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14. ABSTRACT : Direct reinforcement of composite materials is typically promoted through use of high aspect ratio, stiff inclusions whose interface is modified to mediate stress transfer. Here a markedly different mechanism, indirect reinforcement, is achieved through use of weakly-interacting, spherical nanoparticles. Mechanical property enhancement is achieved by nanoparticles introducing density fluctuations to affect the polymer packing near their interface. This results in two phenomena; first a lower bulk modulus is found meaning the material becomes more				
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## Report Title

Nanocomposites

### ABSTRACT

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We have considered many new nanoparticle systems and have had to take a very careful route to ensure the results we have obtained are not spurious. We reproduced most previous results conducted by a previous student in the group since some results were found to be at fault and are sure what we have now measured are correct. Several manuscripts are now in preparation.

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J. J. Wie	1.00	
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**Sub Contractors (DD882)**

**Inventions (DD882)**

**Scientific Progress**

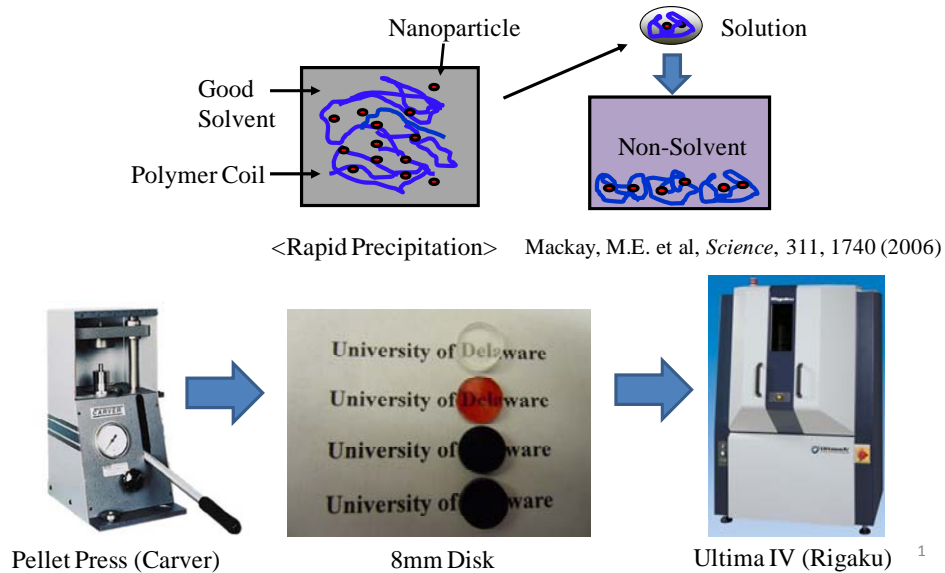
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**Technology Transfer**

**Summary:** Direct reinforcement of composite materials is typically promoted through use of high aspect ratio, stiff inclusions whose interface is modified to mediate stress transfer. Here a markedly different mechanism, indirect reinforcement, is achieved through use of weakly-interacting, spherical nanoparticles. Mechanical property enhancement is achieved by nanoparticles introducing density fluctuations to affect the polymer packing near their interface. This results in two phenomena; first a lower bulk modulus is found meaning the material becomes more compressible. Secondly, the tensile modulus increases. The first phenomenon occurs (we think) because the nanoparticle is surrounded by a low density region. The second is due to high density regions surrounding the low density promoting a higher tensile modulus.

We have considered many new nanoparticle systems and have had to take a very careful route to ensure the results we have obtained are not spurious. We reproduced most previous results conducted by a previous student in the group since some results were found to be at fault and are sure what we have now measured are correct. Several manuscripts are now in preparation.

**Discussion:** Dispersion of the nanoparticles into the polymer is a critical step. Figure 1 below shows how we did this. The nanoparticles and polymer were dissolved in a good solvent then precipitated into a mutual non-solvent and the precipitant was dried for a week at 40°C. An 8 mm diameter disk was formed under vacuum in a modified FT-IR pellet press and then characterization could be performed, such as with the Rigaku Ultima IV Small Angle X-Ray Scattering (SAXS) instrument in our laboratory.



