



Integrity ★ Service ★ Excellence

Energy Conversion and Combustion Sciences

Date: 08 March 2013

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AFOSR/RTE**

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Report Documentation Page

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2013 AFOSR SPRING REVIEW



NAME: **CHIPING LI**

BRIEF DESCRIPTION OF PORTFOLIO:

Meet *Basic Combustion Challenges* with *New Approaches*
Explore *New Energy Conversion Opportunities* for
Next Generation Air Force Propulsion Systems of Game-Changing Efficiency and Operability

Key Portfolio Attributes:

- Understand *Fundamentals* in Realms of *Air Force* Interests (understand the nature as it is)
- Quantify *Rate-Controlling* Processes and Scales in Multi-Physics, Multi-Scale Phenomena (find ways to control complex phenomena)

LIST SUB-AREAS IN PORTFOLIO:

- 1. Combustion Chemistry** (underlying chemistry, new approaches -- working with Drs. Berman/RTE, ARO & DOE/BES)
- 2. Turbulent Flame Properties/Models** (nonlinear flow-chemistry interaction, new thinking/tools)
- 3. Combustion Numerics** (new tools -- collaborating with Dr. Fariba/RTA)
- 4. Combustion Diagnostics** (new tools -- collaborating with Dr. Parra/RTB)
- 5. Game-Changing Energy Conversion Concepts** (new opportunities, Drs. Berman/RTE, Luginsland/RTB, & ONR)

Sub-areas are multi-disciplinary: collaboration with other POs & Agencies are essential.

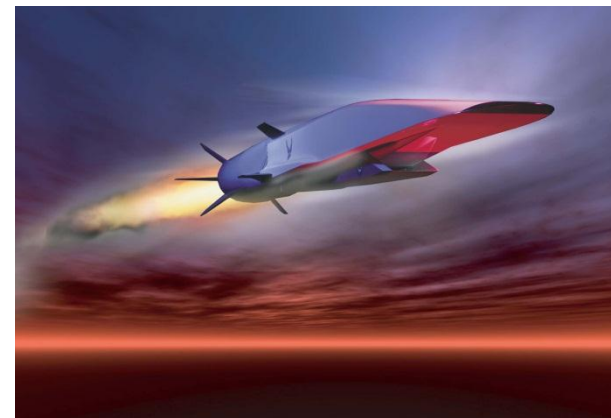
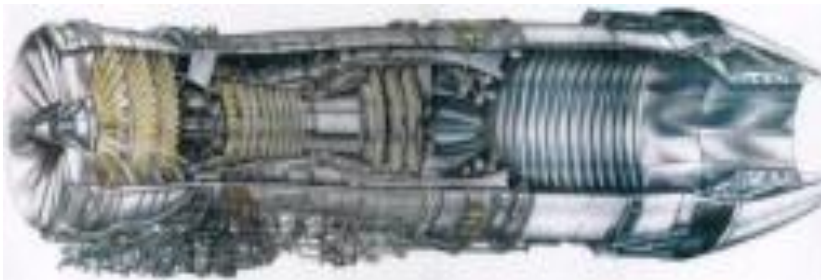
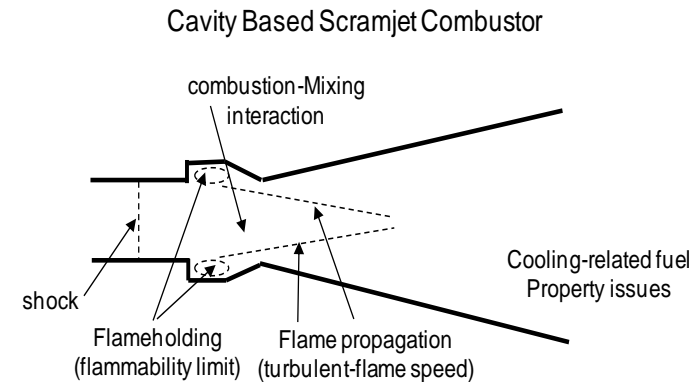
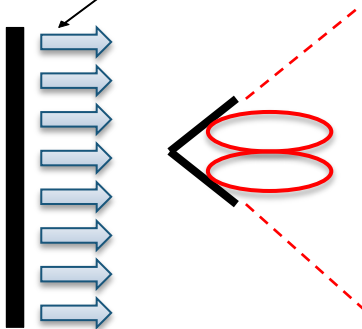


Combustion – the Central Process in Converting Chemical to Mechanical Energy in AF Propulsion Systems



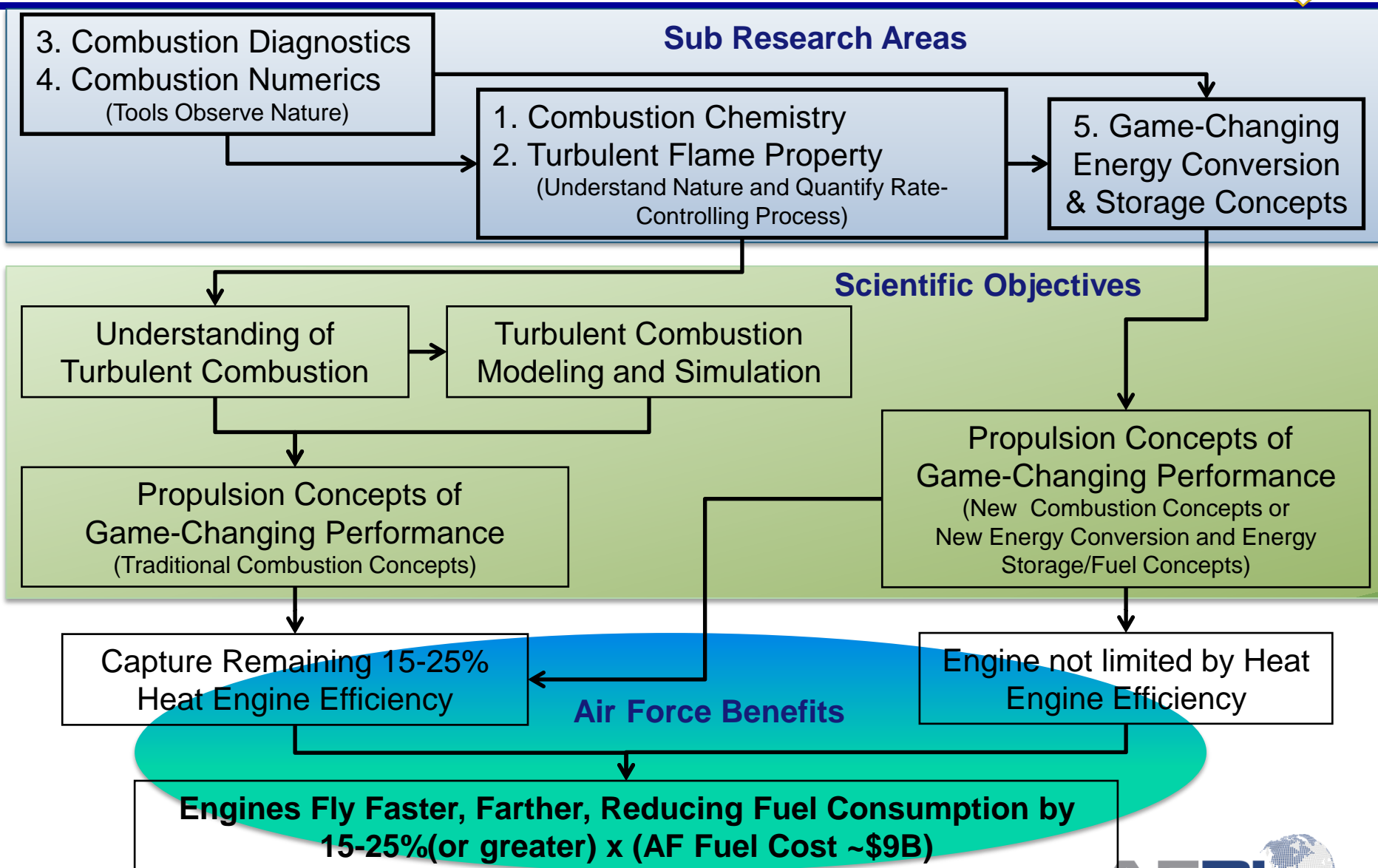
- **Most Important** determining factors of operability and performance;
- **Least Understood** areas in basic combustion research, with large uncertainties;
- Confluence of “grand-old” fundamental science challenges, immediate needs and long-term interests.

