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ECBC-TR-618

MASS SPECTRAL FRAGMENTATION OF VX

Dennis K. Rohrbaugh

RESEARCH AND TECHNOLOGY DIRECTORATE

May 2008

20080527153

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REPORT DOCUMENTATION PAGE				Form Approved OMB No. 0704-0188	
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1. REPORT DATE (DD-MM-YYYY) XX-05-2008		2. REPORT TYPE Final		3. DATES COVERED (From - To) Feb 2000	
4. TITLE AND SUBTITLE Mass Spectral Fragmentation of VX				5a. CONTRACT NUMBER	
				5b. GRANT NUMBER	
				5c. PROGRAM ELEMENT NUMBER	
6. AUTHOR(S) Rohrbaugh, Dennis K.				5d. PROJECT NUMBER None	
				5e. TASK NUMBER	
				5f. WORK UNIT NUMBER	
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) DIR, ECBC, ATTN: AMSRD-ECB-RT-CF, APG, MD 21010-5424				8. PERFORMING ORGANIZATION REPORT NUMBER ECBC-TR-618	
9. SPONSORING / MONITORING AGENCY NAME(S) AND ADDRESS(ES)				10. SPONSOR/MONITOR'S ACRONYM(S)	
				11. SPONSOR/MONITOR'S REPORT NUMBER(S)	
12. DISTRIBUTION / AVAILABILITY STATEMENT Approved for public release; distribution is unlimited.					
13. SUPPLEMENTARY NOTES					
14. ABSTRACT The objective of this study was to provide structural identification of VX fragment ions formed during mass spectrometric analysis, elucidation of fragmentation pathways, and a compilation of tandem mass spectral product ion spectra to aid in detection and confirmation of trace levels of VX in complex matrices. Fragmentation was observed and is reported here for both electron ionization and chemical ionization (methane) conditions. Isotopic labeling analysis using d ₅ -VX was performed to provide confirmation of selected structural assignments, particularly for the m/z 139 ion for which two structures have been proposed in the literature.					
15. SUBJECT TERMS Mass spectrometry Electron ionization Chemical ionization VX Fragmentation pathway Collision-induced dissociation					
16. SECURITY CLASSIFICATION OF:			17. LIMITATION OF ABSTRACT	18. NUMBER OF PAGES	19a. NAME OF RESPONSIBLE PERSON
a. REPORT	b. ABSTRACT	c. THIS PAGE			Sandra J. Johnson
U	U	U	UL	29	19b. TELEPHONE NUMBER (include area code) (410) 436-2914

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PREFACE

The work described in this report was started and completed in February 2000.

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MASS SPECTRAL FRAGMENTATION OF VX

1. INTRODUCTION

Detection and identification of chemical warfare (CW) agents is important for monitoring the future destruction of chemical stockpiles, environmental cleanup of past storage sites, and allegations of CW agent use as prohibited by the Chemical Weapons Convention.¹ Detection of O-ethyl S-[2-diisopropylamino)ethyl] methylphosphonothiolate (VX) at low levels (ppb) by electron ionization/mass spectrometry (EI/MS) is especially difficult because extensive fragmentation occurs. The application of EI and chemical ionization (CI) mass spectrometry for the analysis of VX has been previously reported.^{2,3} Tandem mass spectrometry (MS/MS) has been identified as a potential method that provides the sensitivity and selectivity via matrix elimination to make low level VX detection feasible and provides an additional verification method for treaty analysis.⁴ Bell and coworkers have reported the use of tandem mass spectrometry to elucidate VX fragmentation under electrospray ionization conditions.⁵ The objective of this study was to provide elucidation of fragmentation pathways and structural identification of VX fragment ions formed under EI and CI conditions, and to provide a compilation of MS/MS product ion spectra to aid in detection and confirmation of trace levels of VX in complex matrices in future efforts.

2. EXPERIMENTAL PROCEDURES

2.1 Materials

The Chemical Agent Standard Analytical Reference Material (CASARM) program provided the VX as lot VX-U-2128-CTF-N (vial 187). The gas chromatography/mass spectrometry (GC/MS) purity was 94.6%. The d₅-VX, containing deuterium atoms on the ethyl group, was synthesized in-house.

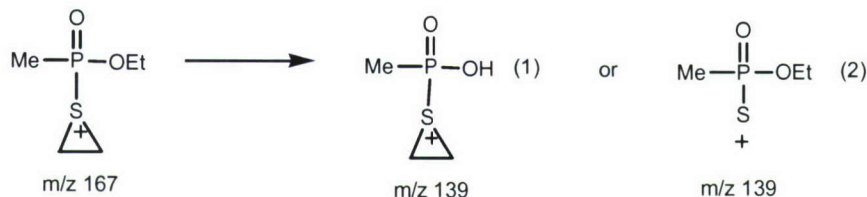
2.2 Instrumentation

Product ion spectra of VX and d₅-VX were obtained using a Finnigan (ThermoQuest, San Jose, CA) TSQ-7000 triple-stage quadrupole mass spectrometer equipped with a 30m x 0.25mm DB-5 column (J&W Scientific, Folsom, CA). Helium carrier gas flow was 1 cc/min, injection temperature 250 °C, interface temperature 250 °C, manifold temperature 70 °C, source temperature 150 °C, and oven temperature 60-270 °C at 15 °C/min with a 5-min hold at 270 °C. The electron energies and emission currents were 70 eV and 400 µA for EI and 200 eV and 300 µA for CI, respectively. Methane was used as the CI reagent gas at a pressure of 3500 mTorr. All collision-induced dissociation (CID) MS/MS experiments were performed using Argon as the collision gas with a collision pressure of 1.5 mT and collision energy of 15V.

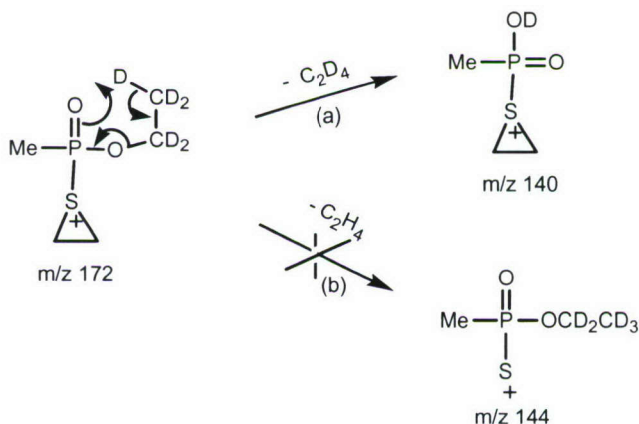
3. RESULTS AND DISCUSSION

The EI and methane CI mass spectra for VX and d₅-VX are provided in Figures 1 and 2. Proposed structures for each ion based on MS/MS fragmentation data are listed in the Table. An overall scheme of proposed VX fragmentation pathways consistent with MS/MS data is provided in the Scheme. Although listed as originating from a neutral VX molecule, fragmentation can occur either from neutral VX (m/z 267) or protonated VX (m/z 268), depending upon whether electron or chemical ionization is used. Tandem MS product ion spectra for each ion are provided in Figures 3-22. Spectra are labeled to identify the mode of analysis (EI and CI). In cases where the same mass ion was subjected to CID analysis under both EI and CI conditions, similar product ion spectra were obtained.

Isotopic labeling using VX labeled with deuterium on the ethyl group was used to conclusively identify the ion at m/z 139. The full scan EI mass spectrum of VX-d₅ (Figure 1) shows that the m/z 167 ion is shifted to m/z 172, indicating the d₅-ethyl group is still intact and providing support for the proposed structure of this fragment. The m/z 137 ion can potentially be formed from the m/z 167 ion by two pathways, via elimination of ethylene from either the O-ethyl group or from the sulfonium ion group.

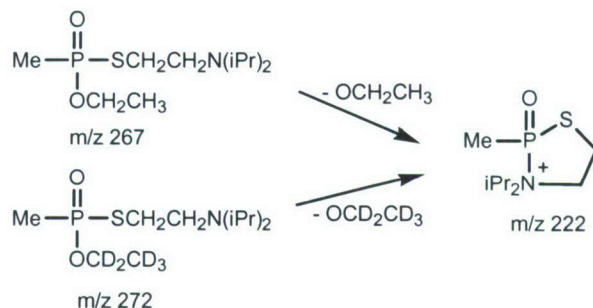


The most logical route is the formation of fragment 1 due to the expected sulfonium ion stabilization of the positive charge. This route is indeed confirmed, as the EI spectrum of the labeled compound shows the m/z 139 ion is shifted to m/z 140 as expected from mechanism a and not m/z 144 as would be expected from mechanism b shown below. The observed mechanism of formation using VX-d₅ is shown as follows:

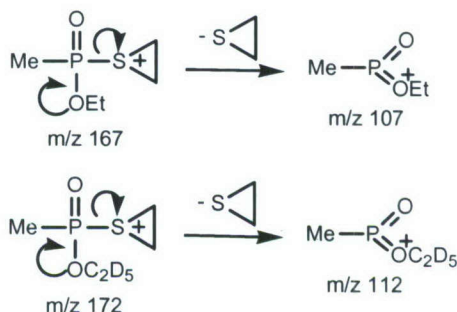


Comparison of the EI and CI Spectra of d₅-VX and unlabelled VX also provides support for other proposed fragment identifications. The data indicates the ethyl group is intact

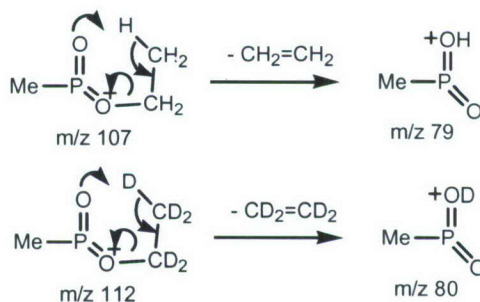
in the fragment ions at m/z 252 and 224 (m/z 257 and 229 in the d_5 -VX spectra). The ion at m/z 222 appears in both unlabeled and labeled spectra, consistent with the proposed formation from loss of the O-ethyl group as follows:



The ion at m/z 107 in the unlabeled EI spectrum is shifted to m/z 112 in the labeled EI spectrum, consistent with the following pathway and structure assignments:



The observation that the ion observed at m/z 79 in the unlabeled EI spectrum is shifted to m/z 80 in the labeled spectrum is consistent with the following mechanism:



In addition, it was observed that m/z 97 was formed as a product ion during CID fragmentation of the m/z 79 ion. As proposed by Bell and coworkers,⁵ formation of this ion can logically be explained as reaction of the m/z 79 ion with trace levels of water:

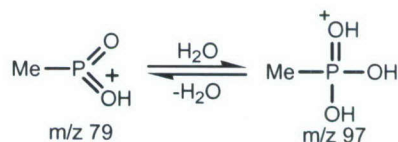


Table. Proposed Structures of VX Mass Spectral Fragmentation Ions

M/z	Proposed Structure
268	$\begin{array}{c} \text{O} \\ \parallel \\ \text{Me}-\text{P}-\text{SCH}_2\text{CH}_2\text{N}^+\text{HiPr}_2 \\ \\ \text{OEt} \end{array}$
252	$\begin{array}{c} \text{O} \\ \parallel \\ \text{Me}-\text{P}-\text{SCH}_2\text{CH}_2\text{N}^+\text{iPr} \\ \quad \diagdown \\ \text{OEt} \quad \text{CH-CH}_3 \end{array}$
224	$\begin{array}{c} \text{O} \\ \parallel \\ \text{Me}-\text{P}-\text{SCH}_2\text{CH}_2\text{N}^+\text{iPr} \\ \\ \text{OEt} \end{array} \rightleftharpoons \begin{array}{c} \text{O} \\ \parallel \\ \text{Me}-\text{P}-\text{SCH}_2\text{CH}_2\text{NH}^+=\text{C}(\text{CH}_3)_2 \\ \\ \text{OEt} \end{array}$
222	$\begin{array}{c} \text{O} \quad \text{S} \\ \parallel \quad \diagup \\ \text{Me}-\text{P} \quad \text{N}^+\text{iPr}_2 \\ \quad \diagdown \\ \text{iPr}_2\text{N}^+ \quad \text{CH}_2\text{CH}_2 \end{array} \rightleftharpoons \begin{array}{c} \text{O} \\ \parallel \\ \text{Me}-\text{P}^+=\text{SCH}_2\text{CH}_2\text{NiPr}_2 \end{array}$
210	$\begin{array}{c} \text{O} \\ \parallel \\ \text{Me}-\text{P}-\text{SCH}_2\text{CH}_2\text{NH}^+=\text{CHCH}_3 \\ \\ \text{OEt} \end{array}$
180	$\begin{array}{c} \text{O} \quad \text{S} \\ \parallel \quad \diagup \\ \text{Me}-\text{P} \quad \text{N}^+\text{iPr} \\ \quad \diagdown \\ \text{iPr}-\text{HN}^+ \quad \text{CH}_2\text{CH}_2 \end{array} \rightleftharpoons \begin{array}{c} \text{O} \\ \parallel \\ \text{Me}-\text{P}^+=\text{SCH}_2\text{CH}_2\text{NHiPr} \end{array}$
167	$\begin{array}{c} \text{O} \\ \parallel \\ \text{Me}-\text{P}-\text{S}^+ \\ \\ \text{OEt} \end{array} \rightleftharpoons \begin{array}{c} \text{O} \\ \parallel \\ \text{Me}-\text{P}-\text{S}-\text{CH}_2\text{CH}_2^+ \\ \\ \text{OEt} \end{array}$
160	$\text{iPr}_2\text{N}^+\text{HC}_2\text{H}_3\text{S}$
144	$\begin{array}{c} \text{iPr} \\ \diagup \\ \text{CH}_3-\text{CH}=\text{N}^+\text{C}_2\text{H}_3\text{S} \end{array}$
139	$\begin{array}{c} \text{O} \\ \parallel \\ \text{Me}-\text{P}-\text{S}^+ \\ \\ \text{OH} \end{array} \rightleftharpoons \begin{array}{c} \text{O} \\ \parallel \\ \text{Me}-\text{P}-\text{SCH}_2\text{CH}_2^+ \\ \\ \text{OH} \end{array}$
138	$\begin{array}{c} \text{O} \quad \text{S} \\ \parallel \quad \diagup \\ \text{Me}-\text{P} \quad \text{N}^+\text{H}_2 \\ \quad \diagdown \\ \text{NH}_2 \quad \text{CH}_2\text{CH}_2 \end{array} \rightleftharpoons \begin{array}{c} \text{O} \\ \parallel \\ \text{Me}-\text{P}^+=\text{S}-\text{CH}_2\text{CH}_2\text{NH}_2 \end{array}$
128	$\text{iPr}_2\text{N}^+ \rightleftharpoons \text{iPr}_2\text{NCH}_2\text{CH}_2^+$
127	$[\text{CH}_2=\text{CHN}(\text{iPr})_2]^+$

Table. Proposed Structures of VX Mass Spectral Fragmentation Ions (Continued)

M/z	Proposed Structure
118	$\text{iPrNH}_2\text{C}_2\text{H}_3\text{S}^+$
114	$\text{H}_2\text{C}=\text{N}^+\begin{smallmatrix} \text{iPr} \\ \text{iPr} \end{smallmatrix} \longleftrightarrow \text{H}_2\text{C}^+-\text{N}\begin{smallmatrix} \text{iPr} \\ \text{iPr} \end{smallmatrix}$
112	$\text{H}_2\text{C}=\text{CH}-\text{N}^+\begin{smallmatrix} \text{iPr} \\ \text{CHCH}_3 \end{smallmatrix}$
107	$\text{Me}-\text{P}^+\begin{smallmatrix} \text{O} \\ \text{OEt} \end{smallmatrix} \longleftrightarrow \text{Me}-\text{P}^+\begin{smallmatrix} \text{O} \\ \text{OEt} \end{smallmatrix}$
102	$\text{CH}_3-\text{CH}=\text{NHC}_2\text{H}_3\text{S}^+$
100	$\text{iPr}_2\text{N}^+ \rightleftharpoons \begin{smallmatrix} \text{H}_3\text{C} & \text{iPr} \\ & \text{C}=\text{NH}^+ \\ \text{H}_3\text{C} & \end{smallmatrix} \rightleftharpoons \begin{smallmatrix} \text{H}_3\text{C} & \text{iPr} \\ & \text{C}=\text{NH}_2^+ \\ \text{H}_2\text{C} & \end{smallmatrix}$
98	$[\text{C}_6\text{H}_{12}\text{N}]^+$
95	$\text{Me}-\text{P}^+\begin{smallmatrix} \text{OH} \\ \text{S} \end{smallmatrix} \rightleftharpoons \text{Me}-\text{P}^+\begin{smallmatrix} \text{O} \\ \text{SH} \end{smallmatrix}$
86	$\text{iPrNH}^+ \triangle \longleftrightarrow \text{iPrNHCH}_2\text{CH}_2^+$
84	$\text{CH}_2=\text{CHN}^+\text{iPr} \rightleftharpoons \text{CH}_2=\text{CHNH}=\text{C}(\text{CH}_3)_2^+ \rightleftharpoons \text{CH}_2=\text{CHNH}_2^+ \begin{smallmatrix} \text{CH}_2 \\ \text{CH}_3 \end{smallmatrix}$
83	$[\text{C}_5\text{H}_9\text{N}]^+$
79	$\text{Me}-\text{P}^+\begin{smallmatrix} \text{O} \\ \text{OH} \end{smallmatrix} \longleftrightarrow \text{Me}-\text{P}^+\begin{smallmatrix} \text{O} \\ \text{OH} \end{smallmatrix}$

Table. Proposed Structures of VX Mass Spectral Fragmentation Ions (Continued)

M/z	Proposed Structure
72	$\text{H}_2\text{C}=\text{NH}^+-\text{iPr}$
70	$\text{H}_2\text{C}=\text{CH}-\text{N}^+=\text{CHCH}_3$
69	$[\text{C}_4\text{H}_7\text{N}]^+$
61	$\text{HS}^+ \longleftrightarrow {}^+\text{CH}_2\text{CH}_2\text{SH}$
58	$\text{iPrNH}^+ \rightleftharpoons \begin{array}{c} \text{H}_3\text{C} \\ \diagup \\ \text{C}=\text{NH}_2^+ \\ \diagdown \\ \text{H}_3\text{C} \end{array} \rightleftharpoons \begin{array}{c} \text{H}_2\text{C} \\ \diagup \\ \text{C}=\text{NH}_3^+ \\ \diagdown \\ \text{H}_3\text{C} \end{array}$
56	$[\text{C}_3\text{H}_6\text{N}]^+$
55	$[\text{C}_3\text{H}_5\text{N}]^+$
47	${}^+\text{P}=\text{O} \longleftrightarrow \text{P}\equiv\text{O}^+$
44	$\text{H}_2\text{N}^+ \longleftrightarrow \text{H}_2\text{NCH}_2\text{CH}_2^+$
43	$\begin{array}{c} \text{CH}_3 \\ \diagup \\ \text{HC}^+ \\ \diagdown \\ \text{CH}_3 \end{array}$
42	${}^+\text{CH}_2-\text{CH}=\text{NH} \longleftrightarrow \text{CH}_2=\text{CH}-\text{NH}^+$
30	$\text{H}_2\text{C}=\text{NH}_2^+$
28	$\text{HC}\equiv\text{NH}^+$

4. CONCLUSIONS

Tandem MS product ion spectra are provided for 20 VX fragment ions. Based on these spectra, proposed ion structures and pathways for VX fragmentation are provided. Spectra obtained for d5-VX provide support for the identifications and in particular provide conclusive evidence that the m/z 139 ion formed is the cyclic sulfonium ion resulting from loss of ethylene from the O-ethyl moiety rather than loss from the sulfonium ion moiety. The MS/MS product ion spectra are provided here as tools for the detection of VX trace levels in environmental samples and other complex matrices.