**Figure 1:** Description of the dispersion process required to make a homogeneous blend of nanoparticles in a polymer.

We used the SAXS data to determine the degree of nanoparticle dispersion in our systems with data given in Figure 2. The overall scattering intensity ( $I$ ) as a function of wave vector ( $q$ ) is consistent with well dispersed blends for both an industrial 10 nm magnetite nanoparticle system (EMG1300, FerroTech) and smaller 5 nm magnetite system synthesized in the group of Prof. J. Pyun at the University of Arizona.

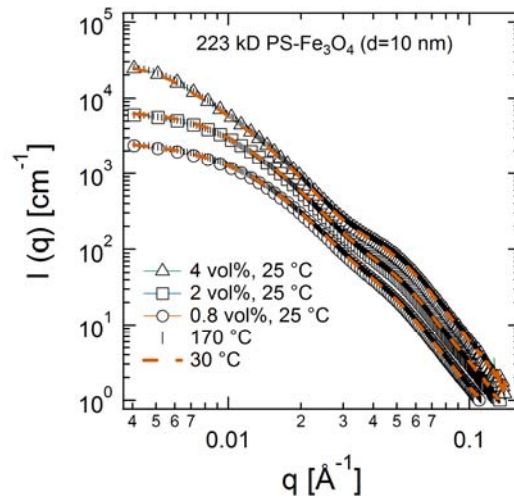
SAXS measurements were carried out to estimate the interaction between particles and thermodynamic stability in polymer blends using Zimm plots. Equation 1 was used to construct SAXS Zimm plots containing electron density contrast ( $\Delta\rho_e$ ) as an adjustable parameter

$$\frac{\phi_c V_c}{I(q, \phi_c)} = \left[ 1 + \frac{5}{9} \left( \frac{q d_z}{2} \right)^2 \right] \left[ \frac{1}{\Delta\rho_e^2} + \frac{8\bar{B}_2}{\Delta\rho_e^2} \phi_c \right] \quad (1)$$

The other parameters are defined in the original publication<sup>1</sup> with the important ones for this report being the Z-average diameter ( $d_z$ ) and the second virial coefficient ( $\overline{B}_2$ ).

The thermodynamic stability of nanoparticles in a polymer melt is essential to ensure good dispersion during the annealing process. Annealing is required before rheological and mechanical measurements are performed to achieve thermodynamically equilibrated states. Then, SAXS measurements were carried out at 25 °C, 170 °C, and 30 °C at the Advanced Photon Source (APS). No significant difference in the scattering profiles and thermodynamic parameters were found for all three measurement temperatures as shown in Figure 2 and Table 1. This demonstrates that quenching the nanocomposite from the annealing temperature (which is representative of the processing temperature if a product were being made) effectively freezes the thermodynamic state.

Here, the second virial coefficient was made dimensionless by normalizing with the hard sphere value. Positive  $B_2$  values indicate thermodynamic stability of magnetite nanoparticles in polystyrene which is critical to the success in making true nanocomposites.



**Figure 2:** Small angle X-ray scattering profiles of 223 kD PS -  $\text{Fe}_3\text{O}_4$  ( $d=10$  nm) nanocomposites at different measurement temperatures.

**Table 1.** Zimm plot results at different measurement temperatures for 223 kD PS- $\text{Fe}_3\text{O}_4$  ( $d=10$  nm) nanocomposites.

Measurement	$d_z$	$\Delta\rho_e \times 10^{11}$	$\overline{B}_2$	$\epsilon$
Temperature	(nm)	( $\text{cm}^{-2}$ )		( $k_B T$ )
25 °C	$28.7 \pm 0.4$	$2.35 \pm 0.01$	$0.0042 \pm 0.0002$	0.35
170 °C	$26.0 \pm 1.7$	$2.38 \pm 0.03$	$0.0046 \pm 0.0001$	0.35
30 °C	$27.5 \pm 2.7$	$2.37 \pm 0.04$	$0.0041 \pm 0.0001$	0.35

