.80	9.03	8.78	11.43	11.69	10.41	11.68	11.60	11.33	12.84
Fe ₂ O ₃	4.98	5.10	4.35	4.13	4.07	4.23	4.36	6.00	4.58	4.31	4.47	4.89
New Total	99.58	100.72	100.51	100.64	100.60	99.57	100.53	99.72	100.76	100.29	100.37	100.65
Cations/4 Oxygens												
Si	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Ti	0.013	0.012	0.003	0.002	0.002	0.007	0.007	0.007	0.007	0.002	0.002	0.002
Al	1.050	1.058	1.168	1.158	1.189	0.786	0.793	0.786	0.826	0.955	0.926	0.943
Cr	0.840	0.835	0.756	0.771	0.741	1.127	1.118	1.095	1.081	0.968	0.994	0.969
Fe	0.389	0.258	0.304	0.305	0.296	0.394	0.400	0.409	0.402	0.388	0.386	0.435
Mn	0.007	0.006	0.006	0.005	0.006	0.008	0.009	0.007	0.008	0.008	0.007	0.007
Mg	0.736	0.867	0.791	0.785	0.791	0.709	0.705	0.741	0.707	0.710	0.715	0.678
Ca	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.001	0.002	0.000	0.001	0.001
Ni	0.007	0.004	0.008	0.007	0.008	0.006	0.006	0.006	0.007	0.005	0.007	0.007
Cat. Tot.	3.042	3.041	3.035	3.033	3.033	3.037	3.038	3.052	3.039	3.036	3.038	3.042
O Tot.	3.945	3.946	3.954	3.956	3.957	3.951	3.950	3.931	3.948	3.952	3.950	3.945
Fe ²⁺	0.274	0.146	0.208	0.214	0.207	0.292	0.296	0.265	0.293	0.287	0.281	0.319
Fe ³⁺	0.110	0.108	0.092	0.088	0.086	0.097	0.099	0.137	0.104	0.096	0.100	0.109
New Cat. Tot.	3.036	3.038	3.032	3.030	3.029	3.032	3.033	3.045	3.034	3.032	3.033	3.036
New O Tot.	4.001	4.000	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001
Cr#	44.46	44.09	39.29	39.97	38.39	58.92	58.49	58.21	56.69	50.34	51.77	50.68
Mg#Fe ²⁺	72.90	85.56	79.17	78.61	79.29	70.85	70.47	73.70	70.69	71.18	71.76	67.97

Table A1.4. Major element compositions of spinels continued.

Sample Grain ID	OB93-291			OB93-297			OB93-302			OB93-305			OB93-307	
	sp7	sp6	sp11	sp1	sp2	sp11	sp8	sp9	sp10, disc.	sp4	sp3	sp1	sp8	n=2
wt. %														
SiO ₂	0.00	0.06	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
TiO ₂	0.19	0.20	0.23	0.53	0.67	0.63	0.10	0.09	0.12	0.14	0.14	0.17	0.04	0.04
Al ₂ O ₃	24.86	23.74	25.41	25.69	25.25	26.47	28.34	28.42	27.57	26.33	27.36	27.71	35.08	35.08
Cr ₂ O ₃	43.91	44.89	42.81	42.59	42.39	41.26	40.38	40.74	40.85	41.89	40.90	40.84	33.78	33.78
FeO*	14.54	14.64	14.98	14.81	15.74	15.18	13.29	12.69	13.78	14.80	14.44	14.53	11.49	11.49
MnO	0.28	0.27	0.31	0.28	0.32	0.28	0.29	0.24	0.28	0.26	0.28	0.28	0.25	0.25
MgO	16.29	16.19	15.93	16.50	15.99	16.32	17.14	17.30	16.49	16.08	16.40	16.47	18.46	18.46
CaO	0.00	0.07	0.01	0.00	0.00	0.01	0.03	0.04	0.02	0.03	0.02	0.03	0.02	0.02
NiO	0.25	0.25	0.25	0.25	0.28	0.25	0.27	0.27	0.23	0.23	0.27	0.23	0.30	0.30
Total	100.31	100.31	99.94	100.65	100.63	100.40	99.84	99.81	99.33	99.76	99.80	100.27	99.42	99.42
FeO	10.88	10.87	11.38	11.09	11.87	11.51	9.83	9.59	10.61	11.21	10.86	11.00	8.69	8.69
Fe ₂ O ₃	4.06	4.19	4.00	4.14	4.30	4.07	3.85	3.44	3.53	3.99	3.97	3.93	3.12	3.12
New Total	100.72	100.73	100.34	101.07	101.06	100.80	100.22	100.16	99.69	100.16	100.19	100.66	99.74	99.74
<i>Cations/4 Oxygens</i>														
Si	0.000	0.002	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Ti	0.004	0.005	0.005	0.012	0.015	0.014	0.002	0.002	0.003	0.003	0.003	0.004	0.001	0.001
Al	0.880	0.845	0.902	0.903	0.892	0.930	0.988	0.988	0.971	0.932	0.962	0.969	1.183	1.183
Cr	1.043	1.071	1.020	1.004	1.005	0.973	0.944	0.950	0.965	0.994	0.965	0.958	0.764	0.764
Fe	0.365	0.369	0.377	0.369	0.395	0.379	0.329	0.313	0.344	0.372	0.360	0.360	0.275	0.275
Mn	0.007	0.007	0.008	0.007	0.008	0.007	0.007	0.006	0.007	0.007	0.007	0.007	0.006	0.006
Mg	0.729	0.729	0.715	0.734	0.715	0.725	0.755	0.761	0.734	0.720	0.729	0.728	0.788	0.788
Ca	0.000	0.002	0.000	0.000	0.000	0.000	0.001	0.001	0.001	0.001	0.001	0.001	0.001	0.001
Ni	0.006	0.006	0.006	0.006	0.007	0.006	0.006	0.006	0.005	0.005	0.006	0.006	0.007	0.007
Cat. Tot.	3.034	3.036	3.034	3.035	3.036	3.034	3.032	3.029	3.030	3.034	3.033	3.033	3.025	3.025
O Tot.	3.955	3.953	3.955	3.954	3.952	3.955	3.958	3.962	3.961	3.955	3.956	3.957	3.967	3.967
Fe ²⁺	0.270	0.271	0.283	0.273	0.294	0.284	0.240	0.234	0.262	0.278	0.268	0.270	0.206	0.206
Fe ³⁺	0.091	0.094	0.090	0.092	0.096	0.090	0.085	0.076	0.079	0.089	0.088	0.087	0.067	0.067
New Cat. Tot.	3.030	3.031	3.030	3.031	3.032	3.030	3.029	3.026	3.026	3.030	3.029	3.029	3.023	3.023
New O Tot.	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.000	4.000
Ct#	54.23	55.92	53.06	52.66	52.97	51.11	48.87	49.02	49.85	51.62	50.07	49.72	39.24	39.24
Mg#Fe ³⁺	72.97	72.87	71.62	72.85	70.84	71.87	75.86	76.45	73.68	72.11	73.12	72.97	79.25	79.25

Table A1.4. Major element compositions of spinels continued.

Sample Grain ID	OB93-307			OB93-309			OB93-310			OB93-314		
	sp9 n=2	sp10 n=2	sp6 n=2	sp0 n=2	sp2 n=2	sp5 n=2	sp8 n=3	sp9 n=4	sp11 n=2	sp12 n=2	sp13 n=2	sp5 n=4
wt. %												
SiO ₂	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
TiO ₂	0.04	0.05	0.20	0.21	0.24	0.25	0.04	0.03	0.09	0.07	0.09	0.08
Al ₂ O ₃	36.18	36.32	20.82	21.76	21.22	25.93	33.70	32.57	29.06	29.95	28.41	31.24
Cr ₂ O ₃	33.43	33.00	46.72	45.40	46.64	41.25	34.96	36.34	40.60	39.49	40.88	38.23
FeO*	11.05	11.13	16.36	16.65	16.34	16.22	11.85	12.04	12.83	12.41	13.26	12.15
MnO	0.21	0.22	0.33	0.33	0.29	0.29	0.23	0.25	0.30	0.23	0.25	0.25
MgO	18.74	18.98	14.82	15.03	15.36	15.88	18.21	17.95	17.29	17.62	16.89	17.82
CaO	0.02	0.00	0.02	0.01	0.01	0.01	0.00	0.01	0.03	0.01	0.01	0.01
NiO	0.30	0.32	0.24	0.23	0.31	0.30	0.32	0.31	0.25	0.28	0.24	0.27
Total	99.95	100.03	99.51	99.62	100.42	100.12	99.31	99.49	100.43	100.06	100.04	100.04
FeO	8.64	8.32	12.18	12.09	11.75	11.62	8.86	9.10	9.90	9.44	10.35	9.35
Fe ₂ O ₃	2.67	3.12	4.64	5.06	5.09	5.12	3.32	3.26	3.26	3.29	3.23	3.11
New Total	100.22	100.34	99.97	100.13	100.93	100.63	99.65	99.82	100.76	100.39	100.37	100.35
<i>Cations/4 Oxygens</i>												
Si	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
Ti	0.001	0.001	0.005	0.005	0.006	0.006	0.001	0.001	0.002	0.002	0.002	0.002
Al	1.207	1.210	0.762	0.792	0.768	0.920	1.146	1.113	1.003	1.031	0.988	1.068
Cr	0.748	0.738	1.147	1.109	1.132	0.982	0.798	0.833	0.940	0.912	0.954	0.877
Fe	0.262	0.263	0.425	0.430	0.419	0.408	0.286	0.292	0.314	0.303	0.327	0.295
Mn	0.005	0.005	0.009	0.009	0.008	0.007	0.006	0.006	0.007	0.006	0.006	0.006
Mg	0.791	0.800	0.686	0.693	0.703	0.713	0.783	0.775	0.755	0.767	0.743	0.771
Ca	0.001	0.000	0.001	0.000	0.000	0.000	0.000	0.000	0.001	0.000	0.000	0.000
Ni	0.007	0.007	0.006	0.006	0.008	0.007	0.007	0.007	0.006	0.007	0.006	0.006
Cat. Tot.	3.021	3.025	3.041	3.044	3.044	3.043	3.027	3.027	3.027	3.027	3.027	3.025
O Tot.	3.972	3.967	3.947	3.942	3.942	3.943	3.964	3.965	3.964	3.964	3.964	3.966
Fe ²⁺	0.203	0.195	0.312	0.308	0.297	0.288	0.212	0.219	0.240	0.229	0.253	0.225
Fe ³⁺	0.057	0.066	0.107	0.116	0.116	0.114	0.072	0.071	0.071	0.072	0.071	0.067
New Cat. Tot.	3.020	3.023	3.035	3.038	3.038	3.038	3.025	3.024	3.024	3.024	3.024	3.023
New O Tot.	4.000	4.000	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001	4.001
Cr#	38.27	37.87	60.09	58.33	59.58	51.63	41.04	42.81	48.38	46.94	49.11	45.09
Mg#Fe ²⁺	79.56	80.39	68.73	69.22	70.27	71.20	78.70	78.00	75.85	77.04	74.59	77.41

Table A1.4. Major element compositions of spinels continued.

Sample Grain ID	OB93-317			
	sp1 n=2	sp2	sp7 n=2	sp5 n=2
wt. %				
SiO ₂	0.00	0.00	0.00	0.00
TiO ₂	0.15	0.14	0.11	0.13
Al ₂ O ₃	24.85	25.35	24.51	26.97
Cr ₂ O ₃	44.08	43.84	44.35	42.71
FeO*	13.92	13.59	14.51	13.03
MnO	0.29	0.29	0.29	0.26
MgO	16.26	16.61	16.03	16.94
CaO	0.02	0.00	0.01	0.01
NiO	0.25	0.25	0.27	0.24
Total	99.82	100.06	100.09	100.30
FeO	10.69	10.32	11.02	10.16
Fe ₂ O ₃	3.59	3.63	3.88	3.19
New Total	100.18	100.42	100.48	100.62
<i>Cations/4 Oxygens</i>				
Si	0.000	0.000	0.000	0.000
Ti	0.003	0.003	0.002	0.003
Al	0.882	0.895	0.871	0.941
Cr	1.050	1.038	1.058	1.000
Fe	0.351	0.340	0.366	0.323
Mn	0.007	0.007	0.007	0.007
Mg	0.730	0.741	0.721	0.748
Ca	0.001	0.000	0.000	0.000
Ni	0.006	0.006	0.006	0.006
Cat. Tot.	3.030	3.031	3.033	3.027
O Tot.	3.960	3.959	3.956	3.965
Fe ²⁺	0.267	0.256	0.275	0.249
Fe ³⁺	0.080	0.081	0.087	0.070
New Cat. Tot.	3.027	3.027	3.029	3.024
New O Tot.	4.001	4.001	4.001	4.001
Cr#	54.34	53.71	54.83	51.51
Mg#Fe ²⁺	73.25	74.35	72.39	74.99

Table A1.5 REE contents of clinopyroxenes

Lac Supérieur												
Sample grain	Lherzolites					Harzburgites						
	OB93-52					OB93-51						
	Notes	cpx1	cpx2	cpx3	cpx1	cpx2	cpx3s	cpx4s	OB93-53 avg n=3	1 S.D.	OB93-56 avg n=4	OB93-58 avg n=5
La	2.0	1.5	1.8	16.40	23.0	9.66	0.72	0.04	0.01	0.12	0.07	0.04
Ce	6.4	5.3	6.2	31.10	45.9	14.00	0.64	0.05	0.01	0.30	0.08	0.06
Nd	6.1	4.1	5.5	7.38	11.4	2.14	0.10	0.05	0.01	0.22	0.08	0.03
Sm	2.35	1.85	2.19	0.95	1.53	0.32	0.03	0.03	0.00	0.10	0.10	0.05
Eu	0.90	0.75	0.90	0.29	0.47	0.09	0.01	0.01	0.00	0.04	0.02	0.01
Dy	3.23	2.38	3.02	0.38	0.60	0.10	0.03	0.03	0.00	0.10	0.06	0.03
Er	1.97	1.47	1.83	0.16	0.27	0.11	0.03	0.04	0.00	0.07	0.08	0.02
Yb	2.05	1.63	2.02	0.17	0.22	0.10	0.09	0.08	0.02	0.12	0.09	0.05
Ti	5290	4789	4977	31.7	36	35.5	23	45	3	33	n.d	5
V	287	274	292	114	114	157	97.8	131	3	114	n.d	6
Cr	6093	5383	7629	4364	4200	4267	3134	4511	312	4003	n.d	574
Sr	126	133	121	322	413	273	3.7	1.1	0.05	3.2	n.d	0.5
Y	17	16	17.2	1.6	2.2	0.5	0	0.2	0.1	0.4	n.d	0.1
Zr	29.8	26.5	29.5	2.9	8.9	1.3	0.4	0.7	0.4	0.9	n.d	0.1
Normalized												
La	8.56	6.48	7.80	69.88	98.00	41.16	3.06	0.19		0.51	0.31	
Ce	10.63	8.72	10.23	51.56	76.09	23.21	1.07	0.09		0.50	0.14	
Nd	13.46	9.13	12.09	16.31	25.20	4.73	0.22	0.11		0.48	0.19	
Sm	15.98	12.58	14.89	6.46	10.40	2.16	0.21	0.23		0.65	0.65	
Eu	16.00	13.39	16.05	5.09	8.41	1.68	0.26	0.18		0.64	0.40	
Dy	13.31	9.81	12.44	1.55	2.48	0.40	0.13	0.13		0.39	0.26	
Er	12.40	9.25	11.52	1.01	1.72	0.70	0.22	0.25		0.43	0.49	
Yb	12.62	10.03	12.43	1.06	1.35	0.58	0.53	0.50		0.72	0.53	
Ti	12.13	10.98	11.42	0.07	0.08	0.08	0.05	0.10		0.08		
Sr	16.15	17.05	15.51	41.28	52.95	35.00	0.47	0.15		0.41		
Zr	7.56	6.73	7.49	0.74	2.26	0.33	0.10	0.17		0.22		
La/Yb	0.68	0.65	0.63	66.02	72.71	70.40	5.81	0.38		0.71	0.59	
La/Sm	0.54	0.51	0.52	10.81	9.42	19.04	14.92	0.81		0.78	0.48	