fuel Injection





Portfolio Logic and Strategy





Portfolio Directions/Trends



1. Combustion Chemistry

- Reaction-pathway centric approaches (based on ab initial methods and exp. data) ↑
- Traditional detailed reaction-rate-constant centric approaches ↓

2. Turbulent Flame Properties/Models

- Turbulent flame experiments in realm of AF interests (high-Re, compressible) ↑
- Laminar and weakly turbulent flame experiments ↓
- Scale interaction models based on DNS, experimental data and new math approach ↑
- Models based on assumptions not directly verifiable by experiments ↓

3. Combustion related Numerical Techniques

- Coupled simulation-experiment approaches ↑
- Computational methods for studying stochastic pathways ↑
- Computational methods for reduction of “large” detailed combustion chemistry models ↓

4. Combustion Diagnostics →

5. Game-Changing Energy Conversion Concepts

- New combustion concepts →
- Direct/partially direct conversion from chemical energy to mechanical energy ↗
- New energy-storage/ fuel concepts for propulsion application ↗



Coordination with Other Agencies



1. **Strong collaboration is continuously being forged in following areas:**
 - Diagnostics (Mainly DoE, NASA)
 - Numerical (DoE, NASA, ARO)
 - Combustion Chemistry (DoE, ARO, NSF)
 - Innovative Combustion Concept (ONR, ARO)

2. **Dividing problems and condition areas according to each interests:**
 - **AFOSR combustion portfolio:**
 - **Turbulence combustion area: Air-Force relevant realms, i.e. compressible, high-Re conditions for propulsion applications**
 - **Combustion Chemistry: Reaction-pathway centric approaches**
 - DOE -- a well funded combustion program focusing on basic energy research:
 - turbulence combustion area: ground-base energy systems and auto-engine types of applications at relatively low-speed and low-Re conditions (TNF etc.)
 - combustion chemistry: large, detailed reaction-rate-constant centric approach
 - NASA -- a modest combustion program focusing:
 - "Very-high" speed (space access) region
 - Overlapping interests and close coordination with AF programs (scramjet, rockets etc.).
 - NSF -- a modest combustion program:
 - Covers broad ranges of combustion problems

3. **Multi-Agency Coordinate Committee of Combustion Research (MACCCR)**
 - Functioning well and its positive roles will continue

Multi-Agency Collaboration Benefits Every One

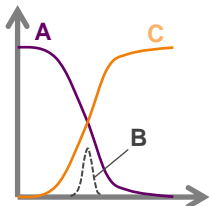
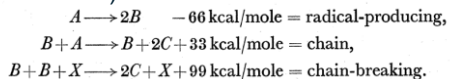


Combustion Chemistry a New Direction



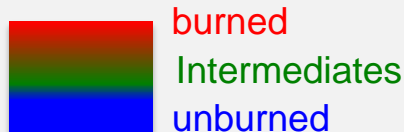
Combustion Chemistry: History and Recent Progress

Spalding used 3-step "fictitious" reactions to describe the global reaction kinetics of a flame (*Philos. Trans. Royal Soc. London A* 1956)



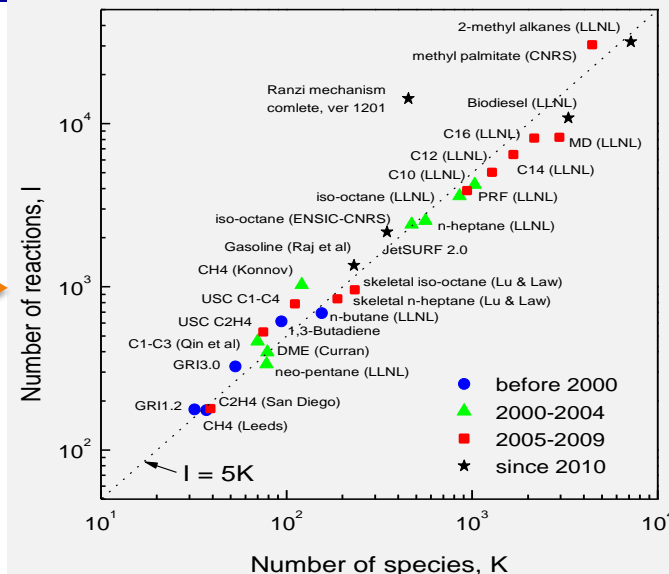
add physics

Late 60s-early 70s
remove empiricism by using ~10-20 step elementary chemistry



Graham Dixon-Lewis

add physics
or just follow
the formula?

