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UCRL-50555

**CHEMICAL ANALYSIS OF SAMPLES
FROM INTEROCEANIC CANAL ROUTE 17**

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December 1968

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CHEMICAL ANALYSIS OF SAMPLES FROM INTEROCEANIC CANAL ROUTE 17

Abstract

This report contains data on the chemical analysis of core samples taken along the Atlantic-Pacific Interoceanic Canal Route 17. Route 17 in Panama is one of the principal routes now being considered for excavation of a canal by nuclear means.

This report contains a brief discussion of sampling procedures, analytical

methods, sources of contamination, and the accuracy of analytical results. Lawrence Radiation Laboratory analyzed each of the samples for 84 constituents and a U.S. Army Corps of Engineers laboratory analyzed the samples for major elements. Four sets of analyses from the two laboratories are compared, and the significance of the comparisons is discussed.

Introduction

Chemical analyses were obtained for 39 core samples taken along the Atlantic-Pacific Interoceanic Canal Route 17. These analyses were obtained in support of the Atlantic-Pacific Interoceanic Canal Study Commission. Figure 1 is the map of the route showing the location of the core holes and an outline of the geology of the area.

Chemical analyses obtained under the direction of Lawrence Radiation Laboratory for 84 constituents in each of 39

core samples are given in Appendix A. Appendix B contains the major element analyses obtained by the U.S. Army Engineering Division Laboratory, South Atlantic Corps of Engineers (Corps of Engineers), Marietta, Georgia.¹

¹A. G. Sutton, Jr., Subsurface Geology Data Collection, Raw Data—Laboratory Results, Atlantic-Pacific Interoceanic Canal Study Commission, Balboa Heights, Canal Zone, Memorandum IOCS-FD-60 (1968), Part IV, Chemical Tests, Route 17.

Sampling

Core samples were taken under the direction of the U.S. Army Corps of Engineers. Bentonite mud was used as a coolant for the drill bits. Cores were

protected from water loss in several ways, which included dipping the bare core in paraffin, wrapping the core in cloth and dipping in paraffin, wrapping in plastic

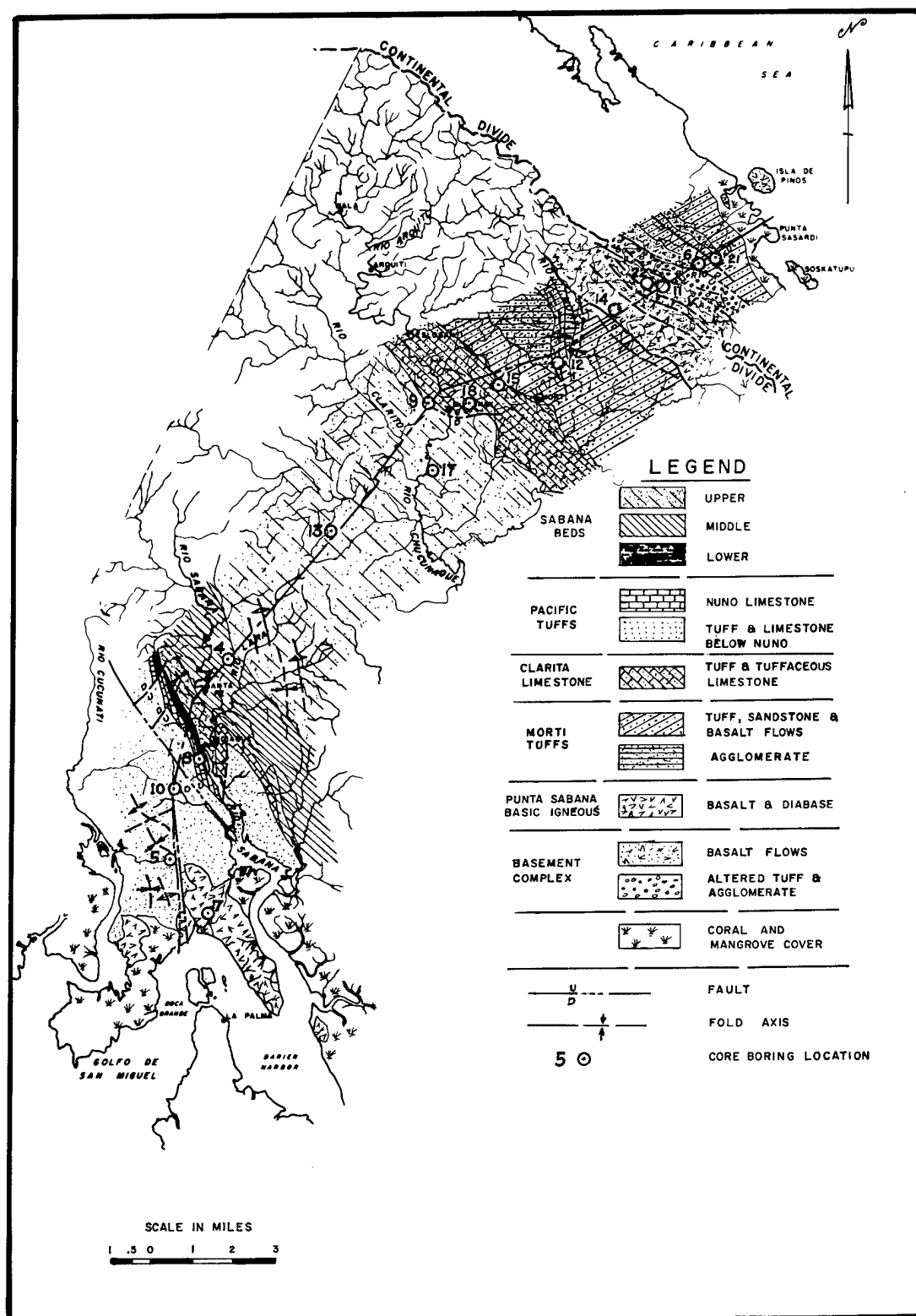


Fig. 1. Geologic map of Atlantic and Pacific Interoceanic Canal Route 17.

and dipping in paraffin, and wrapping in aluminum foil and dipping in paraffin.

The criteria applied in selecting core samples for chemical analysis were as follows²:

1. Select approximately four samples for each major lithologic type likely to occur at shot depths along the Route 17 alignment. (Actual number of samples taken for each rock type ranged from two to seven, depending upon the relative importance of that lithology in the overall stratigraphic section and the degree of compositional variability due to facies changes and/or alteration.)
2. Samples selected should reflect the range of compositional variations within each lithologic type due to facies changes and/or alteration.
3. At least one sample should be taken from each borehole to facilitate recognition of possible compositional changes on a regional scale.
4. Specimens selected for analysis should include those for which other data (physical properties tests, slaking tests, and/or petrographic analyses) are already available.

Samples for chemical analysis were taken from the pieces of core that remained after specimens had been removed for physical properties tests, slaking tests, and petrographic analysis. These pieces of core ranged from 6 to 24 in. in

length. The soft cores of clay and shale were sliced into disks with a knife. The outside edges of the disks were then pared off to remove contamination introduced by the coring operation. All cores of hard rocks, except portions taken for water analysis, were sliced into disks with a diamond wheel. The outside edges of these disks were chipped off with a hammer and chisel to remove contamination from the coring operation.

Samples for water analysis were broken off the core with a hammer and chisel immediately after the core was unwrapped. These samples were promptly crushed in a chipmunk crusher and sealed in plastic bags. Each bagged sample was stored in a sealed paint can until it was analyzed for water.

Two portions of each sample were pulverized for elemental analysis. One portion, pulverized in a chrome-steel unit, was used to obtain the tungsten and cobalt analyses. The second portion, pulverized in a tungsten carbide unit, was used to obtain the iron and chromium analyses.

The procedure for preparing the samples was designed to minimize sample contamination, but some sources of contamination remain. Contamination from minerals in the water used to cool the cutoff saw is difficult to determine but probably negligible, because the cores were generally nonporous. Traces of Fe, Ni, Cr, Cu, Zn, Pb, and Sn could have been introduced from the cutoff saw. The iron contamination from this source is probably negligible compared with the Fe content of the samples, but the Ni, Cr, Cu, Zn, Pb, and Sn results will be biased high.

²P. R. Fisher, U. S. Army Corps of Engineers Nuclear Cratering Group, Livermore, Calif., private communication (1968).

Analysis

All samples were dried at 110°C for all analyses except the free- and bound-water analyses.

The free-water content of six samples was determined by heating the samples to 110°C in a vacuum. The water was collected and weighed. The accuracy of this method is about ± 5 percent of the free-water content of the sample. However, the free-water content of rock samples is very dependent on sample history. Free water may be either gained or lost in coring and subsequent handling operations. Of the four methods used to protect the cores from water loss, the method of wrapping in aluminum foil and dipping in paraffin is the only one adequate to prevent loss of free water during long-term (1-6 months) storage. All of the aluminum-foil-wrapped cores in this group of 39 samples had been opened prior to sampling for chemical analysis. Therefore, the six analyses for free water were run on samples preserved by one of the other three methods. Results for these six samples are probably low.

The bound-water results show the amount of water evolved by the samples when they were heated between 110 and 1000°C in a vacuum. The water was collected and weighed. In the absence of hydrocarbons, the accuracy of this method is about ± 5 percent of the bound-water content of the sample. At temperatures approaching 1000°C, hydrocarbons in the rock samples can react with metallic oxides to form water. This reaction probably occurred on the 781-ft sample from Hole No. 17-CH-13, because oil was

collected in the water trap during the analysis. When this reaction occurs, the bound water results are biased high. The bound-water content of rock samples is not very dependent on sample history, although overheating during the coring operation can cause a loss of bound water. Except for the sample mentioned above, the bound-water results probably represent the actual bound-water content of the in situ rock.

To determine the ratio of Fe(II) to Fe(III), samples were leached in a solution of HCl and HF under an inert atmosphere. The Fe(II) and Fe(III) contents of the solution were determined by titration. The accuracy of the titration is about ± 1 percent of the Fe(II)-Fe(III) ratio in the solution. For samples which dissolve completely and are free of interfering elements, this method gives results which approximate the Fe(II)-Fe(III) ratio of the in situ rock. However, rock samples seldom dissolve completely in the HCl-HF leach solution and almost always contain some interfering elements.

During the determination of total iron, it became evident that the iron in many of these samples was not leached out of the rock completely by the HCl-HF leach solution. The total iron was determined on separate samples that were completely dissolved in a solution of HNO_3 , HClO_4 , and HF. Table I compares the total iron content of these samples with that of the HCl-HF leach solutions. These data indicate that the iron was not always completely leached out of the rock by the HCl-HF leach solution.

Table I. Comparison of total-iron-analysis results for samples leached in HCl-HF solution and samples completely dissolved in HNO₃-HClO₄-HF solution.

Sample No.	Depth, ft	Total Fe, %		Sample No.	Depth, ft	Total Fe, %	
		Leached	Dissolved			Leached	Dissolved
17-CH-4 ^a	113	2.54	4.59	17-CH-12	179.8-182.3	6.52	6.71
17-CH-4 ^a	385	3.30	4.68	17-CH-12	288.2-289.0	5.80	6.13
17-CH-5 ^a	195.2-196.8	1.33	1.36	17-CH-12	385.6-387.1	6.97	7.18
17-CH-5	284.0-284.7	1.46	1.45	17-CH-12	970	5.15	5.37
17-CH-5	400	1.85	1.83	17-CH-13 ^a	321	4.70	5.56
17-CH-6A	650	3.52	3.46	17-CH-13 ^a	781	4.58	5.48
17-CH-7	104.3-106.8	7.12	7.22	17-CH-14	496.2-497.4 and 500.5-501.2	5.92	6.02
17-CH-7	647.6-648.6	6.86	6.88	17-CH-14	576.5-578.2	4.89	5.06
17-CH-7 ^a	672	1.29	2.25	17-CH-14	601.4-603.9	5.16	5.35
17-CH-8 ^a	260.8-262.7	3.01	3.50	17-CH-14	715	4.98	5.19
17-CH-9 ^a	352	4.08	5.40	17-CH-15 ^a	272.5-274.0, 275.2-275.9, and 278.9-279.2	4.23	4.83
17-CH-9 ^a	460-462	1.26	2.13	17-CH-15 ^a	416.8-417.8 and 418.9-420.2	4.61	4.92
17-CH-9	648-658	2.32	2.73	17-CH-17 ^a	113	4.84	5.72
17-CH-9 ^a	873	3.56	4.35	17-CH-18 ^a	208.2-209.8	2.39	2.52
17-CH-10 ^a	363.7-364.2 and 365.3-366.5	0.53	0.57	17-CH-21	150.5-154.5	5.27	5.37
17-CH-10 ^a	495	1.40	2.14	17-CH-22	500	7.33	7.37
17-CH-11	86.1-89.0	5.60	5.78	17-CH-22	1443	6.10	6.49
17-CH-11	496.5-499.0	6.12	6.24				
17-CH-11	801.8-803.2	6.98	7.05				
17-CH-11	925.8-928.7	6.20	6.21				
17-CH-12	118.4-119.1 and 120.2-120.8	5.11	5.42				

^aContains more than 0.1 percent sulfur.

Many of these samples contained more than 0.1 percent sulfur. Part of this sulfur is probably present as sulfide, which reduces Fe(III) to Fe(II) during the leaching process. This causes results for the Fe(II)-Fe(III) ratio to be biased high.

The analytical results for core samples with less than 0.1 percent sulfur and good agreement between the two analyses

for total iron are probably within 5 percent of the actual Fe(II)-Fe(III) ratio in the samples. Results for samples with more than 0.1 percent sulfur or poor agreement between the two total-iron analyses probably do not represent the actual Fe(II)-Fe(III) ratios of their respective cores.

Methods used to obtain the remaining analyses include wet chemistry, atomic

absorption spectroscopy, emission spectrography, and spark-source mass spectrography. Results are shown in Appendix A.

The samples that were analyzed represent only the 6-24 in. of the core from which they were taken. Their relation to the in situ rock formation depends primarily on the homogeneity of the rock mass, which can be estimated by comparing the major element analyses obtained by the Corps of Engineers (Appendix B) with the corresponding analyses obtained by LRL.

On some samples, the Corps of Engineers analyzed alternate slices of the same core used by LRL. For these samples, the Corps of Engineers lab and LRL sample numbers, the core hole and boring numbers, and the depth of sample correspond. On other samples, the Corps of Engineers used a different section of core than LRL. For these samples, the various numbers and depths are different.

A comparison of the results shown in Table II illustrates the differences in composition that may be expected of samples from rock formations that vary from homogeneous to heterogeneous. The analyses shown for Core Hole 17-CH-13 at a depth of 321 ft typify a homogeneous core sample. The analyses shown for Core Hole 17-CH-13 at a depth of 321 ft typify a homogenous core sample. The analyses shown for Core Hole 17-CH-5 at a depth of 400 ft indicate that the core sample is heterogeneous. The heterogeneity in this sample could be caused by pebbles of one rock type included in a matrix of a different type, by lamination of two or more rock types, or by a transition from one

type of rock to another within the length of the core.

The two samples taken from Core Hole 17-CH-13 at 781 and 789 ft represent a formation that has a fairly uniform composition throughout this interval of depth. For Core Hole 17-CH-10, the sample taken at 500 ft is a distinctly different type of rock than the sample taken at 495 ft. Some knowledge of the geology associated with the core hole is required to determine whether one of these samples represents a matrix and the other represents an included boulder or whether the two samples were taken from different rock formations.

Water analyses for two sets of samples are also given in Table II. The hygroscopic water reported by the Corps of Engineers represents the weight lost by the sample when it was dried in an oven at 105°C. The water of crystallization they report is the water evolved by the sample between 105 and 850°C. The sample was heated in a tube furnace. The water of crystallization was collected on magnesium perchlorate and weighed. For the 400-ft sample from Core Hole 17-CH-5, the difference between the Corps of Engineers hygroscopic water and the LRL free water, as well as the difference between the Corps of Engineers water of crystallization and the LRL bound water, is due to the different analytical techniques. The total water content reported by the two laboratories agrees within the limits of analytical error.

The difference between the Corps of Engineers water of crystallization and the LRL bound water on the 789-ft and 781-ft samples from Core Hole 17-CH-13 is probably due to the difference in analytical

Table II. Comparison of results obtained by the Corps of Engineers and Lawrence Radiation Laboratory. All results are in percent.

