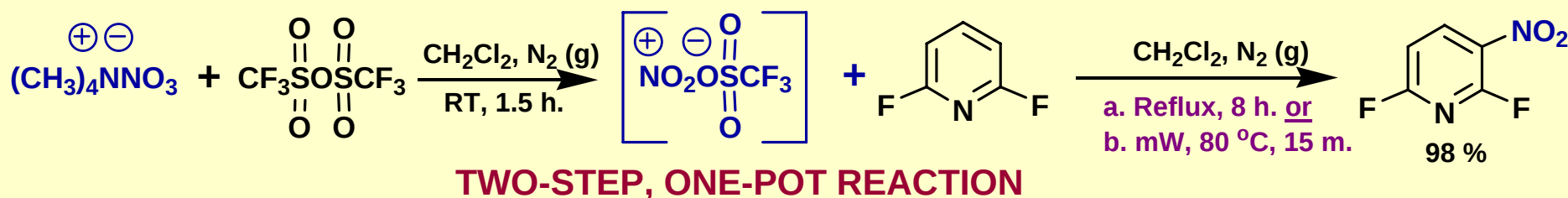




ANHYDROUS NITRONIUM TRIFLATE NITRATION OF AROMATIC and HETEROAROMATIC COMPOUNDS: CONVENTIONAL BENCHTOP & MICROWAVE- ASSISTED CONDITIONS

Scott A. Shackelford, Wade W. Grabow, Ashwani Vij

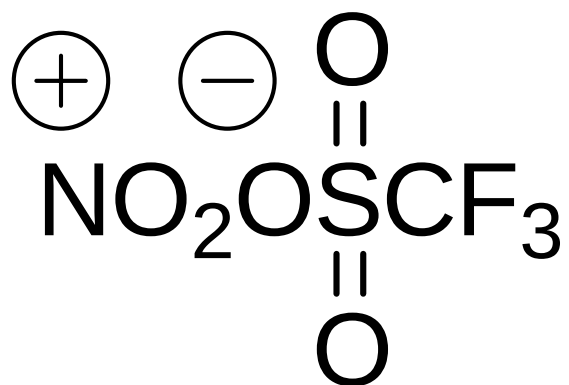
Air Force Research Laboratory
Propellants Branch
10 East Saturn Drive
Edwards AFB, CA 93524-7680



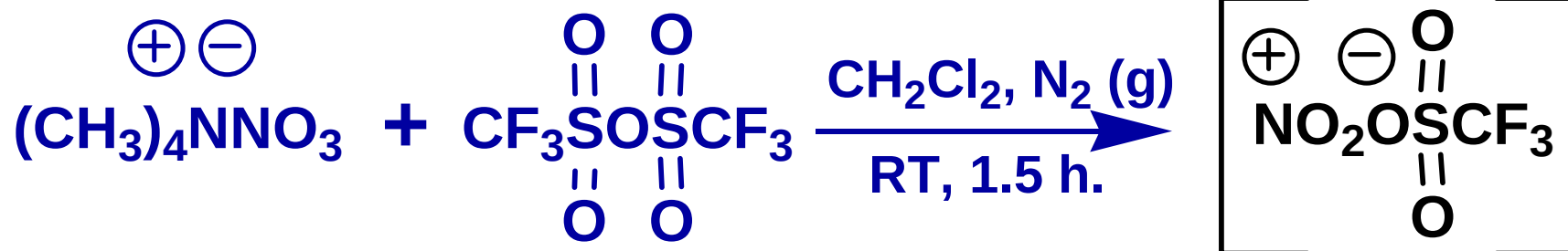
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ANHYDROUS NITRONIUM TRIFLATE NITRATION



CONVENIENT IN-SITU GENERATION





OVERVIEW



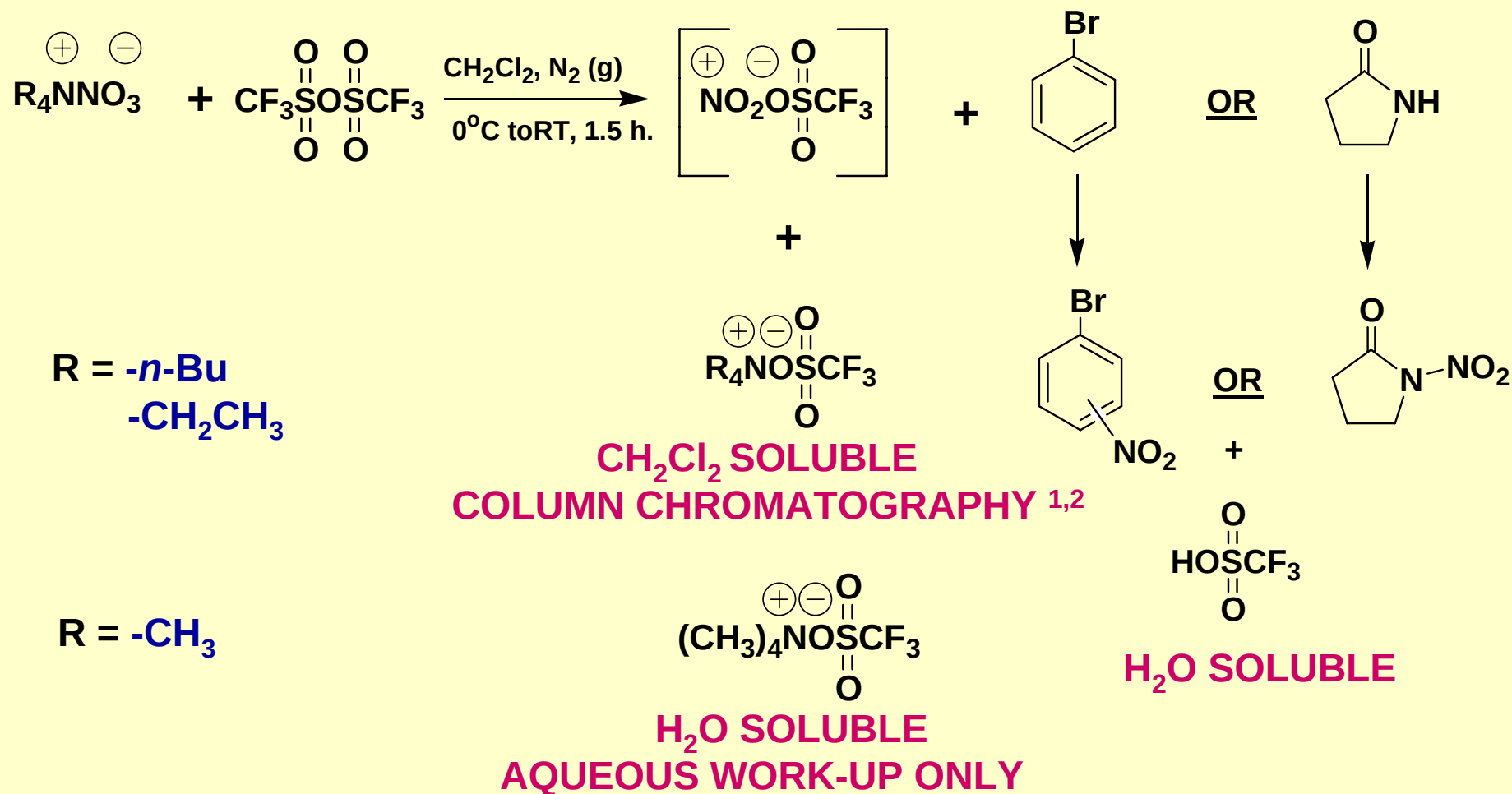
- IMPROVED $\text{NO}_2\text{OSO}_2\text{CF}_3$ NITRATING REAGENT *IN-SITU* FORMATION
 - $(n\text{-Bu})_4\text{NNO}_3$ GENERATED REAGENT
 - $(\text{CH}_3)_4\text{NNO}_3$ GENERATED REAGENT

- CONVENTIONAL BENCHTOP & MICROWAVE-ASSISTED $\text{NO}_2\text{OSO}_2\text{CF}_3$ NITRATION RESULTS
 - CONVENTIONAL BENCHTOP LIMITATIONS
 - MICROWAVE-ASSISTED EXTENTIONS

- PRELIMINARY MICROWAVE-ASSISTED $\text{NO}_2\text{OSO}_2\text{CF}_3$ NITRATION SCREENING REACTIONS
 - DIRECTIONAL EFFECTS OF 5-MEMBERED HETEROAROMATICS
 - POSITIONAL NITRATION REACTIVITY