Table A1.5

Sample grain Notes	OB93-66		OB93-74		OB93-77		OB93-78		OB93-79		OB93-80	
	avg	1 S.D.	avg	1 S.D.	avg	1 S.D.	avg	1 S.D.	avg	1 S.D.	avg	1 S.D.
	n=5						n=3					
La	0.03	0.01	0.40	0.01	0.14	0.18	0.02	0.01	0.05	0.01	0.06	0.06
Ce	0.05	0.00	1.20	0.00	0.58	0.44	0.02	0.01	0.07	0.01	0.14	0.14
Nd	0.05	0.03	0.91	0.03	0.65	0.41	0.03	0.01	0.07	0.01	0.13	0.13
Sm	0.05	0.05	0.27	0.05	0.22	0.31	0.04	0.02	0.06	0.02	0.07	0.07
Eu	0.01	0.01	0.09	0.01	0.08	0.05	0.01	0.00	0.02	0.00	0.02	0.02
Dy	0.03	0.01	0.34	0.01	0.24	0.31	0.02	0.01	0.04	0.01	0.06	0.06
Er	0.04	0.01	0.27	0.01	0.20	0.20	0.04	0.02	0.05	0.02	0.06	0.06
Yb	0.08	0.03	0.17	0.03	0.27	0.17	0.04	0.02	0.13	0.02	0.11	0.11
Ti	35	2	253	2	108	111	n.d.		61		n.d.	
V	125	8	147	8	132	134	n.d.		134		n.d.	
Cr	4010	524	5229	524	4383	5116	n.d.		4064		n.d.	
Sr	1.30	0.22	10.60	0.22	5.50	5.20	n.d.		1.20		n.d.	
Y	0.2	0.1	1.3	0.1	1.0	0.8	n.d.		0.0		n.d.	
Zr	0.6	0.2	4.7	0.2	2.9	3.0	n.d.		0.0		n.d.	
Normalized												
La	0.14		1.69		0.61	0.78	0.78		0.21		0.28	
Ce	0.08		1.99		0.96	0.73	0.81		0.11		0.24	
Nd	0.11		2.02		1.44	0.90	1.19		0.15		0.28	
Sm	0.35		1.81		1.49	2.08	1.75		0.40		0.51	
Eu	0.26		1.55		1.39	0.98	1.57		0.30		0.44	
Dy	0.14		1.42		1.00	1.28	1.26		0.17		0.24	
Er	0.26		1.71		1.24	1.27	0.98		0.33		0.37	
Yb	0.51		1.03		1.65	1.05	1.47		0.82		0.67	
Ti	0.08		0.58		0.25	0.25	0.26					
Sr	0.17		1.36		0.71	0.67	0.54					
Zr	0.15		1.19		0.74	0.76	0.81					
La/Yb	0.28		1.65		0.37	0.74	0.53		0.26		0.41	
La/Sm	0.41		0.93		0.41	0.37	0.45		0.52		0.54	

Table A1.5

Sample grain Notes	OB93-81c		OB93-83		OB93-85			OB93-86		OB93-89		
	1 S.D.	cpx	cpx1s	cpx3s	cpx3fs	cpx7	cpx1s	cpx	avg n=3	1 S.D.	avg n=7	1 S.D.
La	0.02	0.11	48.20	42.50	52.20	37.70	36.60	0.07	0.10	0.04	0.11	0.03
Ce	0.04	0.28	165.00	149.00	153.00	135.00	129.00	0.11	0.17	0.04	0.26	0.04
Nd	0.04	0.26	127.00	118.00	111.00	110.00	92.10	0.13	0.25	0.00	0.24	0.04
Sm	0.03	0.12	36.70	32.70	32.10	32.10	25.70	0.05	0.15	0.02	0.16	0.04
Eu	0.01	0.04	11.00	10.50	9.83	10.10	8.20	0.02	0.08	0.01	0.05	0.01
Dy	0.02	0.13	19.20	18.60	18.90	17.30	15.00	0.04	0.14	0.03	0.16	0.02
Er	0.02	0.08	7.48	7.45	6.36	6.95	6.43	0.04	0.19	0.02	0.10	0.02
Yb	0.03	0.14	6.89	6.39	6.11	5.61	5.27	0.12	0.18	0.04	0.16	0.02
Ti		125	663	619	527	646		26	99	5	75	4
V		131	117	127	91	126		107	128	7	124	7
Cr		4817	12055	8778	8703	8482		3835	4004	557	3848	408
Sr		2.60	655.00	678.00	306.00	655.00		1.90	2.23	0.31	2.78	0.11
Y		0.6	78.3	75.6	48.7	73.9		0.3	0.6	0.2	0.6	0.1
Zr		1.3	1535.0	1322.0	1250.0	1372.0		0.4	2.1	0.2	1.4	0.3
Normalized												
La		0.49	205.37	181.08	222.41	160.63	155.94	0.28	0.43		0.47	
Ce		0.46	273.54	247.02	253.65	223.81	213.86	0.19	0.28		0.44	
Nd		0.57	280.73	260.83	245.36	243.15	203.58	0.29	0.56		0.53	
Sm		0.83	249.49	222.30	218.22	218.22	174.71	0.35	1.05		1.10	
Eu		0.71	196.43	187.50	175.54	180.36	146.43	0.27	1.50		0.90	
Dy		0.52	79.11	76.64	77.87	71.28	61.80	0.17	0.57		0.64	
Er		0.52	47.07	46.88	40.03	43.74	40.47	0.23	1.23		0.65	
Yb		0.84	42.40	39.32	37.60	34.52	32.43	0.76	1.10		0.96	
Ti		0.29	1.52	1.42	1.21	1.48		0.06			0.17	
Sr		0.33	83.97	86.92	39.23	83.97		0.24			0.36	
Zr		0.33	389.59	335.53	317.26	348.22		0.10			0.36	
La/Yb		0.58	4.84	4.60	5.92	4.65	4.81	0.37	0.39		0.49	
La/Sm		0.59	0.82	0.81	1.02	0.74	0.89	0.78	0.41		0.43	

Table A1.5

Sample grain	Notes	OB93-95		OB93-98		OB93-99		Dunites				OB93-64	
		gm 2	avg others	1 S.D.	avg	1 S.D.	cpx1s	cpx2	cpx3s	cpx4	OB93-61c	cpx6	cpx
		n=6		n=6									
La		0.01	0.19	0.03	0.06	0.00	0.68	9.85	0.54	8.63	1.61	1.48	0.84
Ce		0.02	0.58	0.13	0.16	0.04	1.54	16.50	1.31	13.20	5.34	4.81	2.61
Nd		0.09	0.45	0.05	0.11	0.03	1.25	3.69	0.93	2.94	3.72	3.06	2.30
Sm		0.16	0.31	0.15	0.09	0.04	0.48	0.62	0.32	0.62	1.06	0.86	1.01
Eu		0.03	0.09	0.02	0.03	0.01	0.19	0.25	0.16	0.22	0.35	0.43	0.35
Dy		0.07	0.29	0.11	0.07	0.02	0.54	0.56	0.55	0.45	1.50	1.18	1.39
Er		0.05	0.13	0.02	0.06	0.02	0.31	0.38	0.36	0.42	0.97	0.89	0.89
Yb		0.04	0.21	0.06	0.11	0.04	0.30	0.36	0.41	0.37	0.98	1.11	0.75
Ti			115	1616	30	4	619	733	673	747	2881	2743	1764
V			129	63	101	7	166	168	160	169	256	245	157
Cr			4326	340	3628	319	7035	6045	6504	7138	6884	6923	6996
Sr			5.03	21.07	1.83	0.29	16.20	138.00	12.00	234.00	86.70	65.30	35.00
Y			0.9	4.8	0.2	0.1	2.6	3.0	3.2	4.1	10.8	8.7	8.2
Zr			2.5	11.5	0.9	0.2	6.0	8.5	6.9	10.4	40.4	36.8	22.8
Normalized													
La		0.06	0.80		0.27		2.88	41.97	2.32	36.77	6.86	6.31	3.56
Ce		0.03	0.95		0.27		2.55	27.35	2.17	21.88	8.85	7.97	4.33
Nd		0.20	0.99		0.24		2.76	8.16	2.05	6.50	8.22	6.76	5.08
Sm		1.06	2.09		0.61		3.28	4.18	2.14	4.24	7.21	5.84	6.87
Eu		0.46	1.57		0.49		3.30	4.46	2.89	3.98	6.30	7.71	6.23
Dy		0.29	1.19		0.29		2.22	2.32	2.26	1.87	6.18	4.86	5.73
Er		0.33	0.83		0.36		1.97	2.40	2.23	2.61	6.12	5.58	5.59
Yb		0.23	1.29		0.69		1.87	2.22	2.53	2.30	6.02	6.83	4.64
Ti							1.42	1.68	1.54	1.71	6.61	6.29	4.05
Sr							2.08	17.69	1.54	30.00	11.12	8.37	4.49
Zr							1.52	2.16	1.75	2.64	10.25	9.34	5.79
La/Yb		0.27	0.62		0.39		1.54	18.94	0.92	15.98	1.14	0.92	0.77
La/Sm		0.06	0.38		0.45		0.88	10.04	1.08	8.67	0.95	1.08	0.52

Table A1.5

Sample grain Notes	OB93-82 cpx	Pyroxenites										
		OB93-75					OB93-81b					
		cpx1,pt1	cpx1,pt2	cpx1,pt3	cpx2,pt1	cpx2,pt2	gr1	gr2	gr2,2	gr3	gr1	gr2
La	4.54	3.74	2.86	4.40	3.33	4.59	1.06	1.05	0.94	0.88	4.21	4.06
Ce	11.20	10.04	7.63	11.02	9.59	11.74	3.26	3.52	3.07	3.33	11.04	10.94
Nd	6.17	3.88	2.64	4.47	4.11	5.22	2.89	2.65	2.31	2.42	7.28	6.34
Sm	2.67	1.60	1.18	1.53	1.34	1.98	1.42	1.36	1.28	1.40	1.66	1.80
Eu	0.90	0.35	0.20	0.26	0.31	0.52	0.50	0.41	0.28	0.38	0.48	0.52
Dy	3.02	1.34	0.88	1.48	1.36	1.99	1.74	1.60	1.43	1.41	1.32	1.33
Er	1.78	1.04	0.48	1.05	0.91	1.37	1.14	1.20	0.87	1.31	1.01	0.89
Yb	1.78	0.65	0.44	0.82	0.59	0.96	0.97	0.95	0.88	0.81	0.63	0.71
Ti	4390						3131	3151		3129	2351	2296
V	272						305	298		295	231	224
Cr	5520						2574	2574		2621	6939	6267
Sr	79.60						24.40	20.40		21.60	81.40	77.50
Y	18.2						13.1	12.1		12.5	7.0	6.7
Zr	48.4						23.6	21.0		27.1	27.1	26.1
Normalized												
La	19.34	15.93	12.19	18.76	14.20	19.56	4.52	4.48	3.99	3.73	17.94	17.32
Ce	18.57	16.65	12.65	18.26	15.90	19.47	5.40	5.84	5.08	5.52	18.30	18.13
Nd	13.64	8.57	5.83	9.88	9.09	11.55	6.40	5.86	5.12	5.35	16.10	14.01
Sm	18.15	10.86	8.01	10.38	9.10	13.49	9.67	9.25	8.67	9.52	11.26	12.22
Eu	16.02	6.21	3.54	4.67	5.49	9.36	8.85	7.23	5.06	6.71	8.57	9.31
Dy	12.44	5.51	3.62	6.10	5.60	8.19	7.15	6.59	5.87	5.83	5.44	5.50
Er	11.20	6.52	2.99	6.59	5.76	8.63	7.18	7.57	5.49	8.22	6.36	5.60
Yb	10.95	4.01	2.69	5.07	3.65	5.93	6.00	5.82	5.44	4.99	3.85	4.36
Ti	10.07						7.18	7.23		7.18	5.47	5.27
Sr	10.21						3.13	2.62		2.77	10.58	9.94
Zr	12.28						5.99	5.33		6.88	6.65	6.62
La/Yb	1.77	3.97	4.53	3.70	3.89	3.30	0.75	0.77	0.73	0.75	4.66	3.97
La/Sm	1.07	1.47	1.52	1.81	1.56	1.45	0.47	0.48	0.46	0.39	1.59	1.42