Sampling procedure:	Alternate slices of same core sample				Different core samples			
Core hole:	17-CH-13		17-CH-5		17-CH-13		17-CH-10	
Depth, ft:	321		400		789	781	500	495
Laboratory:	C of E ^a	LRL	C of E ^a	LRL	C of E ^a	LRL	C of E ^a	LRL
SiO ₂	47.30	48.37	54.23	49.44	50.71	46.74	71.94	48.56
Al ₂ O ₃	16.14	16.52	14.16	7.28	17.51	16.52	3.10	7.80
Fe ₂ O ₃	7.79	7.95	5.69	2.62	8.48	7.84	1.23	3.06
MnO	0.05	0.06	0.06	0.05	0.95	1.03	0.08	0.10
TiO ₂	0.73	1.25	0.39	0.43	0.05	1.28	0.08	0.30
P ₂ O ₅	1.60	1.37	0.23	0.09	0.25	0.18	0.06	0.06
CaO	6.46	6.32	9.72	18.66	4.77	7.36	10.77	19.22
MgO	3.93	3.00	2.44	1.16	3.40	3.25	0.83	1.69
Na ₂ O	1.60	1.75	1.54	1.27	1.40	1.89	0.42	1.48
K ₂ O	1.55	1.69	0.76	0.90	0.97	1.45	0.28	0.98
CO ₂	5.69	5.90	5.08	12.20	3.60	4.99	9.39	13.30
Total	92.84	94.18	94.30	93.92	92.09	92.53	98.18	96.55
Hygroscopic H ₂ O			1.53		4.78			
Free H ₂ O				4.02		16.96		
H ₂ O of Crystallization			6.18		6.88			
Bound H ₂ O				3.62		4.84		
Total H ₂ O			7.71	7.62	11.66	21.80		

^aCorps of Engineers' results for oxides are corrected to an oven-dried (105°C) basis.

techniques. However, the difference between the Corps of Engineers hygroscopic water and the LRL free water, together with the difference in total water found by the two laboratories, indicates that the two samples actually had different water contents. Because the samples were taken 8 ft apart, it is possible that the

actual water content of the in situ rock differed by a factor of 1.87. However, it is more probable that the difference in the total water content of the two samples is the difference in the amount of water lost by the two samples due to poor packaging techniques.

Acknowledgments

The analyses presented in this report represent the cooperative efforts of many people. Dr. Clifford Chunn, U.S. Army Engineering Division Laboratory, South Atlantic Corps of Engineers, Marietta, Georgia, provided the facilities and personnel for sampling the cores. P. R. Fisher, U.S. Army Engineers Nuclear Cratering Group, Livermore, and D. R. Stephens, Lawrence Radiation Laboratory, Livermore, obtained missing core samples in Panama. The samples

were analyzed by W. G. Boyle, L. J. Gregory, R. Lim, C. G. Morris, William F. Morris, R. L. Morrison, William E. Sunderland, and E. G. Walters, Lawrence Radiation Laboratory, Livermore. The manuscript was critically read by E. H. Fleming, J. W. Frazer, and J. E. Harrar. The author wishes to acknowledge the efforts of the people mentioned above and also to express his appreciation to others who also contributed but are not mentioned.

Appendix A

ELEMENTAL ANALYSIS OF CORE SAMPLES BY LAWRENCE RADIATION LABORATORY

These data were obtained by wet chemistry, atomic absorption spectroscopy, emission spectrography, and spark-source mass spectrography. For the analyses by the first three methods, the approximate accuracy is given as a percentage of the concentration of the element in the sample. Where the accuracy is not indicated, the results were obtained by spark-source mass spectrography. A comparison of results indicates that spark-source mass-spectrographic results generally agree within a factor of 2 with the results obtained by more accurate methods. Spark-source mass-spectrographic results that disagree with more accurate results by more than a factor of 2 are usually biased high.

Elements for which no results are given are present in concentrations of less than 1 ppm.

ANALYTICAL REPORT

Calcareous shale
Material with silty interbeds

LRL Sample No. 222/183

17-CH-4 113'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.1 ±
2. He	na ±	±	51. Sb	±	0.05 ±
3. Li	±	90 ± 5%	52. Te	±	±
4. Be	na ±	±	53. I	±	±
5. B	±	70 ± 50%	55. Cs	±	0.8 ±
6. C	na ±	±	56. Ba	±	480 ±
7. N	na ±	±	57. La	±	8 ±
8. O	na ±	±	58. Ce	±	13 ±
9. F	na ±	±	59. Pr	±	2 ±
11. Na	1.0 ± 5%	±	60. Nd	±	6 ±
12. Mg	1.04 ± 2%	±	61. Pm	±	±
13. Al	8.9 ± 2%	±	62. Sm	±	2 ±
14. Si	24.57 ± 2%	±	63. Eu	±	0.4 ±
15. P	0.008 ± 10%	±	64. Gd	±	2 ±
16. S	3.40 ± 5%	±	65. Tb	±	0.3 ±
17. Cl	na ±	±	66. Dy	±	2 ±
19. K	0.81 ± 5%	±	67. Ho	±	±
20. Ca	3.02 ± 2%	±	68. Er	±	1 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.63 ± 5%	±	70. Yb	±	1 ±
23. V	±	250 ±	71. Lu	±	±
24. Cr	±	50 ± 50	72. Hf	±	±
25. Mn	0.04 ± 20%	±	73. Ta	na ±	±
26. Fe	4.59 ± 5%	±	74. W	±	±
27. Co	±	6 ±	75. Re	na ±	±
28. Ni	±	105 ±	76. Os	±	0.3 ±
29. Cu	±	120 ±	77. Ir	±	±
30. Zn	±	50 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	2 ±	81. Tl	±	±
34. Se	±	0.3 ±	82. Pb	±	4 ±
35. Br	±	2 ±	83. Bi	±	±
37. Rb	±	8 ±	84. Po	±	±
38. Sr	±	400 ±	85. At	±	±
39. Y	±	20 ±	86. Rn	±	±
40. Zr	±	55 ±	87. Fr	±	±
41. Nb	±	2 ±	88. Ra	±	±
42. Mo	±	9 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	±	0.1 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	1 ±
46. Pd	±	±	93. Np	±	±
47. Ag	±	0.1 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	0.05 ±	96. Cm	±	±
*CO ₂	2.63 ± 5%	±		±	±
Fe ⁺² /Fe ⁺³	0.75 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

**Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Calcareous shale
17-CH-4 385'LRL Sample No. 111

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.05 ±
2. He	na ±	±	51. Sb	±	0.1 ±
3. Li	±	71 ±	52. Te	±	±
4. Be	na ±	±	53. I	±	2 ±
5. B	±	80 ±	55. Cs	±	2 ±
6. C	na ±	±	56. Ba	±	1450 ±
7. N	na ±	±	57. La	±	14 ±
8. O	na ±	±	58. Ce	±	16 ±
9. F	na ±	±	59. Pr	±	5 ±
11. Na	1.6 ± 5%	±	60. Nd	±	2 ±
12. Mg	1.82 ± 2%	±	61. Pm	±	±
13. Al	8.90 ± 2%	±	62. Sm	±	7 ±
14. Si	25.14 ± 2%	±	63. Eu	±	1 ±
15. P	0.049 ± 5%	±	64. Gd	±	5 ±
16. S	2.45 ± 5%	±	65. Tb	±	0.9 ±
17. Cl	na ±	±	66. Dy	±	4 ±
19. K	0.97 ± 5%	±	67. Ho	±	0.4 ±
20. Ca	3.54 ± 2%	±	68. Er	±	2.5 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.55 ± 5%	±	70. Yb	±	2.4 ±
23. V	±	450 ±	71. Lu	±	0.4 ±
24. Cr	±	60 ±	72. Hf	±	2 ±
25. Mn	0.06 ± 5%	±	73. Ta	na ±	±
26. Fe	4.68 ± 5%	±	74. W	±	1.5 ±
27. Co	±	38 ±	75. Re	na ±	±
28. Ni	±	210 ±	76. Os	±	±
29. Cu	±	115 ±	77. Ir	±	±
30. Zn	±	110 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	4 ±	81. Tl	±	0.4 ±
34. Se	±	18 ±	82. Pb	±	2 ±
35. Br	±	6 ±	83. Bi	±	±
37. Rb	±	14 ±	84. Po	±	±
38. Sr	±	760 ±	85. At	±	±
39. Y	±	24 ±	86. Rn	±	±
40. Zr	±	75 ±	87. Fr	±	±
41. Nb	±	4 ±	88. Ra	±	±
42. Mo	±	44 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	1 ±
44. Ru	±	0.05 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	3 ±
46. Pd	±	0.2 ±	93. Np	±	±
47. Ag	±	0.1 ±	94. Pu	±	±
48. Cd	±	0.7 ±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂	3.69 ± 5%	±		±	±
Fe ⁻² /Fe ⁺³	0.87 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

**Expressed as a percent of the concentration.

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Slightly weathered silty
Material and tuffaceous limestoneLRL Sample No. 1 M 1200

17-CH-5 195.2-196.8'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	±
2. He	na ±	±	51. Sb	±	< 0.05 ±
3. Li	±	11 ±10%	52. Te	±	±
4. Be	na ±	±	53. I	±	0.2 ±
5. B	±	5 ±	55. Cs	±	0.05 ±
6. C	na ±	±	56. Ba	±	2050 ±
7. N	na ±	±	57. La	±	3 ±
8. O	na ±	±	58. Ce	±	3 ±
9. F	na ±	±	59. Pr	±	.8 ±
11. Na	0.90 ± 5%	±	60. Nd	±	3 ±
12. Mg	0.58 ± 2%	±	61. Pm	±	±
13. Al	3.47 ± 2%	±	62. Sm	±	2 ±
14. Si	27.42 ± 2%	±	63. Eu	±	0.5 ±
15. P	0.006 ± 20%	±	64. Gd	±	1 ±
16. S	0.101 ± 5%	±	65. Tb	±	0.4 ±
17. Cl	na ±	±	66. Dy	±	0.9 ±
19. K	0.64 ± 5%	±	67. Ho	±	0.2 ±
20. Ca	10.66 ± 2%	±	68. Er	±	±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.16 ± 5%	±	70. Yb	±	0.8 ±
23. V	±	40 ±	71. Lu	±	±
24. Cr	±	10 ±50%	72. Hf	±	±
25. Mn	0.04 ± 20%	±	73. Ta	na ±	±
26. Fe	1.33 ± 5%	±	74. W	±	2 ±
27. Co	±	13 ±	75. Re	na ±	±
28. Ni	±	1 ±	76. Os	±	±
29. Cu	±	65 ±	77. Ir	±	±
30. Zn	±	1 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.05 ±	81. Tl	±	±
34. Se	±	0.3 ±	82. Pb	±	0.8 ±
35. Br	±	0.5 ±	83. Bi	±	±
37. Rb	±	1 ±	84. Po	±	±
38. Sr	±	225 ±	85. At	±	±
39. Y	±	4 ±	86. Rn	±	±
40. Zr	±	11 ±	87. Fr	±	±
41. Nb	±	0.2 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	< 0.05 ±
44. Ru	±	< 0.05 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	< 0.05 ±	93. Np	±	±
47. Ag	±	< 0.05 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	±	96. Cm	±	±
*CO ₂ /Fe ⁺² /Fe ⁺³	9.89 ± 5%	±		±	±
	0.35 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Altered tuffLRL Sample No. 115

17-CH-5

262-263'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	< 0.05 ±
2. He	na ±	±	51. Sb	±	0.05 ±
3. Li	±	10 ± 10%	52. Te	±	±
4. Be	na ±	±	53. I	±	0.9 ±
5. B	±	15 ± 50%	55. Cs	±	0.2 ±
6. C	na ±	±	56. Ba	±	3200 ±
7. N	na ±	±	57. La	±	8 ±
8. O	na ±	±	58. Ce	±	5 ±
9. F	na ±	±	59. Pr	±	2 ±
11. Na	0.74 ± 5%	±	60. Nd	±	6 ±
12. Mg	0.42 ± 2%	±	61. Pm	±	±
13. Al	2.72 ± 2%	±	62. Sm	±	4 ±
14. Si	29.69 ± 2%	±	63. Eu	±	1 ±
15. P	0.025 ± 5%	±	64. Gd	±	1 ±
16. S	0.074 ± 5%	±	65. Tb	±	0.5 ±
17. Cl	na ±	±	66. Dy	±	1 ±
19. K	0.67 ± 5%	±	67. Ho	±	0.3 ±
20. Ca	9.93 ± 2%	±	68. Er	±	0.7 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.11 ± 5%	±	70. Yb	±	0.8 ±
23. V	±	80 ±	71. Lu	±	±
24. Cr	±	15 ± 50%	72. Hf	±	< 0.05 ±
25. Mn	0.02 ± 20%	±	73. Ta	na ±	±
26. Fe	0.95 ± 5%	±	74. W	±	1.5 ±
27. Co	±	7 ±	75. Re	na ±	±
28. Ni	±	4 ±	76. Os	±	±
29. Cu	±	24 ±	77. Ir	±	±
30. Zn	±	41 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.05 ±	81. Tl	±	±
34. Se	±	0.3 ±	82. Pb	±	0.6 ±
35. Br	±	0.3 ±	83. Bi	±	±
37. Rb	±	2 ±	84. Po	±	±
38. Sr	±	330 ±	85. At	±	±
39. Y	±	0.8 ±	86. Rn	±	±
40. Zr	±	14 ±	87. Fr	±	±
41. Nb	±	0.2 ±	88. Ra	±	±
42. Mo	±	0.05 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	±	2 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	0.05 ±	93. Np	±	±
47. Ag	±	0.05 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂ +3	9.45 ± 5%	±		±	±
Fe ⁺² /Fe ⁺³	0.46 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

**Expressed as a percent of the concentration
na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Fresh Calcareous TuffLRL Sample No. LM 100617-CH-5 284.0-284.7'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na	± **	50. Sn	±	±
2. He	na	±	51. Sb	±	±
3. Li	±	11	52. Te	±	±
4. Be	na	±	53. I	±	0.3
5. B	±	10	55. Cs	±	0.2
6. C	na	±	56. Ba	±	280
7. N	na	±	57. La	±	8
8. O	na	±	58. Ce	±	9
9. F	na	±	59. Pr	±	2.5
11. Na	1.2	± 5%	60. Nd	±	10
12. Mg	0.66	± 2%	61. Pm	±	±
13. Al	4.23	± 2%	62. Sm	±	5
14. Si	29.14 ^q	± 2%	63. Eu	±	2
15. P	0.017	± 5%	64. Gd	±	3
16. S	0.033	± 5%	65. Tb	±	0.9
17. Cl	na	±	66. Dy	±	2.5
19. K	0.99	± 5%	67. Ho	±	0.5
20. Ca	6.99	± 2%	68. Er	±	2
21. Sc	na	±	69. Tm	na	±
22. Ti	0.25	± 5%	70. Yb	±	2
23. V	±	40	71. Lu	±	0.3
24. Cr	±	10	72. Hf	±	1.5
25. Mn	0.02	± 20%	73. Ta	na	±
26. Fe	1.46	± 5%	74. W	±	3
27. Co	±	12	75. Re	na	±
28. Ni	±	9	76. Os	±	±
29. Cu	±	8	77. Ir	±	±
30. Zn	±	17	78. Pt	±	±
31. Ga	na	±	79. Au	±	±
32. Ge	na	±	80. Hg	±	±
33. As	±	0.2	81. Tl	±	±
34. Se	±	0.3	82. Pb	±	2
35. Br	±	0.5	83. Bi	±	±
37. Rb	±	3	84. Po	±	±
38. Sr	±	230	85. At	±	±
39. Y	±	30	86. Rn	±	±
40. Zr	±	25	87. Fr	±	±
41. Nb	±	0.5	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	0.5
44. Ru	±	10	91. Pa	±	±
45. Rh	na	±	92. U	±	±
46. Pd	±	±	93. Np	±	±
47. Ag	±	±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	±	96. Cm	±	±
*CO ₂	5.55	± 5%		±	±
Fe ²⁺ /Fe ³⁺	0.38	± 5%		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Bedded tuff
17-CH-5 400'