ANHYDROUS NITRONIUM TRIFLATE NITRATION



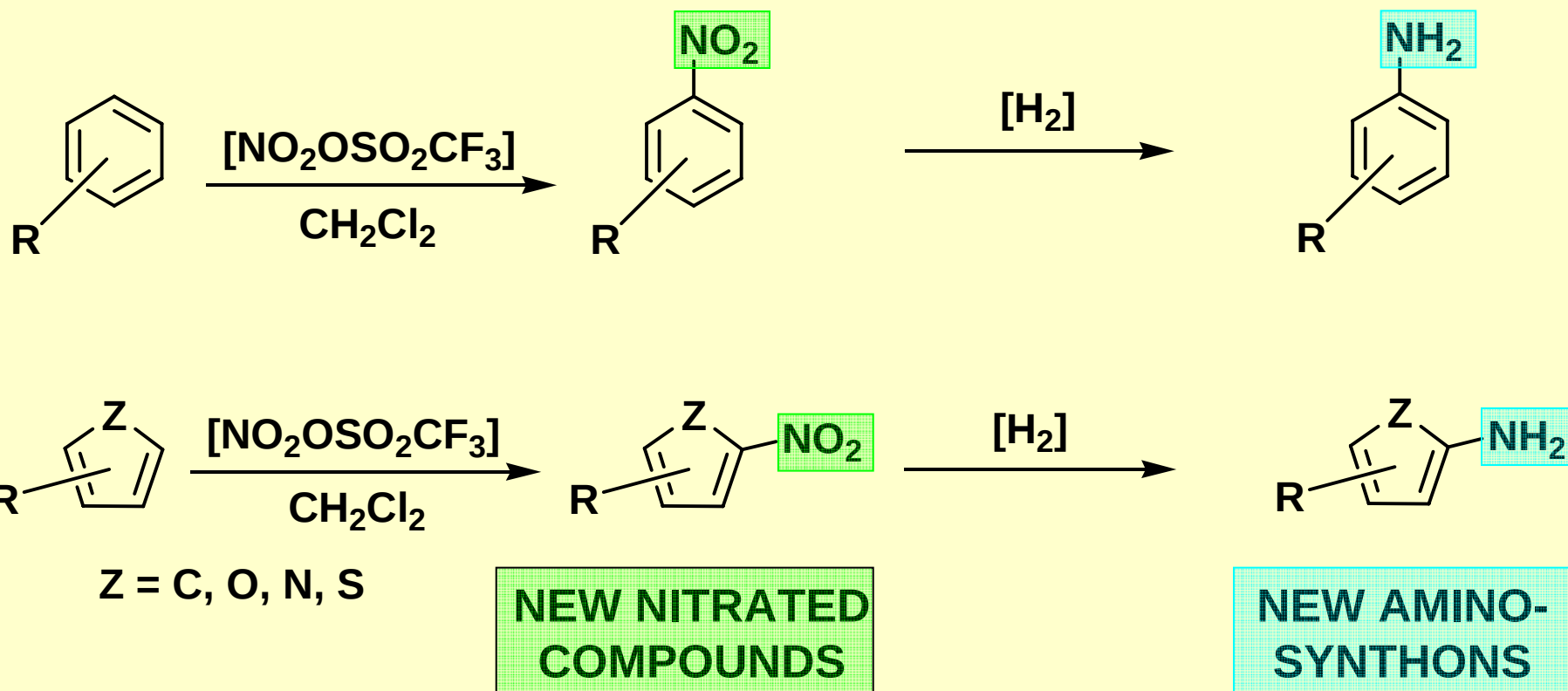
1. Shackelford, S. A.; Adams, C. M.; Sharts, C. M. 11th Rocky Mountain Region ACS Mtg., Albuquerque, NM, 10-12 Jul 1992.
2. Adams, C. M.; Sharts, C. M.; Shackelford, S. A. *Tetrahedron Lett.* **1993**, 34, 6669-6671.