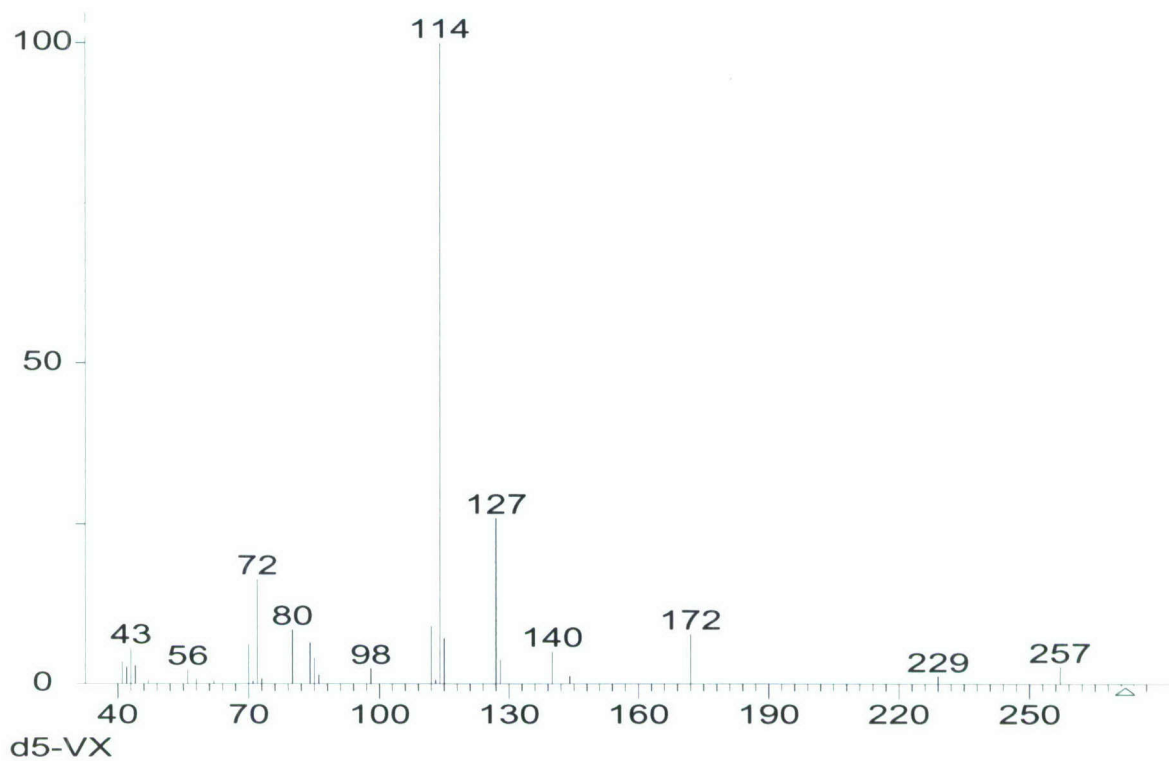
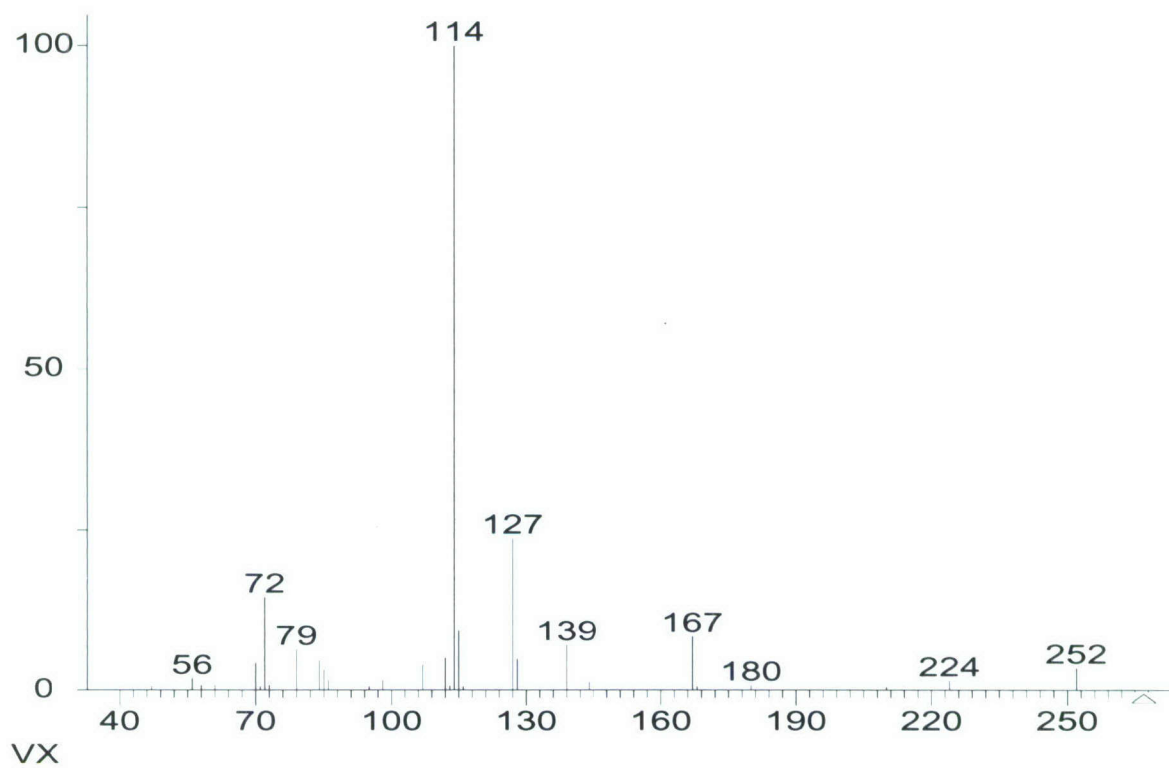


Figure 1. EI Mass Spectra of VX and d₅-VX

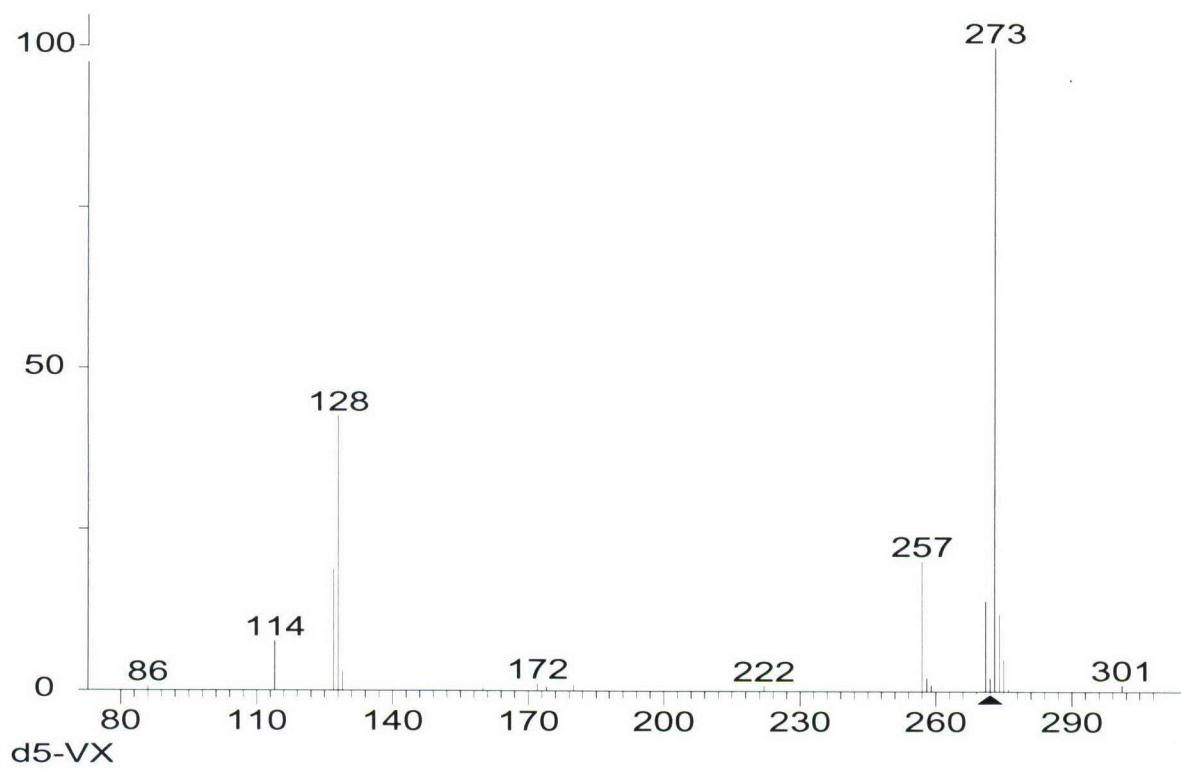
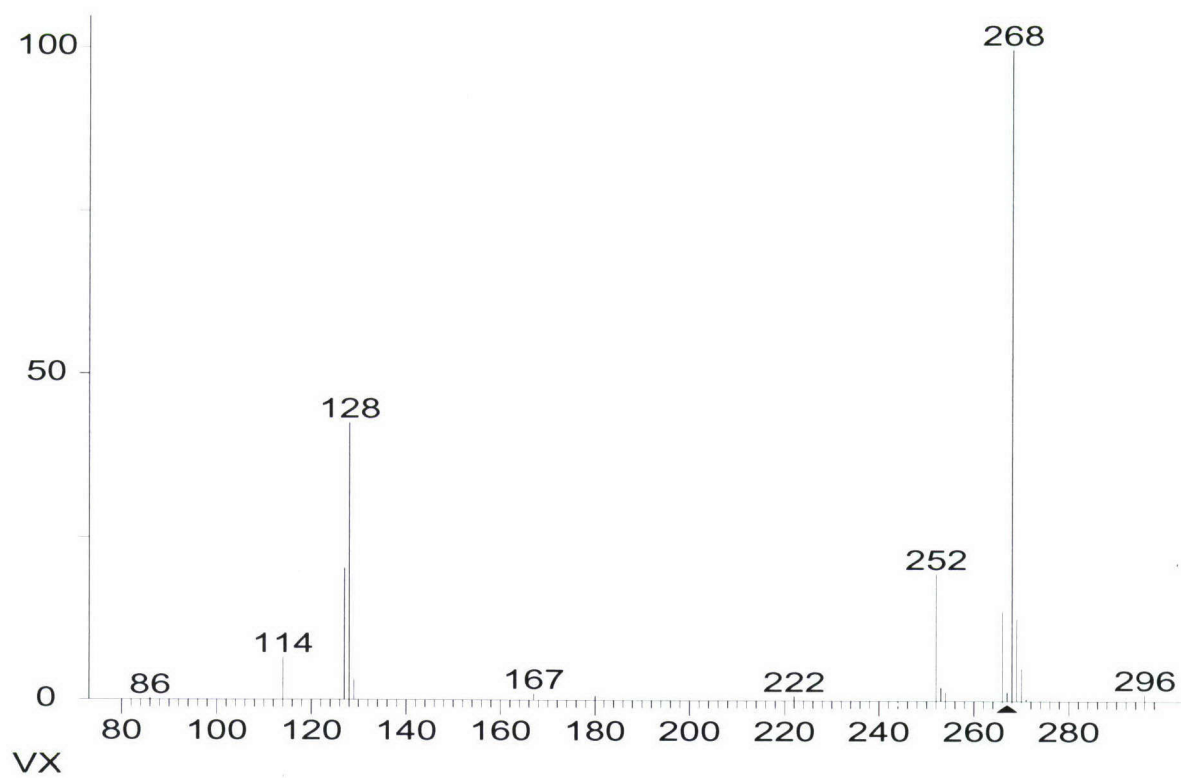


Figure 2. Methane CI Mass Spectra of VX and d₅-VX

v268ci #248-249 RT: 2.96-2.97 AV: 2 NL: 1.81E6
T: + c EISRM ms2 268.00@ -15.00 [19.96-300.00]