LRL Sample No. 113

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	± 0.05	±
2. He	na ±	±	51. Sb	± < 0.05	±
3. Li	±	9 ± 10%	52. Te	±	±
4. Be	na ±	±	53. I	± 0.5	±
5. B	±	5 ±	55. Cs	± 0.3	±
6. C	na ±	±	56. Ba	± 3400	±
7. N	na ±	±	57. La	± 18	±
8. O	na ±	±	58. Ce	± 11	±
9. F	na ±	±	59. Pr	± 3	±
11. Na	0.94 ± 5%	±	60. Nd	± 12	±
12. Mg	0.70 ± 2%	±	61. Pm	±	±
13. Al	3.85 ± 2%	±	62. Sm	± 5	±
14. Si	23.10 ± 2%	±	63. Eu	± 1	±
15. P	0.040 ± 5%	±	64. Gd	± 3	±
16. S	0.022 ± 5%	±	65. Tb	± 0.8	±
17. Cl	na ±	±	66. Dy	± 3	±
19. K	0.75 ± 5%	±	67. Ho	± 0.5	±
20. Ca	13.34 ± 2%	±	68. Er	± 1.5	±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.26 ± 5%	±	70. Yb	± 2	±
23. V	±	85 ±	71. Lu	± 0.3	±
24. Cr	±	10 ± 50%	72. Hf	± 2	±
25. Mn	0.04 ± 20%	±	73. Ta	na ±	±
26. Fe	1.85 ± 5%	±	74. W	± 6	±
27. Co	±	28 ±	75. Re	na ±	±
28. Ni	±	6 ±	76. Os	±	±
29. Cu	±	40 ±	77. Ir	±	±
30. Zn	±	60 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.3 ±	81. Tl	±	±
34. Se	±	0.5 ±	82. Pb	± 2	±
35. Br	±	0.5 ±	83. Bi	±	±
37. Rb	±	4 ±	84. Po	±	±
38. Sr	±	415 ±	85. At	±	±
39. Y	±	30 ±	86. Rn	±	±
40. Zr	±	45 ±	87. Fr	±	±
41. Nb	±	0.8 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	± 0.5	±
44. Ru	±	4 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	± 0.5	±
46. Pd	±	< 0.05 ±	93. Np	±	±
47. Ag	±	0.1 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	0.05 ±	96. Cm	±	±
Fe ⁺² *CO ₂ +3	12.2 ± 5%	±	Free H ₂ O	4.02 ± 5%	±
Fe ⁺² /Fe	0.35 ± 5%	±	Bound H ₂ O	3.62 ± 5%	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

RL-3635

ANALYTICAL REPORT

Material Porphyritic basaltLRL Sample No. 12517-CH-6A 650'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.05 ±
2. He	na ±	±	51. Sb	±	< 0.05 ±
3. Li	±	5 ± 20%	52. Te	±	±
4. Be	±	0.5 ±	53. I	±	±
5. B	±	< 5 ±	55. Cs	±	0.8 ±
6. C	na ±	±	56. Ba	±	450 ± 50%
7. N	na ±	±	57. La	±	8 ± 50%
8. O	na ±	±	58. Ce	±	20 ±
9. F	na ±	±	59. Pr	±	5 ±
11. Na	2.5 ± 5%	±	60. Nd	±	15 ±
12. Mg	0.93 ± 2%	±	61. Pm	±	±
13. Al	7.04 ± 2%	±	62. Sm	±	6 ±
14. Si	29.56 ± 2%	±	63. Eu	±	0.9 ±
15. P	0.082 ± 5%	±	64. Gd	±	4 ±
16. S	0.022 ± 5%	±	65. Tb	±	0.7 ±
17. Cl	na ±	±	66. Dy	±	4 ±
19. K	0.47 ± 5%	±	67. Ho	±	0.7 ±
20. Ca	2.57 ± 2%	±	68. Er	±	2.5 ±
21. Sc	±	15 ± 50%	69. Tm	na ±	±
22. Ti	0.68 ± 5%	4250 ±	70. Yb	±	2 ±
23. V	±	20 ± 50%	71. Lu	±	< 0.05 ±
24. Cr	±	< 1 ±	72. Hf	±	3 ±
25. Mn	0.06 ± 20%	±	73. Ta	na ±	±
26. Fe	3.52 ± 5%	±	74. W	±	±
27. Co	±	8 ±	75. Re	na ±	±
28. Ni	±	< 1 ±	76. Os	±	±
29. Cu	±	50 ± 50%	77. Ir	±	±
30. Zn	±	40 ± 50%	78. Pt	±	±
31. Ga	±	< 10 ± 50%	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.05 ±	81. Tl	±	±
34. Se	±	0.4 ±	82. Pb	±	0.7 ±
35. Br	±	0.8 ±	83. Bi	±	±
37. Rb	±	1.4 ±	84. Po	±	±
38. Sr	±	250 ± 50%	85. At	±	±
39. Y	±	20 ± 50%	86. Rn	±	±
40. Zr	±	200 ± 50%	87. Fr	±	±
41. Nb	±	2 ±	88. Ra	±	±
42. Mo	±	< 0.05 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	0.5 ±
44. Ru	±	< 0.05 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	0.2 ±	93. Np	±	±
47. Ag	±	0.1 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
* CO ₂	0.07 ± 10%	±		±	±
Fe ⁺² /Fe ⁺²⁺³	0.26 ± 5%	±		±	±

Method of Analysis:

* CO₂ from carbonate** Expressed as a percent of the concentration
na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Slightly weathered
Material and altered diabase
17-CH-7 104.3-106.8'

LRL Sample No. 1 M 1089

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.1 ±
2. He	na ±	±	51. Sb	±	±
3. Li	±	4 ± 20%	52. Te	±	1 ±
4. Be	±	< 0.5 ±	53. I	±	±
5. B	±	< 5 ±	55. Cs	±	0.3 ±
6. C	na ±	±	56. Ba	±	70. ± 50%
7. N	na ±	±	57. La	±	0.9 ±
8. O	na ±	±	58. Ce	±	3 ±
9. F	na ±	±	59. Pr	±	0.3 ±
11. Na	1.8 ± 5%	±	60. Nd	±	3 ±
12. Mg	4.95 ± 2%	±	61. Pm	±	±
13. Al	8.00 ± 2%	±	62. Sm	±	3 ±
14. Si	22.17 ± 2%	±	63. Eu	±	0.4 ±
15. P	0.0004 ± 50%	±	64. Gd	±	3 ±
16. S	0.027 ± 5%	±	65. Tb	±	0.5 ±
17. Cl	na ±	±	66. Dy	±	1 ±
19. K	0.056 ± 5%	±	67. Ho	±	0.3 ±
20. Ca	9.28 ± 2%	±	68. Er	±	0.3 ±
21. Sc	±	25 ± 50%	69. Tm	na ±	±
22. Ti	0.90 ± 5%	±	70. Yb	±	1.7 ±
23. V	±	200 ± 50%	71. Lu	±	±
24. Cr	±	200 ± 50%	72. Hf	±	±
25. Mn	0.09 ± 10%	±	73. Ta	na ±	±
26. Fe	7.12 ± 5%	±	74. W	±	±
27. Co	±	±	75. Re	na ±	±
28. Ni	±	70 ± 50%	76. Os	±	±
29. Cu	±	120 ± 50%	77. Ir	±	±
30. Zn	±	40 ± 50%	78. Pt	±	±
31. Ga	±	20 ± 50%	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	1 ±	81. Tl	±	±
34. Se	±	2 ±	82. Pb	±	0.6 ±
35. Br	±	1 ±	83. Bi	±	±
37. Rb	±	1 ±	84. Po	±	±
38. Sr	±	75 ± 50%	85. At	±	±
39. Y	±	10 ± 50%	86. Rn	±	±
40. Zr	±	75 ± 50%	87. Fr	±	±
41. Nb	±	1 ±	88. Ra	±	±
42. Mo	±	0.7 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	±	±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	±	93. Np	±	±
47. Ag	±	±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	3 ±	96. Cm	±	±
*CO ₂	0.09 ± 10%	±		±	±
Fe ⁺² /Fe ⁺³	1.89 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Slightly weathered
Material altered diabase
17-CH-7 647.6 - 648.6'

LRL Sample No. 1M 1093

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.2 ±
2. He	na ±	±	51. Sb	±	±
3. Li	±	6 ± 20%	52. Te	±	±
4. Be	na ±	±	53. I	±	±
5. B	±	20 ± 50%	55. Cs	±	±
6. C	na ±	±	56. Ba	±	1600 ±
7. N	na ±	±	57. La	±	3 ±
8. O	na ±	±	58. Ce	±	5 ±
9. F	na ±	±	59. Pr	±	0.8 ±
11. Na	1.6 ± 5%	±	60. Nd	±	3 ±
12. Mg	4.81 ± 2%	±	61. Pm	±	±
13. Al	7.74 ± 2%	±	62. Sm	±	5 ±
14. Si	21.59 ± 2%	±	63. Eu	±	0.8 ±
15. P	0.006 ± 20%	±	64. Gd	±	3 ±
16. S	0.056 ± 5%	±	65. Tb	±	1 ±
17. Cl	na ±	±	66. Dy	±	6 ±
19. K	0.20 ± 5%	±	67. Ho	±	1 ±
20. Ca	10.03 ± 2%	±	68. Er	±	5 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.86 ± 5%	±	70. Yb	±	5 ±
23. V	±	250 ±	71. Lu	±	±
24. Cr	±	200 ± 50%	72. Hf	±	2 ±
25. Mn	0.08 ± 10%	±	73. Ta	na ±	±
26. Fe	6.86 ± 5%	±	74. W	±	7 ±
27. Co	±	25 ±	75. Re	na ±	±
28. Ni	±	75 ±	76. Os	±	±
29. Cu	±	150 ±	77. Ir	±	±
30. Zn	±	250 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.2 ±	81. Tl	±	±
34. Se	±	1 ±	82. Pb	±	1 ±
35. Br	±	2 ±	83. Bi	±	±
37. Rb	±	1 ±	84. Po	±	±
38. Sr	±	100 ±	85. At	±	±
39. Y	±	35 ±	86. Rn	±	±
40. Zr	±	25 ±	87. Fr	±	±
41. Nb	±	0.6 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	±	6 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	±	93. Np	±	±
47. Ag	±	0.2 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	0.2 ±	96. Cm	±	±
*CO ₂	0.11 ± 5%	±		±	±
Fe ⁺² /Fe ⁺³	1.38 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

na - not analyzed

** Expressed as a percent of the concentration

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Calcareous tuff

LRL Sample No. 114

17-CH-7 672'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.1 ±
2. He	na ±	±	51. Sb	±	0.1 ±
3. Li	±	3 ±50%	52. Te	±	±
4. Be	na ±	±	53. I	±	0.3 ±
5. B	±	< 5 ±	55. Cs	±	±
6. C	na ±	±	56. Ba	±	44 ±
7. N	na ±	±	57. La	±	7 ±
8. O	na ±	±	58. Ce	±	2.5 ±
9. F	na ±	±	59. Pr	±	1 ±
11. Na	0.44 ± 5%	±	60. Nd	±	4 ±
12. Mg	0.70 ± 2%	±	61. Pm	±	±
13. Al	1.72 ± 2%	±	62. Sm	±	2 ±
14. Si	21.46 ± 2%	±	63. Eu	±	0.3 ±
15. P	0.041 ± 5%	±	64. Gd	±	1 ±
16. S	1.45 ± 5%	±	65. Tb	±	0.2 ±
17. Cl	na ±	±	66. Dy	±	1.5 ±
19. K	0.046 ± 5%	±	67. Ho	±	0.3 ±
20. Ca	17.50 ± 2%	±	68. Er	±	1 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.28 ± 5%	±	70. Yb	±	1 ±
23. V	±	115 ±	71. Lu	±	< 0.05 ±
24. Cr	±	75 ±50%	72. Hf	±	< 0.05 ±
25. Mn	0.03 ± 20%	±	73. Ta	na ±	±
26. Fe	2.25 ± 5%	±	74. W	±	±
27. Co	±	10 ±	75. Re	na ±	±
28. Ni	±	31 ±	76. Os	±	±
29. Cu	±	48 ±	77. Ir	±	±
30. Zn	±	70 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	2 ±	81. Tl	±	0.1 ±
34. Se	±	0.6 ±	82. Pb	±	1 ±
35. Br	±	0.3 ±	83. Bi	±	±
37. Rb	±	0.05 ±	84. Po	±	±
38. Sr	±	75 ±	85. At	±	±
39. Y	±	20 ±	86. Rn	±	±
40. Zr	±	18 ±	87. Fr	±	±
41. Nb	±	0.5 ±	88. Ra	±	±
42. Mo	±	18 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	±	10 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	0.5 ±
46. Pd	±	±	93. Np	±	±
47. Ag	±	2 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂ /Fe ⁺² /Fe ⁺³	18.0 ± 5%	±		±	±
	2.06 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

RL-3635

ANALYTICAL REPORT

Tuffaceous fossiliferous

Material limestone

IRL Sample No. 1M 1177

17-CH-8 260.8-262.7'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na	± **	50. Sn	±	0.05
2. He	na	±	51. Sb	±	0.05
3. Li	±	11	52. Te	±	±
4. Be	na	±	53. I	±	0.8
5. B	±	5	55. Cs	±	0.5
6. C	na	±	56. Ba	±	640
7. N	na	±	57. La	±	14
8. O	na	±	58. Ce	±	11
9. F	na	±	59. Pr	±	3
11. Na	0.52	± 5%	60. Nd	±	9
12. Mg	1.26	± 2%	61. Pm	±	±
13. Al	3.07	± 2%	62. Sm	±	4
14. Si	10.73	± 2%	63. Eu	±	0.8
15. P	0.077	± 5%	64. Gd	±	2.5
16. S	1.10	± 5%	65. Tb	±	0.5
17. Cl	na	±	66. Dy	±	3
19. K	0.76	± 5%	67. Ho	±	0.4
20. Ca	24.44	± 2%	68. Er	±	1
21. Sc	na	±	69. Tm	na	±
22. Ti	0.70	± 5%	70. Yb	±	1
23. V	±	115	71. Lu	±	<0.05
24. Cr	±	45	72. Hf	±	2
25. Mn	0.02	± 20%	73. Ta	na	±
26. Fe	3.50	± 5%	74. W	±	±
27. Co	±	115	75. Re	na	±
28. Ni	±	30	76. Os	±	±
29. Cu	±	35	77. Ir	±	±
30. Zn	±	1.5	78. Pt	±	±
31. Ga	na	±	79. Au	±	±
32. Ge	na	±	80. Hg	±	±
33. As	±	0.3	81. Tl	±	<0.05
34. Se	±	0.3	82. Pb	±	2
35. Br	±	0.9	83. Bi	±	±
37. Rb	±	9	84. Po	±	±
38. Sr	±	900	85. At	±	±
39. Y	±	24	86. Rn	±	±
40. Zr	±	42	87. Fr	±	±
41. Nb	±	5	88. Ra	±	±
42. Mo	±	4	89. Ac	±	±
43. Tc	±	±	90. Th	±	<0.05
44. Ru	±	7	91. Pa	±	±
45. Rh	na	±	92. U	±	2
46. Pd	±	<0.05	93. Np	±	±
47. Ag	±	0.3	94. Pu	±	±
48. Cd	±	<0.05	95. Am	±	±
49. In	±	<0.05	96. Cm	±	±
*Co ₂	25.2%	± 5%		±	±
Fe ⁺² /Fe ⁺³	0.57	± 5%		±	±

Method of Analysis:

*Co₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab.

Lab. Sample No.