ANHYDROUS NITRONIUM TRIFLATE NITRATION

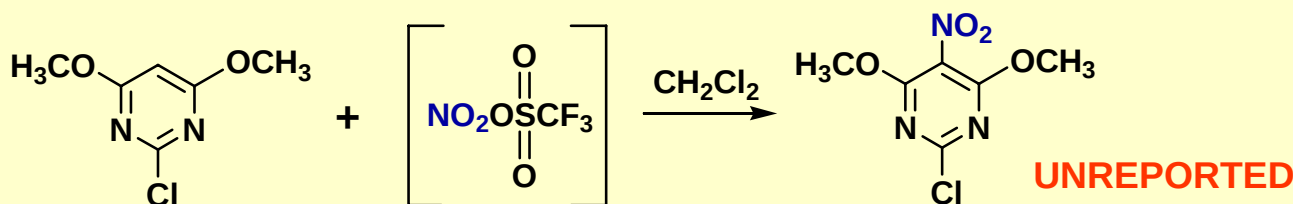


OBJECTIVE: NEW AROMATICS & HETEROAROMATICS

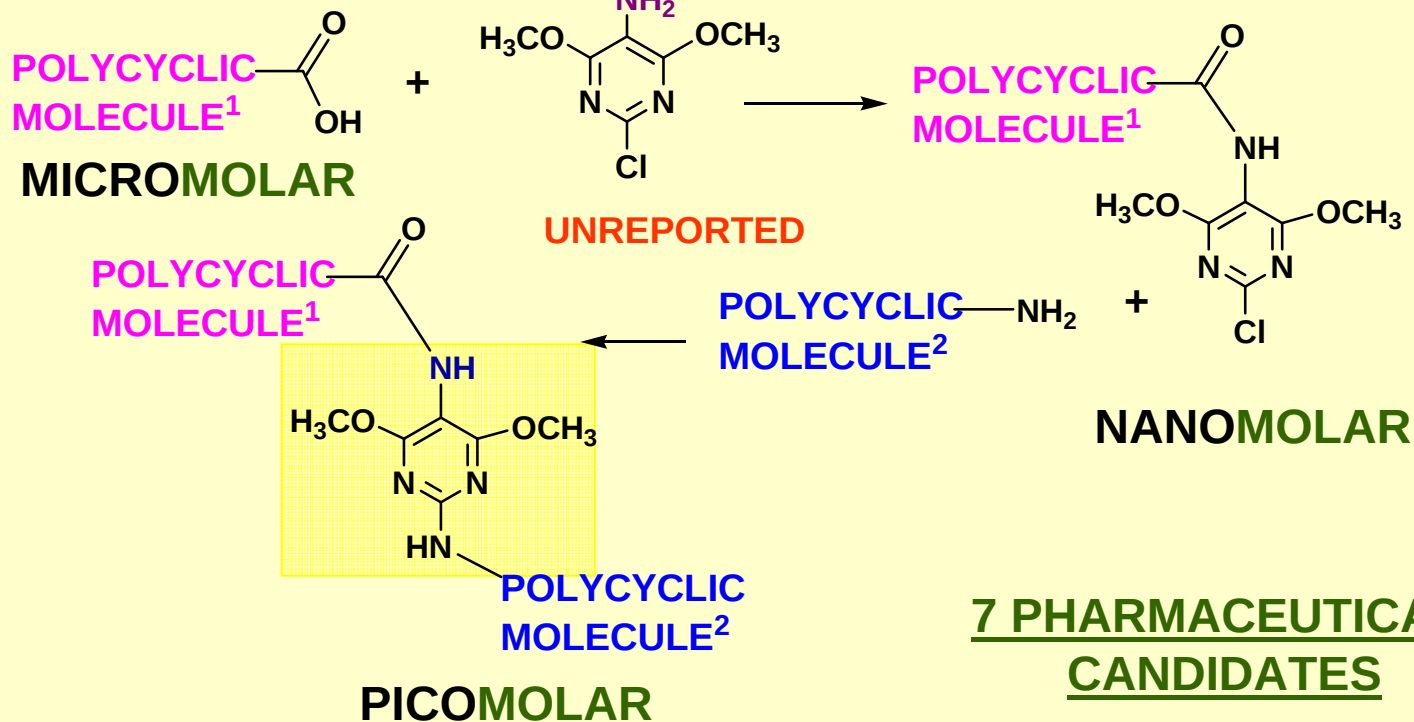




ANHYDROUS NITRONIUM TRIFLATE NITRATION



**BIOASSAY
RESPONSES**

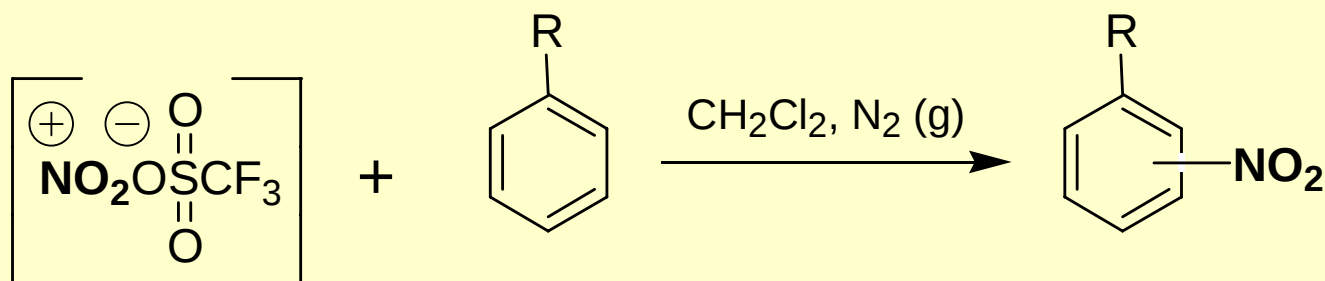




ANHYDROUS NITRONIUM TRIFLATE NITRATION



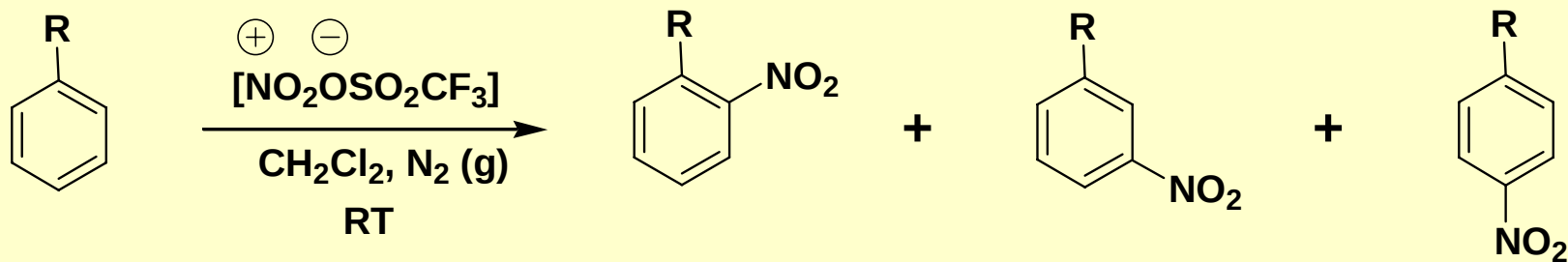
➤ NITRATION CONDITIONS:



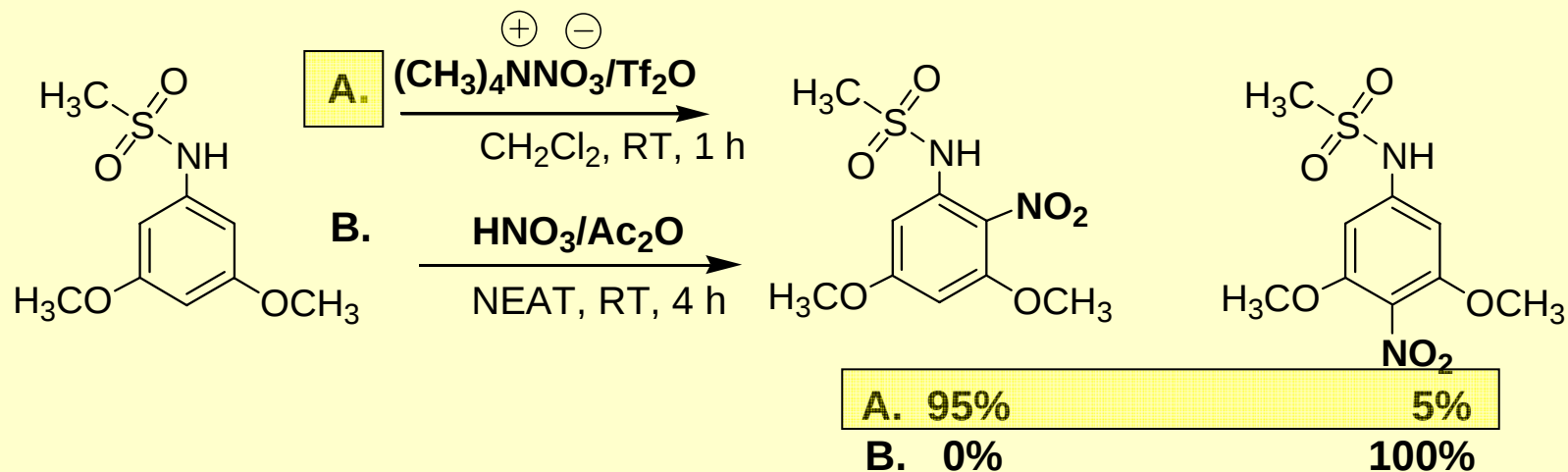
	<u>EQUIVALENTS</u>	<u>CONCENTRATION</u>	<u>TEMPERATURE</u>	<u>METHOD</u>
<u>DILUTE:</u>	1.05 to 1.7	0.2 to 0.3M	-78 °C to RT	CONVENTIONAL BENCHTOP
<u>CONC.:</u>	1.5 to 2.0	1.1 to 3.6M	RT to >REFLUX	CONVENTIONAL BENCHTOP & MICROWAVE- ASSISTED



CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION



R = -OCH ₃	63%	5%	32%
-OH	90	0	10
-CH ₃	62	0	38
-Br	33	0	67
-CO ₂ H	16	78	6
-CF ₃	8%	88%	4%



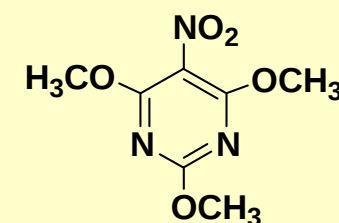
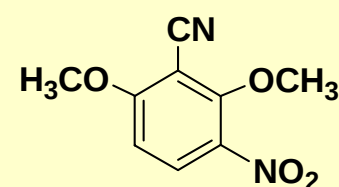
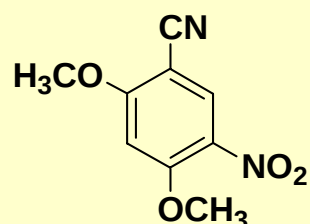
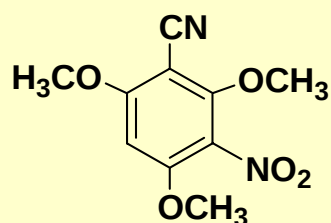
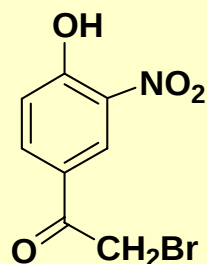


CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION



➤ NITRATION RESULTS:

PRODUCT:



UNREPORTED

ONLY ISOMER

DIRECT NITRATION

SCALE

(mmol):

8 93

8 100

100

100

371

ISOLATED

YIELD:

94% 91%

86% 82%

91%

100%

82%

PURITY:

91% 88%

95% 95%

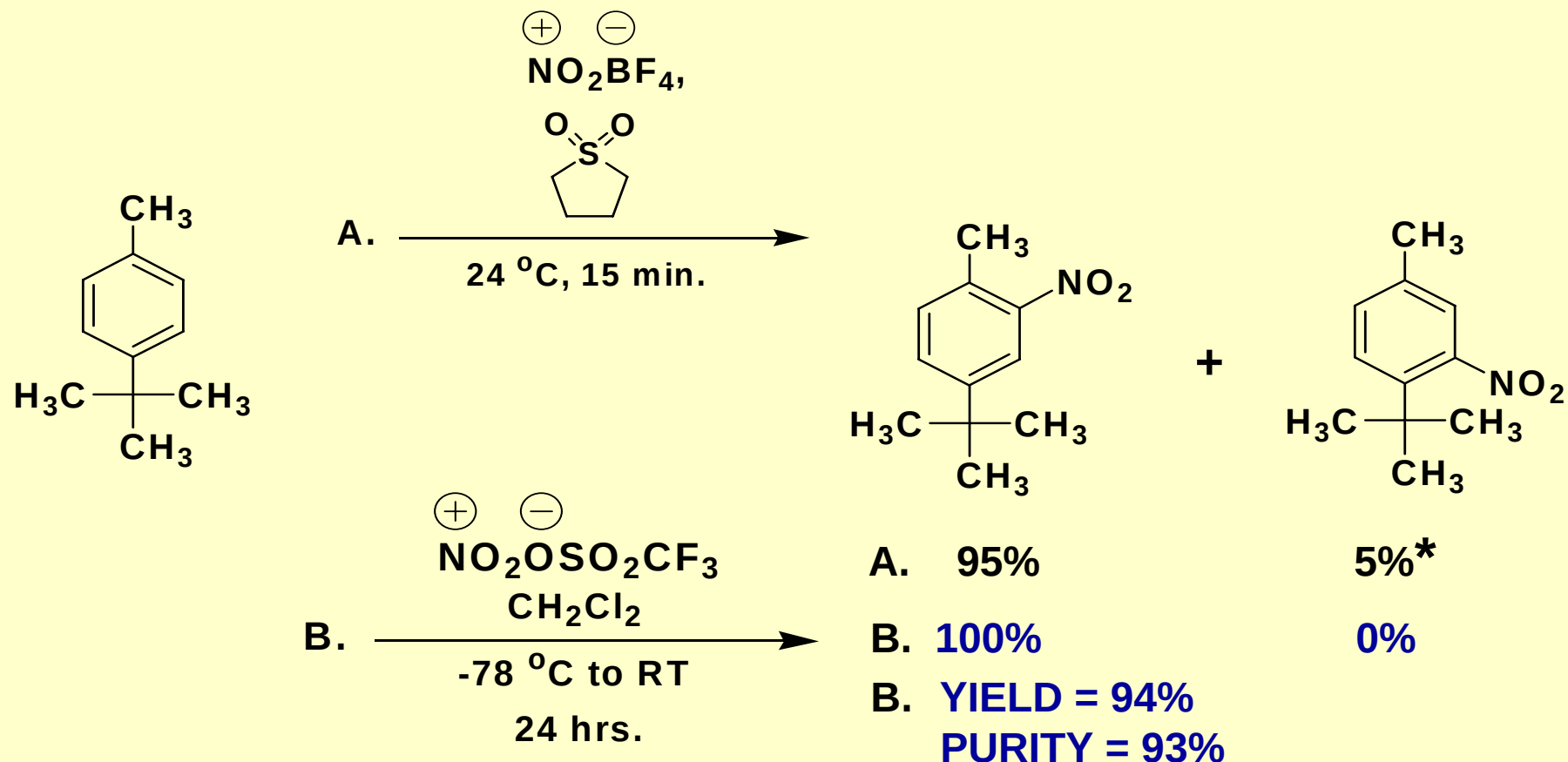
82%

97%

98%



➤ NITRATION SELECTIVITY (82 mmol):



