

Managing Temporal and Spatial Variability in Vapor Intrusion Data

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**Environmental Monitoring and Data Quality Workshop
La Jolla, CA
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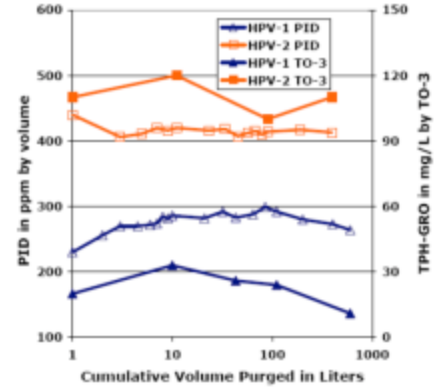
Report Documentation Page

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New Sampling Approaches

	Old	New
Temporal Variability	<p>8 to 24-Hour Time Weighted Average</p> 	<p>3 to 30 day Time Weighted Average</p> 
Spatial Variability	<p>1L Volume Weighted Average</p> 	<p>High Purge Volume Sampling</p> 

Old

New

Temporal Variability

8 to 24-Hour Time Weighted Average



3 to 30 day Time Weighted Average

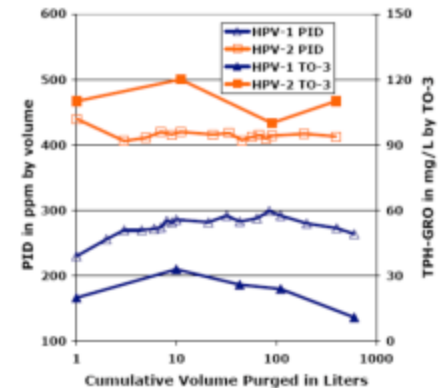


Spatial Variability

1L Volume Weighted Average

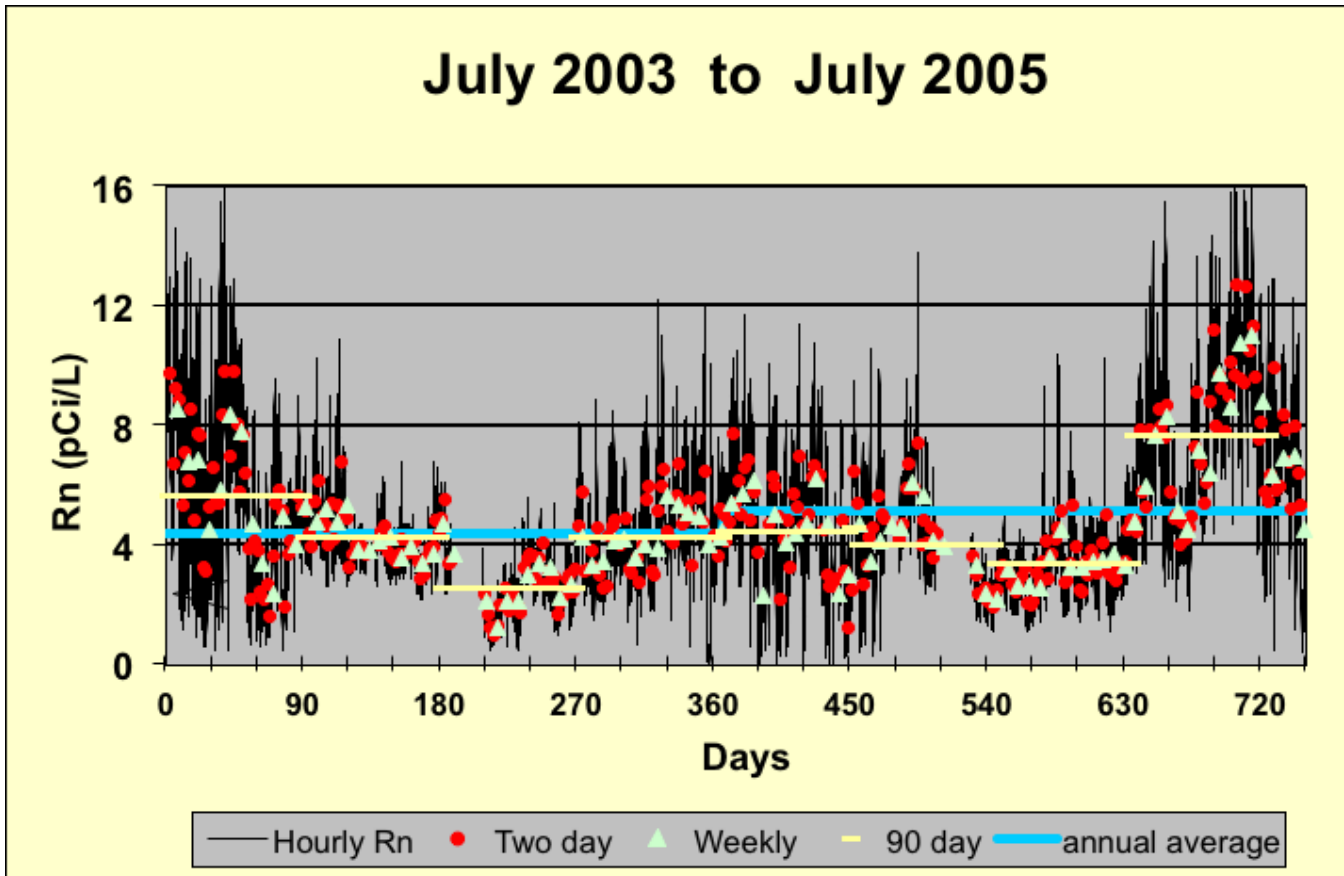


High Purge Volume Sampling



Temporal Variability (Indoor Air)

Single Day Radon Samples Provide Poor Estimates of Annual Average Radon Concentrations



Radon guidance recommends longer-term samples to manage temporal variability

Passive Samplers



ATD Tubes



3M OVM 3500

The mass (M) and time (t) are measured accurately. Key is to know the uptake rate (k^{-1})

$$C_0 = \frac{M}{k^{-1}t}$$

SKC Ultra II



Radiello™

Waterloo Membrane Sampler™



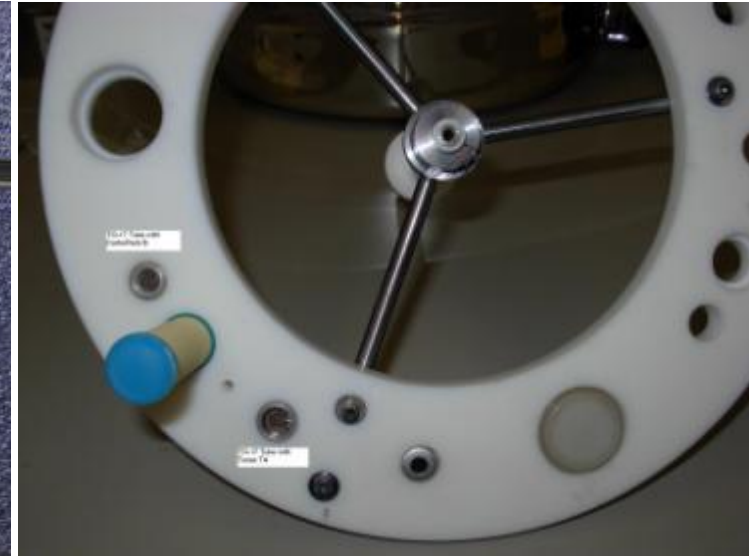
Benefits of Passive Sampling

- Simple (minimal training, less risk of leaks)
- Time-weighted average concentration
(up to a week or a month if needed)
- Low reporting limits with no premium cost
- Smaller – easy to ship, discrete to deploy
- Long history of use in Industrial Hygiene
- Less expensive
- Other benefits unique to each sampler

Laboratory Test Compound List

Analyte	Koc (mL/g)	OSWER indoor conc. at 10 ⁻⁶ risk (ppb)	Vapour pressure (atm)	Water solubility (g/l)
1,1,1-Trichloroethane	110	400	0.16	1.33
1,2,4-Trimethylbenzene	472	1.2	0.00197	0.0708
1,2-Dichloroethane	174	0.023	0.107	8.52
2-Butanone (MEK)	134	340	0.1026	~ 256
Benzene	59	0.10	0.125	1.75
Carbon tetrachloride	174	0.026	0.148	0.793
Naphthalene	2,000	0.57	0.000117	0.031
n-Hexane	3,000	57	0.197	0.0128
Tetrachloroethene	155	0.12	0.0242	0.2
Trichloroethene	166	0.22	0.0948	1.1

Experimental Apparatus



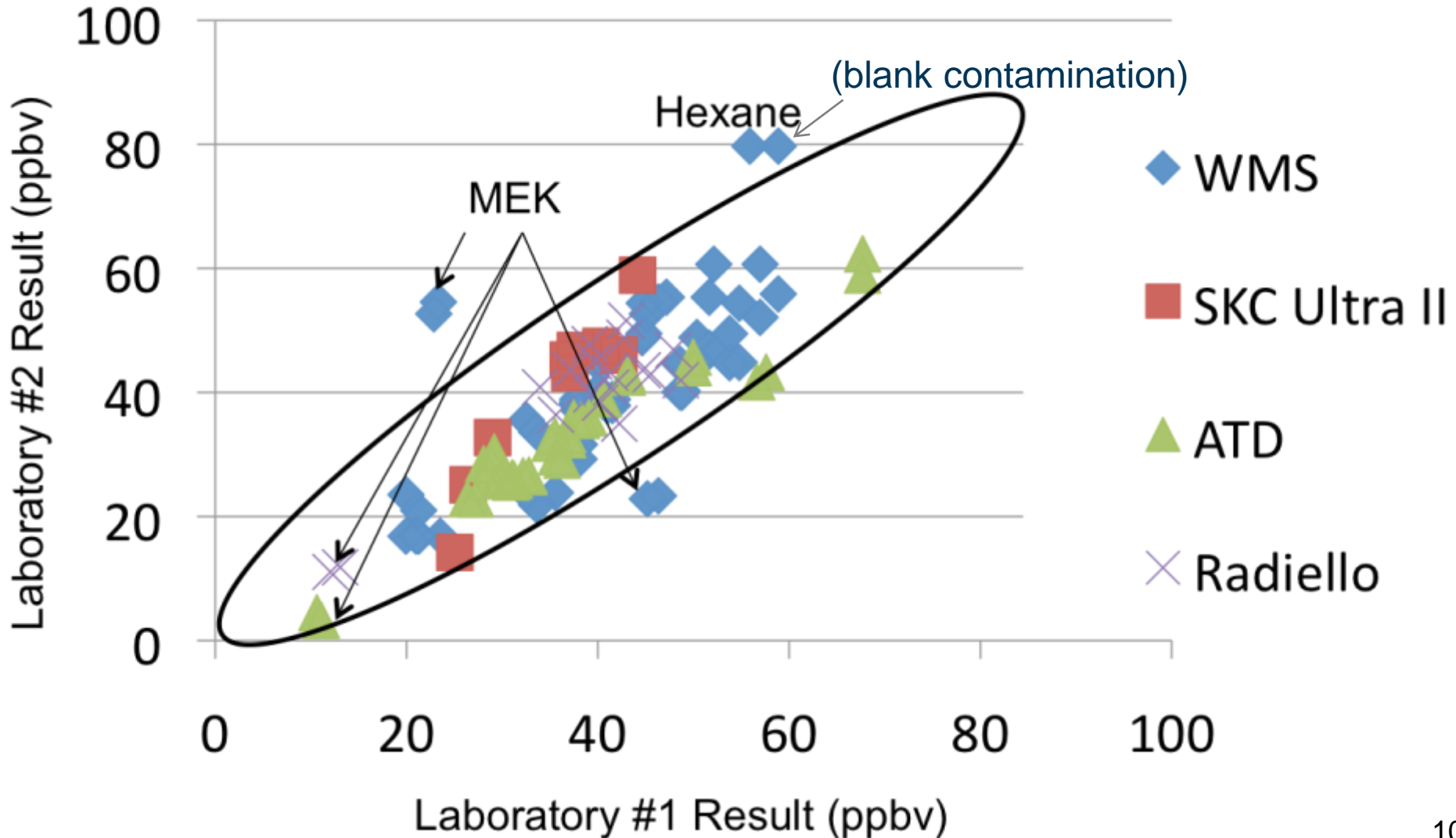
24 chambers x
5 sampler types x
3 replicates x
10 chemicals
= 3600 measurements



Inter-Laboratory Testing

Sampler Type	Home Laboratory	Secondary Laboratories	# of Samplers to Each Laboratory
Waterloo Membrane Sampler	University of Waterloo	Air Toxics Ltd Airzone One	2
ATD Tubes with Tenax TA	Air Toxics Ltd	Columbia Analytical Services University of Waterloo	2
ATD Tubes with CarboPack B	Air Toxics Ltd	Columbia Analytical Services University of Waterloo	2
SKC Ultra	Columbia Analytical Services	Air Toxics Ltd Airzone One	2
Radiello	Fondazione Salvatore Maugeri	Columbia Analytical Services Air Toxics Ltd	2

Interlab Test – Youden Plot



Fractional Factorial Testing

- A series of experiments strategically changing the 5 key factors

Run #	Approximate Concentration (ppbv)	Approximate Temperature (°C)	Face Velocity (m/s)	Duration (days)	Approximate Humidity (%R.H.)
1	100	17	0.41	1	90
2	1	17	0.014	1	90
3	100	30	0.41	1	30
4	1	30	0.014	1	30
5	100	30	0.41	7	90
6	1	30	0.014	7	90
7	100	17	0.41	7	30
8	1	17	0.014	7	30
9	50	20	0.23	4	60
10	50	20	0.23	4	60
11	100	17	0.014	1	30
12	1	17	0.41	1	30
13	100	17	0.014	7	90
14	1	17	0.41	7	90
15	100	30	0.014	7	30
16	1	30	0.41	7	30
17	100	30	0.014	1	90
18	1	30	0.41	1	90

Concentration

Temperature

Face Velocity

Sample Time

Humidity

24 chambers x
5 sampler types x
3 replicates x
10 chemicals
= 3600 measurements

ANOVA Analysis of the 5 Factors

Table 1: Statistical Significance of the Main Factors in the Fractional Factorial Experiments