Table A1.5

Sample grain	Mt Trapez Harzburgite, OB93-280										n=2
	gr2,2	OB93-87 gr1	gr2	gr4	OB93-94		OB93-100		OB93-102		avg cpx2
Notes					cpx1	cpx2	cpx3	cpx1	cpx2	cpx1	cpx2
La	4.43	0.86	1.00	0.90	2.59	3.21	5.97	0.45	0.39	0.75	0.89
Ce	10.70	2.90	2.77	3.12	7.22	8.22	14.80	1.89	1.71	2.90	3.07
Nd	5.78	1.89	1.93	2.61	5.08	5.20	7.78	2.02	2.15	2.89	3.39
Sm	1.42	0.98	1.06	1.35	1.58	1.35	1.62	1.10	1.10	1.29	1.45
Eu	0.43	0.32	0.26	0.51	0.49	0.44	0.48	0.56	0.44	0.65	0.56
Dy	1.30	0.99	1.08	2.25	1.33	1.13	1.26	2.11	2.14	2.14	2.32
Er	0.86	0.72	0.73	1.11	0.79	0.83	0.72	0.99	0.98	1.27	1.27
Yb	0.66	0.63	0.63	1.47	0.78	0.74	0.67	1.04	1.03	1.31	1.15
Ti	2262	3095	3117	2586	2115	1472	1077	2447	2507	3063	3237
V	220	301	302	286	248	235	213	316	329	295	306
Cr	6333	2756	2635	2655	7246	5686	6388	3357	3416	2471	2557
Sr	72.70	20.90	21.20	20.10	61.70	73.20	89.70	19.60	19.00	23.60	24.50
Y	5.7	13.6	12.9	11.2	6.2	6.5	6.2	10.9	10.8	12.0	12.2
Zr	25.6	23.1	25.7	19.3	19.2	23.4	39.2	18.6	16.8	22.2	26.8
Normalized											
La	18.88	3.66	4.25	3.82	11.04	13.68	25.44	1.92	1.66	3.21	3.80
Ce	17.74	4.81	4.60	5.17	11.97	13.63	24.54	3.14	2.84	4.81	5.10
Nd	12.79	4.19	4.27	5.77	11.23	11.49	17.20	4.48	4.76	6.38	7.48
Sm	9.63	6.66	7.18	9.20	10.74	9.18	11.01	7.45	7.48	8.74	9.82
Eu	7.76	5.68	4.59	9.10	8.71	7.93	8.52	10.06	7.89	11.55	9.91
Dy	5.37	4.06	4.45	9.28	5.48	4.66	5.19	8.70	8.81	8.82	9.57
Er	5.42	4.56	4.62	6.95	5.00	5.20	4.52	6.22	6.16	8.01	7.99
Yb	4.06	3.87	3.88	9.07	4.80	4.58	4.13	6.39	6.33	8.06	7.06
Ti	5.19	7.10	7.15	5.93	4.85	3.38	2.47	5.61	5.75	7.02	7.43
Sr	9.32	2.68	2.72	2.58	7.91	9.38	11.50	2.51	2.44	3.03	3.14
Zr	6.50	5.86	6.52	4.90	4.87	5.94	9.95	4.72	4.26	5.63	6.80
La/Yb	4.65	0.94	1.10	0.42	2.30	2.99	6.16	0.30	0.26	0.40	0.54
La/Sm	1.96	0.55	0.59	0.42	1.03	1.49	2.31	0.26	0.22	0.37	0.39
											6.75
											2.40

Table A1.5

Sample grain Notes	1 S.D. avg cpx5s n=2		1 S.D.		OB93-281		OB93-282	
	1 S.D.	avg cpx5s n=2	cpx1	cpx3s	cpx 4	cpx1	cpx2	cpx1
La	0.11	4.31	2.53	9.67	2.27	1.52	0.41	0.41
Ce	0.36	8.45	5.02	17.83	4.94	1.39	0.86	0.86
Nd	0.09	2.96	2.52	4.58	2.37	0.27	0.26	0.45
Sm	0.12	0.90	0.84	0.74	0.72	0.09	0.12	0.51
Eu	0.00	0.22	0.25	0.30	0.27	0.02	0.05	0.12
Dy	0.08	0.49	0.63	0.66	0.50	0.15	0.16	0.80
Er	0.06	0.32	0.26	0.30	0.22	0.17	0.16	0.39
Yb	0.02	0.28	0.21	0.30	0.28	0.25	0.22	0.31
Ti			138			n.d.		
V			180			n.d.		
Cr			7072			n.d.		
Sr			31.90			n.d.		
Y			2.2			n.d.		
Zr			6.8			n.d.		
Normalized								
La	18.36		10.76	41.19	9.68	6.49	1.74	1.75
Ce	14.01		8.33	29.55	8.19	2.31	1.43	1.42
Nd	6.54		5.58	10.13	5.24	0.59	0.58	1.00
Sm	6.12		5.74	5.01	4.90	0.58	0.81	3.44
Eu	3.84		4.42	5.43	4.77	0.37	0.97	2.18
Dy	2.02		2.59	2.71	2.04	0.62	0.65	3.30
Er	2.02		1.63	1.86	1.36	1.07	1.02	2.45
Yb	1.70		1.30	1.88	1.74	1.51	1.37	1.93
Ti			0.32				3.78	10.09
Sr			4.09				1.91	6.12
Zr			1.73				1.03	2.27
							0.91	2.85
							0.64	2.16
							0.54	2.80
							1.74	2.58
							1.98	2.81
La/Yb	10.81		8.29	21.95	5.55	4.30	1.27	0.90
La/Sm	3.00		1.88	8.22	1.97	11.23	2.14	0.51
							4.18	3.55
							1.91	3.59
							4.18	3.55
							1.91	0.90
							4.18	0.51
							1.91	0.69
							4.18	1.09

Table A1.5

Sample grain Notes	OB93-283		OB93-284		OB93-286		OB93-287		cpx3		
	cpx6	cpx2	cpx7	cpx1	cpx1	cpx8s	cpx1	cpx2		cpx4s	cpx5
La	4.16	1.84	6.31	4.93	1.72	1.75	10.65	7.87	3.80	5.93	10.17
Ce	11.12	4.46	15.95	13.48	4.02	3.40	17.48	11.56	5.16	9.54	15.00
Nd	6.39	2.31	9.48	9.10	1.99	1.80	3.17	1.69	0.73	1.70	2.48
Sm	2.00	0.61	2.70	2.79	0.53	0.66	0.38	0.28	0.17	0.36	0.64
Eu	0.75	0.21	0.96	1.04	0.25	0.18	0.14	0.07	0.04	0.08	0.21
Dy	1.87	0.62	3.03	2.65	0.83	0.66	0.32	0.29	0.18	0.32	0.83
Er	1.04	0.44	1.51	1.35	0.57	0.56	0.20	0.25	0.20	0.21	0.71
Yb	0.90	0.44	1.55	1.33	0.61	0.53	0.23	0.24	0.20	0.20	0.55
Ti	n.d.				299	303	245	237		210	n.d.
V	n.d.				226	226	121	112		110	n.d.
Cr	n.d.				8293	7941	4026	4039		3652	n.d.
Sr	n.d.				28.30	30.20	27.50	18.60		15.10	n.d.
Y	n.d.				4.6	4.2	1.1	0.8		1.0	n.d.
Zr	n.d.				4.1	4.8	5.1	1.7		2.6	n.d.
Normalized											
La	17.73	7.83	26.87	20.99	7.31	7.45	45.37	33.55	16.20	25.26	43.32
Ce	18.44	7.39	26.44	22.35	6.67	5.64	28.98	19.17	8.55	15.82	24.87
Nd	14.13	5.10	20.96	20.12	4.41	3.97	7.00	3.74	1.61	3.76	5.49
Sm	13.57	4.17	18.33	18.98	3.60	4.50	2.60	1.92	1.17	2.42	4.35
Eu	13.45	3.71	17.20	18.60	4.43	3.22	2.46	1.34	0.66	1.51	3.82
Dy	7.69	2.55	12.47	10.91	3.40	2.74	1.32	1.18	0.72	1.31	3.41
Er	6.51	2.79	9.48	8.49	3.59	3.54	1.27	1.59	1.26	1.29	4.46
Yb	5.55	2.69	9.53	8.17	3.74	3.27	1.43	1.47	1.25	1.25	3.38
Ti					0.68	0.69	0.56	0.54		0.48	
Sr					3.63	3.87	3.53	2.38		1.94	
Zr					1.04	1.22	1.29	0.43		0.66	
La/Yb	3.19	2.91	2.82	2.57	1.96	2.28	31.84	22.85	12.97	20.27	12.81
La/Sm	1.31	1.88	1.47	1.11	2.03	1.66	17.44	17.48	13.91	10.43	9.96

Table A1.5

Sample grain Notes	OB93-288			OB93-289 cpx1 avg n=2		1 S.D.	OB93-291		cpx8
	cpx6	cpx5s	cpx1	cpx2	cpx4		cpx3	cpx7s	
La	3.15	16.03	1.34	1.51	1.59	0.06	0.56	0.42	5.20
Ce	4.26	26.17	2.57	3.01	3.26	0.03	0.94	0.61	6.06
Nd	1.10	5.05	0.90	1.21	1.35	0.01	0.50	0.20	0.98
Sm	0.66	0.37	0.28	0.34	0.38	0.01	0.35	0.15	0.47
Eu	0.21	0.21	0.09	0.12	0.16	0.00	0.11	0.05	0.17
Dy	0.71	0.57	0.23	0.22	0.25	0.01	0.30	0.17	0.75
Er	0.51	0.49	0.15	0.19	0.19	0.03	0.30	0.20	0.42
Yb	0.65	0.53	0.18	0.25	0.31	0.01	0.33	0.28	0.49
Ti			371	363		n.d.			454
V			183	184		n.d.			185
Cr			7250	7651		n.d.			6273
Sr			29.90	35.70		n.d.			36.70
Y			1.9	2.0		n.d.			2.5
Zr			9.7	3.5		n.d.			7.2
Normalized									
La	13.43	68.29	5.73	6.44	6.78	2.11	2.40	1.81	22.14
Ce	7.07	43.38	4.26	4.98	5.41	1.44	1.55	1.02	10.04
Nd	2.43	11.16	1.99	2.67	2.99	0.68	1.11	0.45	2.16
Sm	4.47	2.49	1.93	2.28	2.60	1.12	2.40	0.99	3.19
Eu	3.68	3.71	1.54	2.06	2.86	0.88	1.97	0.83	3.00
Dy	2.93	2.35	0.95	0.90	1.05	1.06	1.24	0.69	3.08
Er	3.23	3.10	0.95	1.23	1.21	1.69	1.90	1.23	2.66
Yb	4.01	3.28	1.14	1.53	1.90	1.64	2.02	1.70	3.02
Ti			0.85	0.83					1.04
Sr			3.83	4.58					4.71
Zr			2.46	0.89					1.83
La/Yb	3.35	20.81	5.03	4.22	3.56	1.29	1.19	1.06	7.33
La/Sm	3.01	27.39	2.96	2.82	2.60	1.88	1.00	1.83	6.95

Table A1.5

Sample grain Notes	OB93-297		OB93-298		OB93-299		OB93-301	
	cpx1	cpx2	cpx1	cpx2	avg cpx1 n=2	I S.D. avg cpx3s n=2	avg cpx4s n=2	I S.D. cpx1
La	13.19	10.48	6.23	3.82	2.28	0.59	1.36	0.02
Ce	21.92	16.38	12.62	8.24	2.40	0.79	2.05	0.05
Nd	7.24	6.63	3.87	2.90	0.33	0.04	0.59	0.04
Sm	2.08	1.54	0.61	0.36	0.03	0.01	0.12	0.05
Eu	0.57	0.61	0.23	0.11	0.02	0.00	0.04	0.01
Dy	1.52	1.70	0.37	0.30	0.10	0.02	0.11	0.00
Er	0.88	0.80	0.24	0.24	0.17	0.03	0.15	0.01
Yb	0.78	0.90	0.34	0.27	0.24	0.03	0.20	0.05
Ti	1543	1368	258	282	68	63	63	63
V	212	209	181	171	156	155	169	151
Cr	10826	7473	5589	6776	5287	6245	5764	4467
Sr	128.40	102.70	73.80	80.20	12.20	4.10	24.40	19.20
Y	7.1	6.7	1.6	2.0	0.7	0.6	0.7	0.9
Zr	78.4	66.9	9.7	11.6	0.7	0.4	6.2	5.3
Normalized								
La	56.20	44.66	26.53	16.28	9.70	5.79	15.69	14.25
Ce	36.35	27.15	20.92	13.66	3.97	3.40	12.52	11.67
Nd	16.00	14.66	8.56	6.42	0.73	1.30	6.32	6.40
Sm	14.14	10.50	4.12	2.46	0.24	0.83	4.07	3.22
Eu	10.13	10.91	4.13	1.97	0.39	0.77	3.45	3.29
Dy	6.26	7.01	1.53	1.23	0.42	0.45	0.89	0.80
Er	5.52	5.05	1.54	1.53	1.06	0.97	0.91	1.88
Yb	4.81	5.55	2.06	1.68	1.48	1.26	1.73	1.75
Ti	3.54	3.14	0.59	0.65	0.16	0.15	0.14	0.16
Sr	16.46	13.17	9.46	10.28	1.56	0.53	3.13	2.46
Zr	19.90	16.98	2.46	2.94	0.18	0.10	1.58	1.35
La/Yb	11.68	8.05	12.87	9.69	6.57	4.61	9.08	8.14
La/Sm	3.98	4.25	6.44	6.61	40.99	6.95	3.85	4.42

Table A1.5

Sample grain Notes	OB93-302			OB93-305			OB93-306			OB93-307		
	cpx3	cpx8	cpx2	cpx7	cpx1	cpx5	cpx2	cpx3	cpx1, pt2	cpx1,pt1	cpx2	avg1 n=4
La	2.78	3.52	6.23	2.70	3.74	3.74	3.21	2.56	2.40	1.76	0.22	0.04
Ce	4.85	8.43	11.51	4.40	5.80	9.78	9.64	8.08	6.45	5.12	0.58	0.05
Nd	1.28	4.44	2.83	1.13	1.20	4.81	6.83	6.33	5.13	4.36	0.59	0.12
Sm	0.65	0.99	0.46	0.27	0.35	1.25	2.36	1.58	1.75	1.76	0.34	0.12
Eu	0.16	0.35	0.12	0.07	0.08	0.39	0.81	0.67	0.80	0.69	0.11	0.03
Dy	0.19	0.51	0.25	0.15	0.10	0.61	1.50	1.23	2.57	2.34	0.41	0.12
Er	0.39	0.35	0.24	0.26	0.21	0.40	0.67	0.70	1.36	1.29	0.25	0.20
Yb	0.39	0.41	0.19	0.24	0.23	0.38	0.87	0.56	1.50	1.20	0.27	0.25
Ti	74	87	169	168		166	781	276	4655		4071	66
V	150	156	192	222		150	155	150	348		350	146
Cr	4668	5255	6229	6457		6357	5526	5629	3934		4864	4933
Sr	9.80	28.50	73.90	36.00		91.30	128.70	114.20	92.40		80.80	0.70
Y	0.7	1.2	1.2	0.7		2.3	6.9	4.4	14.7		13.3	0.5
Zr	0.9	17.2	8.3	3.9		11.2	20.9	15.4	43.5		40.3	0.6
Normalized												
La	11.87	14.99	26.55	11.51	15.93	15.93	13.67	10.93	10.24	7.52	0.96	0.16
Ce	8.04	13.97	19.09	7.29	9.62	16.21	15.98	13.39	10.69	8.49	0.97	0.08
Nd	2.84	9.82	6.25	2.50	2.66	10.62	15.09	13.99	11.34	9.64	1.31	0.27
Sm	4.41	6.76	3.15	1.85	2.37	8.52	16.03	10.75	11.90	11.95	2.34	0.85
Eu	2.90	6.18	2.15	1.30	1.50	7.03	14.51	12.00	14.28	12.34	1.94	0.47
Dy	0.80	2.08	1.03	0.64	0.42	2.52	6.17	5.06	10.57	9.64	1.71	0.49
Er	2.43	2.17	1.52	1.66	1.35	2.54	4.19	4.39	8.53	8.09	1.59	1.23
Yb	2.41	2.54	1.19	1.45	1.40	2.32	5.36	3.46	9.21	7.40	1.64	1.56
Ti	0.17	0.20	0.39	0.39		0.38	1.79	0.63	10.68		9.34	0.15
Sr	1.26	3.65	9.47	4.62		11.71	16.50	14.64	11.85		10.36	0.09
Zr	0.23	4.37	2.11	0.99		2.84	5.30	3.91	11.04		10.23	0.15
La/Yb	4.93	5.89	22.34	7.95	11.40	6.87	2.55	3.16	1.11	1.02	0.58	0.10
La/Sm	2.69	2.22	8.44	6.22	6.72	1.87	0.85	1.02	0.86	0.63	0.41	0.19