Date:

Signed:

ANALYTICAL REPORT

Material Calcareous Shale
17-CH-9 352'LRL Sample No. 117

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	±
2. He	na ±	±	51. Sb	±	±
3. Li	±	67 ± 5%	52. Te	±	±
4. Be	na ±	±	53. I	±	0.1 ±
5. B	±	200 ± 50%	55. Cs	±	0.8 ±
6. C	na ±	±	56. Ba	±	290 ±
7. N	na ±	±	57. La	±	2 ±
8. O	na ±	±	58. Ce	±	5 ±
9. F	na ±	±	59. Pr	±	0.7 ±
11. Na	1.4 ± 5%	±	60. Nd	±	2 ±
12. Mg	1.53 ± 2%	±	61. Pm	±	±
13. Al	9.44 ± 2%	±	62. Sm	±	2 ±
14. Si	24.37 ± 2%	±	63. Eu	±	0.3 ±
15. P	0.061 ± 5%	±	64. Gd	±	0.6 ±
16. S	2.19 ± 5%	±	65. Tb	±	0.2 ±
17. Cl	na ±	±	66. Dy	±	0.3 ±
19. K	1.5 ± 5%	±	67. Ho	±	±
20. Ca	3.14 ± 2%	±	68. Er	±	0.4 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.64 ± 5%	±	70. Yb	±	±
23. V	±	110 ±	71. Lu	±	±
24. Cr	±	70 ± 50%	72. Hf	±	±
25. Mn	0.07 ± 20%	±	73. Ta	na ±	±
26. Fe	5.40 ± 5%	±	74. W	±	±
27. Co	±	6 ±	75. Re	na ±	±
28. Ni	±	35 ±	76. Os	±	±
29. Cu	±	50 ±	77. Ir	±	±
30. Zn	±	75 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	1 ±	81. Tl	±	±
34. Se	±	0.3 ±	82. Pb	±	0.8 ±
35. Br	±	2 ±	83. Bi	±	±
37. Rb	±	4 ±	84. Po	±	±
38. Sr	±	165 ±	85. At	±	±
39. Y	±	7 ±	86. Rn	±	±
40. Zr	±	18 ±	87. Fr	±	±
41. Nb	±	0.2 ±	88. Ra	±	±
42. Mo	±	1 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	±	0.9 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	±	93. Np	±	±
47. Ag	±	0.1 ±	94. Pu	±	±
48. Cd	±	< 0.05 ±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂	2.63 ± 5%	±	Free H ₂ O	13.05 ± 5%	±
Fe ²⁺ /Fe ³⁺	0.72 ± 5%	±	Bound H ₂ O	5.02 ± 5%	±

Method of Analysis:

* CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Fresh Tuffaceous Limestone
17-CH-9 460-462'

LRL Sample No. 1M 1043

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.05 ±
2. He	na ±	±	51. Sb	±	0.4 ±
3. Li	±	19 ± 10%	52. Te	±	±
4. Be	na ±	±	53. I	±	2 ±
5. B	±	50 ± 50%	55. Cs	±	0.6 ±
6. C	na ±	±	56. Ba	±	2560 ±
7. N	na ±	±	57. La	±	19 ±
8. O	na ±	±	58. Ce	±	28 ±
9. F	na ±	±	59. Pr	±	3 ±
11. Na	1.2 ± 5%	±	60. Nd	±	12 ±
12. Mg	0.69 ± 2%	±	61. Pm	±	±
13. Al	4.21 ± 2%	±	62. Sm	±	4 ±
14. Si	18.99 ± 2%	±	63. Eu	±	0.6 ±
15. P	0.065 ± 5%	±	64. Gd	±	2.5 ±
16. S	1.47 ± 5%	±	65. Tb	±	0.6 ±
17. Cl	na ±	±	66. Dy	±	4 ±
19. K	0.64 ± 5%	±	67. Ho	±	0.3 ±
20. Ca	15.17 ± 2%	±	68. Er	±	1.5 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.41 ± 5%	±	70. Yb	±	2 ±
23. V	±	230 ±	71. Lu	±	<0.05 ±
24. Cr	±	30 ± 50%	72. Hf	±	2 ±
25. Mn	0.02 ± 20%	±	73. Ta	na ±	±
26. Fe	2.13 ± 5%	±	74. W	±	1.5 ±
27. Co	±	125 ±	75. Re	na ±	±
28. Ni	±	105 ±	76. Os	±	±
29. Cu	±	90 ±	77. Ir	±	±
30. Zn	±	140 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	4 ±	81. Tl	±	±
34. Se	±	0.5 ±	82. Pb	±	2 ±
35. Br	±	2 ±	83. Bi	±	±
37. Rb	±	13 ±	84. Po	±	±
38. Sr	±	1900 ±	85. At	±	±
39. Y	±	42 ±	86. Rn	±	±
40. Zr	±	115 ±	87. Fr	±	±
41. Nb	±	9 ±	88. Ra	±	±
42. Mo	±	9 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	0.5 ±
44. Ru	±	< 0.05 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	3 ±
46. Pd	±	0.05 ±	93. Np	±	±
47. Ag	±	0.3 ±	94. Pu	±	±
48. Cd	±	< 0.05 ±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂	14.3 ± 5%	±		±	±
Fe ⁺² /Fe ⁺³	0.65 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Slightly weathered siliceous shale
17-CH-9 648-658'LRL Sample No. 121

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	< 0.05 ±
2. He	na ±	±	51. Sb	±	< 0.05 ±
3. Li	±	17 ± 10%	52. Te	±	< 0.05 ±
4. Be	na ±	±	53. I	±	0.5 ±
5. B	±	50 ± 50%	55. Cs	±	0.3 ±
6. C	na ±	±	56. Ba	±	935 ±
7. N	na ±	±	57. La	±	8 ±
8. O	na ±	±	58. Ce	±	9 ±
9. F	na ±	±	59. Pr	±	2 ±
11. Na	1.4 ± 5%	±	60. Nd	±	7 ±
12. Mg	0.97 ± 2%	±	61. Pm	±	±
13. Al	4.94 ± 2%	±	62. Sm	±	4 ±
14. Si	20.32 ± 2%	±	63. Eu	±	0.5 ±
15. P	0.064 ± 5%	±	64. Gd	±	2 ±
16. S	0.90 ± 5%	±	65. Tb	±	0.4 ±
17. Cl	na ±	±	66. Dy	±	2.5 ±
19. K	0.87 ± 5%	±	67. Ho	±	0.5 ±
20. Ca	13.89 ± 2%	±	68. Er	±	1 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.56 ± 5%	±	70. Yb	±	2 ±
23. V	±	220 ±	71. Lu	±	0.2 ±
24. Cr	±	30 ± 50%	72. Hf	±	1 ±
25. Mn	0.07 ± 20%	±	73. Ta	na ±	±
26. Fe	2.73 ± 5%	±	74. W	±	±
27. Co	±	45 ±	75. Re	na ±	±
28. Ni	±	20 ±	76. Os	±	±
29. Cu	±	50 ±	77. Ir	±	±
30. Zn	±	45 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	1 ±	81. Tl	±	±
34. Se	±	0.4 ±	82. Pb	±	0.8 ±
35. Br	±	0.9 ±	83. Bi	±	±
37. Rb	±	6 ±	84. Po	±	±
38. Sr	±	590 ±	85. At	±	±
39. Y	±	4 ±	86. Rn	±	±
40. Zr	±	55 ±	87. Fr	±	±
41. Nb	±	3 ±	88. Ra	±	±
42. Mo	±	0.6 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	0.1 ±
44. Ru	±	6 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	0.5 ±
46. Pd	±	0.1 ±	93. Np	±	±
47. Ag	±	0.3 ±	94. Pu	±	±
48. Cd	±	< 0.05 ±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂	13.4 ± 5%	±		±	±
Fe+2/Fe+3	0.42 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Shaley limestoneIRL Sample No. 12217-CH-9 873'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	< 0.05 ±
2. He	na ±	±	51. Sb	±	< 0.05 ±
3. Li	±	28 ± 5%	52. Te	±	±
4. Be	na ±	±	53. I	±	0.4 ±
5. B	±	70 ± 50%	55. Cs	±	0.5 ±
6. C	na ±	±	56. Ba	±	640 ±
7. N	na ±	±	57. La	±	13 ±
8. O	na ±	±	58. Ce	±	19 ±
9. F	na ±	±	59. Pr	±	2.5 ±
11. Na	2.0 ± 5%	±	60. Nd	±	9 ±
12. Mg	1.91 ± 2%	±	61. Pm	±	±
13. Al	6.40 ± 2%	±	62. Sm	±	3 ±
14. Si	25.01 ± 2%	±	63. Eu	±	0.5 ±
15. P	0.088 ± 5%	±	64. Gd	±	2 ±
16. S	1.34 ± 5%	±	65. Tb	±	0.3 ±
17. Cl	na ±	±	66. Dy	±	2 ±
19. K	1.1 ± 5%	±	67. Ho	±	0.4 ±
20. Ca	7.56 ± 2%	±	68. Er	±	1 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.64 ± 5%	±	70. Yb	±	0.8 ±
23. V	±	155 ±	71. Lu	±	< 0.05 ±
24. Cr	±	40 ± 50%	72. Hf	±	< 0.05 ±
25. Mn	0.05 ± 20%	±	73. Ta	na ±	±
26. Fe	4.35 ± 5%	±	74. W	±	< 0.05 ±
27. Co	±	35 ±	75. Re	na ±	±
28. Ni	±	30 ±	76. Os	±	±
29. Cu	±	50 ±	77. Ir	±	±
30. Zn	±	45 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	1 ±	81. Tl	±	< 0.05 ±
34. Se	±	0.4 ±	82. Pb	±	2 ±
35. Br	±	1 ±	83. Bi	±	±
37. Rb	±	3 ±	84. Po	±	±
38. Sr	±	770 ±	85. At	±	±
39. Y	±	16 ±	86. Rn	±	±
40. Zr	±	55 ±	87. Fr	±	±
41. Nb	±	2 ±	88. Ra	±	±
42. Mo	±	0.9 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	0.5 ±
44. Ru	±	1.5 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	0.5 ±
46. Pd	±	< 0.05 ±	93. Np	±	±
47. Ag	±	0.1 ±	94. Pu	±	±
48. Cd	±	< 0.05 ±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂ / Fe ²⁺ / Fe ³⁺	4.97 ± 5%	±		±	±
	0.63 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Fresh Tuffaceous Limestone

LRL Sample No. 1M 1213

17-CH-10 363.7 - 364.2'
and 365.3 - 366.5'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	± 0.7	±
2. He	na ±	±	51. Sb	± 0.1	±
3. Li	±	8 ± 10%	52. Te	±	±
4. Be	na ±	±	53. I	± 1	±
5. B	±	< 5	55. Cs	± 1	±
6. C	na ±	±	56. Ba	± 680	±
7. N	na ±	±	57. La	± 55	±
8. O	na ±	±	58. Ce	± 9	±
9. F	na ±	±	59. Pr	± 9	±
11. Na	0.28 ± 5%	±	60. Nd	± 35	±
12. Mg	0.32 ± 2%	±	61. Pm	±	±
13. Al	0.80 ± 2%	±	62. Sm	± 12	±
14. Si	20.11 ± 2%	±	63. Eu	± 3	±
15. P	0.036 ± 5%	±	64. Gd	± 8	±
16. S	0.20 ± 5%	±	65. Tb	± 1	±
17. Cl	na ±	±	66. Dy	± 8	±
19. K	0.32 ± 5%	±	67. Ho	± 1	±
20. Ca	21.26 ± 2%	±	68. Er	± 3	±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.10 ± 5%	±	70. Yb	± 4	±
23. V	±	110	71. Lu	± 0.2	±
24. Cr	±	< 10	72. Hf	± 0.7	±
25. Mn	0.02 ± 20%	±	73. Ta	na ±	±
26. Fe	0.53 ± 5%	±	74. W	± 0.3	±
27. Co	±	80	75. Re	na ±	±
28. Ni	±	25	76. Os	±	±
29. Cu	±	110	77. Ir	±	±
30. Zn	±	175	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	1.5	81. Tl	±	±
34. Se	±	2.5	82. Pb	± 3	±
35. Br	±	4	83. Bi	±	±
37. Rb	±	13	84. Po	±	±
38. Sr	±	455	85. At	±	±
39. Y	±	65	86. Rn	±	±
40. Zr	±	55	87. Fr	±	±
41. Nb	±	1.5	88. Ra	±	±
42. Mo	±	7	89. Ac	±	±
43. Tc	±	±	90. Th	± 5	±
44. Ru	±	±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	1	93. Np	±	±
47. Ag	±	0.3	94. Pu	±	±
48. Cd	±	2	95. Am	±	±
49. In	±	0.1	96. Cm	±	±
*CO ₂	21.5 ± 5%	±		±	±
Fe+2/Fe+3	0.76 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Fresh tuffaceous
Material limestone

LRL Sample No. 116

17-CH-10 495'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
**					
1. H	na	±	50. Sn	±	< 0.05
2. He	na	±	51. Sb	±	< 0.05
3. Li	±	10	52. Te	±	< 0.05
4. Be	na	±	53. I	±	0.1
5. B	±	5	55. Cs	±	0.5
6. C	na	±	56. Ba	±	1850
7. N	na	±	57. La	±	18
8. O	na	±	58. Ce	±	12
9. F	na	±	59. Pr	±	3
11. Na	1.1	± 5%	60. Nd	±	10
12. Mg	1.02	± 2%	61. Pm	±	±
13. Al	4.13	± 2%	62. Sm	±	4
14. Si	22.69	± 2%	63. Eu	±	0.7
15. P	0.026	± 5%	64. Gd	±	3
16. S	0.13	± 5%	65. Tb	±	0.7
17. Cl	na	±	66. Dy	±	2
19. K	0.81	± 5%	67. Ho	±	0.4
20. Ca	13.74	± 2%	68. Er	±	1
21. Sc	na	±	69. Tm	na	±
22. Ti	0.18	± 5%	70. Yb	±	2
23. V	±	48	71. Lu	±	0.2
24. Cr	±	10	72. Hf	±	0.4
25. Mn	0.08	± 10%	73. Ta	na	±
26. Fe	2.14	± 5%	74. W	±	2
27. Co	±	19	75. Re	na	±
28. Ni	±	5	76. Os	±	±
29. Cu	±	38	77. Ir	±	±
30. Zn	±	120	78. Pt	±	0.6
31. Ga	na	±	79. Au	±	±
32. Ge	na	±	80. Hg	±	±
33. As	±	0.1	81. Tl	±	±
34. Se	±	0.4	82. Pb	±	1
35. Br	±	0.5	83. Bi	±	±
37. Rb	±	4	84. Po	±	±
38. Sr	±	400	85. At	±	±
39. Y	±	53	86. Rn	±	±
40. Zr	±	50	87. Fr	±	±
41. Nb	±	4	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	0.5
44. Ru	±	1.5	91. Pa	±	±
45. Rh	na	±	92. U	±	0.3
46. Pd	±	0.1	93. Np	±	±
47. Ag	±	0.1	94. Pu	±	±
48. Cd	±	< 0.05	95. Am	±	±
49. In	±	< 0.05	96. Cm	±	±
*CO ₂ /Fe ⁺² /Fe ⁺³	13.3	± 5%	Free H ₂ O	4.10	± 5%
	0.95	± 5%	Bound H ₂ O	2.46	± 5%

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab.

Lab. Sample No.