Sampler Type	Analyte	Relative Humidity	Temperature	Face Velocity	Exposure Time	Concentration
ATD Carbopack	1,1,1-Trichloroethane	0.0778	0.0281	0.0106	0.0003	<.0001
ATD Carbopack	1,2,4-Trimethylbenzene	0.3181	0.0009	0.1245	0.5664	0.0011
ATD Carbopack	1,2-Dichloroethane	0.0012	0.6819	0.7406	<.0001	0.1371
ATD Carbopack	2-Butanone (MEK)	0.0693	0.4097	0.0603	0.7378	0.0119
ATD Carbopack	Hexane	0.7999	0.2913	0.4002	0.0272	0.1177
ATD Carbopack	Benzene	0.4718	0.2468	0.0547	0.0023	0.0331
ATD Carbopack	Carbon tetrachloride	0.0434	0.2975	0.3501	<.0001	<.0001
ATD Carbopack	Naphthalene	0.2629	0.6088	0.293	0.007	0.0778
ATD Carbopack	Trichloroethene	0.0113	0.2781	0.0002	<.0001	0.9484
ATD Carbopack	Tetrachloroethene	0.8513	0.004	0.0071	0.8484	0.0727
ATD Tenax	1,1,1-Trichloroethane	<.0001	0.2715	0.0021	<.0001	<.0001
ATD Tenax	1,2,4-Trimethylbenzene	0.9169	0.8868	0.0121	0.0296	0.2864
ATD Tenax	1,2-Dichloroethane	0.9154	0.8908	0.4733	<.0001	<.0001
ATD Tenax	2-Butanone (MEK)	0.7719	0.0799	0.1479	<.0001	<.0001
ATD Tenax	Hexane	0.6362	0.21	0.6114	<.0001	0.1148
ATD Tenax	Benzene	0.8106	0.0059	0.438	<.0001	0.0442
ATD Tenax	Carbon tetrachloride	<.0001	0.0229	0.0159	<.0001	<.0001
ATD Tenax	Naphthalene	0.311	0.2147	0.565	0.025	0.0347
ATD Tenax	Trichloroethene	0.5875	0.0002	0.0153	<.0001	0.475
ATD Tenax	Tetrachloroethene	0.3221	0.4522	0.11	<.0001	0.9827
RADIELLO	1,1,1-Trichloroethane	0.1005	0.0261	0.003	0.0899	0.0548
RADIELLO	1,2,4-Trimethylbenzene	0.6688	0.0007	<.0001	0.1133	0.0451
RADIELLO	1,2-Dichloroethane	0.0005	0.054	0.0002	0.0327	<.0001
RADIELLO	2-Butanone (MEK)	<.0001	0.5801	0.0003	0.0738	<.0001
RADIELLO	Hexane	0.1795	0.0066	0.0021	<.0001	0.0035
RADIELLO	Benzene	0.0047	0.0496	0.0012	<.0001	0.6113
RADIELLO	Carbon tetrachloride	0.4994	0.0143	0.0513	0.1724	0.9018
RADIELLO	Naphthalene	0.6635	0.0008	0.933	0.1183	0.0005
RADIELLO	Trichloroethene	0.001	0.0032	<.0001	0.0002	0.0169
RADIELLO	Tetrachloroethene	0.2158	0.0023	<.0001	0.3477	0.9109
SKC	1,1,1-Trichloroethane	0.0906	0.1691	0.0055	0.0096	0.0001
SKC	1,2,4-Trimethylbenzene	0.1362	0.3054	0.0012	0.0004	<.0001
SKC	1,2-Dichloroethane	<.0001	0.5187	0.1033	0.9879	0.6424
SKC	2-Butanone (MEK)	<.0001	0.2819	0.3914	0.0073	0.0028
SKC	Hexane	0.0006	0.0398	0.012	0.4921	0.1584
SKC	Benzene	0.0318	0.0551	0.9085	0.0218	0.0125
SKC	Carbon tetrachloride	0.0223	0.2682	0.032	<.0001	<.0001
SKC	Naphthalene	0.1182	0.1437	0.6579	<.0001	0.1122
SKC	Trichloroethene	<.0001	0.9977	0.0306	0.5618	<.0001
SKC	Tetrachloroethene	0.4868	0.0368	0.018	0.0097	0.1261
WMS	1,1,1-Trichloroethane	0.0224	0.9489	0.0042	0.6355	0.4719
WMS	1,2,4-Trimethylbenzene	0.7716	0.7992	<.0001	0.1467	0.0194
WMS	1,2-Dichloroethane	0.7347	0.1749	0.0054	0.0325	0.1887
WMS	2-Butanone (MEK)	0.5881	0.3369	0.14	0.0319	0.0027
WMS	Hexane	0.6198	0.4942	0.022	0.0003	0.0001
WMS	Benzene	0.5712	0.9017	0.0328	0.0012	0.0099
WMS	Carbon tetrachloride	0.0016	0.3838	0.0035	0.0766	0.0553
WMS	Naphthalene	0.9025	0.4298	<.0001	0.5432	0.006
WMS	Trichloroethene	0.6289	0.0325	0.0006	0.8376	0.0124
WMS	Tetrachloroethene	0.5923	0.1477	<.0001	0.9894	0.0074

red highlighted cells indicate statistical significance when alpha=0.05, therefore, p-value<0.05 = significant

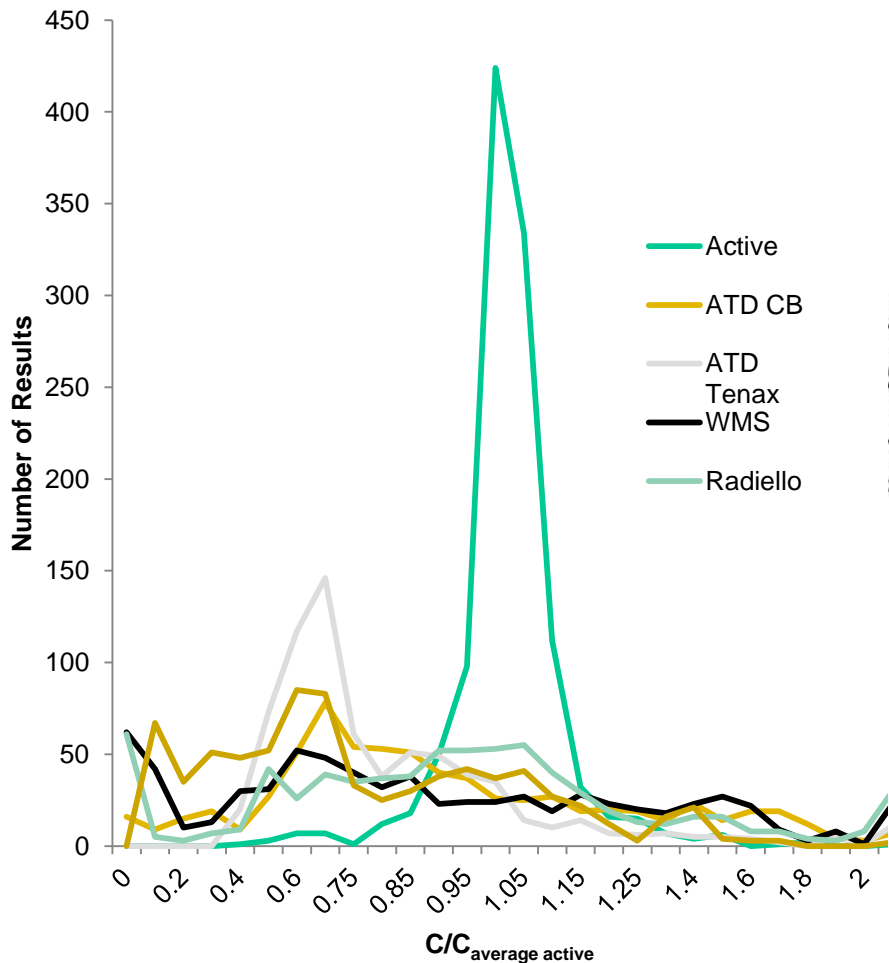
The five factors tested showed statistically significant effects on the concentrations measured with passive samplers.

Need to think about whether “statistically significant” is also “practically significant”