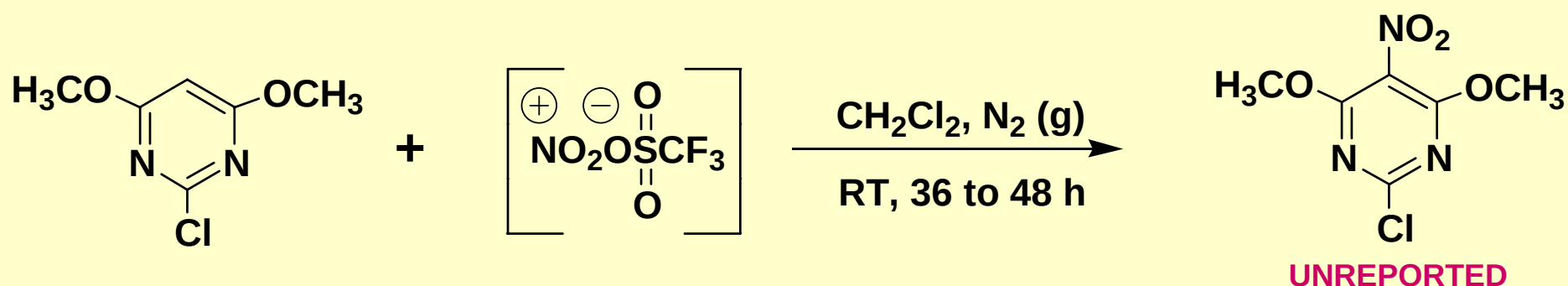
* Olah, G. A.; Kuhn, S. *J. Am. Chem. Soc.*, **1964**, 86, 1067-1070.



CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION



➤ $\text{NO}_2\text{OSO}_2\text{CF}_3$ EQUIVALENTS & NITRATION SCALE-UP:



<u>CONDITIONS</u>	<u>SCALE (mmol)</u>	<u>$\text{NO}_2\text{OSO}_2\text{CF}_3$ EQUIV.</u>	<u>% CONVERTED</u>	<u>ISOLATED YIELD.</u>	<u>PURITY</u>
[0.2M], 48 h	8	1.05	68	-----	68%
	8	1.25	88	-----	88%
	8	1.50	100	98%	99%*
[3.6M], 36 h	2864	1.5	100	94%	>95%

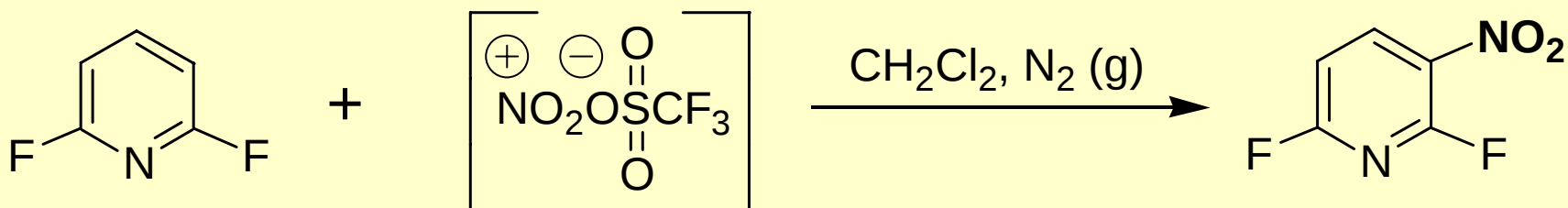
* ANALYTICALLY PURE BY C, H, Cl, and N, ELEMENTAL ANALYSES



CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION



➤ NITRATION REACTIVITY:



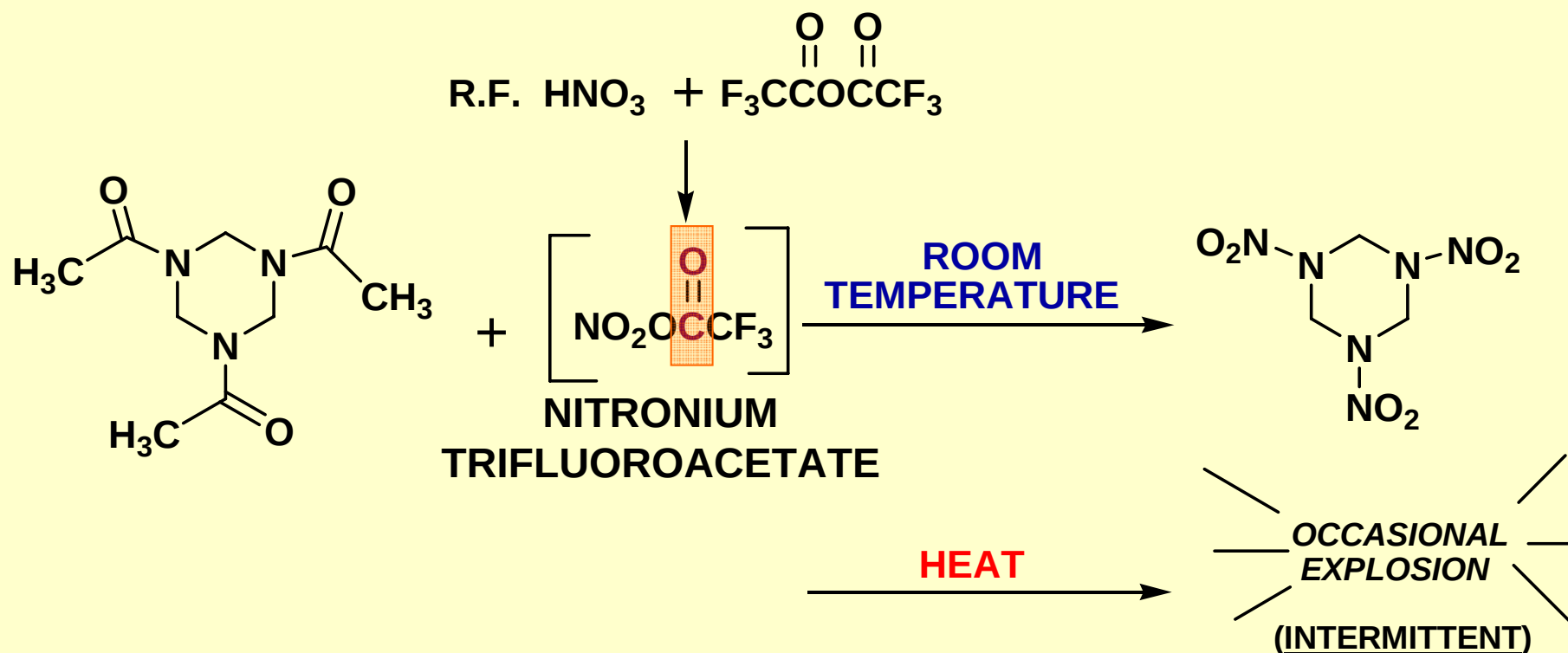
<u>CONDITIONS</u>	<u>NO₂OSO₂CF₃ EQUIV.</u>	<u>[REACTANT CONC.]</u>	<u>% CONVERTED</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>
RT, 42 h	1.6	0.25M	69%	-----	<69%
RT, 24 h	1.5	1.09M	78%	-----	78%



CONVENTIONAL BENCHTOP NITRONIUM TRIFLUOROACETATE NITRATION



➤ OTHER NITRATION REAGENT REACTIVITY:





MICROWAVE-ASSISTED NITRONIUM TRIFLATE NITRATION



➤ ORIGINAL MICROWAVE INSTRUMENTS EMPLOYED:

■ PERSONAL MICROWAVE™

SMALL SCALE REACTIONS
3.4 mmol (0.5 gram)



■ MILESTONE ETHOS™

LARGER-SCALE REACTIONS
54.5 mmol (10 grams)