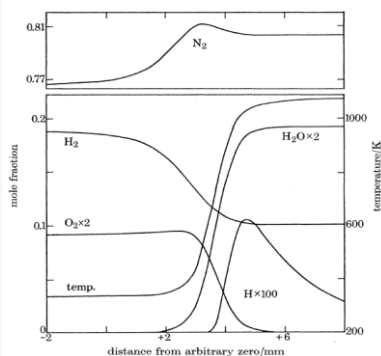


1950s'

Three-step, global chemistry with detailed transport



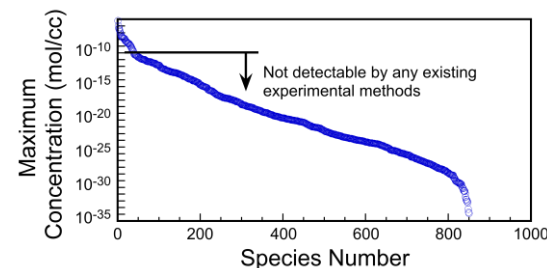
D. Brian Spalding



Dixon-Lewis' calculation of the structure of a hydrogen flame, including intermediate radical concentrations, using detailed chemistry (*Proc. R. Soc. Lond. A* 1970)

Today: following the formula, an extrapolation of Dixon-Lewis' work, but this is sure not what Dixon-Lewis had in mind.

- $O(10^4)$ species and $O(10^5)$ reactions with rates and pathways unverifiable.
- Back to the empirical past: > 90% species considered are not detectable by any experimental means.

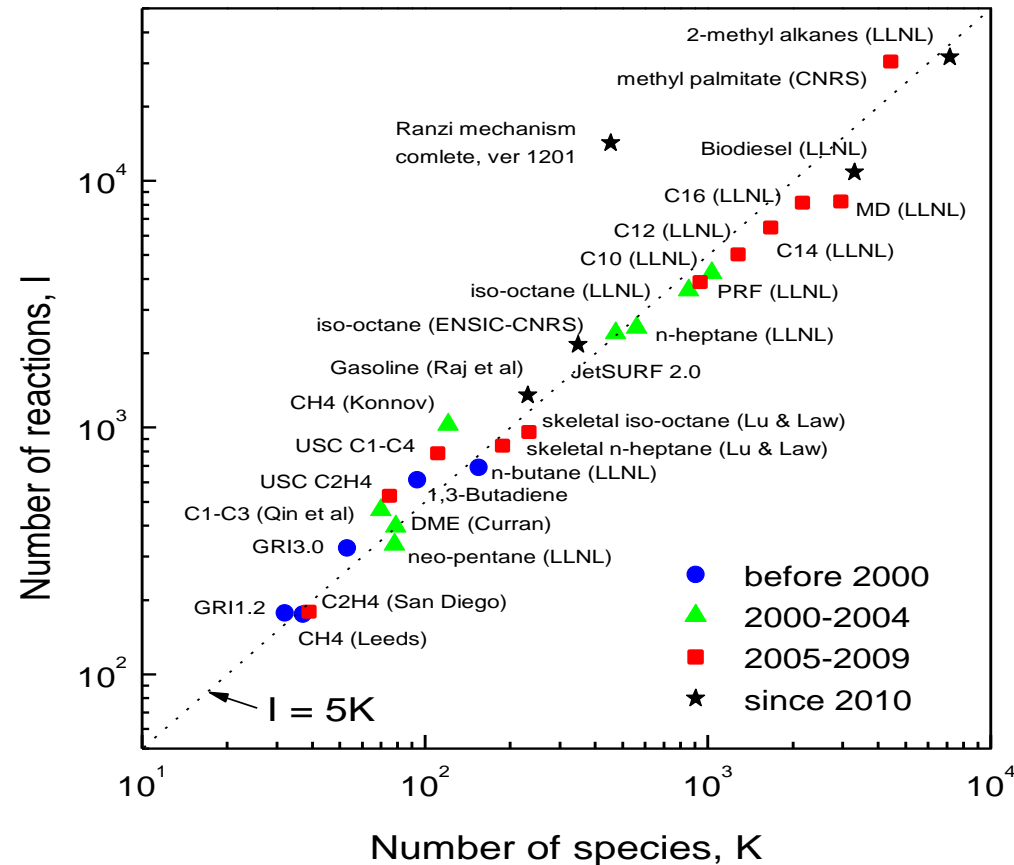


Ranked maximum concentrations of species computed as an initial value problem using a typical reaction mechanism ($T_0 = 1400 \text{ K}$ and $P = 1 \text{ atm}$)





Uncertainty vs. Model Size



The larger, the better – **Maybe Not**
Does the model uncertainty reduce as the model size grows – **Not Necessarily**

- Uncertainties of Model inputs
- Uncertainties with model complexity
- Relationship between model size and its uncertainty

Find out optimal model size for minimal uncertainty --- new direction in UQ for the combustion chemistry model



Two Distinctive, Complementary Approaches for Real HC Fuel



Traditional/State of Art “Rate Constants Centric”

- INCLUDE:** combinatory approaches (up to $\sim 10^4$ - 10^5 reaction steps for common HC fuels)
- ESTIMATE:** rate constants for most and calculate and measure for some
- SELECT:** through sensitivity analysis, targeting \sim order of 10^2 reaction steps

Note1: large uncertainty in steps 1 & 2

Note2: There is a confusion between approximate natures of those large, complex reaction sets and exact nature of the human Genome set.

In the combustion process, reactions follow uncertain **pathways**, step by step.

$$k(T) = AT^n e^{-Ea/RT}$$

At each step in traditional Arrhenius model, the reaction rate is controlled by **rate constants**.

No.	reaction ^b	$k = AT^n \exp(-E/RT)^c$			references/ comments	
		A	n	E		
<i>Reactions of propene</i>						
1	$aC_3H_5 + H (+M) = C_3H_6 (+M)$	2.00×10^{14}			k_{ex}, d	
		1.33×10^{60}	-12.0	5968	k_0	
		$a=0.020$	$T^{**}=1097$	$T^*=1097$	$T^{**}=6860$	e
2	$CH_3 + C_2H_3 (+M) = C_3H_6 (+M)$	2.50×10^{13}			k_{ex}, f	
		4.27×10^{58}	-11.94	9770	k_0	
		$a=0.175$	$T^{**}=1341$	$T^*=60000$	$T^{**}=10140$	e
3	$C_3H_6 + H = C_3H_5 + CH_3$	1.60×10^{22}	-2.39	11180	1 atm, g	
4	$C_3H_6 + H = aC_3H_5 + H_2$	1.70×10^{05}	2.5	2490	[33]	
5	$C_3H_6 + H = CH_3CCH_2 + H_2$	4.00×10^{05}	2.5	9790	[33]	
6	$C_3H_6 + O = CH_2CO + CH_3 + H$	1.20×10^{08}	1.65	327	[33]	
7	$C_3H_6 + O = C_2H_5 + HCO$	3.50×10^{07}	1.65	-972	[33]	
8	$C_3H_6 + O = aC_3H_5 + OH$	1.80×10^{11}	0.7	5880	[33]	
9	$C_3H_6 + O = CH_3CCH_2 + OH$	6.00×10^{10}	0.7	7630	[33]	
10	$C_3H_6 + OH = aC_3H_5 + H_2O$	3.10×10^{06}	2.0	-298	[33]	
11	$C_3H_6 + OH = CH_3CCH_2 + H_2O$	1.10×10^{06}	2.0	1450	[33]	
12	$C_3H_6 + HO_2 = aC_3H_5 + H_2O_2$	9.60×10^{03}	2.6	13910	[33]	
13	$C_3H_6 + CH_3 = aC_3H_5 + CH_4$	2.20×10^{00}	3.5	5675	[33]	
14	$C_3H_6 + CH_3 = CH_3CCH_2 + CH_4$	8.40×10^{-01}	3.5	11660	[33]	