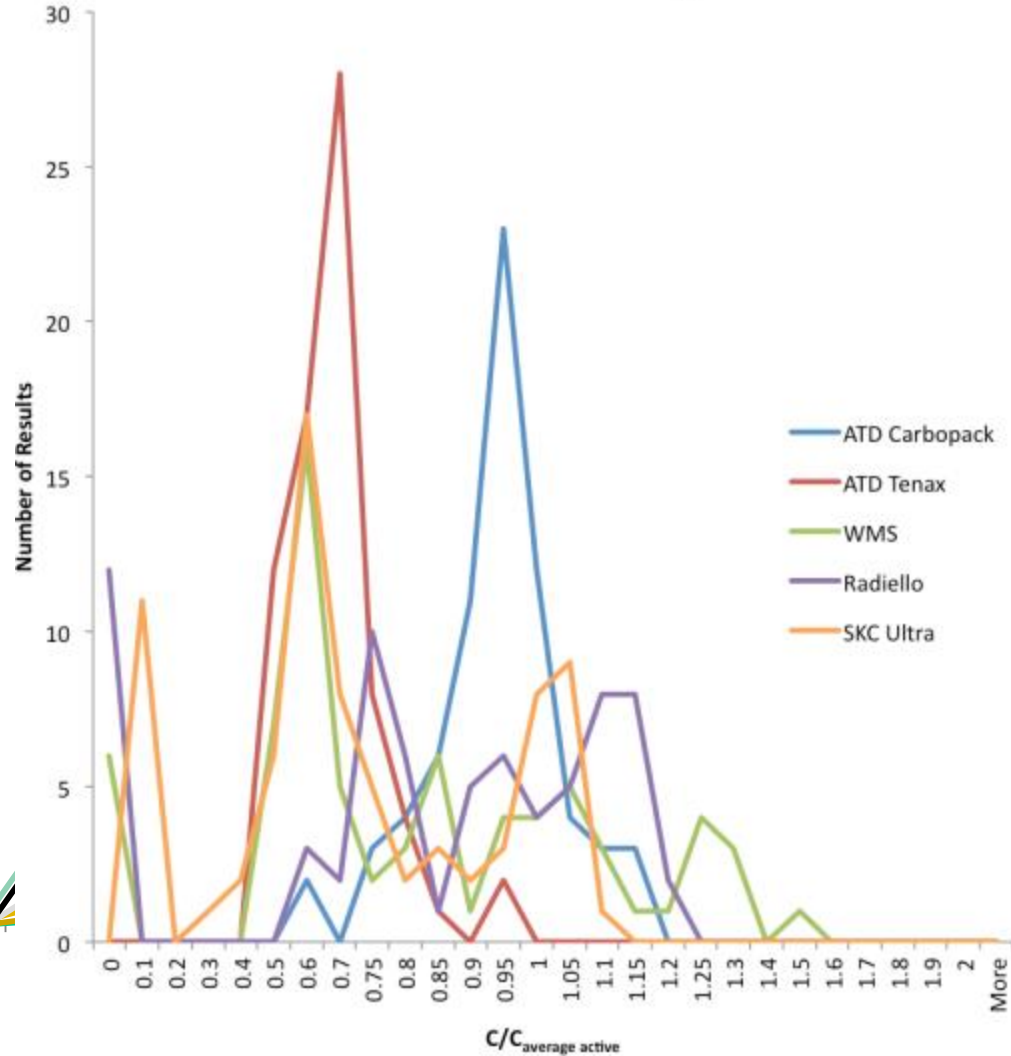
Red cells are significant at 95% level

Low Concentration Lab Tests

Compiled Low Concentration Laboratory Chamber Testing Data



Trichloroethene Histogram



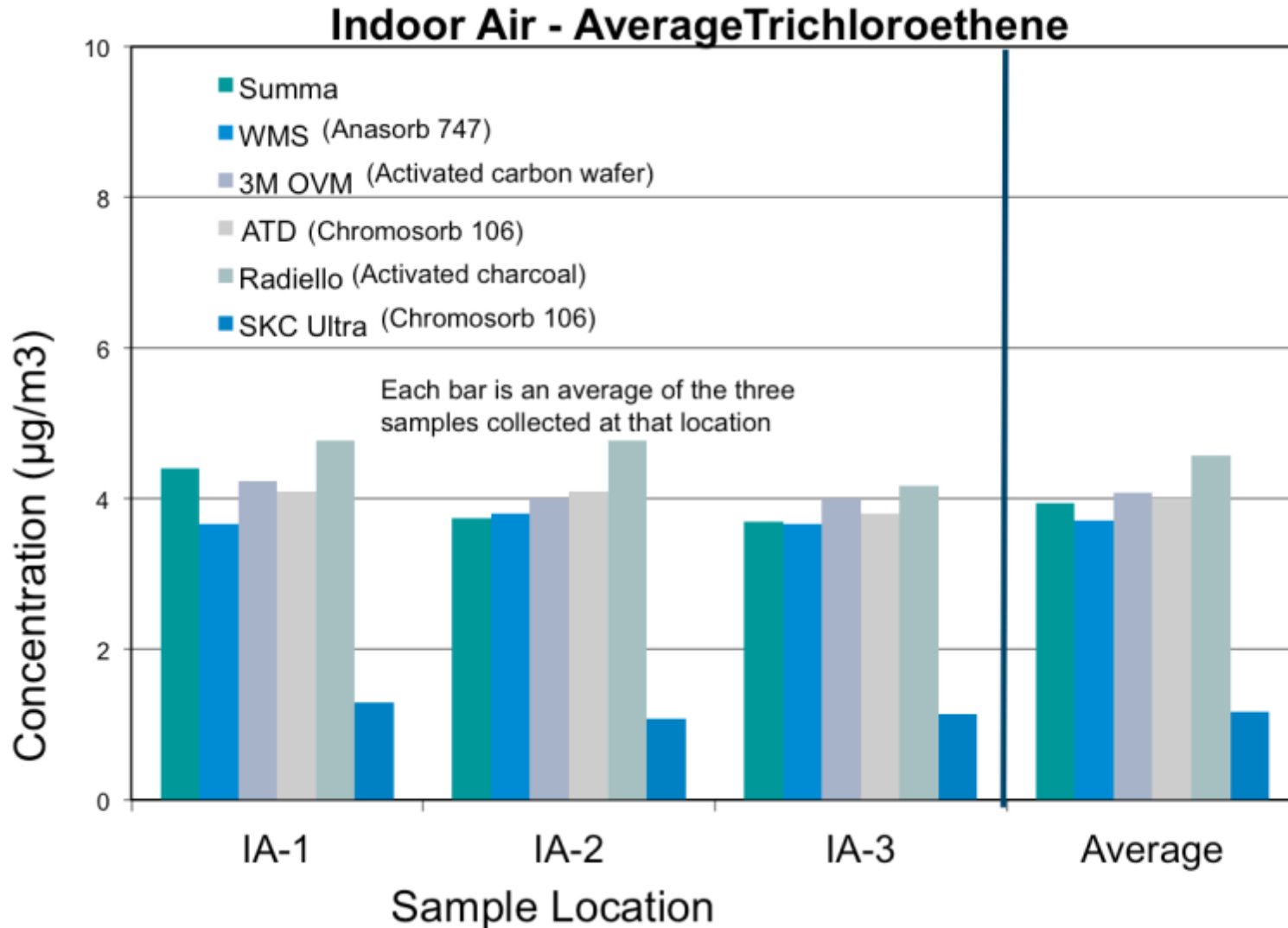
Field Testing of Indoor Air



Navy San Diego, CA
Cherry Point, NC
CRREL, NH

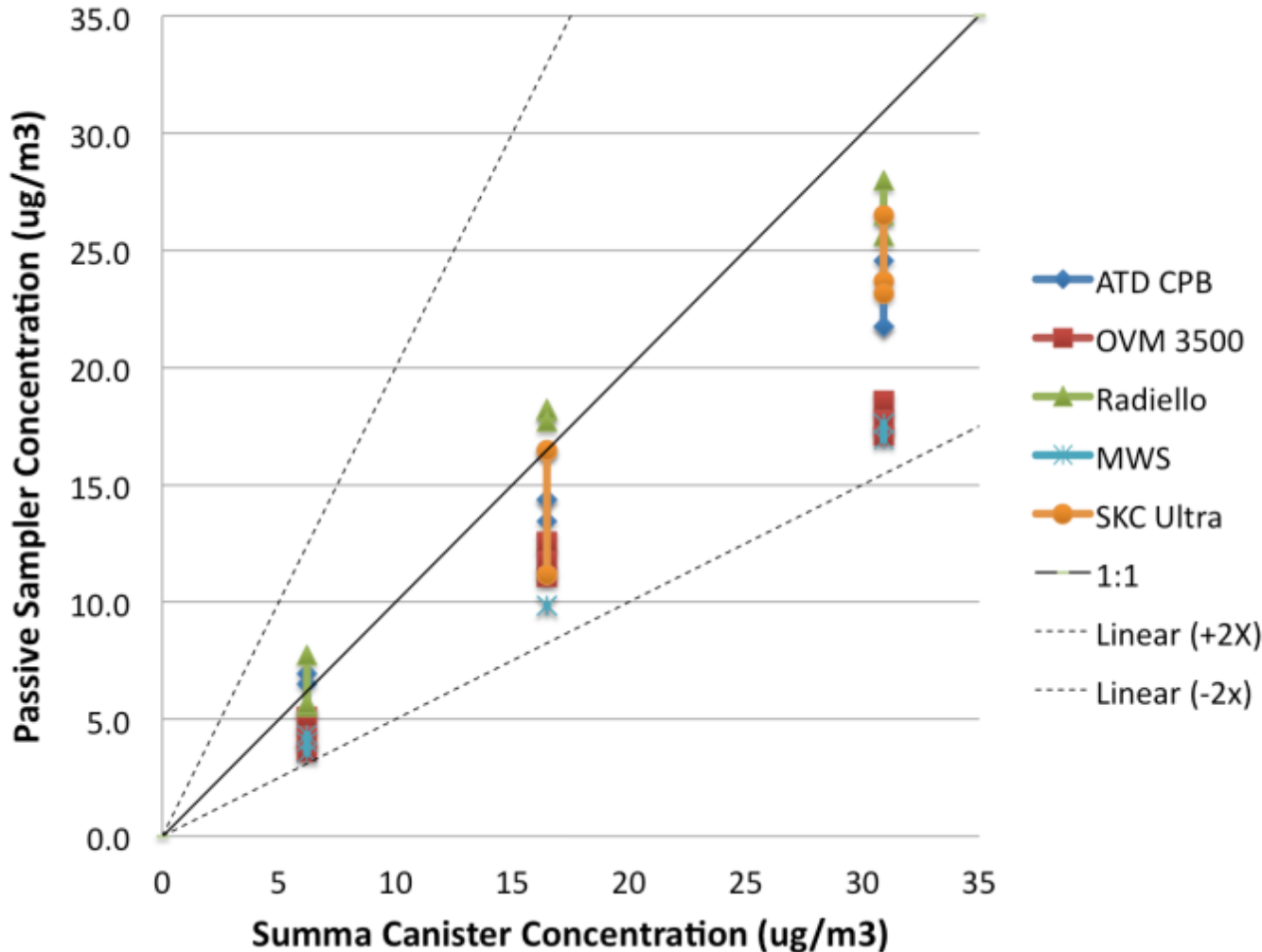
3 locations/site
5 passive samplers
Summa cans
Triplicates of each

Indoor Air TCE at San Diego



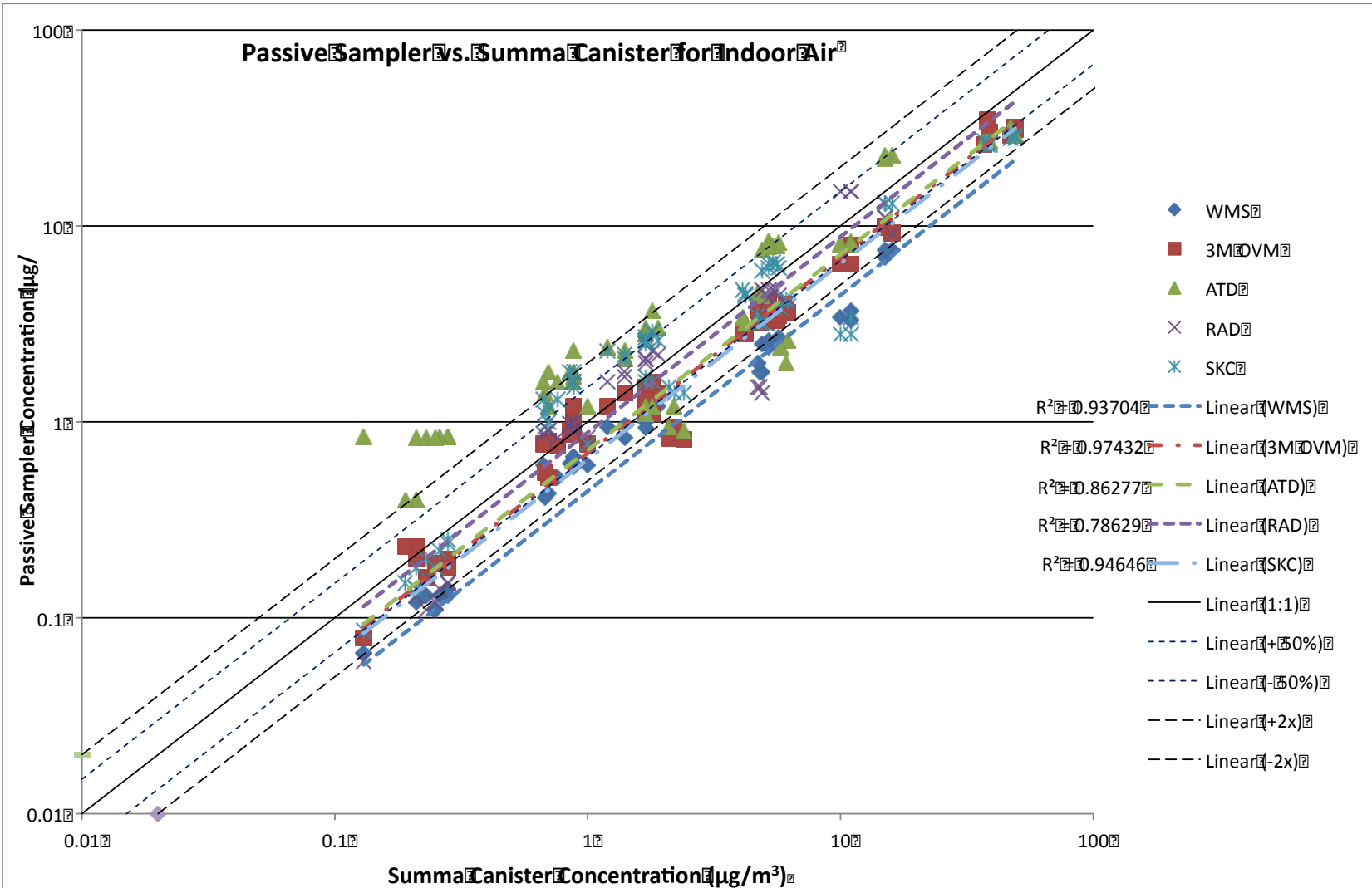
Indoor Air at CRREL

TCE Concentration in Passive Samplers vs Summa Canisters



All passive sampler results were within 2X of Summa canister data for TCE

Indoor Air VOCs at Cherry Point



Broader range (>100X), but still almost all passive data are within 2X of Summa canisters

High Concentration Lab Tests

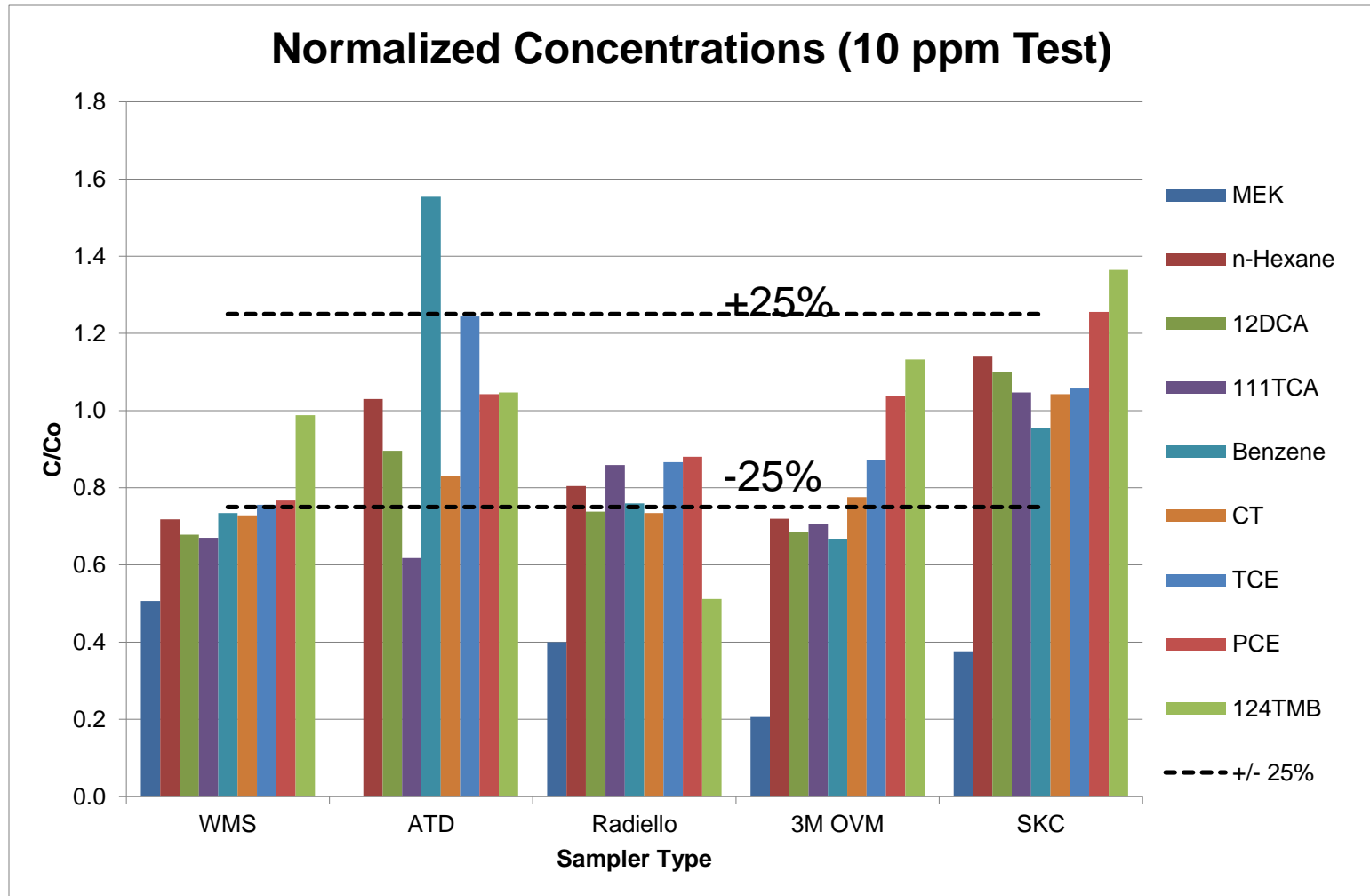
(To mimic soil gas conditions)