Date:

Signed:

ANALYTICAL REPORT

Fresh to slightly
Material weathered diabase

LRL Sample No. 1M1411

17-CH-11 86.1-89.0'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.05 ±
2. He	na ±	±	51. Sb	±	< 0.05 ±
3. Li	±	10 ± 10%	52. Te	±	< 0.05 ±
4. Be	na ±	±	53. I	±	±
5. B	±	< 5 ±	55. Cs	±	0.1 ±
6. C	na ±	±	56. Ba	±	14 ±
7. N	na ±	±	57. La	±	1 ±
8. O	na ±	±	58. Ce	±	3 ±
9. F	na ±	±	59. Pr	±	0.5 ±
11. Na	2.1 ± 5%	±	60. Nd	±	2.5 ±
12. Mg	6.77 ± 5%	±	61. Pm	±	±
13. Al	8.42 ± 5%	±	62. Sm	±	1 ±
14. Si	22.65 ± 5%	±	63. Eu	±	0.3 ±
15. P	0.008 ± 10%	±	64. Gd	±	0.5 ±
16. S	0.009 ± 10%	±	65. Tb	±	0.3 ±
17. Cl	na ±	±	66. Dy	±	1 ±
19. K	0.069 ± 5%	±	67. Ho	±	0.3 ±
20. Ca	6.98 ± 5%	±	68. Er	±	1 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.98 ± 5%	±	70. Yb	±	1 ±
23. V	±	170 ±	71. Lu	±	±
24. Cr	±	400 ± 50%	72. Hf	±	±
25. Mn	0.04 ± 20%	±	73. Ta	na ±	±
26. Fe	5.78 ± 5%	±	74. W	±	±
27. Co	±	14 ±	75. Re	na ±	±
28. Ni	±	65 ±	76. Os	±	±
29. Cu	±	25 ±	77. Ir	±	±
30. Zn	±	75 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.1 ±	81. Tl	±	±
34. Se	±	0.3 ±	82. Pb	±	0.3 ±
35. Br	±	0.6 ±	83. Bi	±	±
37. Rb	±	0.1 ±	84. Po	±	±
38. Sr	±	60 ±	85. At	±	±
39. Y	±	16 ±	86. Rn	±	±
40. Zr	±	16 ±	87. Fr	±	±
41. Nb	±	0.1 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	±	2 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	±	93. Np	±	±
47. Ag	±	±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO	0.14 ± 5%	±		±	±
Fe ⁺² /Fe ⁺²⁺³	1.82 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

01.2638

ANALYTICAL REPORT

Porphyritic amygdaloidal altered

Material and slightly weathered basalt

IRL Sample No. 1 M 1419

17-CH-11 496.5-499.0'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	< 0.05 ±
2. He	na ±	±	51. Sb	±	< 0.05 ±
3. Li	±	9 ±10%	52. Te	±	< 0.05 ±
4. Be	na ±	±	53. I	±	0.1 ±
5. B	±	7 ±50%	55. Cs	±	0.1 ±
6. C	na ±	±	56. Ba	±	15 ±
7. N	na ±	±	57. La	±	4 ±
8. O	na ±	±	58. Ce	±	8 ±
9. F	na ±	±	59. Pr	±	2 ±
11. Na	2.1 ± 5%	±	60. Nd	±	6 ±
12. Mg	2.43 ± 5%	±	61. Pm	±	±
13. Al	9.25 ± 5%	±	62. Sm	±	2 ±
14. Si	23.14 ± 5%	±	63. Eu	±	0.3 ±
15. P	0.010 ±10%	±	64. Gd	±	1 ±
16. S	0.012 ±10%	±	65. Tb	±	0.4 ±
17. Cl	na ±	±	66. Dy	±	1.5 ±
19. K	0.093 ± 5%	±	67. Ho	±	0.3 ±
20. Ca	6.29 ± 5%	±	68. Er	±	0.7 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.69 ± 5%	±	70. Yb	±	1 ±
23. V	±	620 ±	71. Lu	±	±
24. Cr	±	< 10 ±	72. Hf	±	< 0.05 ±
25. Mn	0.03 ±20%	1000 ±	73. Ta	na ±	±
26. Fe	6.24 ± 5%	±	74. W	±	±
27. Co	±	35 ±	75. Re	na ±	±
28. Ni	±	14 ±	76. Os	±	±
29. Cu	±	115 ±	77. Ir	±	±
30. Zn	±	140 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.4 ±	81. Tl	±	±
34. Se	±	0.8 ±	82. Pb	±	0.6 ±
35. Br	±	2 ±	83. Bi	±	±
37. Rb	±	0.3 ±	84. Po	±	±
38. Sr	±	80 ±	85. At	±	±
39. Y	±	21 ±	86. Rn	±	±
40. Zr	±	30 ±	87. Fr	±	±
41. Nb	±	0.2 ±	88. Ra	±	±
42. Mo	±	0.8 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	< 0.05 ±
44. Ru	±	9 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	< 0.05 ±	93. Np	±	±
47. Ag	±	< 0.05 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂ /Fe ⁺² /Fe ⁺³	0.36 ± 5%	±		±	±
	0.93 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

**Expressed as a percent of the concentration

na - not analyzed

Analytical Lab.

Lab. Sample No.

Date:

Signed:

ANALYTICAL REPORT

Altered and weathered
Material porphyritic basalt

IRL Sample No. 1 M 1113

17-CH-11

901.8-903.2'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.3 ±
2. He	na ±	±	51. Sb	±	±
3. Li	±	15 ± 10%	52. Te	±	±
4. Be	na ±	±	53. I	±	±
5. B	±	5 ±	55. Cs	±	±
6. C	na ±	±	56. Ba	±	35 ±
7. N	na ±	±	57. La	±	9 ±
8. O	na ±	±	58. Ce	±	20 ±
9. F	na ±	±	59. Pr	±	5 ±
11. Na	2.0 ± 5%	±	60. Nd	±	20 ±
12. Mg	3.08 ± 2%	±	61. Pm	±	±
13. Al	9.75 ± 2%	±	62. Sm	±	10 ±
14. Si	21.58 ± 2%	±	63. Eu	±	2 ±
15. P	0.051 ± 5%	±	64. Gd	±	6 ±
16. S	0.022 ± 5%	±	65. Tb	±	1 ±
17. Cl	na ±	±	66. Dy	±	5 ±
19. K	0.11 ± 5%	±	67. Ho	±	3 ±
20. Ca	5.92 ± 5%	±	68. Er	±	4 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.74 ± 5%	±	70. Yb	±	3 ±
23. V	±	350 ±	71. Lu	±	1 ±
24. Cr	±	< 10 ±	72. Hf	±	±
25. Mn	0.10 ± 10%	±	73. Ta	na ±	±
26. Fe	6.98 ± 5%	±	74. W	±	±
27. Co	±	20 ±	75. Re	na ±	±
28. Ni	±	30 ±	76. Os	±	±
29. Cu	±	65 ±	77. Ir	±	±
30. Zn	±	75 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.6 ±	81. Tl	±	±
34. Se	±	0.8 ±	82. Pb	±	1 ±
35. Br	±	1 ±	83. Bi	±	±
37. Rb	±	0.6 ±	84. Po	±	±
38. Sr	±	75 ±	85. At	±	±
39. Y	±	35 ±	86. Rn	±	±
40. Zr	±	60 ±	87. Fr	±	±
41. Nb	±	0.5 ±	88. Ra	±	±
42. Mo	±	3 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	±	15 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	±	93. Np	±	±
47. Ag	±	±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	0.3 ±	96. Cm	±	±
*CO ₂ +3	0.39 ± 5%	±		±	±
Fe ⁺² /Fe	0.87 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Fresh altered tuffIRL Sample No. 1 M 147517-CH-11 925.8-928.7'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.05 ±
2. He	na ±	±	51. Sb	±	±
3. Li	±	9 ± 10%	52. Te	±	±
4. Be	±	0.5 ±	53. I	±	±
5. B	±	20 ± 50%	55. Cs	±	0.1 ±
6. C	na ±	±	56. Ba	±	200 ± 50%
7. N	na ±	±	57. La	±	5 ±
8. O	na ±	±	58. Ce	±	20 ±
9. F	na ±	±	59. Pr	±	4 ±
11. Na	1.8 ± 5%	±	60. Nd	±	15 ±
12. Mg	1.68 ± 2%	±	61. Pm	±	±
13. Al	7.30 ± 2%	±	62. Sm	±	5 ±
14. Si	26.65 ± 2%	±	63. Eu	±	1 ±
15. P	0.038 ± 5%	±	64. Gd	±	6 ±
16. S	0.019 ± 5%	±	65. Tb	±	0.5 ±
17. Cl	na ±	±	66. Dy	±	4 ±
19. K	0.47 ± 5%	±	67. Ho	±	1 ±
20. Ca	5.13 ± 2%	±	68. Er	±	4 ±
21. Sc	±	25 ± 50%	69. Tm	na ±	±
22. Ti	0.91 ± 5%	±	70. Yb	±	3 ±
23. V	±	250 ± 50%	71. Lu	±	0.4 ±
24. Cr	±	20 ± 50%	72. Hf	±	2 ±
25. Mn	0.06 ± 20%	±	73. Ta	na ±	±
26. Fe	6.20 ± 5%	±	74. W	±	±
27. Co	±	±	75. Re	na ±	±
28. Ni	±	15 ± 50%	76. Os	±	±
29. Cu	±	120 ± 50%	77. Ir	±	±
30. Zn	±	40 ± 50%	78. Pt	±	±
31. Ga	±	20 ± 50%	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	2 ±	81. Tl	±	±
34. Se	±	0.3 ±	82. Pb	±	2 ±
35. Br	±	6 ±	83. Bi	±	±
37. Rb	±	2 ±	84. Po	±	±
38. Sr	±	75 ± 50%	85. At	±	±
39. Y	±	16 ± 50%	86. Rn	±	±
40. Zr	±	150 ± 50%	87. Fr	±	±
41. Nb	±	0.5 ±	88. Ra	±	±
42. Mo	±	1 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	1 ±
44. Ru	±	7 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	±	93. Np	±	±
47. Ag	±	0.1 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	±	96. Cm	±	±
*CO ₂	0.24 ± 5%	±		±	±
Fe ⁺² /Fe ⁺³	0.74 ± 5%	±		±	±

Method of Analysis:

* CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Fresh tuffaceous sandstoneIRL Sample No. 1 M 126217-CH-12 and 118.4-119.1'
120.2-120.8'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	±
2. He	na ±	±	51. Sb	±	±
3. Li	±	12 ± 10%	52. Te	±	±
4. Be	na ±	±	53. I	±	±
5. B	±	5 ±	55. Cs	±	0.1 ±
6. C	na ±	±	56. Ba	±	400 ±
7. N	na ±	±	57. La	±	6 ±
8. O	na ±	±	58. Ce	±	12 ±
9. F	na ±	±	59. Pr	±	1.5 ±
11. Na	1.6 ± 5%	±	60. Nd	±	6 ±
12. Mg	2.44 ± 2%	±	61. Pm	±	±
13. Al	10.44 ± 2%	±	62. Sm	±	4 ±
14. Si	23.07 ± 2%	±	63. Eu	±	2 ±
15. P	0.009 ± 10%	±	64. Gd	±	4 ±
16. S	0.041 ± 5%	±	65. Tb	±	0.9 ±
17. Cl	na ±	±	66. Dy	±	4 ±
19. K	0.40 ± 5%	±	67. Ho	±	3 ±
20. Ca	6.29 ± 2%	±	68. Er	±	2 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.62 ± 5%	±	70. Yb	±	0.6 ±
23. V	±	200 ±	71. Lu	±	±
24. Cr	±	45 ± 50%	72. Hf	±	±
25. Mn	0.04 ± 20%	±	73. Ta	na ±	±
26. Fe	5.42 ± 5%	±	74. W	±	1 ±
27. Co	±	18 ±	75. Re	na ±	±
28. Ni	±	12 ±	76. Os	±	±
29. Cu	±	85 ±	77. Ir	±	±
30. Zn	±	130 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.3 ±	81. Tl	±	±
34. Se	±	0.2 ±	82. Pb	±	2 ±
35. Br	±	1.5 ±	83. Bi	±	±
37. Rb	±	1.5 ±	84. Po	±	±
38. Sr	±	450 ±	85. At	±	±
39. Y	±	18 ±	86. Rn	±	±
40. Zr	±	30 ±	87. Fr	±	±
41. Nb	±	0.3 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	±	2 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	±	93. Np	±	±
47. Ag	±	±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	±	96. Cm	±	±
* CO ₂	0.14 ± 5%	±		±	±
Fe ⁺² /Fe ⁺³	0.25 ± 5%	±		±	±

Method of Analysis:

* CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Fresh tuffaceous
Material siltstoneIRL Sample No. 1 M 1281

17-CH-12 179.8-182.3'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.1 ±
2. He	na ±	±	51. Sb	±	±
3. Li	±	14 ± 10%	52. Te	±	±
4. Be	na ±	±	53. I	±	0.2 ±
5. B	±	10 ± 50%	55. Cs	±	0.05 ±
6. C	na ±	±	56. Ba	±	550 ±
7. N	na ±	±	57. La	±	5 ±
8. O	na ±	±	58. Ce	±	12 ±
9. F	na ±	±	59. Pr	±	2 ±
11. Na	1.7 ± 5%	±	60. Nd	±	9 ±
12. Mg	2.77 ± 5%	±	61. Pm	±	±
13. Al	9.70 ± 2%	±	62. Sm	±	4 ±
14. Si	24.24 ± 2%	±	63. Eu	±	0.4 ±
15. P	0.031 ± 5%	±	64. Gd	±	3 ±
16. S	0.109 ± 5%	±	65. Tb	±	0.4 ±
17. Cl	na ±	±	66. Dy	±	3 ±
19. K	0.44 ± 5%	±	67. Ho	±	0.5 ±
20. Ca	5.66 ± 2%	±	68. Er	±	1 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.74 ± 5%	±	70. Yb	±	2.5 ±
23. V	±	850 ±	71. Lu	±	±
24. Cr	±	20 ± 50%	72. Hf	±	2 ±
25. Mn	0.08 ± 10%	±	73. Ta	na ±	±
26. Fe	6.71 ± 5%	±	74. W	±	0.5 ±
27. Co	±	10 ±	75. Re	na ±	±
28. Ni	±	11 ±	76. Os	±	±
29. Cu	±	235 ±	77. Ir	±	±
30. Zn	±	75 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.4 ±	81. Tl	±	±
34. Se	±	0.8 ±	82. Pb	±	2 ±
35. Br	±	1 ±	83. Bi	±	±
37. Rb	±	5 ±	84. Po	±	±
38. Sr	±	210 ±	85. At	±	±
39. Y	±	35 ±	86. Rn	±	±
40. Zr	±	50 ±	87. Fr	±	±
41. Nb	±	0.3 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	±	5 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	±	93. Np	±	±
47. Ag	±	±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	±	96. Cm	±	±
*CO ₂	0.14 ± 5%	±		±	±
Fe ⁺² /Fe ⁺³	0.20 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Agglomerate
17-CH-12 288.2 - 289.0'

LRL Sample No. 1M 1073

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	±
2. He	na ±	±	51. Sb	±	±
3. Li	±	8 ± 10%	52. Te	±	±
4. Be	na ±	±	53. I	±	0.3 ±
5. B	±	< 5	55. Cs	±	±
6. C	na ±	±	56. Ba	±	55 ±
7. N	na ±	±	57. La	±	1.5 ±
8. O	na ±	±	58. Ce	±	8 ±
9. F	na ±	±	59. Pr	±	0.8 ±
11. Na	1.4 ± 5%	±	60. Nd	±	2 ±
12. Mg	3.02 ± 2%	±	61. Pm	±	±
13. Al	11.52 ± 2%	±	62. Sm	±	2 ±
14. Si	20.83 ± 2%	±	63. Eu	±	0.7 ±
15. P	0.005 ± 20%	±	64. Gd	±	1 ±
16. S	0.008 ± 10%	±	65. Tb	±	0.4 ±
17. Cl	na ±	±	66. Dy	±	1 ±
19. K	0.17 ± 5%	±	67. Ho	±	0.4 ±
20. Ca	8.86 ± 2%	±	68. Er	±	2 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.59 ± 5%	±	70. Yb	±	±
23. V	±	350 ±	71. Lu	±	±
24. Cr	±	35 ± 50%	72. Hf	±	±
25. Mn	0.06 ± 20%	±	73. Ta	na ±	±
26. Fe	6.13 ± 5%	±	74. W	±	±
27. Co	±	12 ±	75. Re	na ±	±
28. Ni	±	20 ±	76. Os	±	1 ±
29. Cu	±	40 ±	77. Ir	±	±
30. Zn	±	60 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.1 ±	81. Tl	±	±
34. Se	±	0.5 ±	82. Pb	±	0.6 ±
35. Br	±	2 ±	83. Bi	±	±
37. Rb	±	0.7 ±	84. Po	±	±
38. Sr	±	160 ±	85. At	±	±
39. Y	±	12 ±	86. Rn	±	±
40. Zr	±	25 ±	87. Fr	±	±
41. Nb	±	0.2 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	na ±	±	91. Pa	±	±
45. Rh	±	±	92. U	±	±
46. Pd	±	±	93. Np	±	±
47. Ag	±	±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	0.2 ±	96. Cm	±	±
*CO ₂	0.15 ± 5%	±		±	±
Fe ⁺² /Fe ⁺³	0.39 ± 5%	±		±	±

Method of Analysis:

* CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Agglomerate with
tuffaceous matrixLRL Sample No. 1 M 132817-CH-12 385.6-387.1'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	< 0.05 ±
2. He	na ±	±	51. Sb	±	0.2 ±
3. Li	±	14 ±10%	52. Te	±	0.1 ±
4. Be	na ±	±	53. I	±	±
5. B	±	5 ±	55. Cs	±	0.3 ±
6. C	na ±	±	56. Ba	±	32 ±
7. N	na ±	±	57. La	±	2 ±
8. O	na ±	±	58. Ce	±	5 ±
9. F	na ±	±	59. Pr	±	0.8 ±
11. Na	1.0 ± 5%	±	60. Nd	±	4.5 ±
12. Mg	3.04 ± 2%	±	61. Pm	±	±
13. Al	10.11 ± 2%	±	62. Sm	±	3 ±
14. Si	21.90 ± 2%	±	63. Eu	±	0.5 ±
15. P	0.003 ± 50%	±	64. Gd	±	2 ±
16. S	0.006 ± 10%	±	65. Tb	±	0 ±
17. Cl	na ±	±	66. Dy	±	3 ±
19. K	0.082 ± 5%	±	67. Ho	±	0.5 ±
20. Ca	8.10 ± 2%	±	68. Er	±	1.5 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.59 ± 5%	±	70. Yb	±	2 ±
23. V	±	155 ±	71. Lu	±	0.2 ±
24. Cr	±	90 ±50%	72. Hf	±	1 ±
25. Mn	0.08 ± 10%	±	73. Ta	na ±	±
26. Fe	7.18 ± 5%	±	74. W	±	0.7 ±
27. Co	±	8 ±	75. Re	na ±	±
28. Ni	±	16 ±	76. Os	±	±
29. Cu	±	25 ±	77. Ir	±	±
30. Zn	±	75 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.2 ±	81. Tl	±	±
34. Se	±	1.5 ±	82. Pb	±	1 ±
35. Br	±	1 ±	83. Bi	±	±
37. Rb	±	0.1 ±	84. Po	±	±
38. Sr	±	90 ±	85. At	±	±
39. Y	±	10 ±	86. Rn	±	±
40. Zr	±	12 ±	87. Fr	±	±
41. Nb	±	0.5 ±	88. Ra	±	±
42. Mo	±	1 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	±	7 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	< 0.05 ±	93. Np	±	±
47. Ag	±	< 0.05 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	0.1 ±	96. Cm	±	±
*CO ₂ /Fe ⁺² /Fe ⁺³	0.16 ± 5%	±		±	±
	0.37 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

**Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Fresh tuffaceous sandstoneIRL Sample No. 12317-CH-12970'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	< 0.05 ±
2. He	na ±	±	51. Sb	±	±
3. Li	±	17 ± 10%	52. Te	±	±
4. Be	na ±	±	53. I	±	±
5. B	±	25 ± 50%	55. Cs	±	0.5 ±
6. C	na ±	±	56. Ba	±	190 ±
7. N	na ±	±	57. La	±	11 ±
8. O	na ±	±	58. Ce	±	12 ±
9. F	na ±	±	59. Pr	±	3 ±
11. Na	3.1 ± 5%	±	60. Nd	±	10 ±
12. Mg	2.15 ± 2%	±	61. Pm	±	±
13. Al	10.20 ± 2%	±	62. Sm	±	4 ±
14. Si	24.61 ± 2%	±	63. Eu	±	0.5 ±
15. P	0.032 ± 5%	±	64. Gd	±	2 ±
16. S	0.079 ± 5%	±	65. Tb	±	0.3 ±
17. Cl	na ±	±	66. Dy	±	2 ±
19. K	0.69 ± 5%	±	67. Ho	±	0.3 ±
20. Ca	4.99 ± 2%	±	68. Er	±	1 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.56 ± 5%	±	70. Yb	±	0.8 ±
23. V	±	170 ±	71. Lu	±	< 0.05 ±
24. Cr	±	60 ± 50%	72. Hf	±	2 ±
25. Mn	0.05 ± 20%	±	73. Ta	na ±	±
26. Fe	5.37 ± 5%	±	74. W	±	±
27. Co	±	9 ±	75. Re	na ±	±
28. Ni	±	8 ±	76. Os	±	±
29. Cu	±	50 ±	77. Ir	±	±
30. Zn	±	22 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.2 ±	81. Tl	±	±
34. Se	±	0.3 ±	82. Pb	±	0.9 ±
35. Br	±	0.6 ±	83. Bi	±	±
37. Rb	±	21 ±	84. Po	±	±
38. Sr	±	75 ±	85. At	±	±
39. Y	±	17 ±	86. Rn	±	±
40. Zr	±	32 ±	87. Fr	±	±
41. Nb	±	0.2 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	0.5 ±
44. Ru	±	2 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	0.2 ±
46. Pd	±	< 0.05 ±	93. Np	±	±
47. Ag	±	< 0.05 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂	0.20 ± 5%	±		±	±
Fe ⁺² /Fe ⁺²⁺³	0.31 ± 5%	±		±	±

Method of Analysis:

* CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Fresh Tuffaceous
Material Limestone

LRL Sample No. 1M 1248

17-CH-13 208.2 - 209.8'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	<0.05 ±
2. He	na ±	±	51. Sb	±	<0.05 ±
3. Li	±	17 ± 10%	52. Te	±	±
4. Be	na ±	±	53. I	±	0.4 ±
5. B	±	15 ± 50%	55. Cs	±	0.5 ±
6. C	na ±	±	56. Ba	±	2300 ±
7. N	na ±	±	57. La	±	21 ±
8. O	na ±	±	58. Ce	±	29 ±
9. F	na ±	±	59. Pr	±	4 ±
11. Na	1.4 ± 5%	±	60. Nd	±	9 ±
12. Mg	1.03 ± 2%	±	61. Pm	±	±
13. Al	4.33 ± 2%	±	62. Sm	±	4 ±
14. Si	18.85 ± 2%	±	63. Eu	±	1 ±
15. P	0.010 ± 10%	±	64. Gd	±	4 ±
16. S	0.19 ± 5%	±	65. Tb	±	1 ±
17. Cl	na ±	±	66. Dy	±	4 ±
19. K	0.93 ± 5%	±	67. Ho	±	0.6 ±
20. Ca	15.75 ± 2%	±	68. Er	±	2 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.51 ± 5%	±	70. Yb	±	2 ±
23. V	±	192 ±	71. Lu	±	0.3 ±
24. Cr	±	20 ± 50%	72. Hf	±	1.5 ±
25. Mn	0.02 ± 20%	±	73. Ta	na ±	±
26. Fe	2.52 ± 5%	±	74. W	±	0.5 ±
27. Co	±	58 ±	75. Re	na ±	±
28. Ni	±	21 ±	76. Os	±	±
29. Cu	±	96 ±	77. Ir	±	±
30. Zn	±	105 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.65 ±	81. Tl	±	±
34. Se	±	< 0.05 ±	82. Pb	±	1.5 ±
35. Br	±	2 ±	83. Bi	±	±
37. Rb	±	6 ±	84. Po	±	±
38. Sr	±	1900 ±	85. At	±	±
39. Y	±	42 ±	86. Rn	±	±
40. Zr	±	62 ±	87. Fr	±	±
41. Nb	±	3 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	0.5 ±
44. Ru	±	4 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	0.8 ±
46. Pd	±	0.2 ±	93. Np	±	±
47. Ag	±	±	94. Pu	±	±
48. Cd	±	< 0.05 ±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂	14.6 ± 5%	±		±	±
Fe+2/Fe+3	0.36 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Moderately weathered
Material calcareous shale
17-CH-13 321'

LRL Sample No. 222/64

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na	±	50. Sn	±	0.05
2. He	na	±	51. Sb	±	0.05
3. Li	±	96	52. Te	±	±
4. Be	na	±	53. I	±	10
5. B	±	50	55. Cs	±	2
6. C	na	±	56. Ba	±	520
7. N	na	±	57. La	±	8
8. O	na	±	58. Ce	±	13
9. F	na	±	59. Pr	±	2
11. Na	1.3	± 5%	60. Nd	±	8
12. Mg	1.81	± 2%	61. Pm	±	±
13. Al	8.74	± 2%	62. Sm	±	3
14. Si	22.60	± 2%	63. Eu	±	0.5
15. P	0.061	± 5%	64. Gd	±	1
16. S	1.46	± 5%	65. Tb	±	0.3
17. Cl	na	±	66. Dy	±	2
19. K	1.4	± 5%	67. Ho	±	0.2
20. Ca	4.52	± 2%	68. Er	±	1
21. Sc	na	±	69. Tm	na	±
22. Ti	0.75	± 5%	70. Yb	±	1
23. V	±	420	71. Lu	±	±
24. Cr	±	100	72. Hf	±	±
25. Mn	0.05	± 20%	73. Ta	na	±
26. Fe	5.56	± 5%	74. W	±	1
27. Co	±	12	75. Re	na	±
28. Ni	±	75	76. Os	±	0.5
29. Cu	±	60	77. Ir	±	±
30. Zn	±	60	78. Pt	±	±
31. Ga	na	±	79. Au	±	±
32. Ge	na	±	80. Hg	±	±
33. As	±	2	81. Tl	±	±
34. Se	±	1	82. Pb	±	2
35. Br	±	20	83. Bi	±	±
37. Rb	±	8	84. Po	±	±
38. Sr	±	260	85. At	±	±
39. Y	±	20	86. Rn	±	±
40. Zr	±	80	87. Fr	±	±
41. Nb	±	5	88. Ra	±	±
42. Mo	±	3	89. Ac	±	±
43. Tc	±	±	90. Th	±	2
44. Ru	±	2	91. Pa	±	±
45. Rh	na	±	92. U	±	1
46. Pd	±	±	93. Np	±	±
47. Ag	±	0.2	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	±	96. Cm	±	±
*CO ₂	5.90	± 5%		±	±
Fe ⁺² /Fe ⁺³	1.58	± 5%		±	±

Method of Analysis:

* CO₂ From carbonate

** Expressed as a percent of the concentration

na - Not analyzed

Analytical Lab.

Lab. Sample No.

Date:

Signed:

ANALYTICAL REPORT

Material Fresh shale
17-CH-13 781'LRL Sample No. 112

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	± 0.05	±
2. He	na ±	±	51. Sb	± 0.5	±
3. Li	±	94 ± 5%	52. Te	±	±
4. Be	±	0.5 ± 50%	53. I	± 30	±
5. B	±	80 ± 50%	55. Cs	± 2	±
6. C	na ±	±	56. Ba	± 400	± 50%
7. N	na ±	±	57. La	± 5	±
8. O	na ±	±	58. Ce	± 21	±
9. F	na ±	±	59. Pr	± 3	±
11. Na	1.4 ± 5%	±	60. Nd	± 4	±
12. Mg	1.96 ± 2%	±	61. Pm	±	±
13. Al	8.74 ± 2%	±	62. Sm	± 3	±
14. Si	21.84 ± 2%	±	63. Eu	± 0.4	±
15. P	0.078 ± 5%	±	64. Gd	± 2	±
16. S	1.83 ± 5%	±	65. Tb	± 0.3	±
17. Cl	na ±	±	66. Dy	± 3	±
19. K	1.2 ± 5%	±	67. Ho	± 0.3	±
20. Ca	5.26 ± 2%	±	68. Er	± 1	±
21. Sc	±	20 ± 50%	69. Tm	na ±	±
22. Ti	0.73 ± 5%	±	70. Yb	± 1.5	±
23. V	±	100 ± 50%	71. Lu	± 0.2	±
24. Cr	±	100 ± 50%	72. Hf	± 1	±
25. Mn	0.08 ± 10%	±	73. Ta	na ±	±
26. Fe	5.48 ± 5%	±	74. W	± 1	±
27. Co	±	50 ±	75. Re	na ±	±
28. Ni	±	50 ± 50%	76. Os	±	±
29. Cu	±	100 ± 50%	77. Ir	±	±
30. Zn	±	40 ± 50%	78. Pt	±	±
31. Ga	±	10 ± 50%	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	11 ±	81. Tl	± 0.3	±
34. Se	±	1 ±	82. Pb	± 3	±
35. Br	±	65 ±	83. Bi	±	±
37. Rb	±	19 ±	84. Po	±	±
38. Sr	±	250 ± 50%	85. At	±	±
39. Y	±	8 ± 50%	86. Rn	±	±
40. Zr	±	100 ± 50%	87. Fr	±	±
41. Nb	±	4 ±	88. Ra	±	±
42. Mo	±	4 ±	89. Ac	±	±
43. Tc	±	±	90. Th	± 1	±
44. Ru	±	< 0.05 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	± 2	±
46. Pd	±	< 0.05 ±	93. Np	±	±
47. Ag	±	0.1 ±	94. Pu	±	±
48. Cd	±	< 0.05 ±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂	4.99 ± 5%	±	Free H ₂ O	16.96 ± 5%	±
Fe ⁺² /Fe ⁺³	1.27 ± 5%	±	Bound H ₂ O	4.84 ± 5%	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Slightly weathered
Material lapelli tuffLRL Sample No. 1 M 114417-CH-14 496.2-497.4'
and 500.5-501.2'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.1 ±
2. He	na ±	±	51. Sb	±	< 0.05 ±
3. Li	±	17 ± 10%	52. Te	±	±
4. Be	na ±	±	53. I	±	0.2 ±
5. B	±	5 ±	55. Cs	±	0.5 ±
6. C	na ±	±	56. Ba	±	560 ±
7. N	na ±	±	57. La	±	20 ±
8. O	na ±	±	58. Ce	±	45 ±
9. F	na ±	±	59. Pr	±	5 ±
11. Na	1.8 ± 5%	±	60. Nd	±	20 ±
12. Mg	3.66 ± 2%	±	61. Pm	±	±
13. Al	8.23 ± 2%	±	62. Sm	±	5 ±
14. Si	24.33 ± 2%	±	63. Eu	±	0.7 ±
15. P	0.082 ± 5%	±	64. Gd	±	3 ±
16. S	0.019 ± 5%	±	65. Tb	±	0.5 ±
17. Cl	na ±	±	66. Dy	±	4 ±
19. K	0.95 ± 5%	±	67. Ho	±	0.5 ±
20. Ca	6.00 ± 2%	±	68. Er	±	1.5 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.76 ± 5%	±	70. Yb	±	1 ±
23. V	±	500 ±	71. Lu	±	0.2 ±
24. Cr	±	50 ± 50%	72. Hf	±	2.5 ±
25. Mn	0.10 ± 10%	±	73. Ta	na ±	±
26. Fe	6.02 ± 5%	±	74. W	±	3 ±
27. Co	±	20 ±	75. Re	na ±	±
28. Ni	±	20 ±	76. Os	±	±
29. Cu	±	100 ±	77. Ir	±	±
30. Zn	±	115 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.2 ±	81. Tl	±	0.05 ±
34. Se	±	0.4 ±	82. Pb	±	2 ±
35. Br	±	2 ±	83. Bi	±	±
37. Rb	±	6 ±	84. Po	±	±
38. Sr	±	770 ±	85. At	±	±
39. Y	±	35 ±	86. Rn	±	±
40. Zr	±	100 ±	87. Fr	±	±
41. Nb	±	0.2 ±	88. Ra	±	±
42. Mo	±	0.8 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	2 ±
44. Ru	±	±	91. Pa	±	0.7 ±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	0.05 ±	93. Np	±	±
47. Ag	±	< 0.05 ±	94. Pu	±	±
48. Cd	±	< 0.05 ±	95. Am	±	±
49. In	±	0.5 ±	96. Cm	±	±
*CO ₂ +3	0.15 ± 5%	±		±	±
Fe ⁺² /Fe	0.40 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Fresh tuffLRL Sample No. 1 M 111717-CH 14 576.5-578.2'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.05 ±
2. He	na ±	±	51. Sb	±	< 0.05 ±
3. Li	±	21 ± 5%	52. Te	±	±
4. Be	na ±	±	53. I	±	0.05 ±
5. B	±	7 ±	55. Cs	±	0.5 ±
6. C	na ±	±	56. Ba	±	320 ±
7. N	na ±	±	57. La	±	24 ±
8. O	na ±	±	58. Ce	±	40 ±
9. F	na ±	±	59. Pr	±	5 ±
11. Na	1.6 ± 5%	±	60. Nd	±	17 ±
12. Mg	2.58 ± 2%	±	61. Pm	±	±
13. Al	8.03 ± 2%	±	62. Sm	±	4 ±
14. Si	25.14 ± 2%	±	63. Eu	±	0.6 ±
15. P	0.080 ± 10%	±	64. Gd	±	3 ±
16. S	0.018 ± 5%	±	65. Tb	±	0.5 ±
17. Cl	na ±	±	66. Dy	±	3 ±
19. K	1.2 ± 5%	±	67. Ho	±	0.6 ±
20. Ca	6.06 ± 2%	±	68. Er	±	1.5 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.59 ± 5%	±	70. Yb	±	1 ±
23. V	±	460 ±	71. Lu	±	< 0.05 ±
24. Cr	±	20 ± 50%	72. Hf	±	2 ±
25. Mn	0.12 ± 10%	±	73. Ta	na ±	±
26. Fe	5.06 ± 5%	±	74. W	±	±
27. Co	±	32 ±	75. Re	na ±	±
28. Ni	±	16 ±	76. Os	±	±
29. Cu	±	51 ±	77. Ir	±	±
30. Zn	±	44 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.2 ±	81. Tl	±	±
34. Se	±	0.5 ±	82. Pb	±	1.5 ±
35. Br	±	1 ±	83. Bi	±	±
37. Rb	±	8 ±	84. Po	±	±
38. Sr	±	380 ±	85. At	±	±
39. Y	±	32 ±	86. Rn	±	±
40. Zr	±	130 ±	87. Fr	±	±
41. Nb	±	0.6 ±	88. Ra	±	±
42. Mo	±	0.5 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	1 ±
44. Ru	±	2 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	0.3 ±
46. Pd	±	0.1 ±	93. Np	±	±
47. Ag	±	0.05 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂ /Fe ⁺² /Fe ⁺³	2.52 ± 5%	±		±	±
	0.30 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Interbedded tuffaceous
Material agglomerate and sandstone
17-CH-14 601.4-603.9'