Table A1.5

Sample grain Notes	avg2 n=4		avg3 n=2		avg10s n=2		OB93-308 avg1 n=3		cpx2	
	1 S.D.	1 S.D.	1 S.D.	1 S.D.	1 S.D.	1 S.D.	1 S.D.	1 S.D.	1 S.D.	1 S.D.
La	0.02	0.04	0.03	0.16	0.08	0.03	0.04	0.02	5.05	0.39
Ce	0.04	0.04	0.02	0.14	0.07	0.03	0.05	0.04	9.73	1.22
Nd	0.13	0.05	0.01	0.07	0.02	0.05	0.11	0.08	3.22	0.15
Sm	0.13	0.05	0.01	0.05	0.02	0.02	0.08	0.09	0.50	0.02
Eu	0.02	0.02	0.01	0.01	0.00	0.02	0.03	0.03	0.18	0.03
Dy	0.08	0.06	0.02	0.09	0.02	0.09	0.09	0.11	0.24	0.04
Er	0.12	0.12	0.03	0.17	0.04	0.11	0.19	0.16	0.18	0.04
Yb	0.10	0.20	0.04	0.25	0.04	0.17	0.28	0.22	0.21	0.01
Ti				68			73		3187	
V				152			145		125	
Cr				5712			5172		60	
Sr				1.10			0.40		48.10	
Y				0.7			0.5		27.0	
Zr				0.7					1.1	
Normalized										
La	0.16			0.69		0.11	0.19	0.10	21.53	18.23
Ce	0.06			0.23		0.06	0.07	0.06	16.13	12.01
Nd	0.10			0.16		0.12	0.24	0.18	7.11	4.42
Sm	0.36			0.34		0.15	0.55	0.59	3.37	3.54
Eu	0.40			0.26		0.36	0.55	0.47	3.13	2.86
Dy	0.24			0.38		0.36	0.39	0.45	0.97	1.86
Er	0.75			1.08		0.70	1.21	1.03	1.15	1.59
Yb	1.22			1.51		1.06	1.71	1.34	1.32	2.14
Ti				0.16			0.17		7.31	0.70
Sr				0.14			0.05		6.17	4.28
Zr				0.18			0.00		0.28	7.21
La/Yb	0.13			0.45		0.10	0.11	0.07	16.30	8.51
La/Sm	0.44			2.04		0.74	0.34	0.16	6.38	5.14
										7.85

Table A1.5

Sample grain Notes	OB93-309			OB93-310			OB93-310			OB93-310		
	avg3(s) n=2	1 S.D.	cpx4	cpx1	cpx2	cpx6s	avg2 n=3	1 S.D.	cpx3	cpx5s	cpx6	cpx7
La	3.16	0.50	4.85	4.97	1.40	1.01	0.02	0.02	0.01	0.01	0.01	0.01
Ce	4.99	0.71	6.84	7.60	2.15	1.64	0.03	0.02	0.01	0.01	0.02	0.03
Nd	1.71	0.28	1.55	1.79	1.05	0.76	0.07	0.05	0.02	0.04	0.01	0.04
Sm	0.48	0.04	0.44	0.38	0.32	0.35	0.10	0.07	0.02	0.03	0.01	0.04
Eu	0.15	0.01	0.10	0.16	0.12	0.08	0.02	0.01	0.01	0.01	0.02	0.01
Dy	0.51	0.02	0.41	0.45	0.40	0.34	0.09	0.03	0.07	0.04	0.03	0.02
Er	0.26	0.02	0.23	0.29	0.17	0.27	0.12	0.04	0.11	0.07	0.04	0.03
Yb	0.40	0.05	0.27	0.28	0.19	0.20	0.17	0.01	0.16	0.08	0.08	0.02
Ti	357		306	413	388	354	30		30			
V	168		170	157	175	151	125		134			
Cr	7423		6294	7887	12288	7292	4968		5889			
Sr	40.00		33.40	39.30	21.20	13.80	0.42		0.40			
Y	3.1		2.9	2.6	2.2	1.9	0.3		0.5			
Zr	11.0		28.4	7.5	5.8	4.7						
Normalized												
La	13.45		20.68	21.17	5.96	4.32	0.09		0.03	0.03	0.06	0.05
Ce	8.27		11.33	12.60	3.56	2.72	0.05		0.02	0.02	0.04	0.05
Nd	3.77		3.44	3.97	2.33	1.68	0.16		0.05	0.09	0.02	0.10
Sm	3.26		3.02	2.60	2.15	2.37	0.69		0.14	0.18	0.08	0.27
Eu	2.69		1.86	2.81	2.08	1.51	0.43		0.19	0.18	0.40	0.15
Dy	2.10		1.69	1.85	1.65	1.41	0.36		0.31	0.15	0.11	0.08
Er	1.63		1.46	1.81	1.04	1.72	0.78		0.68	0.45	0.25	0.18
Yb	2.48		1.64	1.71	1.19	1.22	1.07		1.00	0.49	0.51	0.15
Ti	0.82		0.70	0.95	0.89	0.81	0.07		0.07			
Sr	5.13		4.28	5.04	2.72	1.77	0.05		0.05			
Zr	2.79		7.21	1.90	1.47	1.19	0.00		0.00			
La/Yb	5.43		12.59	12.39	5.00	3.55	0.08		0.03	0.07	0.12	0.32
La/Sm	4.12		6.84	8.15	2.77	1.82	0.13		0.25	0.18	0.80	0.17

Table A1.5

Sample grain	OB93-313			OB93-314			OB93-317		
	avg1 n=3	1 S.D. n=3	avg2 n=3	1 S.D. n=2	avg cpx n=4	1 S.D. n=4	cpx1	cpx4s	cpx7
Notes									
La	8.82	1.77	8.96	1.43	10.17	0.57	0.30	0.01	0.83
Ce	16.00	2.41	17.99	2.75	14.19	0.88	0.81	0.05	1.93
Nd	3.48	0.33	5.33	0.61	2.52	0.65	0.69	0.08	1.28
Sm	0.56	0.09	0.99	0.09	0.60	0.16	0.35	0.15	1.13
Eu	0.23	0.05	0.38	0.05	0.23	0.09	0.10	0.04	0.21
Dy	0.69	0.08	0.92	0.03	0.96	0.14	0.30	0.11	0.88
Er	0.55	0.03	0.60	0.10	0.50	0.08	0.25	0.08	0.71
Yb	0.60	0.03	0.64	0.04	0.58	0.09	0.30	0.10	0.57
Ti	1305				1262		167		268
V	230				210		167		182
Cr	8245				7730		7436		10589
Sr	74.30				63.10		6.65		21.90
Y	5.4				4.5		1.1		2.7
Zr	18.8				20.6		2.2		8.4
Normalized									
La	37.58		38.18		43.33		1.30		3.70
Ce	26.53		29.83		23.53		1.35		3.20
Nd	7.69		11.78		5.57		1.52		2.83
Sm	3.77		6.71		4.06		2.41		4.66
Eu	4.10		6.84		4.07		1.74		4.19
Dy	2.86		3.78		3.95		1.22		3.22
Er	3.49		3.80		3.16		1.54		4.29
Yb	3.67		3.93		3.59		1.86		2.39
Ti	2.99				2.90		0.38		0.82
Sr	9.53				8.09		0.85		3.23
Zr	4.77				5.23		0.55		4.14
La/Yb	10.24		9.71		12.07		0.70		1.55
La/Sm	9.96		5.69		10.66		0.54		0.79

Table A1.5

Sample grain Notes	OB93-318 cpx2	cpx3	avg4s n=2	1 S.D.
La	0.43	0.78	1.91	0.15
Ce	0.63	1.22	4.15	0.01
Nd	0.18	0.52	1.21	0.01
Sm	0.08	0.27	0.30	0.16
Eu	0.04	0.06	0.08	0.00
Dy	0.19	0.24	0.25	0.01
Er	0.24	0.29	0.27	0.00
Yb	0.32	0.37	0.30	0.01
Ti	268	257		
V	187	204		
Cr	6888	9936		
Sr	22.00	9.30		
Y	1.6	1.6		
Zr	2.8	1.8		
Normalized				
La	1.81	3.34	8.13	
Ce	1.04	2.02	6.87	
Nd	0.39	1.15	2.67	
Sm	0.54	1.83	2.03	
Eu	0.64	1.08	1.42	
Dy	0.78	1.00	1.02	
Er	1.52	1.83	1.73	
Yb	1.95	2.27	1.84	
Ti	0.61	0.59	0.00	
Sr	2.82	1.19	0.00	
Zr	0.71	0.46	0.00	
La/Yb	0.93	1.47	4.43	
La/Sm	3.39	1.82	4.02	

**Appendix 2. Complete major and trace element compositions of
minerals in the granulites.**

Table A2.1. Major element compositions of granulite clinopyroxenes. Each analysis is a separate grain, unless noted (e.g. pt1, pt2, etc).

Sample Point ID	OB93-50			OB93-54			OB93-57								
	1	2		1	2	3	2	3	4	5	6				
wt. %															
SiO ₂	50.11	50.75		50.30	50.67	50.65	50.76	49.55	50.13	50.24	50.06				
TiO ₂	0.47	0.44		0.61	0.54	0.54	0.72	0.79	0.82	0.81	0.48				
Al ₂ O ₃	7.34	6.95		5.17	5.11	5.24	6.31	6.67	6.65	7.04	7.67				
Cr ₂ O ₃	0.28	0.27		0.03	0.02	0.02	0.16	0.18	0.18	0.15	0.01				
FeO*	4.76	4.80		9.85	9.87	9.80	8.74	6.53	6.70	6.85	6.88				
MnO	0.10	0.13		0.10	0.13	0.12	0.16	0.11	0.14	0.12	0.13				
MgO	13.82	14.19		11.78	12.16	12.00	15.53	13.16	13.18	13.13	12.76				
CaO	21.74	21.96		20.19	20.37	20.04	16.98	20.61	20.94	20.91	20.51				
Na ₂ O	0.74	0.78		1.35	1.23	1.41	0.87	1.02	0.89	1.03	1.16				
Total	99.37	100.27		99.39	100.10	99.82	100.24	98.62	99.63	100.29	99.65				
Cations/6 Oxygens															
Si	1.845	1.853		1.892	1.892	1.894	1.862	1.852	1.855	1.848	1.850				
Ti	0.013	0.012		0.017	0.015	0.015	0.020	0.022	0.023	0.023	0.013				
Al	0.319	0.299		0.229	0.225	0.231	0.273	0.294	0.290	0.305	0.334				
Cr	0.008	0.008		0.001	0.001	0.001	0.005	0.005	0.005	0.005	0.000				
Fe	0.147	0.147		0.310	0.308	0.307	0.268	0.204	0.207	0.211	0.213				
Mn	0.003	0.004		0.003	0.004	0.004	0.005	0.003	0.004	0.004	0.004				
Mg	0.759	0.772		0.660	0.677	0.669	0.849	0.733	0.727	0.720	0.703				
Ca	0.858	0.859		0.814	0.815	0.803	0.667	0.825	0.830	0.824	0.812				
Na	0.053	0.056		0.099	0.089	0.102	0.062	0.074	0.064	0.074	0.083				
Total	4.005	4.009		4.025	4.025	4.026	4.011	4.013	4.007	4.012	4.011				
Mg#	83.80	84.05		68.06	68.72	68.57	76.00	78.21	77.81	77.35	76.78				
EQ Liq Mg#	56.93	57.38		35.26	35.96	35.80	44.74	47.85	47.26	46.60	45.79				
Molar percentage of endmembers															
Wo	48.64	48.32		45.60	45.28	45.15	37.39	46.82	47.04	46.96	47.01				
En	43.04	43.44		37.02	37.61	37.61	47.58	41.60	41.21	41.03	40.68				
Fs	8.32	8.24		17.37	17.11	17.24	15.02	11.59	11.75	12.02	12.31				
Notes											augite		fine sympl.		

EQ Liq Mg#: Mg# of cpx equilibrium melt, calculated from $[(Fe^{2+}/Mg)cpx]/(Fe^{2+}/Mg)liq = 0.23$ (Grove, and Bryan, 1983); and $Fe^{2+}/Fe^{3+} = 0.9$.
 Symp.: symplectite

Table A2.1 Clinopyroxenes continued.

Sample Point ID	OB93-60							OB93-61A			
	7	8	9	10	11	12		1, pt1	1, pt2	2, pt1	2, pt2
wt. %											
SiO ₂	50.05	50.55	50.79	50.66	50.85	50.70		48.29	49.50	49.01	49.07
TiO ₂	0.43	0.43	0.40	0.44	0.30	0.46		1.20	1.27	1.16	1.18
Al ₂ O ₃	7.00	7.15	6.54	7.29	7.19	7.06		7.23	7.21	7.12	6.91
Cr ₂ O ₃	0.23	0.27	0.12	0.27	0.04	0.25		0.09	0.10	0.09	0.10
FeO*	4.75	4.77	4.90	4.95	4.85	4.71		6.34	6.59	6.40	6.48
MnO	0.12	0.11	0.10	0.10	0.07	0.12		0.13	0.15	0.17	0.14
MgO	14.05	13.91	14.30	14.04	14.05	14.04		12.89	13.14	13.14	13.41
CaO	21.64	21.48	21.90	21.93	21.41	21.44		21.58	21.05	21.45	20.76
Na ₂ O	0.77	0.83	0.77	0.81	0.84	0.84		0.99	0.88	0.97	0.93
Total	99.04	99.50	99.82	100.50	99.58	99.63		98.75	99.89	99.51	98.99
<i>Cations/6 Oxygens</i>											
Si	1.850	1.857	1.863	1.846	1.864	1.859		1.811	1.829	1.821	1.829
Ti	0.012	0.012	0.011	0.012	0.008	0.013		0.034	0.035	0.032	0.033
Al	0.305	0.310	0.283	0.313	0.310	0.305		0.319	0.314	0.312	0.304
Cr	0.007	0.008	0.004	0.008	0.001	0.007		0.003	0.003	0.003	0.003
Fe	0.147	0.147	0.150	0.151	0.149	0.144		0.199	0.204	0.199	0.202
Mn	0.004	0.004	0.003	0.003	0.002	0.004		0.004	0.005	0.005	0.005
Mg	0.774	0.762	0.782	0.763	0.767	0.768		0.721	0.724	0.728	0.745
Ca	0.857	0.845	0.861	0.856	0.841	0.842		0.867	0.833	0.854	0.829
Na	0.056	0.059	0.055	0.057	0.060	0.060		0.072	0.063	0.070	0.067
Total	4.010	4.002	4.011	4.010	4.002	4.002		4.030	4.009	4.024	4.018
Mg#	84.05	83.86	83.86	83.48	83.78	84.16		78.36	78.03	78.54	78.67
EQ Liq Mg	57.39	57.04	57.05	56.35	56.91	57.60		48.06	47.58	48.33	48.52
<i>Molar percentage of endmembers</i>											
Wo	48.19	48.20	48.01	48.37	47.87	48.02		48.53	47.34	47.97	46.68
En	43.54	43.44	43.60	43.10	43.68	43.75		40.33	41.09	40.86	41.95
Fs	8.26	8.36	8.39	8.53	8.45	8.23		11.14	11.57	11.17	11.37
Notes	coarse symp.										