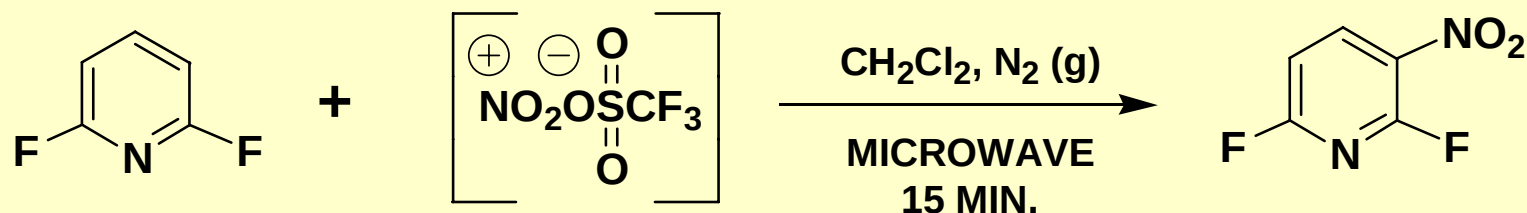




MICROWAVE-ASSISTED NITRONIUM TRIFLATE NITRATION



➤ MICROWAVE TEMPERATURE SCREENING:



<u>RUN NO.</u>	<u>NO₂OSO₂CF₃ EQUIV.</u>	<u>TEMP (°C)</u>	<u>% CONVERTED</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>
1	1.3	60	91	-----	-----
8	1.5	60	96	85%	-----
5	1.5	110	100	91%	98%
7	1.5	100	100	91%	98%
9	1.5	80	100	94%	100%*

*ANALYTICALLY PURE BY C, H, F, & N ELEMENTAL ANALYSIS

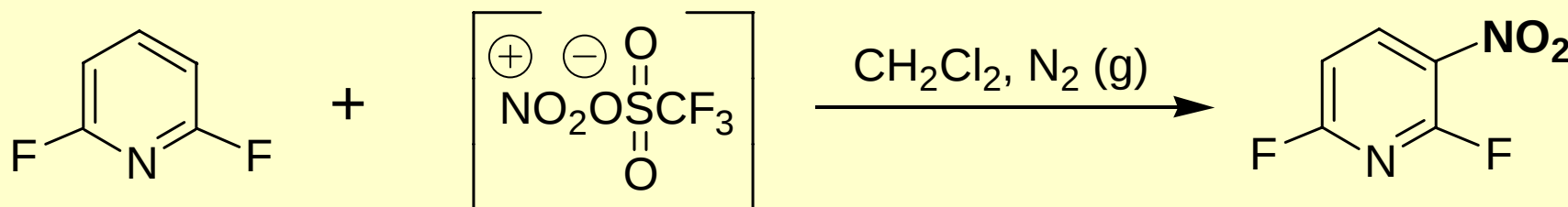
- INCREASED NO₂OSO₂CF₃ EQUIVALENTS IMPROVE REACTANT CONVERSION (RUNS 1 & 8)
- INCREASED TEMPERATURE RAISES REACTANT CONVERSION (RUN 5)
- MINIMUM TEMPERATURE (80 °C) EXISTS FOR COMPLETE REACTION (RUN 9)



CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION



➤ REFLUX TEMPERATURE NITRATION:



<u>CONDITIONS</u>	<u>$\text{NO}_2\text{OSO}_2\text{CF}_3$ EQUIV.</u>	<u>[REACTANT CONC.]</u>	<u>% CONVERTED</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>
RT, 42 h	1.6	0.25M	69%	-----	<69%
RT, 24 h	1.5	1.09M	78%	98%	78%
REFLUX, 8 h	1.5	1.09M	100%	98%	99%*

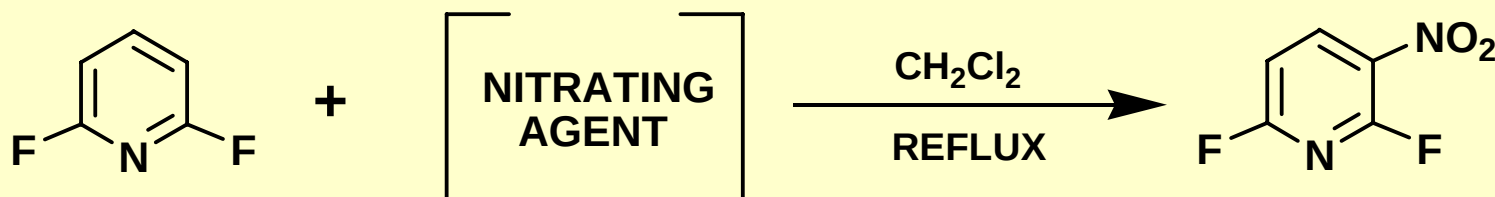
*ANALYTICALLY PURE BY C, H, F, and N, ELEMENTAL ANALYSES



CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION



➤ $\text{NO}_2\text{OSO}_2\text{CF}_3$ NITRATION EFFECTIVENESS:



	<u>TIME (h)</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>
$[\text{NO}_2\text{BF}_4^{\text{A}}]$	24	20%	-----
$[\text{NO}_2\text{OSO}_2\text{CF}_3]$	8	98%	99%*
*ANALYTICALLY PURE BY C, H, F, and N ELEMENTAL ANALYSES			
$[\text{HNO}_3 (100\%) / \text{H}_2\text{SO}_4^{\text{B}}]$	23	78% ^C	-----
^B NEAT, 0 °C to RT		^C DISTILLED YIELD	

A. Duffy, J. L.; Laali, K. K. *J. Org. Chem.* **1991**, 56, 3006-3009.

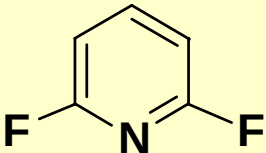
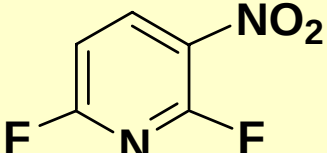
B. Mutterer, F.; Weiss, C. D. *Helv. Chem. Acta* **1976**, 59, 229-235.



CONVENTIONAL BENCHTOP NITRONIUM TRIFLATE NITRATION

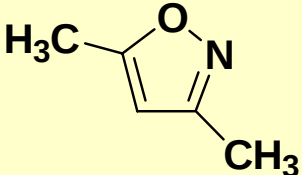
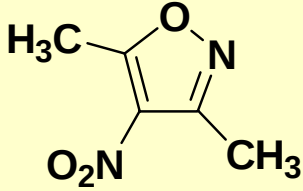


➤ REFLUX REACTION TIMES:

<u>REACTANT</u>	<u>TIME (h)</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>	<u>PRODUCT</u>
	8*	98%	99%**	
	24	94%	99%**	

* MINIMUM REACTION TIME

** ANALYTICALLY PURE BY C, H, F, and N ELEMENTAL ANALYSES

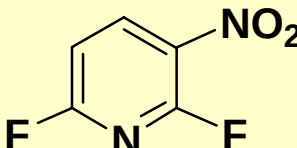
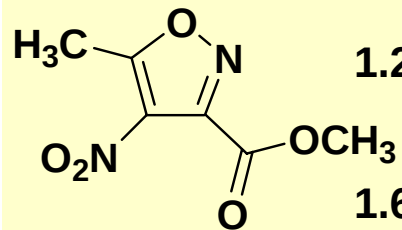
	3.5	92%	95%	
	16	90%	95%	



CONVENTIONAL VS. MICROWAVE $\text{NO}_2\text{OSO}_2\text{CF}_3$ NITRATION COMPARISON



➤ COMPARATIVE PRODUCT RESULTS:

	<u>CONC.</u>	<u>CONDITIONS</u>	<u>SCALE (mmol)</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>
	1.2M	PERSONAL MICROWAVE 80 °C, 15 min.	3.4	94%	100%*
	1.3M	MILESTONE MICROWAVE 80 °C, 15 min.	54.5	98%	98%*
	1.1M	CONVENTIONAL BENCHTOP REFLUX, 8 h	16.4	98%	99%*
	1.2M	PERSONAL MICROWAVE 60 °C, 15 min.	3.4	100%	99%*
	1.6M	MILESTONE MICROWAVE 60 °C, 15 min.	27.0	100%	99%*
	UNREPORTED				
	1.2M	CONVENTIONAL BENCHTOP REFLUX, 24 h	17.2	100%	99%**