IRL Sample No. 1 M 1131

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.2 ±
2. He	na ±	±	51. Sb	±	0.1 ±
3. Li	±	12 ± 10%	52. Te	±	±
4. Be	na ±	±	53. I	±	±
5. B	±	5 ±	55. Cs	±	0.6 ±
6. C	na ±	±	56. Ba	±	1300 ±
7. N	na ±	±	57. La	±	55 ±
8. O	na ±	±	58. Ce	±	115 ±
9. F	na ±	±	59. Pr	±	8 ±
11. Na	1.8 ± 5%	±	60. Nd	±	40 ±
12. Mg	2.34 ± 2%	±	61. Pm	±	±
13. Al	8.48 ± 2%	±	62. Sm	±	10 ±
14. Si	24.60 ± 2%	±	63. Eu	±	1.5 ±
15. P	0.116 ± 5%	±	64. Gd	±	8 ±
16. S	0.048 ± 5%	±	65. Tb	±	0.9 ±
17. Cl	na ±	±	66. Dy	±	5 ±
19. K	2.0 ± 5%	±	67. Ho	±	0.9 ±
20. Ca	5.02 ± 2%	±	68. Er	±	4 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.96 ± 5%	±	70. Yb	±	4 ±
23. V	±	500 ±	71. Lu	±	2 ±
24. Cr	±	20 ± 50%	72. Hf	±	5 ±
25. Mn	0.06 ± 20%	±	73. Ta	na ±	±
26. Fe	5.35 ± 5%	±	74. W	±	±
27. Co	±	60 ±	75. Re	na ±	±
28. Ni	±	40 ±	76. Os	±	±
29. Cu	±	220 ±	77. Ir	±	±
30. Zn	±	240 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	1 ±	81. Tl	±	±
34. Se	±	0.5 ±	82. Pb	±	3 ±
35. Br	±	4 ±	83. Bi	±	±
37. Rb	±	25 ±	84. Po	±	±
38. Sr	±	880 ±	85. At	±	±
39. Y	±	70 ±	86. Rn	±	±
40. Zr	±	255 ±	87. Fr	±	±
41. Nb	±	2 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	2 ±
44. Ru	±	5 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	0.4 ±
46. Pd	±	±	93. Np	±	±
47. Ag	±	±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	0.2 ±	96. Cm	±	±
*CO ₂	1.69 ± 5%	±		±	±
Fe ⁺² /Fe ⁺³	0.28 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material AgglomerateLRL Sample No. 13117-CH-14 715'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ±**	±	50. Sn	±	< 0.05 ±
2. He	na ±	±	51. Sb	±	0.05 ±
3. Li	±	14 ±10%	52. Te	±	< 0.05 ±
4. Be	±	0.5 ±	53. I	±	±
5. B	±	5 ±	55. Cs	±	0.2 ±
6. C	na ±	±	56. Ba	±	400 ±50%
7. N	na ±	±	57. La	±	5 ±50%
8. O	na ±	±	58. Ce	±	16 ±
9. F	na ±	±	59. Pr	±	3 ±
11. Na	2.6 ± 5%	±	60. Nd	±	10 ±
12. Mg	3.07 ± 2%	±	61. Pm	±	±
13. Al	9.32 ± 2%	±	62. Sm	±	4 ±
14. Si	25.21 ± 2%	±	63. Eu	±	0.7 ±
15. P	0.044 ± 5%	±	64. Gd	±	4 ±
16. S	0.009 ±10%	±	65. Tb	±	0.4 ±
17. Cl	na ±	±	66. Dy	±	3 ±
19. K	1.1 ± 5%	±	67. Ho	±	0.3 ±
20. Ca	5.47 ± 2%	±	68. Er	±	1 ±
21. Sc	±	30 ±50%	69. Tm	na ±	±
22. Ti	0.59 ± 5%	±	70. Yb	±	2 ±
23. V	±	250 ±50%	71. Lu	±	< 0.05 ±
24. Cr	±	80 ±50%	72. Hf	±	1 ±
25. Mn	0.05 ±20%	±	73. Ta	na ±	±
26. Fe	5.19 ± 5%	±	74. W	±	±
27. Co	±	11 ±	75. Re	na ±	±
28. Ni	±	25 ±50%	76. Os	±	±
29. Cu	±	100 ±50%	77. Ir	±	±
30. Zn	±	40 ±50%	78. Pt	±	±
31. Ga	±	20 ±50%	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.3 ±	81. Tl	±	0.2 ±
34. Se	±	0.6 ±	82. Pb	±	2 ±
35. Br	±	1 ±	83. Bi	±	±
37. Rb	±	21 ±	84. Po	±	±
38. Sr	±	300 ±50%	85. At	±	±
39. Y	±	10 ±50%	86. Rn	±	±
40. Zr	±	100 ±50%	87. Fr	±	±
41. Nb	±	0.2 ±	88. Ra	±	±
42. Mo	±	0.7 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	0.5 ±
44. Ru	±	7 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	0.1 ±	93. Np	±	±
47. Ag	±	0.1 ±	94. Pu	±	±
48. Cd	±	< 0.05 ±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂	0.18 ± 5%	±		±	±
Fe ⁺² /Fe ⁺²⁺³	0.48 ± 5%	±		±	±

Method of Analysis:

* CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Fresh Calcareous tuffLRL Sample No. 1 M 146117-CH-15 272.5-274.0'
275.2-275.9'
278.9-279.2'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	±
2. He	na ±	±	51. Sb	±	±
3. Li	±	21 ± 5%	52. Te	±	±
4. Be	na ±	±	53. I	±	2 ±
5. B	±	50 ± 50%	55. Cs	±	±
6. C	na ±	±	56. Ba	±	650 ±
7. N	na ±	±	57. La	±	6 ±
8. O	na ±	±	58. Ce	±	8 ±
9. F	na ±	±	59. Pr	±	1.5 ±
11. Na	2.2 ± 5%	±	60. Nd	±	7 ±
12. Mg	1.44 ± 2%	±	61. Pm	±	±
13. Al	6.88 ± 2%	±	62. Sm	±	2 ±
14. Si	20.03 ± 2%	±	63. Eu	±	0.4 ±
15. P	0.005 ± 20%	±	64. Gd	±	1 ±
16. S	0.51 ± 5%	±	65. Tb	±	0.3 ±
17. Cl	na ±	±	66. Dy	±	1 ±
19. K	1.5 ± 5%	±	67. Ho	±	1 ±
20. Ca	10.09 ± 2%	±	68. Er	±	2 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.48 ± 5%	±	70. Yb	±	±
23. V	±	300 ±	71. Lu	±	±
24. Cr	±	90 ± 50%	72. Hf	±	2 ±
25. Mn	0.02 ± 20%	±	73. Ta	na ±	±
26. Fe	4.83 ± 5%	±	74. W	±	±
27. Co	±	20 ±	75. Re	na ±	±
28. Ni	±	65 ±	76. Os	±	±
29. Cu	±	40 ±	77. Ir	±	±
30. Zn	±	85 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	8 ±	81. Tl	±	±
34. Se	±	9 ±	82. Pb	±	1.5 ±
35. Br	±	8 ±	83. Bi	±	±
37. Rb	±	7 ±	84. Po	±	±
38. Sr	±	820 ±	85. At	±	±
39. Y	±	18 ±	86. Rn	±	±
40. Zr	±	18 ±	87. Fr	±	±
41. Nb	±	0.3 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	±	5 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	0.8 ±
46. Pd	±	±	93. Np	±	±
47. Ag	±	0.4 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	±	96. Cm	±	±
*CO ₂ /Fe ^{+2/+3}	9.56 ± 5%	±		±	±
Fe ^{+2/+3}	0.51 ± 5%	±		±	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Fresh calcareous
Material tuffaceous sandstone

LRL Sample No. 1 M 1379

17-CH-15 416.8-417.8'
418.9-420.2'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	±
2. He	na ±	±	51. Sb	±	< 0.05 ±
3. Li	±	28 ± 5%	52. Te	±	±
4. Be	na ±	±	53. I	±	1 ±
5. B	±	50 ± 50%	55. Cs	±	0.3 ±
6. C	na ±	±	56. Ba	±	450 ±
7. N	na ±	±	57. La	±	5 ±
8. O	na ±	±	58. Ce	±	7 ±
9. F	na ±	±	59. Pr	±	3 ±
11. Na	3.4 ± 5%	±	60. Nd	±	4 ±
12. Mg	1.89 ± 2%	±	61. Pm	±	±
13. Al	7.82 ± 2%	±	62. Sm	±	1 ±
14. Si	23.39 ± 2%	±	63. Eu	±	0.3 ±
15. P	0.007 ± 20%	±	64. Gd	±	1 ±
16. S	0.46 ± 5%	±	65. Tb	±	0.3 ±
17. Cl	na ±	±	66. Dy	±	1.5 ±
19. K	0.83 ± 5%	±	67. Ho	±	0.3 ±
20. Ca	6.15 ± 2%	±	68. Er	±	0.7 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.63 ± 5%	±	70. Yb	±	0.5 ±
23. V	±	350 ±	71. Lu	±	±
24. Cr	±	50 ± 50%	72. Hf	±	0.8 ±
25. Mn	0.07 ± 20%	±	73. Ta	na ±	±
26. Fe	4.92 ± 5%	±	74. W	±	±
27. Co	±	16 ±	75. Re	na ±	±
28. Ni	±	34 ±	76. Os	±	±
29. Cu	±	51 ±	77. Ir	±	±
30. Zn	±	70 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	2.5 ±	81. Tl	±	±
34. Se	±	0.3 ±	82. Pb	±	1 ±
35. Br	±	2 ±	83. Bi	±	±
37. Rb	±	4 ±	84. Po	±	±
38. Sr	±	640 ±	85. At	±	±
39. Y	±	14 ±	86. Rn	±	±
40. Zr	±	22 ±	87. Fr	±	±
41. Nb	±	< 0.05 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	< 0.05 ±
44. Ru	±	±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	0.4 ±
46. Pd	±	±	93. Np	±	±
47. Ag	±	< 0.05 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂	4.88 ± 5%	±		±	±
Fe ⁺² /Fe ⁺²⁺³	0.41 ± 5%	±		±	±

Method of Analysis:

* CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab.

Lab. Sample No.

Date:

Signed:

ANALYTICAL REPORT

Material Fresh Fossiliferous shale
17-CH-17 113'

LRL Sample No. 222/159

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	0.05 ±
2. He	na ±	±	51. Sb	±	0.05 ±
3. Li	±	88 ± 5%	52. Te	±	±
4. Be	na ±	±	53. I	±	1 ±
5. B	±	60 ± 50%	55. Cs	±	1 ±
6. C	na ±	±	56. Ba	±	740 ±
7. N	na ±	±	57. La	±	4 ±
8. O	na ±	±	58. Ce	±	9 ±
9. F	na ±	±	59. Pr	±	1 ±
11. Na	1.5 ± 5%	±	60. Nd	±	2.5 ±
12. Mg	2.13 ± 2%	±	61. Pm	±	±
13. Al	8.74 ± 2%	±	62. Sm	±	2 ±
14. Si	25.58 ± 2%	±	63. Eu	±	0.2 ±
15. P	0.008 ± 10%	±	64. Gd	±	±
16. S	1.47 ± 5%	±	65. Tb	±	±
17. Cl	na ±	±	66. Dy	±	±
19. K	1.1 ± 5%	±	67. Ho	±	±
20. Ca	2.56 ± 2%	±	68. Er	±	±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.74 ± 5%	±	70. Yb	±	0.7 ±
23. V	±	270 ±	71. Lu	±	±
24. Cr	±	120 ± 50%	72. Hf	±	±
25. Mn	0.04 ± 20%	±	73. Ta	na ±	±
26. Fe	5.72 ± 5%	±	74. W	±	±
27. Co	±	6.4 ±	75. Re	na ±	±
28. Ni	±	30 ±	76. Os	±	0.5 ±
29. Cu	±	35 ±	77. Ir	±	±
30. Zn	±	35 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.3 ±	81. Tl	±	±
34. Se	±	0.05 ±	82. Pb	±	1 ±
35. Br	±	3 ±	83. Bi	±	±
37. Rb	±	10 ±	84. Po	±	±
38. Sr	±	400 ±	85. At	±	±
39. Y	±	18 ±	86. Rn	±	±
40. Zr	±	120 ±	87. Fr	±	±
41. Nb	±	0.8 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	1.5 ±
44. Ru	±	±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	0.4 ±
46. Pd	±	±	93. Np	±	±
47. Ag	±	0.05 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	±	96. Cm	±	±
*CO ₂	2.22% ± 5%	±		±	±
Fe ²⁺ /Fe ³⁺	1.22 ± 5%	±		±	±

Method of Analysis:

*CO₂ From carbonate

**Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Altered porphyrites

Material basaltLRL Sample No. 1M 140517-CH-21 150.5-154.5'

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na	±	50. Sn	±	< 0.05
2. He	na	±	51. Sb	±	< 0.05
3. Li		±	52. Te	±	±
4. Be		±	53. I	±	0.1
5. B		±	55. Cs	±	0.8
6. C	na	±	56. Ba	±	400
7. N	na	±	57. La	±	4
8. O	na	±	58. Ce	±	8
9. F	na	±	59. Pr	±	1
11. Na	2.5	± 5%	60. Nd	±	3
12. Mg	3.73	± 2%	61. Pm	±	±
13. Al	9.08	± 2%	62. Sm	±	1
14. Si	25.54	± 2%	63. Eu	±	0.3
15. P	0.005	± 20%	64. Gd	±	1
16. S	0.012	± 10%	65. Tb	±	0.2
17. Cl	na	±	66. Dy	±	1
19. K	0.82	± 5%	67. Ho	±	0.4
20. Ca	5.33	± 2%	68. Er	±	0.5
21. Sc		±	69. Tm	na	±
22. Ti	0.55	± 5%	70. Yb	±	4
23. V		±	71. Lu	±	< 0.05
24. Cr		±	72. Hf	±	±
25. Mn	0.09	± 10%	73. Ta	na	±
26. Fe	5.37	± 5%	74. W	±	±
27. Co		±	75. Re	na	±
28. Ni		±	76. Os	±	±
29. Cu		±	77. Ir	±	±
30. Zn		±	78. Pt	±	±
31. Ga		±	79. Au	±	±
32. Ge	na	±	80. Hg	±	±
33. As		±	81. Tl	±	0.1
34. Se		±	82. Pb	±	2
35. Br		±	83. Bi	±	±
37. Rb		±	84. Po	±	±
38. Sr		±	85. At	±	±
39. Y		±	86. Rn	±	±
40. Zr		±	87. Fr	±	±
41. Nb		±	88. Ra	±	±
42. Mo		±	89. Ac	±	±
43. Tc		±	90. Th	±	±
44. Ru		±	91. Pa	±	±
45. Rh	na	±	92. U	±	±
46. Pd		±	93. Np	±	±
47. Ag		±	94. Pu	±	±
48. Cd		±	95. Am	±	±
49. In		±	96. Cm	±	±
*CO ₂	0.25	± 5%		±	±
Fe ⁺² /Fe ⁺³	0.58	± 5%		±	±

Method of Analysis:

*CO₂ from carbonate

**Expressed as a percent of the concentration.

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Tuffaceous siltstone
17-CH-22 500'LRL Sample No. 124