Table A2.1 Clinopyroxenes continued.

Sample Point ID	OB93-68				OB93-69			OB93-71			OB93-72		
	22	23	24	wt. %	19, pt1	20, pt2	21	1	2	3	13	17	
SiO ₂	48.73	48.58	49.18		50.19	49.87	49.87	50.52	50.66	51.75	50.36	50.66	
TiO ₂	1.50	1.56	1.36		1.00	1.30	1.20	0.61	0.70	0.59	0.60	0.53	
Al ₂ O ₃	7.49	7.14	7.18		6.67	6.14	6.17	7.14	7.30	7.06	7.59	6.97	
Cr ₂ O ₃	0.13	0.12	0.09		0.27	0.29	0.28	0.20	0.25	0.18	0.20	0.17	
FeO*	6.84	7.02	7.47		6.80	7.33	7.07	4.39	4.23	4.33	4.87	5.03	
MnO	0.11	0.11	0.13		0.14	0.13	0.14	0.13	0.12	0.11	0.11	0.07	
MgO	12.66	12.81	12.28		12.45	12.76	12.71	13.89	14.02	14.15	13.52	13.77	
CaO	21.41	21.32	21.27		21.15	21.30	21.37	20.92	21.23	21.24	20.99	20.69	
Na ₂ O	0.93	0.99	1.04		0.98	0.77	0.95	1.29	1.26	1.22	0.85	0.92	
Total	99.81	99.65	100.00		99.65	99.89	99.76	99.09	99.77	100.64	99.10	98.82	
<i>Cations/6 Oxygens</i>													
Si	1.809	1.809	1.825		1.859	1.850	1.852	1.860	1.853	1.872	1.855	1.871	
Ti	0.042	0.044	0.038		0.028	0.036	0.034	0.017	0.019	0.016	0.017	0.015	
Al	0.328	0.314	0.314		0.291	0.269	0.270	0.310	0.315	0.301	0.329	0.303	
Cr	0.004	0.004	0.003		0.008	0.008	0.008	0.006	0.007	0.005	0.006	0.005	
Fe	0.212	0.219	0.232		0.211	0.227	0.220	0.135	0.129	0.131	0.150	0.155	
Mn	0.004	0.004	0.004		0.004	0.004	0.004	0.004	0.004	0.004	0.003	0.002	
Mg	0.700	0.711	0.679		0.687	0.706	0.703	0.762	0.764	0.763	0.742	0.758	
Ca	0.852	0.851	0.846		0.839	0.847	0.850	0.825	0.832	0.824	0.828	0.819	
Na	0.067	0.071	0.075		0.070	0.055	0.068	0.092	0.089	0.085	0.061	0.066	
Total	4.017	4.024	4.016		3.998	4.003	4.010	4.012	4.012	4.001	3.991	3.993	
Mg#	76.73	76.49	74.56		76.53	75.63	76.20	84.95	85.54	85.33	83.18	82.99	
EQ Liq Mg	45.73	45.40	42.82		45.45	44.23	45.00	59.05	60.19	59.78	55.82	55.49	
<i>Molar percentage of endmembers</i>													
Wo	48.27	47.79	48.15		48.31	47.58	47.96	47.90	48.21	47.94	48.14	47.27	
En	39.69	39.94	38.66		39.56	39.64	39.66	44.25	44.30	44.42	43.13	43.76	
Fs	12.04	12.28	13.19		12.13	12.77	12.39	7.84	7.49	7.64	8.72	8.97	
Notes					fine syml. in core of				coarse syml.				
					pt 19								

Table A2.1 Clinopyroxenes continued.

Sample	OB93-90				OB93-96			
Point ID	25	26	27	28	29	30	31	
wt. %								
SiO ₂	51.01	50.46	50.30	50.19	50.50	50.77	51.41	
TiO ₂	0.52	0.55	0.55	0.52	0.44	0.47	0.35	
Al ₂ O ₃	7.19	7.37	7.54	6.95	7.29	6.94	6.72	
Cr ₂ O ₃	0.21	0.26	0.24	0.16	0.26	0.26	0.07	
FeO*	4.73	4.73	4.65	4.59	4.77	4.55	4.68	
MnO	0.07	0.09	0.11	0.05	0.12	0.13	0.06	
MgO	14.19	14.18	13.94	14.51	14.22	13.91	14.36	
CaO	21.53	21.33	21.45	21.50	21.78	21.49	21.77	
Na ₂ O	0.84	0.79	0.86	0.80	0.74	0.78	0.69	
Total	100.28	99.76	99.64	99.28	100.12	99.29	100.12	
<i>Cations/6 Oxygens</i>								
Si	1.857	1.848	1.845	1.848	1.845	1.866	1.873	
Ti	0.014	0.015	0.015	0.015	0.012	0.013	0.010	
Al	0.308	0.318	0.326	0.302	0.314	0.301	0.289	
Cr	0.006	0.007	0.007	0.005	0.008	0.008	0.002	
Fe	0.144	0.145	0.143	0.141	0.146	0.140	0.143	
Mn	0.002	0.003	0.003	0.002	0.004	0.004	0.002	
Mg	0.770	0.774	0.762	0.796	0.775	0.762	0.780	
Ca	0.840	0.837	0.843	0.848	0.853	0.846	0.850	
Na	0.059	0.056	0.061	0.057	0.052	0.056	0.049	
Total	4.001	4.002	4.005	4.013	4.008	3.995	3.997	
Mg#	84.24	84.24	84.24	84.92	84.15	84.49	84.53	
EQ Liq Mg	57.74	57.73	57.73	59.00	57.57	58.20	58.28	
<i>Molar percentage of endmembers</i>								
Wo	47.89	47.66	48.23	47.50	48.09	48.41	47.94	
En	43.90	44.09	43.61	44.58	43.68	43.59	44.01	
Fs	8.21	8.25	8.16	7.92	8.23	8.00	8.05	
Notes	cpx core in				in sympl.			
	opx							

Table A2.2. Major element compositions of granulite orthopyroxenes. Each analysis is a separate grain, unless noted.

Sample	OB93-50				OB93-54				OB93-57			
Point ID	1	2	3	4	2	5	4	5, pt	6, pt2	7, pt3	8	1
wt. %												
SiO ₂	52.70	52.35	52.17	52.02	51.43	52.35	51.05	51.60	51.66	50.80	51.09	51.86
TiO ₂	0.08	0.08	0.08	0.07	0.15	0.09	0.10	0.08	0.06	0.07	0.15	0.19
Al ₂ O ₃	6.01	6.13	6.96	7.17	2.32	2.24	6.79	5.54	5.21	6.75	5.40	5.03
Cr ₂ O ₃	0.01	0.06	0.01	0.02	0.03	0.03	0.02	0.01	0.01	0.01	0.06	0.11
FeO*	12.65	12.40	12.58	12.49	23.69	22.78	16.59	16.66	17.26	16.95	16.29	16.95
MnO	0.20	0.15	0.19	0.19	0.31	0.24	0.28	0.28	0.30	0.27	0.24	0.28
MgO	28.33	27.95	27.95	28.13	21.97	22.22	24.84	25.08	25.63	24.86	25.08	25.66
CaO	0.53	0.55	0.55	0.52	0.60	0.60	0.56	0.60	0.61	0.60	1.00	0.57
Na ₂ O	0.06	0.07	0.08	0.04	0.01	0.00	0.14	0.08	0.07	0.10	0.12	0.04
Total	100.56	99.74	100.56	100.65	100.51	100.54	100.37	99.91	100.82	100.41	99.43	100.68
<i>Cations/6 Oxygens</i>												
Si	1.863	1.864	1.844	1.836	1.920	1.941	1.843	1.872	1.864	1.838	1.864	1.871
Ti	0.002	0.002	0.002	0.002	0.004	0.002	0.003	0.002	0.002	0.002	0.004	0.005
Al	0.250	0.257	0.290	0.298	0.102	0.098	0.289	0.237	0.222	0.288	0.232	0.214
Cr	0.000	0.002	0.000	0.001	0.001	0.001	0.001	0.000	0.000	0.000	0.002	0.003
Fe	0.374	0.369	0.372	0.369	0.740	0.706	0.501	0.505	0.521	0.513	0.497	0.511
Mn	0.006	0.005	0.006	0.006	0.010	0.008	0.009	0.009	0.009	0.008	0.008	0.009
Mg	1.493	1.483	1.472	1.480	1.223	1.228	1.337	1.356	1.379	1.341	1.364	1.380
Ca	0.020	0.021	0.021	0.020	0.024	0.024	0.022	0.023	0.024	0.023	0.039	0.022
Na	0.004	0.005	0.005	0.003	0.001	0.000	0.010	0.006	0.005	0.007	0.008	0.003
Total	4.012	4.007	4.012	4.014	4.024	4.007	4.014	4.010	4.026	4.020	4.019	4.017
Mg#	79.97	80.06	79.84	80.06	62.31	63.48	72.74	72.86	72.57	72.33	73.30	72.96
<i>Molar percentage of endmembers</i>												
Wo	1.07	1.12	1.11	1.06	1.21	1.21	1.17	1.23	1.23	1.24	2.06	1.14
En	79.11	79.17	78.95	79.21	61.55	62.71	71.89	71.97	71.68	71.44	71.79	72.12
Fs	19.82	19.71	19.94	19.73	37.24	36.08	26.94	26.81	27.09	27.32	26.15	26.73
Notes	coarse symp.		coarse symp.	coarse symp.			fine symp.			fine symp.		
Symp.: symplectite												

Table A2.2 Orthopyroxenes continued.

Sample	OB93-60				OB93-61A			OB93-68			
Point ID	9	10	11	12	3	4	19	20	21	22	
wt. %											
SiO ₂	52.42	51.67	52.96	52.77	51.37	51.99	51.54	53.40	51.54	51.20	
TiO ₂	0.05	0.08	0.07	0.06	0.13	0.14	0.13	0.72	0.21	0.14	
Al ₂ O ₃	7.25	7.66	5.96	6.30	5.23	5.31	4.76	1.76	5.57	5.48	
Cr ₂ O ₃	0.06	0.01	0.17	0.12	0.07	0.05	0.06	0.07	0.08	0.05	
FeO*	12.37	11.97	12.41	12.31	15.59	16.01	17.48	16.46	17.07	17.35	
MnO	0.17	0.18	0.23	0.24	0.29	0.28	0.32	0.33	0.23	0.28	
MgO	27.95	27.74	28.03	28.05	25.85	25.73	25.03	25.65	25.03	24.67	
CaO	0.54	0.53	0.53	0.57	0.50	0.50	0.51	2.02	0.70	0.44	
Na ₂ O	0.06	0.05	0.07	0.06	0.02	0.03	0.07	0.05	0.10	0.06	
Total	100.87	99.88	100.45	100.48	99.07	100.04	99.91	100.44	100.55	99.66	
<i>Cations/6 Oxygens</i>											
Si	1.844	1.833	1.872	1.864	1.872	1.877	1.879	1.935	1.863	1.869	
Ti	0.001	0.002	0.002	0.002	0.004	0.004	0.004	0.020	0.006	0.004	
Al	0.301	0.320	0.248	0.262	0.225	0.226	0.204	0.075	0.238	0.236	
Cr	0.002	0.000	0.005	0.003	0.002	0.002	0.002	0.002	0.002	0.002	
Fe	0.364	0.355	0.367	0.364	0.475	0.483	0.533	0.499	0.516	0.530	
Mn	0.005	0.006	0.007	0.007	0.009	0.009	0.010	0.010	0.007	0.009	
Mg	1.465	1.467	1.477	1.476	1.404	1.385	1.360	1.385	1.349	1.342	
Ca	0.020	0.020	0.020	0.022	0.020	0.019	0.020	0.078	0.027	0.017	
Na	0.004	0.003	0.005	0.004	0.002	0.002	0.005	0.004	0.007	0.004	
Total	4.006	4.006	4.002	4.004	4.012	4.006	4.017	4.008	4.015	4.011	
Mg#	80.10	80.51	80.10	80.23	74.72	74.12	71.85	73.53	72.32	71.70	
<i>Molar percentage of endmembers</i>											
Wo	1.10	1.09	1.09	1.16	1.04	1.02	1.05	3.99	1.43	0.91	
En	79.22	79.63	79.23	79.30	73.94	73.36	71.10	70.59	71.29	71.05	
Fs	19.68	19.28	19.68	19.54	25.02	25.61	27.86	25.41	27.28	28.04	
Notes	coarse symp.	coarse symp.						fine symp.			

Table A2.2 Orthopyroxenes continued.

Sample Point ID	OB93-69				OB93-71			OB93-72			
	16	17	18	19	8	9		13	14	15	
wt. %											
SiO ₂	51.31	51.70	51.57	51.66	53.85	54.41		51.43	53.68	51.39	
TiO ₂	0.14	0.10	0.10	0.15	0.07	0.08		0.09	0.07	0.08	
Al ₂ O ₃	4.38	5.13	4.32	4.35	4.51	4.07		6.98	5.98	7.61	
Cr ₂ O ₃	0.11	0.04	0.14	0.18	0.05	0.06		0.10	0.03	0.06	
FeO*	19.54	19.52	18.95	19.83	11.48	11.42		12.90	12.87	12.96	
MnO	0.32	0.32	0.31	0.28	0.20	0.23		0.20	0.15	0.21	
MgO	23.80	23.64	23.85	23.44	28.97	29.52		27.01	26.29	26.49	
CaO	0.57	0.50	0.46	0.79	0.51	0.59		0.60	0.59	0.60	
Na ₂ O	0.06	0.07	0.03	0.01	0.07	0.05		0.07	0.08	0.08	
Total	100.24	101.02	99.71	100.69	99.73	100.43		99.37	99.74	99.49	
<i>Cations/6 Oxygens</i>											
Si	1.883	1.879	1.895	1.889	1.907	1.913		1.844	1.907	1.839	
Ti	0.004	0.003	0.003	0.004	0.002	0.002		0.003	0.002	0.002	
Al	0.190	0.220	0.187	0.188	0.188	0.169		0.295	0.250	0.321	
Cr	0.003	0.001	0.004	0.005	0.002	0.002		0.003	0.001	0.002	
Fe	0.600	0.593	0.582	0.607	0.340	0.336		0.387	0.382	0.388	
Mn	0.010	0.010	0.010	0.009	0.006	0.007		0.006	0.004	0.007	
Mg	1.302	1.281	1.306	1.278	1.529	1.547		1.443	1.393	1.413	
Ca	0.022	0.020	0.018	0.031	0.019	0.022		0.023	0.023	0.023	
Na	0.004	0.005	0.002	0.001	0.005	0.003		0.005	0.005	0.005	
Total	4.019	4.010	4.007	4.010	3.998	4.001		4.007	3.968	4.000	
Mg#	68.46	68.35	69.16	67.81	81.81	82.17		78.87	78.46	78.47	
<i>Molar percentage of endmembers</i>											
Wo	1.16	1.03	0.95	1.61	1.03	1.16		1.25	1.25	1.27	
En	67.67	67.64	68.51	66.72	80.97	81.22		77.89	77.47	77.47	
Fs	31.17	31.33	30.54	31.67	18.00	17.62		20.87	21.28	21.26	
Notes	fine sympl.				rims plag (pt 15)				coarse symp.		