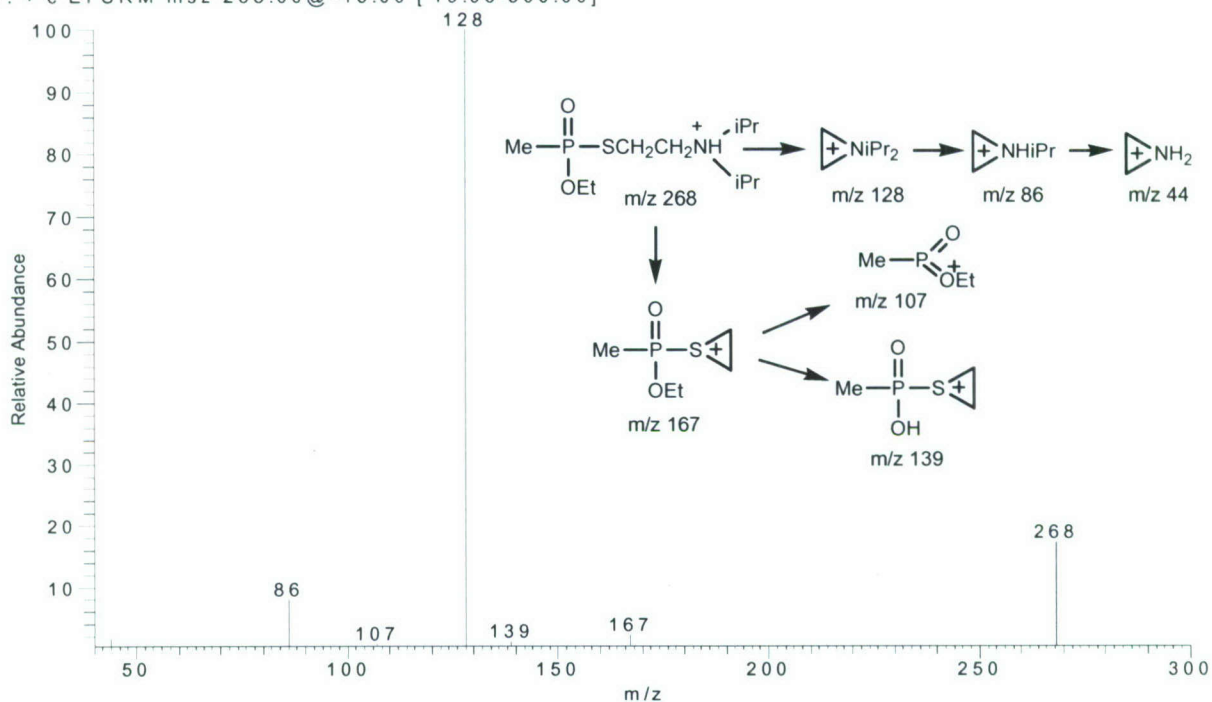


Figure 3. Product Ion Mass Spectrum of VX m/z 268 (CI)

v252c #251-252 RT: 3.01-3.02 AV: 2 NL: 2.21E5
T: + c EISRM ms2 252.00@ -15.00 [44.99-300.00]

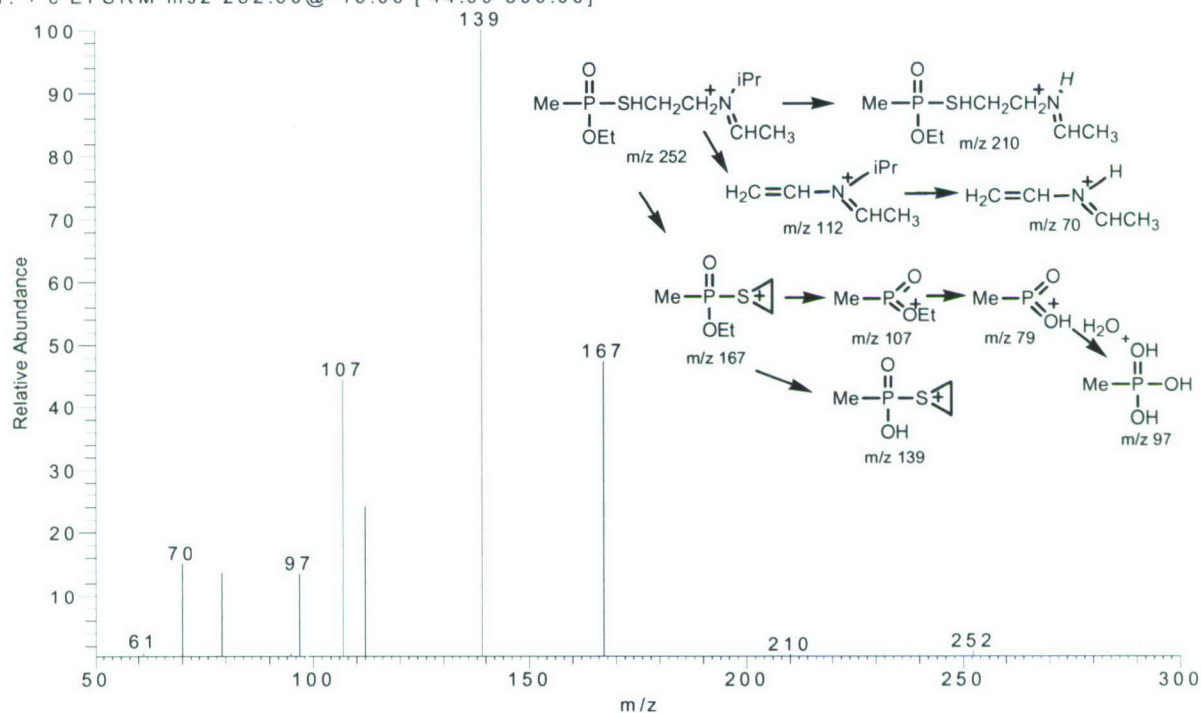


Figure 4. Product Ion Mass Spectrum of VX m/z 252 (EI)

v224 #248-250 RT: 2.97-2.99 AV: 3 NL: 3.70E4
T: + c EI SRM ms2 224.00@ -15.00 [44.99-300.00]

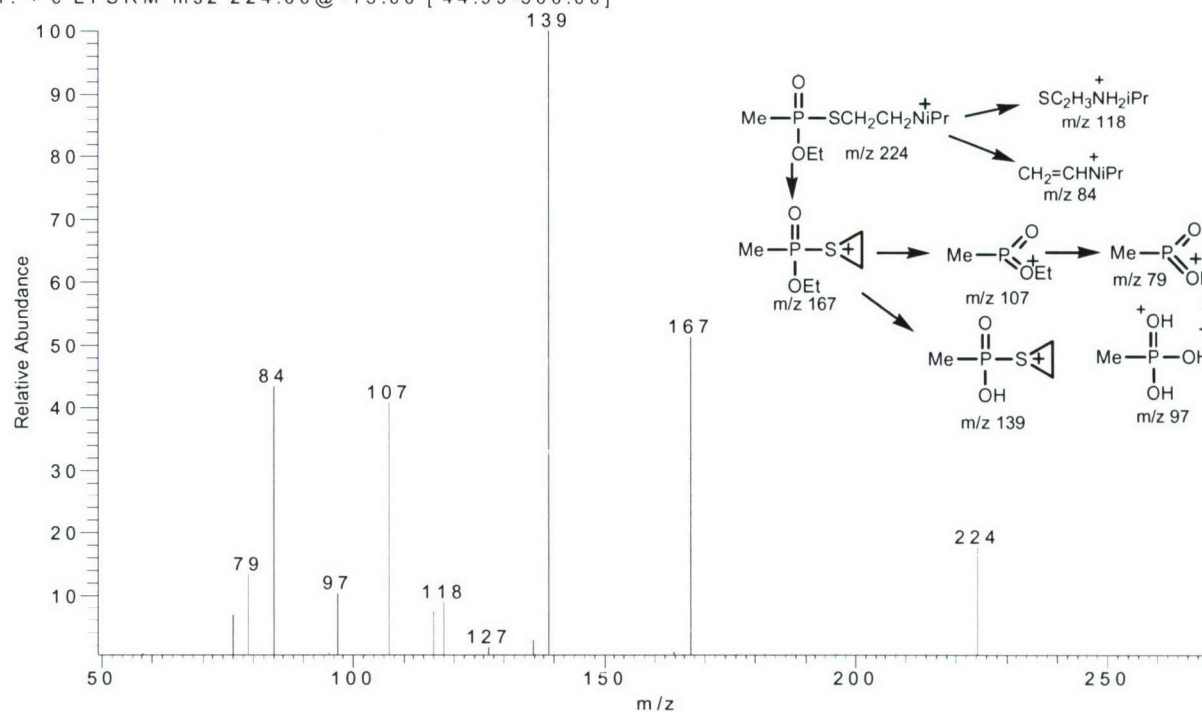


Figure 5. Product Ion Mass Spectrum of VX m/z 224 (EI)

v222ci #249-250 RT: 2.98-2.99 AV: 2 NL: 4.07E4
T: + c EI SRM ms2 222.00@ -15.00 [19.96-300.00]

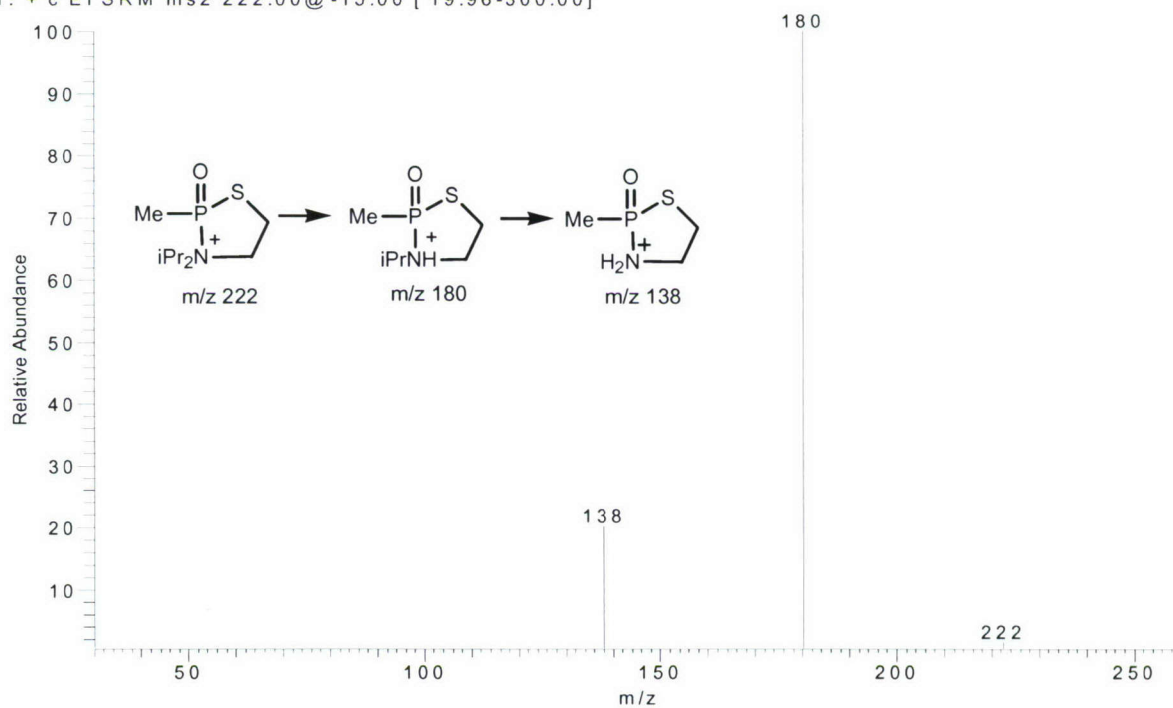


Figure 6. Product Ion Mass Spectrum of VX m/z 222 (CI)

v180ci #249-251 RT: 2.98-3.00 AV: 3 NL: 1.65E4
T: + c EISRM ms2 180.00@ -15.00 [19.96-300.00]