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	< 0.05 ±
2. He	na ±	±	51. Sb	±	< 0.05 ±
3. Li	±	6 ± 10%	52. Te	±	±
4. Be	na ±	±	53. I	±	±
5. B	±	5 ±	55. Cs	±	< 0.05 ±
6. C	na ±	±	56. Ba	±	145 ±
7. N	na ±	±	57. La	±	4 ±
8. O	na ±	±	58. Ce	±	8 ±
9. F	na ±	±	59. Pr	±	2 ±
11. Na	3.8 ± 5%	±	60. Nd	±	7 ±
12. Mg	2.35 ± 2%	±	61. Pm	±	±
13. Al	8.52 ± 2%	±	62. Sm	±	3 ±
14. Si	25.89 ± 2%	±	63. Eu	±	0.4 ±
15. P	0.052 ± 5%	±	64. Gd	±	2 ±
16. S	0.018 ± 5%	±	65. Tb	±	±
17. Cl	na ±	±	66. Dy	±	2 ±
19. K	0.81 ± 5%	±	67. Ho	±	0.4 ±
20. Ca	4.46 ± 2%	±	68. Er	±	1 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.89 ± 5%	±	70. Yb	±	1 ±
23. V	±	230 ±	71. Lu	±	< 0.05 ±
24. Cr	±	< 10 ±	72. Hf	±	1.5 ±
25. Mn	0.10 ± 10%	±	73. Ta	na ±	±
26. Fe	7.33 ± 5%	±	74. W	±	±
27. Co	±	12 ±	75. Re	na ±	±
28. Ni	±	10 ±	76. Os	±	±
29. Cu	±	65 ±	77. Ir	±	±
30. Zn	±	40 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.1 ±	81. Tl	±	±
34. Se	±	0.3 ±	82. Pb	±	0.5 ±
35. Br	±	0.5 ±	83. Bi	±	±
37. Rb	±	0.5 ±	84. Po	±	±
38. Sr	±	105 ±	85. At	±	±
39. Y	±	20 ±	86. Rn	±	±
40. Zr	±	29 ±	87. Fr	±	±
41. Nb	±	0.2 ±	88. Ra	±	±
42. Mo	±	±	89. Ac	±	±
43. Tc	±	±	90. Th	±	0.1 ±
44. Ru	±	6 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	0.05 ±	93. Np	±	±
47. Ag	±	±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
* CO ₂ +3	0.07 ± 10%	±	Free H ₂ O	1.48 ± 5%	±
+2/Fe	1.54 ± 5%	±	Bound H ₂ O	2.18 ± 5%	±

Method of Analysis:

*CO₂ from carbonate

** Expressed as a percent of the concentration

na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

ANALYTICAL REPORT

Material Fresh altered tuff
17-CH-22 1443'LRL Sample No. 132

Z	Weight %	Weight PPM	Z	Weight %	Weight PPM
1. H	na ± **	±	50. Sn	±	< 0.05 ±
2. He	na ±	±	51. Sb	±	0.1 ±
3. Li	±	13 ± 10%	52. Te	±	< 0.05 ±
4. Be	na ±	±	53. I	±	±
5. B	±	10 ± 50%	55. Cs	±	0.1 ±
6. C	na ±	±	56. Ba	±	50 ±
7. N	na ±	±	57. La	±	1 ±
8. O	na ±	±	58. Ce	±	3 ±
9. F	na ±	±	59. Pr	±	0.5 ±
11. Na	0.80 ± 5%	±	60. Nd	±	3 ±
12. Mg	6.01 ± 2%	±	61. Pm	±	±
13. Al	6.90 ± 2%	±	62. Sm	±	2 ±
14. Si	23.00 ± 2%	±	63. Eu	±	0.4 ±
15. P	0.025 ± 5%	±	64. Gd	±	1 ±
16. S	0.012 ± 10%	±	65. Tb	±	0.3 ±
17. Cl	na ±	±	66. Dy	±	1 ±
19. K	0.41 ± 5%	±	67. Ho	±	0.3 ±
20. Ca	5.76 ± 2%	±	68. Er	±	1 ±
21. Sc	na ±	±	69. Tm	na ±	±
22. Ti	0.82 ± 5%	±	70. Yb	±	1 ±
23. V	±	310 ±	71. Lu	±	±
24. Cr	±	400 ± 50%	72. Hf	±	0.3 ±
25. Mn	0.04 ± 20%	±	73. Ta	na ±	±
26. Fe	6.49 ± 5%	±	74. W	±	±
27. Co	±	12 ±	75. Re	na ±	±
28. Ni	±	70 ±	76. Os	±	±
29. Cu	±	24 ±	77. Ir	±	±
30. Zn	±	37 ±	78. Pt	±	±
31. Ga	na ±	±	79. Au	±	±
32. Ge	na ±	±	80. Hg	±	±
33. As	±	0.4 ±	81. Tl	±	±
34. Se	±	1 ±	82. Pb	±	0.4 ±
35. Br	±	1 ±	83. Bi	±	±
37. Rb	±	0.7 ±	84. Po	±	±
38. Sr	±	420 ±	85. At	±	±
39. Y	±	18 ±	86. Rn	±	±
40. Zr	±	26 ±	87. Fr	±	±
41. Nb	±	0.2 ±	88. Ra	±	±
42. Mo	±	2 ±	89. Ac	±	±
43. Tc	±	±	90. Th	±	±
44. Ru	±	3 ±	91. Pa	±	±
45. Rh	na ±	±	92. U	±	±
46. Pd	±	< 0.05 ±	93. Np	±	±
47. Ag	±	0.1 ±	94. Pu	±	±
48. Cd	±	±	95. Am	±	±
49. In	±	< 0.05 ±	96. Cm	±	±
*CO ₂	0.47 ± 5%	±	Free H ₂ O	7.74 ± 5%	±
Fe ⁺² /Fe ⁺²⁺³	0.90 ± 5%	±	Bound H ₂ O	6.86 ± 5%	±

Method of Analysis:

*CO₂ from carbonate** Expressed as a percent of the concentration
na - not analyzed

Analytical Lab. _____

Lab. Sample No. _____

Date: _____

Signed: _____

Appendix B

CHEMICAL ANALYSIS OF CORE SAMPLES BY CORPS OF ENGINEERS

These data are reproduced from an IOCS Memorandum on the subsurface geology of Route 17.*

*See Ref. 1.

PART IV

CHEMICAL TESTS

<u>Rock type</u>	<u>Boring number</u>	<u>Depth (feet)</u>	<u>Lab number</u>	<u>Sample description</u>
1	17-CH-4	113	222/183	Calcareous shale with silty interbeds
1	17-CH-4	383	1M1310	Calcareous shale
1	17-CH-13	321	222/64	Moderately weathered calcareous shale
1	17-CH-13	789	1M1316	Fresh shale
1	17-CH-17	113	222/159	Fresh fossiliferous shale
2	17-CH-5	400	1M1311	Bedded tuff
2	17-CH-7	675	1M1312	Calcareous tuff
2	17-CH-8	261-263	1M1177	Tuffaceous fossiliferous limestone
2	17-CH-5	257-258	1M1006	Fresh calcareous tuff
3	17-CH-5	197-198	1M1200	Slightly weathered silty and tuffaceous limestone
3	17-CH-10	364-365	1M1213 & 1M1215	Fresh tuffaceous limestone
3	17-CH-10	500	1M1314	Fresh tuffaceous limestone
4	17-CH-9	460-462	1M1043	Fresh tuffaceous limestone
4	17-CH-9	680	1M1313	Slightly weathered siliceous shale
4	17-CH-13	207-208	1M1248	Fresh tuffaceous limestone
5	17-CH-12	288-291	1M1073	Agglomerate
5	17-CH-12	383-384	1M1328	Agglomerate with tuffaceous matrix

Chemical tests - continued

<u>Rock type</u>	<u>Boring number</u>	<u>Depth (feet)</u>	<u>Lab number</u>	<u>Sample description</u>
6	17-CH-12	117-118	1M1262 & 1M1264	Fresh tuffaceous sandstone
6	17-CH-12	180-182	1M1281	Fresh tuffaceous siltstone
6	17-CH-12	970	1M1315	Fresh tuffaceous sandstone
6	17-CH-15	418-419	1M1379 & 1M1381	Fresh calcareous tuffaceous sandstone
7	17-CH-7	104-107	1M1089	Slightly weathered and altered diabase
7	17-CH-7	652-654	1M1093	Slightly weathered altered diabase
8	17-CH-11	85-86	1M1411	Fresh to slightly weathered diabase
8	17-CH-11	497-499	1M1419	Porphyritic amygdaloidal altered and slightly weathered basalt
8	17-CH-11	897-899	1M1113	Altered and weathered porphyritic basalt
8	17-CH-21	151-155	1M1405	Altered porphyritic basalt
9	17-CH-15	276-277	1M1461 & 1M1465	Fresh calcareous tuff
10	17-CH-11	926-929	1M1475	Fresh altered tuff
10	17-CH-14	577-578	1M1117	Fresh tuff
10	17-CH-14	600-604	1M1131	Interbedded tuffaceous agglomerate and sandstone
11	17-CH-14	489-491	1M1147 & 1M1144	Slightly weathered lapilli tuff
12	17-CH-22	1143	1M1317	Fresh altered tuff

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Magnesium as MgO - - - - -	6.45 6.31	8.88 8.88																																																																																	
Sodium as Na ₂ O - - - - -	1.52 1.32	0.74 0.84																																																																																	
Potassium as K ₂ O - - - - -	0.32 0.28	0.00 0.00																																																																																	
Water, Crystallization - - - - -	4.54 4.49	5.51 5.70																																																																																	
Water, Hygroscopic - - - - -	4.26 4.46	6.44 6.50																																																																																	
Carbonates as CO ₂ - - - - -	0.26 0.26	0.14 0.20																																																																																	
Ignition Loss (850°C) - - - - -	7.34 7.24	10.74 11.16																																																																																	
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Lab. No. ----- <u>1M1147-1M1144</u> Core Hole ----- <u>17-CH-14</u> Depth (ft.) ----- <u>489-491</u> Rock Type ----- <u>11*</u> Run ----- <u>A B</u> <table style="width: 100%; border-collapse: collapse;"> <tr><td style="width: 60%;">Silicon as SiO₂ -----</td><td style="width: 20%; text-align: right;">48.07</td><td style="width: 20%; text-align: right;">48.00</td></tr> <tr><td>Aluminum as Al₂O₃ -----</td><td style="text-align: right;">16.32</td><td style="text-align: right;">16.42</td></tr> <tr><td>Iron as Fe₂O₃ -----</td><td style="text-align: right;">7.85</td><td style="text-align: right;">7.77</td></tr> <tr><td>Manganese as MnO -----</td><td style="text-align: right;">0.10</td><td style="text-align: right;">0.11</td></tr> <tr><td>Titanium as TiO₂ -----</td><td style="text-align: right;">0.50</td><td style="text-align: right;">0.53</td></tr> <tr><td>Phosphorus as P₂O₅ -----</td><td style="text-align: right;">0.07</td><td style="text-align: right;">0.07</td></tr> <tr><td>Calcium as CaO -----</td><td style="text-align: right;">8.30</td><td style="text-align: right;">8.25</td></tr> <tr><td>Magnesium as MgO -----</td><td style="text-align: right;">6.12</td><td style="text-align: right;">6.14</td></tr> <tr><td>Sodium as Na₂O -----</td><td style="text-align: right;">1.28</td><td style="text-align: right;">1.40</td></tr> <tr><td>Potassium as K₂O -----</td><td style="text-align: right;">0.56</td><td style="text-align: right;">0.64</td></tr> <tr><td>Water, Crystallization -----</td><td style="text-align: right;">5.31</td><td style="text-align: right;">5.12</td></tr> <tr><td>Water, Hygroscopic -----</td><td style="text-align: right;">6.26</td><td style="text-align: right;">6.42</td></tr> <tr><td>Carbonates as CO₂ -----</td><td style="text-align: right;">0.00</td><td style="text-align: right;">0.00</td></tr> <tr><td>Ignition Loss (850°C) -----</td><td style="text-align: right;">9.38</td><td style="text-align: right;">9.36</td></tr> </table>				Silicon as SiO ₂ -----	48.07	48.00	Aluminum as Al ₂ O ₃ -----	16.32	16.42	Iron as Fe ₂ O ₃ -----	7.85	7.77	Manganese as MnO -----	0.10	0.11	Titanium as TiO ₂ -----	0.50	0.53	Phosphorus as P ₂ O ₅ -----	0.07	0.07	Calcium as CaO -----	8.30	8.25	Magnesium as MgO -----	6.12	6.14	Sodium as Na ₂ O -----	1.28	1.40	Potassium as K ₂ O -----	0.56	0.64	Water, Crystallization -----	5.31	5.12	Water, Hygroscopic -----	6.26	6.42	Carbonates as CO ₂ -----	0.00	0.00	Ignition Loss (850°C) -----	9.38	9.36
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		CONTRACT NO.																																											
		DATE REPORTED 31 July 1968																																											
DESCRIPTION See below		WORK ORDER NO. 4965																																											
SOURCE See below		RECON. NO. 06-123-ENG-108-680																																											
FOR USE AS:		BASE UNIT COST																																											
TESTED FOR: Chemical Analysis (% by weight)		DATE SAMPLE RECEIVED																																											
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Lab. No. ----- 1M1317 Core Hole ----- 17-CH-22 Depth (ft.) ----- 1443 Rock Type ----- 12* Run ----- A B <table style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 50%;">Silicon as SiO₂ -----</td> <td style="width: 25%; text-align: right;">43.28</td> <td style="width: 25%; text-align: right;">43.52</td> </tr> <tr> <td>Aluminum as Al₂O₃ -----</td> <td style="text-align: right;">11.58</td> <td style="text-align: right;">12.15</td> </tr> <tr> <td>Iron as Fe₂O₃ -----</td> <td style="text-align: right;">7.98</td> <td style="text-align: right;">7.40</td> </tr> <tr> <td>Manganese as MnO -----</td> <td style="text-align: right;">0.14</td> <td style="text-align: right;">0.13</td> </tr> <tr> <td>Titanium as TiO₂ -----</td> <td style="text-align: right;">0.74</td> <td style="text-align: right;">0.70</td> </tr> <tr> <td>Phosphorus as P₂O₅ -----</td> <td style="text-align: right;">0.11</td> <td style="text-align: right;">0.10</td> </tr> <tr> <td>Calcium as CaO -----</td> <td style="text-align: right;">7.20</td> <td style="text-align: right;">7.22</td> </tr> <tr> <td>Magnesium as MgO -----</td> <td style="text-align: right;">10.48</td> <td style="text-align: right;">10.25</td> </tr> <tr> <td>Sodium as Na₂O -----</td> <td style="text-align: right;">0.55</td> <td style="text-align: right;">0.55</td> </tr> <tr> <td>Potassium as K₂O -----</td> <td style="text-align: right;">0.60</td> <td style="text-align: right;">0.65</td> </tr> <tr> <td>Water, Crystallization -----</td> <td style="text-align: right;">8.91</td> <td style="text-align: right;">8.52</td> </tr> <tr> <td>Water, Hygroscopic -----</td> <td style="text-align: right;">7.11</td> <td style="text-align: right;">7.13</td> </tr> <tr> <td>Carbonates as CO₂ -----</td> <td style="text-align: right;">0.76</td> <td style="text-align: right;">0.59</td> </tr> <tr> <td>Ignition Loss (850°C) -----</td> <td style="text-align: right;">15.02</td> <td style="text-align: right;">14.65</td> </tr> </table>				Silicon as SiO ₂ -----	43.28	43.52	Aluminum as Al ₂ O ₃ -----	11.58	12.15	Iron as Fe ₂ O ₃ -----	7.98	7.40	Manganese as MnO -----	0.14	0.13	Titanium as TiO ₂ -----	0.74	0.70	Phosphorus as P ₂ O ₅ -----	0.11	0.10	Calcium as CaO -----	7.20	7.22	Magnesium as MgO -----	10.48	10.25	Sodium as Na ₂ O -----	0.55	0.55	Potassium as K ₂ O -----	0.60	0.65	Water, Crystallization -----	8.91	8.52	Water, Hygroscopic -----	7.11	7.13	Carbonates as CO ₂ -----	0.76	0.59	Ignition Loss (850°C) -----	15.02	14.65
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