Table A2.2 Orthopyroxenes continued.

Sample	OB93-90				OB93-96			
Point ID	23	25	26	27	28	29		
wt. %								
SiO ₂	52.15	51.32	52.10	52.81	52.90	51.92		
TiO ₂	0.08	0.07	0.09	0.06	0.08	0.07		
Al ₂ O ₃	6.29	7.37	7.28	6.17	5.89	7.29		
Cr ₂ O ₃	0.11	0.07	0.13	0.02	0.03	0.04		
FeO*	12.50	12.11	12.30	12.40	12.02	12.11		
MnO	0.18	0.15	0.19	0.15	0.16	0.19		
MgO	28.01	27.52	27.67	28.41	28.12	27.60		
CaO	0.55	0.56	0.56	0.51	0.53	0.50		
Na ₂ O	0.08	0.04	0.06	0.06	0.04	0.05		
Total	99.93	99.21	100.40	100.57	99.79	99.76		
<i>Cations/6 Oxygens</i>								
Si	1.855	1.835	1.842	1.863	1.877	1.845		
Ti	0.002	0.002	0.003	0.002	0.002	0.002		
Al	0.264	0.311	0.304	0.256	0.246	0.305		
Cr	0.003	0.002	0.004	0.001	0.001	0.001		
Fe	0.372	0.362	0.364	0.366	0.357	0.360		
Mn	0.005	0.005	0.006	0.005	0.005	0.006		
Mg	1.485	1.467	1.458	1.494	1.487	1.461		
Ca	0.021	0.022	0.021	0.019	0.020	0.019		
Na	0.006	0.003	0.004	0.004	0.003	0.003		
Total	4.012	4.008	4.004	4.009	3.999	4.002		
Mg#	79.98	80.20	80.03	80.32	80.65	80.24		
<i>Molar percentage of endmembers</i>								
Wo	1.11	1.16	1.16	1.02	1.08	1.03		
En	79.09	79.27	79.10	79.50	79.78	79.42		
Fs	19.80	19.57	19.74	19.48	19.14	19.55		
Notes	rims cpx	coarse				coarse		
	(pt 17)	symp.				symp.		

Table A2.3. Major element compositions of granulite plagioclase. Each analysis is a separate grain unless otherwise noted.

Sample	OB93-50				OB93-54				OB93-57			
	Point ID	1	3	4, pt 1	4, pt 2	2		5	6	7	8	4
wt. %												
SiO ₂		47.80	47.18	47.63	47.46	47.58	n.d.	49.73	50.00	50.10	50.40	50.78
Al ₂ O ₃		34.07	33.82	33.93	33.90	33.96	n.d.	32.02	31.79	31.41	30.88	31.53
FeO*		0.08	0.15	0.08	0.09	0.11	n.d.	0.06	0.08	0.08	0.20	0.10
MgO		0.03	0.05	0.03	0.03	0.02	n.d.	0.02	0.01	0.01	0.44	0.02
CaO		16.68	15.59	16.56	16.57	16.58	n.d.	14.29	14.01	13.76	13.17	13.74
Na ₂ O		2.00	2.46	2.11	2.05	2.01	n.d.	3.02	3.31	3.38	3.62	3.46
K ₂ O		0.06	0.04	0.04	0.05	0.04	n.d.	0.10	0.11	0.12	0.12	0.10
Total		100.71	99.28	100.38	100.16	100.30		99.24	99.30	98.86	98.83	99.73
<i>Cations/8 Oxygens</i>												
Si		2.175	2.176	2.175	2.173	2.174		2.281	2.292	2.305	2.319	2.314
Al		1.827	1.838	1.826	1.829	1.829		1.731	1.717	1.703	1.674	1.694
Fe		0.003	0.006	0.003	0.004	0.004		0.002	0.003	0.003	0.008	0.004
Mg		0.002	0.004	0.002	0.002	0.002		0.002	0.001	0.001	0.030	0.001
Ca		0.813	0.771	0.810	0.813	0.812		0.702	0.688	0.678	0.649	0.671
Na		0.177	0.220	0.187	0.182	0.178		0.269	0.294	0.302	0.323	0.306
K		0.003	0.002	0.002	0.003	0.002		0.006	0.006	0.007	0.007	0.006
Total		5.001	5.016	5.006	5.006	5.001		4.991	5.000	4.998	5.009	4.995
<i>Mole percentage of endmembers</i>												
Ab		17.79	22.12	18.70	18.25	17.92		27.53	29.74	30.57	32.98	31.12
An		81.89	77.64	81.07	81.42	81.84		71.90	69.62	68.73	66.31	68.28
Or		0.32	0.24	0.23	0.32	0.24		0.57	0.64	0.70	0.72	0.60

n.d.: not determined

Table A2.3. Plagioclase continued.

Sample Point ID	OB93-60				OB93-61A				OB93-68			
	10	12	9	11	1, pt1	1, pt2	2, pt1	2, pt2	19	20	21	22
wt. %												
SiO ₂	47.41	47.04	47.38	47.30	50.26	49.46	49.91	49.14	50.86	50.96	50.14	50.69
Al ₂ O ₃	33.71	33.71	33.32	33.95	31.61	31.87	31.41	32.04	31.36	31.39	31.27	32.02
FeO*	0.10	0.13	0.06	0.08	0.15	0.13	0.14	0.15	0.16	0.10	0.39	0.19
MgO	0.04	0.26	0.05	0.04	0.02	0.01	0.01	0.02	0.03	0.02	1.00	0.05
CaO	16.48	16.55	16.15	16.21	13.94	14.35	13.94	14.48	13.60	13.54	13.81	13.76
Na ₂ O	1.91	1.94	2.23	1.92	3.41	3.28	3.40	3.20	3.69	3.57	3.04	3.40
K ₂ O	0.04	0.04	0.04	0.05	0.32	0.29	0.33	0.26	0.13	0.10	0.24	0.11
Total	99.68	99.68	99.22	99.55	99.71	99.38	99.15	99.30	99.82	99.69	99.88	100.22
<i>Cations/8 Oxygens</i>												
Si	2.179	2.165	2.187	2.175	2.298	2.272	2.296	2.261	2.318	2.322	2.289	2.300
Al	1.826	1.829	1.813	1.839	1.703	1.726	1.703	1.738	1.684	1.686	1.682	1.712
Fe	0.004	0.005	0.002	0.003	0.006	0.005	0.005	0.006	0.006	0.004	0.015	0.007
Mg	0.002	0.018	0.003	0.003	0.002	0.001	0.001	0.001	0.002	0.001	0.068	0.003
Ca	0.811	0.816	0.799	0.798	0.683	0.706	0.687	0.714	0.664	0.661	0.675	0.669
Na	0.170	0.173	0.200	0.171	0.302	0.292	0.304	0.286	0.326	0.316	0.269	0.299
K	0.002	0.002	0.003	0.003	0.019	0.017	0.019	0.015	0.008	0.006	0.014	0.007
Total	4.995	5.009	5.007	4.993	5.011	5.019	5.014	5.021	5.007	4.996	5.011	4.997
<i>Mole percentage of endmembers</i>												
Ab	17.32	17.45	19.95	17.62	30.10	28.75	30.05	28.16	32.66	32.13	28.06	30.72
An	82.46	82.30	79.80	82.06	68.03	69.60	68.03	70.34	66.56	67.28	70.50	68.61
Or	0.22	0.24	0.25	0.32	1.86	1.66	1.92	1.50	0.78	0.59	1.44	0.68

Table A2.3. Plagioclase continued.

Sample Point ID	OB93-69			OB93-71			OB93-72			OB93-90			OB93-96		
	16	18	17				13	14	15	13	14	15	16	17	18
wt. %															
SiO ₂	49.40	50.05	49.69	n.d.			47.60	47.79	47.90	47.95	48.25	47.85	47.48	47.93	47.46
Al ₂ O ₃	32.41	32.12	32.27	n.d.			33.32	33.22	33.37	33.40	33.39	33.52	33.88	34.17	34.08
FeO*	0.17	0.12	0.15	n.d.			0.08	0.05	0.08	0.06	0.06	0.06	0.11	0.07	0.12
MgO	0.05	0.02	0.03	n.d.			0.03	0.05	0.03	0.03	0.03	0.09	0.03	0.02	0.03
CaO	14.81	14.55	14.53	n.d.			15.97	15.68	15.62	15.63	16.01	15.78	16.40	16.50	16.72
Na ₂ O	3.02	3.01	3.01	n.d.			2.21	2.46	2.30	2.52	2.18	2.27	1.98	1.98	1.99
K ₂ O	0.10	0.08	0.07	n.d.			0.07	0.06	0.06	0.06	0.07	0.08	0.04	0.05	0.03
Total	99.95	99.94	99.74				99.27	99.31	99.34	99.64	99.98	99.64	99.91	100.71	100.44
<i>Cations/8 Oxygens</i>															
Si	2.256	2.281	2.270				2.195	2.202	2.204	2.201	2.207	2.196	2.177	2.179	2.167
Al	1.744	1.725	1.737				1.811	1.804	1.809	1.807	1.800	1.813	1.831	1.831	1.834
Fe	0.006	0.005	0.006				0.003	0.002	0.003	0.002	0.002	0.002	0.004	0.003	0.005
Mg	0.003	0.001	0.002				0.002	0.004	0.002	0.002	0.002	0.006	0.002	0.001	0.002
Ca	0.725	0.711	0.711				0.789	0.774	0.770	0.769	0.785	0.776	0.806	0.804	0.818
Na	0.268	0.266	0.267				0.198	0.220	0.205	0.224	0.193	0.202	0.176	0.175	0.176
K	0.006	0.004	0.004				0.004	0.003	0.003	0.003	0.004	0.004	0.002	0.003	0.002
Total	5.008	4.992	4.997				5.001	5.008	4.996	5.009	4.992	5.000	4.997	4.995	5.005
<i>Mole percentage of endmembers</i>															
Ab	26.82	27.10	27.14				19.98	22.06	20.96	22.49	19.67	20.58	17.85	17.78	17.68
An	72.61	72.45	72.41				79.62	77.61	78.70	77.17	79.90	78.97	81.91	81.92	82.14
Or	0.57	0.45	0.45				0.39	0.33	0.34	0.34	0.43	0.45	0.23	0.30	0.18

Table A2.4. Major element compositions of granulite oxides. Each analysis is a separate grain, unless noted by pt1, pt2, etc.

Sample	OB93-50										OB93-54									
Point ID	1	2	3	4	1	9	10	2	3	4	6	7	8							
wt. %																				
SiO ₂	0.07	0.06	0.02	0.09	0.22	0.16	0.08	0.01	0.00	0.09	0.00	0.00	0.07							
TiO ₂	0.01	0.02	0.01	0.05	0.48	0.34	8.24	20.86	50.46	20.98	20.15	49.95	20.45							
Al ₂ O ₃	66.30	65.05	64.72	66.27	44.06	46.08	5.20	2.10	0.20	3.21	2.44	0.18	3.55							
Cr ₂ O ₃	0.32	0.34	0.99	0.43	0.06	0.05	1.32	0.49	0.02	0.40	0.73	0.08	0.43							
FeO*	15.69	15.37	16.12	15.79	46.61	45.30	77.96	70.49	43.80	69.15	71.24	45.54	70.65							
MnO	0.09	0.10	0.08	0.08	0.56	0.52	0.21	0.47	0.41	0.39	0.34	0.27	0.33							
MgO	18.66	18.76	18.00	18.61	5.66	6.08	1.68	2.75	4.24	2.87	2.02	3.09	2.22							
CaO	0.02	0.12	0.02	0.00	0.12	0.14	0.00	0.00	0.00	0.04	0.00	0.00	0.01							
NiO	0.30	0.31	0.32	0.31	0.00	0.02	0.09	0.04	0.00	0.00	0.05	0.02	0.05							
Total	101.46	100.12	100.27	101.62	97.77	98.68	94.78	97.21	99.14	97.13	96.98	99.11	97.76							
FeO	14.16	13.25	14.47	14.35	29.98	29.71	36.66	46.06	37.41	46.29	46.62	39.14	47.08							
Fe ₂ O ₃	1.70	2.36	1.83	1.60	18.49	17.33	45.90	27.15	7.10	25.40	27.36	7.12	26.19							
New Total	101.63	100.36	100.45	101.78	99.63	100.42	99.38	99.93	99.85	99.68	99.72	99.82	100.39							
Cations/3 or 4 Oxygens																				
Si	0.002	0.002	0.001	0.002	0.007	0.005	0.004	0.000	0.000	0.004	0.000	0.000	0.003							
Ti	0.000	0.000	0.000	0.001	0.011	0.008	0.272	0.633	0.952	0.629	0.616	0.951	0.613							
Al	1.965	1.955	1.952	1.963	1.634	1.672	0.269	0.100	0.006	0.151	0.117	0.005	0.167							
Cr	0.006	0.007	0.020	0.008	0.002	0.001	0.046	0.016	0.000	0.013	0.024	0.002	0.014							
Fe	0.330	0.328	0.345	0.332	1.226	1.167	2.857	2.378	0.919	2.305	2.421	0.964	2.354							
Mn	0.002	0.002	0.002	0.002	0.015	0.014	0.008	0.016	0.009	0.013	0.012	0.006	0.011							
Mg	0.700	0.713	0.687	0.697	0.265	0.279	0.110	0.165	0.159	0.170	0.122	0.117	0.132							
Ca	0.001	0.003	0.000	0.000	0.004	0.005	0.000	0.000	0.000	0.002	0.000	0.000	0.000							
Ni	0.006	0.006	0.006	0.006	0.000	0.001	0.003	0.001	0.000	0.000	0.002	0.000	0.002							
Cat. Tot.	3.012	3.017	3.013	3.011	3.164	3.151	3.568	3.309	2.045	3.286	3.314	2.045	3.294							
O Tot.	3.984	3.978	3.982	3.985	3.793	3.809	3.364	3.626	2.934	3.652	3.621	2.934	3.643							
Fe ²⁺	0.297	0.281	0.308	0.300	0.748	0.729	1.130	1.408	0.768	1.409	1.435	0.810	1.428							
Fe ³⁺	0.032	0.045	0.035	0.030	0.415	0.382	1.273	0.747	0.131	0.696	0.757	0.133	0.715							
New Cat. Tot.	3.011	3.015	3.012	3.010	3.101	3.095	3.113	3.087	2.025	3.085	3.084	2.024	3.084							
New O Tot.	4.000	4.000	4.000	4.000	4.010	4.009	4.030	4.035	3.008	4.033	4.036	3.009	4.034							
Cr#	0.32	0.35	1.01	0.43	0.10	0.07	14.57	13.63	5.93	7.75	16.77	22.35	7.56							
Mg#Fe ²⁺	70.23	71.74	69.01	69.88	26.20	27.70	8.86	10.51	17.12	10.78	7.87	12.58	8.46							
No. of Cat.	3	3	3	3	3	3	3	3	2	3	3	2	3							
No of O	4	4	4	4	4	4	4	4	3	4	4	3	4							
Notes	opx sympl.	cpx sympl.		opx sympl.	in sympl.	in sympl.	unzoned	Ti-mt core	ilm core	Ti-mt rim	Ti-mt core	ilm core	Ti-mt rim							

Table A2.4. Oxides continued.