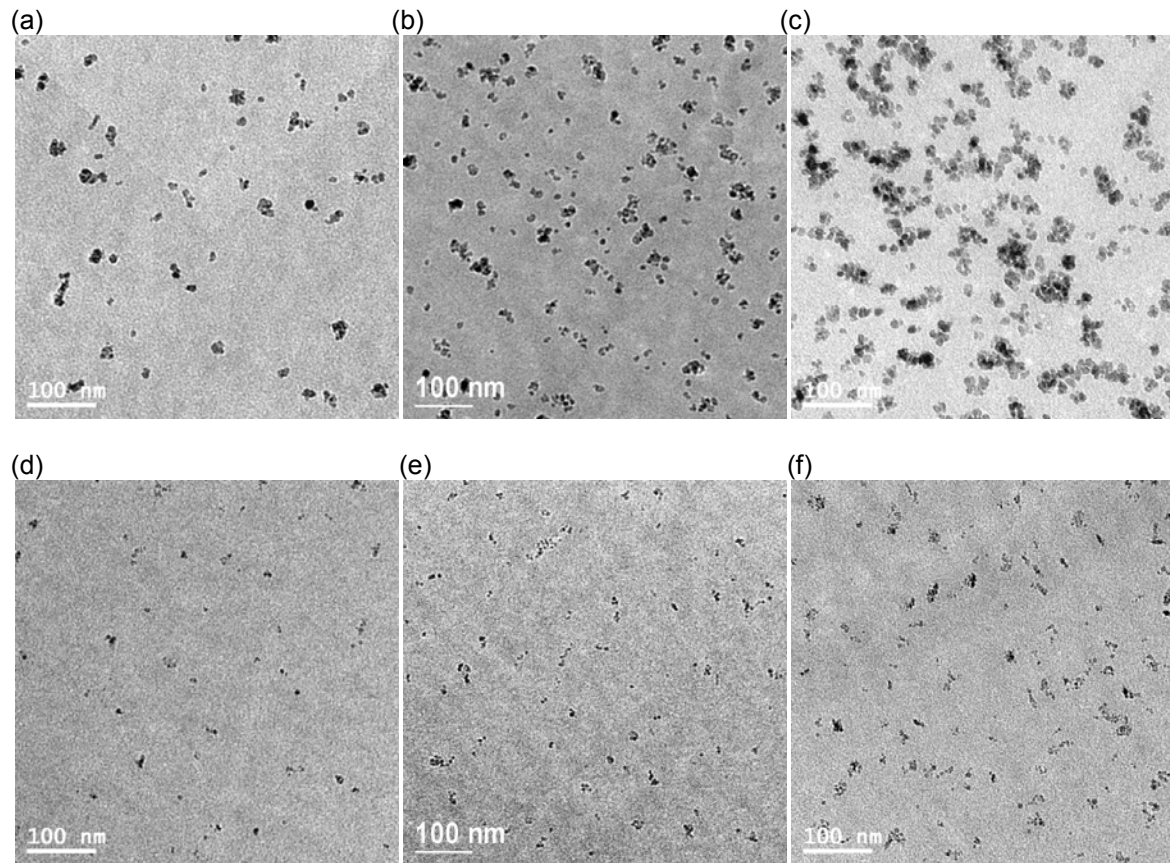
Since the ambient measurement conditions could capture the thermodynamic stability of the polymer melts, by quenching pre-annealed polymer nanocomposites, the rest of the systems in this study were measured at room temperature after quenching the pre-annealed nanocomposites using a Rigaku Ultima IV X-ray instrument. The Zimm plot results of 393 kD PS systems are summarized in Table 2.

**Table 2.** Zimm plot results measured at 25°C for 393 kD PS-Fe<sub>3</sub>O<sub>4</sub> (d=5 and 10 nm) nanocomposites.