*ANALYTICALLY PURE

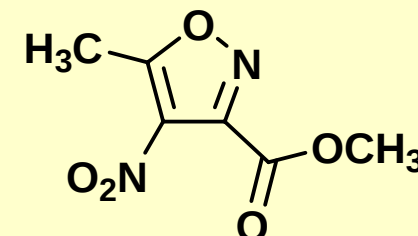
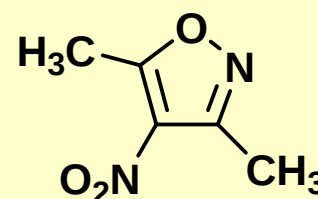
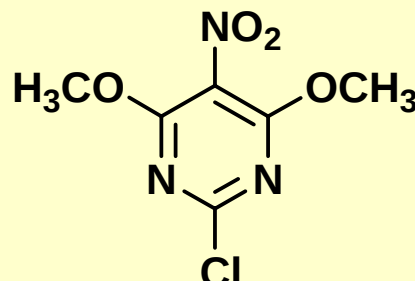
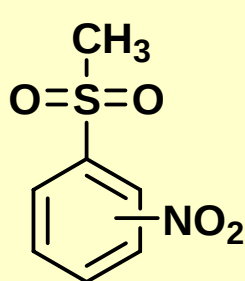
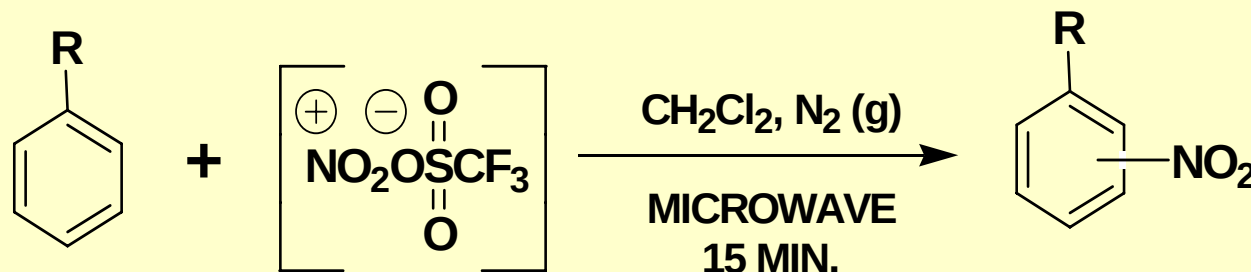
**ELEMENTAL ANALYSIS NOT CONDUCTED



MICROWAVE-ASSISTED NITRONIUM TRIFLATE NITRATION



➤ MICROWAVE-ASSISTED PRODUCT RESULTS:



TEMP.°C)

80

60

60

60

ISOLATED
YIELD

94%

91%

92%

100%

(O/M/P = 17/79/4)

PURITY

100%*

100%**

100%**

100%**

*ISOMERIC MIXTURE ANALYTICALLY PURE

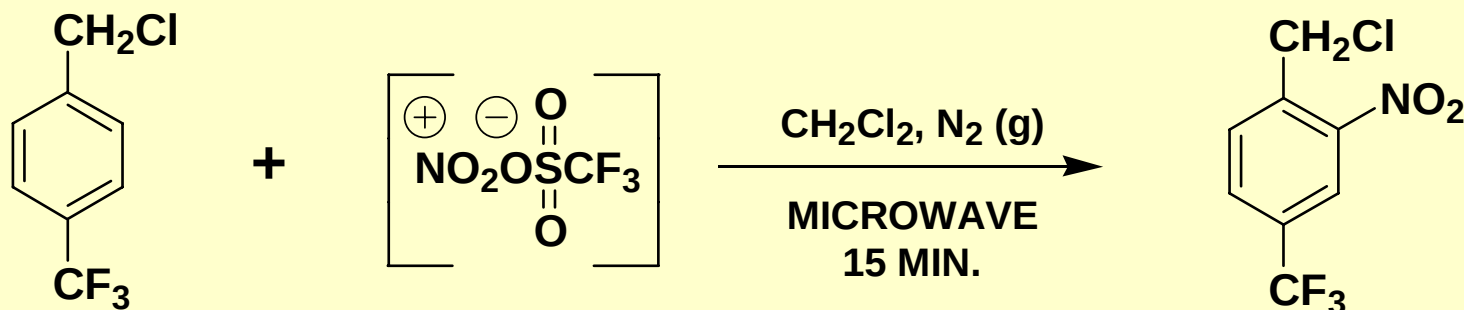
**ANALYTICALLY PURE



MICROWAVE-ASSISTED NITRONIUM TRIFLATE NITRATION



➤ MICROWAVE-ASSISTED PRODUCT RESULTS:



<u>RUN NO.</u>	<u>$\text{NO}_2\text{OSO}_2\text{CF}_3$ EQUIV.</u>	<u>TEMP (°C)</u>	<u>% CONVERTED</u>	<u>ISOLATED YIELD</u>	<u>PURITY</u>
1	1.5	80	82	-----	-----
2	1.5	100	82	-----	-----
3	2.0	80	100	98%	95%

- MINIMUM CONVERSION TEMPERATURE AT LEAST 80 °C (RUNS 1 & 2)
- $\text{NO}_2\text{OSO}_2\text{CF}_3$ EQUIVALENT INCREASE NEED FOR COMPLETE CONVERSION (RUN 3)



MICROWAVE & CONVENTIONAL $\text{NO}_2\text{OSO}_2\text{CF}_3$ NITRATION

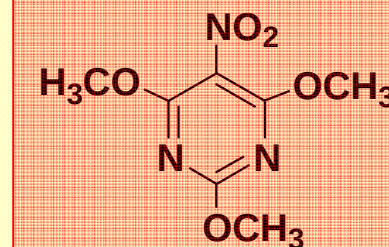
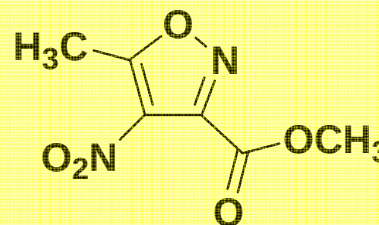
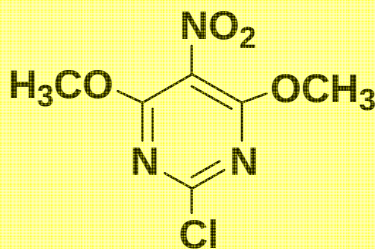
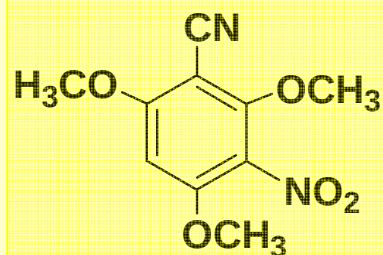


➤ $(\text{CH}_3)_4\text{NNO}_3$ BASED NITRONIUM TRIFLATE NITRATION:

▪ SIMPLE & CONVENIENT REACTION METHOD

- ✓ ONE-POT REACTION
- ✓ EASY AQUEOUS WORK-UP
- ✓ ISOMERIC SELECTIVITY
- ✓ EXCELLENT PRODUCT YIELDS
- ✓ HIGH (ANALYTICAL) "CRUDE" PRODUCT PURITY
- ✓ DIRECTLY SCALEABLE

▪ NEW COMPOUNDS & NEW SYNTHESIS ROUTE (DIRECT NITRATION)



Shackelford, S. A.; Anderson, M. B.; Christie, L. C.; Goetzen, T.; Guzman, M. C.; Hananel, M. A.; Kornreich, W. D.; Haitao, L.; Pathak, V. P.; Rabinovich, A. K.; Rajapakse, R. J.; Truesdale, L. K.; Tsank, S. M.; Vazir H. N.
J. Org. Chem., 2003, 68, 267-275.



MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



BIOTAGE MICROWAVE INITIATOR™



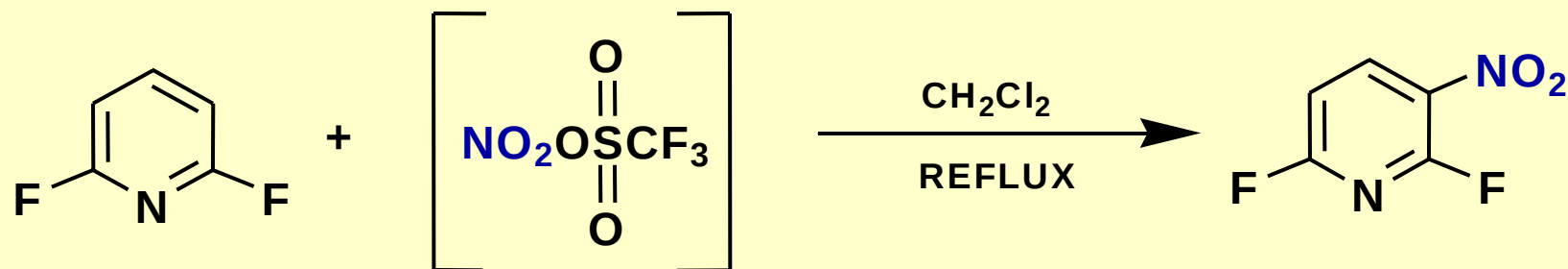
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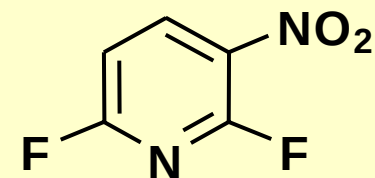
MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



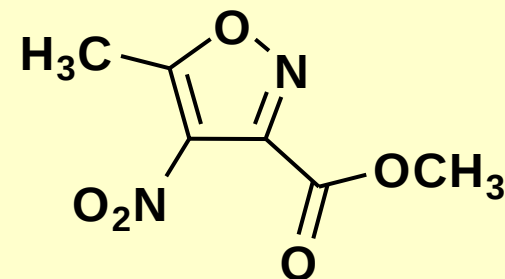
➤ MINIMUM NITRATION REACTION TIME:



<u>TEMP.</u>	<u>TIME (MIN)</u>	<u>PRODUCT CONVERSION</u>
<i>MW SET</i>	2	79%
= 80°C	3	98% (2 RUNS)
<i>ACTUAL</i>	4	100%
= 98-80°C		



<i>MW SET</i>	3	93%
= 60°C	5	100%
<i>ACTUAL</i>		
= 68-60°C		



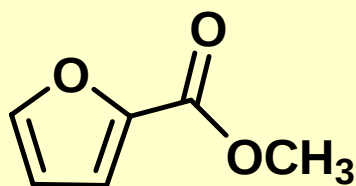
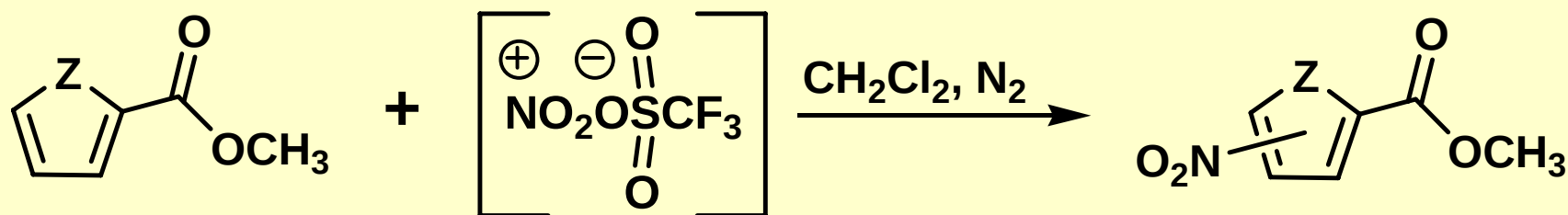


MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



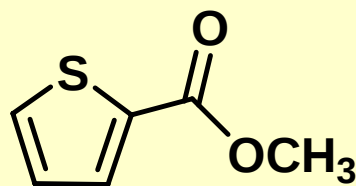
➤ 5-MEMBERED HETEROAROMATICS (ONE HETEROOATOM)

- DIFFERENT HETEROCYCLE REACTIVITY (Z = O, S, N)

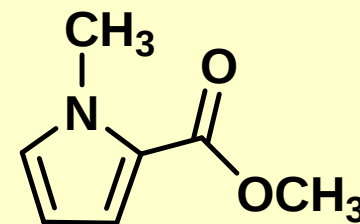


REACTIVE AT RT

NON-REACTIVE AT
-78°C



NON-REACTIVE AT
-78°C



VERY REACTIVE AT RT

VERY REACTIVE AT
-78 °C



MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



■ HETERO-RING DIRECTING EFFECTS

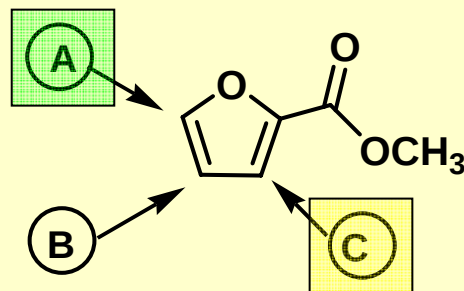
CONDITIONS

RT to 60°C
MICROWAVE
1.1 M (5 MIN)

-78°C to 60°C
MICROWAVE
0.7M (5 MIN)

-78°C to RT
CONVENTIONAL
0.7M (3 HR)

-78°C to RT
CONVENTIONAL
0.09M
(27-48 HRS)

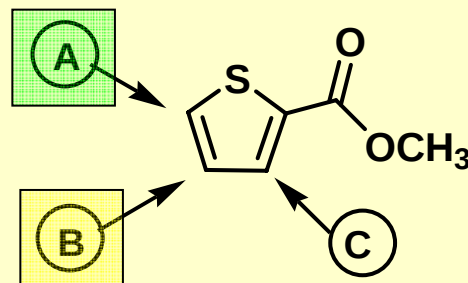


A	B	C
95%	0%	5%

94%	0%	6%*
*2% UNREACTED		

97%	0%	3%
-----	----	----

91-92%	0	9-8%
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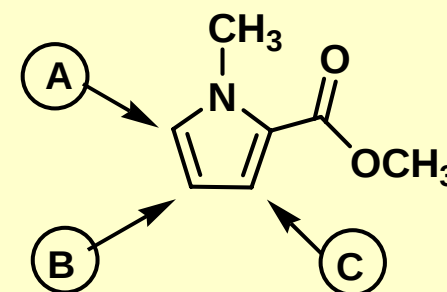


A	B	C
----	----	----

59%	36%	5%(?)
62%	38%	----

----	-----	----
------	-------	------

62%	38%	0%
-----	-----	----



**NO NITRATED
PRODUCT**

FURAN: SLIGHT
CONC. EFFECT

THIOPHENE:
SLIGHT HEAT
and./or CONC.
EFFECT

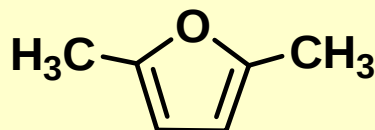
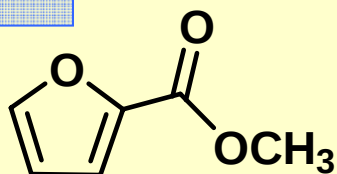


MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



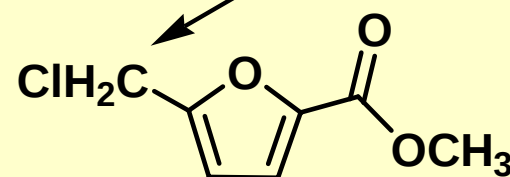
■ RING POSITIONAL REACTIVITY

SELECTIVE
-NO₂ POSITION

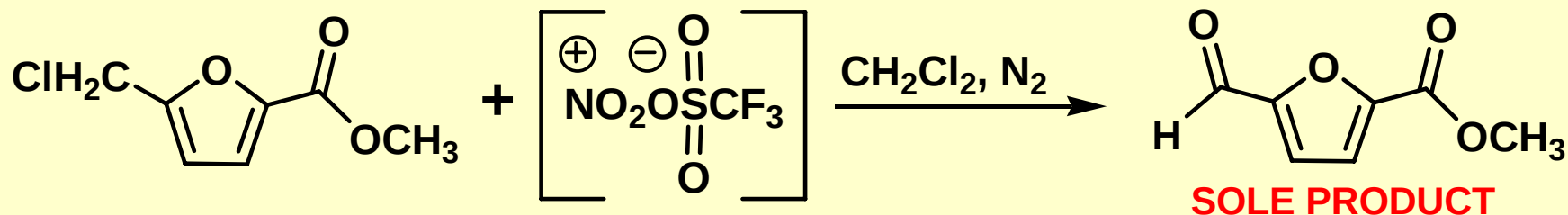


VERY REACTIVE AT
RT & -78°C, NO PRODUCT

BENZYLIC-LIKE
POSITION



NON-REACTIVE
AT -78°C



CONVENTIONAL [0.19M, -78°C to RT, 76 HR]:

71% CONVERSION
(NOT ISOLATED)

MICROWAVE [0.7M, -78°C to 70°C, 10 MIN]:

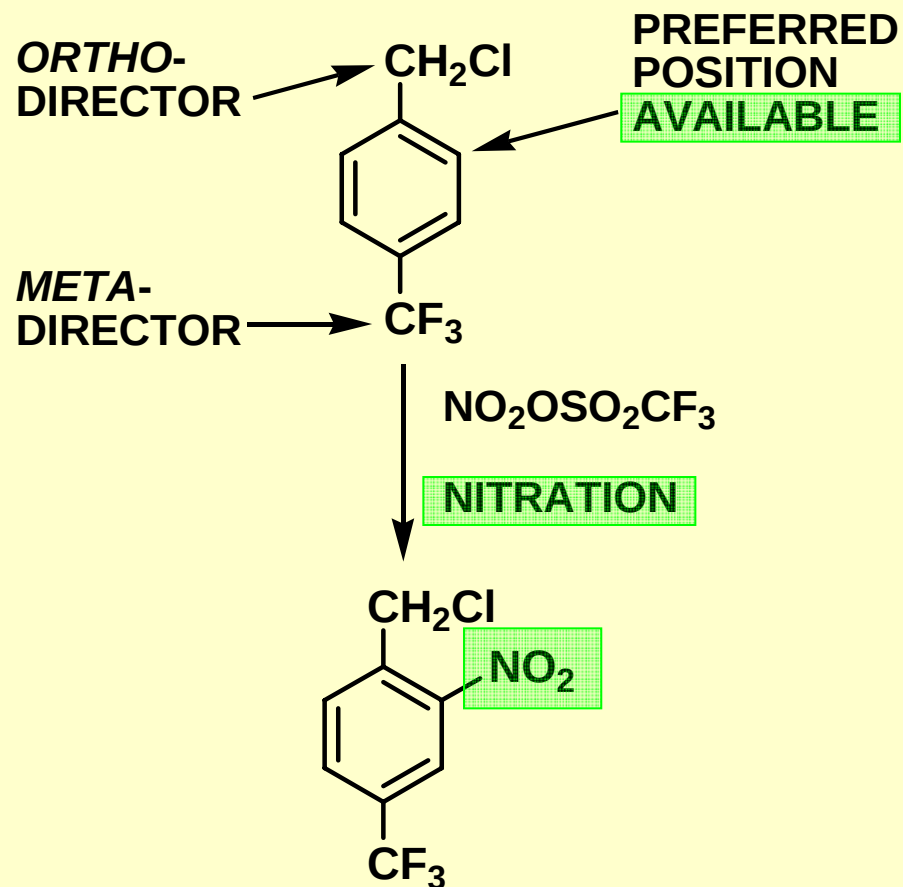
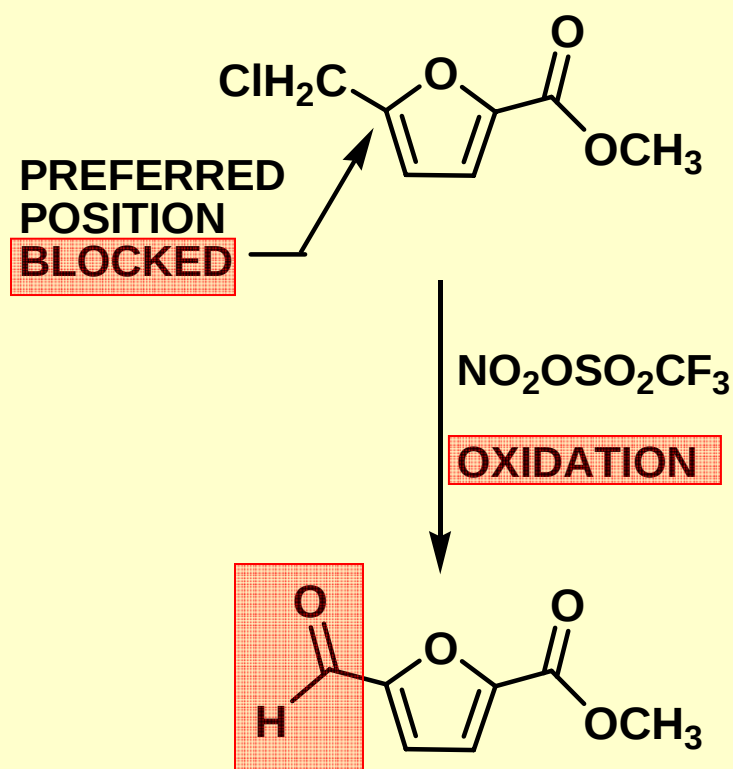
100% CONVERSION
64% YIELD (NMR PURE)
LT. YELLOW SOLID



MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



■ RING POSITION REACTIVITY



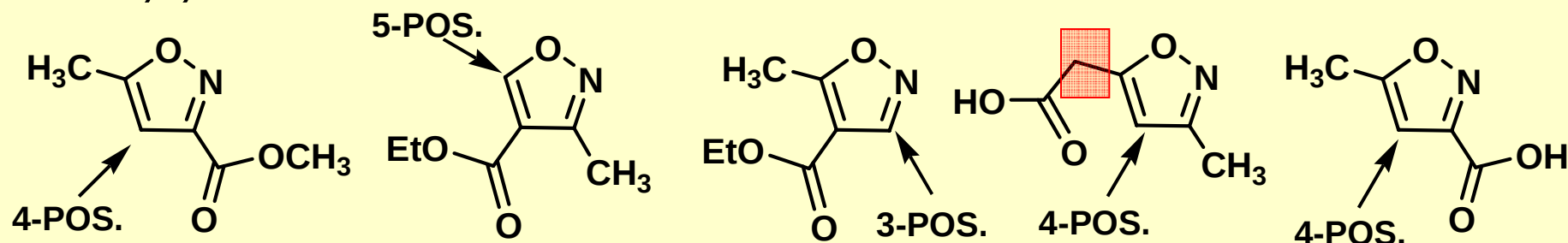


MICROWAVE NITRONIUM TRIFLATE SCREENING REACTIONS



➤ 5-MEMBERED HETEROAROMATICS (TWO HETEROATOMS)

▪ 3,4,5-POSITION REACTIVITY PREFERENCE



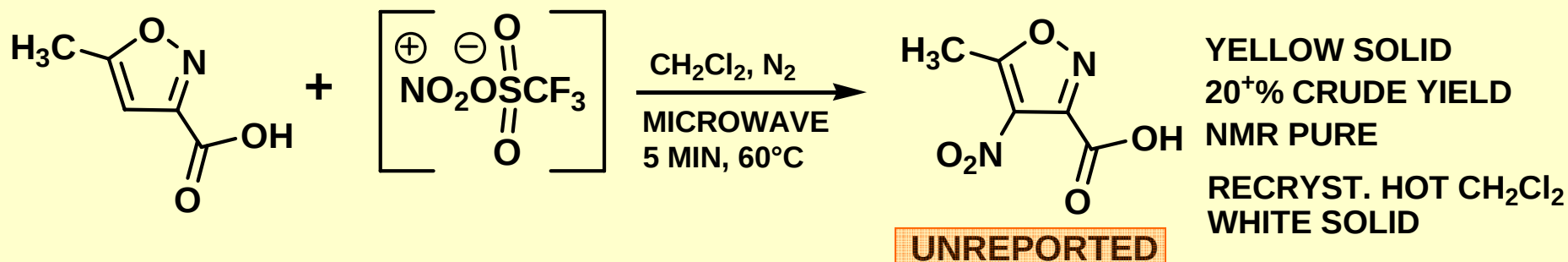
NON-REACTIVE
AT RT
HIGH YIELD
NITRATION
60°C, 5-15 MIN

NON-REACTIVE
AT RT
POSSIBLE TRACE
NITRATION,
60-100°C, 6-10 MIN

VERY REACTIVE
AT RT
NON-REACTIVE
AT -78°C
NO NITRATION

VERY REACTIVE
AT -78°C
NO NITRATION
OBSERVED

SEE
BELOW

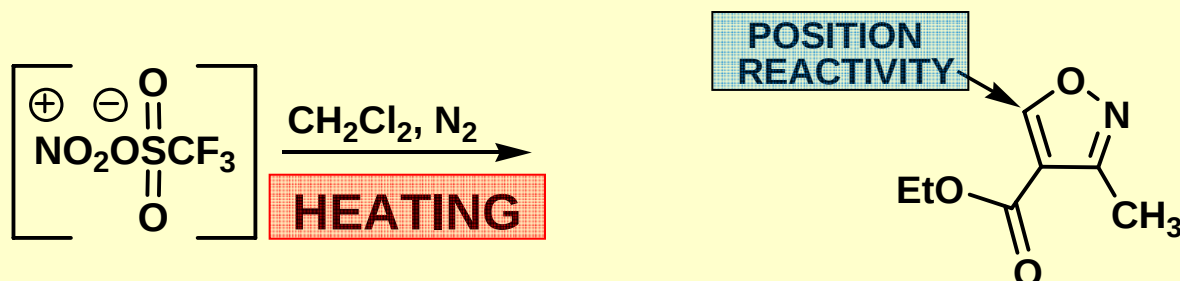




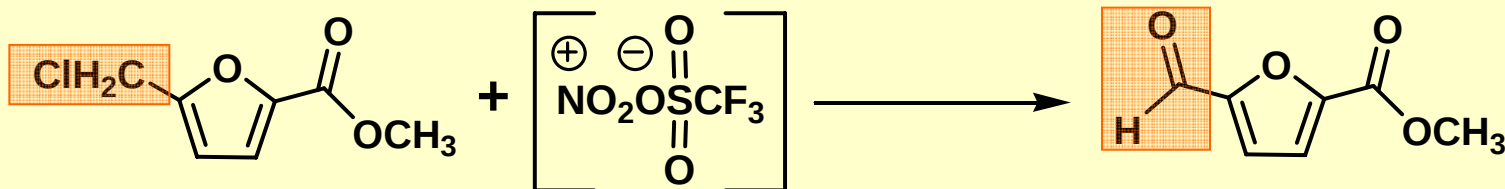
SUMMARY



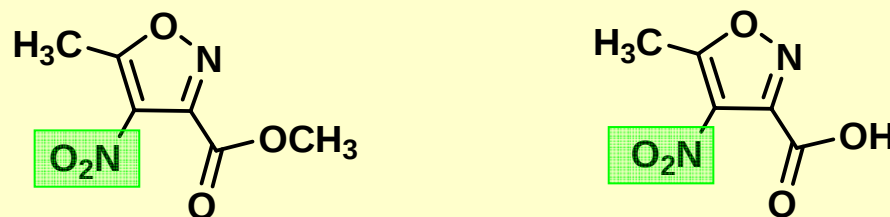
- **MICROWAVE-ASSISTED SYNTHESIS A USEFUL TOOL**
 - **SCREENING NITRONIUM TRIFLATE NITRATION SAFETY / REACTIVITY**



- **COMPLETING NEW NITRONIUM TRIFLATE REACTION PATHWAY**



- **OBTAINING NOVEL NITRO-HETEROAROMATICS w. NITRONIUM TRIFLATE**





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