Sample	OB93-57				OB93-60				OB93-68				
	4	5	6	7	8	9	10	22	23	24	25		
wt. %													
SiO ₂	0.11	0.45	0.06	0.08	0.03	0.00	0.06	0.11	0.11	0.10	0.18		
TiO ₂	0.12	0.08	0.12	0.14	0.05	0.04	0.00	52.42	50.75	1.76	19.12		
Al ₂ O ₃	62.18	61.78	60.55	60.84	64.77	64.64	65.77	0.24	0.00	47.93	7.86		
Cr ₂ O ₃	0.17	0.22	1.04	0.93	1.07	1.22	0.44	1.06	0.19	5.84	4.33		
FeO*	21.52	21.54	23.26	21.68	15.74	16.22	15.93	39.96	45.31	31.15	60.05		
MnO	0.13	0.13	0.10	0.12	0.11	0.11	0.08	0.46	0.28	0.18	0.43		
MgO	15.20	14.99	14.68	14.75	17.72	17.75	18.05	5.56	2.88	12.65	6.19		
CaO	0.18	0.03	0.03	0.19	0.01	0.01	0.01	0.07	0.09	0.08	0.11		
NiO	0.23	0.26	0.29	0.26	0.30	0.30	0.33	0.18	0.38	0.28	0.26		
Total	99.83	99.48	100.13	98.99	99.80	100.30	100.65	100.07	100.00	99.97	98.53		
FeO	18.29	18.99	19.04	18.46	14.78	14.85	14.71	36.64	39.87	21.61	40.39		
Fe ₂ O ₃	3.58	2.83	4.69	3.58	1.07	1.53	1.35	3.69	6.04	10.60	21.85		
New Total	100.19	99.77	100.60	99.35	99.91	100.45	100.79	100.44	100.60	101.03	100.72		
<i>Cations/3 or 4 Oxygens</i>													
Si	0.003	0.012	0.002	0.002	0.001	0.000	0.001	0.003	0.003	0.003	0.007		
Ti	0.002	0.002	0.002	0.003	0.001	0.001	0.000	0.961	0.957	0.038	0.531		
Al	1.933	1.926	1.900	1.916	1.959	1.951	1.969	0.007	0.000	1.615	0.342		
Cr	0.003	0.005	0.022	0.020	0.022	0.025	0.009	0.020	0.004	0.132	0.126		
Fe	0.475	0.477	0.518	0.485	0.338	0.347	0.338	0.814	0.950	0.745	1.855		
Mn	0.003	0.003	0.002	0.003	0.002	0.002	0.002	0.009	0.006	0.004	0.013		
Mg	0.598	0.591	0.583	0.588	0.678	0.678	0.684	0.202	0.108	0.539	0.341		
Ca	0.005	0.001	0.001	0.005	0.000	0.000	0.000	0.002	0.002	0.002	0.004		
Ni	0.005	0.006	0.006	0.006	0.006	0.006	0.007	0.004	0.008	0.006	0.008		
Cat. Tot.	3.027	3.021	3.035	3.027	3.008	3.011	3.010	2.023	2.038	3.086	3.228		
O Tot.	3.965	3.972	3.954	3.964	3.990	3.985	3.987	2.967	2.944	3.889	3.718		
Fe ²⁺	0.400	0.417	0.419	0.409	0.316	0.317	0.312	0.739	0.821	0.502	1.160		
Fe ³⁺	0.071	0.056	0.093	0.071	0.021	0.029	0.026	0.067	0.112	0.222	0.564		
New Cat. Tot.	3.022	3.018	3.029	3.023	3.007	3.010	3.009	2.013	2.020	3.065	3.097		
New O Tot.	4.001	4.001	4.001	4.001	4.000	4.000	4.000	3.004	3.008	4.004	4.022		
Cr#	0.18	0.24	1.14	1.02	1.09	1.25	0.44	74.51	100.00	7.56	26.98		
Mg#Fe ²⁺	59.91	58.62	58.17	58.97	68.18	68.14	68.69	21.48	11.62	51.77	22.71		
No. of Cat.	3	3	3	3	3	3	3	2	2	3	3		
No of O	4	4	4	4	4	4	4	3	3	4	4		
Notes	cpx sympl.	opx sympl.		opx sympl.		coarse opx sympl.	fine cpx sympl.	ilm	ilm, included in	sp core	Ti-mt, rim of sp core		

Table A2.4. Oxides continued.

Sample	OB93-69					OB93-71				OB93-72		
	15	16, pt1	17, pt2	18	19	20	21	2	13	11	12	13
wt. %												
SiO ₂	0.10	0.08	0.09	0.04	0.11	0.04	0.19	0.02	0.00	0.02	0.00	0.00
TiO ₂	0.53	0.40	0.49	0.28	53.38	0.20	0.54	0.13	0.11	0.02	0.04	0.06
Al ₂ O ₃	56.01	54.91	55.53	52.50	0.70	56.43	60.83	63.47	61.58	64.96	64.42	65.09
Cr ₂ O ₃	2.34	3.00	3.05	7.23	0.60	4.86	0.72	2.75	2.63	1.22	1.20	1.29
FeO*	27.86	28.34	27.78	31.54	38.29	28.22	22.32	16.70	16.86	16.71	16.37	16.69
MnO	0.16	0.19	0.14	0.22	0.30	0.19	0.09	0.10	0.13	0.07	0.11	0.09
MgO	12.74	11.76	12.65	8.94	7.44	10.97	15.60	18.10	17.70	17.04	17.09	17.25
CaO	0.04	0.07	0.10	0.02	0.06	0.02	0.03	0.00	0.00	0.01	0.03	0.10
NiO	0.32	0.25	0.33	0.28	0.21	0.32	0.24	0.28	0.28	0.30	0.29	0.28
Total	100.11	99.00	100.16	101.04	101.09	101.26	100.56	101.56	99.01	100.35	99.54	100.85
FeO	21.64	22.47	21.60	26.87	34.29	24.37	18.42	14.69	14.38	16.02	15.55	15.79
Fe ₂ O ₃	6.91	6.52	6.87	5.19	4.45	4.28	4.34	2.24	2.75	0.77	0.91	1.00
New Total	100.80	99.66	100.84	101.56	101.53	101.68	100.99	101.78	99.28	100.42	99.63	100.95
<i>Cations/3 or 4 Oxygens</i>												
Si	0.003	0.002	0.002	0.001	0.003	0.001	0.005	0.001	0.000	0.000	0.000	0.000
Ti	0.011	0.008	0.010	0.006	0.956	0.004	0.011	0.003	0.002	0.000	0.001	0.001
Al	1.814	1.809	1.802	1.743	0.020	1.818	1.889	1.906	1.901	1.963	1.961	1.957
Cr	0.051	0.066	0.066	0.161	0.011	0.105	0.015	0.055	0.054	0.025	0.025	0.026
Fe	0.640	0.663	0.640	0.743	0.762	0.645	0.492	0.356	0.369	0.358	0.354	0.356
Mn	0.004	0.005	0.003	0.005	0.006	0.004	0.002	0.002	0.003	0.001	0.002	0.002
Mg	0.522	0.490	0.519	0.375	0.264	0.447	0.613	0.688	0.691	0.651	0.658	0.656
Ca	0.001	0.002	0.003	0.001	0.002	0.001	0.001	0.000	0.000	0.000	0.001	0.003
Ni	0.007	0.006	0.007	0.006	0.004	0.007	0.005	0.006	0.000	0.006	0.006	0.006
Cat. Tot.	3.054	3.051	3.053	3.041	2.027	3.033	3.032	3.016	3.020	3.006	3.007	3.007
O Tot.	3.930	3.933	3.930	3.946	2.961	3.956	3.957	3.979	3.973	3.993	3.991	3.990
Fe ²⁺	0.489	0.517	0.489	0.624	0.673	0.551	0.401	0.311	0.313	0.343	0.335	0.336
Fe ³⁺	0.140	0.135	0.140	0.108	0.079	0.087	0.085	0.043	0.054	0.015	0.018	0.019
New Cat. Tot.	3.042	3.040	3.042	3.031	2.017	3.026	3.027	3.014	3.018	3.005	3.006	3.006
New O Tot.	4.002	4.003	4.002	4.003	3.004	4.002	4.001	4.001	4.001	4.000	4.000	4.000
Cr#	2.73	3.53	3.55	8.46	36.46	5.46	0.79	2.82	2.79	1.24	1.24	1.31
Mg#Fe ²⁺	51.65	48.69	51.52	37.55	28.16	44.79	60.42	68.84	68.83	65.51	66.25	66.12
No. of Cat.	3	3	3	3	2	3	3	3	3	3	3	3
No of O	4	4	4	4	3	4	4	4	4	4	4	4
Notes	fine opx			embayed, rimmed by			ilm			cpx sympl.		
	symp			rimmed by			symp.			fine opx sympl.		

Table A2.4. Oxides continued.

Sample	OB93-90				OB93-96			
	26	27	28	29	30	31, pl	32, pl2	
wt. %								
SiO ₂	0.04	0.02	0.04	0.02	0.11	0.03	0.04	
TiO ₂	0.07	0.04	0.03	0.03	0.03	0.05	0.02	
Al ₂ O ₃	65.05	64.87	65.32	65.53	63.73	65.75	65.26	
Cr ₂ O ₃	1.89	2.05	1.64	1.00	0.96	0.76	0.74	
FeO*	16.37	16.30	16.00	15.85	16.19	15.87	15.83	
MnO	0.11	0.08	0.09	0.09	0.09	0.12	0.07	
MgO	17.57	17.69	17.69	17.83	18.93	18.00	18.14	
CaO	0.00	0.02	0.00	0.03	0.02	0.03	0.04	
NiO	0.31	0.29	0.32	0.35	0.31	0.27	0.33	
Total	101.41	101.34	101.13	100.72	100.37	100.88	100.47	
FeO	15.60	15.35	15.35	14.95	13.08	14.85	14.40	
Fe ₂ O ₃	0.85	1.05	0.72	1.00	3.46	1.13	1.59	
New Total	101.50	101.45	101.20	100.82	100.72	101.00	100.63	
<i>Cations/3 or 4 Oxygens</i>								
Si	0.001	0.000	0.001	0.000	0.003	0.001	0.001	
Ti	0.001	0.001	0.001	0.001	0.001	0.001	0.000	
Al	1.945	1.941	1.954	1.963	1.924	1.965	1.959	
Cr	0.038	0.041	0.033	0.020	0.019	0.015	0.015	
Fe	0.347	0.346	0.340	0.337	0.347	0.337	0.337	
Mn	0.002	0.002	0.002	0.002	0.002	0.003	0.001	
Mg	0.665	0.670	0.669	0.676	0.723	0.680	0.689	
Ca	0.000	0.000	0.000	0.001	0.001	0.001	0.001	
Ni	0.006	0.006	0.006	0.007	0.006	0.006	0.007	
Cat. Tot.	3.006	3.008	3.005	3.007	3.025	3.008	3.011	
O Tot.	3.992	3.990	3.993	3.990	3.967	3.989	3.985	
Fe ²⁺	0.330	0.325	0.325	0.317	0.278	0.314	0.306	
Fe ³⁺	0.016	0.020	0.014	0.019	0.066	0.022	0.030	
New Cat. Tot.	3.005	3.007	3.005	3.006	3.022	3.007	3.010	
New O Tot.	4.000	4.000	4.000	4.000	4.001	4.000	4.000	
Cr#	1.91	2.08	1.66	1.01	1.00	0.77	0.75	
Mg#Fe ²⁺	66.79	67.31	67.30	68.06	72.23	68.42	69.26	
No. of Cat.	3	3	3	3	3	3	3	
No of O	4	4	4	4	4	4	4	
Notes								
	opx sympl.		coarse opx sympl.		fine opx sympl.		opx sympl.	

Table A2.5. Trace element compositions of granulite clinopyroxenes. Each analysis is a separate grain, unless noted (e.g. pt1, pt2).
The subscript "s" denotes symplectite.

Sample	OB93-50				OB93-54				OB93-57			
	Point ID	cpx1	cpx3	sym cpx, 50u raster	cpx1	cpx2	cpx3	A	B	cpx1	cpx2	cpx3
<i>ppm</i>												
La		0.26	0.24	0.28	2.00	2.03	2.41			0.41	0.53	0.60
Ce		1.13	1.35	1.26	11.82	12.51	15.90			2.22	2.73	2.76
Nd		1.96	1.86	1.42	16.28	17.23	22.54			2.88	3.12	2.62
Sm		1.32	1.08	0.73	4.73	4.93	6.56			1.51	1.74	1.47
Eu		0.47	0.46	0.41	1.82	2.00	2.84			0.65	0.77	0.73
Dy		1.54	1.41	1.08	2.14	3.07	6.32			2.08	2.54	1.37
Er		0.94	1.02	0.53	0.70	1.06	2.20			1.25	1.56	1.21
Yb		0.78	0.63	0.55	0.71	1.08	2.25			1.16	1.30	0.99
Ti		2555					2804	5464	9415	4001	4266	
V		444					517	414	483	396	419	
Cr		1861					251	219	229	1246	1312	
Sr		17					25	23	22	16	16	
Y		8					16	21	16	12	12	
Zr		9					120	141	142	17	17	
<i>CI Normalized</i>												
La		1.12	1.00	1.21	8.51	8.64	10.28			1.75	2.24	2.56
Ce		1.87	2.24	2.08	19.60	20.73	26.36			3.68	4.53	4.57
Nd		4.34	4.12	3.14	35.98	38.08	49.83			6.36	6.90	5.79
Sm		8.94	7.32	4.98	32.17	33.53	44.61			10.25	11.81	10.02
Eu		8.45	8.13	7.26	32.47	35.77	50.66			11.67	13.66	13.03
Dy		6.33	5.83	4.44	8.81	12.63	26.04			8.58	10.48	5.65
Er		5.93	6.43	3.31	4.38	6.64	13.83			7.88	9.79	7.63
Yb		4.79	3.90	3.36	4.38	6.64	13.87			7.15	8.01	6.07
La/Yb		0.23	0.26	0.36	1.94	1.30	0.74			0.24	0.28	0.42
La/Sm		0.12	0.14	0.24	0.26	0.26	0.23			0.17	0.19	0.26

Table A2.5. Clinopyroxene trace elements continued.