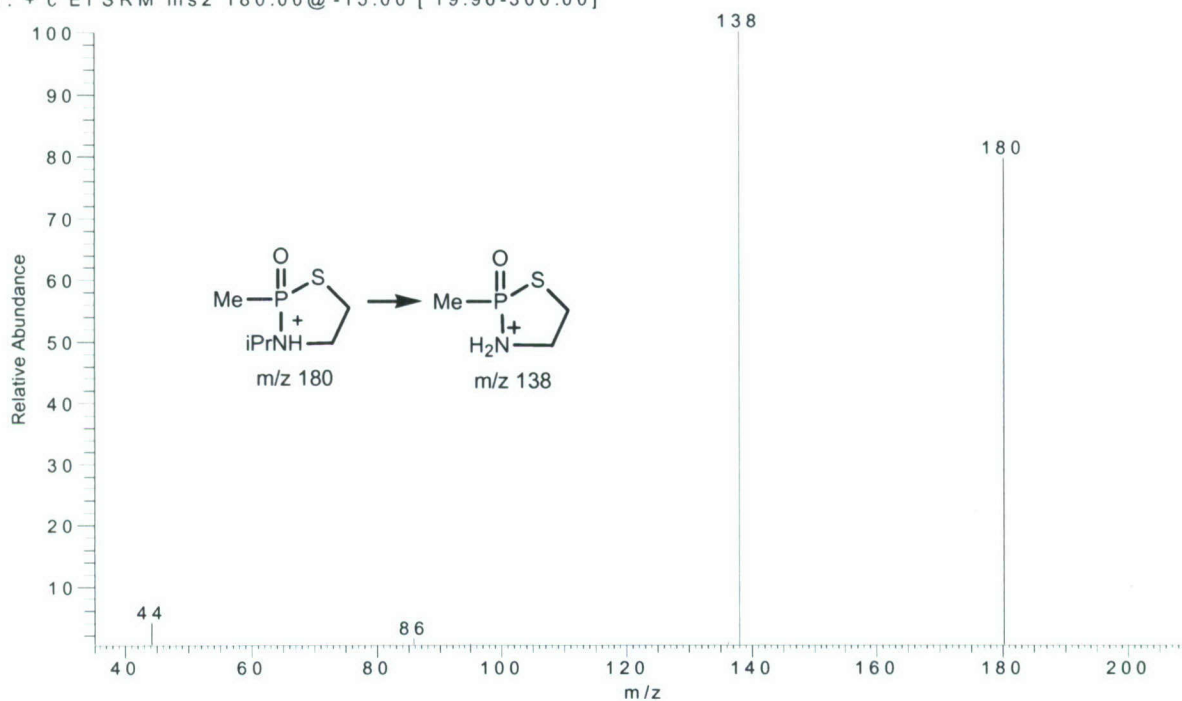


Figure 7. Product Ion Mass Spectrum of VX m/z 180 (EI)

v167 #176-178 RT: 2.12-2.14 AV: 3 NL: 3.96E5
T: + c EISRM ms2 167.00@ -15.00 [44.99-300.00]

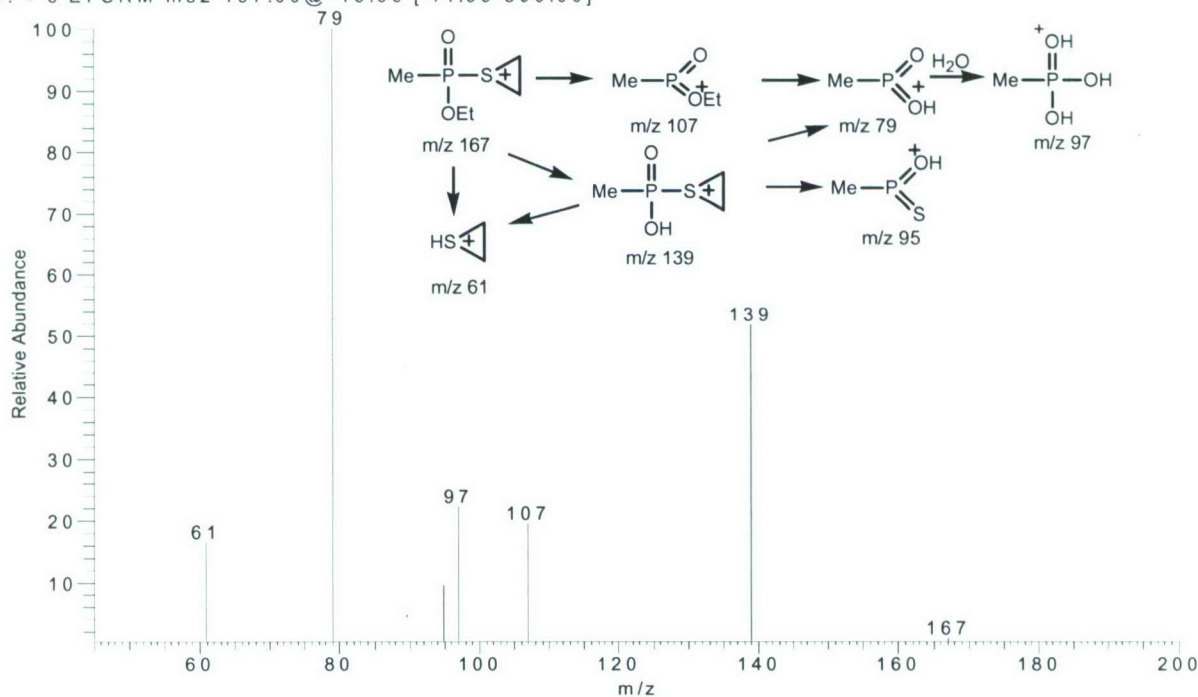


Figure 8. Product Ion Mass Spectrum of VX m/z 167 (EI)

v160ci #250-251 RT: 2.99-3.00 AV: 2 NL: 2.66E3
T: + c EISRM ms2 160.00@ -15.00 [19.96-300.00]

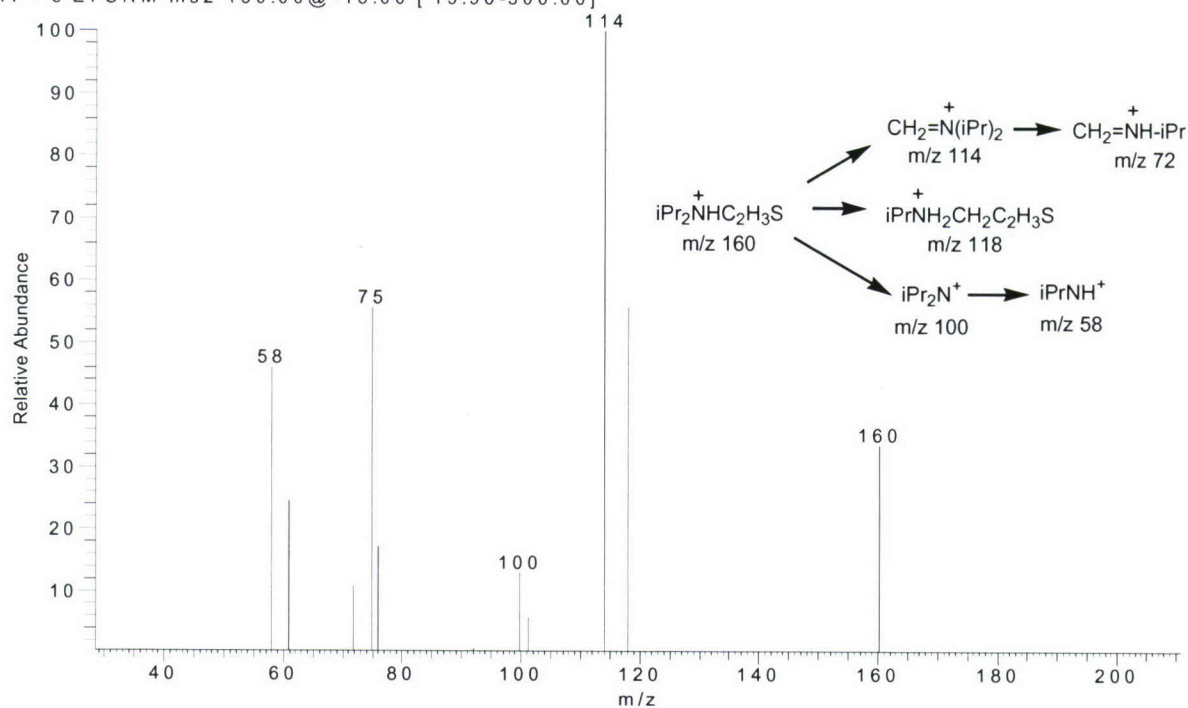


Figure 9. Product Ion Mass Spectrum of VX m/z 160 (CI)

v144 #249-252 RT: 2.98-3.01 AV: 4 NL: 5.21E4
T: + c EISRM ms2 144.00@ -15.00 [44.99-300.00]

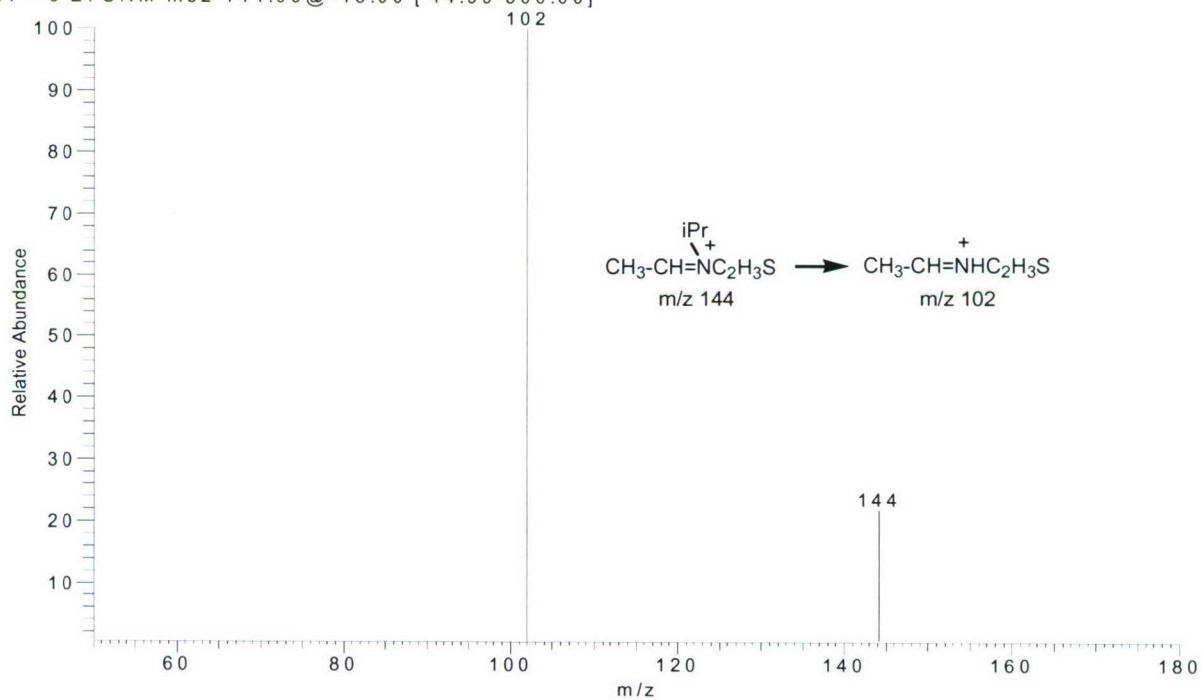


Figure 10. Product Ion Mass Spectrum of VX m/z 144 (EI)

v139 #251-253 RT: 3.00-3.02 AV: 3 NL: 2.37E5
T: + c EI SRM ms2 139.00@ -15.00 [44.99-300.00]