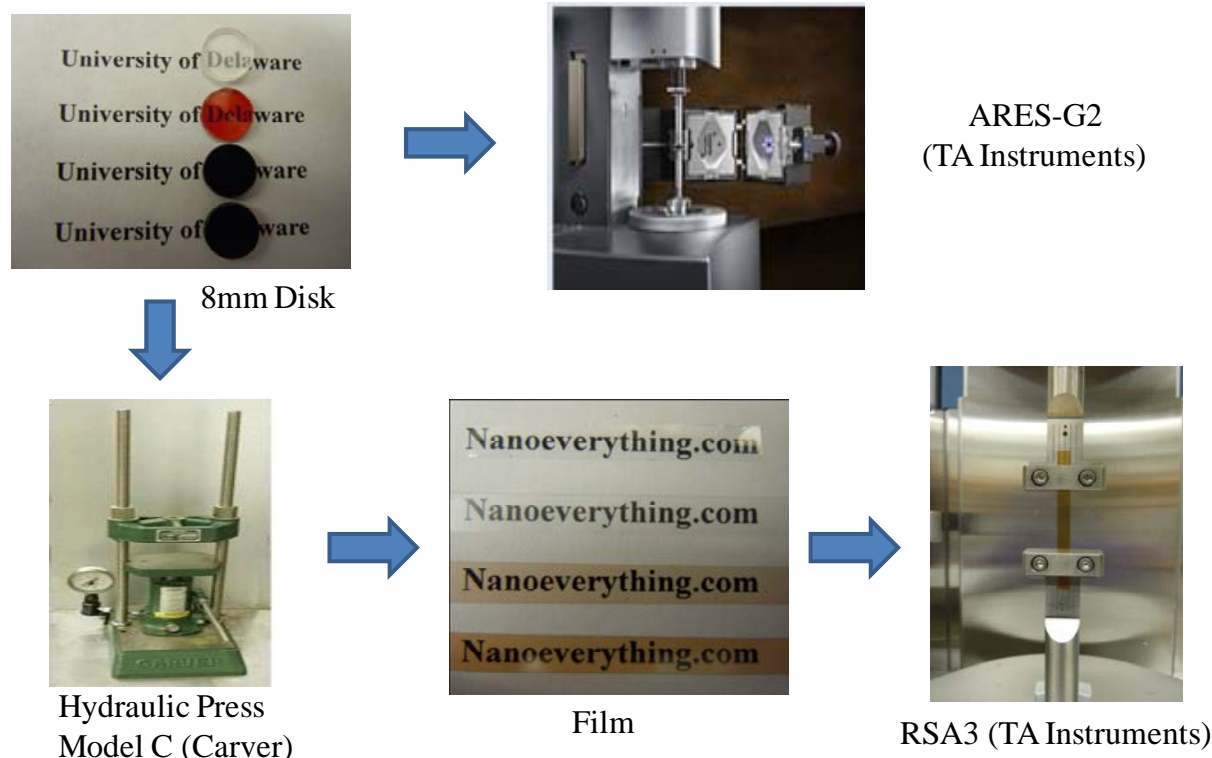
PS MW (kD)/ Fe <sub>3</sub> O <sub>4</sub> d (nm)	$d_{z,DLS}$ (nm)	$d_{z,SAXS}$ (nm)	$\Delta\rho_e \times 10^{11}$ ( $cm^{-2}$ )	$\overline{B_2}$	$\epsilon$ ( $k_B T$ )
393/5	$8.7 \pm 2.1$	$8.8 \pm 0.9$	$3.81 \pm 0.27$	$0.0258 \pm 0.0018$	0.34
393/10	$18.8 \pm 7.5$	$18.0 \pm 1.0$	$2.03 \pm 0.11$	$0.0583 \pm 0.0033$	0.33

The z-averaged diameter of the 393 kD-nanoparticle systems are almost equal to those extracted from dynamic light scattering experiments, suggesting this system exhibits minimal clustering and is thermodynamically stable.