Sample		OB93-61A			OB93-68			OB93-69			
Point ID	OB93-60	cpx1s	cpx2	cpx1	cpx2	cpx3		cpx4	cpx5	cpx1	cpx2
ppm											
La	0.30	0.35		2.38	2.43	4.56		2.39	1.96	0.79	0.97
Ce	1.36	1.50		10.01	9.82	20.64		12.12	9.53	3.57	3.91
Nd	1.87	1.91		7.97	8.01	19.92		14.56	12.25	4.87	4.96
Sm	0.87	1.29		2.47	2.47	7.02		5.27	4.83	1.86	2.36
Eu	0.44	0.61		0.84	0.91	2.27		1.68	1.46	0.74	0.89
Dy	1.68	1.80		2.82	2.87	6.83		5.11	4.70	3.03	3.13
Er	0.85	1.08		1.50	1.47	3.60		2.67	2.21	2.04	1.99
Yb	0.99	0.98		1.48	1.33	3.43		2.48	2.62	1.67	1.82
Ti	2274	2626				6461	8348				
V	323	452				404	387				
Cr	434	1954				1784	1640				
Sr	15	15				33	30				
Y	7	8				24	23				
Zr	8	9				84	94				
CI Normalized											
La	1.29	1.50		10.15	10.34	19.43		10.18	8.33	3.37	4.11
Ce	2.26	2.49		16.60	16.28	34.22		20.09	15.80	5.92	6.48
Nd	4.13	4.22		17.62	17.70	44.04		32.18	27.08	10.76	10.95
Sm	5.93	8.79		16.80	16.79	47.70		35.83	32.83	12.66	16.06
Eu	7.80	10.87		14.93	16.30	40.49		30.04	26.06	13.26	15.84
Dy	6.93	7.43		11.63	11.82	28.14		21.07	19.35	12.48	12.89
Er	5.36	6.80		9.43	9.28	22.66		16.81	13.93	12.84	12.53
Yb	6.11	6.03		9.11	8.20	21.13		15.26	16.10	10.25	11.18
La/Yb	0.21	0.25		1.11	1.26	0.92		0.67	0.52	0.33	0.37
La/Sm	0.22	0.17		0.60	0.62	0.41		0.28	0.25	0.27	0.26

Table A2.5. Clinopyroxene trace elements continued.

Table A2.3. Chondroite-rich meteorite OB93-69											
Sample OB93-69		OB93-71				OB93-72					
Point ID	cpx3 pt1, core	cpx3 pt 2, rim	cpx4	cpx1, pt1	cpx1, pt2	cpx2s	cpx3	symp1	cpx3	cpx1, pt1	cpx1, pt2
ppm											
La	0.33	0.46	0.45	1.16	1.13	1.11	1.28	0.76	0.31	0.52	0.47
Ce	1.55	1.86	1.99	3.63	3.71	3.01	4.05	2.05	1.42	2.39	1.97
Nd	2.78	3.36	3.55	3.06	3.30	2.18	3.47	1.31	1.48	2.46	2.59
Sm	1.82	1.85	1.64	1.24	1.21	0.83	1.50	0.62	0.74	1.47	0.93
Eu	0.76	0.85	0.95	0.57	0.57	0.47	0.59	0.92	0.32	0.67	0.52
Dy	2.72	2.93	3.21	1.83	1.89	1.13	2.18	0.92	1.13	2.56	1.87
Er	1.63	1.54	1.80	0.95	1.07	0.66	1.20	0.47	0.63	1.52	0.90
Yb	1.67	1.65	1.76	1.00	1.03	0.66	1.11	0.44	0.60	1.27	1.07
CI Normalized											
La	1.40	1.97	1.91	4.92	4.80	4.73	5.44	3.24	1.33	2.22	1.99
Ce	2.57	3.08	3.30	6.01	6.16	4.99	6.72	3.40	2.35	3.96	3.26
Nd	6.15	7.42	7.84	6.77	7.29	4.81	7.66	2.89	3.27	5.45	5.74
Sm	12.37	12.61	11.15	8.44	8.21	5.65	10.17	4.23	5.00	9.98	6.35
Eu	13.57	15.25	16.95	10.15	10.19	8.33	10.55	16.38	5.71	11.92	9.29
Dy	11.20	12.07	13.23	7.52	7.78	4.64	8.98	3.80	4.67	10.54	7.69
Er	10.27	9.69	11.31	6.00	6.71	4.16	7.56	2.94	3.99	9.55	5.64
Yb	10.30	10.16	10.85	6.16	6.33	4.09	6.82	2.73	3.67	7.79	6.61
La/Yb	0.14	0.19	0.18	0.80	0.76	1.16	0.80	1.19	0.36	0.29	0.30
La/Sm	0.11	0.16	0.17	0.58	0.58	0.84	0.53	0.77	0.27	0.22	0.31

Table A2.5. Clinopyroxene trace elements continued.

Sample		OB93-96				
Point ID	OB93-90	cpx1	cpx2	cpx1	cpx2s	cpx3
<i>ppm</i>						
La		0.40	0.33	0.25	0.30	0.16
Ce		1.63	1.33	1.22	1.04	0.92
Nd		2.10	1.66	1.49	1.00	1.36
Sm		1.33	1.03	1.33	0.77	0.91
Eu		0.40	0.35	0.49	0.35	0.27
Dy		1.44	1.14	1.55	1.14	1.12
Er		0.75	0.70	0.80	0.63	0.81
Yb		0.77	0.67	0.88	0.51	0.64
Ti		2658	1813	2409	2326	2415
V		235	134	466	442	473
Cr		1640	376	2036	1949	2015
Sr		15	15	20	17	18
Y		8	6	8	7	7
Zr		14	12	10	7	7
<i>CI Normalized</i>						
La		1.70	1.39	1.07	1.30	0.68
Ce		2.70	2.21	2.01	1.73	1.52
Nd		4.63	3.67	3.30	2.22	3.01
Sm		9.04	6.97	9.02	5.21	6.17
Eu		7.06	6.26	8.68	6.21	4.77
Dy		5.93	4.70	6.38	4.71	4.60
Er		4.70	4.38	5.05	3.98	5.07
Yb		4.75	4.14	5.39	3.13	3.97
La/Yb		0.36	0.33	0.20	0.41	0.17
La/Sm		0.19	0.20	0.12	0.25	0.11

Table A2.5. Trace element compositions of granulite clinopyroxenes. Each analysis is a separate grain, unless noted (e.g. pt1, pt2).
The subscript "s" denotes symplectite.

OB93-54														OB93-57			
Sample	OB93-50																
Point ID	cpx1	cpx3	sym cpx, 50u raster	cpx1	cpx2	cpx3	A	B									
ppm																	
La	0.26	0.24	0.28	2.00	2.03	2.41				0.41	0.53		0.60				
Ce	1.13	1.35	1.26	11.82	12.51	15.90				2.22	2.73		2.76				
Nd	1.96	1.86	1.42	16.28	17.23	22.54				2.88	3.12		2.62				
Sm	1.32	1.08	0.73	4.73	4.93	6.56				1.51	1.74		1.47				
Eu	0.47	0.46	0.41	1.82	2.00	2.84				0.65	0.77		0.73				
Dy	1.54	1.41	1.08	2.14	3.07	6.32				2.08	2.54		1.37				
Er	0.94	1.02	0.53	0.70	1.06	2.20				1.25	1.56		1.21				
Yb	0.78	0.63	0.55	0.71	1.08	2.25				1.16	1.30		0.99				
Ti	2555					2804	5464	9415		4001	4266						
V	444					517	414	483		396	419						
Cr	1861					251	219	229		1246	1312						
Sr	17					25	23	22		16	16						
Y	8					16	21	16		12	12						
Zr	9					120	141	142		17	17						
CI Normalized																	
La	1.12	1.00	1.21	8.51	8.64	10.28				1.75	2.24		2.56				
Ce	1.87	2.24	2.08	19.60	20.73	26.36				3.68	4.53		4.57				
Nd	4.34	4.12	3.14	35.98	38.08	49.83				6.36	6.90		5.79				
Sm	8.94	7.32	4.98	32.17	33.53	44.61				10.25	11.81		10.02				
Eu	8.45	8.13	7.26	32.47	35.77	50.66				11.67	13.66		13.03				
Dy	6.33	5.83	4.44	8.81	12.63	26.04				8.58	10.48		5.65				
Er	5.93	6.43	3.31	4.38	6.64	13.83				7.88	9.79		7.63				
Yb	4.79	3.90	3.36	4.38	6.64	13.87				7.15	8.01		6.07				
La/Yb	0.23	0.26	0.36	1.94	1.30	0.74				0.24	0.28		0.42				
La/Sm	0.12	0.14	0.24	0.26	0.26	0.23				0.17	0.19		0.26				

Table A2.5. Clinopyroxene trace elements continued.

Sample	OB93-61A				OB93-68				OB93-69			
	Point ID	cpx1s	cpx2	cpx1	cpx2	cpx3	cpx4	cpx5	cpx1	cpx2		
<i>ppm</i>												
La		0.30	0.35	2.38	2.43	4.56	2.39	1.96	0.79	0.97		
Ce		1.36	1.50	10.01	9.82	20.64	12.12	9.53	3.57	3.91		
Nd		1.87	1.91	7.97	8.01	19.92	14.56	12.25	4.87	4.96		
Sm		0.87	1.29	2.47	2.47	7.02	5.27	4.83	1.86	2.36		
Eu		0.44	0.61	0.84	0.91	2.27	1.68	1.46	0.74	0.89		
Dy		1.68	1.80	2.82	2.87	6.83	5.11	4.70	3.03	3.13		
Er		0.85	1.08	1.50	1.47	3.60	2.67	2.21	2.04	1.99		
Yb		0.99	0.98	1.48	1.33	3.43	2.48	2.62	1.67	1.82		
Ti		2274	2626			6461						
V		323	452			404						
Cr		434	1954			1784						
Sr		15	15			33						
Y		7	8			24						
Zr		8	9			84						
<i>CI Normalized</i>												
La		1.29	1.50	10.15	10.34	19.43	10.18	8.33	3.37	4.11		
Ce		2.26	2.49	16.60	16.28	34.22	20.09	15.80	5.92	6.48		
Nd		4.13	4.22	17.62	17.70	44.04	32.18	27.08	10.76	10.95		
Sm		5.93	8.79	16.80	16.79	47.70	35.83	32.83	12.66	16.06		
Eu		7.80	10.87	14.93	16.30	40.49	30.04	26.06	13.26	15.84		
Dy		6.93	7.43	11.63	11.82	28.14	21.07	19.35	12.48	12.89		
Er		5.36	6.80	9.43	9.28	22.66	16.81	13.93	12.84	12.53		
Yb		6.11	6.03	9.11	8.20	21.13	15.26	16.10	10.25	11.18		
La/Yb		0.21	0.25	1.11	1.26	0.92	0.67	0.52	0.33	0.37		
La/Sm		0.22	0.17	0.60	0.62	0.41	0.28	0.25	0.27	0.26		

Table A2.5. Clinopyroxene trace elements continued.

Sample OB93-69				OB93-71			OB93-72				
Point ID	cpx3 pt1, core	cpx3 pt 2, rim	cpx4	cpx1, pt1	cpx1, pt2	cpx2s	cpx3	symp1	cpx3	cpx1, pt1	cpx1, pt2
ppm											
La	0.33	0.46	0.45	1.16	1.13	1.11	1.28	0.76	0.31	0.52	0.47
Ce	1.55	1.86	1.99	3.63	3.71	3.01	4.05	2.05	1.42	2.39	1.97
Nd	2.78	3.36	3.55	3.06	3.30	2.18	3.47	1.31	1.48	2.46	2.59
Sm	1.82	1.85	1.64	1.24	1.21	0.83	1.50	0.62	0.74	1.47	0.93
Eu	0.76	0.85	0.95	0.57	0.57	0.47	0.59	0.92	0.32	0.67	0.52
Dy	2.72	2.93	3.21	1.83	1.89	1.13	2.18	0.92	1.13	2.56	1.87
Er	1.63	1.54	1.80	0.95	1.07	0.66	1.20	0.47	0.63	1.52	0.90
Yb	1.67	1.65	1.76	1.00	1.03	0.66	1.11	0.44	0.60	1.27	1.07
CI Normalized											
Ti	7213		6611						2692	2758	
V	546		530						230	282	
Cr	2536		2453						1537	1794	
Sr	17		16						14	14	
Y	19		18						8	10	
Zr	25		25						16	17	
La	1.40	1.97	1.91	4.92	4.80	4.73	5.44	3.24	1.33	2.22	1.99
Ce	2.57	3.08	3.30	6.01	6.16	4.99	6.72	3.40	2.35	3.96	3.26
Nd	6.15	7.42	7.84	6.77	7.29	4.81	7.66	2.89	3.27	5.45	5.74
Sm	12.37	12.61	11.15	8.44	8.21	5.65	10.17	4.23	5.00	9.98	6.35
Eu	13.57	15.25	16.95	10.15	10.19	8.33	10.55	16.38	5.71	11.92	9.29
Dy	11.20	12.07	13.23	7.52	7.78	4.64	8.98	3.80	4.67	10.54	7.69
Er	10.27	9.69	11.31	6.00	6.71	4.16	7.56	2.94	3.99	9.55	5.64
Yb	10.30	10.16	10.85	6.16	6.33	4.09	6.82	2.73	3.67	7.79	6.61
La/Yb	0.14	0.19	0.18	0.80	0.76	1.16	0.80	1.19	0.36	0.29	0.30
La/Sm	0.11	0.16	0.17	0.58	0.58	0.84	0.53	0.77	0.27	0.22	0.31

Table A2.5. Clinopyroxene trace elements continued.

Sample OB93-90		OB93-96			
Point ID	cpx1	cpx2	cpx1	cpx2s	cpx3
<i>ppm</i>					
La	0.40	0.33	0.25	0.30	0.16
Ce	1.63	1.33	1.22	1.04	0.92
Nd	2.10	1.66	1.49	1.00	1.36
Sm	1.33	1.03	1.33	0.77	0.91
Eu	0.40	0.35	0.49	0.35	0.27
Dy	1.44	1.14	1.55	1.14	1.12
Er	0.75	0.70	0.80	0.63	0.81
Yb	0.77	0.67	0.88	0.51	0.64
Ti	2658	1813	2409	2326	2415
V	235	134	466	442	473
Cr	1640	376	2036	1949	2015
Sr	15	15	20	17	18
Y	8	6	8	7	7
Zr	14	12	10	7	7
<i>CI Normalized</i>					
La	1.70	1.39	1.07	1.30	0.68
Ce	2.70	2.21	2.01	1.73	1.52
Nd	4.63	3.67	3.30	2.22	3.01
Sm	9.04	6.97	9.02	5.21	6.17
Eu	7.06	6.26	8.68	6.21	4.77
Dy	5.93	4.70	6.38	4.71	4.60
Er	4.70	4.38	5.05	3.98	5.07
Yb	4.75	4.14	5.39	3.13	3.97
La/Yb	0.36	0.33	0.20	0.41	0.17
La/Sm	0.19	0.20	0.12	0.25	0.11

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