High Concentration Lab Tests



High Concentrations Test Results

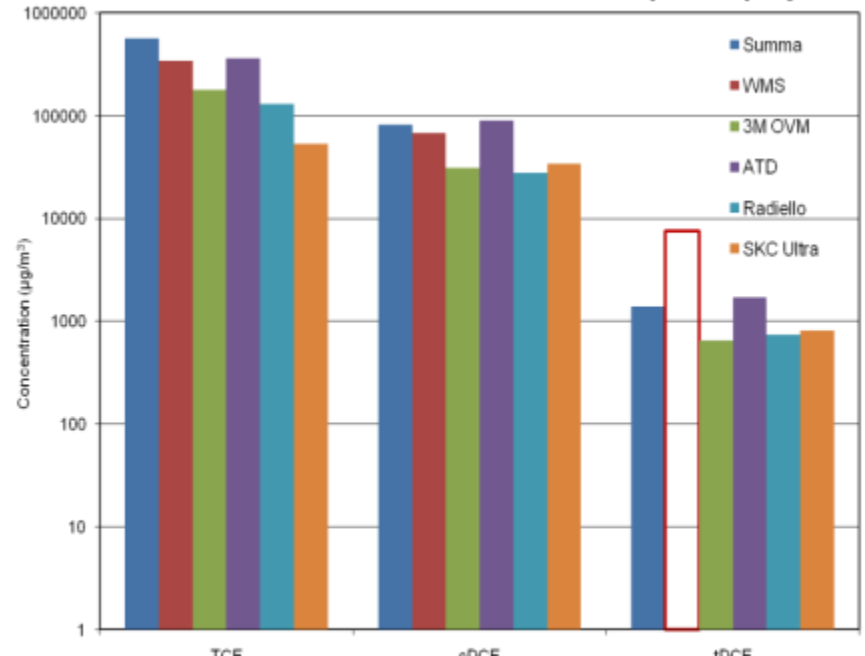
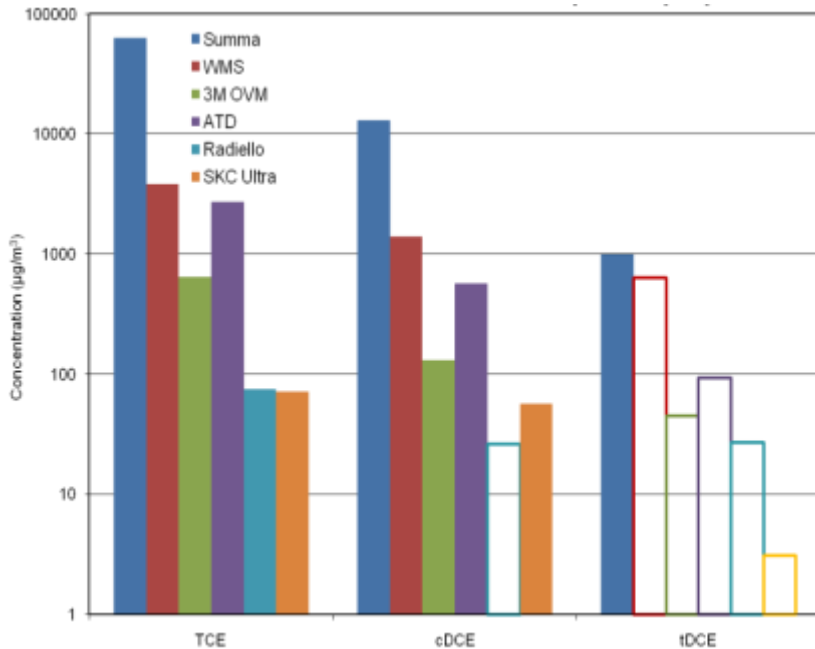


Sub-Slab – Navy San Diego



Sub-slab samples only
Fully-passive and with PID purging (flow-through)

Starvation proportional to uptake rate
Less starvation for semi-passive samples



Modified Uptake Rates

Lower uptake rate = less starvation



SKC Ultra II and 12-hole Cap



ATD Tube & Pinhole Cap

Sorbent Selection

Carbopack B

(Graphitized Carbon Black)

Surface Area: 100 m²/g

Desorption Temperature: 330 °C

	Challenge Volume (Liters)					
	0.2	1	5	10	20	100
Halocarbon 12	Green	Green	Green	Green	Green	Green
Chloromethane	Green	Green	Green	Green	Green	Green
Halocarbon 114	Green	Green	Green	Green	Green	Green
Vinyl chloride	Green	Green	Green	Green	Green	Green
1,3-Butadiene	Green	Green	Green	Green	Green	Green
Bromomethane	Green	Green	Green	Green	Green	Green
Chloroethane	Green	Green	Green	Green	Green	Green
Halocarbon 11	Green	Green	Green	Green	Green	Green
Acrylonitrile	Green	Green	Green	Green	Green	Green
1,1-Dichloroethene	Green	Yellow	Red	Red	Red	Red
Methylene chloride	Green	Green	Green	Green	Green	Green
3-Chloropropene	Green	Yellow	Red	Red	Red	Red
Halocarbon 113	Green	Green	Green	Green	Green	Green
1,1-Dichloroethane	Green	Green	Green	Green	Green	Green
cis-1,2-Dichloroethene	Green	Green	Green	Green	Green	Green
Chloroform	Green	Green	Green	Green	Green	Green
1,2-Dichloroethane	Green	Green	Green	Green	Green	Green
1,1,1-Trichloroethane	Green	Green	Green	Green	Green	Green
Benzene	Green	Green	Green	Green	Green	Green
Carbon tetrachloride	Green	Green	Green	Green	Green	Red
1,2-Dichloropropane	Green	Green	Green	Green	Green	Red
Trichloroethene	Green	Green	Green	Green	Green	Yellow
cis-1,3-Dichloropropene	Green	Green	Green	Green	Green	Red
trans-1,3-Dichloropropene	Green	Green	Green	Green	Green	Red
1,1,2-Trichloroethane	Green	Green	Green	Green	Green	Green
Toluene	Green	Green	Green	Green	Green	Green
1,2-Dibromoethane	Green	Green	Green	Green	Green	Yellow
Tetrachloroethene	Green	Green	Green	Green	Green	Green
Chlorobenzene	Green	Green	Green	Green	Green	Green
Ethylbenzene	Green	Green	Green	Green	Green	Green
m & p-Xylene	Green	Green	Green	Green	Green	Green
Styrene	Green	Green	Green	Green	Green	Green
1,1,2,2-Tetrachloroethane	Green	Green	Green	Green	Green	Green
o-Xylene	Green	Green	Green	Green	Green	Green
4-Ethyltoluene	Green	Green	Green	Green	Green	Green
1,3,5-Trimethylbenzene	Green	Green	Green	Green	Green	Green
1,2,4-Trimethylbenzene	Green	Green	Green	Green	Green	Green
1,3-Dichlorobenzene	Green	Green	Green	Green	Green	Green
1,4-Dichlorobenzene	Green	Green	Green	Green	Green	Green
1,2-Dichlorobenzene	Green	Green	Green	Green	Green	Green
1,2,4-Trichlorobenzene	Green	Green	Green	Green	Green	Green
Hexachlorobutadiene	Green	Green	Green	Green	Green	Green

Performance Key

Safe to use: Recovery is greater than 50%
Caution: Recovery is between 21 to 79%
Not Recommended: Recovery is less than 20%
* indicates this analyte was strongly adsorbed



Carbopack X

(Graphitized Carbon Black)

Surface Area: 240 m²/g

Desorption Temperature: 330 °C

	Challenge Volume (Liters)					
	0.2	1	5	10	20	100
Halocarbon 12	Green	Green	Green	Green	Green	Green
Chloromethane	Green	Green	Green	Green	Green	Green
Halocarbon 114	Green	Green	Green	Green	Green	Green
Vinyl chloride	Green	Green	Green	Green	Green	Green
1,3-Butadiene	Green	Green	Green	Green	Green	Green
Bromomethane	Green	Green	Green	Green	Green	Green
Chloroethane	Green	Green	Green	Green	Green	Green
Halocarbon 11	Green	Green	Green	Green	Green	Green
Acrylonitrile	Green	Green	Green	Green	Green	Green
1,1-Dichloroethene	Green	Green	Green	Green	Green	Green
Methylene chloride	Green	Green	Green	Green	Green	Green
3-Chloropropene	Green	Green	Yellow	Red	Red	Red
Halocarbon 113	Green	Green	Green	Green	Green	Green
1,1-Dichloroethane	Green	Green	Green	Green	Green	Green
cis-1,2-Dichloroethene	Green	Green	Green	Green	Green	Green
Chloroform	Green	Green	Green	Green	Green	Green
1,2-Dichloroethane	Green	Green	Green	Green	Green	Green
1,1,1-Trichloroethane	Green	Green	Green	Green	Green	Green
Benzene	Green	Green	Green	Green	Green	Green
Carbon tetrachloride	Green	Green	Green	Green	Green	Red
1,2-Dichloropropane	Green	Green	Green	Green	Green	Red
Trichloroethene	Green	Green	Green	Green	Green	Yellow
cis-1,3-Dichloropropene	Green	Green	Green	Green	Green	Red
trans-1,3-Dichloropropene	Green	Green	Green	Green	Green	Red
1,1,2-Trichloroethane	Green	Green	Green	Green	Green	Green
Toluene	Green	Green	Green	Green	Green	Green
1,2-Dibromoethane	Green	Green	Green	Green	Green	Yellow
Tetrachloroethene	Green	Green	Green	Green	Green	Green
Chlorobenzene	Green	Green	Green	Green	Green	Green
Ethylbenzene	Green	Green	Green	Green	Green	Green
m & p-Xylene	Green	Green	Green	Green	Green	Green
Styrene	Green	Green	Green	Green	Green	Green
1,1,2,2-Tetrachloroethane	Green	Green	Green	Green	Green	Green
o-Xylene	Green	Green	Green	Green	Green	Green
4-Ethyltoluene *	Green	Green	Green	Green	Green	Green
1,3,5-Trimethylbenzene *	Green	Green	Green	Green	Green	Green
1,2,4-Trimethylbenzene *	Green	Green	Green	Green	Green	Green
1,3-Dichlorobenzene *	Green	Green	Green	Green	Green	Green
1,4-Dichlorobenzene *	Green	Green	Green	Green	Green	Green
1,2-Dichlorobenzene *	Green	Green	Green	Green	Green	Green
1,2,4-Trichlorobenzene *	Green	Green	Green	Green	Green	Green
Hexachlorobutadiene	Green	Green	Green	Green	Green	Green

Performance Key

Safe to use: Recovery is greater than 50%
Caution: Recovery is between 21 to 79%
Not Recommended: Recovery is less than 20%
* indicates this analyte was strongly adsorbed



Soil Gas @ 12 ft – Hill AFB

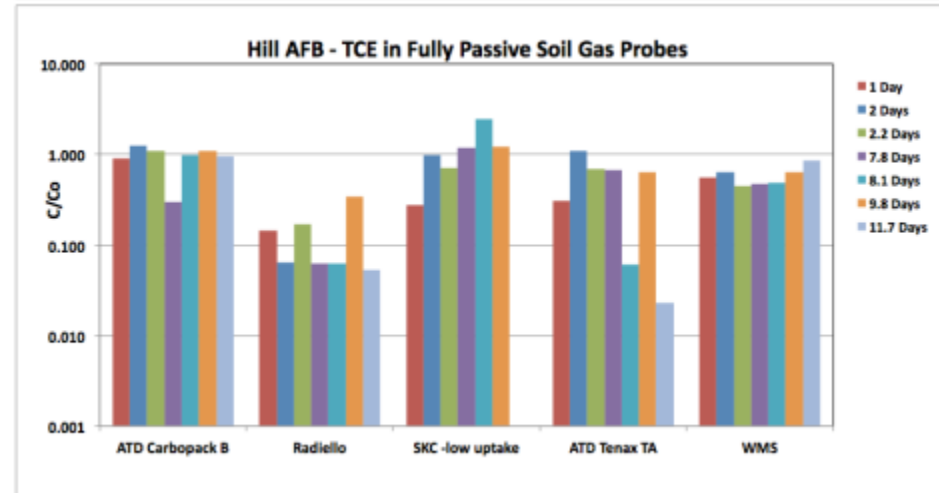
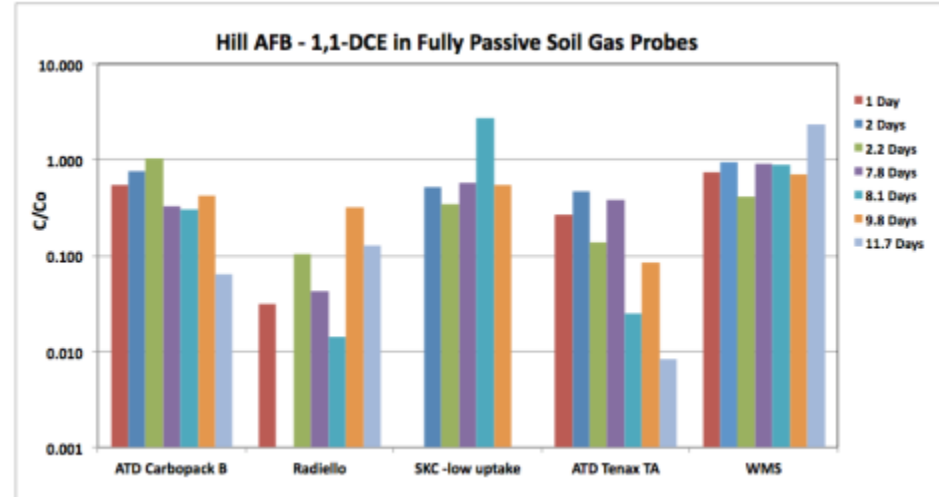