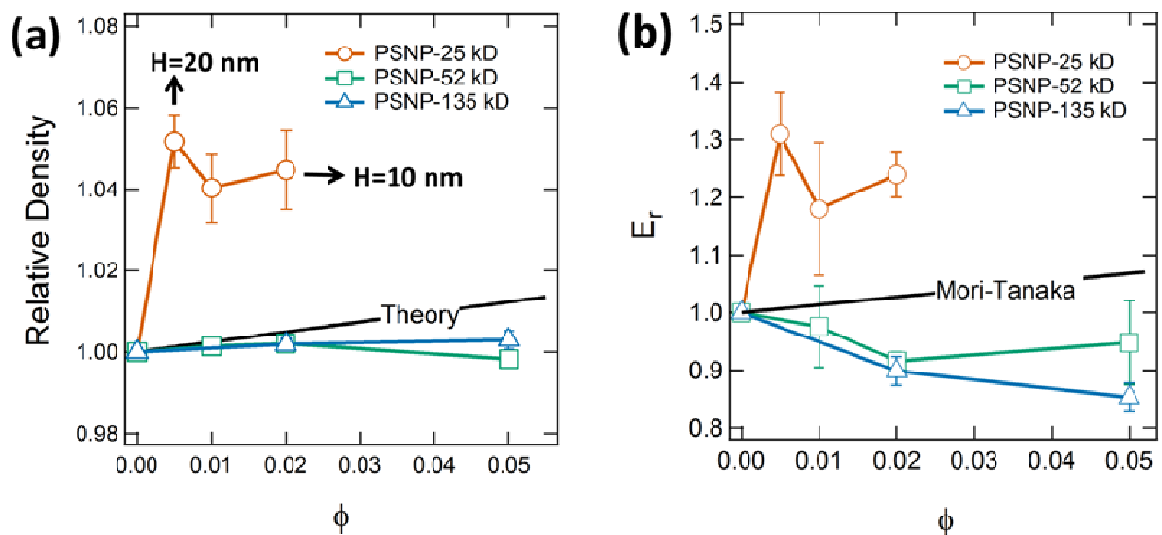
For reference, Transmission Electron Micrographs (TEMs) shown in Figure 3 also supported the finding from SAXS. Good dispersion was found for lower concentrations however a significant amount of clustering was found at higher nanoparticle concentrations.



**Figure 3:** 393 kD polystyrene nanocomposites with 10 and 5 nm iron oxide nanoparticles (a) 0.8 vol% Fe<sub>3</sub>O<sub>4</sub> (d=10 nm) (b) 2 vol% Fe<sub>3</sub>O<sub>4</sub> (d=10 nm) (c) 4 vol% Fe<sub>3</sub>O<sub>4</sub> (d=10 nm) (d) 0.2 vol% Fe<sub>3</sub>O<sub>4</sub> (d=5 nm) (e) 0.4 vol% Fe<sub>3</sub>O<sub>4</sub> (d=5 nm) (f) 0.8 vol% Fe<sub>3</sub>O<sub>4</sub> (d=5 nm)



**Figure 4:** Preparation of samples for melt rheological testing (ARES-G2) or tensile testing (RSA3).



**Figure 5:** (a) Relative density of three different molecular weights of PSNP systems accurately determined by a Helium gas pycnometer at room temperature ( $\sim 25^\circ\text{C}$ ). Theory indicates theoretical value calculated from the polymer and nanoparticle density information.  $H$  is interparticle distance between nanoparticles. (b) Relative tensile modulus of 393 kD PS nanocomposites against nanoparticle volume fraction. Mori-Tanaka indicates predicted values by a conventional continuum micromechanics model.

Disks for tensile testing were pressed to make films as shown in Figure 4. Results from the tensile testing experiments are shown in Figure 5. Nanoparticle volume fractions were corrected by measuring the density of polymer and nanoparticles. Here it is found that nanoparticles having a size of 3 – 5 nm provide the best reinforcement at low volume fraction. Our hypothesis is that as the size of the nanoparticle approaches the persistence length of the polymer (~ 3 nm for polystyrene) The polymer can not effectively pack around the particle. When this happens the polymer has density fluctuations within the bulk material.

We know the density fluctuations occur since we find the nanocomposites become more compressible, as described in the previous interim report, and an increase in compressibility means the formation of void region. The introduction of voidage without formation of high density regions for PS nanoparticle (PSNP) systems should produce a slight density reduction regardless of molecular weight since PSNPs have a density of 1.30 g/cc which is a little higher than linear polystyrene 1.05 g/cc. However, the results in Figure 5a shows the nanocomposite density remaining essentially unchanged for 52 kD and 135 kD PSNPs, suggesting that any voids had been compensated by the higher density regions. Remarkably, the inclusion of 25 kD PSNP resulted in significant densification, which is indicative of a large amount of high density regions and/or substantially high density for high density regions.