New/Start Exploring “Follow the Pathway”

- SELECT:** identify important pathways following PES
- INCLUDE:** only include most relevant ones – targeting no more than \sim order of 10^2 reaction steps
- OBTAIN:** rate constants from experimental measure and ab initio calculations

Note1: understanding of initial fuel break-up is most important

Note2: made possible for recent develop in diagnostics and ab. initio chemistry calculation method

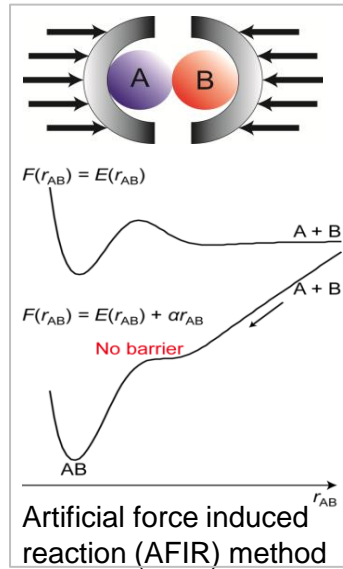
↓ **? – Just Start Exploring**

Combustion Chemistry Models of limited reaction steps with acceptable uncertainties, usable for reactive CFD tools

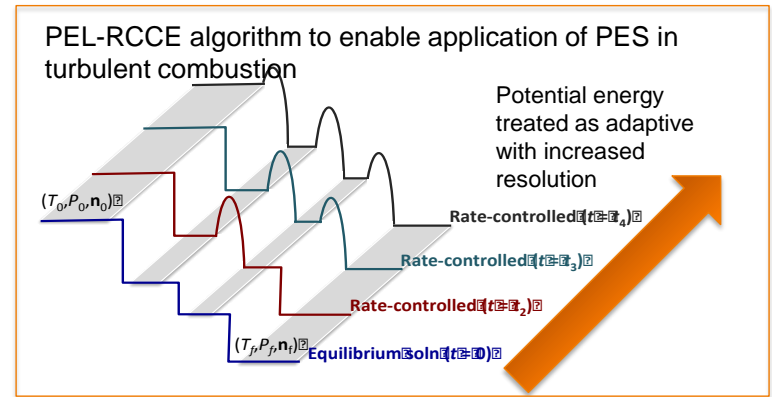
???? – Have Explored for more than forty years



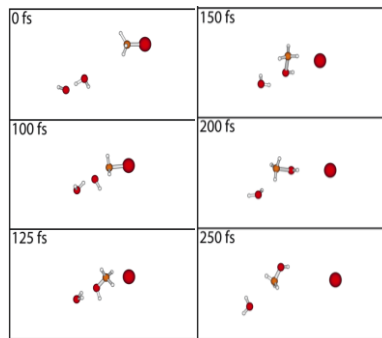
Ab. Initio Methods to Identify Key Pathways



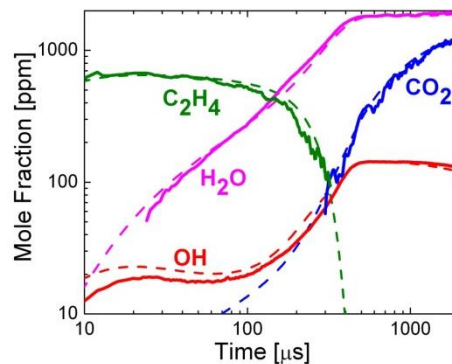
automated
 PES/PEL
 generation



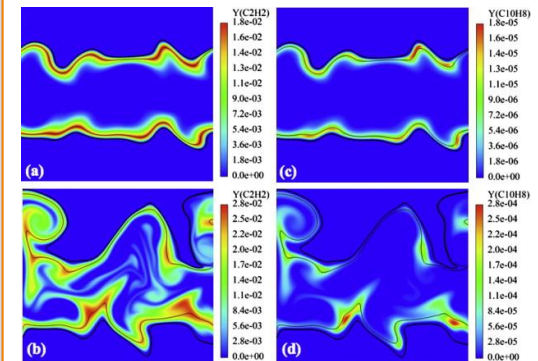
Chemical Dynamics



Experimental Validation



Combustion





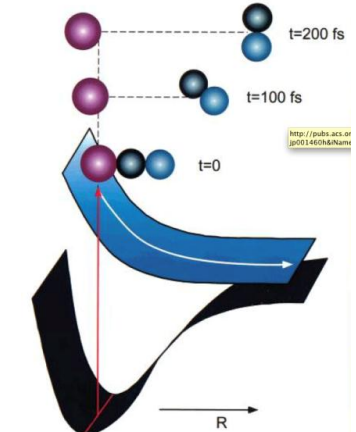
Diagnostics in Combustion Chemistry



Femtochemistry



Zewail, JPC A 104, 5660 (2000)



Heart beat

Picture refresh in LED TV

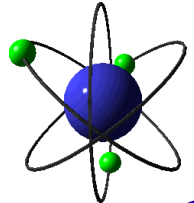
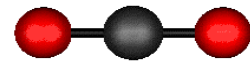
Period of AM radio