6 probes -12 ft deep

Latin Square Design

1 to 12 day exposures

C_o Measured using combination of Summa and Hapsite GC/MS

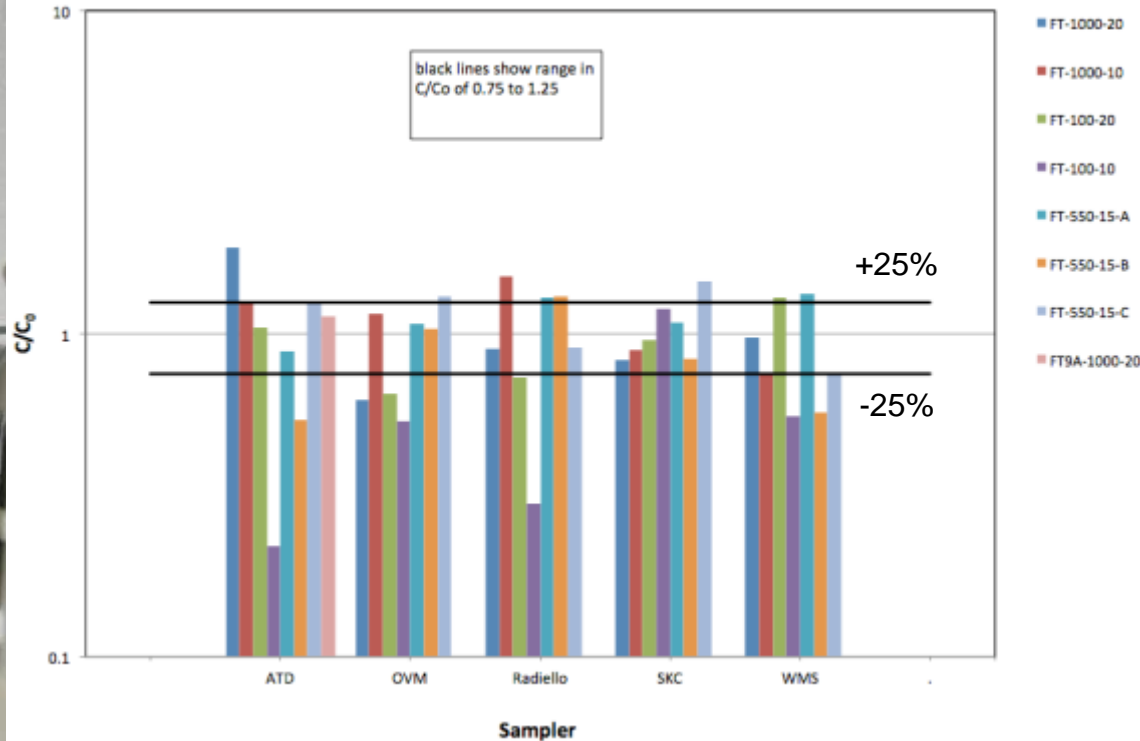


Negative bias for long duration with ATD-Tenax
 Negative bias for high uptake rate (Radiello)
 Otherwise, encouraging results for TCE and DCE

Flow-Through Cell – CRREL

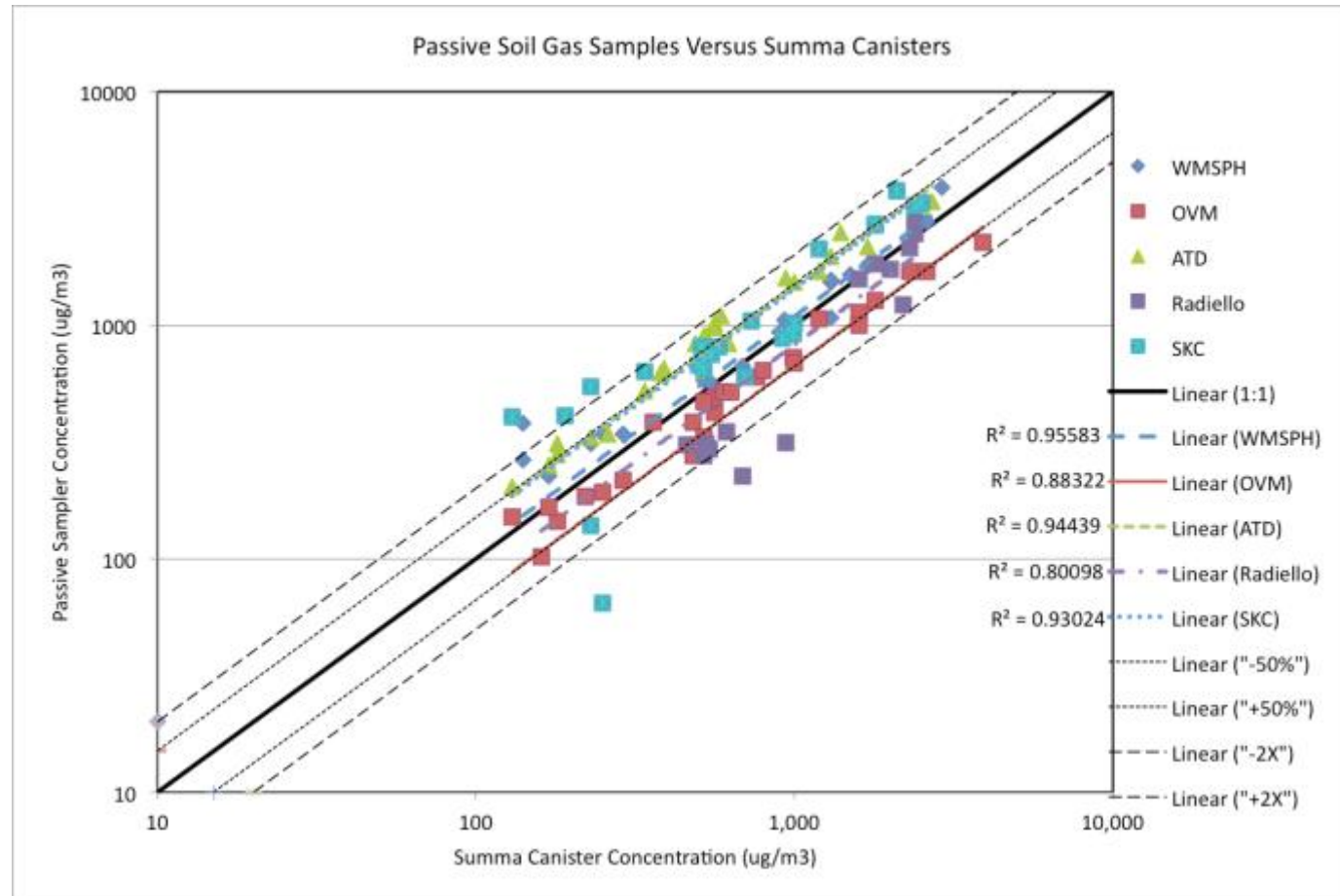


CRREL Sub-Slab Flow-Through Cell - TCE



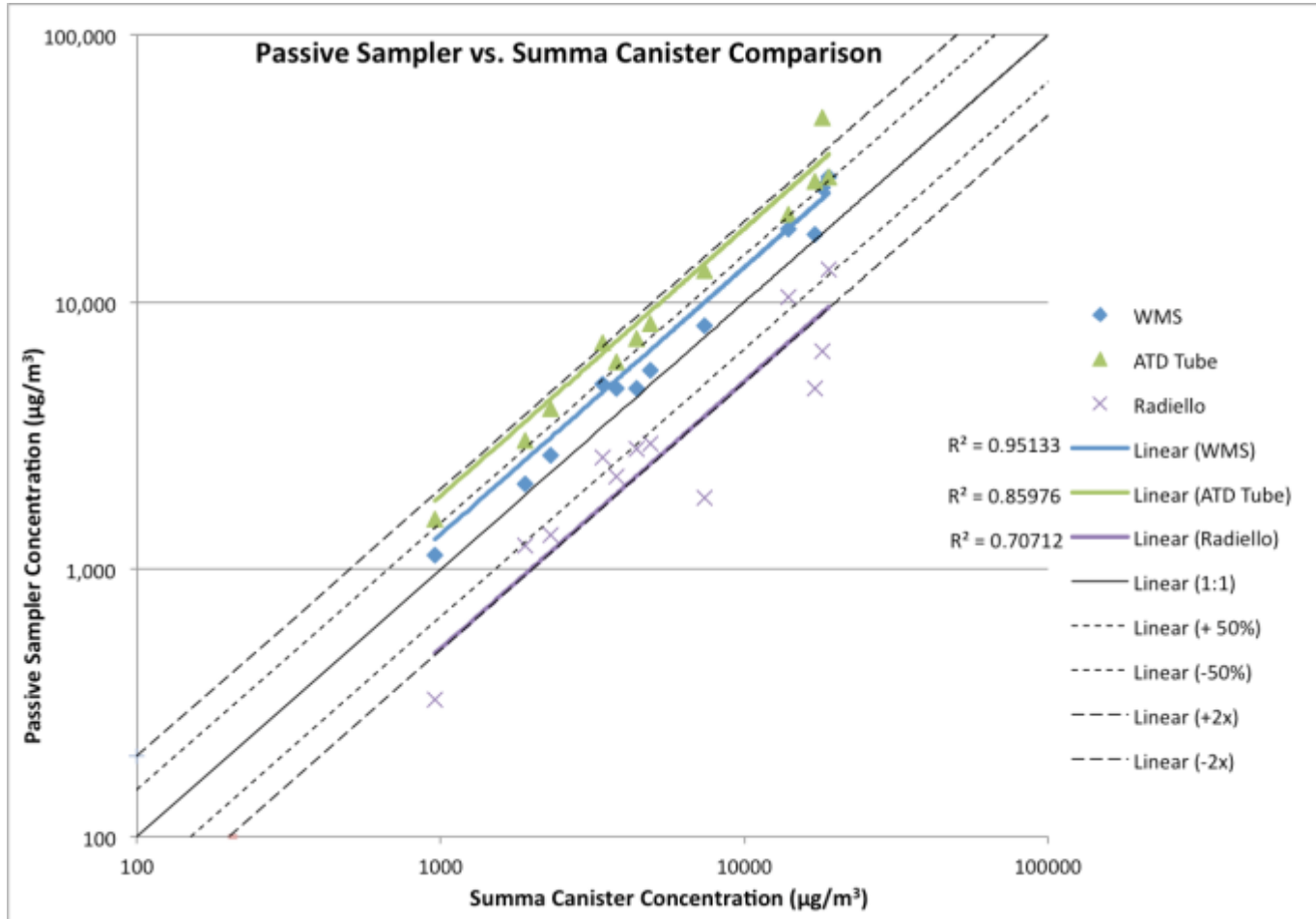
Flow-through cell to avoid starvation by design
 No starvation for high-uptake rate samplers
 Negative bias only for short duration/low-flow
 (insufficient purging)

Soil Vapor Sampling – NAS JAX



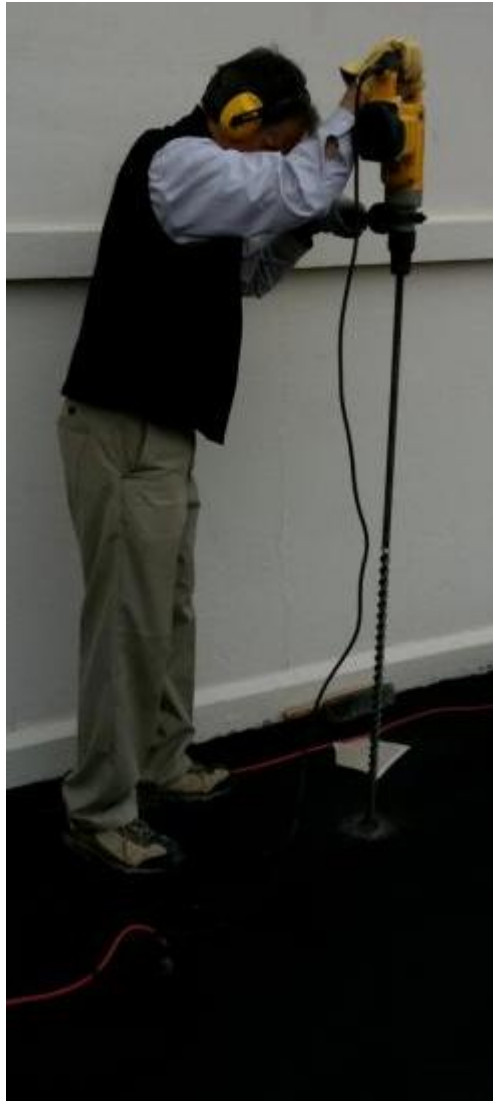
Probes to 3-4 feet deep, exposure durations of 20, 40 and 60 minutes
Strong correlations, regression slopes all near 1.0