The relative tensile modulus results of 393 kD PS nanocomposites against nanoparticle volume fraction are presented in Figure 5b. The PSNP can yield a large reinforcement or a reduction depending on their molecular weight (25, 52, or 135 kD). General trend of tensile modulus was similar to density results shown in Figure 5a. In uniaxial tension, here we offer a continuum explanation of how it may be possible to have a tensile modulus increase. Consider first where there is no density fluctuations as shown in Figure 6a. There are three regions that operate in series and parallel to yield the following tensile modulus, E,

$$\frac{1}{E} = \frac{\phi_{m1}}{E_m} + \frac{\phi_2}{\phi_{m2}^* E_m + \phi_{NP}^* E_{NP}} + \frac{\phi_{m3}}{E_m} \quad (2)$$

Here,  $E_m$  and  $E_{NP}$  are the matrix and nanoparticle modulus, respectively. We note the volume fraction of the matrix in region 1 ( $\phi_{m1}$ ) is equal to that in region 3 ( $\phi_{m3}$ ) where

$$\phi_{m1} = \phi_{m3} = \frac{[H+d]^2 \frac{H}{2}}{[H+d]^3} = \frac{1}{2} \frac{H}{H+d} \quad (3)$$

The distance between the nanoparticle surfaces is given by

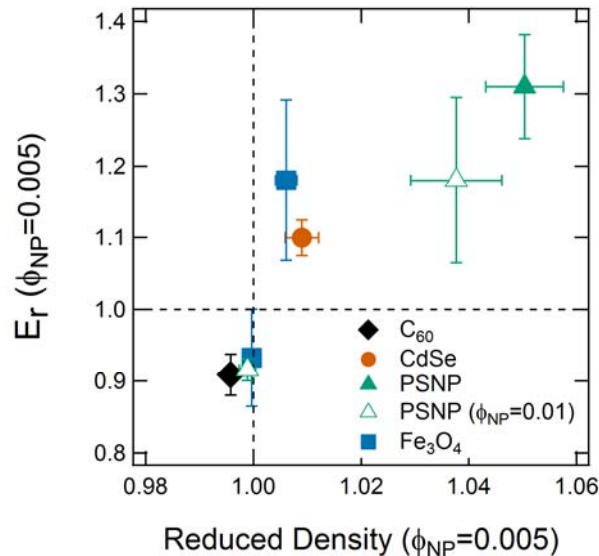
$$\frac{H}{d} = \left[ \frac{1}{\phi_{NP}} \right]^{1/3} - 1 \quad (4)$$

where  $\phi_{NP}$  is the nanoparticle volume fraction. One can now write

$$\phi_{m1} = \phi_{m3} = \frac{1}{2} [1 - \phi_{NP}^{1/3}] \quad (5)$$



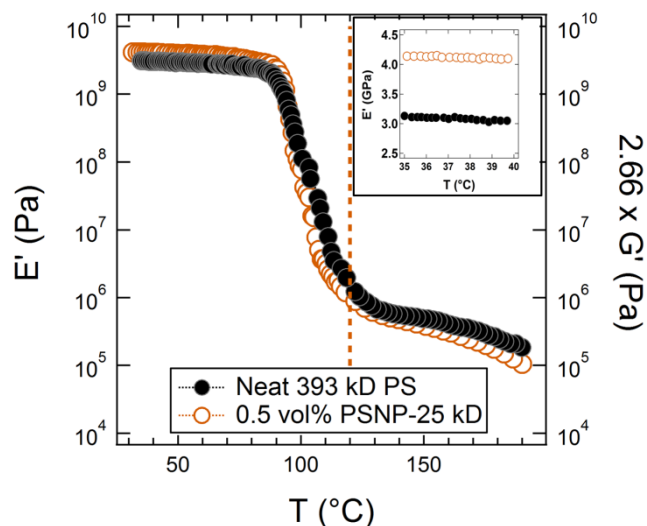




**Figure 8.** (a) The relative tensile modulus of 393 kD PS nanocomposites is plotted versus reduced density. The nanocomposites were prepared by the addition of various nanoparticles at  $\phi \sim 0.005$ . Nanoparticle diameters for  $C_{60}$ , CdSe, and PSNP were 0.76 nm, 3 nm, and 5 nm, respectively. Unfilled triangles are PSNPs at  $\phi = 0.01$  and the PSNPs next to  $C_{60}$  had molecular weights of 52 kD while both top right two PSNPs had molecular weight of 25 kD.  $Fe_3O_4$  nanoparticle next to  $C_{60}$  had 10 nm and one next to CdSe nanoparticle had 5 nm.