Computer processor speed

Molecular rotation

Molecular vibration

Electronic motion



1 second

10⁻³ milisec

10⁻⁶ microsec

10⁻⁹ nanosec

10⁻¹² picosec

10⁻¹⁵ femtosec

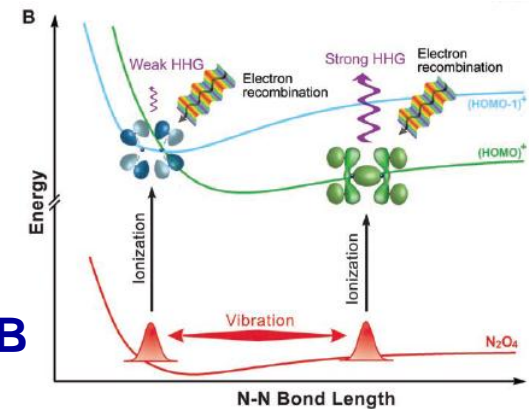
10⁻¹⁸ attosec

10⁻²¹ zeptosec

Attosecond physics/chemistry



Murnane, Science 322, 1207 (2008)





Combustion Chemistry: Where We Are



With recent developments in combustion diagnostics (especially ultra-fast laser based diagnostics) and ab. initio chemistry methods, we have unprecedented opportunities in combustion chemistry, --- AFOSR is leading the charge

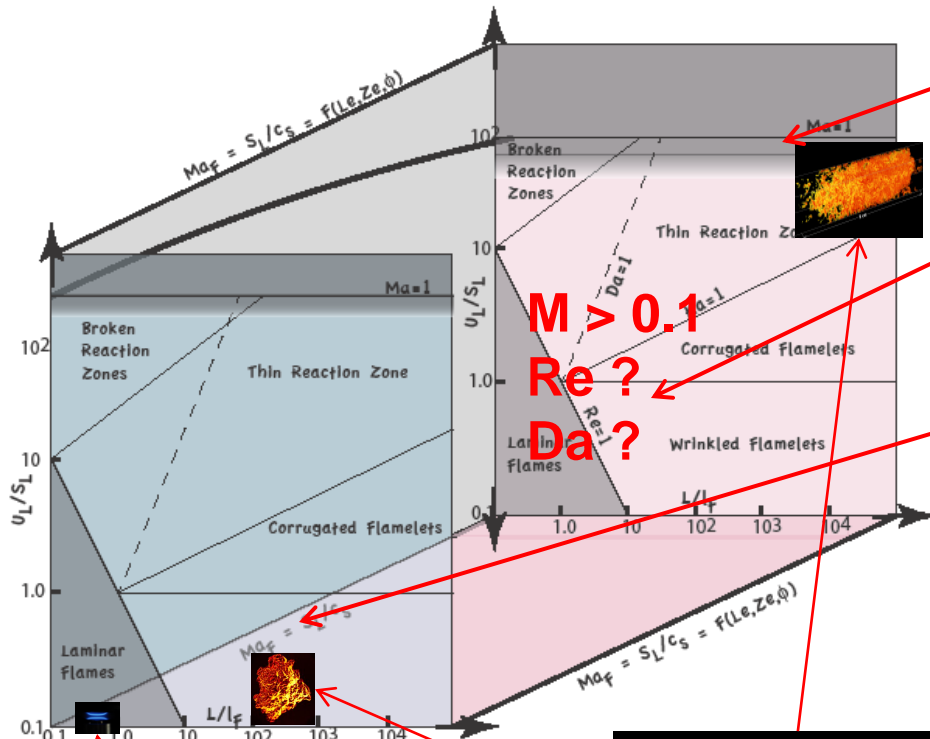
leading to usable models with acceptable uncertainty to revolutionize Air Force propulsion system development.



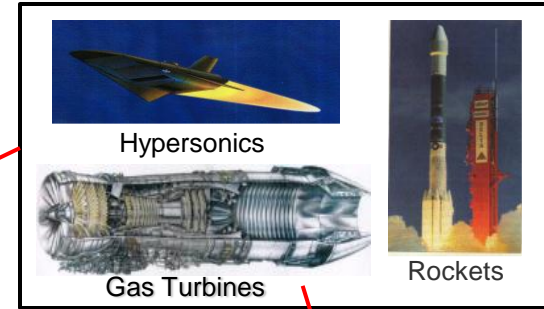
Turbulent Flame Property in Air-Force-Relevant Realms



Turbulence Combustion: Fundamental Structures, Critical Scales and Relevant Conditions



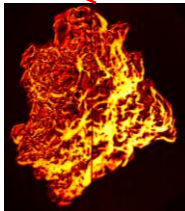
PGC



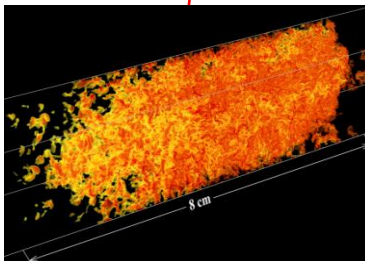
Auto Engines



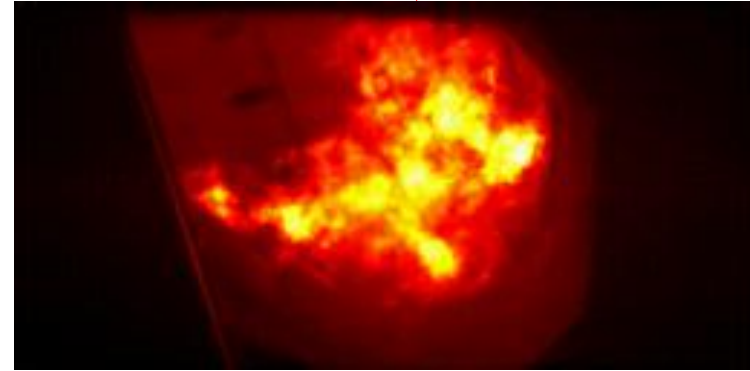
laminar flame



wrinkled flame ball



Hi-M, Hi-Re (flame brushes)



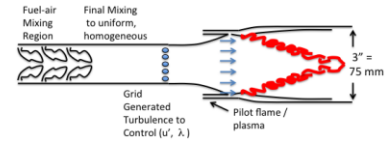
(1) Little Understanding and Data Available at AF Relevant Compressible, High-Re Conditions;
 (2) Needs for Better Definition of Re-Conditions in Regions of Interests



High-Re, Compressible Turbulence Flame Experiments at AF Relevant Condition Ranges

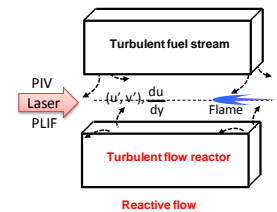
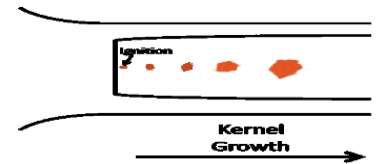


- Focus on key combustion properties and characteristics such as:
 - **Flame propagation,**
 - **Flammability limit**
 - **Combustion instability**
- Multi-phase conditions **applicable** to Air Force propulsion systems
- Made possible by diagnostics developed by this portfolio up to date



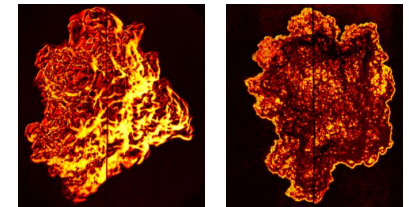
Key Requirements (Experimental Data Objectives):

1. **Understanding** the above key combustion phenomena and characteristics;
2. **Quantifying rate-controlling processes and scales** that govern those phenomena and characteristics;
3. Developing and validating as directly as possible **basic model assumptions**
4. Controlling and quantifying turbulence properties are **essential**.