Passive Sub-Slab – NAS JAX

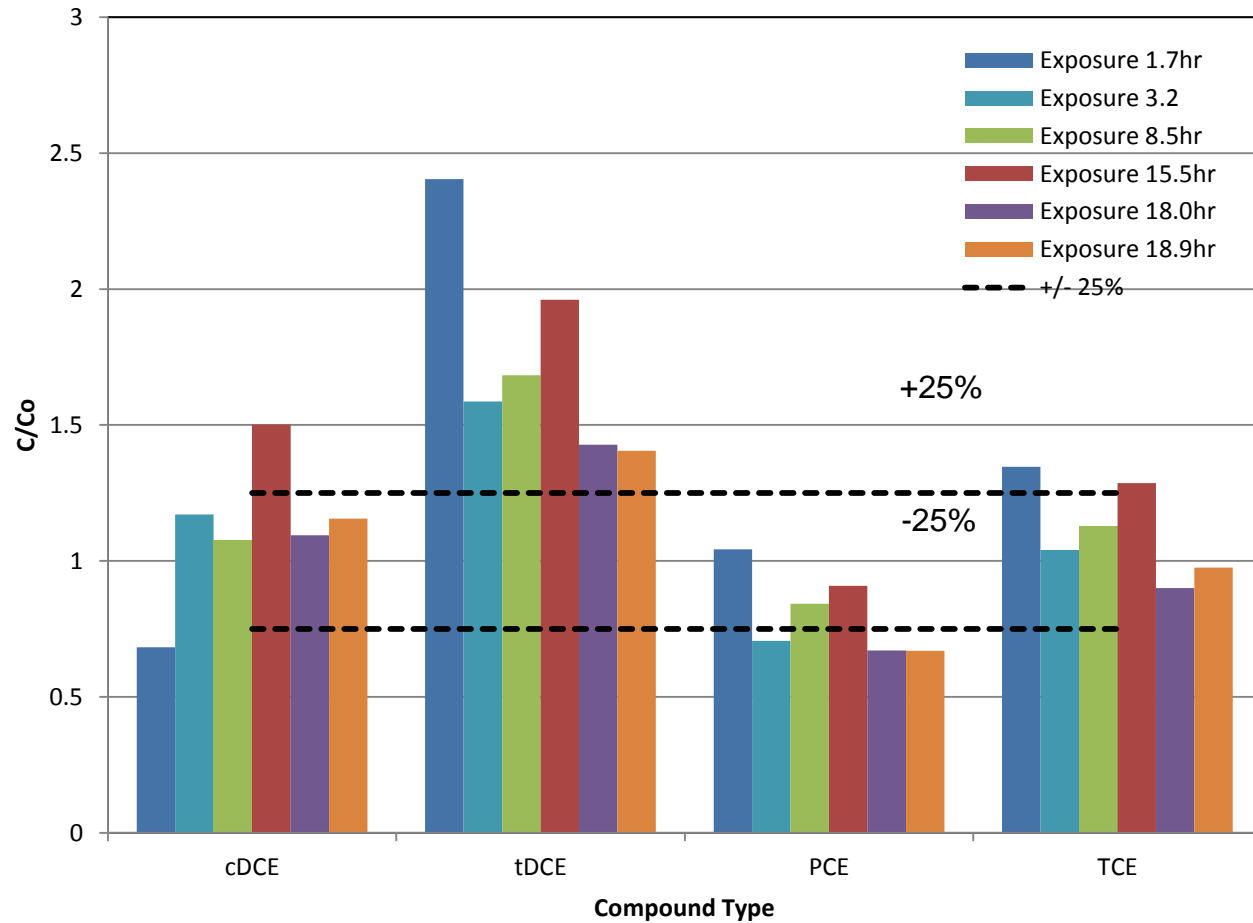


Limited to 1-inch diameter or less – Low-Uptake Rate Samplers

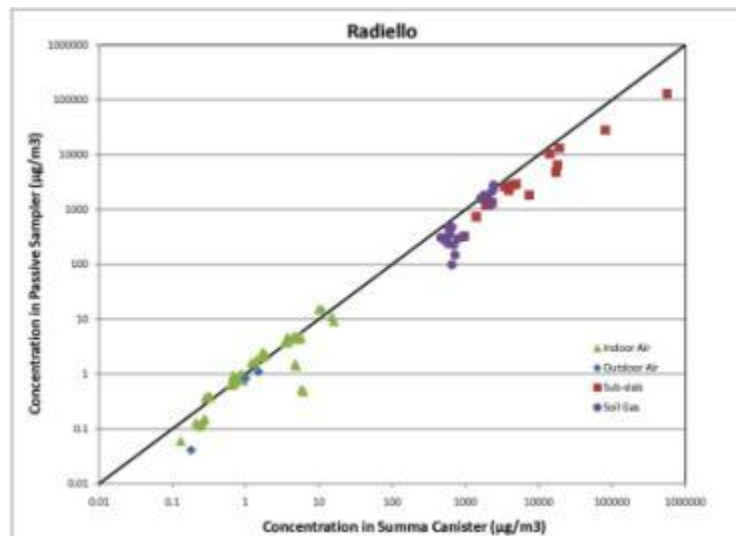
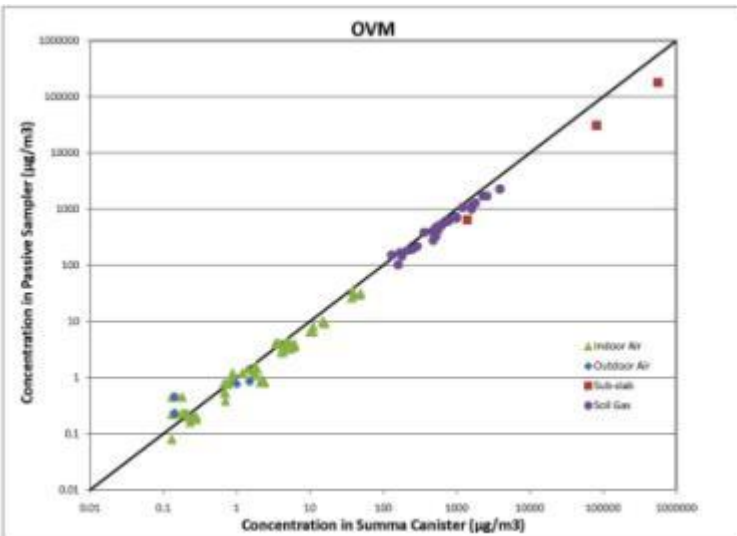
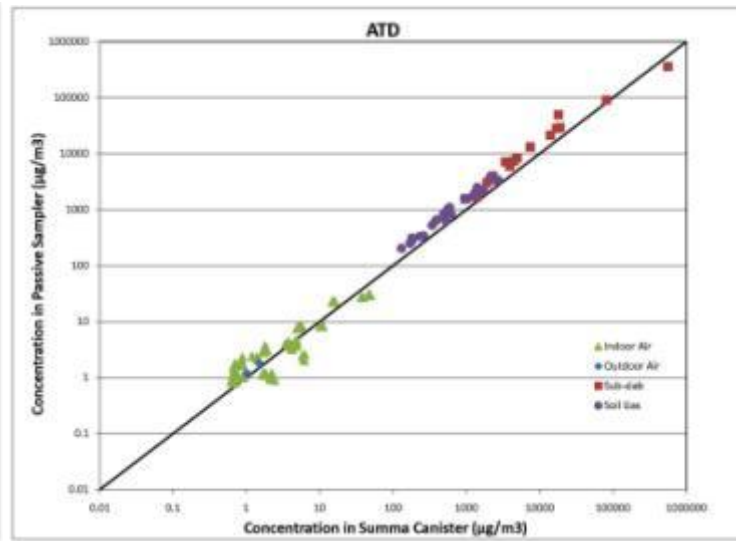
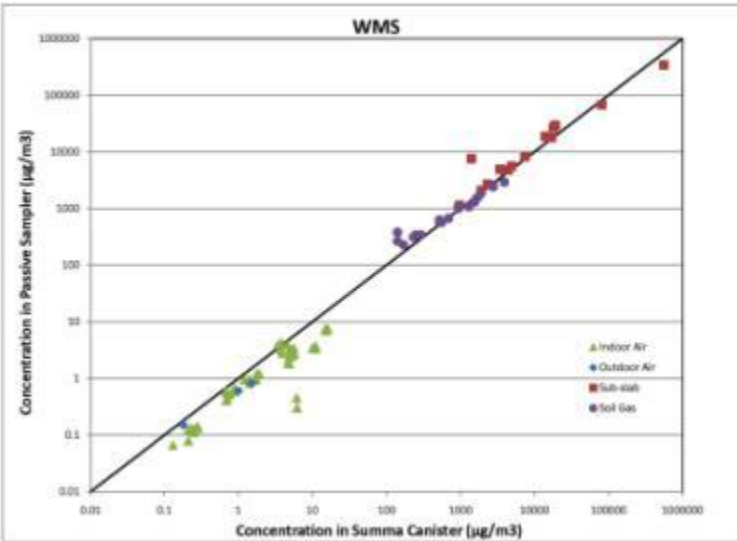
Temporary Passive - NAS JAX



WMS-PH Samples vs Summa Canisters



Overall Correlation between Passive and Active Samplers



Strong correlation to conventional samples over 6+ orders of magnitude

Quantitative results for soil vapor (a breakthrough)

Maybe we don't need to be using so many Summa Canisters



Cost Comparison

Simple comparison:

6 indoor samples

2 outdoor samples

6 sub-slab samples

Summa	WMS	Radiello	ATD	3M OVM	SKC
\$6,810	\$3,670	\$3,590	\$3,590	\$3,610	\$4,100

Ballpark 50% cost for passive samplers versus Summa cans

(even with some side-by-side Summa cans for benchmarking, you can still save a lot of money)

Case Study – Air Force Base

- TCE concentrations in Area of Interest:
 - Groundwater up to 100,000 ug/L
 - Soil Vapor up to 6,000,000 ug/m³
- Used Waterloo Membrane Samplers (4 weeks)
 - No VOCs above RBSLs or ambient background
- Open Bay Doors – huge ventilation rate
- Screen out, or reduce to low priority for VI

Even if there is no significant risk, it still needs to be documented

Regulators & occupants often prefer indoor air data



High Purge Volume Sampling



Fan or Vacuum

Bleed Valve

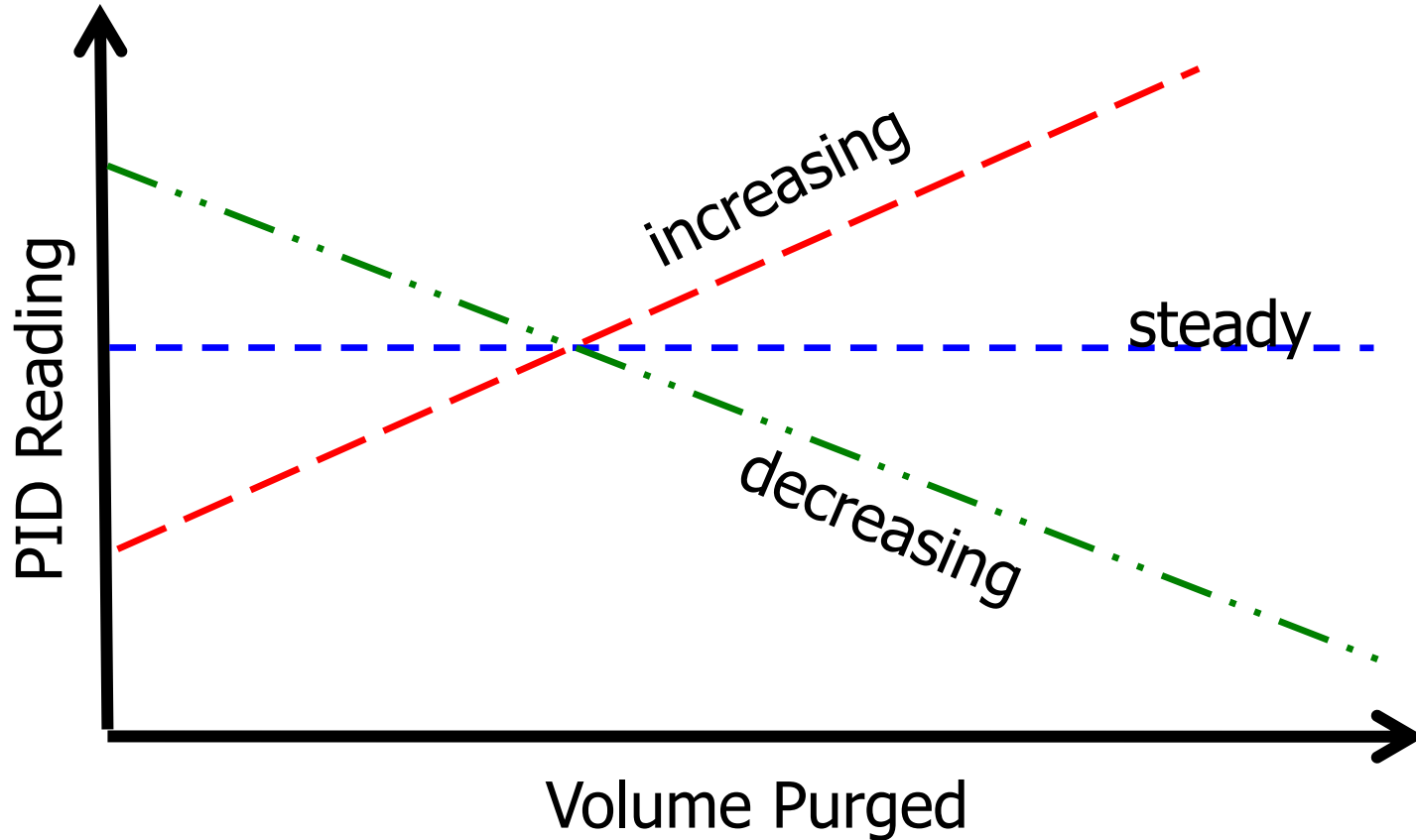
Sample Port

Vacuum Gauge

Cored Hole

Lung Box

PID Reading vs Volume Purged



Infer concentrations versus distance based on C versus V trend
Minimize risk of missing a localized “hot spot”

High Purge Volume (HPV) Testing

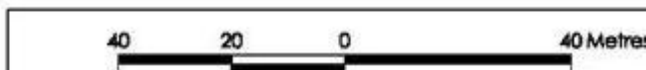
Semi-conductor manufacturer, roughly 100,000 ft² (i.e. 100 sub-slab samples?)

- Conducted 12 HPV Tests
- 2 in soil gas probes
- 7 in sub-slab probes
- 3 in monitoring wells



Note "Block F"