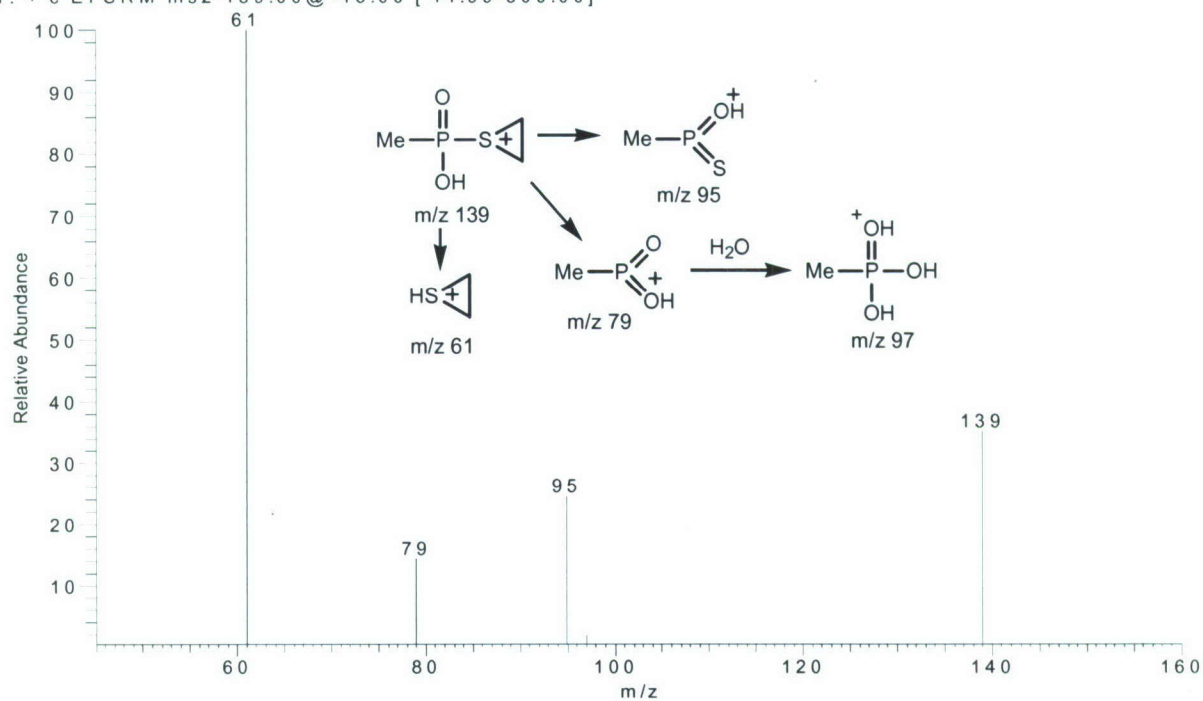


Figure 11. Product Ion Mass Spectrum of VX m/z 139 (EI)

v128ci #249-251 RT: 2.97-3.00 AV: 3 NL: 1.54E5
T: + c EI SRM ms2 128.00@ -15.00 [19.96-300.00]

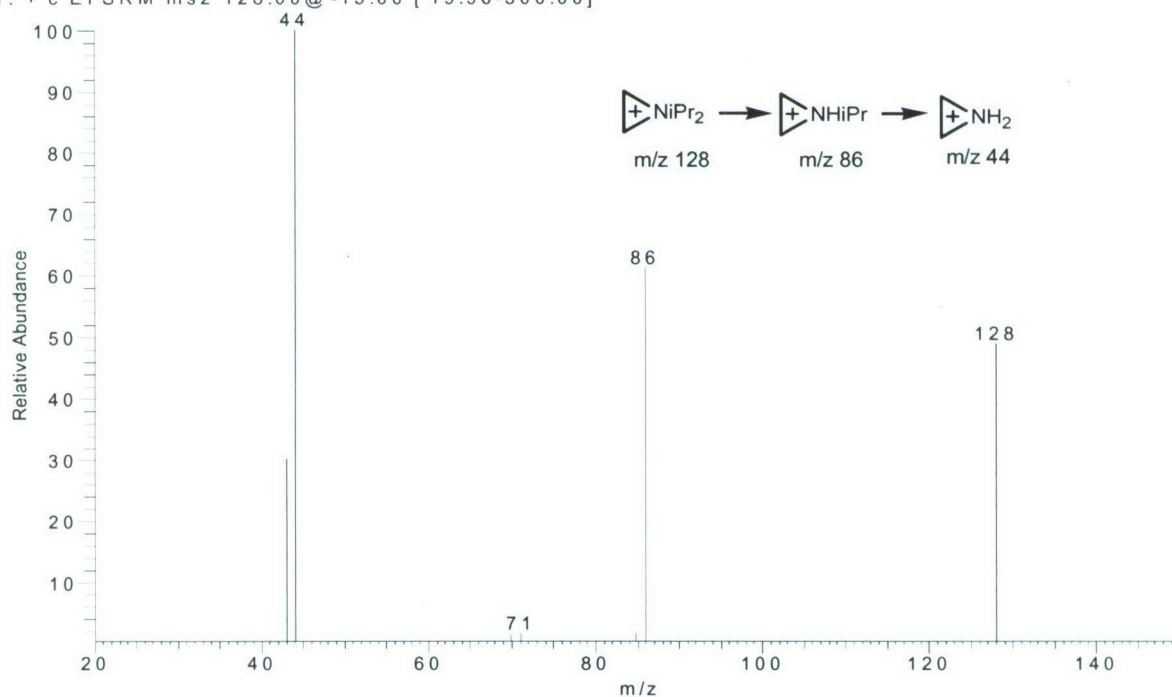


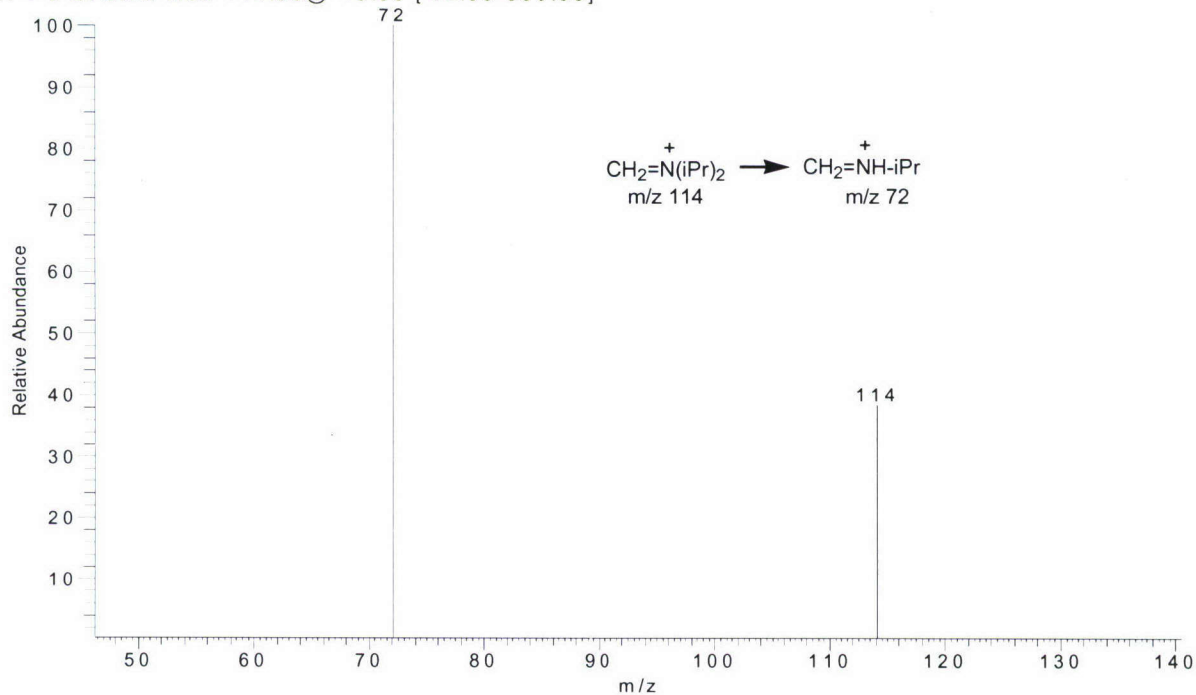
Figure 12. Product Ion Mass Spectrum of VX m/z 128 (CI)

The mass spectrum shows relative abundance versus m/z from 50 to 150. The base peak is at m/z 84. Other significant peaks are at m/z 70, 56, 112, and 127.

The fragmentation scheme illustrates the following pathways:

- The precursor ion $[CH_2=CHN(iPr)_2]^{++}$ at m/z 127 fragments to form the m/z 98 ion: $H_3C-C \equiv N^+=C(CH_3)_2$.
- The precursor ion at m/z 127 also fragments to form the m/z 84 ion: $CH_2=CHN-iPr$.
- The m/z 84 ion fragments to form the m/z 112 ion: $H_2C=CH-N^+(iPr)=CHCH_3$.
- The m/z 112 ion fragments to form the m/z 70 ion: $H_2C=CH-N^+=CHCH_3$.

v114 #249-251 RT: 2.97-2.99 AV: 3 NL: 6.98E5
T: + c EI SRM ms2 114.00@ -15.00 [44.99-300.00]



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v112 #250-253 RT: 2.99-3.02 AV: 4 NL: 1.51E5
T: + c EI SRM ms2 112.00@ -15.00 [44.99-300.00]