Proposals are being considered and funded for:

- Defining relevant conditions and Studying Critical Scales (1 funded in FY12)
- Relevant Experiments in different configurations (4 funded in FY12)



Understanding Nature from Observation and Data

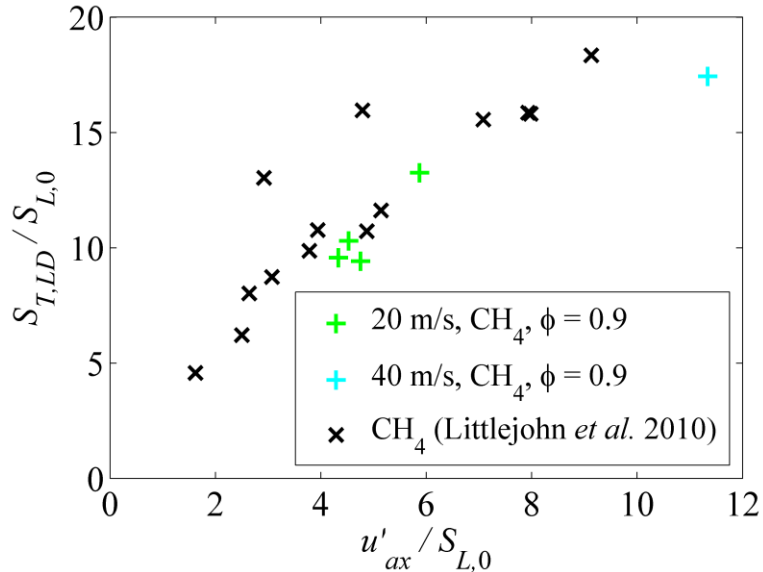


More Turbulent Flame Experiments

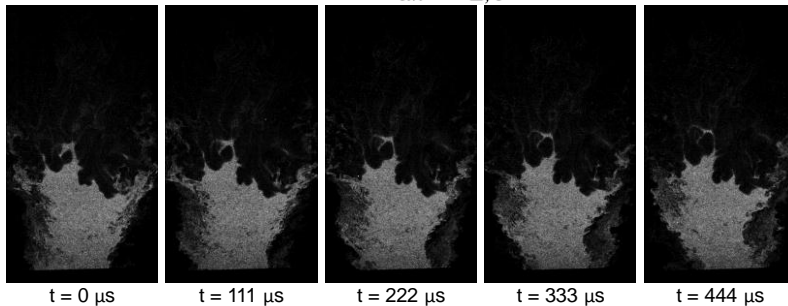
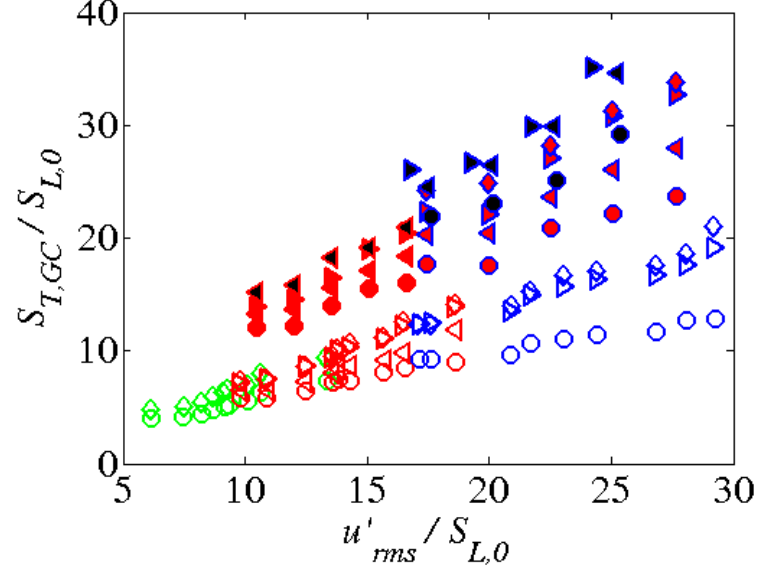


Speed and Free Propagating Turbulent Flame and Swirl Flame: (PI: Lieuwen, Georgia Tech)

swirl burner (jet-engine similar) data



sister program Bunsen burner data



Major data set obtained over range of velocities (4-70 m/s), pressures (1-20 atm), turbulence intensities, fuel compositions. S_T/S_L can be $\gg 100$.

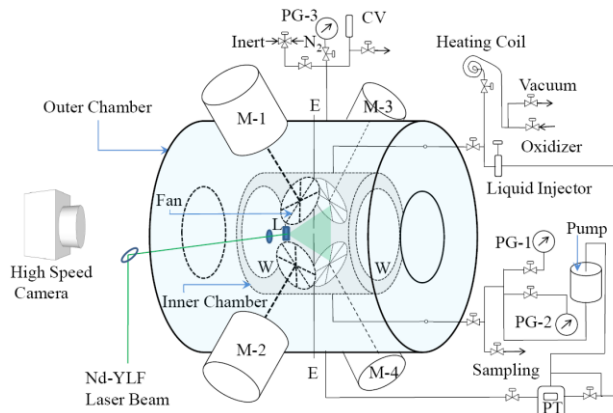
No upper limits observed in turbulence flame speed!!!
Much more efficient and compacted combustor can be designed.