To further investigate the particle size effect on densification and tensile modulus reinforcement, different sized and type nanoparticles were studied in addition to PSNPs at a low nanoparticle volume fraction,  $\phi = 0.005$ . The relative tensile modulus was plotted against reduced density and shown in Figure 5.7. Here, the reduced density is normalized by the theoretical density from the regular rule of mixtures by knowing the density information of the nanoparticle and polymer. Thus, a reduced density larger than unity indicates densification of polymer nanocomposites.

Fullerene ( $C_{60}$ ) has a 0.76 nm diameter, which is hypothesized to be too small to perturb the polymer chain conformation and so little reinforcement or densification occurs. The iron oxide ( $Fe_3O_4$ ) nanoparticle data point next to  $C_{60}$  in the figure had a 10 nm diameter and was hypothesized to be too large to distort the polymer chain similar to what was found with the larger PSNPs discussed above. The data for the 52 kD PSNP with a diameter of 6.2 nm ( $\phi = 0.01$ ) was also included in Figure 8 for comparison and the data point is next to the  $C_{60}$  data point in the figure. Neither demonstrated densification nor reinforcement. Meanwhile, the 3 nm diameter cadmium selenide (CdSe) and 5 nm  $Fe_3O_4$  nanoparticles whose size is comparable to the polymer Kuhn length and both exhibited densification and reinforcement. Remarkably, the 5 nm diameter PSNP had the largest densification and reinforcement as Bedrov et al. reported maximized densification occurs at the athermal condition. [3] Although only a few nanoparticles having different chemical components were tested due to their limited availability, we have found a definite particle size dependent density and tensile modulus. The effect is subtle, yet, quite robust as various systems, inorganic or organic, produce the same phenomenon.



**Figure 9.** The inclusion of smallest PSNP enhanced mechanical properties in the glassy state while the melt viscosity decreased compared to neat PS. The  $E'$  is matched with  $G'$  at 120 °C with factor of 2.66 using the Poisson's ratio of 0.33 for PS as  $E=2G(1+\nu)$  where  $\nu$  is Poisson's ratio. The inset shows the magnified elastic storage modulus data in the glassy states.

We previously reported a melt viscosity decrease by the addition of PSNP's to linear PS matrices. Rheological characterization was performed with the nanocomposites to ensure whether the mechanical reinforcement in the glassy state is contradictory to the melt viscosity decrease. Polymer melt properties were probed using a torsional rheometer and matched with solid state properties determined with a dynamic mechanical analyzer (DMA) operating in tension. Measurements were taken at a frequency of 1 rad/s with both instruments. Compared to neat PS, an enhanced elastic storage modulus,  $E'$ , in tension was found by the inclusion of 0.5 vol% 25 kD PSNP in the glassy state (Figure 9). However,  $E'$  around the polymer glass transition temperature ( $T_g$ ) and shear storage modulus,  $G'$  above  $T_g$  of the nanocomposite became less than that of the pure polymer, suggesting a viscosity reduction in the liquid state. Therefore, reinforcement in the glassy state is not contradictory to previous findings of a melt viscosity decrease.

## References

1. Anderson, B.J. and Zukoski, C.F., *Nanoparticle stability in polymer melts as determined by particle second virial measurement*, *Macromolecules* **40** (2007) 5133-5140.
2. Garcia-Fierro, J.L. and Aleman, J.V., *Interactions between water and polystyrene*, *Euro Poly J* **21** (1985) 753-756.
3. Bedrov, D., G.D. Smith, and J.S. Smith, *Matrix-induced nanoparticle interactions in a polymer melt: A molecular dynamics simulation study*. *Journal of Chemical Physics*, 2003. **119**(19): p. 10438-10447.