SUB-SLAB SOIL GAS MONITORING
LOCATION
AVERAGE VOC
GROUNDWATER MONITORING WELL

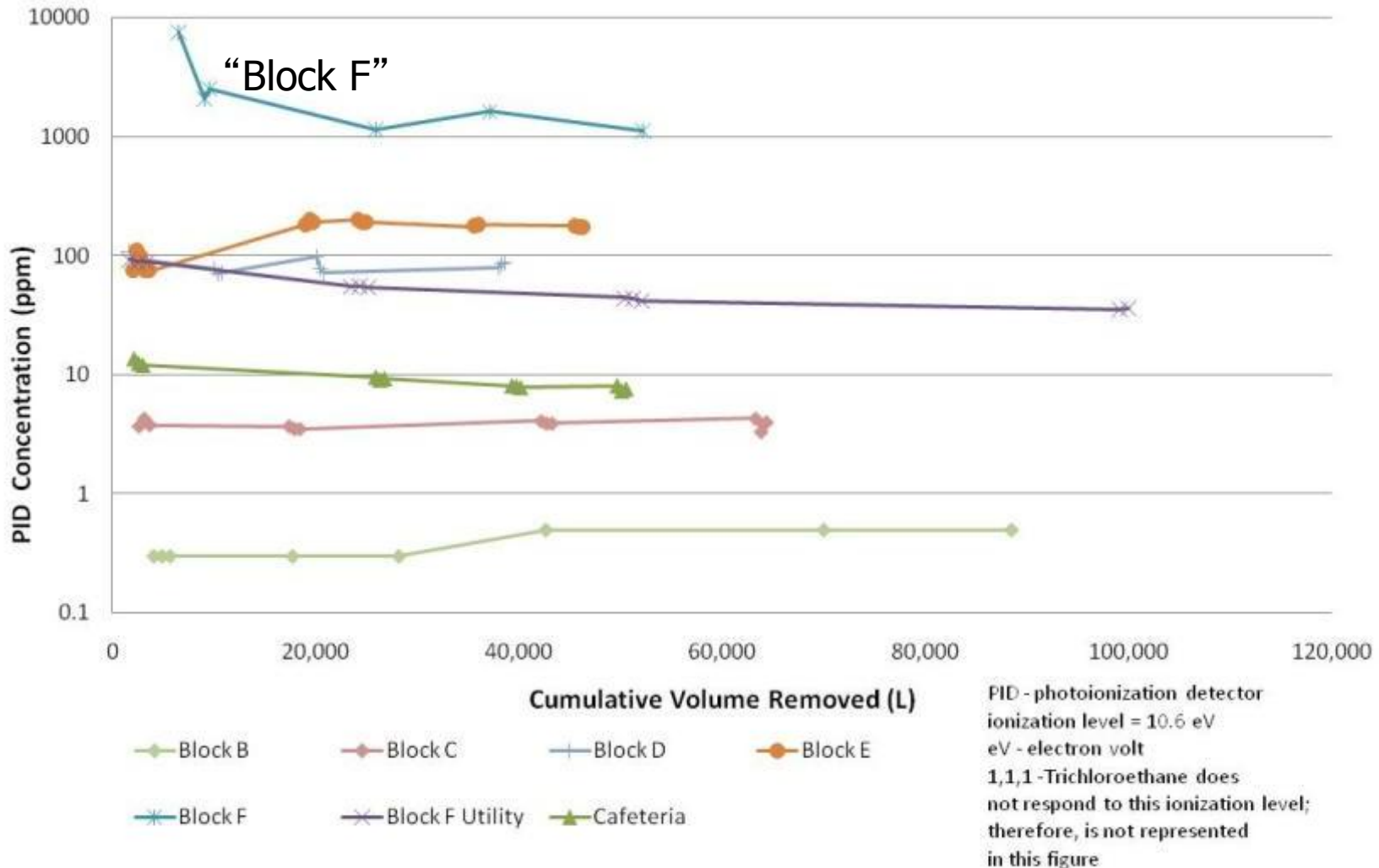




17.08.2009



PID versus Volume Purged



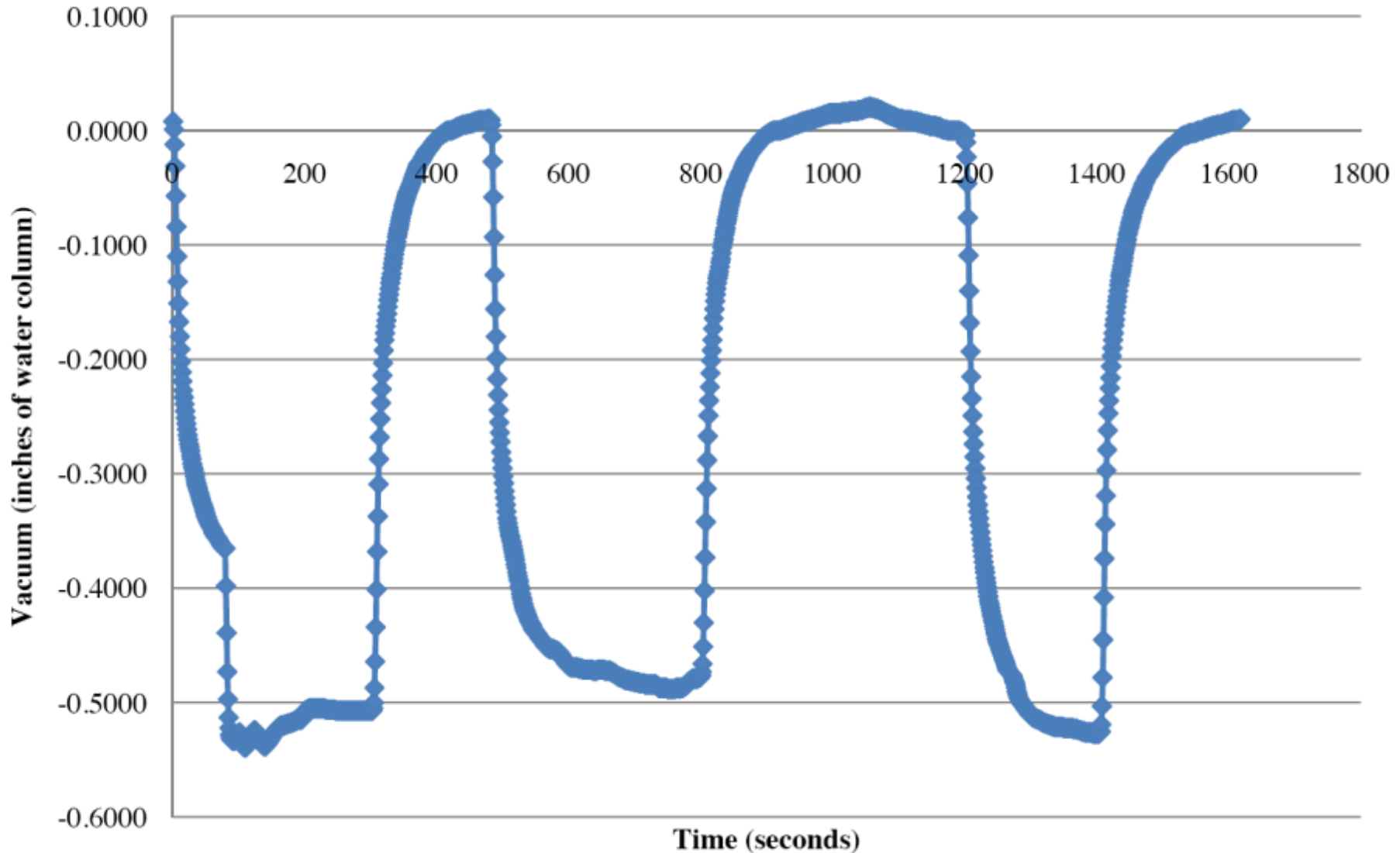
What happened to all the spatial variability in sub-slab data?

Pressure Transducers / Data Loggers

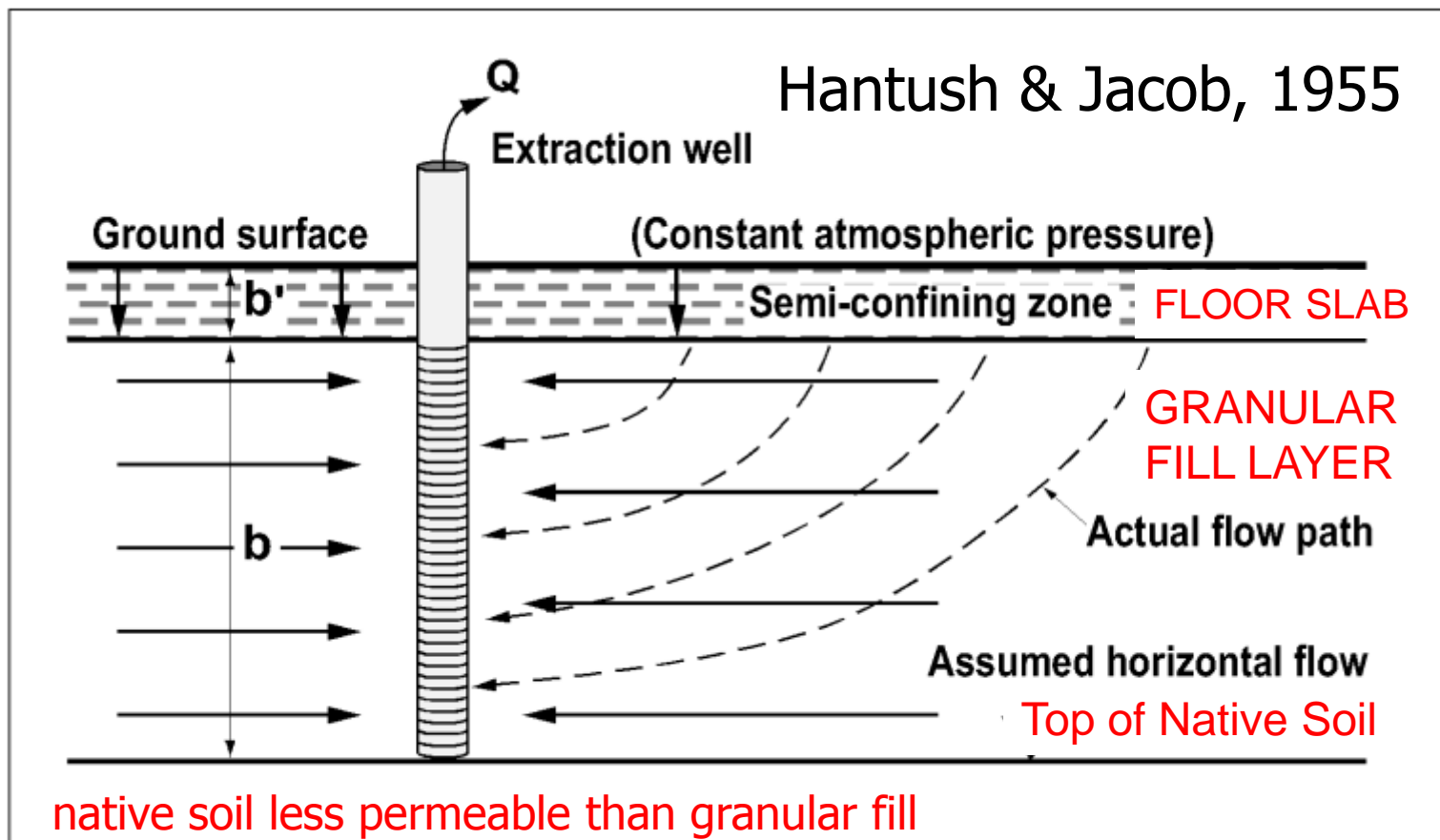


In just a few minutes, you've got "pump-test" data

Drawdown and Recovery



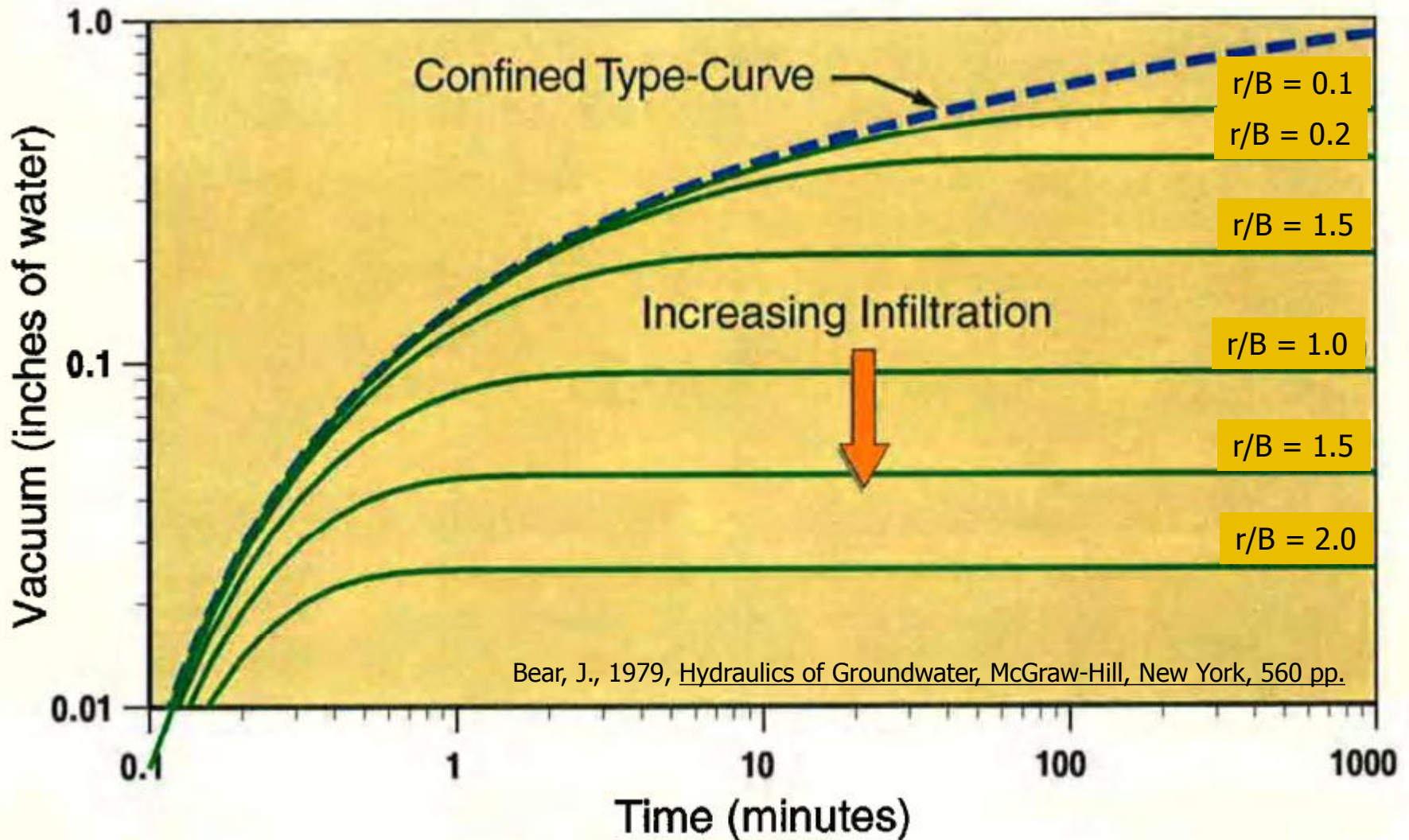
Leaky Aquifer Model for SSD



Thrupp, G.A., Gallinatti, J.D., Johnson, K.A., 1996, "Tools to Improve Models for Design and Assessment of Soil Vapor Extraction Systems", in *Subsurface Fluid-flow (Groundwater and Vadose Zone) Modeling*, ASTM STP 1288, Joseph D. Ritchey and James O. Rambaugh, Eds., American Society for Testing and Materials, Philadelphia. pp 268-2