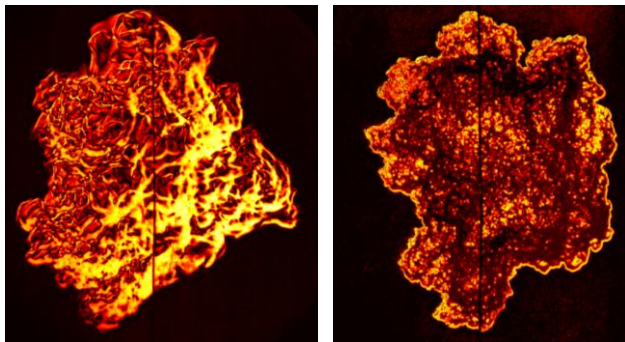
More Turbulent Flame Experiments



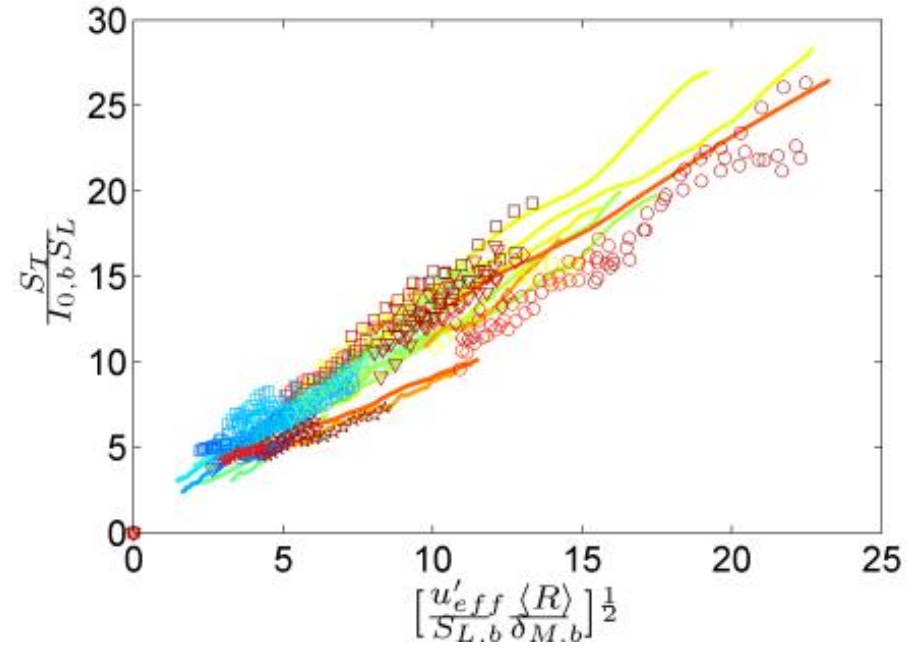
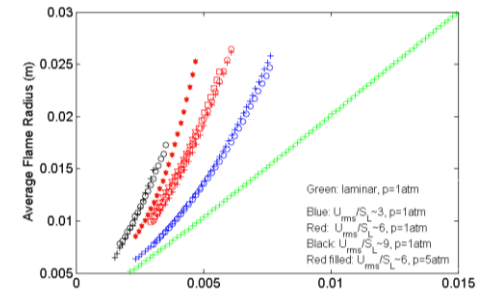
Flame Speed and Self-similar Propagation of Turbulent Premixed Flames: (PI: Law, Princeton)



CV: Check Valve, PG: Pressure Gauge, PT: Pressure Transducer, M: Fan Motor, L: Cylindrical Lens, E: Electrodes, W: Quartz Window



Pressure = 5 atm
 Pressure = 50 atm
 Scanned images of turbulent premixed CH₄ air flames ($\phi=0.9, Le=1$) at same u_{rms}



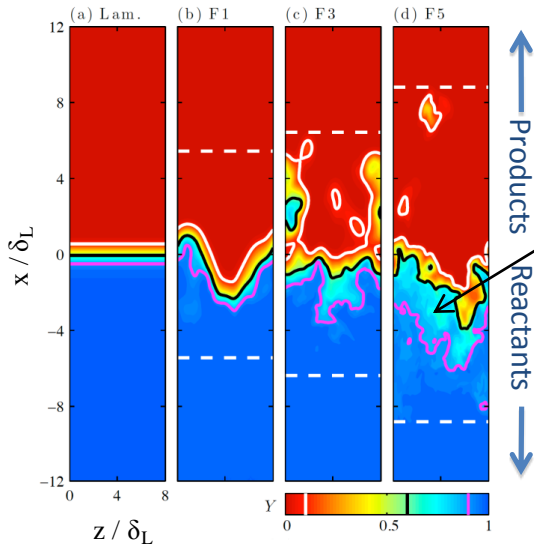
Turbulence flame speed can be scaled, at least partially understood and modeled.



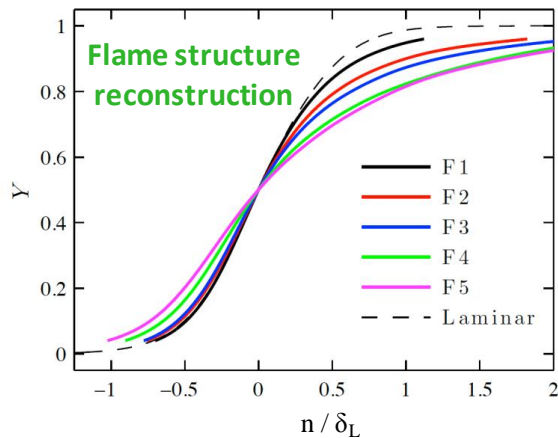
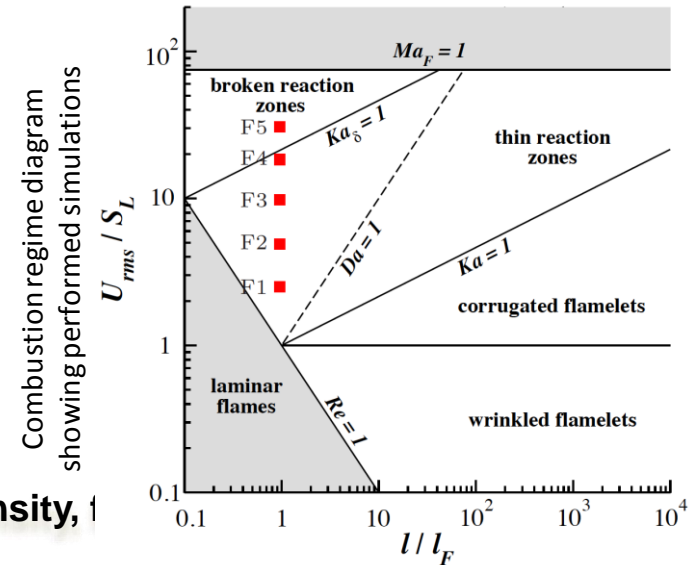
Turbulent Flame Interactions

Turbulent Flame Structure: (PIs: Oran, Poludnenko & Hamlington NRL)