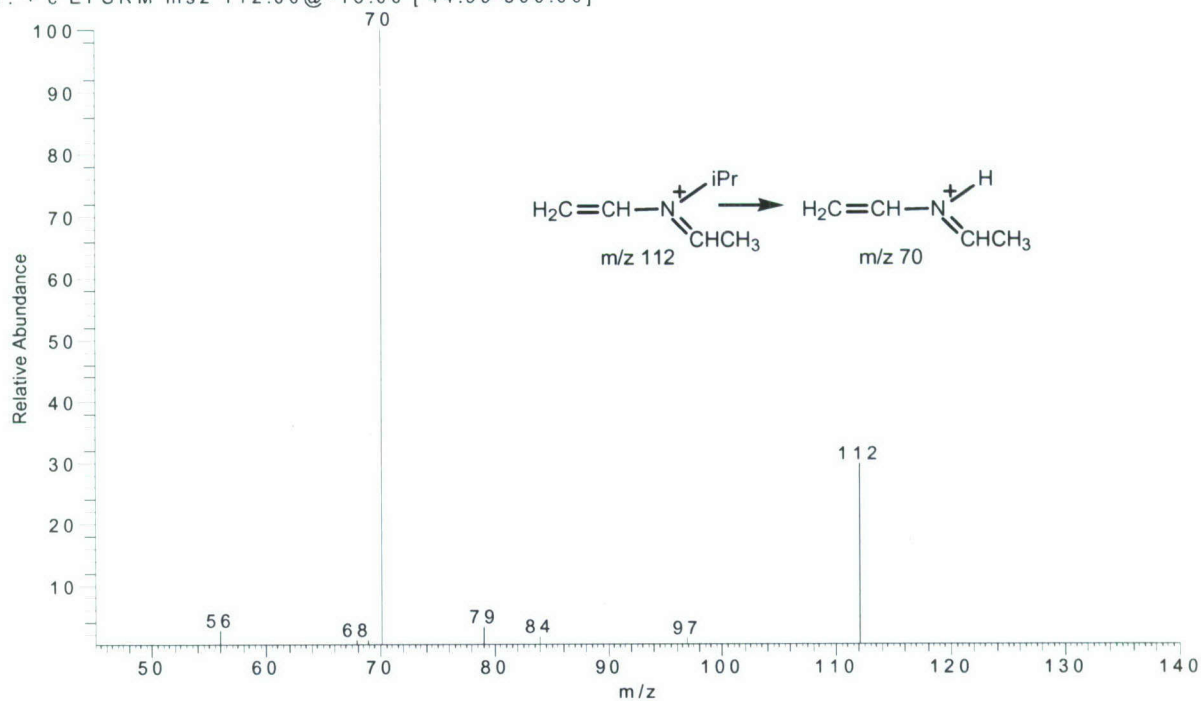


Figure 15. Product Ion Mass Spectrum of VX m/z 112 (EI)

v107 #249-252 RT: 2.97-3.01 AV: 4 NL: 1.57E5
T: + c EI SRM ms2 107.00@ -15.00 [44.99-300.00]

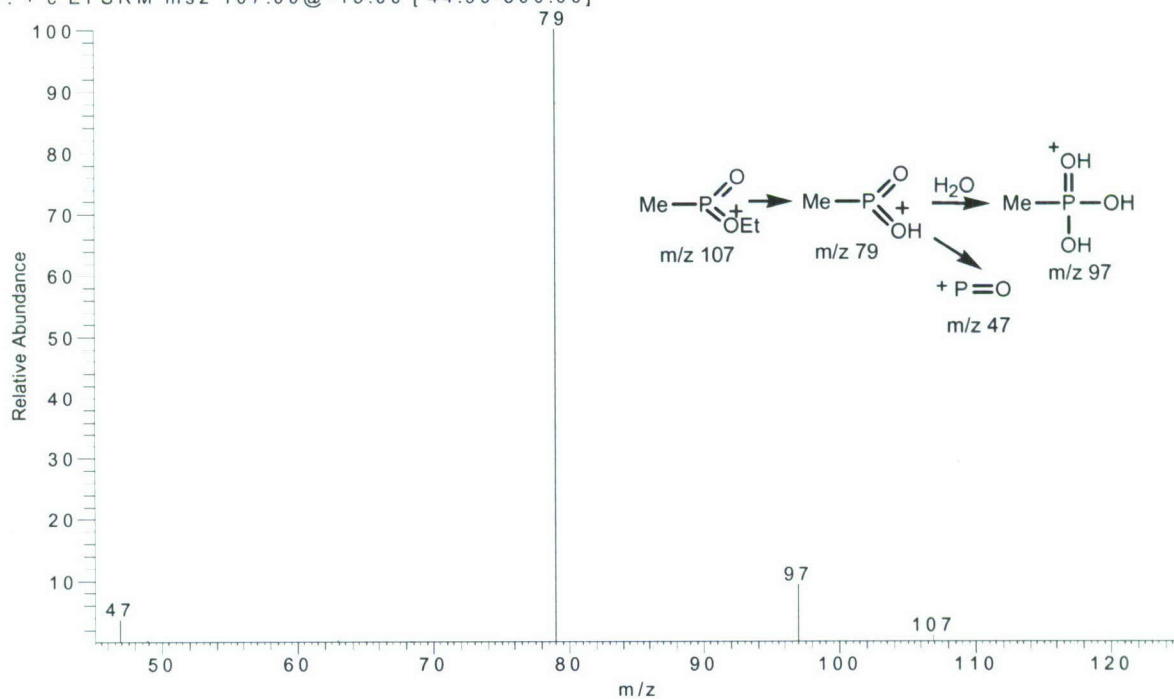


Figure 16. Product Ion Mass Spectrum of VX m/z 107 (EI)

v98 #247-249 RT: 2.96-2.98 AV: 3 NL: 1.25E5
T: + c EISRM ms2 98.00@ -15.00 [19.96-300.00]

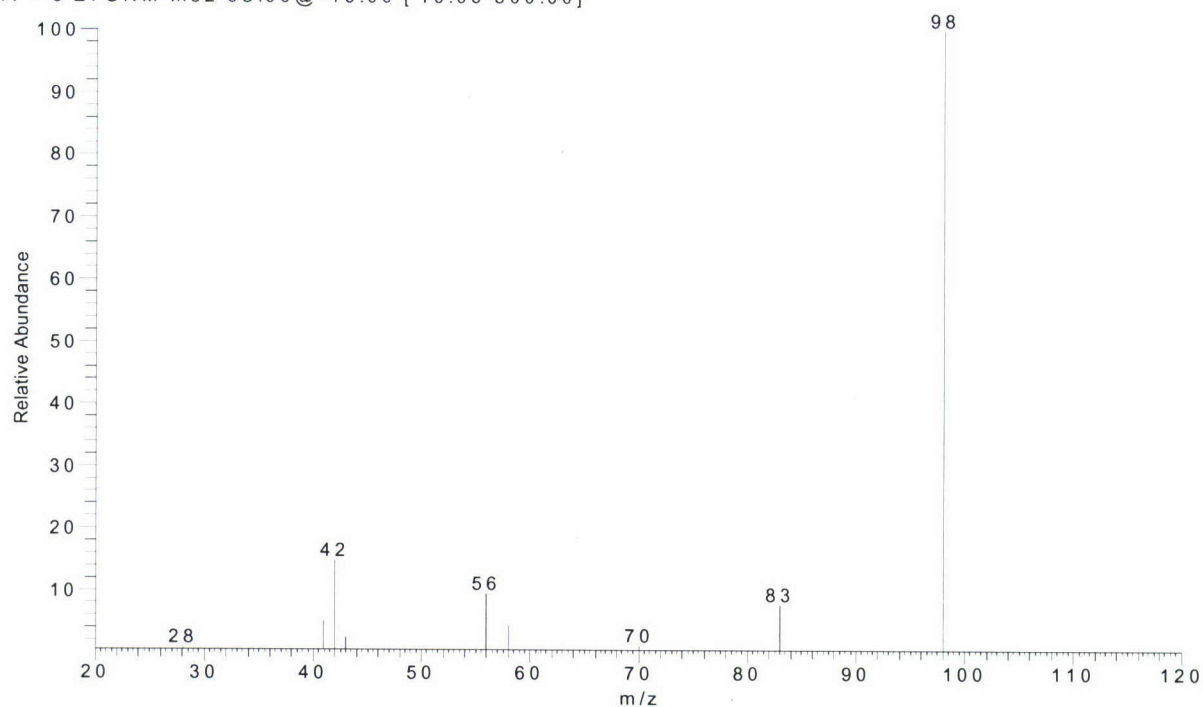


Figure 17. Product Ion Mass Spectrum of VX m/z 98 (EI)

v86ci #251-253 RT: 3.00-3.02 AV: 3 NL: 6.03E3
T: + c EISRM ms2 86.00@ -15.00 [19.96-300.00]

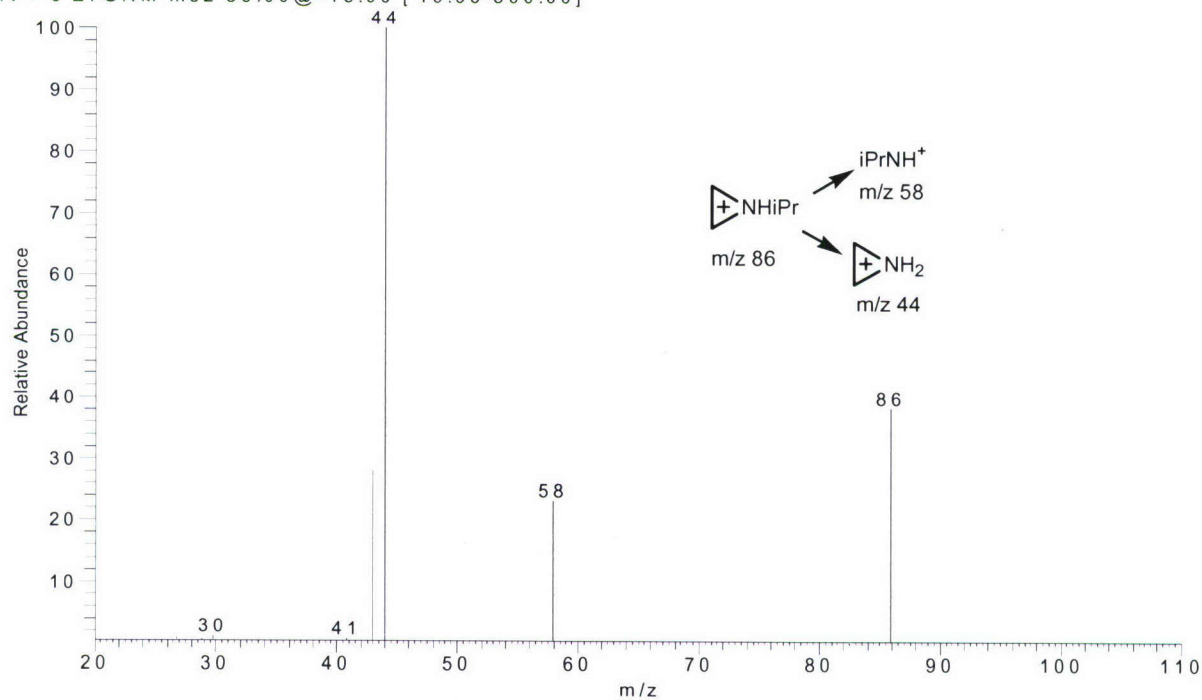


Figure 18. Product Ion Mass Spectrum of VX m/z 86 (CI)

v84 #248-250 RT: 2.96-2.99 AV: 3 NL: 2.29E5
T: + c EISRM ms2 84.00@ -15.00 [19.96-300.00]