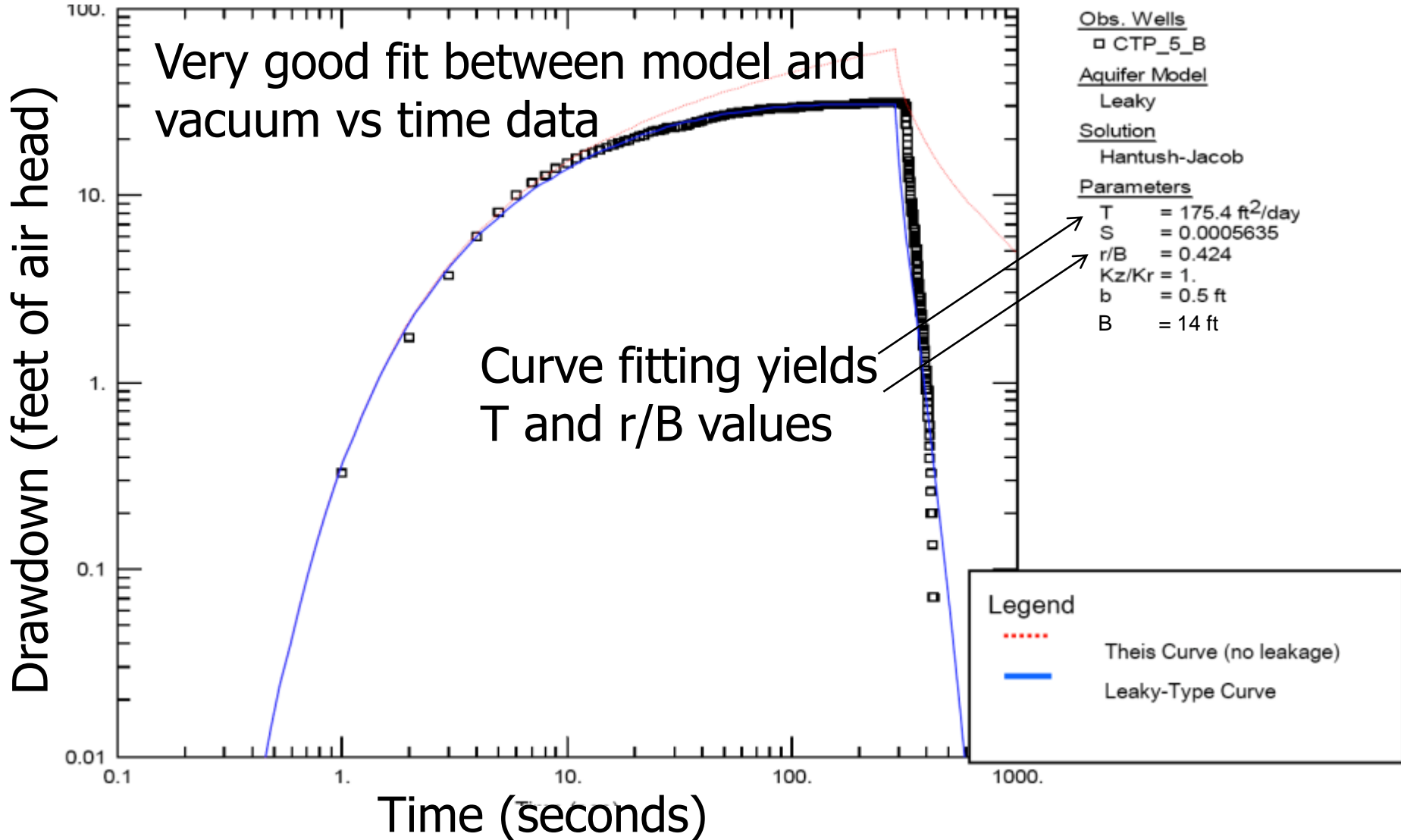
Massman, J. W., 1989, "Applying Groundwater Flow Models to Vapor Extraction System Design," *J. of Environmental Engineering*, Vol. 115, No. 1, pp. 129-149.

Leaky Aquifer Type-Curves



Hantush Jacob Model Fit

Vacuum measurements 6 feet from extraction point



Floor Slab Conductivity

$$K' = \frac{T b'}{B^2}$$

K' = vertical pneumatic conductivity of the floor slab [L/t]

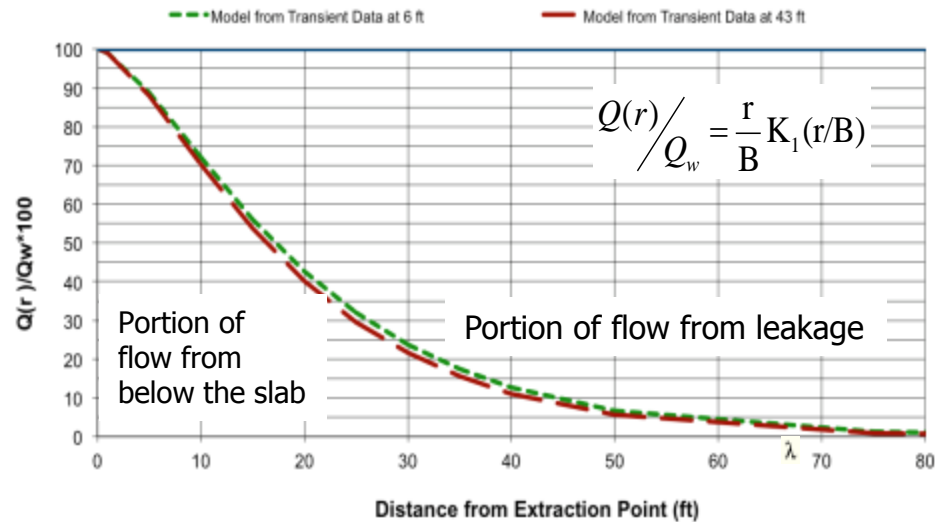
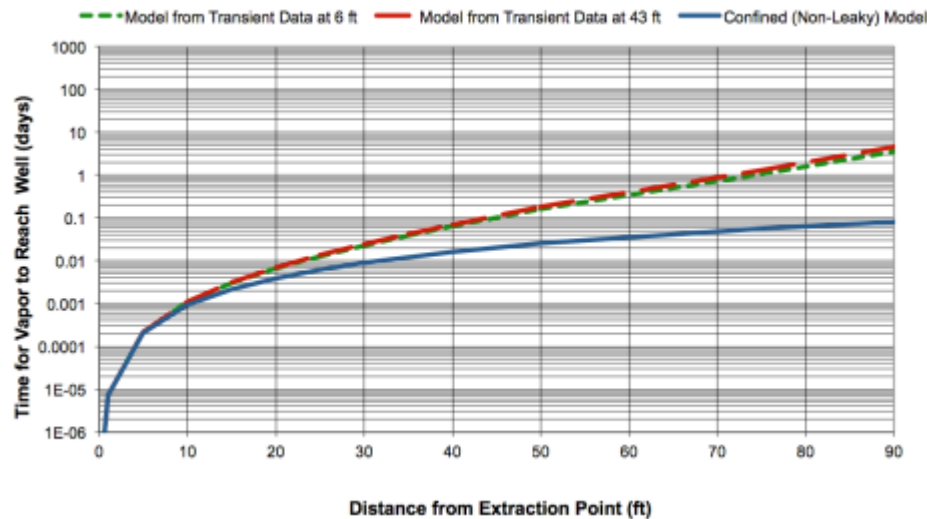
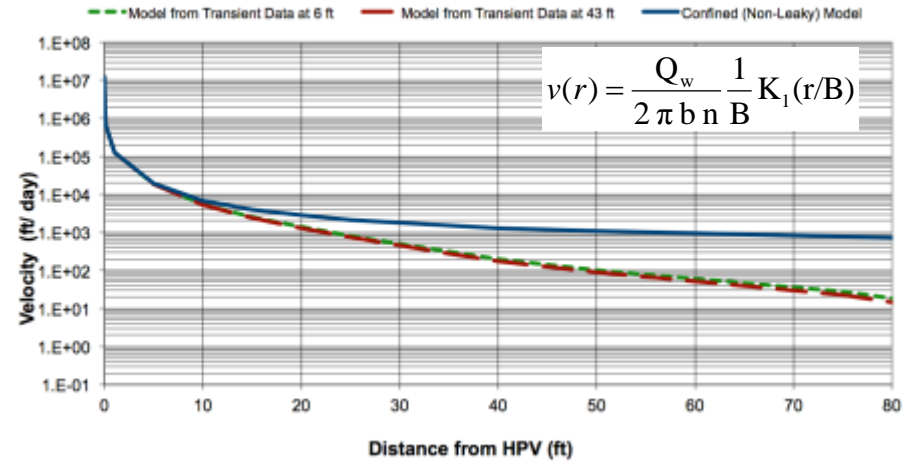
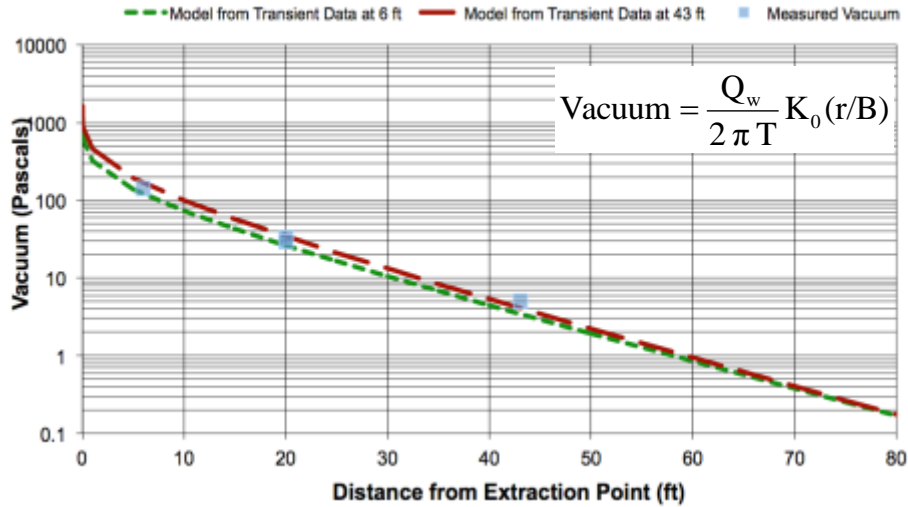
b' = floor slab thickness [L], easily measured

T = transmissivity [L²/t], a direct output of the model

B = leakance [L], also output from the model

Therefore, if you know b' (slab thickness), you can calculate the vertical pneumatic conductivity of the slab

Calculations

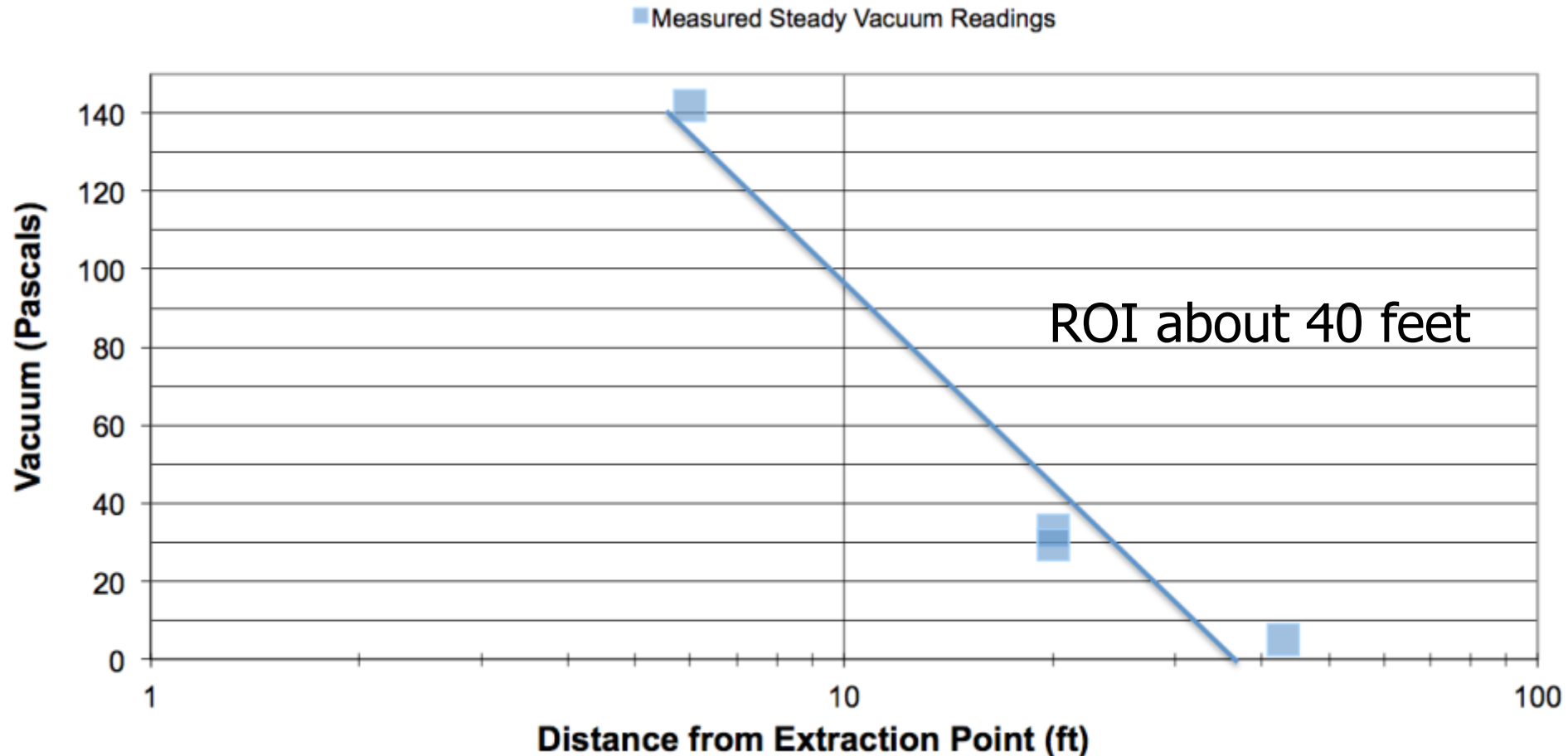


Summary

- Point measurements in space and time are variable
 - We assess risks for a 25 to 30-year exposure scenario
 - Data should be representative and cost effective
- Long-term samples minimize temporal variability
- Large volume samples minimize spatial variability
 - Easy to add pneumatic testing and get design data
- Passive sampling can now give quantitative soil vapor data
- Regulatory acceptance is progressing

Conventional Radius of Influence

Case Study: 100,000 ft² commercial building, slab-on-grade





Questions/Comments?



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