Domain cross-sections



thicken pre-heating (pyrolysis) zone

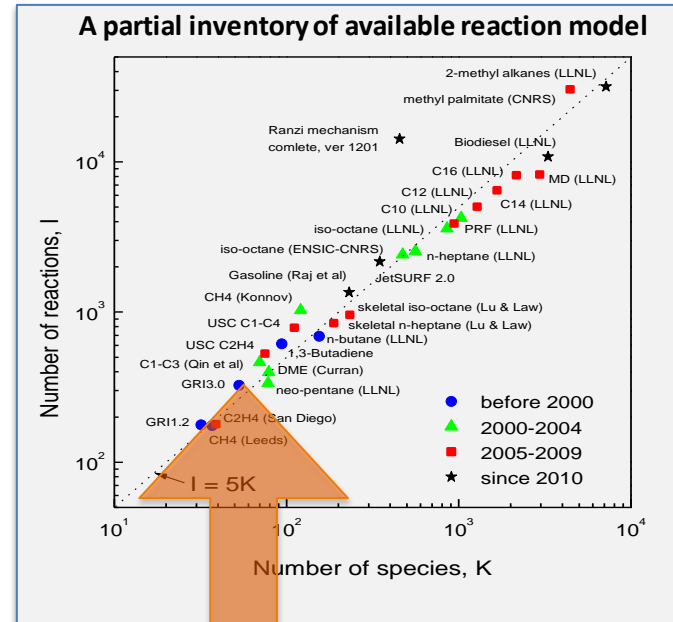
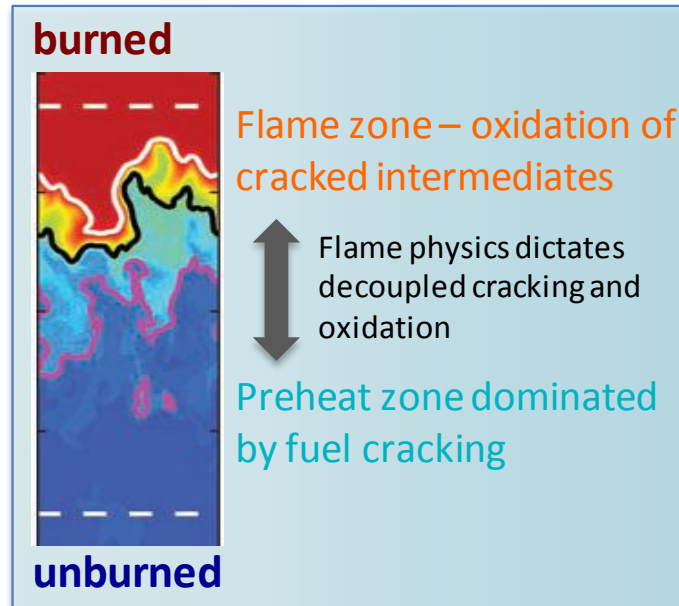


- small turbulent intensity, $l / l_F < 1$
- large intensities, preheat (pyrolysis) zone broadened and reaction zone virtually unaffected
- robustness of reaction zone – turbulent diffusion suppressed in reaction zone by heat release (suppressed small scales)
- tangential strain rate thins flame

We may be able to understand turbulent flame structure after all...



Turbulence-Chemistry Interaction



Stage 1: quantify the pyrolysis process in the thickened pre-heating zone, which leads to ~6 c1-c4 molecular fragments

Stage 2: combined with the c1-c4 combustion chemistry that has been well characterized.

Turbulence-Chemistry Interaction: (PI: Wang, USC)

Turbulent Pre-Heating (Pyrolysis) Zone Makes the Chemistry Model Simpler....



Turbulent Flame Property: Where We Are



With recent developments in combustion diagnostics and numerical simulation for the reactive flow, we begun to observe and understand fundamental attributes of the turbulent flame in Air-Force-relevant realms,

Leading to:

- Quantify of interactions among different scales
- Establish of usable turbulent combustion models
- With acceptable uncertainty to revolutionized Air Force propulsion system development.

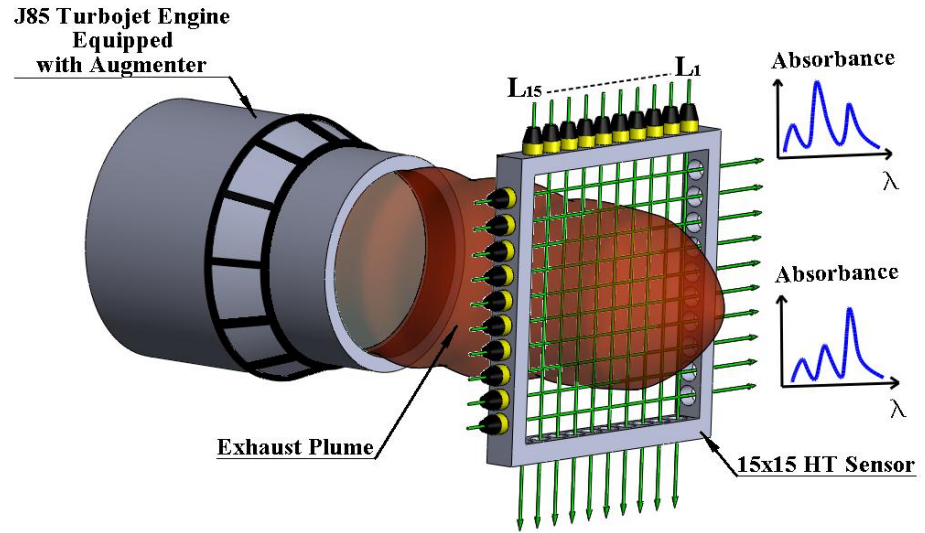


Examples of Continuous Transition



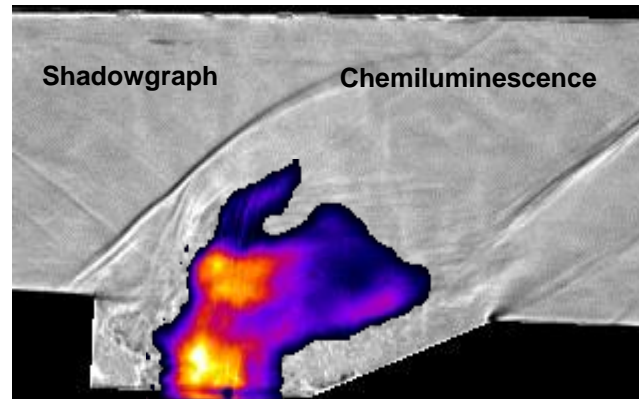
Diagnostics

State-of-Art Optical Fibers, Probes, Single-Beam Techniques



New Ignition Technique

Basic Combustion





Closing Statements



After a year, the portfolio is taking shape.

Supported projects have started showing very encouraging results.

More to come, stay tuned.