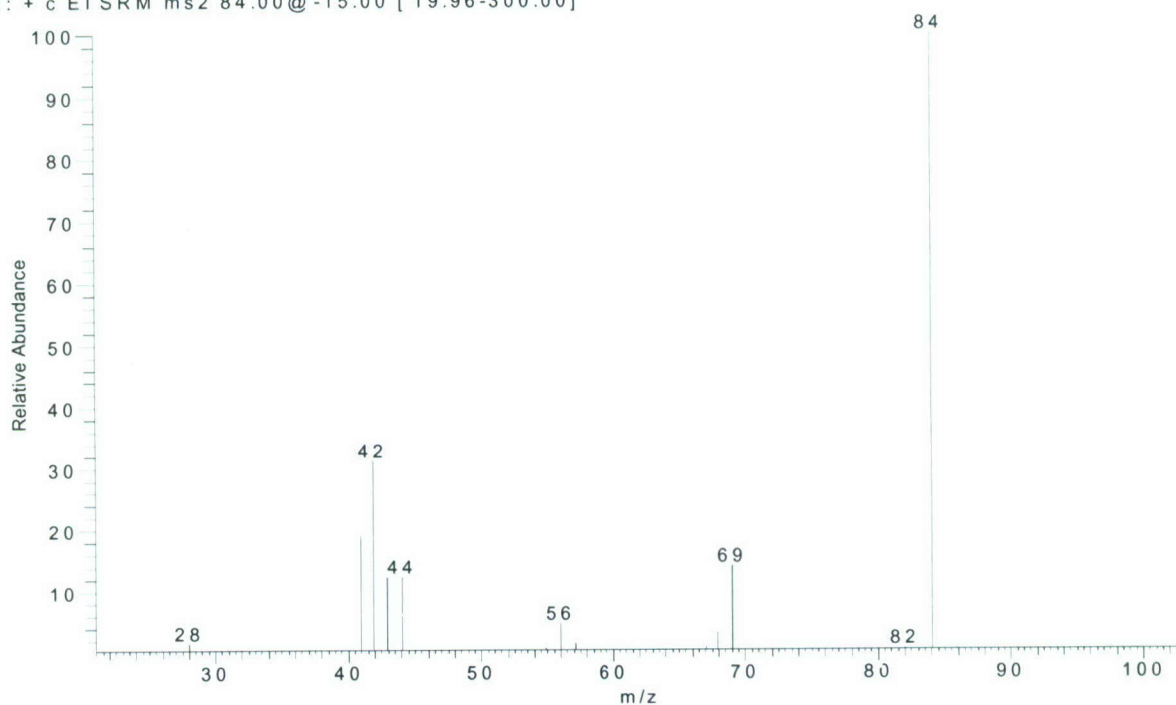


Figure 19. Product Ion Mass Spectrum of VX m/z 84 (EI)

v79 #251-252 RT: 3.00-3.01 AV: 2 NL: 1.16E5
T: + c EISRM ms2 79.00@ -15.00 [44.99-300.00]

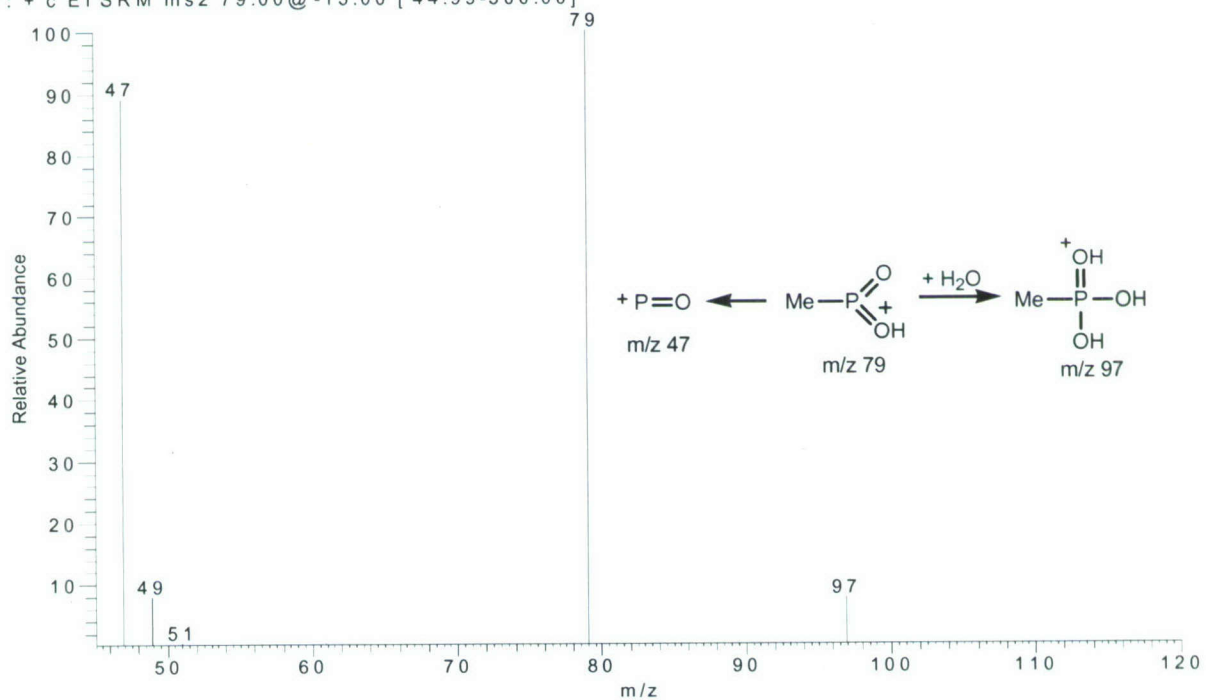


Figure 20. Product Ion Mass Spectrum of VX m/z 79 (EI)

v72 #250-252 RT: 2.99-3.02 AV: 3 NL: 3.21E5
T: + c E I SRM ms2 72.00@ -15.00 [19.96-300.00]

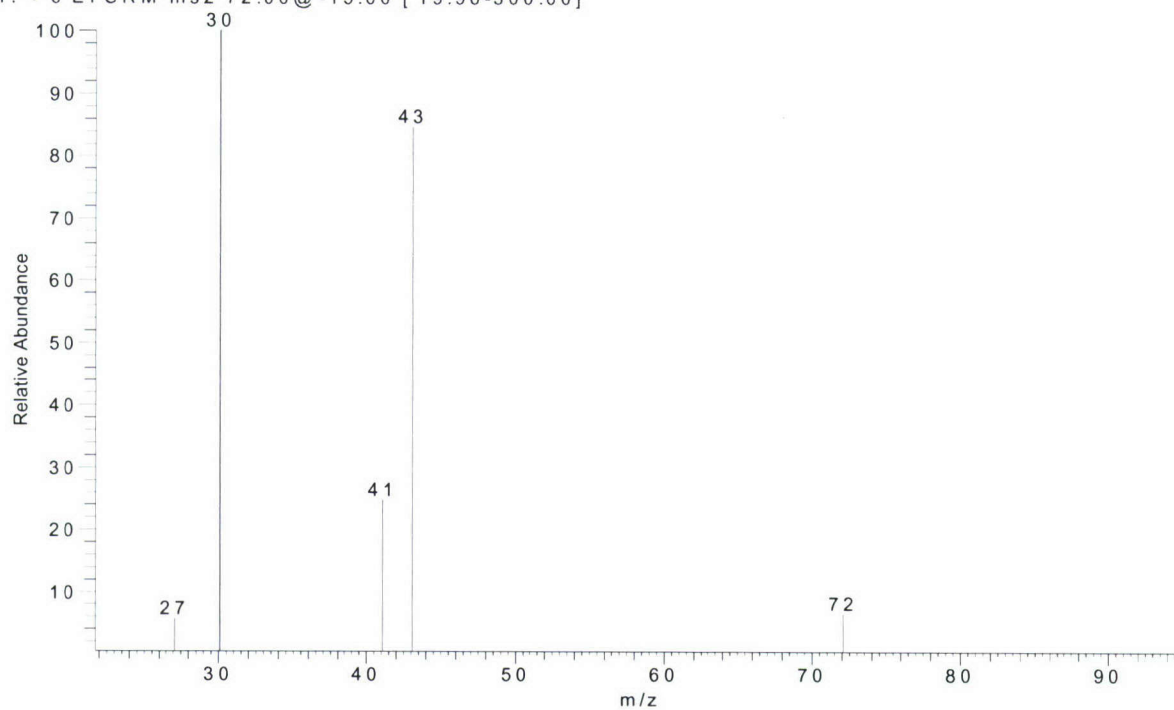


Figure 21. Product Ion Mass Spectrum of VX m/z 72 (EI)

v70 #251-253 RT: 3.00-3.02 AV: 3 NL: 9.39E4
T: + c E I SRM ms2 70.00@ -15.00 [19.96-300.00]

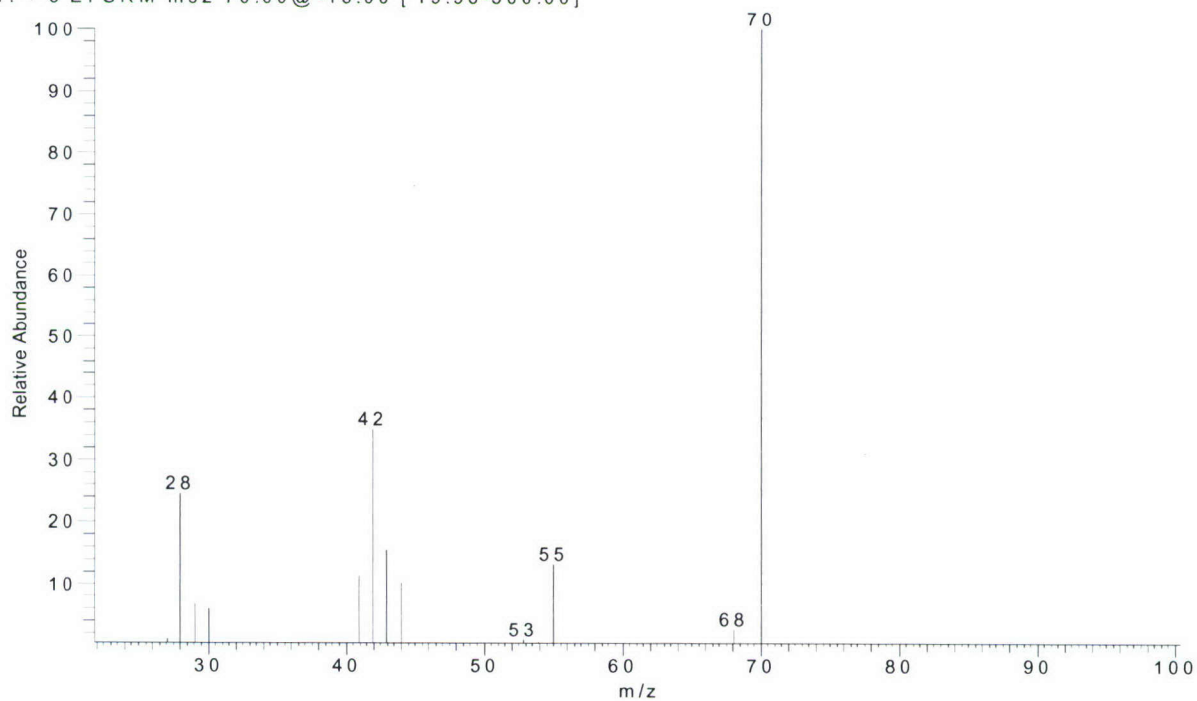


Figure 22. Product Ion Mass Spectrum of VX m/z 70 (EI)

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